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OPEN Dynamic defect correlations dominate activated electronic transport in SrTiO₃

Paul C. Snijders^{1,2}, Cengiz Şen³, Michael P. McConnell², Ying-Zhong Ma⁴, Andrew F. May¹, Andreas Herklotz¹, Anthony T. Wong¹ & T. Zac Ward¹

Strontium titanate (SrTiO₃, STO) is a critically important material for the study of emergent electronic phases in complex oxides, as well as for the development of applications based on their heterostructures. Despite the large body of knowledge on STO, there are still many uncertainties regarding the role of defects in the properties of STO, including their influence on ferroelectricity in bulk STO and ferromagnetism in STO-based heterostructures. We present a detailed analysis of the decay of persistent photoconductivity in STO single crystals with defect concentrations that are relatively low but significantly affect their electronic properties. The results show that photo-activated electron transport cannot be described by a superposition of the properties due to independent point defects as current models suggest but is, instead, governed by defect complexes that interact through dynamic correlations. These results emphasize the importance of defect correlations for activated electronic transport properties of semiconducting and insulating perovskite oxides.

In modern materials design, the perfection of material synthesis capabilities has led to numerous discoveries and technologies ranging from quantum Hall effects in high mobility semiconductor heterostructures^{1,2} to tunable microwave dielectrics in complex oxides³. In these examples, point defects and impurities in the crystalline lattice are often deleterious, because they profoundly affect the electronic properties. However, precisely because of their effect on materials' properties, defects and impurities offer an extremely versatile way to significantly expand the range of phenomena exhibited by a parent material that can be used to control new properties. STO is an excellent example of this: oxygen vacancies can control resistive switching^{4,5} and lead to magnetism⁶ or even superconductivity⁷, strontium vacancies can induce ferroelectricity⁸, and the presence of both cation^{9,10}, and anion^{11,12} point defects has been used to explain emergent electronic properties near STO - perovskites interfaces. Furthermore, in complex oxides beyond STO, oxygen vacancies and other point defects can produce active sites with a high catalytic activity for e.g. oxygen reduction or evolution reactions¹³.

It is often challenging to ascertain the type of defects present, in particular when defect concentrations are low, and most point defects need to be monitored by indirect methods¹⁴. However, even at concentrations below the detection limit, defects can have a tremendous effect in the properties of functional materials as evidenced by the appearance of persistent photoconductivity due to low concentrations of defects¹⁵⁻¹⁷. Persistent photoconductivity can thus be exploited to probe low concentrations of defects, in particular those that are relevant for the electronic properties of the material. Specifically, exposing semiconductors or insulators to light with energies near and above the band gap photoexcites charge carriers, which are subsequently trapped in defect states within the gap¹⁵⁻¹⁷. Persistent electronic conduction happens through hopping mechanisms between these defect states, and its decay with time offers insight into the energy barriers that are involved in de-trapping the carriers responsible for the observed conductivity¹⁵⁻¹⁷. The altered electronic properties of defected crystalline materials are usually considered in terms of isolated, independent point defects such as the oxygen vacancies in STO mentioned above¹⁸. However, interactions between isolated defects can create defect complexes that have a lower formation energy than isolated point defects¹⁹⁻²² and thus dominate the properties of defected materials. The static electronic structure of these composite defect structures cannot be described by a superposition of individual defect

¹Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA. ²Department of Physics and Astronomy, University of Tennessee Knoxville, Tennessee 37996, USA. ³Department of Physics, Lamar University, Beaumont, Texas 77710, USA. ⁴Chemical Sciences Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA. Correspondence and requests for materials should be addressed to P.C.S. (email: snijderspc@ornl.gov) or T.Z.W. (email: wardtz@ornl.gov)

electronic spectra²³. Moreover, the nature of the interactions also allows for dynamic correlations in response to changes in electron occupation of defect levels. These dynamic correlations should then be apparent in correlated electronic relaxations upon creating non-equilibrium electron energy distributions ubiquitous in electronic transport experiments. Such dynamic correlations are conceptually comparable to dynamic correlations that constrain structural relaxations in solids, leading to cascading relaxation processes²⁴, but their influence on effective activation energies has not been recognized yet in basic electronic transport experiments.

