SCIENTIFIC REPORTS

OPEN

Piezoelectric scattering limited mobility of hybrid organic-inorganic perovskites $\text{CH}_3\text{NH}_3\text{PbI}_3$

Ying-Bo Lu^{1,2}, Xianghua Kong², Xiaobin Chen², David G. Cooke² & Hong Guo^{2,3}

Received: 03 October 2016

Accepted: 29 December 2016

Published: 02 February 2017

Carrier mobility is one of the most important parameters for semiconducting materials and their use in optoelectronic devices. Here we report a systematic first principles analysis of the acoustic phonon scattering mechanism that limits the mobility of $\text{CH}_3\text{NH}_3\text{PbI}_3$ (MAPbI₃) perovskites. Due to the unique hybrid organic-inorganic structure, the mechanical, electronic and transport properties are dominated by the same factor, *i.e.* the weak interatomic bond and the easy rotation of methylammonium (MA) molecules under strain. Both factors make MAPbI₃ soft. Rotation of MA molecule induces a transverse shift between Pb and I atoms, resulting in a very low deformation potential and a strong piezoelectricity in MAPbI₃. Hence the carrier mobility of pristine MAPbI₃ is limited by the piezoelectric scattering, which is consistent to the form of its temperature dependence. Our calculations suggest that in the pristine limit, a high mobility of about several thousand $\text{cm}^2 \text{V}^{-1} \text{S}^{-1}$ is expected for MAPbI₃.

Hybrid organic-inorganic perovskites (HOIPs) have been known for a long time, but there has been a recent explosion of interest in these materials due to their exceptional performance as the active material in a solar cell. Within five years, the power conversion efficiency (PCE) of HOIPs solar cell has increased from 3.8% to 22.1%^{1–5}. The excellent performance mainly originates from the hybrid organic-inorganic structure which possesses favorable characteristics such as high electrical mobility, band gap tunability, excellent optical absorption and low fabrication cost⁶. Listed among the several favourable properties amenable to optoelectronic devices, the excellent transport properties of HOIPs are partly responsible for the large PCEs, which combined with their long carrier lifetimes, governs the large diffusion lengths characteristic of these materials^{7,8}. Charge carrier mobilities in high quality polycrystalline films are consistently found in the 10–60 $\text{cm}^2 \text{V}^{-1} \text{S}^{-1}$ range, determined by time-resolved THz spectroscopy (TRTS)^{9–11}, time-resolved microwave¹² and Hall measurements^{13,14}. For large single crystal samples, Stoumpos *et al.* reported¹⁵ that MASnI_3 possesses a mobility exceeding 2000 $\text{cm}^2 \text{V}^{-1} \text{S}^{-1}$; and more recently using ultra-broadband TRTS Cooke *et al.* measured a pure Drude response and a remarkably high transient mobility of 500–800 $\text{cm}^2 \text{V}^{-1} \text{S}^{-1}$ on picosecond time scales for methylammonium (MA) lead iodide ($\text{CH}_3\text{NH}_3\text{PbI}_3$ or MAPbI₃) materials¹⁶. These and some other reported mobilities of HOIPs are tabulated in Table S1 in the supplementary material. We note that these reported high mobilities are comparable to that of traditional semiconductors, *e.g.* GaN and AlN¹⁷.

The most intensively investigated HOIPs to date is MAPbI₃, however the intrinsic limitations of electrical transport in this material are still under experimental and theoretical investigation. For most semiconductors, various scattering mechanisms between charge carriers and phonons or impurities limit their transport properties. For example, from the broadening of the photoluminescence linewidth of MAPbI₃, a polar optical phonon limited carrier mobility was proposed^{14,18–20}. However, since MAPbI₃ is a light-sensitive material, several effects including photostriction²¹, degradation due to above-band-gap illumination^{22,23}, band structure variation caused by spin-orbital coupling (SOC)²⁴ etc., influence the measured photoluminescence lineshape. It is also difficult to separate competing contributions from acoustic phonons and optical phonons. For MAPbI₃, several experiments reported^{11,14,25–27} a temperature dependence of the carrier mobility to be $\mu \propto T^{-1.5}$, which could not be easily explained if the scattering mechanism is due solely to the polar optical phonon. On the other hand, such a temperature dependence suggested a scattering mechanism due to deformation potentials, *i.e.* one of the acoustic phonon scattering mechanisms, because a $T^{-1.5}$ factor does appear in the mobility formula^{28–32}. Nevertheless, in the same mobility formula, the carrier effective mass m^* may also be temperature dependent, for instance

¹School of Space Science and Physics, Shandong University, Weihai 264209, China. ²Department of Physics, McGill University, Montreal, QC H3A 2T8, Canada. ³School of Physics and Energy, Shenzhen University, Shenzhen 518060, China. Correspondence and requests for materials should be addressed to Y.-B.L. (email: lyb@sdu.edu.cn)

Figure 1. (a) Top view and (b) side view of the geometric structure of tetragonal MAPbI₃. The yellow, brown, blue, grey and white balls represent Pb, I, N, C and H atoms, respectively. α , β , γ and δ in (a,b) denote angles of I-Pb-I, which show the distortion of PbI₆ octahedra.

$m^* \propto T^{0.4}$ in lead chalcogenides^{33,34}. Therefore if this temperature dependence were also true for MAPbI₃, the mobility governed by deformation potential would become $\mu \propto T^{-1.5}/(m^*)^{2.5} \propto T^{-2.5}$, which is markedly different from the reported experimental data^{11,14,25–27} of $\mu \propto T^{-1.5}$. Most importantly, in fact some smaller exponents were also observed experimentally¹¹, $\mu \propto [T^{-1.2}, T^{-1.4}]$. Therefore it appears that the deformation potential scattering mechanism alone cannot explain the experimentally observed temperature dependence of MAPbI₃ mobility. In fact, the mobility measured in optically pumped samples may be affected by electron-hole scattering to alter the temperature dependence^{11,25,26,35}. With these and other complications, so far no clear consensus has emerged concerning the dominating scattering mechanism that limits the carrier mobility of pristine MAPbI₃.

