# communications chemistry

# ARTICLE

https://doi.org/10.1038/s42004-022-00797-y

OPEN

Check for updates

# Surface engineering of zinc phthalocyanine organic thin-film transistors results in part-per-billion sensitivity towards cannabinoid vapor

Zachary J. Comeau (1,2, Rosemary R. Cranston<sup>1</sup>, Halynne R. Lamontagne (1,2, Cory S. Harris<sup>2,3</sup>, Adam J. Shuhendler<sup>2,3,4 \vee &</sup> Benoît H. Lessard (1,5 \vee )

Phthalocyanine-based organic thin-film transistors (OTFTs) have been demonstrated as sensors for a range of analytes, including cannabinoids, in both liquid and gas phases. Detection of the primary cannabinoids,  $\Delta^9$ -tetrahydrocannabinol (THC) and cannabidiol (CBD), is necessary for quality control and regulation, however, current techniques are often not readily available for consumers, industry, and law-enforcement. The OTFT characteristics, X-ray diffraction (XRD) spectra, and grazing incident wide angle x-ray scattering (GIWAXS) spectra of two copper and three zinc phthalocyanines, with varying degrees of peripheral fluorination, were screened to determine sensitivity to THC vapor. Unsubstituted ZnPc was found to be the most sensitive material and, by tuning thin-film morphology, crystal polymorphs, and thickness through altered physical vapor deposition conditions, we increased the sensitivity to THC by 100x. Here we demonstrate that deposition conditions, and the resulting physical film characteristics, play a significant role in device sensitization.

<sup>&</sup>lt;sup>1</sup> Department of Chemical and Biological Engineering, University of Ottawa, 161 Louis Pasteur, K1N 6N5 Ottawa, ON, Canada. <sup>2</sup> Department of Chemistry and Biomolecular Sciences, University of Ottawa, 150 Louis Pasteur, K1N 6N5 Ottawa, ON, Canada. <sup>3</sup> Department of Biology, University of Ottawa, 30 Marie Curie, K1N 6N5 Ottawa, ON, Canada. <sup>4</sup> University of Ottawa Heart Institute, 40 Ruskin St, Ottawa K1Y 4W7 ON, Canada. <sup>5</sup> School of Electrical Engineering and Computer Science, University of Ottawa, 800 King Edward Ave., K1N 6N5 Ottawa, ON, Canada. <sup>Se</sup>email: Adam.Shuhendler@uottawa.ca; benoit. lessard@uottawa.ca

hthalocyanines (Pcs) and their derived metal complexes (MPcs) are macrocyclic organic compounds with a variety of industrial applications due to their useful spectral and electronic properties<sup>1,2</sup>. First reported in 1907, Pcs were highlighted for their excellent stability and brilliant color, finding extensive use as dyes and pigments<sup>3-5</sup>. As conjugated aromatic molecules, Pcs are often deposited as charge transport lavers within organic thin-film transistors (OTFTs) and organic photovoltaics (OPVs)<sup>6,7</sup>. Pc-based OTFTs have been demonstrated as sensors for a variety of liquid and gas sensing applications<sup>8–12</sup>, including our groups' recent demonstration of ratiometric detection and differentiation of  $\Delta^9$ -tetrahydrocannbinol (THC) and cannabidiol (CBD)<sup>13,14</sup>. The dominant active cannabinoids in Cannabis sativa smoke/vapour and consumer products, THC and CBD are used for therapeutic and recreational purposes<sup>15,16</sup>, however, as THC and CBD elicit different pharmacological effects, accurate, low-cost quantification and speciation is of interest to industry, law enforcement and consumers<sup>17</sup>. Quantitative commercial speciation can be accomplished with highliquid chromatography performance (HPLC) or gas chromatography-mass spectrometry (GC-MS), though these techniques are often impractical for companies or individuals with limited resources. In our previous works<sup>13,14,18</sup>, we rationally examined a variety of single-use Pc-based OTFT sensors and established that the observed sensing responses were due to a combination of electrochemical interactions and physical effects on thin-film crystallinity. Ultimately, we established a relationship between OTFT sensing characteristics and analyte induced physical thin-film effects, which highlighted the importance of semiconducting material selection.

An advantage of Pcs is the ease by which modifications can be made to the central metal, peripheral, or axial substituents<sup>2,7,19</sup>, which allows tuning of the electronic, colorimetric, and solubility properties of the Pc to enable a range of applications. Peripheral tetra- (F<sub>4</sub>-MPc), octa- (F<sub>8</sub>-MPc), or hexadeca- (F<sub>16</sub>-MPc) Pc fluorination has been demonstrated for tuning thin-film band structures and crystal morphologies<sup>20-22</sup>. Increased fluorination was also found to confer increased solubility and n-type behavior in OTFTs<sup>20,23,24</sup>. With tuneable solubility and high stability, Pcs have been deposited as semiconducting layers for OTFTs both through solution techniques and physical vapor deposition (PVD)<sup>7,22,25</sup>. Examined with the goal of improving charge transport, deposition conditions generally focus on optimizing intermolecular distances while minimizing the negative effects of interface energetics, and grain boundaries, device architecture<sup>22,26-28</sup>. Electrically, improved charge transport is characterized by high mobility ( $\mu$ ), a voltage threshold ( $V_T$ ) near 0 V, low hysteresis, a large on/off current ratio (On/Off), and low defect density (N), all of which can be obtained from the transfer characteristics of an OTFT<sup>7,29,30</sup>. Post-deposition annealing can be accomplished with a range of techniques, most commonly heat or solvent vapor, but the goal of improved charge transport, and the mechanisms by which it's achieved, remain the same<sup>31,32</sup>. Thus, in addition to molecular tuning, Pc thin films can be morphologically tuned through film engineering by altering the deposition surface, deposition conditions, or post-deposition annealing to obtain more favorable spectral characteristics or charge transport conditions<sup>33,34</sup>.

