

<https://doi.org/10.1038/s42003-024-05938-8>

Spatial variability of and effect of light on the coelenteron pH of a reef coral

Check for updates

Lucas Crovetto ^{1,2}, Alexander A. Venn ¹, Duygu Sevilgen¹, Sylvie Tambutté ¹ ✉ & Eric Tambutté ¹

Coral reefs, the largest bioconstruction on Earth, are formed by calcium carbonate skeletons of corals. Coral skeleton formation commonly referred to as calcification occurs in a specific compartment, the extracellular calcifying medium (ECM), located between the aboral ectoderm and the skeleton. Calcification models often assume a direct link between the surrounding seawater and the ECM. However, the ECM is separated from the seawater by several tissue layers and the coelenteron, which contains the coelenteric fluid found in both polyps and coenosarc (tissue connecting the polyps). Symbiotic dinoflagellate-containing cells line the coelenteron and their photosynthetic activity contributes to changes in the chemistry of the coelenteric fluid, particularly with respect to pH. The aim of our study is to compare coelenteron pH between the coenosarc and polyps and to compare areas of high or low dinoflagellate density based on tissue coloration. To achieve this, we use liquid ion exchange (LIX) pH microsensors to profile pH in the coelenteron of polyps and the coenosarc in different regions of the coral colony in light and darkness. We interpret our results in terms of what light and dark exposure means for proton gradients between the ECM and the coelenteron, and how this could affect calcification.

Coral reefs occupy less than 1.2% of the world's continental shelf area¹, but are of great ecological value. They represent the largest bioconstruction on Earth and host 30% of known marine species². The reefs build up over time, creating a variety of ecological niches necessary for the colonisation and survival of many other marine species. Reef-building scleractinian corals, also called the engineers of the reefs, are calcifying organisms that secrete a calcium carbonate (CaCO₃) skeleton through a process known as biomineralization or, more commonly, calcification. Coral skeletons are composite structures containing an organic fraction and a mineral fraction of CaCO₃ in the form of aragonite. The precipitation of CaCO₃ requires a specific chemical environment at the site of calcification, in which pH is a very important parameter³. In the context of current global change, including ocean acidification, many studies have investigated the effects of reduced seawater pH (pH_{SW}) on coral physiological processes^{4–10}.

Detailed descriptions of coral anatomy and histology are reviewed in refs. ^{3,11}. Briefly, reef-building corals are mainly colonial organisms composed of numerous polyps that are linked together by a tissue called the coenosarc. The polyp is the anatomical unit of a coral and consists of a central mouth surrounded by a ring of tentacles. Both polyps and coenosarc contain an internal fluid-filled cavity, the coelenteron, which in the case of the polyp opens to the external seawater via the stomodeum through the

mouth. The coelenteron includes the gastrovascular cavity (the interior space of a coral polyp) and the gastrovascular canals (which connect the gastrovascular cavities of polyps). It separates the oral and aboral tissues, which are both composed of an epithelium and an endothelium or gastroderm separated by a layer of extracellular matrix called mesoglea. The oral epithelium faces the surrounding seawater and the oral and aboral endothelia face the coelenteron. Symbiotic photosynthetic dinoflagellates (family *Symbiodinaceae*¹²) are mostly found in the oral gastroderm and reside within specific cells. The aboral epithelium, also known as the aboral ectoderm or calicoderm, houses the calcifying cells and is located next to the skeleton, playing a key role in its formation. The polyps overlie the calices and the coenosarc overlies the coenosteum.

The calcification process takes place in a semi-enclosed compartment located between the calicoderm and the skeleton, namely the extracellular calcifying medium (ECM). Numerous studies have investigated the chemical composition of the ECM, including measurements of pH, calcium, and carbonate concentrations, as these are important parameters controlling the saturation state in the ECM and thus driving calcification³. Of these parameters, pH has been the most studied using a variety of approaches. Whether by indirect methods (geochemical proxies^{6,13}) or direct methods (pH-sensitive fluorescent dyes^{14,15} or pH microsensors^{5,16,17}), studies all show

¹Marine Biology Department, Centre Scientifique de Monaco, 98000, Monaco. ²Sorbonne Université – ED 515 Complexité du Vivant, 75005 Paris, France.

✉ e-mail: stambutte@centrescientifique.mc

that the pH of the ECM (pH_{ECM}) is more elevated than pH of the seawater (pH_{sw}).

Studies dealing with the calcification process of corals often assume a direct link between the external seawater and the ECM^{7,16,18–20}. Recent studies show that in the ECM, calcification involves particle attachment of amorphous calcium carbonate and ion by ion growth^{21,22}. However, as described above, the ECM is separated from the external seawater by several compartments, including tissue layers and the coelenteron. Recently, it has been shown that pH in the aboral mesoglea, which is at the basal side of calcifying cells, has a different pH than seawater²³. To understand pH gradients across coral compartments (both cellular and extracellular), it is necessary to determine pH values in all of them, including the coelenteron. The coelenteron plays a crucial role by serving multiple functions such as digestion, nutrient distribution, waste removal and structural integrity²⁴. The coelenteron could play an important role in mediating the transport of molecules/ions between the external environment, the mesoglea, and the compartment where calcification occurs (i.e. the ECM).

As described earlier, symbiotic dinoflagellate-containing cells line the coelenteron and their photosynthetic activity contributes to changes in the chemistry of the coelenteron, particularly with respect to pH. Previous studies using pH microsensors in the coelenteron focused only on polyps but pH in the coenosarc has not yet been investigated. Amongst these studies, research using pH microsensors on scleractinian corals have described the variation of pH in the coelenteron (pH_{coel}) on a daily cycle^{4,25–27}. A pH increase is observed in the light due to the photosynthetic activity of dinoflagellates, while a pH decrease is observed in the dark due to the respiration of coral host and symbionts. It is therefore necessary to account for these differences when considering integrated models of physico-chemical gradients between different tissue layers of a coral²⁸. Moreover,

although the polyps are connected by the coenosarc, there are no data in the literature showing whether the composition of the coelenteric fluid is the same in polyps and coenosarc.

The pH in the coelenteron and/or ECM has already been characterised with microsensors in *Montastraea cavernosa*, *Duncanopsammia axifuga*⁴, *Galaxea fascicularis*²⁵, *Orbicella faveolata*, *Turbinaria reniformis*, *Acropora millepora*²⁹ and with microsensors and pH-sensitive dyes in *Stylophora pistillata*^{7,10,14,15,17,30,31}, *Pocillopora damicornis*¹⁵ and *Acropora sp.*^{15,32}. Although data on pH_{coel} are available for several species^{4,25–27,29,33}, they were obtained only for polyps. No previous study has addressed the comparison of pH_{coel} between polyps and coenosarc or the influence of light intensity or zooxanthellae density within a single coral species. In the present study, we chose to work with *Stylophora pistillata* since it is the coral species in which ECM chemistry has been most extensively studied using geochemical proxies, microsensors, or pH-sensitive dyes^{6,10,13,14,17,30,31}.

The aim of our study was to determine if the anatomical region (polyp/coenosarc) and light/dark conditions affect pH_{coel} and could potentially affect pH_{ECM} in a single coral species. We worked with microcolonies of *Stylophora pistillata* growing on glass slides^{34,35}. We used the pH microsensor technique used in ref. 17 for measurements in polyps and coenosarc with different levels of dinoflagellate density based on tissue colouration (Fig. 1). We first performed depth profiles in the polyps and in the coenosarc in tissue with a high dinoflagellate density, under light conditions to determine the variation of pH_{coel} . We then measured pH_{coel} of polyps and coenosarc at eight light intensities, from darkness to strong illumination, which allowed us to derive a pH_{coel} -irradiance curve and evaluate the role of photosynthesis in influencing pH in the coelenteron. Finally, we measured pH_{coel} in the coenosarc under light and dark conditions in two regions of interest characterised by visually different densities of dinoflagellates