Interestingly and known experimentally, MAPbI₃ with tetragonal symmetry (I4/mcm) is ferroelectric and piezoelectric, and a light-enhanced piezoelectricity has been observed^{36–38}. Theoretically, Frost *et al.* predicted a large polarization value of 38 $\mu\text{C}/\text{cm}^2$ for MAPbI₃ materials³⁹. The piezoelectric scattering - a type of the acoustic phonon scattering, is also suggested to play a dominant role for mobilities of certain semiconductors such as GaN with large, built-in internal electric fields^{40,41}. An important and interesting question therefore arises: what is the role of piezoelectric scattering in hybrid organic-inorganic perovskites MAPbI₃? It is the purpose of this work to investigate this issue theoretically.

We focus on elucidating the role of piezoelectric scattering to the carrier mobility of MAPbI₃, using first principles techniques. The mechanical, electronic and transport properties of this material are found to be related and heavily influenced by the weak interatomic bonds and the easy rotation of MA molecules under strain. The acoustic phonons do not manifest deformation potential scattering, but bring about a strong polarized electric field which gives piezoelectric scattering as the dominant mobility limiting scattering mechanism in pristine MAPbI₃. We also explain the observed temperature dependence of mobility through the piezoelectric scattering mechanism.

Results

Geometric structure of tetragonal MAPbI₃. As illustrated in Fig. 1, the tetragonal MAPbI₃ is constituted by networks of corner-sharing PbI₆ octahedra, with organic MA molecules fitting into small cuboctahedral holes. MA molecules provide two important roles to MAPbI₃: maintaining charge neutrality and stabilizing the perovskite structure^{42,43}. Since the MA molecule is dipolar, the crystal structure and thus the electronic structure of MAPbI₃ is extremely sensitive to the orientation of the MA molecule. We have investigated many MAPbI₃ configurations with MA molecules arranged in different patterns (shown in Fig. S1 in supplementary materials) using density functional theory (DFT) total energy method^{43–45}, and find that the most energetically stable structure is one where all the MA molecules are oriented along the [110] direction but arranged perpendicular to each other, as shown in Fig. 1. Because the bonding interactions at two radicals of MA molecules are different, obvious tilting of PbI₆ octahedrons and out-of-plane rotations of MA molecules appear. We therefore use this configuration as the standard model structure of tetragonal MAPbI₃ in the rest of this work unless explicitly specified.

Electronic structure of tetragonal MAPbI₃. Band structures and band gaps of MAPbI₃ calculated by various strategies with DFT are shown in Fig. S2 and Table S2 in supplementary materials, respectively. The band gap calculated by DFT with the generalized gradient approximation (GGA) is 1.552 eV. The inclusion of HSE functional and/or spin-orbit coupling (SOC) interaction provides corresponding gap corrections, and the result agrees well with the experimental value⁴⁶. Another key parameter with respect to transport is the effective mass of carriers (m^*) that can be obtained by parabolic band fitting around the Γ point in the Brillouin zone^{47,48}. The effective carrier mass tensors obtained by various approaches are listed in Table S2 in supplementary materials.

Carrier	Electrons	Holes
Ξ (eV)	0.764	2.319
Ξ_x (eV)	0.467	1.657
Ξ_y (eV)	0.707	1.471
Ξ_z (eV)	1.368	3.900
B (eV/Å ³)	0.110	
C_x (eV/Å ³)	0.173	
C_y (eV/Å ³)	0.213	
C_z (eV/Å ³)	0.294	
μ (cm ² V ⁻¹ s ⁻¹)	0.303×10^6	0.193×10^5
μ_x (cm ² V ⁻¹ s ⁻¹)	0.838×10^6	0.520×10^5
μ_y (cm ² V ⁻¹ s ⁻¹)	0.379×10^6	0.789×10^5
μ_z (cm ² V ⁻¹ s ⁻¹)	0.636×10^6	0.247×10^5

Table 1. Calculated mobilities via formula described by the deformation potential scattering mechanism and corresponding parameters for MAPbI₃.

The inclusion of SOC reduces m^* significantly, especially for electrons, m_e^* decreases from the DFT-GGA value of $\sim 0.900 m_e$ to the DFT-SOC value of $\sim 0.160 m_e$, where m_e is the bare electron mass. This is reasonable because SOC impacts only on heavy elements. The introduction of SOC in the calculation therefore strongly influences the dispersion near the conduction band minimum (CBM) of MAPbI₃, which is governed by the Pb-6p orbitals^{49,50}. Nevertheless, compared with DFT/DFT-SOC functionals, the more advanced HSE/HSE-SOC approaches do not introduce a sizeable change to the effective mass of both electrons and holes. Hereafter in the discussions below, we use the effective mass obtained from DFT-SOC since these values are close to the obtained experimental values⁵¹.

A knowledge of the carrier effective mass is necessary to estimate the charge carrier mobility, however more information is required. Next, we investigate how carriers interact with the host material when they traverse the lattice. In general, the intrinsic mobility is limited by a variety of scattering mechanisms, including impurity scattering, boundary scattering, phonon scattering, etc. Previous work has suggested impurity scattering plays less of a role in limiting the observed MAPbI₃ mobility^{6,11,44,45,52}. The experimentally observed temperature dependence of the mobility was predominantly attributed to acoustic phonon scattering¹⁷ which is composed of deformation potential scattering and piezoelectric scattering. In the following we investigate these scattering mechanisms individually.

Deformation potential scattering. The acoustic phonons induce lattice vibration which results in the periodic shift of band edges. These potential shifts act as a source of carrier scattering, called deformation potential scattering. By Eq. 1 in the Method Section below, we calculate the mobility μ_{dp} of tetragonal MAPbI₃ due to the deformation potential scattering mechanism where the temperature is set at 300 K. The results are listed in Table 1 where the mobility for electrons $\mu_e = 0.303 \times 10^6 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, and for holes $\mu_h = 0.193 \times 10^5 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. As tetragonal MAPbI₃ is not an isotropic material, we calculate mobilities projected into three lattice axis using Eq. 2 in the Method Section and list them in Table 1. We find that the calculated μ_{dp} of MAPbI₃ is extremely large, much larger than the highest experimental value of $800 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. This is caused by two factors: the small carrier effective mass m^* and the low deformation potential Ξ of the material. Particularly, a low deformation potential means the scattering of acoustic phonons with carriers is rather weak, resulting in a relatively large mobility. We conclude that such a very weak deformation potential is likely not the limiting factor to the observed carrier mobilities of MAPbI₃.

