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Large-band-gap non-Dirac quantum spin Hall states and strong Rashba effect in functionalized thallene films

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The quantum spin Hall state materials have recently attracted much attention owing to their potential applications in the design of spintronic devices. Based on density functional theory calculations and crystal field theory, we study electronic structures and topological properties of functionalized thallene films. Two different hydrogenation styles (Tl_2H and Tl_2H_2) are considered, which can drastically vary the electronic and topological behaviors of the thallene. Due to the C_{3v} symmetry of the two systems, the p_x and p_y orbitals at the Γ point have the non-Dirac band degeneracy. With spin-orbit coupling (SOC), topological nontrivial band gaps can be generated, giving rise to non-Dirac quantum spin Hall states in the two thallium hydride films. The nontrivial band gap for the monolayer Tl_2H is very large (855 meV) due to the large on-site SOC of $\text{Tl } p_x$ and p_y orbitals. The band gap in Tl_2H_2 is, however, small due to the band inversion between the $\text{Tl } p_{xy}$ and p_z orbitals. It is worth noting that both the Tl_2H and Tl_2H_2 monolayers exhibit strong Rashba spin splitting effects, especially for the monolayer Tl_2H_2 ($\alpha_r = 2.52 \text{ eV\AA}$), rationalized well by the breaking of the structural inversion symmetry. The Rashba effect can be tuned sensitively by applying biaxial strain and external electric fields. Our findings provide an ideal platform for fabricating room-temperature spintronic and topological electronic devices.

Since Kane and Mele first proposed the quantum spin Hall (QSH) effect in graphene^{1,2}, QSH insulators, also known as one type of two-dimensional (2D) topological insulators (TIs), have attracted great attention in condensed matter physics and material science due to their wide potential applications in spintronics and topological quantum computation^{3–5}. QSH insulators are characterized by an insulating band gap in bulk and non-dissipative, fully spin-polarized gapless helical edge states at the sample boundary, which are protected by time inversion symmetry. As the first predicted 2D TI, graphene¹ opens a topologically nontrivial band gap at the Dirac point, with the consideration of the spin-orbit coupling (SOC) effect. However, besides the factor of the light carbon element, the horizontal mirror symmetry in the graphene structure inhibits the first-order SOC effect between the nearest neighbor carbon atoms. Therefore, the weak second-order SOC makes the QSH effect only appear at an unrealistically low temperature⁶. Till now, the experimental observations of quantized Hall conductance through the QSH effect are only reported in few systems including Bismuthene⁷, WTe_2 ⁸, and HgTe/CdTe ^{9,10} and InAs/GaSb ^{11,12} quantum-wells with an ultralow temperature ($< 10 \text{ K}$), which limits their tempting applications in room temperature spintronic devices.

To achieve QSH effects at room temperature, 2D materials with strong SOC interactions and large topologically nontrivial band gaps are very desired. Some graphene-like 2D layered materials have been proposed to own QSH effects with relatively large band gaps, including group-IVA monolayers for the low-pucker silicene^{13,14}, germanene^{13,15}, stanene¹⁶, and group-VA bismuthene⁷. Based on the existing 2D materials, chemical functionalization has been found being a valid tactic to design the QSH effects with pretty large band gaps^{14,16,17}. These researches indicate that two types of schemes may be employed to produce the QSH states with large band gaps. One is designing the materials containing heavy elements which can bring strong SOC effects. Another approach is through chemical functionalization. Their combination should be a more effective tactic.

Besides the groups IVA Pb and VA Bi, group IIIA TI can also be regarded as a heavy element, whose graphene-like monolayer structure is called thallene^{18,19}. The pristine 2D honeycomb-like thallene is a topological trivial

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semiconductor. The phase transition from a semiconductor to a QSH insulator can be achieved under the application of large biaxial strain and its nontrivial topology comes from a band inversion between p_{xy} and p_z orbitals induced by SOC¹⁸. The thallene monolayer has been successfully prepared in experiments by cooling the 2/3 monolayer of mobile Tl atoms on a single-layer NiSi₂ atop a Si(111) substrate below ~ 150 K¹⁹. Compared with the theoretically assumed free-standing thallene, the thallene structure frozen on the NiSi₂/Si(111) substrate in the experiments undergoes strong tensile strain ($\sim 27\%$) and then enters the topological phase¹⁹. Very recently, high-quality large-scale thallene monolayers with exotic electron bands demonstrating colossal spin-polarization have been fabricated through the decoration of thallene/NiSi₂ interface by Sn interlayers²⁰. Thus, Tl monolayers actually become a new heavy-element material platform, which may be designed to explore the interesting 2D topological electronic states etc.

In this work, based on first-principles calculations, two different configurations of hydrogenated thallene Tl₂H and Tl₂H₂ are built. The electronic structures and topological properties for the two monolayers are systematically studied. Due to the C_{3v} symmetry of the structures, the Tl p_x and p_y orbitals of the Tl₂H and Tl₂H₂ monolayers form quadratic non-Dirac bands. A topologically nontrivial band gap (up to 855 meV) can be generated by the SOC interaction, giving rise to a non-Dirac quantum spin Hall state in the materials. The variation between the electronic states of the Tl₂H and Tl₂H₂ monolayers can be comprehended through the crystal field splitting and band inversion. Particularly, due to the absence of the structural inversion symmetry, the Tl₂H and Tl₂H₂ monolayers both exhibit marked Rashba spin splitting effect, especially in the monolayer Tl₂H₂. The Rashba splitting can be tuned sensitively by applying biaxial strain and external electric fields, beneficial to spintronic applications.

