



OPEN

SUBJECT AREAS:
ELECTRONIC PROPERTIES
AND MATERIALS
ELECTRONIC MATERIALS

Received 28 February 201*4*

> Accepted 17 April 2014

> > Published 8 May 2014

Correspondence and requests for materials should be addressed to W.C. (phycw@nus. edu.sg)

Gap States Assisted MoO₃ Nanobelt Photodetector with Wide Spectrum Response

Du Xiang¹, Cheng Han¹, Jialin Zhang¹ & Wei Chen^{1,2,3,4}

¹Department of Physics, National University of Singapore, 2 Science Drive 3, Singapore, 117542, ²Department of Chemistry, National University of Singapore, 3 Science Drive 3, Singapore, 117543, ³Graphene Research Centre, National University of Singapore, 3 Science Drive 3, Singapore, 117543, ⁴National University of Singapore (Suzhou) Research Institute, 377 Lin Quan Street, Suzhou Industrial Park, Jiang Su, China, 215123.

Molybdenum oxides have been widely investigated for their broad applications ranging from electronics to energy storage. Photodetectors based on molybdenum trioxide (MoO₃), however, were seldom reported owing to their low conductivity and weak photoresponse. Herein we report a photodetector based on single MoO₃ nanobelt with wide visible spectrum response by introducing substantial gap states via H_2 annealing. The pristine MoO₃ nanobelt possessed low electrical conductance and no photoresponse for nearly all visible lights. The H_2 annealing can significantly improve the conductance of MoO₃ nanobelt, and result in a good photodetector with wide visible spectrum response. Under illumination of 680 nm light, the photodetector exhibited high responsivity of ~56 A/W and external quantum efficiency of ~10200%. As corroborated by in situ ultraviolet photoelectron spectroscopy and X-ray photoelectron spectroscopy investigations, such strong wide spectrum photoresponse arises from the largely enriched gap states in the MoO₃ nanobelt after H_2 annealing.

etal oxides nanomaterials have been widely used in optoelectronic nanodevices¹, solar cells² and photocatalysis³. However, their wide bandgap limits the applications in ultraviolet region⁴⁻⁶. It has been demonstrated that generating mid-gap states in these wide bandgap semiconductors can extend the photoactive region to visible or even infrared range, and hence significantly improve the efficiency of the optoelectronic devices and photocatalysts⁷⁻¹⁰. In order to produce substantial gap states in these wide bandgap semiconductors, various approaches have been proposed, including intercalating metal or nonmetal dopants in the wide bandgap semiconductors to introduce donor or acceptor states in various positions above the valance band and altering the degree of doping to modify the gap states 11-16. For metal oxides, one effective way to generate gap states is to remove oxygen ions in the lattice, and hence the formation of oxygen vacancies. These oxygen vacancies are vitally important to determine the electronic and optical properties of metal oxides 17-19. Annealing the nanostructures of metal oxides in reducing gas is effective to obtain such oxygen vacancies^{19,20}. Wang et al. demonstrated that oxygen vacancies were generated in rutile TiO₂ nanowire arrays by annealing the samples in H₂ atmosphere. These oxygen vacancies served as donor states to strongly improve the light absorption⁹. Davazoglou et al. succeeded to utilize oxygen vacancies in WO₃ and MoO₃ films based organic light-emitting diodes and solar cells to improve their performance^{17,18}. Recently, it was found that introducing large amounts of lattice disorder in nanophase TiO2 can generate substantial gap states, and hence extend the light absorption edge to \sim 1200 nm, thereby leading to the remarkably enhanced photocatalytic efficiency^{21,22}.

Attributed to the reduced dimensionality and large surface-to-volume ratio, photodetectors based on one dimensional (1D) nanomaterials possess two major advantages compared to their bulk counterparts, including high sensitivity and high quantum efficiency^{23–25}. However, the photodetectors based on 1D nanomaterials with large bandgap only works under the light with narrow spectra range⁵. Introducing considerable gap states in such wide bandgap 1D nanomaterials can help broadening their photoresponse spectra region.