Post deposition annealing and sensing studies have highlighted zinc phthalocyanine (ZnPc) as a highly tuneable and sensitive material<sup>24,35–38</sup> and we have previously demonstrated that analyte exposure induces structural changes within films, altering, and more often disrupting, charge transport pathways similarly to post deposition annealing techniques<sup>18</sup>. Thus, where analyte exposure can induce altered thin-film nanostructures triggering a larger sensing response, we hypothesize that structurally different

films will have altered sensing responses. Here, we examine and screen the effects of THC vapor on thin-film crystallinity and OTFT performance of ZnPc, zinc tetrafluorophthalocyanine (F<sub>4</sub>-ZnPc), and zinc hexadecafluorophthalocyanine (F<sub>16</sub>-ZnPc), and compare the THC sensing performance to our previously evaluated copper phthalocyanine (CuPc), and copper hexadecafluorophthalocyanine (F<sub>16</sub>-CuPc) devices. We further investigate the most sensitive material by examining the effects of deposition conditions, crystal morphology, and thin-film thickness with the goal of increasing the sensitivity to THC vapor. By X-ray diffraction (XRD), grazing incident wide angle X-ray spectroscopy (GIWAXS), atomic force microscopy (AFM), and scanning electron microscopy (SEM), we interrogate thin-film surface nanostructures and morphology pre- and post-exposure to THC vapor, relating OTFT sensing characteristics to the physical characteristics of Pc thin films. Additionally, we perform realtime electrical characterization with vapor exposure to further elucidate the role of film thickness and polymorphism with sensing response. Ultimately, through material selection and tuned deposition conditions, we achieve a 100x increase in ZnPc device sensitivity to THC vapor, demonstrating the importance and utility of thin-film structures and engineering.

# **Results and discussion**

**Material selection**. CuPcs and ZnPcs, with various degrees of fluorination (CuPc,  $F_{16}$ -CuPc, ZnPc,  $F_{4}$ -ZnPc,  $F_{16}$ -ZnPc) (Fig. 1), were deposited as 400 Å thin-films by PVD at 25 °C with a rate of 0.2 Å/s on OTS treated Si/SiO<sub>2</sub> substrates with and without prepatterned gold electrodes. Pre-exposure GIWAXS and XRD spectra (Figure S1) revealed films with a high intensity (100) plane at q = ~0.50 Å<sup>-1</sup> and a single XRD peak with 2θ between 6–7°, suggesting a preferential orientation of the Pc ring to the substrate surface of ~75°; typical of α-crystal orientations with co-facial herringbone stacking<sup>21,39</sup>. Pre-exposure output and transfer curves (Figure S2) demonstrated OTFT performance in agreement with other sources<sup>7,27,40,41</sup>, with complete peripheral fluorination ( $F_{16}$ -) of both CuPc and ZnPc conferring n-type OTFT behavior.

Pc thin-films were exposed to 4 ppm THC vapor over a period of 90 seconds and the percent change in peak mobility  $(\%\Delta\mu)$ , voltage threshold shift  $(\Delta V_T)$ , defect density  $(\Delta N)$ , hysteresis (%  $\Delta Hys$ ), on/off current ratio ( $\Delta On/Off$ ), and peak XRD intensity (%  $\Delta XRD$ ) was determined for each material and displayed in Fig. 1a. As previously established<sup>18</sup>, analytes, such as THC, when introduced to Pc thin films, in addition to the effects of intermolecular interactions, can also alter physical characteristics of the thin film, such as crystal packing and morphology, which can cause changes in OTFT performance. Here, with the exception of devices made with F<sub>4</sub>-ZnPc, the sign, and magnitude, of the change in XRD peak height correlates to the magnitude of the change in  $\mu$  and N. By XRD and GIWAXS, the absence of any observed new crystal morphologies suggests that, without the formation of new polymorphs, mobility decreases are related to the degree of change in a-crystallinity of the bulk film. Limited changes in hysteresis and negligible OTFT bias stress effects suggests that exposure of OTFTs to THC vapor results in irreversible and relatively stable physical thin-film changes<sup>29</sup>.

In our previous works<sup>14,18</sup>, we found that THC, as a  $\pi$ conjugated molecule with electron-donating properties<sup>42</sup>, interacts with the central metal of MPcs, and that these interactions are deferential to the species of the central metal. The molecular size, electronegativity, and valence structure of the central metal affects analyte interactions, which, in addition to electrochemical effects, can also result in physical effects, such as distortions to the Pc ring, and induce physical charge traps<sup>38</sup>. Here, CuPc and F<sub>16</sub>-



**Fig. 1 Effects of THC vapor on Pc OTFT electrical characteristics and XRD spectra. (a)** CuPc,  $F_{16}$ -CuPc, ZnPc,  $F_4$ -ZnPc, or  $F_{16}$ -ZnPc OTFTs were exposed to 4 ppm THC vapor over a period of 90 seconds. Boxed regions represent a sign change from the median response. Characteristic transfer and XRD spectra demonstrate sampled regions. Dashed lines represent regions of the curve within which slope was measured. (b) General Pc structure and (c) a table of the Pc's studied. Mobility was calculated from the saturation region of the transfer curves while defect density and voltage threshold were estimated from the subthreshold slope, with average values taken from 20 devices.