Results and discussion

Crystal structures and stability

Different from other heavy element monolayers, the thallene monolayer is a fully flat 2D honeycomb lattice without buckling^{18,19}. In other words, the Tl atoms in thallene are aligned in the same plane, with two Tl atoms in each unit cell. The structure has a space group symmetry of $P6/mmm$ (No. 191) with a space inversion symmetry. On this basis, we construct two types of functionalized thallene structures to explore the unique electronic structures: unilateral semi-saturation and unilateral full saturation with hydrogen atoms, which were named as Tl₂H and Tl₂H₂, respectively, as shown in Fig. 1a,b. To simulate the experimental configuration of thallene grown on a substrate¹⁹, merely the unilateral saturation is considered. Due to the unilateral hydrogenation, structural inversion symmetry of the two materials is broken and the space groups of Tl₂H and Tl₂H₂ are $P3m1$ (No.156) and $P6mm$ (No.183), respectively. After hydrogenation, the Tl atoms still own a honeycomb lattice (see the top views of Fig. 1a,b) for Tl₂H and Tl₂H₂, and the Tl atomic layer of monolayer Tl₂H is not a completely planar structure, but a low-buckled structure with a vertical height h between the two Tl atomic layers (see the side view of Fig. 1a). In Tl₂H₂, the Tl atomic layer still maintains a planar structure (see the side view of Fig. 1b). The buckled structural characteristic in Tl₂H is attributed to the different chemical environments around the

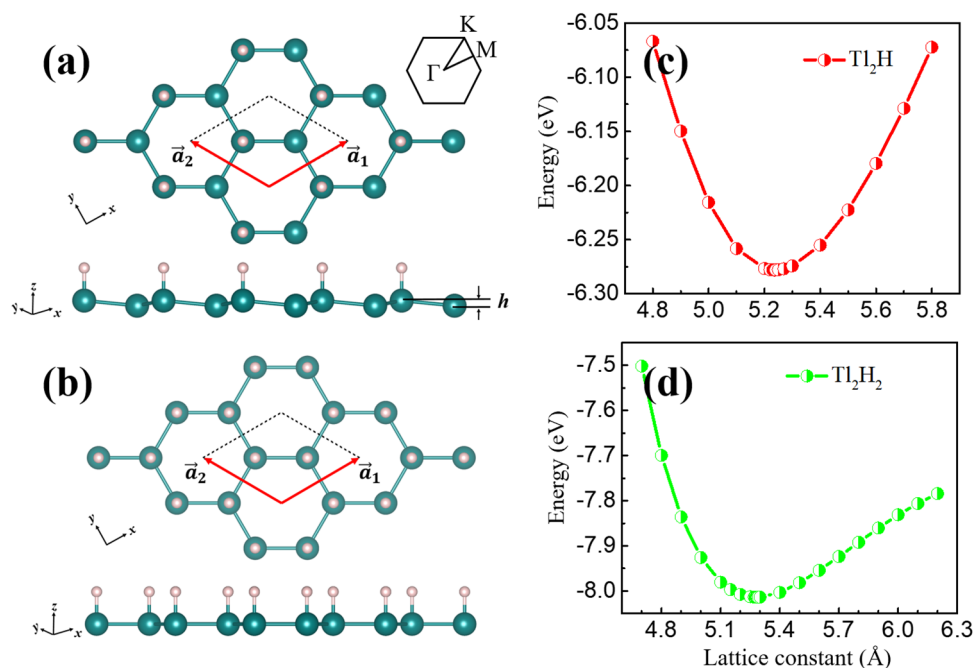


Figure 1. (Color online) The geometry structures of the monolayer (a) Tl₂H and (b) Tl₂H₂ from the top and side views. The buckled height between the Tl atoms is h . The inset shows the first BZ with the high symmetry points. The green and white balls represent Tl and H atoms, respectively. In (a) and (b), the corresponding xyz axis are shown. (c, d) The total energies as a function of the lattice constants for the monolayer Tl₂H and Tl₂H₂, respectively.

two types of Tl atoms located in different positions of the honeycomb lattice. In the unit cell of Tl_2H (Fig. 1a), the left Tl atom is not only bonded to three in-plane neighbor Tl atoms, but also saturated by one H atom. The interaction between Tl and H leads to the Tl atom (bonded to the H atom) approaching the H atom. Thus, a low-buckling structure appears for the Tl atoms in Tl_2H (Fig. 1a). Dissimilarly, in Tl_2H_2 , both of the Tl atoms in the unit cell bond with three neighbor Tl atoms and one H atom, resulting in a planar structure of the Tl atoms in the monolayer (Fig. 1b).