Piezoelectric scattering. Acoustic phonon scattering is usually described by the deformation potential which, as we showed above, does not appear to provide the limiting scattering mechanism for the mobility of MAPbI₃. However for certain semiconductors, namely for non-centrosymmetric crystals, the elastic strain can create macroscopic electric fields that induce a coupling between acoustic waves and free carriers, known as piezoelectric scattering. In semiconductors with high piezoelectric coefficient such as GaN, ZnO etc., piezoelectric scattering actually dominates over the deformation potential scattering^{40,41}. The tetragonal MAPbI₃ belongs to the space group I4/mcm and has a strong piezoelectric response^{38,39,53}. Normally the piezoelectric strain coefficient d_{zz} of MAPbI₃ is $\sim 5\text{--}6 \text{ pm/V}$, but the light-enhanced piezoelectric coefficient dramatically increases up to 25 pm/V ^{36,37}. Therefore the contribution of piezoelectric scattering to carrier mobility in MAPbI₃ cannot be ignored.

Table 2 summarizes the piezoelectric stress coefficients e_{15} , e_{31} and e_{33} obtained by our first principles calculation, together with the elastic constant tensor adopted from experiments⁵⁴. Putting these parameters into Eq. 3 in the Method Section, we obtain the piezoelectric scattering limited mobility μ_{pe} , as listed in Table 2. μ_{pe} for electrons and holes are found to be $11294 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ and $8368 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, respectively. The piezoelectric scattering limited mobility μ_{pe} shows two interesting characters. First, the hole mobility is smaller than that of the electron, due to the larger effective mass of the hole. Second, the mobility obtained from piezoelectric scattering is significantly lower than that obtained by the deformation potential scattering: for electron $\mu_{pe} < \mu_{dp}$ dramatically by two orders of magnitude. We conclude that for carriers in MAPbI₃, piezoelectric scattering dominates over the deformation potential scattering.

Parameters	Values	Results	Values
$C_{11}(\text{GPa})$	20.100	$m_e^*(m_e)$	0.132
$C_{12}(\text{GPa})$	14.600	$m_h^*(m_e)$	0.161
$C_{13}(\text{GPa})$	6.800	$C_i(\text{GPa})$	17.234
$C_{33}(\text{GPa})$	17.900	$C_i(\text{GPa})$	3.567
$C_{44}(\text{GPa})$	1.600	$(e_i^2)_{av} (C^2/m^4)$	0.006
$C_{66}(\text{GPa})$	9.200	$(e_i^2)_{av} (C^2/m^4)$	0.026
$e_{13}(\text{C/m}^2)$	-0.465	$\mu_e(\text{cm}^2 \text{V}^{-1} \text{s}^{-1})$	11294
$e_{15}(\text{C/m}^2)$	0.066	$\mu_h(\text{cm}^2 \text{V}^{-1} \text{s}^{-1})$	8368
$e_{33}(\text{C/m}^2)$	0.170		

Table 2. Calculated mobilities limited by piezoelectric scattering mechanism and the corresponding parameters for MAPbI₃. The elastic constants tensors are employed from ref. 54 directly.

Discussions

It is quite surprising that piezoelectric scattering, rather than deformation potential scattering, is a stronger influence on the mobility of MAPbI₃. Note that the piezoelectric scattering mechanism for the hybrid organohalide Pb-based material has not been considered previously, and some discussions are in order. In particular there are two issues to address: why the deformation potential scattering gives a very high mobility in MAPbI₃, and why piezoelectric scattering plays a critical role. While the small effective mass already points to a large mobility, the form of the deformation potential arising from the unique geometric structure of MAPbI₃ is key to this result and is discussed below.

All existing experimental and theoretical investigations have revealed the extremely soft nature of the HOIPs materials^{54–56}. For example, MAPbI₃ is known to have good ductility and its bulk modulus is only 10–20 GPa. The soft nature is attributed to two reasons: the weak bonding interaction and the easy rotation of the MA molecules under strain. For the optimized MAPbI₃ structure obtained in our calculation, the Pb-I bond lengths, the C-I bond lengths and N-I bond lengths are found to be 3.14–3.22 Å, 3.84–4.00 Å, and 3.61–3.84 Å, respectively. These values are much longer than the bond lengths of traditional semiconductors such as Si, Ge and In₂O₃, etc.⁵⁷. The longer bond length implies a relatively weak interatomic bonds⁵⁸ and thus weak resistance to elastic deformation. Therefore a small elastic constant and a soft character of the MAPbI₃ material are expected.

On the other hand, the influence on mechanical properties by rotations of the MA molecule is seldom considered, even though it is discussed in studies of other physical properties. The weak interatomic bonds between Pb-I and the hydrogen bonding between MA molecules and inorganic units, give rise to a very low rotational barrier for the MA molecules under external strain and/or intrinsic lattice disturbances. The calculated rotation angles of MA molecules are tabulated in Table S3 in supplementary materials, where θ_z , θ_y represent angles between the C-N bond direction and the *xy*-plane or *xz*-plane, respectively. Variations of θ_z and θ_y show that when the strain in the *x*-axis changes from tension to compression, the C-N bond direction (*i.e.*, the direction of the MA molecule) tends to rotate toward the *z*-axis and *y*-axis, respectively. Furthermore, if the strain is along *y*-axis and *z*-axis, the MA molecules also rotate towards the *xz*- and *xy*-planes, respectively. These findings mean that if the MAPbI₃ lattice is subjected to a strain along one specific direction, the MA molecules will rotate toward other two perpendicular directions, which releases resistance to the external strain. Therefore, for a given strain, with MA molecular rotation the MAPbI₃ lattice changes more significantly than without the rotation, contributing further to a low elastic constant and soft nature of the material.