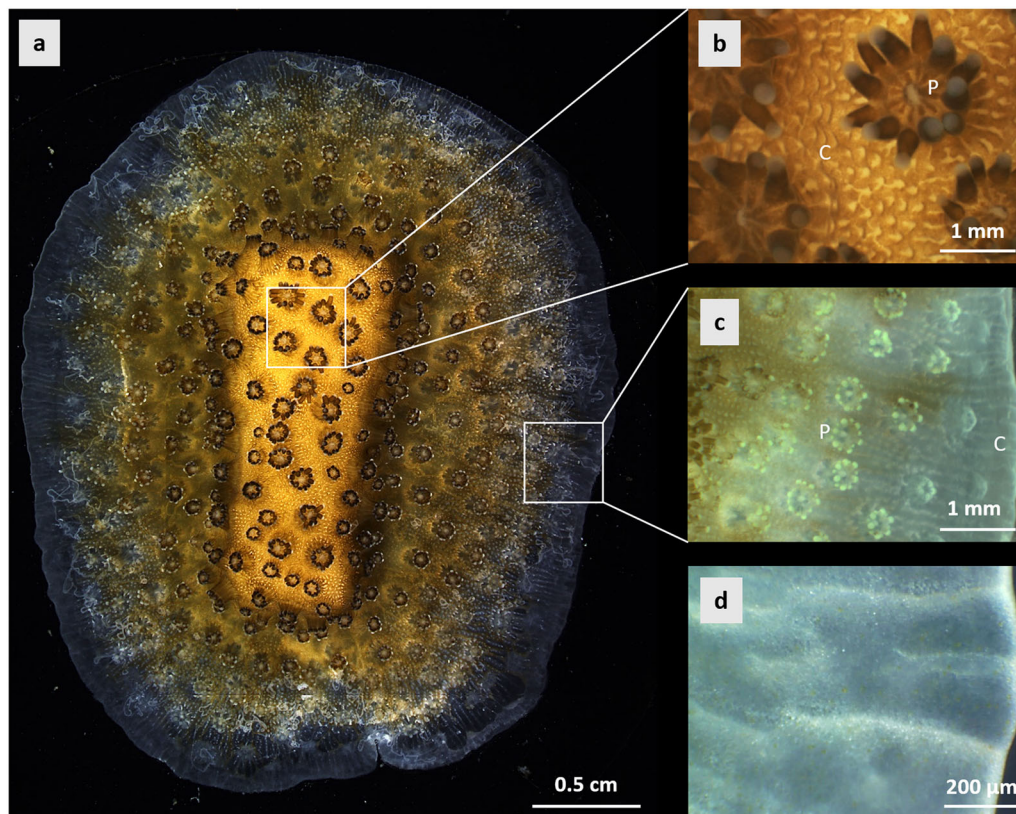


Fig. 1 | Macroscopic images of a microcolony of *Stylophora pistillata* taken from above, looking down on the sample. a Whole image of a microcolony grown on a glass coverslip. **b** Zoom of the white square in **a** indicating the centre of the microcolony with tissue characterised by a high dinoflagellate density (HDD) (brown tissue). Microsensor measurements of pH in the coelenteron of HDD tissue

were made in such area. **c** Zoom of the white square in **a** indicating the growing edge (GE) of the microcolony with tissue characterised by a low dinoflagellate density (LDD) (transparent tissue). **d** Zoom of the coenosarc at the growing edge. Microsensor measurements of pH in the coelenteron of LDD tissues were made in such area. C coenosarc, P polyp.

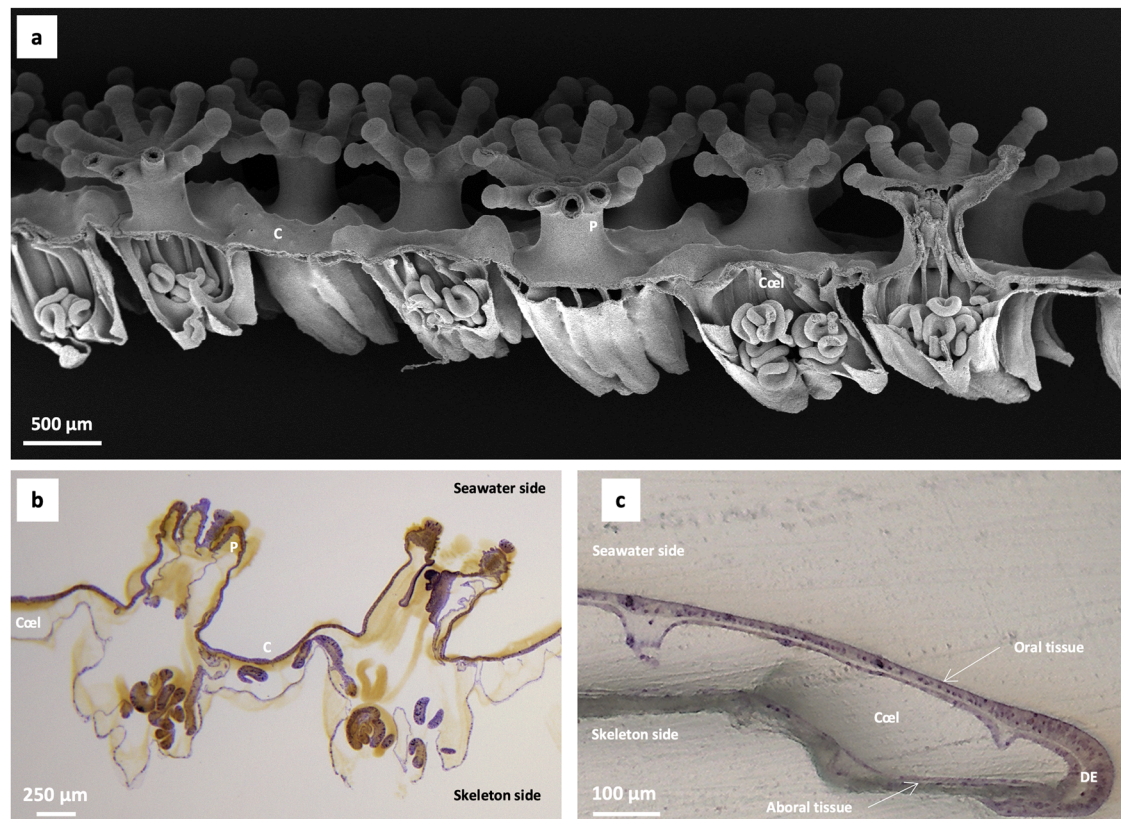


Fig. 2 | General organisation of the tissue of *Stylophora pistillata* grown on glass slides. a Decalcified microcolony prepared for scanning electron microscopy observations showing polyps and coenosarc. **b** Section of a microcolony mounted on

a glass slide stained with toluidine blue in borax showing polyps and coenosarc in HDD tissue. **c** Section of the coenosarc from LDD tissue, colouration with toluidine blue in borax. C coenosarc, Coel coelenteron, P polyp, DE distal edge.

residing within the coral tissue: a high dinoflagellate density (HDD) tissue at the centre of the microcolonies (Fig. 2a, b) growing on glass slides (brown coloured tissue) and a low dinoflagellate density (LDD) tissue at the edge of microcolonies (transparent tissue), a zone called the growing edge³⁶ (Fig. 2c).

Results

pH depth profiles in tissues with a high dinoflagellate density (HDD) under light

Representative depth profiles for pH_{coel} of a polyp and coenosarc in HDD tissue under light are shown in Fig. 3a, b, respectively. Depth profiles of pH in the light (irradiance of $200 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$) were carried out with the microsensor tip progressively inserted through the mouth of the polyp or the tissue of the coenosarc and then through the coelenteron until the maximum depth was reached. For the polyp, profiles were stopped when the polyp started to retract due to further advancement of the microsensor. In the coenosarc, profiles were stopped when the microsensor tip started to lightly bend. All pH data were collected from the surface to the maximum depth by advancing the microsensor downward. During experiments, samples were placed in a temperature-controlled seawater bath (1 L) to maintain a temperature of 25°C , with a seawater pH of 8.08 ± 0.04 (mean \pm SD) (National Bureau of Standards (NBS) scale).

The depth profile of pH in the polyp showed an increase from the mouth (depth $0 \mu\text{m}$) with a pH of 8.48, throughout the stomodeum to the entry into the coelenteron (depth $400 \mu\text{m}$) with a pH of 8.69, displaying a Δ pH of 0.21 units relative to the pH at the mouth. From the upper coelenteron (depth $400 \mu\text{m}$) to the bottom of the coelenteron (depth $1400 \mu\text{m}$), the pH stabilises and showed a variation of only 0.07 pH units between the minimum and the maximum value (pH 8.64 and 8.71, respectively).

The depth profile of pH in the coenosarc showed a similar pattern with an increase from the tissue surface (depth $0 \mu\text{m}$) with a pH of 8.37, through the different cell layers of the oral tissue to the entry into the coelenteron

(depth $100 \mu\text{m}$) with a pH of 8.73, displaying a Δ pH of 0.36 units compared to the pH at the surface of the coenosarc. From the upper coelenteron (depth $100 \mu\text{m}$) to the bottom of the coelenteron (depth $300 \mu\text{m}$), the pH stabilised and showed a variation of only 0.06 pH units between the minimum and the maximum value (pH 8.67 and 8.73, respectively).

Overall, these results show that the only difference between the polyp and the coenosarc lied in the depth at which the coelenteron was reached. Indeed, the pH exhibited a similar pattern in both the polyp and coenosarc, with an increase in the first micrometres (i.e. the stomodeum for the polyp and the oral tissue for the coenosarc) and stabilisation in the coelenteron itself.