The optimized lattice constants of the Tl_2H and Tl_2H_2 are $a = 5.24 \text{ \AA}$ and 5.28 \AA (Fig. 1c,d), respectively, which together with other structural parameters, the binding energies, and the formation energies, are listed in Table 1. The positions of the atoms in the unit cells for the two monolayers are given in the Supplementary Material as Table S1. Compared to the Tl–Tl bond length ($d_{\text{Tl-Tl}}$) in pristine thallene (3.01 \AA ¹⁸ and 3.03 \AA ¹⁹), the Tl–Tl bond lengths in Tl_2H (3.04 \AA) and Tl_2H_2 (3.05 \AA) are increased slightly by the hydrogenation. The trend is reasonable because the repulsive force between the H–H and Tl–H ions appears in the monolayer with H adsorbed. And it becomes larger with the increase of the H concentration. The binding energies and formation energies of the two monolayers are also calculated to investigate their structural stability. The binding energies are calculated by $E_{b/\text{Tl}_2\text{H}} = (2E_{\text{Tl}} + E_{\text{H}}) - E_{\text{Tl}_2\text{H}}$ and $E_{b/\text{Tl}_2\text{H}_2} = 2(E_{\text{Tl}} + E_{\text{H}}) - E_{\text{Tl}_2\text{H}_2}$, where $E_{\text{Tl}_2\text{H}}/E_{\text{Tl}_2\text{H}_2}$ is the total energy per unit cell of the $\text{Tl}_2\text{H}/\text{Tl}_2\text{H}_2$ monolayer and $E_{\text{Tl}}/E_{\text{H}}$ is the total energy of one Tl/H atom. The positive and relatively large values ($> 4.5 \text{ eV}$) of the obtained E_b for Tl_2H and Tl_2H_2 indicate very strong bonding between the atoms. The formation energies for the two monolayers are calculated with $E_{f/\text{Tl}_2\text{H}} = E_{\text{Tl}_2\text{H}} - E_{\text{Tl}} - 1/2E_{\text{H}_2}$ and $E_{f/\text{Tl}_2\text{H}_2} = E_{\text{Tl}_2\text{H}_2} - E_{\text{Tl}} - E_{\text{H}_2}$, where E_{Tl} is the total energy per unit cell of the pristine thallene monolayer and E_{H_2} is the total energy of one H_2 molecule. The obtained negative formation energies for the monolayers (Table 1) show exothermic reactions from the thallene monolayer and H_2 molecules, guaranteeing the feasibility of the experimental synthesis for these functionalized materials.

To explore the dynamical stability of the two monolayers, we calculate the phonon spectra for them. The results are displayed in Fig. S1(a) and (b). There are some negative frequencies in phonon spectra for the both Tl_2H and Tl_2H_2 monolayers. The results are not surprising because the phonon spectrum of the pristine thallene has also negative frequencies, as reported in Ref. 18. To eliminate the negative frequencies, the two monolayers are deposited to SiC substrates. The geometries of the $\text{Tl}_2\text{H}_2/\text{SiC}$ heterostructure are displayed in Fig. S2 (a) and (b). The substrates are helpful to stabilize the dynamical stability for the two materials. There are, however, still some negative frequencies for the $\text{Tl}_2\text{H}/\text{SiC}$ heterostructure, with the lowest value of -0.7 THz (larger than that of the material without the substrate, -2.0 THz). All the soft modes in the Tl_2H_2 monolayer are removed by the SiC substrate, as indicated in Fig. S2(c). Thus, to experimentally observe the interesting electronic states and topological behaviors in the materials, the two monolayers should be placed on the SiC substrates. There are rich examples that the materials are not completely dynamically stable (with some soft vibration modes), which could be, however, fabricated successfully in experiments, for such as stanene^{21,22} and plumbene^{23,24}.

Band structures and strain tuning

Figure 2 shows band structures of the hydrogenated Tl_2H and Tl_2H_2 monolayers without the consideration of SOC. For convenient comparison, the bands of the pristine 2D thallene are also displayed and discussed first. Similar to the case of the low-buckled plumbene¹⁷, a linear Dirac cone composed of $\text{Tl } p_z$ orbitals is found around the K point (at about 1.4 eV in Fig. 2a,d). Due to the C_{3v} symmetry of the honeycomb structure, another set of

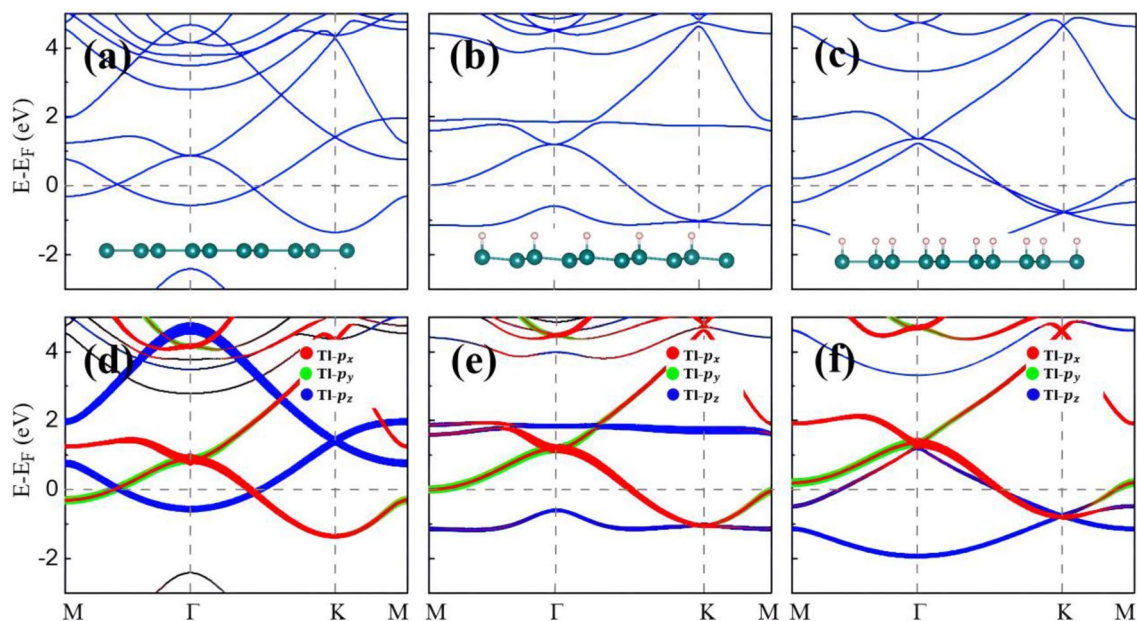


Figure 2. Band structures for (a) Tl, (b) Tl_2H , and (c) Tl_2H_2 monolayers without SOC. (d–f) are orbital projections corresponding to (a–c) band structures, respectively. The insets in (a–c) are the side views of the corresponding structures.

