ARTICLEOPENSynthesis of ultrathin two-dimensional nanosheets and van derWaals heterostructures from non-layered γ-CuI

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Two-dimensional (2D) nanosheets have attracted considerable recent interest for their atomically thin geometry and unique thickness-dependent electronic properties. The 2D nanosheets studied to date are generally limited to intrinsically layered materials, in which the covalently bonded atomic layers are held together by weak van der Waals forces and can be readily exfoliated to single or few-atom thick nanosheets. To prepare 2D nanosheets from non-layered materials can greatly expand the scope of 2D materials, but is much less straightforward. Here, we report the successful synthesis of ultrathin nanosheets from non-layered γ -Cul on SiO₂/Si substrate using a facile physical vapor deposition process. The resulting γ -Cul nanosheets display a triangular and hexagonal geometry with the lateral dimension up to 5 µm and thickness down to 1 nm. Raman spectroscopy, X-ray diffraction, and transmission electron microscopy studies demonstrate the resulting nanosheets retain single-crystalline γ -Cul phase. Additionally, we further show the γ -Cul nanosheets can be readily grown on other 2D materials (e.g., 2D-WS₂, 2D-WS₂) to form van der Waals heterostructures (vdWHs). Optical microscopy images and Raman intensity mappings confirm the formation of γ -Cul/WS₂ and γ -Cul/WS₂ vertical heterostructures. The electrical transport studies show that γ -Cul nanosheets exhibit a low resistivity of ~0.3 Ω cm and γ -Cul/WS₂ vertical heterostructures open a pathway to ultrathin nanosheets and van der Waals heterostructures from non-layered materials and could open up exciting opportunities in electronics and optoelectronics.

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INTRODUCTION

Two-dimensional (2D) materials have attracted significant attention as a new generation of atomically thin material for nextgeneration electronic and optoelectronic devices due to their ultrathin geometry and unique thickness-dependent physical properties. In the past decade, significant efforts have been devoted to synthesizing graphene¹ and graphene-like 2D materials including hexagonal BN², transition metal dichalcogenides (TMDs) (such as MOS_2 , $^{3}WS_2$, 4 and WSe_2 ⁵), metal oxides (such as MoO₃^{,6} WO₃^{,7}), and metal halides (such as PbI₂^{,8} CdI₂^{,9} BiI₃^{,10} and Crl_{3¹¹}) using various approaches. These 2D materials typically feature unconventional physical properties distinct from their bulk crystals, which can open up exciting opportunities for both the fundamental investigation of low dimensional chemistry and physics and potential technological applications at the limits of single atomic thickness.^{12–21} Some available 2D materials may also be flexibly combined to produce new van der Waals heterostructures (vdWHs) with atomic sharp modulation of chemical composition and electronic structure. Such vdWHs can enable the creation of tailored heterojunctions without the traditional lattice matching requirement, and thus offer a much more flexible approach for heterogeneous material integration than the traditional semiconductor heterostructures. It can thus open up a limitless possibility to nearly arbitrarily combine and control different properties, and develop exciting new technologies beyond the reach of existing materials. Recent studies have already demonstrated the exciting potentials of 2D materials and vdWHs in nano-electronics applications, such as atomically thin transistors,^{22,23} vertical field-effect transistors,²⁴ and optoelectronics applications, such as photodetectors²⁵ and light-emitting diodes.^{26,27}

Although it has been relatively straightforward to produce ultrathin 2D nanosheets from intrinsically layered materials using various synthetic approaches, it is considerably more difficult to grow highly anisotropic ultrathin nanosheets from non-layered materials due to their intrinsic three-dimensional lattice structure. Nonetheless, a few examples of 2D nanosheets of some nonlayered materials (such as ZnSe,²⁸ PbS²⁹) and heterostructures (such as CdS/MoS₂,³⁰ PbS/MoS₂³¹) have been reported recently. Copper iodide (Cul), as a non-layered I–VII group compound, crystallizes into three different phases: α , β , and γ and shows different crystal lattices with increasing temperature. Cul undergoes a phase transition from the cubic y-phase to the hexagonal β -phase above 369 °C. The β -phase can be further transformed to the cubic α -phase with increasing temperature above 407 °C.³² Figure 1a, b is the stick-and-ball crystal structure model of y-Cul. The low-temperature γ -phase Cul crystal is a p-type semiconductor³³ and has a cubic zinc blende structure with the space group F-43 m. In the crystal of γ -Cul, each Cu⁺ ion is tetrahedrally surrounded by four I ions (Fig. 1a). Currently, γ -Cul has been