CuPc-based devices both show  $+ \Delta V_T$  with exposure to THC, while the ZnPc,  $F_4$ -ZnPc, and  $F_{16}$ -ZnPc-based devices show  $-\Delta V_T$ , suggesting that the copper-THC interaction induces electron trapping while the zinc-THC interaction creates deep, but short life-time, hole traps<sup>29,43</sup>. The peripherally fluorinated Pcs demonstrate smaller changes in XRD peak intensity,  $\Delta N$ , and  $\Delta On/Off$ , suggesting limitations to both metal-analyte interactions and film restructuring as a result of their increased size or peripheral electronegativity. With the exception of hysteresis, an overall comparison of the transfer curves and XRD spectra demonstrates that ZnPc devices undergo the greatest changes across all metrics with exposure to THC vapor. This further illustrates the critical impacts of analyte-Pc interactions and thinfilm restructuring on over all device performance, highlighting the role of material selection in optimizing sensing responses.

Effects of morphology. With the greatest changes observed for ZnPc devices, we sought to further examine the effects of thinfilm structures by preparing 400 Å  $\alpha$ -ZnPc ( $2\theta = 6.92^{\circ}$ ) OTFTs by altering the PVD rate or deposition substrate temperature, and altering the substrate surface by pre-depositing a monolayer of p-sexiphenyl (p-6P) as a patterning agent to afford varying degrees of  $\alpha$ -crystallinity<sup>44,45</sup>. The fastest deposition rate (1 Å/s) at the lowest deposition substrate temperature (25 °C) resulted in small uniform grains by AFM, low peak intensity by XRD, and poor transfer characteristics (Fig. 2). Decreasing deposition rate, increasing deposition substrate temperature, or deposition on p-6P as a patterning agent, led to larger grain sizes, increased XRD peak intensity, and generally improved OTFT transfer characteristics (Table S1). A small peak, characteristic of  $\beta$ -ZnPc  $(2\theta = 9.32^{\circ})$ , was observed (Figure S3) for the depositions carried out at 180 °C, suggesting that some  $\alpha$ - to  $\beta$ - phase transition was beginning to occur, however the relative intensity to the  $\alpha$ - peak  $(2\theta = 6.92^\circ)$  was less than 2%.

Exposure of the least crystalline films to 4 ppm of THC vapor rendered the OTFTs inoperable. Films were instead exposed to 400 ppb THC vapor over a period of 90 seconds, inducing the greatest changes in the least crystalline films, and manifesting as  $-\%\Delta\mu$ ,  $-\Delta V_T$ , an increase in hysteresis, decreased *On/Off*, and +  $\Delta N$  (Table S1). Increased thin-film crystallinity reduced OTFT sensing responses, with the most crystalline films demonstrating limited changes in charge transport characteristics and negligible changes in XRD intensity. Measured by AFM, the surface area of each film was inversely proportional to the degree of alpha crystallinity, where the very high crystallinity films had the least surface area, while the low crystallinity films had the greatest surface area (very high = 6.43, high = 6.52, medium = 6.61, low =  $6.81 \,\mu\text{m}$  of surface area in a  $2.5 \times 2.5 \,\mu\text{m}$  section). This suggests that smaller grains in the low crystallinity films affords increased surface area for ZnPc-THC interactions and intermolecular forces which facilitate greater changes by XRD and thus greater OTFT sensing responses<sup>46</sup>. Previously, we demonstrated a relationship between different Pcs, their analyte-induced physical film changes, and the measured OTFT sensing responses<sup>18</sup>. Here, these findings demonstrate that this relationship between physical thin-film changes and an OTFT sensor response can be applied to a single material, highlighting the importance of thin-film nanostructures for analytes-sensor interactions.

Film thickness effects and real-time sensing. To assay the effects of thickness on the most sensitive thin-film conditions, we prepared OTFTs with 200 and 800 Å, low-crystallinity a-ZnPc films (deposition rate 1 Å/s at 25 °C) and exposed them to 40 ppb THC vapor over a period of 90 seconds (Figure S4). Pre-exposure, OTFTs of both thicknesses demonstrated similar transfer characteristics to low-crystallinity a-ZnPc 400 Å OTFTs, indicating that film thickness between 200-800 Å does not significantly influence the electrical performance of these low-crystallinity OTFTs. Compared to the 200 Å films, XRD spectra shows an expected, 59% more intense  $\alpha$ - peak ( $2\theta = 6.92^{\circ}$ ) for the 800 Å film, as the additional thickness provides additional diffraction signal<sup>46</sup>. The 2D scattering pattern determined by GIWAXS corroborates an  $\alpha$ - phase orientation with a highly ordered (100) plane at q = 0.49 Å<sup>-1</sup> and a preferential Pc orientation to the surface of ~74°. Exposure to THC vapor showed a similar  $+ \Delta V_T$ , increase in hysteresis,  $-\Delta N$ , and a decrease in XRD peak intensity for OTFTs of both thicknesses. The OTFTs with the thinnest, 200 Å films, in agreement with other works<sup>47,48</sup>, demonstrated a larger decrease in on-current, greater  $\Delta N$ , and greater  $-\%\Delta\mu$ , suggesting greater relative changes in film crystallinity, while



**Fig. 2 Effect of surface morphology on ZnPc OTFT sensitivity to THC vapor.** (a) AFM images, transfer data, and XRD spectra of ZnPc OTFTs with varying degrees of crystallinity (**b**, **d**) pre- and (**c**, **e**) postexposure to 400 ppb THC vapor over 90 seconds. Low crystallinity thin-films were deposited at a rate of 1 Å/s and 25 °C, medium (med) at 0.05 Å/s and 25 °C, high at 0.2 Å/s and 140 °C, and very high at 0.2 Å/s and 180 °C with a pre-deposited monolayer of p-sexiphenyl (p-6P). Scale bars represent 500 nm.

changes in  $V_T$  and hysteresis suggest Pc-THC interactions which are independent of film thickness.