twofold degenerate bands appear at the Γ point around 1.0 eV in the planar thallene, composed of Tl p_x and p_y orbitals (Fig. 2a,d). The dispersion of these p_x and p_y orbitals belongs to quadratic non-Dirac bands since the Γ point is a time-reversal invariant point and the linear k term in the energy eigenvalue at such point is forbidden²⁵. This band feature with two twofold degenerate points respectively at Γ and K also presents in Pb monolayers¹⁷, which happen at the Fermi level (E_F). They are, however, located above the E_F in the Tl monolayer due to the less valence electron number of a Tl atom, compared to a Pb atom. In Tl₂H, since partial Tl p_z orbitals are saturated by hydrogen atoms, the linear Dirac degenerate bands around the K point are opened and become two relatively flat bands (the blue curves in Fig. 2e). The tendency is the same as those in half-hydrogenated Bi honeycomb monolayers in nonmagnetic states²⁶. The flat bands of the Tl p_z orbitals in Tl₂H are, however, not located around the E_F . Thus, spontaneous spin polarization does not happen in Tl₂H, different from many other half-hydrogenated Ge²⁷, Sn²⁷, and Bi²⁶ monolayers etc. The quadratic non-Dirac bands (composed of Tl p_x and p_y orbitals) at the Γ point still exist because of the C_{3v} symmetry unbroken by the hydrogenation. For the electronic structure of Tl₂H₂ (Fig. 2c,f), it can be intuitively understood that the dangling bonds of Tl p_z are all saturated by H atoms now, which makes the upper flat p_z orbitals move down in energy.

The orbital-resolved band structures with SOC interactions for the Tl₂H and Tl₂H₂ monolayers are shown in Fig. 3. As displayed in Fig. 3a,b, the bands around the E_F in Tl₂H are primarily made up of the Tl p orbitals (and also some s orbitals below the E_F). When the SOC interaction is taken into account, the twofold degeneracy at the Γ point is lifted and a direct band gap $\Delta E_d = 1.19$ eV is opened at the Γ point (Fig. 3(b)). This large band gap can be ascribed to the strong on-site SOC interaction contributed by Tl p_x and p_y orbitals²⁸. Due to the band dispersion, a relatively small global band gap $\Delta E_g = 854$ meV is obtained for Tl₂H, which is pretty large and close to the record value of QSH effects reported in the F-decorated Bi monolayer²⁹. For Tl₂H₂, the quadratic non-Dirac degenerate point (red and green colors) is located at 1.3 eV at the Γ point (Fig. 3c), below which there is an energy band primarily composed of the Tl p_z orbitals (blue color). Under the influence of SOC, not only the quadratic non-Dirac point is broken with a band gap opened, but also the band inversion between Tl $p_{x/y}$ and Tl p_z orbitals is induced in Tl₂H₂. Due to the large on-site SOC, the SOC-induced direct band gap of the Tl $p_{x/y}$ is still very large ($\Delta E_d = 1.32$ eV, as marked in Fig. 3d), within which there is the band of the Tl p_z orbitals hybridizing with Tl $p_{x/y}$ orbitals. Thus, the band inversion between Tl $p_{x/y}$ and Tl p_z orbitals causes the global band gap (334 meV) for the monolayer Tl₂H₂ much less than that (854 meV) of Tl₂H. These SOC-induced global band gaps in the two monolayers are expected to be topologically nontrivial³⁰, to be discussed below. Compared to the pristine thallene with a topologically trivial band gap (98 meV)¹⁸, the hydrogenation causes the thallene becoming a metal. Especially, for the full passivated thallene, more bands cross the E_F and the Tl₂H₂ monolayer turns to be a typical metal. This trend is opposite to the full-passivation effect in group IVA monolayer. For example, the full-hydrogenation makes graphene change from a gapless semiconductor to a semiconductor with a large band gap of 3.49 eV³¹ due to the Dirac cone composed of C p_z orbitals located right at E_F in the pristine graphene.

Since the distributions of the featured bands for the Tl₂H and Tl₂H₂ monolayers are different, the mechanisms of the band evolution for the two materials are now analyzed. For Tl₂H and Tl₂H₂ monolayers, the evolutions of the atomic orbitals around the Γ point are both mainly triggered by two aspects: one is chemical bonding and the other is SOC. The difference is that the two Tl p_z bands in the monolayer Tl₂H are far away from the concerned energy degenerate point at Γ (Fig. 2e). Since we mainly focus on the bands near the two-fold degenerate point, only the Tl $p_{x/y}$ orbitals are discussed in Tl₂H. As shown in Fig. 4a, in the absence of SOC, the chemical bonding between Tl-Tl atoms forms bonding and anti-bonding states for the $p_{x/y}$ orbitals. These states are labeled as $|p_{x/y}^{\pm}\rangle$, where the superscript (+/−) denotes the bonding/anti-bonding states and the concerned degenerate energy level

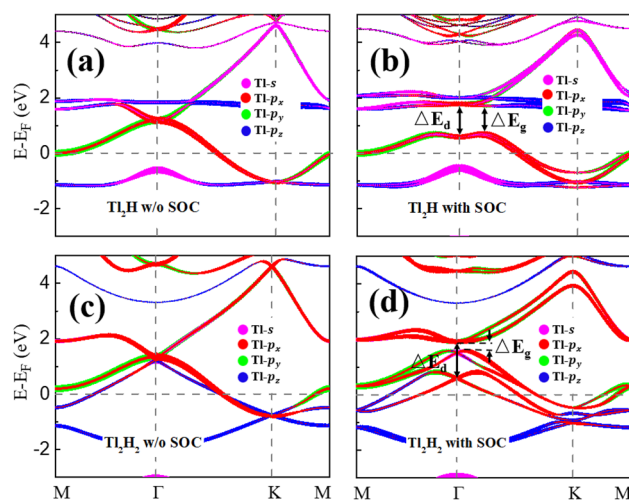


Figure 3. (a) and (b) give the orbital-projected band structures of the Tl₂H without and with SOC, respectively. (c) and (d) give the orbital-projected band structures of the Tl₂H₂ without and with SOC, respectively. The dot size is proportional to the contribution of the corresponding orbitals. In (b) and (d), ΔE_d represents a direct band gap at the Γ point and ΔE_g represents the global band gap.