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extensively applied in fields such as organic catalysts,³⁴ structural template of organic semiconductor,³⁵ solid state dye-sensitized solar cells^{36,37} and hole conductor in perovskite solar cells.^{38,39}

The preparations of γ -Cul nanoparticles,⁴⁰ nanowires,⁴¹ thin films⁴² have been studied using either physical or chemical methods. Furthermore, there are several reports on the synthesis of the y-Cul single crystal nanoplates with the thickness of 60-90 nm via PEG/TSA-assisted solution-based methods.43,44 However, the synthesis of regular and ultrathin y-Cul nanosheets has not been reported. Here, we report a facile physical vapor deposition method to prepare high-quality single-crystalline y-Cul nanosheets with the thickness down to 1 nm. Optical microscopy (OM), atomic force microscopy (AFM) and scanning electron microscopy (SEM) studies show triangular and hexagonal nanosheets can be readily grown on 300 nm SiO₂/Si substrate with the lateral dimension up to 5 µm and thicknesses down to 1 nm. Raman spectroscopy studies of the nanosheets reveal the transverse-optical (TO) mode of γ -Cul at around 122 cm⁻¹. X-ray diffraction (XRD) studies show highly crystalline nature of the obtained y-Cul samples with cubic phase. Energy-dispersive X-ray spectroscopy (EDX) studies reveal that the atomic ratio between Cu and I is about 1:1. Transmission electron microscopy (TEM) and the corresponding selected area electron diffraction (SAED) studies demonstrate that the y-Cul nanosheets are single crystalline. Take a step further, we also show that the y-Cul nanosheets can also be readily grown on typical 2D materials to form y-Cul/ WSe₂ and γ -Cul/WS₂ van der Waals vertical heterostructures (vdWHs). The current–voltage curves show that γ -Cul nanosheets have low resistivity and γ -Cul/WS₂ vertical heterostructure show p–n diode behavior with distinct current rectification behavior.

RESULTS AND DISCUSSION

The y-Cul nanosheets were synthesized on SiO₂/Si substrate using a home-built physical vapor deposition (PVD) system (Figure S1), with y-Cul powder as the precursor and argon as the carrier gas. The as-synthesized y-Cul nanosheets typically display a triangle shape and occasionally a hexagonal shape with lateral size in the range of 1–5 μm (Fig. 2a, Figure S2). The different γ-Cul nanosheets show highly distinct colors, which can be attributed to variable optical interference resulted from the different thicknesses. In addition to the relatively thick nanosheets with obvious contrast to the substrate color, there are a large number of triangular domains with rather weak contrast (highlighted by white arrows in Fig. 2a). The thickness of nanosheets ranges from 300 nm to 1 nm (Figure S2, Fig. 2b). Figure 2b further shows a high resolution OM image of an ultrathin y-Cul nanosheet grown on the 300 nm SiO₂/Si substrate. The triangle nanosheet shows nearly the identical color with substrate due to its ultrathin thickness. The thickness can be further characterized using AFM. Figure 2c reveals the corresponding AFM image where the nanosheet has a



Fig. 1 Stick-and-ball crystal structures of γ-Cul. **a** Side view of the cubic crystal structure of γ-Cul along the (111) plane. **b** Top view of γ-Cul cubic crystal above the (111) plane. Purple and yellow spheres correspond to copper and iodine, respectively



Fig. 2 The basic characterizations of γ -Cul nanosheets. **a** The OM image of the γ -Cul nanosheets grown on the SiO₂/Si substrate; The white arrows highlight the ultrathin nanosheets that are barely visible. Scale bar, 10 µm. **b**, **c** The OM image and AFM image of an ultrathin γ -Cul nanosheet. Scale bar, 2, 1 µm, respectively. **d** Raman spectrum of the γ -Cul nanosheets. Raman experiment was performed in a confocal spectrograph using a 532 nm excitation laser. **e** The Raman intensity mapping of a typical γ -Cul nanosheet with Raman peak located at 122 cm⁻¹. Scale bar, 2 µm. **f** XRD pattern of γ -Cul nanosheets grown on the SiO₂/Si substrate