The initially screened thin-film conditions of 400 Å  $\alpha$ -ZnPc OTFTs, deposited at a rate of 0.2 Å/s and 25 °C, demonstrate a negligible response when exposed to 40 ppb THC vapor (Figure S5). Comparing the response of the initially screened thin-film conditions to the sensitized response of the thinnest low crystallinity films, we demonstrate a 100x increase in post-exposure sensitivity to THC vapor. Thus, altered surface morphology and semiconductor thickness has a significant effect on OTFT sensitivity.

To examine the effects of THC exposure in real time, we operated low crystallinity  $\alpha$ -ZnPc OTFTs at fixed saturation biases while they were continually exposed to 40 ppb THC vapor (Fig. 3). At operating biases of  $V_{SD} = -50$  V and  $V_{GS} = -40$  V, continuous THC vapor exposure caused an immediate, sharp, operating current decrease followed by a sustained decrease in operating current at a constant rate, which was mediated by thin-film thickness (Fig. 3c). Pulsed exposure demonstrated a similar effect (Fig. 3d and S6), where the onset of THC vapor induced an immediate decrease in operating current followed by a sustained, thickness mediated, decrease over time. However, after short exposures (~10 seconds), operating current recovered sharply when vapor application was stopped, though not to pre-exposure levels, and subsequent exposures reduced the magnitude of both the operating current decrease and recovery with vapor onset and offset.

The magnitude of the effects of vapor onset, continued exposure, and vapor offset were inversely related to film thickness, where the thinnest, 200 Å films, demonstrated the largest responses, consistent with OTFT sensors made with Cl-AlPc<sup>47</sup>. Additionally, for both pulsed and continual modes of exposure, after 90 seconds, the operating current is consistent with the operating current of films of the same thickness characterized post-exposure at the same bias conditions, suggesting prolonged exposure limits the observed recovery. Both periodic exposures to forced air and continued device operation in air (Fig. 3b), yielded negligible changes in operating current.

The observed, sharp, but semi-reversible, decreases in operating current with THC vapor onset suggests that the introduction of THC immediately causes hole-trapping effects, lowering the operating current. Sustained, irreversible decreases in operating current with continued THC vapor application then suggests that over time, irreversible structural defects occur within the film, permanently lowering operating current, and limiting repeated sensor use. That both the immediate and sustained decreases in operating current are thickness dependant supports these conclusions, as both the amounts of THC relative to Pc within the film and the THC-induced structural defects will be greatest for the thinnest films. Periodic exposure of the 200 Å films to 40 ppb cigarette smoke yields small onset current decreases with smoke application, slight current recovery when smoke application stops, and a slight decrease in current over time (Fig. 3e). However, the magnitude of the operating current decrease was small and the results noisy, indicating that other vaporized organic compounds have different effects on the electrical performance of the substrates, and demonstrating the innate selectivity of ZnPc for this application. Thus, every step of sensor manufacturing impacts device performance, from material selection, deposition conditions, and thin-film thickness.

Effects of polymorphism. To analyze the effects of crystal polymorphism on sensing response, we prepared β-ZnPc thinfilms by treating 400 Å α-ZnPc thin-films with toluene vapor for 24 h (Fig. 4) resulting in OTFTs with lower peak  $\mu$ , lower  $V_T$ , lower off current, and lower N in comparison to α-ZnPc OTFTs. Additionally, large, µm-scale rectangular crystals are observable by SEM. XRD spectra of the toluene-treated films showed two sharp peaks of equal intensity at  $2\theta = 7.04$  and  $9.32^{\circ}$ , consistent with reported β-Pc morphology<sup>31</sup>, and a small shoulder peak at  $2\theta = 6.84^{\circ}$ . The azimuthally-integrated GIWAXS pattern of the films is in good agreement with the β-ZnPc pattern predicted by single crystal XRD (Figure S7) confirming complete transition from α- to β- phase. The 2D scattering pattern reveals a highly



**Fig. 3 Real-time THC vapor exposure and detection.** (a) Schematic showing the application of THC vapor to a ZnPc OTFT, (b) the effects of heated air, (c) the effects of continual THC vapor exposure on 200, 400, and 800 Å ZnPc OTFTs, (d) the effects of periodic THC exposure on a 200 Å ZnPc OTFT, and (e) the effects of 40 ppb cigarette smoke on 200 Å ZnPc OTFTs. A V<sub>SD</sub> of -50 V was held and a V<sub>GS</sub> of -40 V and was pulsed at a rate of 20 milliseconds on 80 milliseconds off over a period of 120 seconds while 40 ppb THC vapor was continually or periodically flowed over the surface of the OTFT in a 50 mL chamber. Slope was calculated from 20 to 120 seconds.



Fig. 4 Effect of  $\beta$ - morphology on ZnPc OTFT sensitivity to THC vapor. (a) Transfer data, (b) XRD spectra, and (c, d) SEM images of ZnPc OTFTs after exposure to 400 ppb THC vapor over a period of 90 seconds. Scale bars represent 2 µm and red arrows denote sheet formation. (e, f) 2D scattering patterns ( $\theta$  = 0.1°) of  $\beta$ -ZnPc thin films deposited on Si substrates and (g, h) a schematic showing GIWAXS determined preferential phase angle to the substrate pre- and post- exposure to THC vapor.

ordered crystalline thin film with the (100) plane at q = 0.50 and the (10-2) plane at q = 0.68 Å<sup>-1</sup> along the  $q_z$  axis suggesting multiple molecular orientations however with a preference of the Pc ring aligned ~45° to the surface.