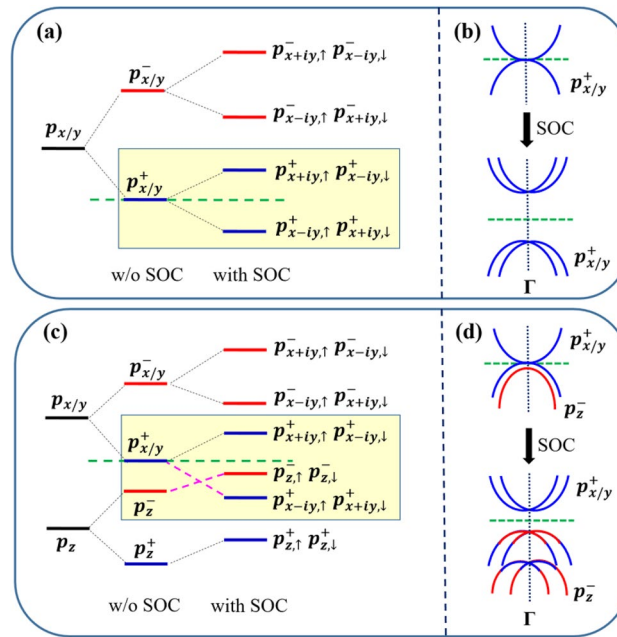


Figure 4. (Color online) (a) and (c) correspond to the TI-orbital evolution diagrams at Γ point near the E_F in the monolayer Tl_2H and Tl_2H_2 , respectively. The signs of '+' and '-' indicate the bonding and anti-bonding states, respectively. The green dotted lines represent the location of the concerned energy degenerate points. (b) and (d) show the schematic band dispersion of the energy regions marked by the rectangles in (a) and (c), respectively. In (b) and (d), the blue curves represent the bonding states of the $p_{x/y}$ orbitals and the red curves represent the anti-bonding states of the p_z orbital.

is mainly contributed by the $p_{x/y}^+$ orbitals. When SOC is turned on, $|p_{x/y}^+ \rangle$ splits into two energy levels. Analogous to the cases of stanene¹⁵ and germanene films¹⁶, the two SOC-induced splitting energy levels can be expressed as $|p_{x+iy,\uparrow}^+, p_{x-iy,\downarrow}^+ \rangle$ and $|p_{x-iy,\uparrow}^+, p_{x+iy,\downarrow}^+ \rangle$ (Fig. 4a), with $|p_{x+iy,\uparrow}^+, p_{x-iy,\downarrow}^+ \rangle$ moving up and the other set of states moving down. Thus, a band gap is opened at the Γ point. Note that the role of SOC here is only to lift the degeneracy or open a band gap. It does not induce band inversion, as displayed in Fig. 4b. The status in Tl_2H_2 is, however, dissimilar to that of Tl_2H . The effect of the Tl p_z orbital cannot be ignored in Tl_2H_2 due to some of its locations close to the two-fold degenerate point at the Γ point (Fig. 2f). As illustrated in Fig. 4c, the chemical bonding between Tl-Tl atoms makes all atomic orbitals split into bonding and anti-bonding states, labeled as $|p_{x/y}^\pm \rangle$ and $|p_z^\pm \rangle$, respectively. Owing to the increase of the H concentration in Tl_2H_2 , more $|p_z^- \rangle$ states are passivated. Hence, the $|p_z^- \rangle$ state moves down in energy and is lower than the two-fold degenerate $|p_{x/y}^+ \rangle$ states (Fig. 4c). When the SOC is taken into account, $|p_{x/y}^+ \rangle$ splitting is similar to that of Tl_2H , where $|p_{x+iy,\uparrow}^+, p_{x-iy,\downarrow}^+ \rangle$ moves down and the anti-bonding state of the p_z orbitals (labeled as $|p_{z,\uparrow}^-, p_{z,\downarrow}^- \rangle$) moves up. The large SOC splitting of the $|p_{x/y}^+ \rangle$ states leads to the band inversion between $p_{x/y}$ and p_z states (Fig. 4d), resulting in a decrease of the band gap in Tl_2H_2 (Fig. 4d), compared to that of Tl_2H . To confirm the results, the electronic structures of the two monolayers are also calculated with the hybrid density functional (HSE06). Similar band structures with the indirect band gaps of 924 and 233 meV are obtained for Tl_2H and Tl_2H_2 , respectively (Fig. S3).

Applying strain is generally an effective strategy for making controls to the structures and electronic states of the monolayer materials. Here, we apply biaxial strain in the range of -5 to 5% to the two monolayers. To check the structural stability of the monolayers under strain, the binding energies of the two structures under strain are calculated. As illustrated in Fig. S4, the strain does not decrease much the binding energies, indicating the structural stability of the monolayers under strain. The trend is consistent with the results reported in Ref.³² that thallene can undergo very large tensile strain ($\sim 27\%$) in experiments. The band gaps opened at the two-fold degenerate point of the non-Dirac bands as a function of the biaxial strain for the monolayers are shown in Fig. 5a. The strain is defined as $(a'-a) \times 100\% / a$, where a' stands for the lattice constant with strain applied and a stands for the equilibrium lattice constant. Due to the different orbital components near the concerned band gaps for the Tl_2H and Tl_2H_2 , the responses to the external tuning are various.