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Fig. 3 SEM and TEM characterizations of γ -Cul nanosheets. **a** SEM image of triangular γ -Cul nanosheets grown on SiO₂/Si substrate; Scale bar, 1 µm. **b** SEM-EDX profile for the obtained γ -Cul nanosheets. **c** LR-TEM image of a typical triangle γ -Cul nanosheet supported on a grid of copper; Scale bar, 50 nm. **d** HR-TEM image of a γ -Cul nanosheet, the inset is the corresponding SAED pattern of the nanosheet; Scale bar, 2 nm, the scale bar in the inset is 2 1/nm

thickness of $\sim 1 \text{ nm}$, indicating the formation of ultrathin 2D structure.

We have next conducted Raman and XRD studies on the resulted y-Cul nanosheets (Fig. 2d-f). The Raman spectra of the resulted y-Cul nanosheets exhibits a prominent characteristic peak around 122 cm^{-1} when excited with a 532 nm laser (Fig. 2d), in agreement with the TO mode of bulk y-Cul at room temperature.⁴⁵ We were not able to observe the longitudinal-optical (LO) mode at 140 cm⁻¹, which can be interpreted by the reason that the LO peak may be sub-merged in the broad TO peak resulting from the existence of the disorder in the structure of y-Cul.⁴⁶ A spatially resolved mapping of the Raman signal (122 cm⁻¹) shows highly uniform contrast throughout the entire nanosheet (Fig. 2e), suggesting highly uniform crystalline guality across the entire nanosheet. The XRD data indicates that the diffraction peaks can be indexed to cubic phase γ -Cul, which has a space group F-43 m with lattice parameters of a = b = c = 6.034 Å (JCPDS no. 76–0207) (Fig. 2f). A strong diffraction peak of the (111) plane suggests that the as-grown y-Cul nanosheets tend to be preferentially oriented along the (111) direction. Additionally, the stability of y-Cul nanosheets was also studied. Figure S3 shows the OM images and Raman spectra of freshly prepared Cul nanosheets and those obtained after a 5 days aging under ambient conditions. The nearly unchanged peak position and weakly reduced peak intensity indicate no significant degradation of the PVD-derived Cul nanosheets.

The synthesized γ -Cul nanosheets are further characterized using SEM and TEM studies (Fig. 3). A typical SEM image reveals the well-faceted triangular nanosheets deposited on SiO₂/Si substrate (Fig. 3a). The SEM-EDX spectroscopy demonstrates that the nanosheets consist of Cu and I elements (the exhibited O and Si elements are from the SiO₂ substrate), with the atomic ratio of Cu and I approximately 1 (Fig. 3b). Figure 3c shows the low-resolution TEM (LR-TEM) image of a typical γ -Cul nanosheet, where well-faced triangle shape of the nanosheet is clearly identified. The high-resolution TEM (HR-TEM) image of the nanosheet shows well-resolved lattice planes with a periodic atomic arrangement, confirming a single-crystalline nature of the resulted 2D

nanosheets. The lattice spacing of 0.34 and 0.20 nm can be assigned to the 1/3(422) and (220) planes of cubic structure (Fig. 3d), respectively. The corresponding SAED pattern (inset of Fig. 3d) shows six-fold rotational symmetry for the diffraction spots, which can be indexed to the cubic-blend structure of γ -Cul along the [111] zone axis. The single set of diffraction spots further demonstrates the single crystalline nature of the γ -Cul nanosheets.

