SCIENTIFIC REPORTS

Received: 09 October 2015 Accepted: 08 April 2016 Published: 04 May 2016

OPEN Enhanced high-frequency absorption of anisotropic Fe₃O₄/ graphene nanocomposites

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Anisotropic Fe₃O₄ nanoparticle and a series of its graphene composites have been successfully prepared as high-frequency absorbers. The crystal structure, morphology and magnetic property of the samples were detailed characterized through X-ray diffractometer (XRD), transmission electron microscopy (TEM) and vibrating sample magnetometer (VSM). The high-frequency absorbing performance of the composites is evaluated within 2.0–18.0 GHz. Combining reduced graphene oxide (RGO) to Fe_3O_4 helps to adjust the permittivity and permeability of the composite, balance the dielectric loss and magnetic loss, consequently improve the absorbing performance in view of the impedance matching characteristic. The optimal reflection loss of the pure Fe₃O₄ sample reaches -38.1 dB with a thickness of 1.7 mm, and it increases to -65.1 dB for the sample grafted with 3 wt.% RGO. The addition of proper content of RGO both improves the reflection loss and expands the absorbing bandwidth. This work not only opens a new method and an idea for tuning the electromagnetic properties and enhancing the capacity of high-efficient absorbers, but also broadens the application of such kinds of lightweight absorbing materials frameworks.

Recently, high-frequency wave absorption materials have attracted a great deal of attention because of their potential applications in the fields of wireless data communication, mobile phones, radar systems, local area networks, satellite television and self-concealing¹⁻⁴. Excellent absorption materials should have strong wave attenuation abilities as well as a wide absorption bandwidth⁵. The attenuation of the electromagnetic wave is mainly in the form of magnetic or dielectric loss by transforming it into thermal energy⁶. Thus, adjusting the electromagnetic parameters of the materials combining kinds of lossing principles and keeping a balance between the dielectric loss and magnetic loss in view of the impedance matching characteristic will improve the absorbing performance. Moreover, considering the bandwidth, it is important for an absorber to exhibit multiple resonance phenomena in the frequencies which may contribute to fulfil wide absorbing range.

Ferrite material has been widely used in magnetic sensor, magnetic resonance, electro-magneto-rheological fluid⁷, microwave absorption⁸, and so on. So many works about its absorbing properties have been conducted and it has been proved to be a prosperous family for wave absorption. For instance9, the minimum reflection loss value of a conductive PANI/MnFe₂O₄ nanocomposite is -15.3 dB at 10.4 GHz with the thickness of 1.4 mm. And it is -12.0 dB at 11.3 GHz with the thickness of 1.5 mm for another conductive PPy/MnFe₂O₄ nanocomposite scattering in resin acrylic¹⁰. For one more example, the optimal reflection loss value of the coin-like α -Fe₂O₃@ $CoFe_2O_4$ core-shell composite can reach -60 dB at 16.5 GHz with a thickness of 2.0 mm¹¹. As a wave absorption material, Fe₃O₄ have been studied extensively for its excellent absorption properties by virtue of strong permeability and relative high resistivity. Many Fe₃O₄ composites also have been reported in recent years: Fe₃O₄@ZnO sphere decorated graphene¹², Fe₃O₄/TiO₂ core-shell nanotubes¹³, Fe₃O₄@TiO₂ yolk-shell microspheres¹⁴, fluorinated polybenzobisoxazole/silica-coated magnetic Fe₃O₄ nanocomposities¹⁵, Fe₃O₄/SiO₂ nanorods¹⁶, graphene@ Fe₃O₄ nanocluster@carbon@MnO₂ nanosheet array composites¹⁷, superparamgnetic Fe₃O₄ nanocrystals¹⁸, $Fe_3O_4@C$ core-shell nanotubes¹⁹, $Fe_3O_4@$ metal-organic framework²⁰, 3D Fe_3O_4 nanocrystals decorating carbon nanotubes21.