The molybdenum trioxide (MoO₃), as an intrinsic n-type II-VI semiconductor with wide bandgap (\sim 3.2 eV), has been extensively utilized in organic electronics as efficient anode interfacial layers owing to its high work function²⁶. Moreover, the MoO₃ nanostructures have also been heavily investigated as effective photocatalyst in pollution degradation^{27,28}. However, due to their low intrinsic conductivity and weak photoresponse²⁹, MoO₃ based optoelectronic nanodevices are rarely reported. In this paper, a photodetector with wide visible spectrum



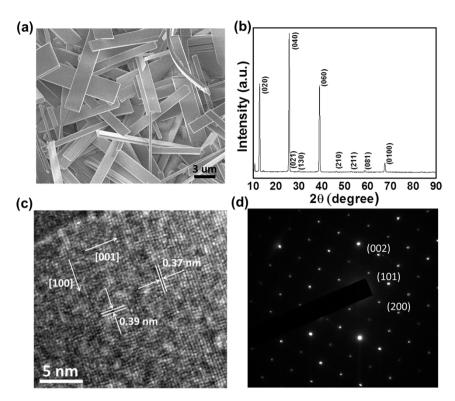


Figure 1 | Morphology and lattice structure of the as-grown MoO_3 nanobelts. (a) SEM image and (b) XRD pattern of the as-grown MoO_3 nanobelts. (c) HRTEM image of a single nanobelt, and (d) the corresponding SAED pattern.

response based on single $\mathrm{MoO_3}$ nanobelt treated by annealing in $\mathrm{H_2}$ was proposed and carefully examined. The intrinsic $\mathrm{MoO_3}$ nanobelt device exhibited low electrical conductance and no photoresponse for the visible spectrum. After $\mathrm{H_2}$ annealing, the conductance of $\mathrm{MoO_3}$ nanobelt was largely enhanced; at the same time, the photodetector possessed wide visible spectrum response. The responsivity and external quantum efficiency of the photodetector under the illumination of 680 nm light can reach as high as 56 A/W and 10200%, respectively. *In situ* ultraviolet photoelectron spectroscopy (UPS) and X-ray photoelectron spectroscopy (XPS) measurements indicate the significant enrichment of gap states in $\mathrm{MoO_3}$ after $\mathrm{H_2}$ annealing, thereby leading to the excellent photoresponse in the wide visible spectra region.

Results

Figure 1a displays a typical SEM image of as-grown MoO $_3$ nanobelts. The sample showed the widths ranging from 1 to 4 um and lengths

from 10 to 25 um. The average thickness of nanobelts was about 100 nm. The XRD pattern (shown in figure 1b) is in good agreement with the orthorhombic structure of MoO_3 phase, with lattice constants of a = 3.96 Å, b = 13.86 Å, and c = 3.7 Å (JCPDS 05-0508). The high-resolution transmission electron microscopy (HRTEM) and selected area electron diffraction (SAED) images of individual nanobelt are shown in figure 1c and 1d, respectively. The TEM information revealed that the nanobelt was single crystalline with longitudinal direction preferentially along the <001> direction.

In order to explore the electrical transport properties of as-grown MoO₃ nanobelts, the single nanobelt was configured with two Cr/Au (50 nm/100 nm) contacts via the conventional e-beam lithography (EBL) process. The inset of Figure 2a shows the SEM image of a typical fabricated device with the conduction channel length of 6 μ m. The typical current-voltage (I–V) characteristic of the fabricated device is illustrated in Figure 2a. The good linearity of I–V curve reveals the ohmic contact between electrodes and MoO₃

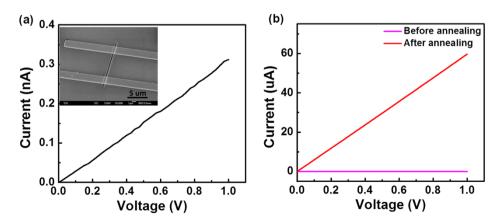
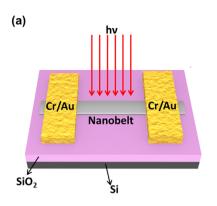


Figure 2 | Electronic measurements of MoO₃ nanobelt device before and after H_2 annealing. (a) I-V curve of the single MoO₃ nanobelt before annealing. The inset shows a SEM image of the single nanobelt device. (b) Comparison of the I-V curves before and after H_2 annealing.