Coincidentally, the combination of weak interatomic bonds and rotations of the MA molecules in MAPbI₃ are the exact reason why MAPbI₃ possesses a rather low deformation potential. The conduction band minimum (CBM) of MAPbI₃ is the anti-bonding state formed by the Pb-6p and I-5s orbitals, while the valence band maximum (VBM) is the anti-bonding states composed of the Pb-6s and I-5p orbitals, as illustrated in Fig. 2(a). Generally speaking, if the volumetric variation only leads to compression (dilation) of the bond length, both CBM and VBM of MAPbI₃ would shift upward (downward), due to the enhanced (weakened) repulsion between the s and p orbitals, as plotted in Fig. 2(b). The deformation potential is inversely proportional to the energy difference between the two constituent atomic orbitals making up the anti-bonding states⁵⁹. A higher deformation potential of the VBM than that of the CBM is obtained due to the relatively smaller energy difference between I-5p and Pb-6s compared to that of I-5s and Pb-6p.

Besides the change of bond length, lattice strain changes the inorganic PbI₆ networks to adapt to the rotation of the MA molecules. We use the I-Pb-I angle to illustrate the distortion of PbI₆ octahedra, where α , β , γ in Fig. 1(a) represent the in-plane tilting angles, and δ in Fig. 1(b) denotes the out-of-plane tilting angle. Data in Table S3 in supplementary materials show that with the increase of strain (from compression to tension), all angles α , β , γ and δ decrease, indicating the distortion of PbI₆ octahedron is reinforced. It also means that the relative transverse shift between Pb and I atoms inside the PbI₆ octahedron is enlarged. In ref. 60, Kim *et al.* demonstrated that the relative transverse displacement between Pb and I atoms raises the band edge levels, because the coupling between Pb-6p and I-5p changes from non-bonding states to anti-bonding states. Figure 2(c,d) illustrate the projected density of states (PDOS) of MAPbI₃ before and after the transverse atomic displacement. We can clearly find that after the atomic shift, both intensities of I-5p states in CBM and Pb-6p states in VBM increase, indicating that the anti-bonding interactions are strengthened for both CBM and VBM. Hence, as aforementioned, the increase of strain (volumetric expansion) leads to an increase of Pb-I bond length as well as a transverse shift between Pb and I atoms. The former factor shifts both CBM and VBM downward, while the latter raises them. In

Figure 2. (a) PDOS of MAPbI₃ materials and the inset represents the schematic diagram for the coupling interaction near band edges. (b) Energy shifts of VBM and CBM caused by the external strain with the presence and absence of rotation of MA molecule, respectively. (c,d) Are PDOS of MAPbI₃ without and with the distortion of PbI₆ octahedra, respectively. Insets in (c,d) illustrate the standard PbI₆ octahedron and tilting PbI₆ octahedron, respectively.

some sense, these two conflicting factors “cancel” as far as the variation of CBM and VBM is concerned, namely their deformation potentials take small values.

To further examine effects induced by distortion of PbI₆ octahedron and transverse shift between Pb and I atoms, we adopt another approach where we only change the Pb-I bond length while preventing the MA molecules from rotating and the transverse shift between Pb and I atoms to change. The calculated deformation potentials of MAPbI₃ by this approach are also illustrated in Fig. 2(b) which are around 8 eV and 10 eV for CBM and VBM, respectively. The deformation potentials without the atomic transverse shift are much larger than the value with atomic shift, revealing that the relatively transverse shift between Pb and I atoms is one of the key reasons to reduce the deformation potential in MAPbI₃. Nevertheless, we should note that in comparison with other materials such as GaN and PbS, the deformation potentials of MAPbI₃ calculated without atomic shift still take lower values even though they are 8.0–10.0 eV³³, arising from the weak interatomic bonding interaction. We conclude that the combined effect of the weak bonding interaction and the atomic transverse shift inside PbI₆ octahedra, produces a low deformation potential. As a result, the deformation potential scattering mechanism plays only a minor role to the carrier mobility of MAPbI₃.

As discussed above, MAPbI₃ exhibits pronounced piezoelectricity, due to a weak bonding interaction and rotation of the MA molecules. The MA molecule has a permanent dipole due to the imbalance of charge distribution between the methyl radical and the ammonia radical, and the piezoelectric effect of MAPbI₃ has been attributed to the dipole variation³⁶. Note that the role of the MA molecule includes not only its own dipole, but also the transverse shift between the Pb and I atoms. The latter enhances the separation between the positive charge center (*i.e.*, Pb atom) and the negative charge center (*i.e.*, I atom) inside the PbI₆ octahedra. This provides another contribution to the piezoelectric response of MAPbI₃. Our calculated Born effective charges of Pb, I, C and N atoms are 4.693 *e*, −2.624 *e*, 0.314 *e*, and −0.620 *e*, respectively. The latter two values produce a dipole on the MA molecule roughly equals ~2.250 Debye. When the strain upon *z*-axis changes from −0.030 to +0.030, the average change of separations between the Pb and I atoms is nearly 0.04 Å, hence the distortion of PbI₆ octahedron gives a change of polarization of about 1.246 μC/cm². The detailed piezoelectric stress coefficients are given in Table 2. The piezoelectric response in MAPbI₃ means that a polarization field can be induced by lattice strain: the strain can be applied externally or induced internally by acoustic phonons. Carriers then interact with the polarization field and are scattered during the transport process. The strong piezoelectricity of MAPbI₃ gives rise to pronounced piezoelectric scattering, thus a much lower carrier mobility than that produced by the deformation potential scattering.

The temperature dependence of mobility can be used as an indicator for elucidating the scattering mechanism. For the mobility of MAPbI₃, several experimental studies reported^{11,25–27} a temperature dependence of $\mu \propto T^{-1.2} - T^{-1.6}$. La-o-vorakiat *et al.*¹⁹ reported that the carrier relaxation time follows a power law of $\tau \propto T^{-0.42}$.