Here we report an experimental study demonstrating that in insulating STO single crystals with low defect concentrations, photo-induced electronic transport does not follow relaxation pathways governed by a superposition of activation energies associated with independent point defects, but instead proceeds through electronic states of defect complexes that interact through dynamic correlations. We chose to study STO because it is a prototypical insulating perovskite transition metal oxide that is widely used as a substrate to grow complex oxide thin films and heterostructures. As such STO, its defects and their properties are a crucial part of the design of functional complex oxide materials and devices as borne out by the intense focus on STO since the seminal work by Ohtomo and Hwang²⁵ describing the high-mobility two-dimensional electron gas at STO and LaAlO₃ interfaces.

In our experiments, annealing treatments were designed to produce low concentrations of different types of defects. In particular, the different annealing conditions aimed to affect the chemical potential of the elements in the samples, which, in turn, is expected to result in different defect formation energies that determine equilibrium defect concentrations^{18,19,21}. The decay of the photoconductivity for these samples varies from a few minutes to a few hours. We analyze fits of the decay profiles to double or triple, as well as stretched exponential (Kohlrausch) decay functions. For each type of fit, we extract the apparent activation energies. A disparity between the activation energies extracted from the two types of fits suggests that the persistent conductivity originates in clusters of interacting defects, or defect complexes, rather than in isolated and independent point defects, and that its decay proceeds subject to dynamical correlations in these defect complexes. Our results indicate that designing STO-based systems requires proper control over sample illumination, and a thorough treatment of defects, their correlations, and the resulting effective activation energies, in order to properly describe their electronic properties.

Results and Discussion

Figure 1 presents the sheet resistance measured at room temperature as a function of time and illumination on the differently treated STO crystals (for treatment conditions, see Table 1 and Experimental Methods) plotted using the same sheet resistance scale to emphasize the different behaviors. In addition to vacancy disorder due to stoichiometric Schottky defects in STO²⁶, these treatments aimed at producing low concentrations of oxygen and titanium vacancies (V_0 and V_{Ti}) and Sr_{Ti} antisite defects²¹, as well as extrinsic Si impurities. The sheet resistances before photoexcitation are consistently 18 to $20 \text{ G}\Omega/\Box$ for 0.5 mm thick samples. Isolated oxygen vacancies, produced by the vacuum annealing, are known to rapidly increase *n*-type conductivity²⁷ due to the formation of a donor state located close to the conduction band minimum of STO^{18,28}. The high resistance of our annealed samples therefore suggests that the oxygen vacancy concentration in our samples is not very large. This is corroborated by X-ray diffraction data (supplementary information) that does not show evidence of a defect-induced unit cell volume increase.

Upon illumination with below band gap light, all samples annealed in a SiO₂ ampoule¹⁷ (samples A, B, C) show conventional (transient) photoconductivity as well as persistent photoconductivity that continues long after illumination has ceased (Fig. 1); only the sample subjected to annealing in an Al₂O₃ boat exhibits no photoconductivity upon illumination (sample D, Fig. 1(d)). This suggests that the measured (persistent) photoconductivity is dependent on the environmental conditions during annealing. It is known that annealing quartz at 1200 °C in vacuum produces SiO vapor²⁹, and hence it is probable that silicon (oxide) is deposited on the STO samples when annealed in SiO₂ ampoules. Indeed, X-ray Photoelectron Spectroscopy (XPS) detects the presence of Si on the surface of sample B, as shown in the inset of Fig. 2. Comparing the integrated intensity of the Si 2*p* photoelectron peak to that of the Ti 2*p* peak and accounting for the order of magnitude different photoionization cross sections for the respective kinetic energies of the photoelectrons from these core levels³⁰, as well as their respective inelastic mean free paths (assuming a homogeneous sample) gives an estimate of the surface Si/Ti atomic ratio of approximately 0.1. While it is not possible to reliably extract the concentration of Si that is incorporated in the STO lattice because most of the silicon observed in the XPS data could be located on the surface, silicon will diffuse into the STO crystal during a high temperature anneal³¹. These results suggest that Si defects such as Si interstitials or Si_{Ti} defects¹⁴ play a role in the observed photoconductivity upon illumination with below bandgap light.