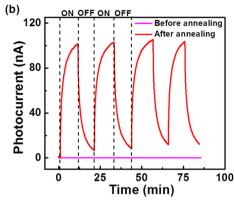


Figure 3 | Photodetector based on single MoO_3 nanobelt. (a) Schematic illustration of MoO_3 nanobelt device for photocurrent measurement. (b) Time dependent photoresponse of MoO_3 device before and after annealing under 660 nm laser illumination at 0.1 V bias voltage.

nanobelt. The conductance was calculated to be $\sim\!\!3.14\times10^{-10}$ S. Through the annealing treatment in H_2 atmosphere, the conductance of the MoO_3 nanobelt device dramatically increased to $\sim\!\!5.96\times10^{-5}$ S by 5 orders of magnitude, as demonstrated in Figure 2b.

The schematic diagram of single MoO₃ nanobelt based photodetector is exhibited in Figure 3a. Figure 3b shows the time dependent photocurrent measurements of MoO₃ nanobelt before and after H₂ annealing via alternately switching on and off a 660 nm laser with the power density of 112.3 mW/cm². The photocurrent (ΔI) is defined as: $\Delta I = I_p - I_d$, where I_d , I_p represents the current at bias voltage of 0.1 V in the dark and under light illumination, respectively. Prior to annealing process, the pristine MoO₃ nanobelt exhibited nearly no photoresponse illuminated by the incident light. In contrast, the significant photoresponse was detected for the H₂ annealed nanobelt with the excellent reproducibility and apparent photocurrent as high as ~100 nA. This remarkable photoresponse of the annealed device was also observed upon the illumination over a wide visible spectrum; while the MoO₃ nanobelt before annealing demonstrated almost zero photocurrent under these visible lights with different wavelength.

The relationship between photocurrent and intensity of incident light for $\rm H_2$ annealed $\rm MoO_3$ nanobelt was also examined. Figure 4a exhibits the real-time photoresponse of annealed nanobelt irradiated by a 560 nm laser with varying intensities. The photocurrent increased from 8.9 nA to 45.6 nA by increasing the laser intensity from 6.5 mW/cm² to 68.5 mW/cm². The corresponding photocurrent versus light intensity plot is shown in Figure 4b. It indicates that the photocurrent increases almost linearly as a function of the intensity of incident light. It is believed that the density of

photo-induced charge carrier and hence the photocurrent linearly depends on the absorbed photo $flux^{23}$, in good agreement with our experimental results. Such linear dependence of photocurrent as a function of light intensity reveals the potential application of the annealed MoO_3 nanobelt as light power detectors.

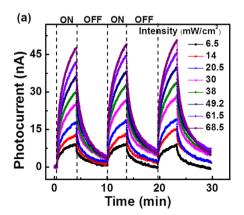
In order to probe the wavelength dependence of the photosensitivity of MoO_3 nanobelt photodetector, the time dependent photoresponse of the annealed device was measured under the exposure to visible lights of selected wavelengths ranging from 400 nm to 700 nm with the same intensity of 5.6 mW/cm² (as shown in Figure 5a–5c). The significant photocurrent under these visible lights with different wavelength indicates the wide spectrum response of the annealed nanobelt photodetector. Moreover, nearly reserved photocurrent through the visible spectrum was observed, suggesting the uniform visible light photoresponse for the annealed MoO_3 nanobelt.