If the effective mass m^* were temperature independent, the deformation potential scattering would indeed give $\mu \propto T^{-1.5}$ through Eq. 1 - in the range of the reported experimental temperature power of 1.2–1.6. However, for many semiconductors, the nonparabolicity effect in the band edge and thermal expansion in the material cause m^* to vary significantly with temperature. For instance, $m^* \propto T^{0.4}$ has been well-known for Si, PbSe and PbS^{34,61,62}. Hence for these materials, the mobility governed by deformation potential scattering is $\mu \propto 1/[(m^*)^{2.5} T^{1.5}] \propto T^{-2.5}$ (see Eq. 1), in excellent consistency with the experimentally measured power law of $\mu \propto T^{-2.20}$ for Si⁶³; $\mu \propto T^{-2.55}$ for PbSe^{33,34}; and $\mu \propto T^{-2.34}$ for PbS. Therefore, if MAPbI₃ were to have a temperature dependence of its effective mass, the deformation potential scattering would not necessarily yield a power law of $\mu \propto T^{-1.5}$.

Next, let's consider piezoelectric scattering where the temperature dependence of the mobility is given by Eq. 3, i.e. $\mu \propto 1/[(m^*)^{1.5} T^{0.5}]$. For MAPbI₃, though there is still no experimentally reported temperature dependence of m^* , we may employ $m^* \propto T^{0.4}$ - known for lead dihalides¹⁷ which have similar mechanical properties as MAPbI₃, and this gives a power law of $\mu \propto T^{-1.1}$. This exponent is rather close to the experimental reported $\mu \propto [T^{-1.2}, T^{-1.6}]$ power law¹¹. In addition, the carrier relaxation time deduced from piezoelectric scattering follows $\tau \propto T^{-0.50}$, consistent with the experimentally observed¹⁹ $\tau \propto T^{-0.42}$. We conclude that the experimentally observed temperature dependence of the mobility of MAPbI₃ appears to be most consistent with the piezoelectric scattering.

Note that the calculated mobility value as limited by the piezoelectric scattering for MAPbI₃ is still about a factor of ten higher than the so far reported experimental values. This difference can be due to several reasons. First, a light-enhanced piezoelectricity of MAPbI₃ was reported experimentally³⁶ where a rapid increase of piezoelectric coefficient under illumination was detected, and this effect could increase the square of piezoelectric stress constant in Eq. 3 by a few times so that the mobility is reduced. Second, in real materials there can be scattering by grain boundaries (GB) to reduce the observed mobility^{12,27}. GB scattering is very complicated but has been well studied in the semiconductor literature⁶⁴. If the carriers are trapped by GBs, one expects an exponential temperature dependence of transport due to the activation energy. On the other hand, carriers can also quantum tunnel through the GBs⁶⁴ which is a process independent of temperature, and GB tunneling will clearly reduce the measured mobility. In this scenario, the temperature dependence of the mobility would be that of the next dominant scattering mechanism, which our results suggest to be the piezoelectric scattering. Indeed, it has been reported experimentally that GB scattering plays an important role in MAPbI₃ films^{12,27}. Therefore, the calculated piezoelectric scattering limited carrier mobility of about 8000–10000 cm² V⁻¹ s⁻¹ can be taken as the upper limit for pristine bulk MAPbI₃.

Summary. We have carried out systematic first principles calculations to determine the possible microscopic acoustic phonon scattering mechanism that limits the carrier mobility of perovskites MAPbI₃. Our calculations reveal a very close relationship between mechanical, electronic and transport properties of these materials. The soft nature of MAPbI₃ originates from the weak Pb-I bond/MA-I bond and the easy rotation of MA molecules, indicating that the structural fluctuations and dynamical disorder could emerge easily in MAPbI₃. The weak Pb-I bond and the rotation of MA molecule also produce a transverse shift between Pb and I atoms inside PbI₆ octahedron under lattice strain, which in turn raises the anti-bonding states near the band edge and reduces the deformation potential of MAPbI₃ to a very low level.

On the other hand, the rotation of MA molecule changes its dipole. The transverse shift between Pb and I atoms separates the positive and negative charge centers. Both factors give rise to the strong piezoelectricity of MAPbI₃. Therefore, our calculations suggest that the main effect of acoustic phonons in MAPbI₃ is to generate a strong piezoelectric scattering. By considering the temperature dependence of the effective mass, we argue that the temperature dependence of mobility - as limited by the piezoelectric scattering, is $\mu \propto T^{-1.1}$, which is close to the reported experimental power law of $T^{-1.2}$ to $T^{-1.6}$. The calculated piezoelectric scattering limited mobility of about 8000–10000 cm² V⁻¹ s⁻¹ can be taken as an upper limit of this important transport parameter, and the high value makes MAPbI₃ a very promising material for solar cell applications. Finally, we note that the complexity of hybrid organic-inorganic structure of MAPbI₃ as well as other perovskite materials may well lead to coexistence of various scattering mechanisms. In general, the dominant scattering mechanism that limits carrier mobility may be revealed by the calculated mobility value combined with other information such as its temperature dependence, as reported here.

Methods

DFT calculations. First principle calculations were performed via density functional theory (DFT) using the Vienna *Ab initio* Simulation Package (VASP)⁶⁵. The exchange-correlation is treated at the generalized gradient approximation (GGA) level with the PBE method, as well as at the hybrid functional level with the HSE method⁶⁶. In particular, HSE method gives the correct band gap. Projector augmented wave method is used to expand the electronic wave functions. The cutoff kinetic energy is set to 400 eV. A 3 × 3 × 3 Monkhorst-Pack k-mesh is used to integrate the first Brillouin zone. To optimize the supercell for tetragonal MAPbI₃, internal positions of atoms and lattice parameters are relaxed until the residual force and total energy difference between the self-consistent steps are less than 0.01 eV/Å and 10⁻⁶ eV, respectively. When calculating the deformation potential under strain, only internal positions of atoms are relaxed. In our calculations, all structures are optimized with optb88-vdW functional⁴⁹ to account for the van der Waals forces. Because there are heavy elements in the supercell, spin-orbit coupling effects are taken into account during the calculation of electronic properties⁴⁹.