Furthermore, we also found that the y-Cul nanosheets can be readily grown on 2D transition metal dichalcogenides (e.g., 2D-WSe₂, 2D-WS₂) to form vertical heterostructures by using a twostep PVD progress. Firstly, we grew the single-crystal WSe₂ (WS₂) nanosheets on the SiO₂/Si substrate using a PVD method.^{24,47,48} Then, we directly grew the Cul nanosheets on the as-grown 2D-WSe₂ (WS₂) substrate (more details are described in the Methods). Interestingly, the y-Cul nanosheets can readily nucleate and grow on the top surface of WSe₂ (WS₂) nanosheets, forming the γ -Cul/ WSe₂ (y-Cul/WS₂) vertical vdWHs (Fig. 4a-c). Figure 4a shows the OM image of the γ -Cul/WSe₂ vertical heterostructures, where the green and light gray domains are the y-Cul nanosheets, and the purple domains are the underlying WSe₂ nanosheets. Figure 4b, c show optical images of the representative y-Cul/WSe₂ and y-Cul/ WS₂ vertical heterostructures. It is noted that γ -Cul/WSe₂ and γ -Cul/WS₂ heterostructures apparently show a parallel orientation between the y-Cul hexagonal domains and the underlying TMDs, suggesting a preferred van der Waals epitaxial relationship.³⁰ Figure 4d–f shows an optical image of a γ -Cul/WSe₂ vertical heterostructure, and the corresponding Raman mappings at 122 cm⁻¹ (for γ -Cul) and 248 cm⁻¹ (for WSe₂). The Raman studies clearly confirm the formation of vertical heterostructures. Besides, similar y-Cul/WS₂ heterostructure have also clearly confirm the formation of vertical heterostructures (Fig. 4g-i). Photoluminescence (PL) spectra on the resulted y-Cul/WSe₂ and y-Cul/WS₂ heterostructures were shown in Fig. 4j, k. The results indicate that the WSe₂ and WS₂ monolayers exhibit strong PL emission with a dominant emission peak locating at 769 and 630 nm, respectively, whereas both γ -Cul/WSe₂ and γ -Cul/WS₂ regions show apparent PL guenching, with essentially no detectable PL. The observed quenching of the PL in the heterostructure may be attributed to



Fig. 4 OM, Raman, and PL characterizations of the γ -Cul/WSe₂ and γ -Cul/WS₂ heterostructures. **a** OM image of the γ -Cul/WSe₂ heterostructures on the SiO₂/Si substrate. Scale bar, 10 µm. **b**, **c** High magnification OM image of the representative γ -Cul/WSe₂ and γ -Cul/WSe₂ vertical heterostructures. Scale bar, 5 and 2 µm, respectively. **d** OM image of a typical γ -Cul/WSe₂ vertical heterostructure used for Raman characterization. Scale bar, 2 µm. **e**, **f** Raman intensity mappings at 122 and 248 cm⁻¹, corresponding to the characteristic Raman peak of Cul and WSe₂, respectively. Scale bar, 2 µm. **g** OM image of a typical γ -Cul/WS₂ vertical heterostructure used for Raman characterization. Scale bar, 2 µm. **g** OM image of a typical γ -Cul/WS₂ vertical heterostructure used for Raman characterization. Scale bar, 2 µm. **g** OM image of a typical γ -Cul/WS₂ vertical heterostructure used for Raman characterization. Scale bar, 2 µm. **g** OM image of a typical γ -Cul/WS₂ vertical heterostructure used for Raman characterization. Scale bar, 2 µm. **h**, **i** Raman intensity mappings at 122 and 350 cm⁻¹, corresponding to the characteristic Raman peak of Cul and WS₂, respectively. Scale bar, 2 µm. **j** PL spectra observed from bare WSe₂ region (black curve) and Cul/WSe₂ heterostructure region (red curve). **k** PL spectra observed from bare WSe₂ heterostructure region (red curve).

the spontaneous separation of charge carriers in the junction region. 49,50

We have further studied the electronic properties of the γ -Cul nanosheets and γ -Cul/WS₂ vertical heterostructure (Fig. 5). The insets of Fig. 5a, b show optical images of two γ -Cul nanosheets devices. The current–voltage (*I–V*) characteristic curves at room temperature show the linear and symmetric relationship, indicating that the good ohmic contacts are formed (Fig. 5a, b). The resulting γ -Cul nanosheets show a relatively low resistivity about 0.28 Ω cm for the 230 nm thick nanosheets and 0.57 Ω cm for the 43 nm nanosheets. The low resistivity of the γ -Cul nanosheets may be largely attributed to copper vacancy, acting as an electron acceptor that creates holes in the valence band.⁵¹ Because γ -Cul is reported to be a p-type semiconductor, WSe₂ is an p-type semiconductor, and WS₂ is an n-type semiconductor, ^{47,51,52} the γ -