The spectra responsivity (R_{λ}) and external quantum efficiency (EQE) are two critical parameters to evaluate the quality of photodetectors, where R_{λ} is defined as the photocurrent generated per unit power of incident light on the effective area of a photodetector, and EQE is the number of electrons detected per incident photon. The large values of R_{λ} and EQE suggest high sensitivity for photodetectors. R_{λ} and EQE can be expressed as³⁰:

$$R_{\lambda} = \Delta I_{\lambda} / (P_{\lambda} S) \tag{1}$$

$$EQE = hcR_{\lambda}/(e\lambda) \tag{2}$$

where ΔI_{λ} is the photocurrent induced by the incident light of wavelength λ , P_{λ} is the light intensity, S is the effective illuminated area,



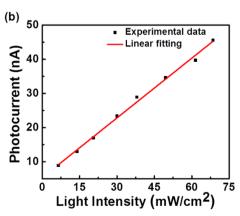


Figure 4 | Photocurrent versus light intensity of annealed MoO₃ nanobelt. (a) Time dependent photoresponse of MoO₃ nanobelt device after annealing under 532 nm laser with varying intensities at bias voltage of 0.1 V. (b) Plot of the photocurrent as a function of laser intensity.



and h, c, e represent the Plank constant, velocity of light, and charge of electron, respectively. At the bias voltage of 0.1 V, the R_{λ} of H₂ annealed MoO3 nanobelt for the selected visible wavelengths was calculated to be in the range of 55 to 56 A/W (shown in figure 5d). This is much higher than many reported photodetectors based on both 1D and two dimensional (2D) materials, such as ZnS nanobelts $(\sim 0.12 \text{ A/W})^{31}$, SbSe₃ nanowires $(\sim 8.0 \text{ A/W})^{32}$, ZnSe nanobelts (~20 A/W)³³ and 2D materials, such as graphene (~1 mA/W)³⁴, single layer MoS₂ (~7.5 mA/W)³⁵, multi-layer GaS nanosheets (~4.2 A/W)³⁶, but still significantly lower than that of In₂Se₃ nanowires (~89 A/W)³⁷, ZnTe nanowires (~360 A/W)³⁸, and CdSe nanobelts (~1400 A/W)³⁹ based photodetectors. Moreover, the EQE of annealed device can be determined as high as 16300% for the wavelength of 420 nm, and gradually decreased to 10200% as the wavelength of incident light increased to 680 nm, revealing superior device performance of the MoO₃ nanobelt based photodetector.

Discussion

To further investigate the mechanism of strong photoresponse to wide visible spectrum for $\rm H_2$ annealed $\rm MoO_3$ nanobelt photodetector, XRD and $in \, situ \, \rm XPS/UPS$ measurements were conducted on asgrown $\rm MoO_3$ nanobelts and thermally deposited $\rm MoO_3$ thin film before and after the $\rm H_2$ annealing process, respectively. As shown in Figure S1, $\rm H_2$ annealing did not induce any crystal structure change of the $\rm MoO_3$ nanobelts. Figure 6a and 6b show the Mo 3d core level XPS spectra of the $in \, situ \, grown \, MoO_3 \, film \, (10 \, nm)$ before and after $\rm H_2$ annealing, respectively. These two core level spectra were fitted with Gaussian/Lorentzian mixed functions. In Figure 6a, the Mo $\rm 3d_{5/2}$ and $\rm 3d_{3/2}$ peaks located at the binding energy of 232.11 eV and 235.21 eV can be assigned to the 6+ oxidation state of $\rm MoO_3$ phase, in accordance with the previous reports⁴⁰. This suggests that $\rm Mo^{6+}$ dominates the $\rm MoO_3$ layer before the $\rm H_2$

annealing. After $\rm H_2$ annealing, the Mo 3d peaks were apparently broadened arising from the appearance of $\rm Mo^{5+}$ oxidation state (as shown in Figure 6b). This reveals that large quantity of oxygen vacancies were introduced in $\rm MoO_3$ through $\rm H_2$ annealing, reducing the Mo atoms neighboring to the oxygen vacancies from the 6+ state to the 5+ state. The corresponding O 1s XPS spectra shown in Figure S2 also indicate the significant enhancement of oxygen vacancies after annealing, in good agreement with Mo 3d XPS spectra. The $\rm H_2$ annealing can also significantly increase the charge carrier (electron) concentration and induce obvious n-type doping of $\rm MoO_3$, thereby significantly enhancing the conductivity of $\rm MoO_3$ nanobelts.