Exposure to 400 ppb THC vapor over a period of 90 seconds resulted in a 10.3  $\mu$ A increase in peak operating current, a -7.4 V in  $\Delta V_T$ , and a 4.1 V increase in hysteresis. By XRD, the intensity

of the peak at  $2\theta = 7.04^{\circ}$  does not change, while the intensity of the peak at  $2\theta = 9.32^{\circ}$  decreases by 53%. The shoulder peak at  $2\theta = 6.84^{\circ}$  increases in intensity by 20-fold, and a new peak at  $2\theta = 7.38^{\circ}$  appears, indicating morphological changes. Visualized by SEM, previously well-ordered, narrow, rectangular crystals become broad and sheet-like post-exposure (red arrows). By GIWAXS we observe a decrease in relative peak intensity

corresponding to the (10-2) plane ( $q = 0.68 \text{ Å}^{-1}$ ) and peak splitting at  $q = \sim 0.50 \text{ Å}^{-1}$ , corresponding to the (100) plane (Figure S7) in  $\beta$ -ZnPc thin films post-exposure to THC vapor. Additionally, the 2D scattering pattern of post-exposure films exhibit a narrowing of the partial arcs and a change in preferential ZnPc orientation to the surface from ~45° to ~74°.

β-ZnPc, despite being highly crystalline, generally demonstrates poor semiconducting performance due to suboptimal charge transport distances caused by close intermolecular spacing and an increased face-on configuration<sup>49</sup>. Thus, the observed increase in operating current as an OTFT "turn-on" response may be a result of partial conversion of  $\beta$ -ZnPc crystals to an  $\alpha$ -, or an a-like, polymorph through exposure to THC vapor and suggests the interstitial incorporation of THC into the ZnPc crystallites. Low defect density ( $N = 5.4 \cdot 10^{-12}$ ), extracted from the subthreshold swing, post-exposure suggests the sheet-like structures provide excellent charge transport pathways, with hysteresis possibly resulting from the large interfacial areas between the sheets<sup>22,50</sup>. A new XRD peak at  $2\theta = 7.38^{\circ}$  in the post-exposure  $\beta$ -ZnPc films, but not in the post-exposure  $\alpha$ - films or the pre-exposure  $\beta$ - films, suggests a tertiary morphology in the THC vapor induced phase transition of  $\beta$ -ZnPc<sup>31</sup>.

Real-time characterization of the  $\beta$ -ZnPc OTFTs with continual 400 ppb THC vapor exposure (Figure S8) demonstrated a turn-off response when operated with  $V_{SD}$  of -50 V and  $V_{GS}$  of -20 V. Like the  $\alpha$ -ZnPc OTFTs, with the onset of THC vapor there is an immediate decrease in operating current, however, rather than a subsequent sustained decrease with constant slope, instead there is a rapid decrease in operating current with a decaying slope. As a large  $V_T$  shift is observed for the  $\beta$ -ZnPc OTFTs characterized post-exposure, a decrease in operating current is expected at these bias conditions. Inconsistencies in the pre-exposure operating current and vapor onset effects for the  $\beta$ -ZnPc OTFTs, in comparison to the  $\alpha$ -ZnPc OTFTs, are likely a result of uneven distribution and orientation of the  $\beta$ -crystals between OTFT electrodes.

With a  $V_{GS}$  bias of -40 V, there is again an immediate decrease in OTFT operating current with the onset of THC vapor with limited change for ~40 seconds, whereupon the operating current begins to fluctuate and trend upwards. Comparing to periodic exposure with the same bias conditions, we observe a similar fluctuating climb in operating current after ~35 seconds of exposure to THC vapor. Fluctuations in operating current suggest that significant µm-scale recrystallization events begin after ~35-40 seconds of exposure, in comparison to the immediate sustained decrease in operating current observed for the a-ZnPc OTFTs. Moreover, sharp decreases in operating current with vapor onset, and recovery with vapor offset, again suggests THC is introducing or acting as a hole-trap within the film. Additionally, where the most crystalline a-ZnPc thin-films demonstrated the least OTFT and XRD changes, highly crystalline β-ZnPc films demonstrated significant re-crystallization with exposure to THC vapor. Thus, the degree of molecular order on the surface is not necessarily indicative of either sensing response or the analyte-induced physical effects. Instead, the lowest crystallinity a-ZnPc OTFTs demonstrated the greatest electrical sensitivity to THC vapor, exhibiting the balance between thinfilm morphology, recrystallization, and surface area for intermolecular interactions.

# Conclusions

The effects of THC vapor exposure on both the central metal and peripheral fluorination demonstrated that non-fluorinated ZnPc OTFTs exhibited the greatest electrical and structural sensitivity compared to isostructural CuPc OTFTs and devices. Increasing peripheral fluorination was found to limit analyte-induced structural changes while the central metal-mediated voltage threshold shifts. Preparing a-ZnPc films with varying degrees of crystallinity revealed that the least crystalline films were the most susceptible to physical alterations upon exposure to analytes and had the largest OTFT electrical changes. Film thickness was also found to effect sensitivity, with 200 Å, low crystallinity films demonstrating sensitivity to 40 ppb THC vapor. Real-time characterization of low crystallinity a-ZnPc OTFTs demonstrated an immediate, reversible hole-trapping effect with the onset of THC vapor and a sustained, irreversible operating current decrease due to physical film restructuring. In contrast, highly crystalline β-ZnPc films demonstrated significant structural changes when exposed to THC vapor. Such films transitioned from large, regular, µm-scale crystals to more sheet-like structures, with a change in preferred substrate angle from 45 to ~74° and demonstrating a turn-on OTFT response. Through film engineering of Pc-based OTFT sensors we achieved a 100x increase in sensitivity over our previously developed CuPc-based devices<sup>13</sup>, illustrating the importance of not only material selection, but also thin-film nanostructures, thickness, and polymorphism in Pc-OTFT sensor implementations.