Cul/WSe₂ vertical heterostructure may be formed a p–p junction whereas the γ -Cul/WS₂ vertical heterostructure is expected to behavior as a p–n diode. Figure S4 shows the band alignment and band diagram of Cul, WS₂, and WSe₂ across the heterostructure. At zero bias, the conduction band minimum and valence band maximum of Cul are located at a higher energy than that of WS₂, forming a type-II heterostructure. Thus, the electrons on the conduction band transfer from Cul to WS₂, while the holes on the valence band transfer from WS₂ to Cul resulting in the efficient charge separation. Indeed, the current–voltage characteristics of a typical γ -Cul/WS₂ vertical heterostructure show obvious current rectification behavior with a rectification ratio reaching up to 338 at the bias voltage of ±1.5 V (Fig. 5c), consistent with the expected p-n diode behavior. The inset shows the optical image of a typical device where the Au electrodes are separately made on γ -Cul and



Fig. 5 Electrical characterization of the γ -Cul nanosheets and γ -Cul/WS₂ vertical heterostructure. **a** Current–voltage characteristic of the γ -Cul nanosheet with the thickness of 230 nm. **b** Current-voltage characteristic of the γ -Cul nanosheet with the thickness of 43 nm. **c** Current–voltage characteristic of the γ -Cul/WS₂ vertical heterostructure p-n diode. The inset shows the OM image of the corresponding devices. Scale bars, 5 μ m

 $\mathsf{WS}_2.$ We have further deduced the ideality factor using Lamber function: 53

$$I(V) = \frac{nV_{\rm T}}{R_{\rm S}} W_{\rm o} \left(\frac{I_{\rm S}R_{\rm S}}{nV_{\rm T}} e^{(V + I_{\rm S}R_{\rm S})/nV_{\rm T}} \right) - I_{\rm S}, \tag{1}$$

where $I_{\rm S}$ is the saturation current, $V_{\rm T}$ is the thermal voltage (≈ 0.026 V at T = 300 K), n is the ideality factor, and $R_{\rm S}$ is the series resistance. An ideality factor n = 2.72 is extracted, which compares well with many other 2D diodes.^{54,55}

In summary, we have reported the successful synthesis of ultrathin nanosheets from non-layered y-Cul and 2D-y-Cul/WSe2, 2D-y-Cul/WS₂ vdWHs on SiO₂/Si substrate using a facile PVD method. The obtained y-Cul nanosheets exhibit triangle shapes and hexagonal shapes with lateral size up to several micrometers and thickness down to 1 nm. The Raman spectra studies show characteristic peak at around 122 cm⁻¹, in accordance with the bulk y-Cul. XRD study suggests the obtained nanosheets are pure cubic phase of γ -Cul crystal with the (111) directions normal to the SiO₂/Si substrate plane. TEM and SAED studies demonstrate the high crystallinity and single crystal nature of the as-synthesized y-Cul nanosheets. The y-Cul/WSe2 and y-Cul/WS2 vertical heterostructures have been prepared by a two-step PVD progress. The electrical transport studies show that y-Cul nanosheets have low revisitivity ~0.3 Ω cm and γ -Cul/WS₂ vertical heterstructure show expected p-n diode behavior. The successful synthesis of 2D nanosheets from non-layered y-Cul and van der Waals heterostructures opens a new path way to explore ultrathin 2D structure from other non-layered materials and could enable new opportunities for electronic and optoelectronic devices.

METHODS

Synthesis of single crystal y-Cul nanosheets

The γ -Cul nanosheets were grown on 300 nm SiO₂/Si substrate using a home-built PVD system. Figure S1 shows the illustration of the PVD setup for γ -Cul growth. Briefly, 0.5 g Cul powder (Alfa, 99.5%) in a small ceramic boat was placed in the center of a horizontal tube furnace with 1-inch diameter quartz tube. The 300 nm SiO₂/Si substrate (~1 cm × 3 cm) was placed in another boat downstream about 10 cm away from the center of the quartz tube. Prior to the growth, the tube was purged with ~400 sccm Ar carrier gas (Rizhen, ~99.999%) flow for 10 min at room temperature to remove residual oxygen in the reactor. The furnace was then ramped to the desired growth temperature 360–410 °C, and kept for 10–40 min under ambient pressure with an argon gas flow of 100–225 sccm. Note that the temperature of the growth substrate at the downstream end is 193–234 °C when the center heating zone is set at 360–410 °C. The growth was terminated by shutting off the power of the furnace, and the system was naturally cooled down to room temperature.