H₂ annealing of MoO₃ can induce the formation of substantial gap states to facilitate the aforementioned wide-range visible light response in MoO₃ nanobelt based photodetectors. This hypothesis can be corroborated by in-situ UPS measurements. Figure 7a shows the UPS spectra of MoO₃ thin film before and after H₂ annealing at the low binding energy region near the Fermi level. By linear extrapolation of the low binding energy onset, the valence band edge of MoO₃ layer without annealing was measured to be ~2.56 eV. After H₂ annealing, the valence band edge was located at ~2.89 eV below the Fermi level. This indicates that the Femi level moved 0.33 eV closer to the conduction band and hence a more significant n-type doping of MoO₃, in agreement with the XPS results. After H₂ annealing, the intensity of the gap states located between the Femi level and the valence band edge was significantly enhanced. Moreover, these gap states substantially extended towards the Fermi level. As shown by the energy level diagram of MoO₃ before and after annealing in Figure 7b, upon light illumination, such annealing process induced gap states offer many possible routes for electrons to be excited from gap states to the conduction band. This can significantly improve the photoresponse under the illumination of visible lights with different wavelength, making H₂ annealed MoO₃ nanobelt as an effective photodetector with wide spectrum response.

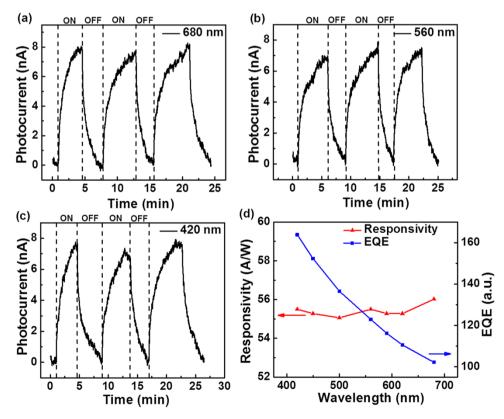


Figure 5 | Wavelength dependence of annealed MoO_3 nanobelt photodetector. Photoresponse of annealed MoO_3 nanobelt device under the light with different wavelength: (a) 680 nm (b) 560 nm (c) 420 nm. The intensity of light is kept the same at 5.6 mW/cm². (d) Plot of the responsivity and EQE versus light wavelength.



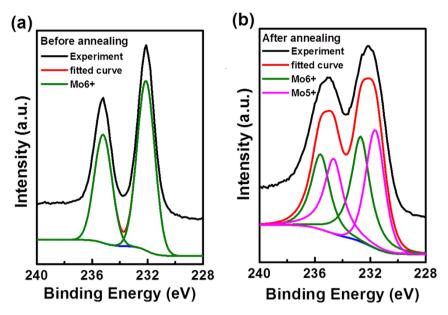


Figure 6 | XPS investigation. XPS spectra of MoO_3 film for Mo 3d core level (a) before and (b) after annealing. The experiment data are fitted with the Gaussian/Lorentzian mixed functions.

In conclusion, we report a MoO_3 nanobelt based photodetector with wide spectrum response in the visible light region assisted through the H_2 annealing induced gap states with high density. The as-grown MoO_3 nanobelt exhibited low conductance and nearly no photoresponse under visible light irradiation. After H_2 annealing, the conductance of MoO_3 nanobelt was dramatically enhanced; moreover, the photodetector possessed wide visible spectrum photoresponse with high responsivity and EQE. As corroborated by *in situ* XPS and UPS measurements, such excellent photodetector with wide spectrum response mainly resulted from the significantly enriched gap states in H_2 annealed MoO_3 nanobelt. This work demonstrates the possibility to extend the wide bandgap metal oxide nanomaterials based optoelectronics devices or photocatalysts with efficient visible light response through the introduction of the high intensity of carefully engineered gap states.

Methods

Material Preparation and Characterization. The MoO $_3$ nanobelts were synthesized by adopting the previously reported method⁴¹. A molybdenum foil (size of 10 mm \times

 $10\,$ mm \times 0.05 mm, 99.9% Mo) was used as the Mo source to grow MoO3 nanobelts. Firstly, the molybdenum foil was polished to remove the oxide layer and washed in acetone and distilled water via sonication. It was then placed on a ceramic digital stirring hotplate with a glass slide (35 mm \times 50 mm \times 150 um in size) covering on it. The hotplate was heated at 480°C for 2 days in the air ambient. After heating, the hotplate was allowed to cool down to room temperature. MoO3 nanobelts were grown on the glass slide. The nanostructures were characterized by scanning electron microscope (JEOL JSM-6400F), X-ray diffraction (Philip PW 127), and transmission electron microscope (JEOL TEM 2010F).