Since pH values in both polyp and coenosarc remained stable from the top ($400 \mu\text{m}$ for the polyp and $100 \mu\text{m}$ for the coenosarc) to the bottom ($1400 \mu\text{m}$ for the polyp and $300 \mu\text{m}$ for the coenosarc) of the coelenteron, we used the mean value of the profile ($=\text{pH}_{\text{coel}}$, mean \pm SD) to make a comparison between pH_{coel} in polyp and coenosarc in the light. Thus, a mean pH_{coel} value for both the polyp and the coenosarc was determined for each sample (at least three repeated measurements were made in both polyp and coenosarc), representing the average of the repeated measurements. Figure 3c shows boxplots with the mean pH_{coel} of the five different samples of both the polyp and coenosarc. We found that the mean pH_{coel} was 8.67 ± 0.27 ($n = 5$) in the polyp, and 8.66 ± 0.18 ($n = 5$) in the coenosarc, representing a pH variation of 0.59 and 0.58 units, respectively, compared to the pH_{SW} . No statistically significant difference in pH_{coel} was found between these two anatomical regions, but pH_{coel} was significantly elevated above pH_{SW} in both regions (Wilcoxon test: $W = 0$, $P < 0.05$).

Effect of light intensity on pH in the coelenteron: pH_{coel} -irradiance curve

With pH_{coel} in the light being stable throughout most of the profile in the polyp and the coenosarc of HDD tissue (Fig. 3), it allows us to compare these

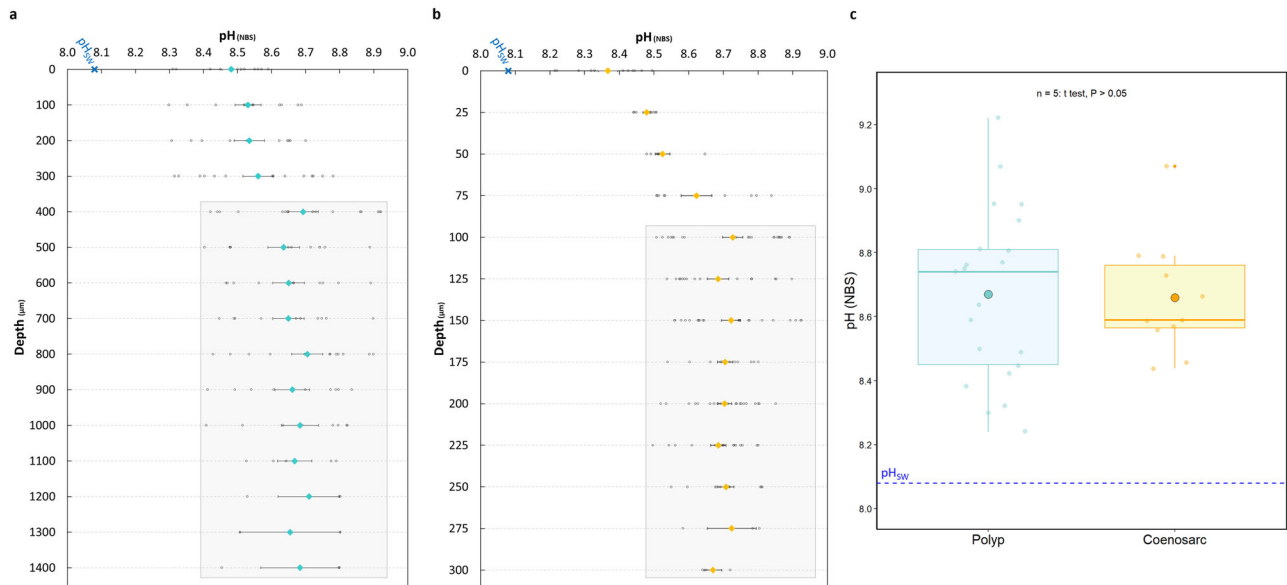


Fig. 3 | pH measurement in the coelenteron with a LIX microsensors in polyp and coenosarc (tissue with a high dinoflagellate density (HDD)) of a *S. pistillata* microcolony in the light ($200 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$). For clarity, pH depth profiles of polyp and coenosarc are separated as the depth scales are different (not the same total depth), and individual data points are shown. **a** Representative pH depth profile recorded on a polyp in HDD tissue, considering the sensor tip inserted through the mouth as reference depth 0; depths represent the interior of the polyp from the mouth to reaching the maximum depth (1400 μm); values are mean \pm SE calculated from depth profiles made on five different microcolonies, at least three profiles were made per microcolony. Light grey rectangle represents the depths at

which the microsensors is inside the coelenteron. **b** Representative pH depth profile recorded on the coenosarc in HDD tissue with the sensor tip inserted into the tissue considered as reference depth 0; depths represent the interior of the coenosarc until reaching the maximum depth (300 μm); values are mean \pm SE calculated from depth profiles made on five different microcolonies, at least three profiles were made per microcolony. Light grey rectangle represents the depths at which the microsensors is inside the coelenteron. **c** Box and whisker plots show the mean (\pm SD); the first, second (median) and third quartile; and respective whiskers (lowest and highest data point) of coelenteron pH (pH_{cel}) in the polyp and the coenosarc; the blue dotted line represents the pH_{SW} ; paired *t*-test: $t = 0.20365$; $df = 3$; $P > 0.05$, $n = 5$.

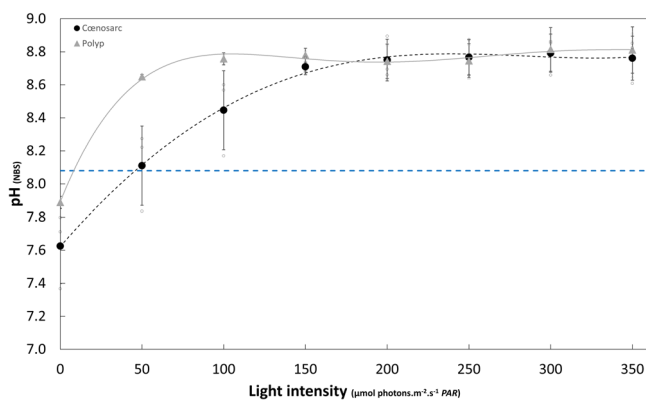


Fig. 4 | Effect of light intensity on coelenteron pH (tissue with a high dinoflagellate density (HDD)) of a *S. pistillata* microcolony sample. pH_{cel} values in the polyp (\blacktriangle) and the coenosarc (\bullet) of HDD tissue were collected at eight different light intensities during time series lasting between 40 and 60 min, pH values were taken once the signal was stabilised ($n = 3$); data were mean \pm SD and individual data points are shown; a polynomial regression curve was fitted to the data for both the polyp and the coenosarc; the cyan dotted line represents the pH_{SW} . Spearman correlation test: Polyp, $S = 16.598$; $P < 0.05$; $\rho = 0.80$. Spearman correlation test: Coenosarc, $S = 6$; $P < 0.05$; $\rho = 0.93$.

two regions at different light intensities. Measurements were carried out at eight light intensities during a time series that lasted between 40 and 60 min ($\text{pH}_{\text{SW}} = 8.08 \pm 0.04$). At each light intensity except darkness for polyps (see Methods below), we allowed the pH_{cel} to stabilise for at least 20 min before recording the values. Measurements were made in the polyp and coenosarc of three samples. Figure 4 shows the pH_{cel} -irradiance curve obtained in the coelenteron of polyps and coenosarc in HDD tissue, where pH_{cel} corresponds to the mean \pm SD. pH_{cel} increased from darkness (pH_{cel} polyp =

7.89 ± 0.04 ; pH_{cel} coenosarc = 7.63 ± 0.23) until $100 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ in the polyp ($\text{pH}_{\text{cel}} = 8.76 \pm 0.04$) and $150 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ in the coenosarc ($\text{pH}_{\text{cel}} = 8.71 \pm 0.05$) and then reached a plateau. At lower light intensities ($< 100 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$) pH_{cel} , even though not statistically significantly different, is slightly higher in the polyps than in the coenosarc which could be due to a higher density of dinoflagellates per coral biomass. Even though pH_{cel} plateaued at different light intensities, the values in both the polyp and coenosarc were stable at higher irradiance. A Spearman correlation test was performed on the data and showed a strong positive relationship between light intensity and pH of the coelenteron, in both the polyp and coenosarc (Spearman correlation test: Polyp, $S = 16.598$, $P < 0.05$, $\rho = 0.80$; Coenosarc, $S = 6$, $P < 0.05$, $\rho = 0.93$).