Synthesis of WSe₂ and WS₂ nanosheets

 WSe_2 nanosheets: A ceramic boat loaded with WSe_2 powder (Sigma Aldrich, 99.9%) was placed in the center of the furnace. A 300 nm SiO₂/Si substrate was placed face-up on the other boat downstream. Before heating, high-purity argon gas was continuously introduced into the

quartz tube. The furnace was then heated to 1200 °C and a constant flow of 100 sccm Ar was used as carrier gas. Note that the temperature of the growth substrate at the downstream end is 782 °C when the center heating zone is set at 1200 °C. After stabilizing for 8 min, the growth was terminated by shutting off the power of the furnace, and the system was naturally cooled down to room temperature. WS₂ nanosheets: Similar to the growth of WS₂ nanosheets, a ceramic boat loaded with WS₂ powder (Sigma Aldrich, 99.9%) was placed into a ceramic boat at the center of the furnace. A 300 nm SiO₂/Si substrate was placed at the end of the furnace. During the growth of WSe2 nanosheets, the furnace temperature was heated to 1180 °C in an argon environment and maintained 8 min with a 50 sccm of Ar as carrier gas. Note that the temperature of the substrate at the down stream end is about 754 °C when the center heating zone is set at 1180 °C. In the end, the growth was terminated by shutting off the power of the furnace, and the system was naturally reduced to room temperature.

Synthesis of 2D $\gamma\text{-Cul/WS}_2$ and $\gamma\text{-Cul/WS}_2$ van der Waals heterostructures

The 2D γ -Cul/WSe₂ (γ -Cul/WS₂) vertical heterostructure was grown in an atmospheric pressure PVD system using Cul powder as the source. A ceramic boat loaded with Cul powder was placed into the heating zone of a quartz tube and the SiO₂/Si substrate with as grown WSe₂ (WS₂) nanosheets was placed at the downstream end of the tube furnace. The system was purged with high-purity Ar gas (Rizhen, ~99.999%) before ramping to the desired growth temperature for the growth of γ -Cul nanoplates at 450 °C, and maintained for 5 min with a 150 sccm of Ar as carrier gas. Note that the temperature of the substrate at the downstream end is at 262 °C when the center heating zone is set at 450 °C. Finally, the growth was terminated by shutting off the power of the furnace, and the system was naturally cooled down to room temperature.

Sample characterizations

The γ -Cul nanosheets grown on the SiO₂/Si substrate were first characterized using an optical microscope (DP27, OLYMPUS). The thickness of γ -Cul nanosheets was determined using an atomic force microscope (Bioscope system, BRUCKER). The crystal structure of the sample was studied using a X-ray diffraction spectrometer (XRD-6100, SHIMADZU). The morphology and element composition of the obtained γ -Cul nanosheets were characterized using a SEM microscope (Σ IGMA HD, ZEISS) equipped with an EDX system. TEM characterization was performed using a TEM microscope (JEM-2100F, JEOL), operating at 200 kV. Raman spectra from the γ -Cul nanosheets and the γ -Cul/WSe₂ (WS₂) heterostructures were collected using a confocal microscopic spectrometer (invia-reflex, Renishaw) with 532 nm laser as the excitation source.

Device fabrication and characterization

The γ -Cul nanosheet devices and p–n junction devices were fabricated using e-beam lithography and thermal evaporation of 150 nm Au electrode. The electrical measurements were done at room temperature under dark condition.

Data availability

The data related to the findings of this work are available from the corresponding author on reasonable request.

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AUTHOR CONTRIBUTIONS

X.D.D., J.H., and X.D. designed the research. K.Y. performed the synthesis and characterization of the samples. P.C. fabricated and measured the device. Z.Z. prepared the WS_2 and WSe_2 sample. All authors contributed to experiment research and commented on the manuscript.

ADDITIONAL INFORMATION

Supplementary information accompanies the paper on the *npj 2D Materials and Applications* website (https://doi.org/10.1038/s41699-018-0058-2).

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