# Experimental

**Materials**. CuPc (copper(II) phthalocyanine),  $F_{16}$ -ZnPc (zinc 1,2,3,4,8,9,10,11,15,16,17,18,22,23,24,25-hexadecafluoro phthalocyanine), and tricholoro(octyl)silane were obtained from Sigma Aldrich.  $F_{16}$ -CuPc (copper(II) 1,2,3,4,8,9,10,11,15,16,17,18,22,23,24,25-hexadecafluoro phthalocyanine), and ZnPc (zinc phthalocyanine) were obtained from TCI chemicals.  $F_4$ -ZnPc was synthesized according to ref. <sup>35</sup> and confirmed by mass spectrometry. All Pcs were purified by train sublimation prior to use. Cannabinoid standards were obtained from Toronto Research Chemicals. All solvents were HPLC grade and purchased from Fischer Scientific.

Thin film fabrication. Si substrates with 230 nm thermally grown  $SiO_2$  dielectric and prepatterned gold source-drain electrodes (W = 2000  $\mu$ m, L = 10  $\mu$ m), were purchased from Fraunhofer IPMS and used to fabricate bottom-gate bottom-contact (BGBC) transistors. Si substrates with 300 nm thermally grown SiO<sub>2</sub> were purchased from Ossila and used for fabrication of thin films for XRD and GIWAXS. Wafers were prepared as described in ref. <sup>18</sup> before being transferred to an Angstrom EvoVac thermal evaporator where deposition rate and thin-film thickness was controlled by quartz crystal microbalance and substrate temperature was controlled by thermocouple.

 $\beta$ -ZnPc films were generated by exposing 400 Å ZnPc films deposited at 25 °C at a rate of 0.2 Å/s to 50 °C toluene vapor in a custom vapor chamber for 24 h followed by a baking at 70 °C for 45 minutes in a vacuum oven.

**Vapor exposure**. THC was dissolved in methanol and loaded on to a steel wool frit where it was allowed to dry before being placed in a Volcano Medic vaporizer set to 210 °C temperature. An 8 L balloon was filled completely before being evacuated into a 50 mL volume vapor chamber in which the substrates were placed.

**OTFT characterization**. Organic thin film transistors were characterized by applying a voltage bias between gate and source electrodes with BeCu alloy probe tips and the source drain current ( $I_{SD}$ ) was measured with a Keithley 2614B on a custom electrical probe station. Mobility was calculated using the

following:

$$I_{DS} = \frac{\mu C_I W}{2L} (V_{GS} - V_T)^2$$

Where  $C_i$  is the dielectric capacitance, W and L are the width and length of the semiconducting channel, and  $V_T$  describes the threshold voltage at which  $I_{DS}$  begins to rapidly increase. To mitigate the effects of bias stress, the gate bias was pulsed at 20 millisecond intervals with an 80 millisecond delay. To saturate the devices, six transfer curves were obtained for each device with the final three averaged to yield a characteristic transfer curve. Defect density (N) was determined with the following:

$$N = \left(\frac{Sq}{kTlin(10)} - 1\right)\frac{C_i}{q}$$

Where S is the subthreshold slope, as estimated graphically from the transfer data, q is the electronic charge, k is Boltzmann's constant, and T is temperature.

**XRD and GIWAXS**. XRD characterization was performed using a Rigaku Ultima IV powder diffractometer with a Cu-Ka ( $\lambda = 1.5418$  Å) source. Thin-film substrate measurements were taken with a scan range of 5° < 2 $\theta$  < 11° at a rate of 0.5° min<sup>-1</sup> with a spin rate of 30 rpm.

GIWAXS experiments were performed at the Canadian Light Source (CLS) in Saskatoon, Canada using the Brockhouse Diffraction Sector (BXDS) beamline with a photon energy of 15.1 keV, and the SOLEIL Synchrotron facility in Saint-Aubin, France using the SIRIUS beamline with a photon energy of 10 keV. For data collected at CLS, GIWAXS patterns were collected using a Rayonix MX300 CCD detector (73.242 µm x 73.242 µm pixel size), placed 416 mm from the sample with an angle of incidence of  $\theta = 0.3^{\circ}$ . For data collected at SOLEIL, GIWAXS patterns were collected using a PILATUS3 S 1 M detector (172 µm x 172 µm pixel size), placed 330 mm from the sample with an angle of incidence of  $\theta = 0.1^{\circ}$ . All GIWAXS data was calibrated against a silver behenate standard and a poly(3hexylthiophene-2,5-diyl) standard and analyzed using the GIXSGUI software package in MATLAB, where both polarization and solid-angle corrections were applied<sup>51</sup>.

**AFM and SEM**. SEM measurements were performed with a Tescan Vega II on thin-film Pc devices at 20 kV. AFM measurements were performed with a Bruker Dimension Icon AFM, equipped with ScanAsyst-Air tips, and images were processed with NanoScope Analysis v.1.8. Scans were performed at a rate of 0.814 Hz with multiple sites investigated.

## Data availability

All data is available upon request to the corresponding authors.

Received: 4 October 2022; Accepted: 12 December 2022; Published online: 24 December 2022

#### References

- Lu, H. & Kobayashi, N. Optically Active Porphyrin and Phthalocyanine Systems. *Chem. Rev.* 116, 6184–6261 (2016).
- Zhou, W. et al. From chemical curiosity to versatile building blocks: unmasking the hidden potential of main-group phthalocyanines in organic field-effect transistors. *Mater. Adv.* 2, 165–185 (2021).
- 3. Braun, A. & Tcherniac, J. Über die Produkte der Einwirkung von Acetanhydrid auf Phthalamid. *Berichte der Dtsch. Chem. Ges.* **40**, 2709–2714 (1907).