Graphene has been applied as a new wave absorption material because of its desirable physical and chemical properties. Nevertheless, pure graphene has very weak EM wave absorption properties^{22,23}. Many researchers have synthesized magnetic nanoparticles coupled with graphene that can highly improve the absorption

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Figure 1. Illustration of the synthetic protocol for the Fe₃O₄/RGO nanocomposites.

performance^{24,25}. Wang and *co*-workers²⁶ synthesized graphene/Fe₃O₄/SiO₂/NiO hierarchical nanosheets, of which the minimum reflection loss was up to -51.5 dB at 14.6 GHz with a thickness of only 1.8 mm and the absorption bandwidth with a reflection loss below -10 dB ranged from 12.4–17.5 GHz. Zhu *et al.*²⁷ reported a graphene-carbonyl iron cross-linked composite of 3.0 mm with a minimum reflection loss reaching -52.46 dB at 9.46 GHz.

The size, shape and composite structure play importance roles on the absorption properties of the ferrite materials²⁸⁻³¹. To further investigate the wave absorption property of the Fe₃O₄ nanocomposites, we synthesized anisotropic α -Fe₂O₃ nanoparticles by a facile hydrothermal process. The particles shown spindle-like shape and were then combined with graphene oxide (GO) to form α -Fe₂O₃/GO nanocomposites by different ratio. The anisotropic α -Fe₂O₃ nanoparticles were homogeneously dispersed in the graphene aqueous suspension and embedded into the graphene network. Finally, the Fe₃O₄/RGO nanocomposites were obtained after annealing in H₂/Ar (5%:95%) atmosphere for 2 hours at 500 °C.

Experimental

Synthesis of spindle-like α -**Fe**₂**O**₃ **nanoparticles.** The monodispersed spindle-like α -Fe₂O₃ nanoparticles was parallel to the literature through a refluxing process^{32,33}. Briefly, 1.08 g FeCl₃ · 6H₂O, 10 mg Na₂HPO₄ · 12H₂O and 200 mL deionized water were directly added into a round-bottomed flask. The mixture was heated to 110 °C and refluxed under continuous stirring for 24 hours. After cooling down to room temperature, a red brown homogeneous suspension containing α -Fe₂O₃ nanoparticles was achieved. The final samples were firstly centrifuged, and then washed with deionized water and ethanol three times, respectively.

Synthesis of Fe₃O₄/RGO nanocomposites. A series of Fe₃O₄/RGO nanocomposites were performed by a simple ultrasonic-dispersion method, illustrated in Fig. 1. Typically, 3.0 mg GO powders were added to 50 ml of deionized water and sonicated for 1 hour. Then $0.3 \text{ g} \propto$ -Fe₂O₃ nanoparticles were added into the above GO suspension and sonicated for another 1 hour. The precipitates were collected by centrifugation, followed by annealing in H₂/Ar atmosphere (5%:95%) for 2 hours at 500 °C to obtain Fe₃O₄/RGO (1.0 wt.%) nanocomposites. The other two Fe₃O₄/RGO nanocomposites were achieved by modifying the contents of GO to 9.3 mg and 15.8 mg (3 wt.% and 5 wt.%), respectively.

Characterization. The crystal structure of the samples was analyzed using X-ray diffractometer (XRD, Rigaku D/MAX-2500) with a Cu K_a irradiation ($\lambda = 1.54178$ Å, 40.0 kV, 150.0 mA), recorded from 5° to 90° (2 θ) with a scanning step of 6°/min. Transmission electron microscopy (TEM, JEOL-2100) was used to observe the morphology, size and microstructure of the samples. Room-temperature magnetic properties of the samples were measured by a Riken vibrating sample magnetometer. The complex permittivity and permeability parameters of the composites were measured using an Agilent N5230C network analyzer in the range of 2.0–18.0 GHz, for which the samples containing 50 wt.% obtained composites and 50 wt.% wax were pressed into toroidal shapes ($\Phi_{out} = 7.00$ mm and $\Phi_{in} = 3.04$ mm).



Figure 2. The XRD patterns of the samples obtained from different synthetic steps.