Working at different light intensities also allowed us to determine the optimum irradiance ($200 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$, see methods for details) at which to carry out measurements with respect to polyp behaviour and the visibility of the microsensors tip.

Mapping of pH_{cel} in coenosarc of tissue with different dinoflagellate density

Comparisons of coelenteron pH in tissue with high and low dinoflagellate densities were made exclusively in the coenosarc, because polyps were not fully formed in low dinoflagellate density regions at the growing edge (Fig. 1b, c).

We first performed a depth profile in the coenosarc of a low dinoflagellate density (LDD) tissue to determine pH variation. Results show (Supplementary Fig. 1) a decrease in pH from the tissue surface (depth 0 μm , $\text{pH} = 7.91$), to the bottom of the coenosarc (depth 80 μm , $\text{pH} = 7.81$), displaying a ΔpH of 0.27 units to the external seawater ($\text{pH}_{\text{SW}} = 8.08 \pm 0.04$). From a depth of 50 μm , pH values stabilised to the bottom of the coenosarc. Therefore, when measuring pH_{cel} in LDD tissue, the microsensors was carefully inserted through the tissue and positioned at a depth of 50–70 μm .

pH_{cel} was measured under light and dark conditions, in seven samples for HDD tissue and five samples for LDD tissue. Each sample was firstly

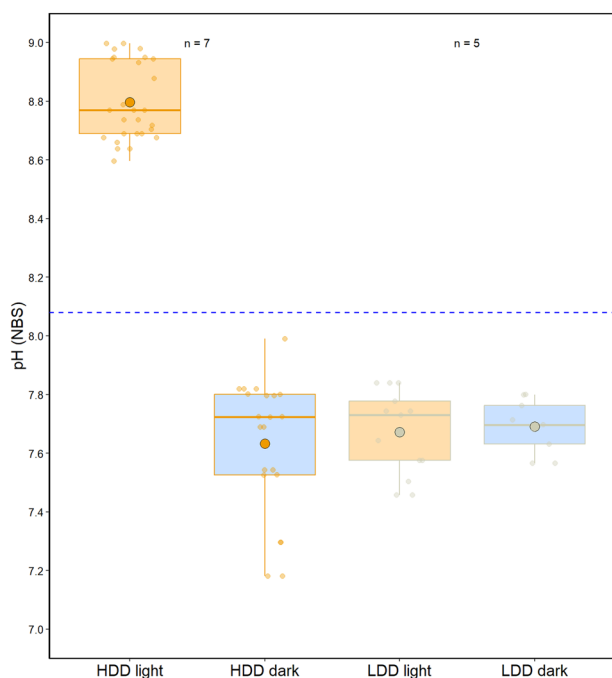


Fig. 5 | Cœlenteron pH measured in the cœnoscarc with high (HDD) or low (LDD) dinoflagellate density of a *S. pistillata* microcolony in the light and in the dark. Box and whisker plots show the mean (\pm SD); the first, second (median) and third quartile; and respective whiskers (lowest and highest data point) of cœlenteron pH ($pH_{\text{cœl}}$) obtained on seven samples for HDD tissue and five samples for LDD tissue. Cœnoscarc, where measurements were made, is shown in Fig. 1b (HDD) and Fig. 1d (LDD). The blue dotted line represents the pH_{SW} . Two-way ANOVA: Area (e.g. HDD or LDD) $F_{1,66} = 202.04$, $P < 0.05$; Light intensity $F_{1,66} = 393.84$, $P < 0.05$; Interaction $F_{1,66} = 186.02$, $P < 0.05$. Letters in superscript indicate subsets determined by Tukey's post hoc analysis.

exposed to an irradiance of $200 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ (light) and then to an irradiance of $0 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ (dark). For each sample, at least three repeated measurements were performed. For each light condition, $pH_{\text{cœl}}$ was allowed to stabilise for at least 20 min before values were taken. Mean pH values (mean \pm SD) are shown in Fig. 5 and $pH_{\text{cœl}}$ data were compared with a two-way analysis of variance and a Tukey post hoc analysis.

First, regarding HDD tissue, $pH_{\text{cœl}}$ in the light was significantly higher than $pH_{\text{cœl}}$ in the dark. $pH_{\text{cœl}}$ reached a value of 8.80 ± 0.13 under light and 7.63 ± 0.23 under dark, respectively displaying a Δ pH of 0.72 higher than seawater and 0.45 units lower than seawater ($pH_{\text{SW}} = 8.08 \pm 0.04$).

Second, regarding LDD tissue, no significant difference was found between light and dark measurements and pH remained stable under all light conditions. $pH_{\text{cœl}}$ reached a value of 7.67 ± 0.14 and 7.69 ± 0.09 in the light and dark, respectively, displaying a Δ pH of 0.41 and 0.39 lower than the external seawater ($pH_{\text{SW}} = 8.08 \pm 0.04$).

The statistical analysis performed on the data showed a significant effect of the region measured (i.e. the density of dinoflagellates within the tissue) and the light condition alone, and an interaction effect of both parameters on $pH_{\text{cœl}}$ (two-way ANOVA: area (e.g. tissue type) $F_{1,66} = 202.04$, $P < 0.05$; light intensity $F_{1,66} = 393.84$, $P < 0.05$; interaction $F_{1,66} = 186.02$, $P < 0.05$). The post hoc analysis revealed two groups: (1) cœnoscarc of high dinoflagellate density tissue under light, and (2) cœnoscarc of high dinoflagellate density tissue under dark and low dinoflagellate density tissue under both light and dark conditions.

Discussion

As the largest internal extracellular compartment in corals, the chemistry of cœlenteron is anticipated to influence pH gradients with the ECM where the coral skeleton forms. In the current study, we focused on the pH of the

cœlenteric fluid because pH is a major parameter affecting coral calcification, together with other parameters, including calcium and dissolved inorganic carbon concentrations. In colonial corals, the cœlenteric fluid is found not only in the polyps but also in the connecting tissue, the cœnosarc (Fig. 2). In the present study, we measured $pH_{\text{cœl}}$ in different anatomical regions of the coral: polyps and cœnosarc and investigated the influence of dinoflagellate density (observed by tissue colouration) in both light and dark conditions in the cœnosarc.

pH depth profiles performed in the polyps of *S. pistillata* microcolonies are consistent with the previous study of ref. ¹⁷, which used a similar experimental set-up (specifically similar seawater, coral species, feeding and light conditions). The increase in pH with depth and stabilisation of values in the cœlenteron (or gastrovascular cavity for some references listed) with little pH variation (Fig. 3) was also observed in other coral species such as *Acropora* sp., *Favia* sp., *Orbicella aveolate* and *Turbinaria reniformis*^{26,29,33}. Although pH rises similarly in the first micrometres after insertion of the microsensor tip through the mouth and then stabilises in the cœlenteron, the depths at which this compartment is reached is species-specific. In *Turbinaria reniformis* and *Acropora millepora*, the increase in pH occurs from a depth of $400\text{--}500 \mu\text{m}$ ^{29,33} and is similar to *S. pistillata* ($400 \mu\text{m}$; present study), while in *Favia* sp., the pH increases rapidly after entering the polyp mouth and reaches its maximum value at a depth of about $300 \mu\text{m}$ ²⁶. These differences in pH depth profiles are probably linked to the tissue/skeletal relationship in these various species, with some belonging to the ‘‘Complex’’ and other to the ‘‘Robust’’ clade. The latter, in which *S. pistillata* belongs, presents heavily calcified skeletons whereas the ‘‘Complex’’ corals (e.g. *Acropora* sp.) tend to be less heavily calcified³⁷. This could have an influence on the fluid chemistry of the gastrovascular cavity with polyps being more or less isolated from each other. In this study, we only focus on *S. pistillata* from the ‘‘Robust’’ clade, but it would be interesting to make a comparative study on coral species belonging to the different clades in future work.

The pH depth profiles performed in the cœnosarc of a microcolony of *S. pistillata* were similar to those of the polyps, only the depth at which the cœlenteron was reached differs (100 and $400 \mu\text{m}$ for cœnosarc and polyp, respectively) (Fig. 3a, b). This result is not surprising and can be explained by looking at the anatomy of *S. pistillata*. Indeed, the cœlenteron of the polyp is much deeper than the cœlenteron of the cœnosarc relative to the surface of the coral.