- Dahlen, M. A. The Phthalocyanines A New Class of Synthetic Pigments and Dyes. Ind. Eng. Chem. 31, 839–847 (2002).
- Claessens, C. G. et al. Phthalocyanines: From outstanding electronic properties to emerging applications. *Chem. Rec.* 8, 75–97 (2008).
- de la Torre, G. et al. Phthalocyanines and Subphthalocyanines: Perfect Partners for Fullerenes and Carbon Nanotubes in Molecular Photovoltaics. *Adv. Energy Mater.* 7, 1601700 (2017).
- Melville, O. A. et al. Phthalocyanine-Based Organic Thin-Film Transistors: A Review of Recent Advances. ACS Appl. Mater. Interfaces 7, 13105–13118 (2015).
- Roberts, M. E. et al. Material and device considerations for organic thin-film transistor sensors. J. Mater. Chem. 19, 3351–3363 (2009).
- Liu, J. et al. Glucose sensor based on organic thin film transistor using glucose oxidase and conducting polymer. *Sens. Actuators B Chem.* 135, 195–199 (2008).
- Seo, G. et al. Rapid Detection of COVID-19 Causative Virus (SARS-CoV-2) in Human Nasopharyngeal Swab Specimens Using Field-Effect Transistor-Based Biosensor. ACS Nano 14, 5135–5142 (2020).
- 11. Torsi, L. et al. Multi-parameter gas sensors based on organic thin-filmtransistors. *Sens. Actuators B Chem.* **67**, 312-316 (2000).
- Boileau, N. T. et al. P and N type copper phthalocyanines as effective semiconductors in organic thin-film transistor based DNA biosensors at elevated temperatures. *RSC Adv.* 9, 2133–2142 (2019).
- Comeau, Z. J. et al. On-the-Spot Detection and Speciation of Cannabinoids Using Organic Thin-Film Transistors. ACS Sens. 4, 2706–2715 (2019).
- Comeau, Z. J. et al. Engineering Cannabinoid Sensors through Solution-Based Screening of Phthalocyanines. ACS Appl. Mater. Interfaces 12, 50692–50702 (2020).
- Russo, E. B. History of Cannabis and Its Preparations in Saga, Science, and Sobriquet. Chem. Biodivers. 4, 1614–1648 (2007).
- Gowran, A. et al. The Multiplicity of Action of Cannabinoids: Implications for Treating Neurodegeneration. CNS Neurosci. Ther. 17, 637–644 (2011).
- Stolker, A. A. M. et al. Determination of cannabinoids in cannabis products using liquid chromatography-ion trap mass spectrometry. *J. Chromatogr. A* 1058, 143–151 (2004).
- Comeau, Z. J. et al. Organic Thin-Film Transistors as Cannabinoid Sensors: Effect of Analytes on Phthalocyanine Film Crystallization. *Adv. Funct. Mater.* 32, 2107138 (2022).
- Cranston, R. R. et al. Highlighting the processing versatility of a silicon phthalocyanine derivative for organic thin-film transistors. *J. Mater. Chem. C.* 10, 485–495 (2022).
- Schwarze, M. et al. Band structure engineering in organic semiconductors. Science (80-) 352, 1446–1449 (2016).
- Boileau, N. T. et al. Metal phthalocyanine organic thin-film transistors: Changes in electrical performance and stability in response to temperature and environment. *RSC Adv.* 9, 21478–21485 (2019).
- Cranston, R. R. & Lessard, B. H. Metal phthalocyanines: thin-film formation, microstructure, and physical properties. *RSC Adv.* 11, 21716–21737 (2021).
- 23. Klyamer, D. et al. Fluorinated metal phthalocyanines: Interplay between fluorination degree, films orientation, and ammonia sensing properties. *Sensors (Switzerland)* **18**, (2018).
- Schwarze, M. et al. Impact of molecular quadrupole moments on the energy levels at organic heterojunctions. *Nat. Commun.* 10, 1–9 (2019).
- Shimoda, T. et al. Solution-processed silicon films and transistors. *Nature* 440, 783–786 (2006).
- Tan, J. et al. Fullerene-derivative as interlayer for high performance organic thin-film transistors. J. Mater. Chem. C. 6, 6052–6057 (2018).
- Ye, R. et al. Improved performance of fluorinated copper phthalocyanine thin film transistors using an organic pn junction: Effect of copper phthalocyanine film thickness. *Thin Solid Films* 517, 3001–3004 (2009).
- Melville, O. A. et al. Ambipolarity and Air Stability of Silicon Phthalocyanine Organic Thin-Film Transistors. Adv. Electron. Mater. 5, 1900087 (2019).
- Sirringhaus, H. Reliability of organic field-effect transistors. Adv. Mater. 21, 3859–3873 (2009).
- Lin, Y. J. & Lin, Y. C. Electrical conduction mechanisms in the transfer characteristics of pentacene thin film transistors. *Appl. Phys. Lett.* 105, 023506 (2014).
- Xu, J. et al. Effect of solvent-vapour annealing on morphology, structure of copper(II) phthalocyanine thin films and device performance. *Bull. Mater. Sci.* 41, 111 (2018).
- Ogunsipe, A. et al. Solvent effects on the photochemical and fluorescence properties of zinc phthalocyanine derivatives. J. Mol. Struct. 650, 131–140 (2003).
- Faris, T. et al. Effects of annealing on device parameters of organic field effect transistors using liquid-crystalline tetrasubstituted zinc phthalocyanine. *Europhys. Lett.* 106, 58002 (2014).
- Padma, N. et al. Effect of post deposition annealing on the performance of copper phthalocyanine based organic thin film transistor. *AIP Conf. Proc.* 1512, 786 (2013).