Device Fabrication and Characterization. Single MoO $_3$ nanobelt based device was fabricated by the standard lithography procedures. The as-grown MoO $_3$ nanobelts on glass were dispersed in ethanol by sonication. The nanobelts suspension was subsequently dropped on the heavily p-doped Si substrate (resistivity < 0.005 Ω ·cm) with 300 nm thermal oxide followed by drying under nitrogen. Two electrodes with bonding pads were precisely pattered on the single nanobelt using the conventional ebeam lithography (EBL) technique, followed by thermal deposition of Cr (50 nm) and Au (100 nm) bilayer as the metal contact. After lift-off process, the fabricated devices were wire-bonded on a LCC chip carrier for electrical measurements. The annealing process of as-made MoO $_3$ nanobelt devices was conducted in H $_2$ /Ar (10%) at 300°C for 1 hour. All the electrical and optoelectronic measurements were carried out in high vacuum (\sim 10 $^{-8}$ mbar) using an Agilent B2912A source measurement unit. The light sources utilized in our experiments contain 660 nm laser, 532 nm

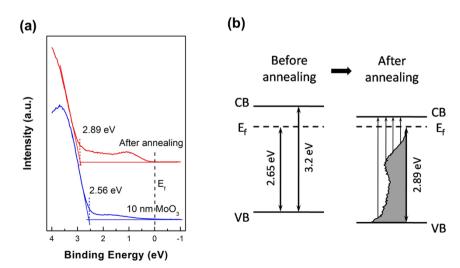


Figure 7 | UPS characterization. (a) UPS spectra of the low binding energy region near the Feimi level for MoO₃ film before and after annealing. (b) Schematic diagram of the energy level alignment for MoO₃ film before and after annealing.



laser, and 500 W xenon light source configured with a monochromator to give a continuous spectrum output. The power of the incident light was calibrated by THORLABS GmbH (PM 100A) power meter.

XPS and UPS Measurements. MoO₃ thin film was grown on the Si (111) substrate coated with native oxide layer (1–10 Ω ·cm) via thermal evaporation in an ultra-high-vacuum (UHV) chamber with a base pressure of $\sim 2 \times 10^{-9}$ mbar. The highly purified MoO₃ source was thermally evaporated onto Si substrate from a Knudsen cell (Creaphys, Germany) at the temperature of 490°C. The thickness of the grown MoO₃ layer was estimated by the attenuation of Si 2p peak and further calibrated by a quartz crystal microbalance (QCM). *In situ* XPS and UPS measurements were carried out in an analysis chamber of base pressure $\sim 1 \times 10^{-10}$ mbar with Al K α (1486.6 eV) and He I (21.2 eV) as the excitation source. The as-grown MoO₃ thin film was *in situ* annealed in H₂ atmosphere at 300°C under the pressure of 5×10^{-5} mbar for 1 hour.