Deformation potential scattering. The mobility limited by deformation potential scattering is calculated by the following equation^{28,29,33}:

$$\mu_{dp} = \frac{2\sqrt{2\pi}}{3} \frac{e\hbar^4 B}{m_f^* (m_b^* k_B T)^{\frac{3}{2}} \Xi^2} \quad (1)$$

where m_f^* is the conductivity effective mass and m_b^* is the density of states effective mass, B is the bulk modulus $B = V_0(\partial^2 E/\partial V^2)$, k_B is the Boltzmann constant, $\hbar$ is the reduced Planck's constant, e the elementary charge and T the temperature. Ξ is the deformation potential coefficient that describes the change in energy of the band edge under elastic deformation, $\Xi = (\Delta E_{VBM/CBM})/(\Delta V/V_0)$. The mobility projected on three axis is calculated by⁶⁷:

$$\mu_{dp}^\beta = \frac{2\sqrt{2\pi}}{3} \frac{e\hbar^4 C_\beta}{(m_\beta^*)^{\frac{5}{2}} (k_B T)^{\frac{3}{2}} (\Xi_\beta)^2} \quad (2)$$

where $\beta = x, y, z$. C_β is the elastic modulus along direction β , and is derived from $(E - E_0)/V_0 = C_\beta(\Delta l/l_0)^2/2$, here E and E_0 are total energies of systems under strain and equilibrium, respectively. V_0 is the lattice volume at equilibrium. $\Xi_\beta = (\Delta E_\beta)/(\Delta l/l_0)$ represents the deformation potential coefficient along the transport path, where ΔE_β is the energy variation of band edges owing to the compression and dilatation along the direction β , l_0 is the equilibrium lattice constant in the transport direction β and Δl is the variation of l_0 .

Piezoelectric scattering. The piezoelectric scattering gives the mobility via the following expression^{40,41,53,68}:

$$\mu_{pe} = \frac{16\sqrt{2\pi}}{3} \frac{\hbar^2}{e(m^*)^{\frac{3}{2}} (k_B T)^{\frac{1}{2}}} \left\{ \sum_{modes} \left[\frac{(e_{lt}^2)_{av}}{C_{lt}\epsilon_s} \right] \right\}^{-1} \quad (3)$$

where ϵ_s is the static dielectric constant, $(e_t^2)_{av}$ and $(e_l^2)_{av}$ are spherical averages of the square of piezoelectric stress constants for longitudinal and transverse waves, respectively. C_l and C_t are the spherical elastic constant for the longitudinal and transverse waves, respectively. The summation is taken over all modes (longitudinal and transverse waves) for acoustic phonons.

References

- Kojima, A., Teshima, K., Shirai, Y. & Miyasaka, T. Organometal halide perovskites as visible-light sensitizers for photovoltaic cells. *Journal of the American Chemical Society* **131**, 6050–6051 (2009).
- Stranks, S. D. & Snaith, H. J. Metal-halide perovskites for photovoltaic and light-emitting devices. *Nature nanotechnology* **10**, 391–402 (2015).
- Bach, U. Perovskite solar cells: Brighter pieces of the puzzle. *Nature chemistry* **7**, 616–617 (2015).
- Sessolo, M. & Bolink, H. J. Perovskite solar cells join the major league. *Science* **350**, 917–917 (2015).
- Green, M. A., Emery, K., Hishikawa, Y., Warta, W. & Dunlop, E. D. Solar cell efficiency tables (version 48). *Progress in Photovoltaics: Research and Applications* **24**, 905–913 (2016).
- Agiorghousis, M. L., Sun, Y.-Y., Zeng, H. & Zhang, S. Strong covalency-induced recombination centers in perovskite solar cell material CH₃NH₃PbI₃. *Journal of the American Chemical Society* **136**, 14570–14575 (2014).
- Brenner, T. M., Egger, D. A., Kronik, L., Hodes, G. & Cahen, D. Hybrid organic/inorganic perovskites: low-cost semiconductors with intriguing charge-transport properties. *Nature Reviews Materials* **1**, 15007 (2016).
- McMeekin, D. P. *et al.* A mixed-cation lead mixed-halide perovskite absorber for tandem solar cells. *Science* **351**, 151–155 (2016).
- Wehrenfennig, C., Eperon, G. E., Johnston, M. B., Snaith, H. J. & Herz, L. M. High charge carrier mobilities and lifetimes in organolead trihalide perovskites. *Advanced materials* **26**, 1584–1589 (2014).
- Ponseca, C. S. Jr. *et al.* Organometal halide perovskite solar cell materials rationalized: ultrafast charge generation, high and microsecond-long balanced mobilities, and slow recombination. *Journal of the American Chemical Society* **136**, 5189–5192 (2014).
- Karakus, M. *et al.* Phonon–electron scattering limits free charge mobility in methylammonium lead iodide perovskites. *The journal of physical chemistry letters* **6**, 4991–4996 (2015).
- Reid, O. G., Yang, M., Kopidakis, N., Zhu, K. & Rumbles, G. Grain-Size Limited Mobility in Methylammonium Lead Iodide Perovskite Thin-Films. *ACS Energy Letters* **1**, 561–565 (2016).
- Chen, Y. *et al.* Extended carrier lifetimes and diffusion in hybrid perovskites revealed by Hall effect and photoconductivity measurements. *Nature Communications* **7**, 12253 (2016).
- Filipetti, A., Mattoni, A., Caddeo, C., Saba, M. & Delugas, P. Low electron-polar optical phonon scattering as a fundamental aspect of carrier mobility in methylammonium lead halide CH₃NH₃PbI₃ perovskites. *Physical Chemistry Chemical Physics* **18**, 15352–15362 (2016).
- Stoumpos, C. C., Malliakas, C. D. & Kanatzidis, M. G. Semiconducting tin and lead iodide perovskites with organic cations: phase transitions, high mobilities, and near-infrared photoluminescent properties. *Inorganic chemistry* **52**, 9019–9038 (2013).
- Valverde-Chávez, D. A. *et al.* Intrinsic femtosecond charge generation dynamics in single crystal CH₃NH₃PbI₃. *Energy & Environmental Science* **8**, 3700–3707 (2015).
- Brenner, T. M. *et al.* Are Mobilities in Hybrid Organic–Inorganic Halide Perovskites Actually High? *The journal of physical chemistry letters* **6**, 4754–4757 (2015).
- Wright, A. D. *et al.* Electron-phonon coupling in hybrid lead halide perovskites. *Nature Communications* **7**, 11755 (2016).
- La-o vorakiat, C. *et al.* Phonon Mode Transformation Across the Orthorhombic–Tetragonal Phase Transition in a Lead Iodide Perovskite CH₃NH₃PbI₃: A Terahertz Time-Domain Spectroscopy Approach. *The journal of physical chemistry letters* **7**, 1–6 (2016).
- Menéndez-Proupin, E., Ros, C. L. B. & Wahnón, P. Nonhydrogenic exciton spectrum in perovskite CH₃NH₃PbI₃. *Physica status solidi (RRL)–Rapid Research Letters* **9**, 559–563 (2015).
- Zhou, Y. *et al.* Giant photostriction in organic-inorganic lead halide perovskites. *Nature communications* **7**, 11193 (2016).
- Leguy, A. *et al.* Dynamic disorder, phonon lifetimes, and the assignment of modes to the vibrational spectra of methylammonium lead halide perovskites. *arXiv preprint arXiv:1606.01841* (2016).
- Ledinsky, M. *et al.* Raman spectroscopy of organic–inorganic halide perovskites. *The journal of physical chemistry letters* **6**, 401–406 (2015).
- Lee, J., Koteles, E. S. & Vassell, M. Luminescence linewidths of excitons in GaAs quantum wells below 150 K. *Physical Review B* **33**, 5512–5516 (1986).