Apart from the likely Si impurities, the presence of SrO in the quartz ampoules¹⁷ is expected to affect the type of defects present in the STO crystals after annealing. The presence of bulk SrO with the STO crystal during vacuum annealing has been predicted to produce oxygen vacancies and Sr_{Ti} antisite defects²¹, and could decrease the relatively high formation energy of Ti vacancies under reducing conditions³². A rudimentary estimate of Sr/Ti ratio we deduce from the Sr 3*d* and Ti 2*p* core level XPS spectra (Fig. 2) is Sr/Ti = 1.5 (after correcting for the difference in photoionization cross section and inelastic mean free path for the Sr 3*d* and Ti 2*p* core levels, assuming a homogeneous sample). This suggests that excess Sr is present on the surface, and is consistent with the smaller Sr/Ti intensity ratio measured on un-treated STO (not shown). Since this off-stoichiometry is much larger than expected for STO, these data indicate that SrO has likely migrated through the gas phase onto the STO surface.

In the following we analyze in more detail the persistent photoconductivity of the STO samples that were annealed under various conditions in quartz ampoules (Fig. 1(a,b,c)) with the aim of extracting the energies of the in-gap defect states governing the persistent photoconductivity. Figure 3 presents the decay profiles of the photoconductivity of these crystals after the illumination was stopped. Specifically, Fig. 3(a) shows the photoconductivity decay of sample A, annealed without SrO, and measured in vacuum (see also Fig. 1(a)). Figure 3(b) shows the decay profile measured on sample B annealed in the presence of SrO, and measured in vacuum (see also





Sample	Annealing environment	Cooling rate	Expected defects		
А	SiO ₂ Ampoule	10 K/min	$V_{\rm O}$, Si imp.		
В	SiO ₂ Ampoule–SrO powder	10 K/min	$V_{\rm O}$, Sr _{Ti} , $V_{\rm Ti}$, Si imp.		
С	SiO ₂ Ampoule–SrO powder	Quenched	$V_{\rm O}$, Sr $_{\rm Ti}$, $V_{\rm Ti}$, Si imp.		
D	Al ₂ O ₃ boat–SrO powder	10 K/min	Vo		

Table 1. Annealing conditions. The annealing conditions, cooling rates, and the resulting expected defect types in the STO samples.

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Fig. 1(b)) as well as in an oxygen atmosphere (see below). Finally, the data in Fig. 3(c) were measured in vacuum on sample C that was quench-cooled from high temperature in the presence of SrO (see also Fig. 1(c)), which should result in an increased defect concentration as compared to an equilibrium cool-down. Clearly the decay rates of the photoconductivity in these samples differ significantly. The longest persistent photoconductivity is observed in samples that were annealed in the presence of SrO, suggesting that, apart from possible Si impurities, intrinsic defects such as Sr_{Ti} antisite defects or Ti vacancies^{21,32} could be essential for long-term persistent photoconductivity in STO. Although in the quenched sample C (Fig. 3(c)) the concentration of these defects is expected to be even larger as a higher temperature equilibrium concentration is frozen into this sample, the persistence of the photoconductivity is considerably shorter in the quench-cooled sample (Fig. 3(c)) than in the slowly cooled samples (Fig. 3(b)). This unexpected finding suggests that an increased defect concentration leads to changes in the nature of the effective trapping states.



Figure 2. Photoelectron spectroscopy data for the Ti 2*p*, Sr 3*d*, and Si 2*p* (inset) core levels measured on sample B. From the integrated intensities of the Ti 2*p* and Sr 3*d* peaks, a Sr/Ti ratio of 1.5 is inferred, suggesting excess Sr is present on the surface. The data in the inset reveal the presence of Si on the surface of the sample.



Figure 3. Decaying sheet photoconductance after the illumination is switched off. (**a**–**c**) present data for samples A–C, measured in vacuum. In (**b**), a measurement in an oxygen atmosphere is shown for comparison. Significantly different relaxation times are apparent, originating from different defects present in the samples. The data are fitted to stretched exponential functions plotted as continuous lines as discussed in the text.