Previous studies on the cœlenteron using pH microsensors found that, on a daily cycle, $pH_{\text{cœl}}$ increases in the light due to photosynthesis, while $pH_{\text{cœl}}$ decreases in the dark due to respiration^{4,25–27}. In addition, there is a positive relationship between coral photosynthesis and calcification under light conditions, a process known as light-enhanced calcification (LEC). There are many hypotheses to explain LEC, and one of them involves the increase of pH in the cœlenteric fluid, which favours the removal of protons from the ECM^{38–40}. Many studies have investigated the photosynthesis-irradiance (PI) relationship in corals and have characterised how photosynthetic rates increase with increasing light intensity until a plateau is reached^{40–42}. However, although $pH_{\text{cœl}}$ is often assumed to be light dependent^{4,25,26}, the $pH_{\text{cœl}}$ -irradiance relationship has never been determined, and therefore the full range of $pH_{\text{cœl}}$ has remained uncharacterised. Therefore, we used multiple light intensities to measure $pH_{\text{cœl}}$ in both polyps and cœnosarc in HDD tissue. The resulting $pH_{\text{cœl}}$ -irradiance curve showed a strong positive relationship between pH and light intensity, reflecting the photosynthetic activity of the symbionts and its effects on pH cœlenteron and provides us with the full range of light-driven $pH_{\text{cœl}}$ changes (Fig. 4).

As mentioned above, symbiotic dinoflagellates are not evenly distributed throughout the coral tissue in *Stylophora pistillata*. Unlike the white-transparent tissue observed in bleached corals⁴³, such tissues are also observed in non-stressful conditions in active growing zones such as at the tip of coral branches^{44,45}. This is also the case at the growing edge prepared with the lateral preparative assay^{14,46}, and this can be clearly seen in Fig. 1a, c, d. Knowing that $pH_{\text{cœl}}$ is directly influenced by the photosynthetic activity

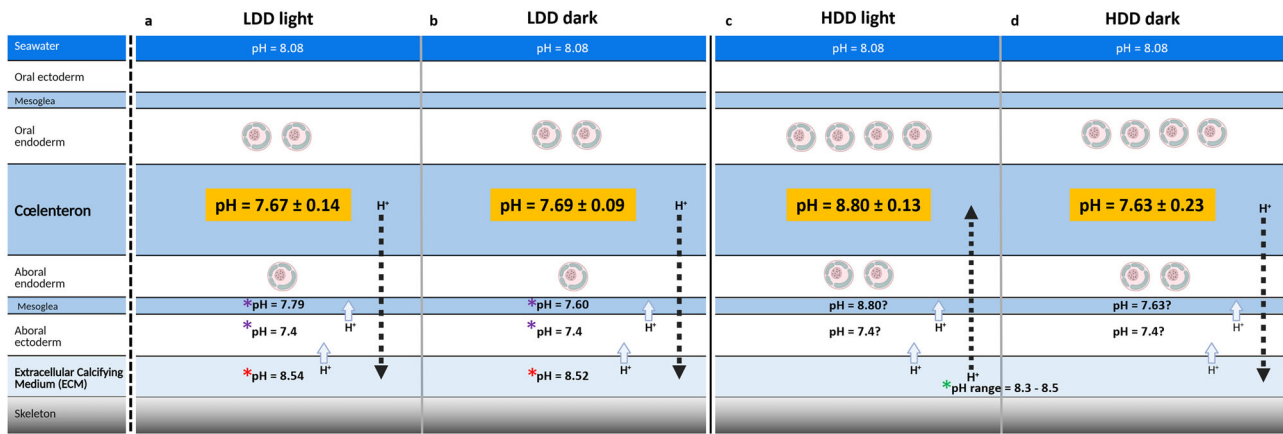


Fig. 6 | Diagram depicting a model of the influence of light and dark on proton gradients across the tissue layers of a microcolony of *S. pistillata*. a, b represent low dinoflagellate density (LDD) tissue in light and dark conditions, respectively. c, d represent high dinoflagellate density (HDD) tissue in light and dark conditions, respectively. pH values in orange correspond to coelenteron pH measured in the present study. * pH_{ECM} (total scale) from skeletal boron isotope analysis^{6,13,51,52}.

* pH of the mesoglea and aboral ectoderm measured with pH-sensitive fluorescent dye from ref. ²³. * pH_{ECM} measured with microsensors from ref. ¹⁷. Symbiotic dinoflagellates are represented in gastrodermis for both LDD and HDD tissues. Dashed arrows represent the paracellular pathway along the concentration gradient between the coelenteron and the ECM. Light blue arrows represent the removal of H^+ from the ECM via active transcellular mechanisms.

of dinoflagellates, we aimed to compare pH_{ceel} in high and low dinoflagellate density tissue under light and dark conditions. The pH_{ceel} in HDD tissue showed values above that of the external seawater under light conditions where photosynthesis occurs and below that of the surrounding seawater under dark conditions where respiration of both host and symbionts occurs (Fig. 5). The results of this study are consistent with previous studies that have shown shifts in pH_{ceel} between light and dark conditions^{4,25,26,47}. However, previous studies did not characterise full pH_{ceel} -irradiance relationships, so it is not known if previous reports of light-driven pH_{ceel} changes represent the full range of pH_{ceel} . When measured in LDD tissue (Fig. 5), pH_{ceel} remains stable in both light ($pH_{ceel} = 7.67 \pm 0.14$ at $200 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$) and dark conditions ($pH_{ceel} = 7.69 \pm 0.09$ at $0 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$), indicating that pH is kept below pH_{SW} ($pH_{SW} = 8.08 \pm 0.04$) when net respiration of both host and symbionts occurs. Even though low numbers of dinoflagellates are present at the growing edge, their combined photosynthetic activity is not sufficient to increase pH_{ceel} in the light. It is noteworthy that the values of pH_{ceel} in LDD tissue are similar to those measured under dark conditions in HDD tissue, suggesting that pH_{ceel} in this latter tissue does not decrease below a threshold value of ~ 7.6 . Such a result could be explained by similar production of CO_2 by respiration in the two zones and presumably similar rates at which CO_2 diffuses to the surrounding seawater. Also, the difference in pH in the light between pH_{ceel} in the coenosarc at the centre of the colony (HDD region) and the growing edge (LDD region) suggests, that coelenteron fluid circulation in the colony is not sufficient to lead to light-driven increases in pH_{ceel} at the growing edge.

Our results show that pH_{ceel} in LDD tissue (at the growing edge) is lower than pH_{ECM} determined in previous studies¹⁷ in light and darkness in *S. pistillata* (Fig. 6). Moreover, our pH_{ceel} values are similar to the pH values determined previously in the mesoglea, which lies between the coelenteron and ECM²³. Together these data suggest that paracellular diffusion of protons from the ECM to the coelenteron is unlikely to occur as it is against the concentration gradient (Fig. 6). Instead, active transcellular mechanisms must be involved to move protons out of the ECM via membrane transporters of the calcicoderm⁴⁸⁻⁵⁰. We recognise that Venn et al. ²³ recorded a small difference in the pH of mesoglea between light and dark conditions, and we did not measure light/dark differences of pH_{ceel} in LDD tissue in our current study. This inconsistency might be due to a difference in symbiont density at the growing edge in the samples used in the two studies.

In HDD tissue, our dark measurements are similar to the values obtained at the growing edge (LDD) in both light and darkness. Direct measurements of pH_{ECM} have never been achieved with optical verification

in the zone of high symbiont density but estimates of pH_{ECM} by skeletal boron isotope analysis (that do not discriminate between light and dark conditions) in *Stylophora pistillata* and other corals indicate that pH_{ECM} is in the range of pH 8.3 to 8.5 (total scale)^{6,13,51,52} (Fig. 6). Contrary to mesoglea pH in LDD tissue (at the growing edge²³), mesoglea pH has not been measured previously in HDD tissue in either light or dark conditions. Since pH_{ceel} was similar to pH mesoglea in the LDD tissue, we assume that this is the same in the HDD tissue. As such, there is likely to be an unfavourable gradient of protons from the ECM to the coelenteron in the dark in this zone. Indeed, paracellular diffusion of protons away from the ECM is unlikely to occur in these circumstances, and active transport mechanisms are likely to be required to maintain elevated pH_{ECM} relative to the coelenteron and seawater.

In HDD tissue in the light, our measurements indicate that pH_{ceel} is higher than previous estimates of pH_{ECM} in this zone (values by skeletal boron isotope analysis as above, Fig. 6). In contrast to dark conditions and both light and darkness in LDD tissue, the pH gradient is favourable for the diffusion of protons from the ECM into the coelenteron. We propose that paracellular and transcellular mechanisms may operate in parallel to move protons produced by calcification in the ECM across the calcicoblastic epithelium to the coelenteron. In the coelenteron, protons are then neutralised by reaction with OH^- , released by photosynthesis^{38,53,54}.