- Ohta, H. & Hosono, H. Transparent oxide optoelectronics. *Mater. Today* 7, 42–51 (2004).
- Jose, R., Thavasi, V. & Ramakrishna, S. Metal Oxides for Dye-Sensitized Solar Cells. J. Am. Ceram. Soc. 92, 289–301 (2009).
- Chen, X., Shen, S., Guo, L. & Mao, S. S. Semiconductor-based Photocatalytic Hydrogen Generation. Chem. Rev. 110, 6503–6570 (2010).
- Linsebigler, A. L., Lu, G. & Yates, J. T. Photocatalysis on TiO₂ Surfaces: Principles, Mechanisms, and Selected Results. *Chem. Rev.* 95, 735–758 (1995).
- Kind, H. et al. Nanowire Ultraviolet Photodetectors and Optical Switches. Adv. Mater. 14, 158–160 (2002).
- Mathur, S. et al. Size-Dependent Photoconductance in SnO₂ Nanowires. Small 1, 713–717 (2005).
- Luque, A. & Martí, A. Increasing the Efficiency of Ideal Solar Cells by Photon Induced Transitions at Intermediate Levels. *Phys. Rev. Lett.* 78, 5014–5017 (1997).
- Asahi, R. et al. Visible-Light Photocatalysis in Nitrogen-Doped Titanium Oxides. Science 293, 269–271 (2001).
- 9. Wang, G. et al. Hydrogen-Treated ${\rm TiO_2}$ Nanowire Arrays for Photoelectrochemical Water Splitting. Nano Lett. 11, 3026–3033 (2011).
- Li, J. et al. 35% efficient nonconcentrating novel silicon solar cell. Appl. Phys. Lett. 60, 2240–2242 (1992).
- Lee, H.-J., Jeong, S.-Y., Cho, C. R. & Park, C. H. Study of diluted magnetic semiconductor: Co-doped ZnO. Appl. Phys. Lett. 81, 4020–4022 (2002).
- Park, C. H., Zhang, S. B. & Wei, S.-H. Origin of p type doping difficulty in ZnO: The impurity perspective. *Phys. Rev. B* 66, 073202 (2002).
- 13. Hoffmann, M. R., Martin, S. T., Choi, W. & Bahnemann, D. W. Environmental Applications of Semiconductor Photocatalysis. *Chem. Rev.* **95**, 69–96 (1995).
- Khan, S. U. M., Al-Shahry, M. & Ingler, W. B. Efficient Photochemical Water Splitting by a Chemically Modified n-TiO₂. Science 297, 2243–2245 (2002).
- 15. Van de Walle, C. G. Hydrogen as a Cause of Doping in Zinc Oxide. *Phys. Rev. Lett.* **85**, 1012–1015 (2000).
- 16. Yin, J. B. & Zhao, X. P. Preparation and Enhanced Electrorheological Activity of TiO₂ Doped with Chromium Ion. Chem. Mater. 16, 321–328 (2003).
- 17. Vasilopoulou, M. et al. The Influence of Hydrogenation and Oxygen Vacancies on Molybdenum Oxides Work Function and Gap States for Application in Organic Optoelectronics. J. Am. Chem. Soc. 134, 16178–16187 (2012).
- Vasilopoulou, M. et al. Reduction of Tungsten Oxide: A Path Towards Dual Functionality Utilization for Efficient Anode and Cathode Interfacial Layers in Organic Light-Emitting Diodes. Adv. Funct. Mater. 21, 1489–1497 (2011).
- Janotti, A. et al. Hybrid functional studies of the oxygen vacancy in TiO₂. Phys. Rev. B 81, 085212 (2010).
- Hsu, H. S. et al. Evidence of oxygen vacancy enhanced room-temperature ferromagnetism in Co-doped ZnO. Appl. Phys. Lett. 88, 242507–242507-3 (2006).
- Chen, X., Liu, L., Yu, P. Y. & Mao, S. S. Increasing Solar Absorption for Photocatalysis with Black Hydrogenated Titanium Dioxide Nanocrystals. *Science* 331, 746–750 (2011).
- 22. Liu, L. et al. Hydrogenation and Disorder in Engineered Black TiO2. Phys. Rev. Lett. 111, 065505 (2013).
- Soci, C. et al. ZnO Nanowire UV Photodetectors with High Internal Gain. Nano Lett. 7, 1003–1009 (2007).
- 24. Li, L. *et al.* Electrical Transport and High-Performance Photoconductivity in Individual ZrS₂ Nanobelts. *Adv. Mater.* **22**, 4151–4156 (2010).
- 25. Lieber, C. M. & Wang, Z. L. Functional Nanowires. MRS Bull. 32, 99–108 (2007).