Results and Discussions

Structure and morphology. The crystalline structures of the as-prepared samples are presented by XRD patterns, shown in Fig. 2. Obviously, all XRD diffraction peaks belonging to crystalline α -Fe₂O₃ can be seen for the first step. Nine peaks at 24.1°, 33.1°, 35.6°, 40.8°, 49.4°, 53.9°, 57.4°, 62.3° and 63.9° are assigned to the reflections from the (012), (104), (110), (113), (024), (116), (018), (214) and (300) crystal planes (JCPDS card no. 24-0072), respectively, which is in good agreement with the reference data for α -Fe₂O₃ phase. No additional peaks belonging to other phases are observed, indicating the good crystallinity and high purity of the original α -Fe₂O₃ nanoparticles. After reduced for 2 hours in H₂/Ar atmosphere, the diffraction peaks for the as-prepared particles are in good agreement with the data for the cubic spinel structured Fe₃O₄ (JCPDS card no. 65-3107), demonstrating this reduction method is efficient for the phase transformation from α -Fe₂O₃ to Fe₃O₄. After grafted on RGO, the intensity of diffraction peaks of the Fe₃O₄/RGO nanocomposites are weakened compared with that of the Fe₃O₄ due to the RGO cover. Compared with Figure S1, a weak and broad peak at around 20° is the typical pattern of amorphous carbon, indicating the RGO structures. Little diffraction peaks belonging to FeO at 36.0°, 41.8°, 60.7°, 72.7°, 76.6° (JCPDS card no. 06-0615) and Fe at 44.6°, 65.0°, 82.3° (JCPDS card no. 06-0696) can be detected in the composites, due to hydrogen as a reducing gas can more easily penetrate the gap within the multistage structures, which help further accomplish the phase transformations from Fe₃O₄ to some FeO and Fe.

The TEM morphologies of a series of nanocomposites are shown in Fig. 3. The hydrolysis of the iron precursor with the help of Na₂HPO₄ leads to the monodispersed anisotropic spindle-like nanocrystalline α -Fe₂O₃ nanostructures, as shown in the inset of Fig. 3(a), with a average length of 200 nm and the outer diameter around 150 nm. After annealing treatment, the uniform dispersed spindle-shape particles are mainly destroyed and developed to bigger irregular Fe₃O₄ structures (Fig. 3(b)). Increasing the contents of GO, the α -Fe₂O₃ nanoparticles can be more evenly dispersed in the graphene layers. The α -Fe₂O₃ nanoparticles react with some polar functional groups such as hydroxyl, carboxyl or oleylamine and are slightly aggregated and grafted on the GO surfaces, ensuring the integrity of spindle-shaped Fe_3O_4 structures after annealing. From the TEM images, the GO has the typical crumpled structures with abundant wrinkles on the surface and scrolling on the edge of the nanosheets. Besides, the GO nanosheets are almost transparent in TEM pictures, indicating that they are very thin. The uniform spindle-like α -Fe₃O₃ nanoparticles shown in Fig. 3(c,e,g) are anchored onto the surfaces of graphene sheets, forming a cross-linked framework structure illustrated as Fig. 4. On one hand, the spindle-like nanoparticles prevent the GO sheets from folding; On the other hand, the curly GO sheets help to separate the α -Fe₂O₃ nanoparticles and consequently, prevent the Fe₃O₄ nanoparticles from agglomerating during annealing to form a homogeneous dispersion as shown in Fig. 3(b,d,h). When the amount of GO increases to 5 wt.%, the final Fe₃O₄ nanoparticles remain spindle-like morphologies as the original α -Fe₂O₃ nanoparticles, dispersing on the RGO nanosheet network.

Magnetic properties. The room-temperature magnetic hysteresis (*M*-*H*) loops in Fig. 5 show the magnetic variation for the samples from different processes. The value of magnetization saturation (*M*s) for the pure α -Fe₂O₃ nanoparticles is only 0.98 emu/g. It increases to 86.56 emu/g for the Fe₃O₄ then gradually deceases to 59.38 emu/g as increasing the ratio of the non-magnetic RGO nanosheets to 5 wt.%. Besides, the remnant magnetization (*M*r) and coercivity (*Hc*) of the samples are also shown as Table 1, do not reveal much variation after combining with RGO.