In Fig. 3(b) a second data set is also plotted, which was illuminated and measured in an oxygen atmosphere to assess the influence of oxygen vacancies. Apart from the vacuum annealing treatment which produces a low concentration of oxygen vacancies, exposure to intense synchrotron UV light in vacuum is known to produce a high concentration of oxygen vacancies near the STO surface, creating a 2-dimensional electron gas (2DEG)^{33,34}.

	Multiple exponential fits				Stretched exponential fits		
Sample	$A_i[S]$	$ au_i[s]$	E_i [eV]	χ^2_{red}	au[s]	β	χ^2_{red}
А	1.5E-12	1.4	0.82	7E-27	1.54E7	0.25	9E-27
	2.1E-12	78.4	0.93				
B _{vac}	4.3E-11	43.2	0.91	9E-25	1.21E3	0.44	7E-25
	6.9E-11	583.9	0.98				
	8.0E-11	4074.2	1.03				
B _{oxy}	3.5E-11	11.0	0.88	4E-25	4.62E3	0.54	4E-25
	6.5E-11	211.0	0.95				
	7.9E-11	1186.2	1.00				
С	1.8E-12	1.3	0.82	2E-26	7.3E9	0.30	3E-26
	9.1E-12	60.7	0.92				
	9.7E-11	76771.9	1.11				

 Table 2. Decay parameters obtained from exponential fits. Fitting parameters from (stretched) exponential fits to the decay in photoconductivity of STO samples with low concentrations of intentional defects.

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Exposing such oxygen-deficient STO samples to oxygen fills the oxygen vacancies again³⁵. It is unlikely that persistent photoconductivity in our experiments is caused by such a 2DEG effect because of the much lower intensity and energy of the light used in this work (3.06 eV in this work vs. ~50 eV in refs 33,34). This is corroborated by the data in Fig. 3(b) which shows that the presence of oxygen does not destroy persistent photoconductivity. The reduction in the persistence by approximately a factor of 2 shows that oxygen vacancies are likely to play a role, however, they are evidently neither a dominant nor independent cause of the persistent photoconductivity as the photoconductivity still persists for orders of magnitude longer than the conventional transient photoconductivity. Note that if the effect of the oxygen atmosphere would be limited to filling (independent) oxygen vacancies in only a surface layer of a sample that is oxygen deficient throughout the bulk, the measured resistance should increase proportional to the decreased volume of oxygen-deficient and semiconducting STO. This is not observed, indicating that the persistent conductivity occurs throughout the bulk of the sample consistent with the findings in ref. 17. Moreover, in a system with independent defects, the characteristic relaxation timescale should not change, because only the quantity of one type of defect is changing, while the types of the defects present in the sample remains the same. This is clearly inconsistent with the observed data in Fig. 3(b).

To gain quantitative insight into what defects dominate the observed behavior, an experimental determination of the energies associated with the defects that dominate the persistent conductivity is necessary. To this end, we first fitted the time dependent conductivity $\sigma(t)$ using a sum of independent exponential (Debye) decays³⁶, i.e. using the common assumption that the data can be described using a superposition of independent defects; $\sigma(t) = \sigma_0 + \sum_{i=1}^n A_i e^{-\frac{t}{\tau_i}}$, where σ_0 is the dark (background) conductivity, A a conductivity amplitude weighing factor, and τ a characteristic relaxation time. Using $(\tau) = k_B T \ln(\tau \nu)^{36.37}$ we extracted the activation energy *E* for each decay process associated with the relaxation times τ determined from these fits. Here $k_{\rm B}$ is the Boltzmann constant, T denotes the temperature (300 K), and ν is the attempt frequency which we take to be a typical phonon frequency of $v = 5 \times 10^{13}$ Hz. The results of these fits are summarized in Table 2. While the STO sample annealed without SrO (Fig. 3(a)) could be satisfactorily fitted using a double exponential function, all the data in Fig. 3(b,c) need triple exponential functions for acceptable fits. The activation energies calculated from the obtained relaxation times lie between 0.82 and 1.11 eV, and are plotted as data points in Fig. 4, with the corresponding relative amplitudes (right y-axis) given by the weighing factors A_i of the individual exponential components. Clearly, these defect states are located deep in the 3.3 eV gap of STO. Comparing these activation energies to recently reported calculated energy levels within the STO band gap for the defects that could be expected from our annealing conditions^{18,21,28,32}, reveals that the extracted activation energies are inconsistent with those of known defect levels (see supplementary information).