Our results and their interpretation in the diagram in Fig. 6 agree with one of the previously proposed mechanisms for light-enhanced calcification^{38,40}. Under light conditions, elevated pH_{ceel} driven by symbiont photosynthesis in HDD zones would enhance proton flux from the ECM. If this resulted in higher pH_{ECM} , then this could increase the CaCO_3 saturation state and thus lead to higher rates of precipitation. Similarly, higher pH_{ECM} would presumably also be favourable to higher pH in macropinocytotic vesicles that engulf ECM, thus potentially favoring the formation of intracellular ACC (amorphous calcium carbonate) precursors in the calcicoblastic cells²¹. It is worth pointing out that higher rates of precipitation would result in higher proton production and therefore decrease pH. As such, pH_{ECM} values found in HDD tissue may not be markedly higher than LDD tissue like the growing edge, as calcification rates may be higher. Further research is required to explore this issue. Changes in proton gradients may also modify membrane/transcellular potential, thus influencing ion transport across membranes. However, it has been shown that the light-mediated electrical potential is independent of the photosynthetic activity of the algal symbionts⁵⁵, thus ruling out this possibility. It is also important to point out that other mechanisms underlying LEC may also operate, including greater energy supply from photosynthesis for active ion

transport⁵⁶ and the supply of organic matrix precursors from the symbionts⁵⁷.

The present study showed the importance of pH_{coel} in proton gradients between the different coral's compartments and its implication on the calcification process. However, measurements were performed under controlled conditions and did not look at the impact of seawater acidification on the coelenteron and its possible impacts on gradients between the surrounding seawater and the coral. A study performed on *M. cavernosa* and *D. axifuga* showed a species-specific response to a decrease in seawater pH but focused only on coelenteron pH⁴. This study suggests that the photosynthetic activity of symbiotic dinoflagellates can partially mitigate the negative effects of ocean acidification on calcification rates. In *S. pistillata*, previous studies focused on the effects of seawater acidification on the pH of the ECM^{7,15,23}. These studies have shown that ocean acidification has a major impact on coral physiology, but the effects depend on the species, light and compartment studied. The ECM is relatively well regulated with respect to pH, but mesoglea is more pH-conforming with respect to the external seawater environment. However, the effects of ocean acidification on coelenteron pH remain unknown. This is an important area for future research as the coelenteron could act as a buffering compartment that mitigates the effects of decreasing pH_{SW} and helps maintain a favourable chemical environment for calcification in the ECM.

In summary, the present study characterised the pH_{coel} of *S. pistillata* in both polyps and coenosarc and in tissue with different dinoflagellate densities under light/dark conditions. The pH_{coel} of HDD tissue exhibits light/dark fluctuations due to the photosynthetic activity of the symbionts. By contrast, the pH_{coel} of LDD tissue measured in the coenosarc does not exhibit light/dark variations and the pH values measured here are much lower than those of seawater and ECM. When $\text{pH}_{\text{coel}} < \text{pH}_{\text{ECM}}$ (Fig. 6a, b, d), paracellular diffusion of protons from the ECM to the coelenteron is unlikely to occur as it is against the concentration gradient. Active transcellular mechanisms must be involved to move protons out of the ECM via membrane transporters of the aboral ectoderm. When $\text{pH}_{\text{coel}} > \text{pH}_{\text{ECM}}$ (Fig. 6c), the concentration gradient could be favourable for diffusion of H^+ from the ECM into the coelenteron.

The inclusion of the coelenteron in calcification models is imperative, with particular attention to its chemical composition, especially in terms of pH. The importance lies in the efficient removal of protons from the calcification site. However, pH is only one parameter that influences calcification. For a comprehensive understanding of the coelenteron carbonate chemistry, including factors such as carbonate and calcium concentration, additional experiments are needed. Furthermore, research into the effects of environmental factors, such as seawater acidification, is crucial for a more sophisticated understanding of the calcification process.

Methods

Stylophora pistillata microcolonies

S. pistillata colonies, maintained at the Centre Scientifique de Monaco, were used to produce microcolonies grown on glass slides according to the technique initially described by ref.³⁴, later referred to as the lateral skeleton preparative assay ref.³⁵, and since then used in many physiological studies^{7,10,14,17}. Briefly, pieces of microcolonies were cut with a razor blade and fixed with resin (Devcon™) on rectangular glass slides. These pieces were then left to grow (Fig. 1a) in long-term coral culture facilities supplied with flowing seawater from the Mediterranean Sea (exchange rate $170\% \cdot \text{h}^{-1}$), at a salinity of 38, temperature of 25 °C, under an irradiance of $175 \mu\text{mol photons} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ (provided by a BLV HQI Light Bulb Nepturion, 150 W) on a 12 h: 12 h photoperiod. Corals were fed both with frozen rotifers (daily) and live *Artemia salina* nauplii (twice per week). During experiments, samples were placed in a temperature-controlled seawater bath (1 L) to maintain a temperature of 25 °C, with a seawater pH of 8.08 ± 0.04 (mean \pm SD) (National Bureau of Standards (NBS) scale), and light intensities ranging from 0 to $350 \mu\text{mol photons} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ (provided by a CL 9000 LED lamp, Zeiss®, Germany and measured with a Walz US-SQS/L Submersible Spherical Micro Quantum Sensor, Heinz Walz GmbH®,

Germany). The seawater bath was filled with water from the coral culture aquaria during experiments.

Microsensor construction and calibration

pH-liquid ion exchange (LIX) microsensors were prepared as described previously¹⁷. Briefly, glass capillary tubes (borosilicate; 8 cm length; 1.5 mm diameter; Science Product, Germany) were pulled on a DMZ Universal puller (Zeitz Instruments). The micropipettes with a tip diameter of 2–5 μm were silanized and backfilled with electrolyte specific for H^+ , then front-filled with the LIX membrane containing the H^+ ionophore and let to dry for several hours to allow them to stabilise prior to measurements. Several microsensors were manufactured at once to have a stock in case of malfunction.

Calibration of the pH microsensors was performed in seawater adjusted to pH 7 to 9 by adding HCl and NaOH in 0.5 pH units (NBS scale) as described previously¹⁷. The pH of seawater was measured using a pH electrode (Mettler Toledo) previously calibrated with three commercially available pH NBS buffers (pH 4, 7, 10; Hannah Instruments Buffer Solutions).

Experimental set-up

All experiments were performed under a Leica Z16 APO macroscope (Leica Microsystems) connected to a camera system and a computer monitor that allowed live macroscopic observations (Archimed® Microvision, France). Macroscopic images of the insertion of a microsensor tip in an *S. pistillata* microcolony are shown in Supplementary Fig. 2. The use of a motorised micromanipulator allows precise movements of the microsensor on the order of micrometres (MUX2, PyroScience GmbH, Germany). The set-up used in this study is the same as that used ref.¹⁷ for depth profiles of pH obtained on polyps of *S. pistillata*. Microsensor signals were recorded every 5 s.

Light microscopy

Coral samples growing on a coverslip are fixed overnight in 4% glutaraldehyde in artificial seawater buffered to pH 7.8 with 0.1 M sodium cacodylate (according to ref.³⁴). The samples were then rinsed in distilled water before being dehydrated through a series of ethanol solutions. The coral was then embedded in EPO-TEK® (Epoxy Technology, France) for sectioning. The sections (1.0 mm) were cut using the Minitom® and a diamond cut-off wheel Minitom® (Struers, France). The section was mounted on glass slides, polished using silicon carbide foils (up to 4000 grades, lubricated with water), and stained with toluidine blue in borax and photographed with a Leica DM750P.

Scanning electron microscopy

Samples of *S. pistillata* growing on a glass slide were processed as described in ref.³. Briefly, samples were fixed overnight at 4 °C with 4% glutaraldehyde in 0.085 M Sorensen phosphate buffer at pH 7.8 with 0.5 M sucrose. Decalcification was achieved by transferring the samples to a mixture of 0.085 M Sorensen phosphate buffer, 0.5 M sucrose containing 2% glutaraldehyde and 0.5 M ethylenediaminetetraacetic acid (EDTA) at pH 7.8 and 4 °C. This solution was renewed until decalcification was completed. Decalcified samples were rinsed in Sorensen buffer, then post-fixed for 1 h at ambient temperature with 1% osmium tetroxide in Sorensen phosphate buffer. Samples were dehydrated by transfer through a graded series of ethanol ending with a concentration of 100%. After dehydration, they were incubated for 15 min in hexamethyldisilazane (HMDS)/ethanol 100% (v/v), then 30 min in HMDS 100% that was subsequently evaporated under a fume hood overnight. Samples were then coated with gold-palladium and observed at 3–5 kV with a JEOL JSM-6010LV.

pH depth profiles: polyp and coenosarc with high dinoflagellate density

Depth profiles were performed only under light conditions (irradiance of $200 \mu\text{mol photons} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$) in polyps and coenosarc of tissue with a high

dinoflagellate density (Fig. 1b) to determine the variation of coelenteron pH in *S. pistillata*. For the polyp profiles, the tip of the microsensor was positioned above the mouth, corresponding to what we defined as depth 0 (μm). The microsensor was then inserted through the mouth until the polyp began to bend and retract. We took this depth as the maximum depth. Profile data were collected from the mouth of the polyp in incremental steps of 100 μm downward to the maximum depth (bottom of the polyp). For profiles in the coenosarc, the tip of the microsensor was positioned at the tissue surface, corresponding to a depth of 0. The microsensor was inserted through the tissue until the tip of the microsensor began to lightly bend. We took this depth as the maximum depth. As with the polyp profiles, data were collected from the tissue surface in incremental steps of 25 μm as we moved downward to the maximum depth (bottom of the coenosarc). Insertion of the microsensor tip into the polyp and coenosarc was checked both visually by macroscopy and by a sudden change in the signal. To compare the pH_{coel} of polyp and coenosarc under light conditions, we used the mean value of the profile ($=\text{pH}_{\text{coel}}$) corresponding to the depths at which pH values remained stable along the depth profile through the coelenteron.