25. Milot, R. L., Eperon, G. E., Snaith, H. J., Johnston, M. B. & Herz, L. M. Temperature-Dependent Charge-Carrier Dynamics in CH₃NH₃PbI₃ Perovskite Thin Films. *Advanced Functional Materials* **25**, 6218–6227 (2015).
26. Savenije, T. J. *et al.* Thermally activated exciton dissociation and recombination control the carrier dynamics in organometal halide perovskite. *The journal of physical chemistry letters* **5**, 2189–2194 (2014).
27. Oga, H., Saeki, A., Ogomi, Y., Hayase, S. & Seki, S. Improved understanding of the electronic and energetic landscapes of perovskite solar cells: high local charge carrier mobility, reduced recombination, and extremely shallow traps. *Journal of the American Chemical Society* **136**, 13818–13825 (2014).
28. He, Y. & Galli, G. Perovskites for solar thermoelectric applications: A first principle study of CH₃NH₃AlI₃ (A = Pb and Sn). *Chemistry of Materials* **26**, 5394–5400 (2014).
29. Chin, X. Y., Cortecchia, D., Yin, J., Bruno, A. & Soci, C. Lead iodide perovskite light-emitting field-effect transistor. *Nature communications* **6**, 7383 (2015).
30. Motta, C., El-Mellouhi, F. & Sanvito, S. Charge carrier mobility in hybrid halide perovskites. *Scientific reports* **5**, 12746 (2015).
31. Wang, Y., Zhang, Y., Zhang, P. & Zhang, W. High intrinsic carrier mobility and photon absorption in the perovskite CH₃NH₃PbI₃. *Physical Chemistry Chemical Physics* **17**, 11516–11520 (2015).
32. Zhao, T., Shi, W., Xi, J., Wang, D. & Shuai, Z. Intrinsic and Extrinsic Charge Transport in CH₃NH₃PbI₃ Perovskites Predicted from First-Principles. *Scientific reports* **7**, 19968 (2016).
33. Wang, H., Pei, Y., LaLonde, A. D. & Snyder, G. J. Weak electron–phonon coupling contributing to high thermoelectric performance in n-type PbSe. *Proceedings of the National Academy of Sciences* **109**, 9705–9709 (2012).
34. Schlichting, U. & Gobrecht, K. The mobility of free carriers in PbSe crystals. *Journal of Physics and Chemistry of Solids* **34**, 753–758 (1973).
35. Heo, S. W., Ko, Y. D., Kim, Y. S. & Moon, D. K. Enhanced performance in polymer light emitting diodes using an indium–zinc–tin oxide transparent anode by the controlling of oxygen partial pressure at room temperature. *J. Mater. Chem. C* **1**, 7009–7019 (2013).
36. Coll, M. *et al.* Polarization switching and light-enhanced piezoelectricity in lead halide perovskites. *The journal of physical chemistry letters* **6**, 1408–1413 (2015).
37. Liu, S., Zheng, F., Grinberg, I. & Rappe, A. M. Photoferroelectric and Photopiezoelectric Properties of Organometal Halide Perovskites. *The journal of physical chemistry letters* **7**, 1460–1465 (2016).
38. Fan, Z. *et al.* Ferroelectricity of CH₃NH₃PbI₃ perovskite. *The journal of physical chemistry letters* **6**, 1155–1161 (2015).
39. Frost, J. M. *et al.* Atomistic origins of high-performance in hybrid halide perovskite solar cells. *Nano letters* **14**, 2584–2590 (2014).
40. Kang, N. L. & Choi, S. D. Scattering effects of phonons in two polymorphic structures of gallium nitride. *Journal of Applied Physics* **106**, 063717 (2009).
41. Vitanov, S., Nedjalkov, M. & Palankovski, V. A Monte Carlo model of piezoelectric scattering in GaN. In *International Conference on Numerical Methods and Applications*, 197–204 (Springer, 2006).
42. Borriello, I., Cantele, G. & Nino, D. Ab initio investigation of hybrid organic–inorganic perovskites based on tin halides. *Physical Review B* **77**, 235214 (2008).
43. Motta, C. *et al.* Revealing the role of organic cations in hybrid halide perovskite CH₃NH₃PbI₃. *Nature communications* **6**, 7026 (2015).
44. Du, M.-H. Efficient carrier transport in halide perovskites: theoretical perspectives. *Journal of Materials Chemistry A* **2**, 9091–9098 (2014).
45. Yin, W.-J., Shi, T. & Yan, Y. Unique properties of halide perovskites as possible origins of the superior solar cell performance. *Advanced Materials* **26**, 4653–4658 (2014).
46. Burschka, J. *et al.* Sequential deposition as a route to high-performance perovskite-sensitized solar cells. *Nature* **499**, 316–319 (2013).
47. Lu, Y.-B. *et al.* Geometric, electronic and optical properties of zinc/tin codoped In₂O₃ modulated by the bixbyite/corundum phase transition. *Journal of Physics D: Applied Physics* **49**, 065105 (2016).
48. Wang, Y. *et al.* Electronic structure of III–V zinc-blende semiconductors from first principles. *Phys. Rev. B* **87**, 235203 (2013).
49. Menéndez-Proupin, E., Palacios, P., Wahnón, P. & Conesa, J. Self-consistent relativistic band structure of the CH₃NH₃PbI₃ perovskite. *Physical Review B* **90**, 045207 (2014).
50. Gao, W. *et al.* Quasiparticle band gap of organic–inorganic hybrid perovskites: Crystal structure, spin-orbit coupling, and self-energy effects. *Physical Review B* **93**, 085202 (2016).
51. Miyata, A. *et al.* Direct measurement of the exciton binding energy and effective masses for charge carriers in organic–inorganic tri-halide perovskites. *Nature Physics* **11**, 582–587 (2015).
52. Yin, W.-J., Yang, J.-H., Kang, J., Yan, Y. & Wei, S.-H. Halide perovskite materials for solar cells: a theoretical review. *Journal of Materials Chemistry A* **3**, 8926–8942 (2015).
53. Kim, Y.-J. *et al.* Piezoelectric properties of CH₃NH₃PbI₃ perovskite thin films and their applications in piezoelectric generators. *Journal of Materials Chemistry A* **4**, 756–763 (2016).
54. Feng, J. Mechanical properties of hybrid organic–inorganic CH₃NH₃ BX₃ (B = Sn, Pb; X = Br, I) perovskites for solar cell absorbers. *APL Materials* **2**, 081801 (2014).
55. Zhou, Y., Game, O. S., Pang, S. & Padture, N. P. Microstructures of organometal trihalide perovskites for solar cells: Their evolution from solutions and characterization. *The journal of physical chemistry letters* **6**, 4827–4839 (2015).
56. Sun, S., Fang, Y., Kieslich, G., White, T. J. & Cheetham, A. K. Mechanical properties of organic–inorganic halide perovskites, CH₃NH₃PbX₃ (X = I, Br and Cl), by nanoindentation. *Journal of Materials Chemistry A* **3**, 18450–18455 (2015).
57. Lu, Y.-B. *et al.* How does the multiple constituent affect the carrier generation and charge transport in multicomponent TCOs of In–Zn–Sn oxide. *Journal of Materials Chemistry C* **3**, 7727–7737 (2015).
58. Gong, K., Zhang, L., Ji, W. & Guo, H. Electrical contacts to monolayer black phosphorus: A first-principles investigation. *Physical Review B* **90**, 125441 (2014).
59. Wei, S.-H. & Zunger, A. Predicted band-gap pressure coefficients of all diamond and zinc-blende semiconductors: Chemical trends. *Physical Review B* **60**, 5404–5411 (1999).
60. Kim, J., Lee, S.-C., Lee, S.-H. & Hong, K.-H. Importance of orbital interactions in determining electronic band structures of organolead iodide. *The Journal of Physical Chemistry C* **119**, 4627–4634 (2015).
61. Kh, A. N., Bankina, V., Poretskaya, L., Shelimova, L. & Skudnova, E. Semiconducting II–VI, IV–VI and V–VI Compounds. NY: Plenum Press **249**, 14 (1969).
62. Riffe, D. M. Temperature dependence of silicon carrier effective masses with application to femtosecond reflectivity measurements. *JOSA B* **19**, 1092–1100 (2002).
63. Lundstrom, M. *Fundamentals of Carrier Transport*, second edn. Cambridge Books Online (Cambridge University Press, 2000).
64. Seto, J. Y. The electrical properties of polycrystalline silicon films. *Journal of Applied Physics* **46**, 5247–5254 (1975).
65. Kresse, G. & Furthmüller, J. Efficient iterative schemes for ab initio total-energy calculations using a plane-wave basis set. *Physical review B* **54**, 11169–11186 (1996).
66. Heyd, J., Scuseria, G. E. & Ernzerhof, M. Erratum: Hybrid functionals based on a screened Coulomb potential[J. Chem. Phys. **118**, 8207 (2003)]. *The Journal of Chemical Physics* **124**, 219906 (2006).
67. Qiao, J., Kong, X., Hu, Z.-X., Yang, F. & Ji, W. High-mobility transport anisotropy and linear dichroism in few-layer black phosphorus. *Nature communications* **5**, 4475 (2014).
68. Hutson, A. Piezoelectric scattering and phonon drag in ZnO and CdS. *Journal of Applied Physics* **32**, 2287–2292 (1961).