Our results therefore suggest that the presence of isolated, independent defects is unable to explain the measured decay of the photoconductivity but that interacting defect complexes should be considered instead^{20,23}. In the case of a perovskite oxide such as STO, the number of possible defect complexes can be significant²¹. This is compounded in the present study by the likely presence of Si interstitials or Si_{Ti} and Si_{Sr} antisite defects, which to our knowledge have not yet been considered theoretically. In this context, it is well-known that relaxations in many disordered systems can be empirically described using a stretched exponential or Kohlrausch function, i.e.

 $(t) = \sigma_0 e^{-(\frac{1}{\tau})^{\circ}}$, $0 < \beta < 1^{38,39}$. While it is commonly assumed that a stretched exponential decay arises because there is a distribution of independent relaxation processes with associated activation energies, it was shown³⁹ that relaxation in interacting disordered materials with hierarchically constrained (i.e. cascading) dynamics also gives rise to stretched exponential decay, where the stretch exponent β has been suggested to be the inverse of the number of interacting or cooperating processes that must occur to lead to a relaxation event⁴⁰. By fitting our data with a stretched exponential function (included in Fig. 3), even with the significantly reduced number of parameters, fits were obtained with a similar quality as compared to double and triple exponential functions. Table 2 lists the resulting stretch exponents, β , and relaxation times, τ . The values for β are relatively small (<0.55), consistent with a large deviation from a single exponential Debye decay. It is worth noting that for sample A, the double (independent) exponential fit suggests only two processes are active in the decay, which should have resulted in



Figure 4. Activation energies for the processes dominating the decay in photoconductivity for samples A–C extracted using the two methods discussed in the text. The data points, connected by dotted lines, are calculated directly from the relaxation times obtained from fitting the decay curves with multiple exponential functions. The continuous curves are obtained through an inverse Laplace transformation of the stretched exponential decay fits.

the largest value of β , i.e. the system is closest to a description with a single exponential decay ($\beta = 1$). Instead, the stretched exponential fit results in an exceedingly small $\beta = 0.25$. If we take β^{-1} to be the number of cooperating or correlated processes^{40,41} associated with specific defect structures, then the observed relaxation of the persistent photoconductivity for the defect configurations analyzed here should involve 2 to 4 defect states, see Table 2. The discrepancy between the number of independent exponentials for each individual dataset and the value of β^{-1} needed to fit the data (Table 2) implies that the stretched exponential decays cannot be explained with a distribution of independent relaxation processes, each with its own relaxation time.

Finally, we have extracted effective activation energy distributions from these stretched exponential decays by calculating the inverse Laplace transform of the decaying photoconductivity^{42–44}. These distributions are plotted as continuous curves in Fig. 4 for the decays presented in Fig. 3, normalized to have the same total area under each curve for better visibility. Comparing these peaked activation energy distributions with the monotonic increase in the relative amplitudes obtained from the multiple (independent) exponential fits (datapoints in Fig. 4) shows that, again, there is a discrepancy between the independent defect model, and the correlated defect model and the data: the data points do not correspond to the envelope of the (continuous) distributions, and the activation energies obtained from the multiple exponential fits (datapoints in Fig. 4) with the largest weight are significantly removed (>100 meV) from the maxima in the distribution curves from the stretched exponential fits.