Effect of light intensities on pH in the coelenteron (HDD tissues)

pH microsensor was positioned at a depth corresponding to stable pH_{coel} in both the polyp and coenosarc of HDD tissues, and measurements were made at eight different light intensities ranging from strong illumination to darkness: 350, 300, 250, 200, 150, 100, 50 and 0 $\mu\text{mol photons.m}^{-2}.\text{s}^{-1}$. For each light intensity, values were recorded from an entire time series that lasted between 40 and 60 min. pH_{coel} for each time series were averaged after the microsensor readings stabilised. In polyps, darkness causes the polyp to retract into the corallite calyx, causing the microsensor to exit it and alter the signal. Therefore, pH_{coel} in the polyp under dark conditions corresponds to the stabilised microsensor readings prior to this complete retraction of the polyp. In addition to pH_{coel} , we also evaluated the behaviour of the polyps (whether they retracted or not), and the resolution of the macroscopic observations (whether the tip of the microsensor was clearly visible or not).

Mapping of coelenteron pH in tissues with high and low dinoflagellate density

We performed measurements under light and dark conditions in the coenosarc in two regions of interest characterised by visually different densities of dinoflagellates residing within the coral tissue: tissue with a high dinoflagellate density (HDD) versus tissue with a low dinoflagellate density (LDD) (Fig. 1b, d). For the region characterised by a low dinoflagellate density (transparent tissue), depth profiles were carried out to determine the depth of stable pH_{coel} values as was performed for polyps and coenosarc in HDD tissue (see Methods above). The microsensor was inserted through the tissue until the maximum depth was reached. Data were collected from the tissue surface to the maximum depth (bottom of the coenosarc) in incremental steps of 10–20 μm , moving downward. The microcolonies were selected so that the growing edge was wide enough for access with microsensors. pH_{coel} was measured within the first 400 μm from the edge of the sample, as has been done previously with confocal microscopy^{14,17}. In this area, the oral and aboral epithelia (including the calciderm) are present (Fig. 2), and a new skeleton is in the process of forming¹⁴. Once the depth of measurement was determined, for each region of interest (coenosarc with high or low density of dinoflagellates), measurements were made at an irradiance of 200 $\mu\text{mol photons.m}^{-2}.\text{s}^{-1}$ (light) and 0 $\mu\text{mol photons.m}^{-2}.\text{s}^{-1}$ (dark). For each sample, pH_{coel} was recorded first under light and then under dark conditions during a time series of 40 to 60 min. The first 20 min of each time series were discarded to allow stabilisation of the signal. After the dark period, the light was turned back on for at least 10 min before the microsensor was removed and positioned in seawater. As previously described, the pH_{coel} values for each time series were averaged for each replicate after the microsensor readings were stabilised. pH_{coel} replicates were obtained in separate samples and averaged ($\pm\text{SD}$) for final values.

Statistics and reproducibility

Seven samples of *S. pistillata* grown in long-term coral culture facilities on glass slides at the Centre Scientifique de Monaco were used for this study. For each sample, at least three replicate measurements were performed under all conditions to allow statistical analysis if required. Calibration curves, graphs and statistical analyses were performed using Excel and the software RStudio⁵⁸. Spearman correlation test, *T*-tests, and two-way analyses of variance (ANOVA) were performed on the data. Post hoc analyses were also performed as needed. All statistical analysis performed in the current study are shown in Supplementary Tables 1, 2.

Reporting summary

Further information on research design is available in the Nature Portfolio Reporting Summary linked to this article.

Data availability

Numerical source data for graphs and charts can be found within the Supplementary Data file. Additional information and relevant data will be available from the corresponding author upon reasonable request. The datasets presented in this study can be found in the online repository: <https://www.pangaea.de/tok/358beef9d2a11b64a8f5e964d6543b1f49ba056b>.

Received: 5 September 2023; Accepted: 19 February 2024;

Published online: 29 February 2024

References

- Spalding, M. D., Ravilious, C. & Green, E. P. *World Atlas of Coral Reefs* (Univ. California Press, 2001).
- Porter, J. W. & Tougas, J. I. Reef ecosystems: threats to their biodiversity. *Encycl. Biodivers.* **5**, 73–95 (2001).
- Tambutté, S. et al. Coral biomineralization: from the gene to the environment. *J. Exp. Mar. Biol. Ecol.* **408**, 58–78 (2011).
- Bove, C. B., Whitehead, R. F. & Szmant, A. M. Responses of coral gastrovascular cavity pH during light and dark incubations to reduced seawater pH suggest species-specific responses to the effects of ocean acidification on calcification. *Coral Reefs* **39**, 1675–1691 (2020).
- Cameron, L. P. et al. Impacts of warming and acidification on coral calcification linked to photosymbiont loss and deregulation of calcifying fluid pH. *J. Mar. Sci. Eng.* **10**, 1106 (2022).
- Holcomb, M. et al. Coral calcifying fluid pH dictates response to ocean acidification. *Sci. Rep.* **4**, 1–4 (2014).
- Venn, A. A. et al. Impact of seawater acidification on pH at the tissue-skeleton interface and calcification in reef corals. *Proc. Natl Acad. Sci. USA* <https://doi.org/10.1073/pnas.1216153110> (2013).
- Comeau, S., Cornwall, C. E. & McCulloch, M. T. Decoupling between the response of coral calcifying fluid pH and calcification to ocean acidification. *Sci. Rep.* **7**, 1–10 (2017).
- Hoegh-Guldberg, O., Poloczanska, E. S., Skirving, W. & Dove, S. Coral reef ecosystems under climate change and ocean acidification. *Front. Mar. Sci.* **4**, 1–20 (2017).
- Tambutté, E. et al. Morphological plasticity of the coral skeleton under CO_2 -driven seawater acidification. *Nat. Commun.* **6**, 7368 (2015).
- Allemand, D., Tambutté, É., Zoccola, D. & Tambutté, S. Coral Calcification, Cells to Reefs. In *Coral Reefs: An Ecosystem in Transition* (eds. Dubinsky, Z. & Stambler, N.) vol. 3 119–150 (Springer Netherlands, Dordrecht, 2011).
- LaJeunesse, T. C. et al. Systematic revision of symbiodiniaceae highlights the antiquity and diversity of coral endosymbionts. *Curr. Biol.* **28**, 2570–2580.e6 (2018).
- Guillermic, M. et al. Thermal stress reduces pocilloporid coral resilience to ocean acidification by impairing control over calcifying fluid chemistry. *Sci. Adv.* **7**, eaba9958 (2021).