- 26. Kyaw, A. K. K. et al. An inverted organic solar cell employing a sol-gel derived ZnO electron selective layer and thermal evaporated MoO₃ hole selective layer. Appl. Phys. Lett. 93, 221107–221107-3 (2008).
- Papp, J., Soled, S., Dwight, K. & Wold, A. Surface Acidity and Photocatalytic Activity of TiO₂, WO₃/TiO₂, and MoO₃/TiO₂ Photocatalysts. *Chem. Mater.* 6, 496–500 (1994).
- Song, K. Y. et al. Preparation of Transparent Particulate MoO₃/TiO₂ and WO₃/TiO₂ Films and Their Photocatalytic Properties. Chem. Mater. 13, 2349–2355 (2001).
- Balendhran, S. et al. Enhanced Charge Carrier Mobility in Two-Dimensional High Dielectric Molybdenum Oxide. Adv. Mater. 25, 109–114 (2013).
- Li, L. et al. Single-Crystalline CdS Nanobelts for Excellent Field-Emitters and Ultrahigh Quantum-Efficiency Photodetectors. Adv. Mater. 22, 3161–3165 (2010)
- Fang, X. et al. Single-Crystalline ZnS Nanobelts as Ultraviolet-Light Sensors. Adv. Mater. 21, 2034–2039 (2009).
- Zhai, T. et al. Single-Crystalline Sb₂Se₃ Nanowires for High-Performance Field Emitters and Photodetectors. Adv. Mater. 22, 4530–4533 (2010).
- Salfi, J. et al. Electrical properties of Ohmic contacts to ZnSe nanowires and their application to nanowire-based photodetection. Appl. Phys. Lett. 89, 261112–261112-3 (2006).
- Xia, F. et al. Photocurrent Imaging and Efficient Photon Detection in a Graphene Transistor. Nano Lett. 9, 1039–1044 (2009).
- 35. Yin, Z. et al. Single-Layer MoS₂ Phototransistors. ACS Nano 6, 74–80 (2011).
- 36. Hu, P. et al. Highly Responsive Ultrathin GaS Nanosheet Photodetectors on Rigid and Flexible Substrates. Nano Lett. 13, 1649–1654 (2013).
- Zhai, T. et al. Fabrication of High-Quality In₂Se₃ Nanowire Arrays toward High-Performance Visible-Light Photodetectors. ACS Nano 4, 1596–1602 (2010).
- Li, Z. et al. Room temperature single nanowire ZnTe photoconductors grown by metal-organic chemical vapor deposition. Appl. Phys. Lett. 97, 063510–063510-3 (2010).
- Dai, Y. et al. High-performance CdSe nanobelt based MESFETs and their application in photodetection. J. Mater. Chem. 22, 18442–18446 (2012).
- Zhong, J. Q. et al. Ionization potential dependent air exposure effect on the MoO₃/ organic interface energy level alignment. Org. Electron. 13, 2793–2800 (2012).
- 41. Xie, Y. L. *et al.* Rainbow-like MoO₃ Nanobelts Fashioned via AFM Micromachining. *J. Phys. Chem. C* **114**, 120–124 (2009).

Acknowledgments

Authors acknowledge the technical support from NUS Graphene Research Centre for the device fabrication, and financial support from Singapore MOE Grants R143-000-505-112, R143-000-542-112 and R143-000-559-112.

Author contributions

D.X. and C.H. contributed equally to this paper. D.X. and W.C. designed the experiments. D.X., C.H. and J.Z. performed the experiments. D.X., C.H. and W.C. wrote the main manuscript text. D.X., C.H. and J.Z. prepared figures 1-7. D.X. prepared figures S1 and S2. All authors reviewed the manuscript.

Additional information

Supplementary information accompanies this paper at http://www.nature.com/scientificreports

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Xiang, D., Han, C., Zhang, J.L. & Chen, W. Gap States Assisted MoO₃ Nanobelt Photodetector with Wide Spectrum Response. *Sci. Rep.* 4, 4891; DOI:10.1038/srep04891 (2014).



This work is licensed under a Creative Commons Attribution 3.0 Unported License. The images in this article are included in the article's Creative Commons license, unless indicated otherwise in the image credit; if the image is not included under the Creative Commons license, users will need to obtain permission from the license holder in order to reproduce the image. To view a copy of this license, visit http://creativecommons.org/licenses/by/3.0/