The global band gaps for Tl_2H exhibit roughly linear and saturated trends with respect to the compressive and tensile strain, respectively (Fig. 5a), comprehended as following. As indicated in Fig. S5, with the strain varying from $-3 \sim 3\%$, the $p_{x/y}$ states of Tl_2H move down in energy while the p_z states (at about 1.6 eV) move up. And a band inversion occurs with $p_{x/y}$ and p_z orbitals around the Γ point at -1% strain (Fig. S5). During this process, the buckled height (h) decreases (Fig. 5b), which can be straightforward ascribed to the increase of the in-plane lattice constants. The h decrease makes the p_z orbitals of the Tl atoms without H atoms more isolated. The interaction between this Tl atom and its three neighbor Tl atoms more tends to form planar sp^2 hybridization instead

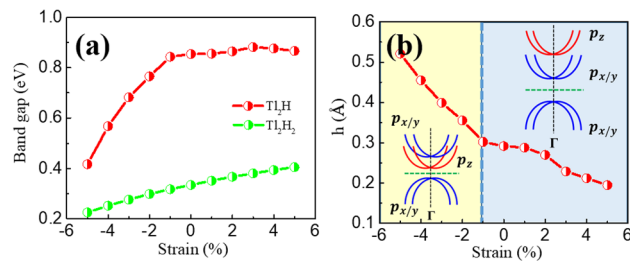


Figure 5. (Color online) (a) The global band gaps as a function of the strain for the monolayer Tl₂H and Tl₂H₂. The red and green colors give the Tl₂H and Tl₂H₂ results, respectively. (b) The buckled heights (*h*) between Tl atoms in the monolayer Tl₂H as a function of the strain. The inset shows the band evolution, in which the blue/red colors express the $p_{x/y}$ / p_z bands.

of three dimensional sp^3 hybridization, giving rise to the Tl p_z orbitals moving up in energy during the process (especially around Γ point). This tendency causes the band inversion between the $p_{x/y}$ and p_z orbitals at -1% strain, as illustrated in the inset of Fig. 5b. Since the SOC split of $p_{x/y}$ orbitals does not change much with the strain (see the black arrows in Fig. S5), the band gap hardly changes with the increase of the tensile strain (Fig. 5a).

Different from the case of Tl₂H, the band gap of Tl₂H₂ increases linearly with the biaxial strain varying from -5 to 5% (Fig. 5a). The overall tendency is, however, relatively weak. The increase of the band gap for Tl₂H₂ can be primarily attributed to two factors. One is that the band dispersion tends to weaken with the increase of the atomic distance. The other is that the on-site SOC of $p_{x/y}$ orbitals is enhanced slightly by the gradual decrease of the p_z state involved in the concerned bands with the strain varying from -5 to 5% (Fig. S6).

Strong Rashba effect

Considerable Rashba SOC interactions³² exist in the monolayer Tl₂H and Tl₂H₂ due to the breaking of vertical reflection symmetry. The strength of the Rashba SOC generally can be manipulated markedly by an external electric field. And the Rashba SOC effect has been employed to fabricate all-electric spintronic devices, such as spin field effect transistors and spin valves (without magnetic fields needed)^{33–35}, promoting the spintronic applications. Very large Rashba spin splitting has been reported for the deep-energy π states (~ 8 eV below E_F) of graphene deposited on metallic substrates³⁶. Here, we merely discuss the Rashba effect of Tl₂H₂ since it is stronger than that of Tl₂H (Fig. 3b,d). The orange bands of Tl₂H₂ in Fig. 6 are focused on. The magnified bands around the Γ point (Fig. 6b) exhibit an obvious Rashba splitting phenomenon similar to that of semiconductor quantum wells and heavy metal surfaces^{37–41}. To examine the Rashba effect, the spin textures in the k_x - k_y plane for Tl₂H₂ are calculated. The spin-projected constant-energy contour plots for the spin textures calculated in the $k_z = 0$ plane are shown in Fig. 7a. The flower-like spin textures appear in Tl₂H₂. For both S_x and S_y spin components, the pair of spin-splitting bands have the same spin orientation. However, for pure 2D Rashba spin splitting, the pair of spin-splitting bands for the both S_x and S_y spin components generally have opposite spin orientations⁴². For Tl₂H₂, the SOC effect not only opens the nontrivial band gap, but also induces the band inversion near E_F (Fig. 4c,d), resulting in an anomalous spin texture. To verify this point, the p_z state is moved down in energy to the position far from the concerned energy point by applying 20% biaxial tensile strain. In this case, the SOC is not sufficient to reverse the bands and the corresponding spin structures of the Tl₂H₂ monolayer are displayed in Fig. 7b. Clearly, the pair of spin-splitting bands for both S_x and S_y spin components have opposite spin orientations. Due to the disappearance of the S_z component, the spin moments of the two rings shown in Fig. 7b have opposite chirality. The large ring is anticlockwise, while the small ring is clockwise. This large Rashba effect has recently been observed in experiments in the thallene²⁰ with a different forming mechanism.