14. Venn, A. A., Tambutté, E., Holcomb, M., Allemand, D. & Tambutté, S. Live tissue imaging shows reef corals elevate pH under their calcifying tissue relative to seawater. *PLoS ONE* **6**, e20013 (2011).
15. Venn, A. A. et al. Effects of light and darkness on pH regulation in three coral species exposed to seawater acidification. *Sci. Rep.* **9**, 1–12 (2019).
16. Ries, J. B. A physicochemical framework for interpreting the biological calcification response to CO₂-induced ocean acidification. *Geochim. Cosmochim. Acta* **75**, 4053–4064 (2011).
17. Sevilgen, D. S. et al. Full in vivo characterization of carbonate chemistry at the site of calcification in corals. *Sci. Adv.* **5**, 1–10 (2019).
18. Cyronak, T., Schulz, K. G. & Jokiel, P. L. The Omega myth: what really drives lower calcification rates in an acidifying ocean. *ICES J. Mar. Sci.* **73**, 558–562 (2016).
19. Hohn, S. & Merico, A. Quantifying the relative importance of transcellular and paracellular ion transports to coral polyp calcification. *Front. Earth Sci.* **2**, 1–11 (2015).
20. Willard, H. F., Deutekom, E. S., Allemand, D., Tambutté, S. & Kaandorp, J. A. Testing hypotheses on the calcification in scleractinian corals using a spatio-temporal model that shows a high degree of robustness. *J. Theor. Biol.* **561**, 111382 (2023).
21. Sun, C. Y. et al. From particle attachment to space-filling coral skeletons. *Proc. Natl Acad. Sci. USA* **117**, 30159–30170 (2020).
22. Gilbert, P. U. P. A. et al. Biomineralization: integrating mechanism and evolutionary history. *Sci. Adv.* **8**, eabl9653 (2022).
23. Venn, A. A., Tambutté, E., Comeau, S. & Tambutté, S. Proton gradients across the coral calcifying cell layer: Effects of light, ocean acidification and carbonate chemistry. *Front. Mar. Sci.* **9**, 1–15 (2022).
24. Shick, J. M. *A Functional Biology of Sea Anemones* (Springer, 1991).
25. Al-Horani, F. A., Al-Moghrabi, S. M. & De Beer, D. The mechanism of calcification and its relation to photosynthesis and respiration in the scleractinian coral *Galaxea fascicularis*. *Mar. Biol.* **142**, 419–426 (2003).
26. Kuhl, M., Cohen, Y., Dalsgaard, T., Jorgensen, B. B. & Revsbech, N. P. Microenvironment and photosynthesis of zooxanthellae in scleractinian corals studied with microsensors for O₂, pH and light. *Mar. Ecol. Prog. Ser.* **117**, 159–177 (1995).
27. Agostini, S. et al. Biological and chemical characteristics of the coral gastric cavity. *Coral Reefs* **31**, 147–156 (2012).
28. Putnam, H. M., Barott, K. L., Ainsworth, T. D. & Gates, R. D. The vulnerability and resilience of reef-building corals. *Curr. Biol.* **27**, R528–R540 (2017).
29. Cai, W. J. et al. Microelectrode characterization of coral daytime interior pH and carbonate chemistry. *Nat. Commun.* **7**, 1–8 (2016).
30. Barott, K. L., Venn, A. A., Thies, A. B., Tambutté, S. & Tresguerres, M. Regulation of coral calcification by the acid-base sensing enzyme soluble adenylyl cyclase. *Biochem. Biophys. Res. Commun.* **525**, 576–580 (2020).
31. Comeau, S. et al. Coral calcifying fluid pH is modulated by seawater carbonate chemistry not solely seawater pH. *Proc. R. Soc. B Biol. Sci.* **284**, 20161669 (2017).
32. Ohno, Y. et al. An aposymbiotic primary coral polyp counteracts acidification by active pH regulation. *Sci. Rep.* **7**, 1–8 (2017).
33. Yuan, X. et al. Quantitative interpretation of vertical profiles of calcium and pH in the coral coelenteron. *Mar. Chem.* **204**, 62–69 (2018).
34. Muscatine, L., Tambutte, E. & Allemand, D. Morphology of coral desmocytes, cells that anchor the calicoblastic epithelium to the skeleton. *Coral Reefs* **16**, 205–213 (1997).
35. Raz-Bahat, M., Erez, J. & Rinkevich, B. In vivo light-microscopic documentation for primary calcification processes in the hermatypic coral *Stylophora pistillata*. *Cell Tissue Res.* **325**, 361–368 (2006).
36. Venn, A. A. et al. Coral calcification at the cellular scale: insight through the ‘window’ of the growing edge. *Front. Invertebr. Physiol. A Collect. Rev.* (in the press).
37. Romano, S. L. & Palumbi, S. R. Evolution of scleractinian corals inferred from molecular systematics. *Science* **271**, 640–642 (1996).
38. Allemand, D. et al. Biomineralisation in reef-building corals: from molecular mechanisms to environmental control. *Comptes Rendus Palevol* **3**, 453–467 (2004).
39. Goreau, T. F. The physiology of skeleton formation in corals. I. A method for measuring the rate of calcium deposition by corals under different conditions. *Biol. Bull.* **116**, 59–75 (1959).
40. Gattuso, J.-P., Allemand, D. & Frankignoulle, M. Photosynthesis and calcification at cellular, organismal and community levels in coral reefs: a review on interactions and control by carbonate chemistry. *Am. Zool.* **39**, 160–183 (1999).
41. Chalker, B. E., Dunlap, W. C. & Oliver, J. K. Bathymetric adaptations of reef-building corals at Davies Reef, Great Barrier Reef, Australia. II. Light saturation curves for photosynthesis and respiration. *J. Exp. Mar. Biol. Ecol.* **73**, 37–56 (1983).
42. Chalker, B. E. Simulating light-saturation curves for photosynthesis and calcification by reef-building corals. *Mar. Biol.* **63**, 135–141 (1981).
43. Brown, B. E. Coral bleaching: causes and consequences. *Coral Reefs* **16**, 129–138 (1997).
44. Jokiel, P. L. Ocean acidification and control of reef coral calcification by boundary layer limitation of proton flux. *Bull. Mar. Sci.* **87**, 639–657 (2011).
45. Pearse, V. B. & Muscatine, L. Role of symbiotic algae (Zooxanthellae) in coral calcification. *Biol. Bull.* **141**, 350–363 (1971).
46. Tambutté, E. et al. Observations of the tissue-skeleton interface in the scleractinian coral *Stylophora pistillata*. *Coral Reefs* **26**, 517–529 (2007).
47. Furla, P., Bénazet-Tambutté, S., Jaubert, J. & Allemand, D. Functional polarity of the tentacle of the sea anemone *Anemonia viridis*: role in inorganic carbon acquisition. *Am. J. Physiol.* **274**, 303–310 (1998).
48. Capasso, L., Ganot, P., Planas-Bielsa, V., Tambutté, S. & Zoccola, D. Intracellular pH regulation: characterization and functional investigation of H⁺ transporters in *Stylophora pistillata*. *BMC Mol. Cell Biol.* **22**, 1–19 (2021).
49. Zoccola, D. et al. Bicarbonate transporters in corals point towards a key step in the evolution of cnidarian calcification. *Sci. Rep.* **5**, 9983 (2015).
50. Zoccola, D. et al. Molecular cloning and localization of a PMCA P-type calcium ATPase from the coral *Stylophora pistillata*. *Biochim. Biophys. Acta* **1663**, 117–126 (2004).
51. Krief, S. et al. Physiological and isotopic responses of scleractinian corals to ocean acidification. *Geochim. Cosmochim. Acta* **74**, 4988–5001 (2010).
52. McCulloch, M., Falter, J., Trotter, J. & Montagna, P. Coral resilience to ocean acidification and global warming through pH up-regulation. *Nat. Clim. Chang.* **2**, 623–627 (2012).
53. Furla, P., Galgani, I., Durand, I. & Allemand, D. Sources and mechanisms of inorganic carbon transport for coral calcification and photosynthesis. *J. Exp. Biol.* **203**, 3445–3457 (2000).
54. Furla, P. et al. The symbiotic anthozoan: a physiological chimera between alga and animal. *Integr. Comp. Biol.* **45**, 595–604 (2005).
55. Taubner, I., Hu, M. Y., Eisenhauer, A. & Bleich, M. Electrophysiological evidence for light-activated cation transport in calcifying corals. *Proc. R. Soc. B Biol. Sci.* **286**, 20182444 (2019).
56. Chalker, B. E. & Taylor, D. L. Light-enhanced calcification, and the role of oxidative phosphorylation in calcification of the coral *Acropora cervicornis*. *Proc. R. Soc. Lond. B. Biol. Sci.* **190**, 323–331 (1975).
57. Muscatine, L. & Cernichiaro, E. Assimilation of photosynthetic products of zooxanthellae by a reef coral. *Biol. Bull.* **137**, 506–523 (1969).
58. RStudio Team. RStudio: integrated development environment for R (RStudio, PBC, 2022).

Acknowledgements

We would like to thank Dominique Desgré and Xavier Maccario for coral maintenance. This study was conducted as part of the Centre Scientifique de Monaco research programme, supported by the Government of the Principality of Monaco.

Author contributions

L.C. and E.T. carried out research, contributed to the conception and design of the study and analysed data. L.C., A.V., S.T. and E.T. wrote the manuscript. All authors contributed to the article and approved the submitted version.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information The online version contains supplementary material available at <https://doi.org/10.1038/s42003-024-05938-8>.

Correspondence and requests for materials should be addressed to Sylvie Tambutté.

Peer review information *Communications Biology* thanks Cesar Pachterres, Alina Szmant and the other, anonymous, reviewer(s) for their

contribution to the peer review of this work. Primary Handling Editors: Linn Hoffmann and David Favero. A peer review file is available.

Reprints and permissions information is available at <http://www.nature.com/reprints>

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons licence, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons licence, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons licence and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this licence, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2024