Acknowledgements

This work is supported by the Natural Science and Engineering Research Council (NSERC) of Canada (D.C. and H.G.), and the National Natural Science Foundation of China with Grant No. 11504202 (Y.-B.L.). We thank high performance computer center of McGill University, Calcul-Quebec and Compute Canada for the substantial computation allocation which made this work possible. Parts calculations in this work have been done on the Supercomputing Center, Shandong University, Weihai.

Author Contributions

Y.-B.L. carried out the research and formulated the research project with H.G. X.K. provided the strategy for calculating deformation potential scattering limited mobility. X.B.C. participated interpretations of data. Y.-B.L., G.H. and D.C. wrote the manuscript and all authors contributed to discussions of results.

Additional Information

Supplementary information accompanies this paper at <http://www.nature.com/srep>

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Lu, Y.-B. *et al.* Piezoelectric scattering limited mobility of hybrid organic-inorganic perovskites $\text{CH}_3\text{NH}_3\text{PbI}_3$. *Sci. Rep.* **7**, 41860; doi: 10.1038/srep41860 (2017).

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

This work is licensed under a Creative Commons Attribution 4.0 International License. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in the credit line; if the material is not included under the Creative Commons license, users will need to obtain permission from the license holder to reproduce the material. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>

© The Author(s) 2017