The Rashba coefficient α_R is also calculated to describe the strength of the Rashba SOC effect. The Rashba coefficient can be obtained from the formula $\alpha_R = 2E_R/K_R$ ⁴¹, where E_R and K_R are defined in Fig. 6b. From Fig. 6b, the E_R , K_R , and α_R are estimated to be 68 meV, 0.054 Å⁻¹, and 2.52 eVÅ, respectively. The obtained α_R

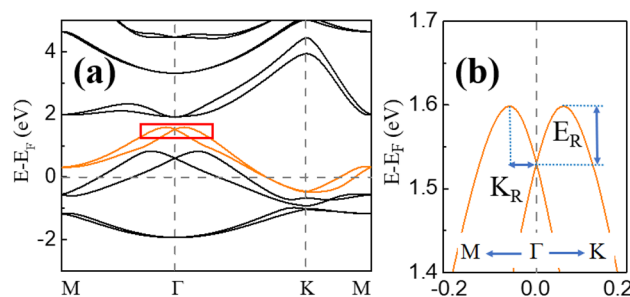


Figure 6. (a) Band structure of the Tl₂H₂ monolayer with SOC. (b) The magnified view of the band structure (in the red rectangle in (a)) of the Tl₂H₂ monolayer. The orange color indicates the concerned bands.

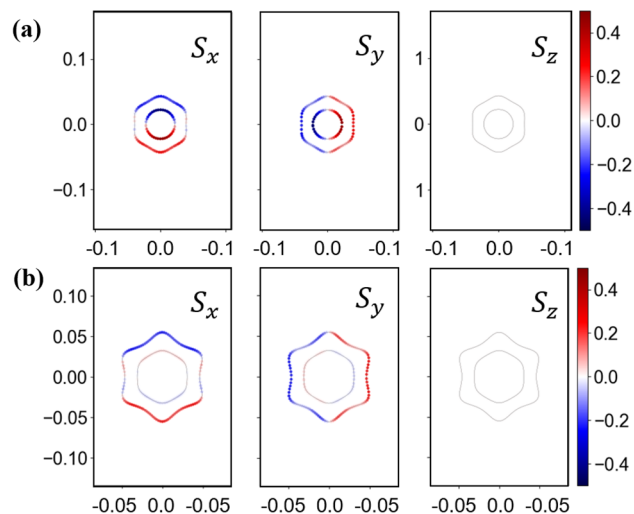


Figure 7. (Color online) Spin textures centered at the Γ point calculated in the $k_z = 0$ plane for the Tl_2H_2 monolayer with (a) the pristine structure and (b) 20% tensile strain, respectively. The red and blue colors show spin-up and spin-down states, respectively. The energy in (a) is set at 1.1 eV.

is significantly higher than those of many heterostructures or surface states of 2D TIs, such as InGaAs/InAlAs ($0.07 \text{ eV}\text{\AA}$)³⁴, Au/W(110) ($0.16 \text{ eV}\text{\AA}$)³⁵, and CSb_3 ($0.83 \text{ eV}\text{\AA}$)⁴³. In Fig. 8, we show the Rashba energy E_R , the momentum offset K_R , and Rashba parameter α_R under different biaxial strain (-5 to 5%) and external electric fields (-0.5 to 0.5 V/\AA). With the increase of the biaxial strain and external electric field, the values of E_R and K_R increase almost linearly and are very sensitive to the biaxial strain. On the contrary, the values of α_R shows a decreasing trend with the increase of the biaxial strain and external electric field, indicating that the application of compressive strain and electric fields along $-z$ axis are more conducive to the Rashba effect of Tl_2H_2 . Under compressive strain, the Tl p_z component of the concerned bands becomes more (Fig. S6), which together with the slight increase of Tl-H bond strength (the bond length becomes short) and the Tl $p_{x/y}$ asymmetric distribution about the Tl-Tl plane results in the enhancement of the Rashba effect with the compressive strain in Tl_2H_2 . As shown in Fig. 1b, the hydrogen atoms are located above the Tl atoms. When a negative electric field (along $-z$ axis) is applied, the electrons of thallene tend to move toward the H atoms. And the charge densities between the Tl-H atoms become more, leading to the increase of the Rashba effect.

Large band-gap quantum spin Hall state

The topological properties of the concerned band gaps for the monolayer Tl_2H and Tl_2H_2 are investigated. Figure 9a,b show the band structures of the monolayer Tl_2H and Tl_2H_2 with SOC. The acquired bands from the Wannier interpolation method are also displayed, which are in very good agreement with those obtained

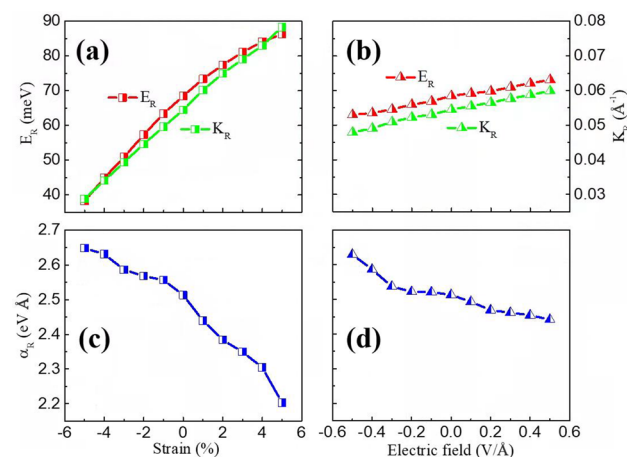


Figure 8. The Rashba energy E_R (red color) and the momentum offset K_R (green color) under biaxial strain (a) and external electric fields (b) in the monolayer Tl_2H_2 . The Rashba parameter α_R under different biaxial strain (c) and external electric fields (d) is also shown.