Absorption properties. A series of the pure Fe_3O_4 and Fe_3O_4/RGO nanocomposites are evaluated as high-frequency absorber. An efficient electromagnetic wave absorbing material should satisfy both strong absorbing and wide absorbing frequency band. The Fe_3O_4 units combined with RGO sheets to build a cross-linked framework may improve the wave absorbing performance. The frequency dependences of the complex permittivity (ε) and the complex permeability (μ) of the samples are shown in Figs 6 and 7, respectively. The real permittivity (ε') and real permeability (μ') represent the storage ability of electromagnetic energy, whereas the imaginary permittivity (ε'') and imaginary permeability (μ'') are connected with the energy dissipation or loss^{34,35}.



Figure 3. The TEM images of (a) the as-prepared α -Fe₂O₃, (b) Fe₃O₄ nanoparticles, (c) α -Fe₂O₃/1 wt.% GO, (d) Fe₃O₄/1 wt.% RGO, (e) α -Fe₂O₃/3 wt.% GO, (f) Fe₃O₄/3 wt.% RGO, (g) α -Fe₂O₃/5 wt.% GO and (h) Fe₃O₄/5 wt.% RGO nanocomposites.



Figure 4. Skeleton of the synthesis process of the Fe₃O₄/RGO composite.



Figure 5. Room-temperature magnetic hysteresis loops of the samples.



Figure 6. The frequency dependences of real and imaginary parts of the complex permittivities of the nanocomposites.

| Sample | Ms (emu/g) | Mr (emu/g) | Hc (Oe) |
|---|------------|------------|---------|
| α -Fe ₂ O ₃ | 0.98 | 0.18 | 285.11 |
| Fe ₃ O ₄ | 86.56 | 23.46 | 243.90 |
| Fe ₃ O ₄ /1 wt.%RGO | 67.46 | 19.94 | 287.19 |
| Fe ₃ O ₄ /3 wt.%RGO | 66.10 | 17.52 | 288.72 |
| Fe ₃ O ₄ /5 wt.%RGO | 59.38 | 18.62 | 290.99 |

Table 1. Magnetic properties of the samples.

SCIENTIFIC REPORTS | 6:25075 | DOI: 10.1038/srep25075



Figure 7. The frequency dependences of real and imaginary parts of the complex permeability of the nanocomposites.

Generally, the complex permittivity of the material shows frequency dispersion behaviour^{28,36}. As shown in Fig. 6, the values of ε' for Fe₃O₄ generally increase from 9.9 to 13.2 when the frequencies increase from 2.0 to 18.0 GHz, while the values of ε'' are almost under 1 and fluctuate as the frequencies increasing and reveal several resonance peaks. The behaviors should be attributed to the permittivity property and the special structure of the Fe₃O₄ nanoparticles. Since ε' is an expression of the polarizability of a material, which consists of dipolar polarization and electric polarization at microwave frequency¹⁷. The high ϵ' for Fe₃O₄ means high levers of the electric polarization and electric conductivity due to the electron transfer between Fe³⁺ and Fe²⁺ irons. And the resonance peaks in the ε'' curve demonstrates multi-relaxations also originating from the dipole polarization. After combined with RGO, the dielectric properties of the composites depend on that of each component and the interaction between. Particularly, the values of ε'' of a series of Fe₃O₄/RGO nanocomposites are higher than that of the pure Fe₃O₄. When the ratio of RGO is 3 wt.%, the ε'' curve have two high resonance peaks at 8.4 GHz and 16.5 GHz. The peaks root in the interfacial polarization, known as Maxwell-Wagner polarization in a heterogeneous media consisting of RGO and different conductivity or permittivity components³⁷. The complex permittivity of the composite with 5 wt.% RGO possessing the lowest values compared with those of 1 wt.% and 3 wt.% RGO is owing to the isolated RGO sheets will connect to each other in the composite when the RGO content is high enough, leading to a reduction of the electric dipole as the similar phenomena reported before²⁷.