# ARTICLE

- Klyamer, D. D. et al. Influence of fluorosubstitution on the structure of zinc phthalocyanine thin films. *Macroheterocycles* 11, 304–311 (2018).
- Zhu, Y. et al. High-Performance NO2 Sensors Based on Ultrathin Heterogeneous Interface Layers. Adv. Mater. Interfaces 7, 1901579 (2020).
- Gonzalez Arellano, D. L. et al. Phase Transition of Graphene-Templated Vertical Zinc Phthalocyanine Nanopillars. J. Am. Chem. Soc. 140, 8185–8191 (2018).
- S Saini, G. S. et al. Zinc phthalocyanine thin film and chemical analyte interaction studies by density functional theory and vibrational techniques. https://doi.org/10.1088/0953-8984/21/22/225006 (2009).
- Dong, H. et al. 25th Anniversary Article: Key Points for High-Mobility Organic Field-Effect Transistors. *Adv. Mater.* 25, 6158–6183 (2013).
- Yan, X. et al. An investigation on air stability of copper phthalocyanine-based organic thin-film transistors and device encapsulation. *Thin Solid Films* 515, 2655–2658 (2006).
- Jiang, H. et al. Molecular Crystal Engineering: Tuning Organic Semiconductor from p-type to n-type by Adjusting Their Substitutional Symmetry. *Adv. Mater.* 29, 1605053 (2017).
- Borges, R. S. et al. Understanding the Molecular Aspects of Tetrahydrocannabinol and Cannabidiol as Antioxidants. *Mol. 2013, Vol. 18, Pages 12663-12674* 18, 12663–12674 (2013).
- 43. Gu, G. et al. Electron traps and hysteresis in pentacene-based organic thinfilm transistors. *Appl. Phys. Lett.* **87**, 243512 (2005).
- Gu, W. et al. Preparing highly ordered copper phthalocyanine thin-film by controlling the thickness of the modified layer and its application in organic transistors. *Solid. State Electron.* 89, 101–104 (2013).
- 45. Wang, T. et al. Weak epitaxy growth and phase behavior of planar phthalocyanines on p-sexiphenyl monolayer film. *J. Phys. Chem. B* **112**, 6786–6792 (2008).
- Chang, J. F. et al. The effects of thickness and operation temperature on ZnO:Al thin film CO gas sensor. Sens. Actuators B Chem. 84, 258–264 (2002).
- Lamontagne, H. et al. Chloro Aluminum Phthalocyanine-based Organic Thin-Film Transistors as Cannabinoid Sensors: Engineering the thin film response. Sensors and Diagnostics, (2022).
- Jiang, Y. et al. Thickness modulation on semiconductor towards high performance gas sensors based on organic thin film transistors. *Mater. Sci. Eng. B* 226, 107–113 (2017).
- McAfee, T. et al. Morphological, Optical, and Electronic Consequences of Coexisting Crystal Orientations in β-Copper Phthalocyanine Thin Films. J. Phys. Chem. C. 120, 18616–18621 (2016).
- Hanwell, M. D. et al. Charge Transport in Imperfect Organic Field Effect Transistors: Effects of Explicit Defects and Electrostatics. J. Phys. Chem. C. 114, 20417–20423 (2010).
- Jiang, Z. GIXSGUI: a MATLAB toolbox for grazing-incidence X-ray scattering data visualization and reduction, and indexing of buried three-dimensional periodic nanostructured films. J. Appl. Crystallogr. 48, 917–926 (2015).

# Acknowledgements

This work was supported by the National Science and Engineering Council of Canada (RGPIN-2015-05796 and RGPIN-2021-03387 to A. J. S.), and the Canada Research Chairs Program (950-230754 to A. J. S.). We acknowledge the Centre for Research in

Photonics at the University of Ottawa (CRPuO) for access to the AFM. The authors would like to thank both the CLS and the SOLEIL Synchrotron facility for providing beamtime, and Adam Leontowich and Arnaud Hemmerle for their expertise and technical support during beamtime at CLS and SOLEIL respectively. CLS is supported by CFI, NSERC, the University of Saskatchewan, the Government of Saskatchewan, Western Economic Diversification Canada, the National Research Council Canada, and the Canadian Institutes of Health Research. SOLEIL is supported by the National Center for Scientific Research (CNRS), the French Alternative Energies and Atomic Energy Commission (CEA), the Île-de-France Regional Council, the Essonne Department Council and the Centre Regional Council.

# Author contributions

Z.J.C performed electrical and thin-film characterization experiments, data analysis, and authored the manuscript. R.R.C and H.R.L. performed GIWAXS experiments, assisted in data analysis, and provided editorial contributions. C.S.H., A.J.S., and B.H.L, provided editorial contributions.

## **Competing interests**

The authors declare no competing interests.

## Additional information

Supplementary information The online version contains supplementary material available at https://doi.org/10.1038/s42004-022-00797-y.

**Correspondence** and requests for materials should be addressed to Adam J. Shuhendler or Benoît H. Lessard.

Peer review information Communications Chemistry thanks Maciej Krzywiecki and the other, anonymous, reviewers for their contribution to the peer review of this work.

Reprints and permission information is available at http://www.nature.com/reprints

**Publisher's note** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

**Open Access** This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/ licenses/by/4.0/.

© The Author(s) 2022