This leads us to the important conclusion that the observed relaxation of the photoconductivity in the STO system cannot be understood by a sum of independent individual processes each with their own activation energy. Instead, the relaxation proceeds through cooperative or hierarchical^{24,39} events that correlate the relaxation processes occurring at different defects. Our main observations intuitively fit this picture: (i) the changing nature (characteristic relaxation time) of the trapping states with increased defect concentration (comparing sample C and Byrac, see Tables 1 and 2) implies that the increase of the concentration of individual defects creates new complexes made up of different sets of interacting defects. The non-trivial way in which the associated defect states depend on the detailed structure and interaction of such complexes²³ naturally explains the changing characteristic relaxation times upon increasing the defect concentration by quenching the sample from the annealing temperature. (ii) The changing relaxation times upon decreasing only the oxygen vacancy concentration (comparing both samples B) corroborates this explanation; eliminating oxygen vacancies from a defect complex does not simply remove an independent defect level associated with an independent oxygen vacancy from the extracted activation energies (Fig. 4). Instead, it alters or shifts the effective activation energy distribution, i.e. it changes the character of the trapping state complexes. (iii) The fact that the effective activation energies extracted from fits based on independent Debye decays do not correspond to deep trap states of the most likely point defects, nor to the distribution of activation energies extracted from the stretched exponential fit, indicates that the effective activation energies governing the dynamic decay present in the samples are indeed non-trivially affected by the interactions between the defects. We conjecture that while Coulomb interactions between charged defects may dominate these interactions in many materials, for the case of STO these interactions are likely mediated through lattice deformations as the high dielectric constant of STO screens the Coulomb interaction between charged defects forming a complex. Indeed, the STO lattice does significantly respond to the presence of defects as evidenced by the presence of polarons near defects^{18,28}.

Conclusions

In our experiments we have shown that photo-activated electron transport in STO single crystals with low defect concentrations cannot be described by a superposition of properties due to independent point defects as current

models suggest. Instead, defect complexes that interact through dynamic correlations determine the effective activation energies for electronic transport. Our experiments therefore reveal that in order to understand how defects affect the electronic properties of STO, its interfaces, and complex oxides and heterostructures in general, it is necessary to consider models beyond those based on independent defects. Our finding bears important implications for defect engineering of complex oxides to obtain desired properties in heterogeneous catalysis and low dimensional electron systems, as well as for gaining an understanding of superconductivity or colossal magnetoresistance in these materials where similar interactions lead to emergent properties. Moreover, the approach demonstrated in this work makes it possible to gain spectroscopic insight into the defect complexes that dominate electronic properties in complex oxides through basic transport experiments.

Experimental Methods

 $5 \text{ mm} \times 5 \text{ mm} \times 0.5 \text{ mm} \text{ SrTiO}_3$ single crystal samples were used in this study. To produce different defect complexes, we placed these samples in different environments during a 1 h 1200 °C anneal in a tube furnace with a Al₂O₃ tube. Apart from creating oxygen vacancies by annealing in vacuum, Si ampoules and Al₂O₃ boats were used to investigate the influence of possible Si-induced defects. Finally, SrO powder was added to prevent STO decomposition during high temperature annealing, and create Ti vacancies¹⁷. Specifically, three samples were placed in sealed SiO₂ ampoules evacuated to <1 mTorr; two of these ampoules were also filled with ~2 grams of SrO powder. The fourth sample was placed in an Al₂O₃ boat with ~2 grams SrO powder within the tube furnace's sample space that was evacuated to <1 mTorr during annealing. All samples were heated from 20 °C to 1200 °C at 10 °C min⁻¹ and all samples were cooled at 10 °C min⁻¹ with the exception of one Si ampoule-SrO powder sample which was quench cooled to 20 °C by directly removing from 1200 °C furnace.

4-probe electrical contacts were fabricated using indium solder with a separation of 3–4 mm between contacts. Photoexcitation was implemented by locating two 405 nm (3.06 eV) ultraviolet LEDs with a luminous intensity of 160 mcd 10 mm above the sample surface. The attenuation of this below band gap light is small, and the photoexcitation thus takes place throughout the bulk of the sample. This device was placed in a vacuum chamber with a base pressure of 10^{-8} Torr. Photoconductivity was measured at room temperature with a constant current mode of $10 \,\mu$ A. X-ray Photoelectron spectra were recorded in ultra-high vacuum using monochromatized Al K α x-rays after a mild degassing at 200 °C for 30 minutes. A Shirley background was subtracted from the spectra.

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Author Contributions

T.Z.W. fabricated samples with help from A.F.M. and carried out the transport experiments with help from A.T.W. and with input from P.C.S. on annealing treatments. C.S. performed the inverse Laplace transformations. M.P.M. carried out the XPS experiments. A.H. did the XRD measurements. P.C.S. analyzed the data and conceived the interpretation. P.C.S. and T.Z.W. supervised the project and wrote the manuscript. Y.Z.M. contributed to understanding optical properties of defect states. All authors contributed to discussions and commented on the manuscript.

Additional Information

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