Figure 7 shows the real part (μ') and imaginary part (μ'') of the complex permeability of the Fe₃O₄ and Fe₃O₄/RGO composites. The μ' of the composites generally decreases as the frequency increasing. It drops from 1.33 to 0.64 for the Fe₃O₄. For 1 wt.%, 3 wt.% and 5 wt.% RGO samples, it does from 1.27, 1.32, 1.33 to 0.97, 1.01, 0.97, respectively. Increasing the RGO ratio helps to remain the μ' value at the frequencies above 8.0 GHz. The μ'' of the composites appear similar trend to the μ' as the frequency increasing, showing large decrease in 4.7–8.5 GHz and serious fluctuations in the 8.5–18.0 GHz.

For most magnetic absorption materials, the magnetic loss could originate from the magnetic hysteresis, domain wall resonance, natural resonance, exchange resonance and eddy current effect^{38,39}. The magnetic hysteresis loss is negligible in weak field. The domain wall resonance usually occurs at a much lower frequency range in multi-domain materials. The eddy current loss is another important factor for electromagnetic microwave absorption. It is related to the electric conductivity (σ) and thickness (d) of the samples, which can be expressed by C_0^{27} :

$$C_0 = \mu''(\mu')^{-2} f^{-1} = 2\pi \mu_0 \sigma d^2 \tag{1}$$

where μ_0 is the permeability in a vacuum, σ is the electric conductivity of the material. If C_0 is a constant with the change of frequency, we can say that the magnetic loss results from the eddy current loss²⁷. As observed in Fig. 8, C_0 decreases with the increasing frequency and have serious fluctuations in the whole frequency range, implying that the eddy current effect has no significant effect on the electromagnetic microwave absorption.

Dielectric loss and magnetic loss are the two mainly possible contributors for the absorption, which can be expressed as $\tan \delta_{\varepsilon} = \varepsilon''/\varepsilon'$ and $\tan \delta_{\mu} = \mu''/\mu'$, respectively. It is very important to adjust the compatibility of the two kinds of loss to improve the absorption. Figure 9 shows the $\tan \delta_{\varepsilon}$ and $\tan \delta_{\mu}$ of the samples. It is clear that the Fe₃O₄/RGO nanocomposites possess higher dielectric losses than that of the Fe₃O₄ sample. The enhanced dielectric loss could stem from the enhanced interfacial polarization between the Fe₃O₄ nanoparticles and RGO sheets. For magnetic loss, the values of the Fe₃O₄/RGO nanocomposites are lower than that of the Fe₃O₄, exhibiting the same variation trend as μ'' . In view of the impedance matching characteristic for an absorber, well balance between the dielectric loss and magnetic loss could help to improve the absorbing performance, suggesting the lightweight graphene plays a key role in the improvement of the dielectric loss, which contributes to the absorption for the Fe₃O₄/RGO nanocomposites.

The reflection loss (RL) values are calculated using the measured complex permittivity ($\varepsilon_r = \varepsilon' - j\varepsilon''$) and complex permeability ($\mu_r = \mu' - j\mu''$) at the given frequencies and the absorber thicknesses according to the transmission line theory as follows⁴⁰:







Figure 9. The dielectric loss and magnetic loss of the nanocomposites.

$$Z_{in} = Z_0 \sqrt{\frac{\mu_r}{\varepsilon_r}} \tanh\left[\left(j\frac{2\pi ft}{c}\right)\sqrt{\mu_r}\frac{\varepsilon_r}{c}\right]$$
(2)

$$R_L = 20 \lg \left[\left| \frac{Z_{in} - Z_0}{Z_{in} + Z_0} \right| \right]$$
(3)

$$Z_0 = \sqrt{\frac{\mu_0}{\varepsilon_0}} \tag{4}$$

where Z_o (377 Ω) is the characteristic impedance of free space, Z_{in} is the input impedance of the absorber. ε_0 and μ_0 are the pemittivity and permeability of the free space, respectively. *f* is the frequency of the wave, *d* is the thickness of the absorber and *c* is the speed of light in free space. The results are illustrated in Fig. 10.

2

Figure 10 illustrates the reflection losses of the composites with different blending ratio. For the pure Fe₃O₄ composite, the image of reflection loss is shown in Fig. 10 (a). The peaks shift to high frequency with decrease of layer thickness and the optimal reflection loss can reach -38.1 dB at 14.8 GHz with a thickness of 1.7 mm. Combining Fe₃O₄ with 1 wt.% RGO, the optimal reflection loss is -28.2 dB at 8.5 GHz with a thickness of 2.7 mm (Fig. 10(b)). Adding 3 wt.% RGO makes it possess an optimal reflection loss of -65.1 dB at 15.2 GHz with a thickness of 1.7 mm (Fig. 10(c)). Further increasing the RGO to 5 wt.%, the optimal reflection loss becomes about -21.0 dB at 5.3 GHz with a thickness of 5.0 mm (Fig. 10(d)). Hence, addition of RGO with a proper content enhances the electromagnetic (EM) performance on the whole, which is ascribed to several aspects as indicated below. First, the RGO provides tremendous electric dipoles which react with high-frequency EM wave and convert EM energy to thermal energy. Second, the interfaces brought in have a dominant role in enhancing dielectric performance and also cause multiple reflections, further consuming the EM energy. Moreover, the introduction of RGO ameliorates the impedance matching to some degree so as to modify the EM absorbing performance³⁶. In general, the composites with RGO exhibit multiple absorbing peaks at several points of the frequency and thickness. That is to say, RGO incorporated may expand the absorbing bandwidth and improve the reflection loss even with a smaller thickness.



Figure 10. The reflection losses of the composites with different ratio between Fe_3O_4 and RGO.



Figure 11. The bandwidths of the composites of 1.7 mm with different ratio between Fe_3O_4 and RGO.

When the thickness is 1.7 mm, the reflection losses versus frequency of the composites are shown in Fig. 11. The bandwidth of the $Fe_3O_4/3$ wt.% RGO composite for which the reflection loss is higher than -10 dB is from 13.4 GHz to over 18.0 GHz that larger than that of other samples, demonstrating wide range absorbing property. Therefore, adding proper content of RGO can increase both the reflection loss and the absorbing bandwidth, demonstrating the anisotropic Fe_3O_4/RGO nanocomposites are of high performance for high-frequency wave absorbing.

Conclusions

In summary, anisotropic Fe_3O_4 nanoparticle and a series of Fe_3O_4/RGO nanocomposites have been successfully prepared. The Fe_3O_4/RGO nanocomposites exhibit high-performance microwave absorption properties over 2.0–18.0 GHz. Combining with RGO, the spindle-like Fe_3O_4 nanoparticles evenly dispersed in the graphene layers and are retarded from aggregating during annealing. The grafted composites possess higher dielectric losses than that of the pure Fe_3O_4 specimen, due to well balance between the dielectric loss and magnetic loss contribute to

the high absorbing performance. The optimal reflection loss of the pure Fe_3O_4 composite is -38.1 dB at 14.8 GHz with a thickness of 1.7 mm. While it reaches -65.1 dB at 15.2 GHz with a thickness of 1.7 mm for the $Fe_3O_4/3 \text{ wt.\%}$ RGO composite. The improved absorption arises from the synergy of dielectric loss and magnetic loss, as well as the enhancement of multiple interfaces among graphene. Adding proper content of RGO can increase both the reflection loss and the absorbing bandwidth, suggesting the $Fe_3O_4/\text{graphene}$ nanocomposites are one kind of the prosperous candidates for EM wave absorbing.

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Acknowledgements

This work was supported by the National Science Foundation of China (Grant Nos 51171007, 51271009, and 61227902).

Author Contributions

Y.Y. prepared the samples and wrote the main manuscript text; M.Z. and R.Y. advised and supported in preparing the manuscript; J.L. supported in TEM image observation; W.T., H.D. and R.X. gave helps in the experiment. All authors have reviewed the manuscript.

Additional Information

Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Yin, Y. *et al.* Enhanced high-frequency absorption of anisotropic Fe₃O₄/graphene nanocomposites. *Sci. Rep.* **6**, 25075; doi: 10.1038/srep25075 (2016).

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