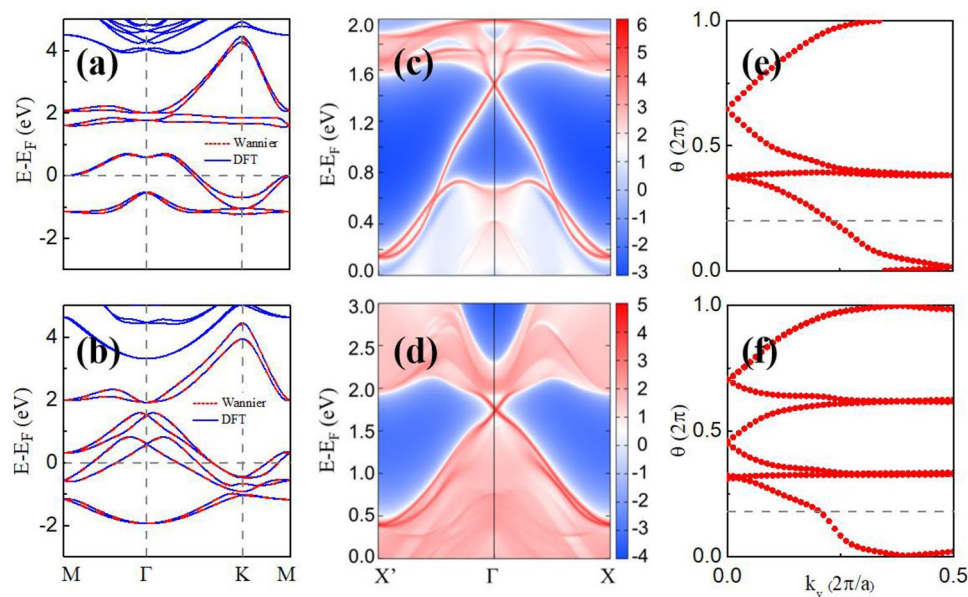


Figure 9. Band structures for the (a) Tl_2H and (b) Tl_2H_2 monolayers with SOC by using Wannier interpolation (red dotted curves) calculations. The DFT results (blue solid curves) are also shown. (c, d) The calculated edge states for the Tl_2H and Tl_2H_2 monolayers. Evolutions of the Wannier function centers for Tl_2H (e) and Tl_2H_2 (f) along k_y , yielding $Z_2 = 1$.

from the DFT calculations. Figure 9c,d give the edge states of the semi-infinite Tl_2H and Tl_2H_2 samples by using the Green's function method and the semi-infinite MLWF Hamiltonian. Obviously, the topological protected gapless helical edge states connect the bands at the two sides of the band gaps, as expected. To further identify the topological properties of the gapped states of the monolayer Tl_2H and Tl_2H_2 , the topological invariant Z_2 is obtained by using the Wannier charge center (WCC) method⁴⁴. The calculated $Z_2 = 1$ for the monolayer Tl_2H and Tl_2H_2 with the E_F set to be located inside the band gaps indicates that the gapped states are QSH states with topological nontrivial band gaps induced by SOC (Fig. 9e–f). The E_F movement can be achieved experimentally through such as carrier doping with the carrier concentration of 4.1×10^{14} and $4.2 \times 10^{14} \text{ cm}^{-2}$ for Tl_2H and Tl_2H_2 , respectively, which can be realized in experiments via current advanced gating technologies⁴⁵. Our research results indicate that the QSH states with large band gaps can be achieved in the thallene film by functionalization. No strong tensile strain is needed to acquire the QSH state in Tl_2H and Tl_2H_2 , superior to the pristine thallene¹⁸. The obtained QSH states are actually very robust against the strain. As indicated in Fig. 5a, the global band gaps for Tl_2H and Tl_2H_2 are both not closed under the strain varying from -5 to 5% . Thus, the QSH states are well maintained even under 5% tensile or compressive strain.

To check the accessibility of the topological states for Tl_2H and Tl_2H_2 in experiments, we deposit Tl_2H and Tl_2H_2 monolayers to a substrate material of SiC (0001). The built geometrical structures of the $\text{Tl}_2\text{H}/\text{SiC}$ and $\text{Tl}_2\text{H}_2/\text{SiC}$ heterostructures are displayed in Fig. 10a,c. Note that the bottom and top surfaces of the SiC substrate are saturated with H atoms to remove the dangling bonds of the surface atoms for the SiC substrates. The lattice mismatch between the $\text{Tl}_2\text{H}/\text{Tl}_2\text{H}_2$ sample and the SiC (0001) substrate is very small (about $1.82\%/0.96\%$). The optimized interface distance between the sample and the substrate is $2.32 \text{ \AA}/2.77 \text{ \AA}$, indicating the van der Waals (vdW) interactions in the interface. The achieved band structures are shown in Fig. 10b,d. The bands near the E_F of Tl_2H and Tl_2H_2 are obviously not affected much by the SiC substrate, due to the weak vdW interactions from the substrates. Therefore, the QSH states displayed in Fig. 9a,b keep well in the samples when depositing on the SiC substrate. The acquired SiC substrate together with the large binding energies ($>4.5 \text{ eV}$) and the negative formation energies for Tl_2H and Tl_2H_2 indicates the experimental accessibility of the unique electronic and topological states in the two materials.

Conclusions

In summary, we have built two hydrogenated thallene Tl_2H and Tl_2H_2 monolayers and explored systematically their electric structures and topological properties based on density functional and crystal field theories. Our results indicate that non-Dirac quantum spin Hall states with large band gaps can be achieved in thallene by unilateral hydrogenation. The topological nontrivial band gap can be up to 855 meV , much higher than that of most of the proposed quantum spin Hall insulator. The QSH states in both monolayers are very robust and well maintained even under 5% tensile or compressive strain. Due to the absence of structural inversion symmetry, the Tl_2H_2 monolayer exhibits very strong Rashba spin splitting characteristics. The non-Dirac quantum spin Hall state and Rashba effect can be tuned efficiently by applying biaxial strain and external electric fields. Our results provide an excellent material platform for realizing room-temperature topological electronic devices.

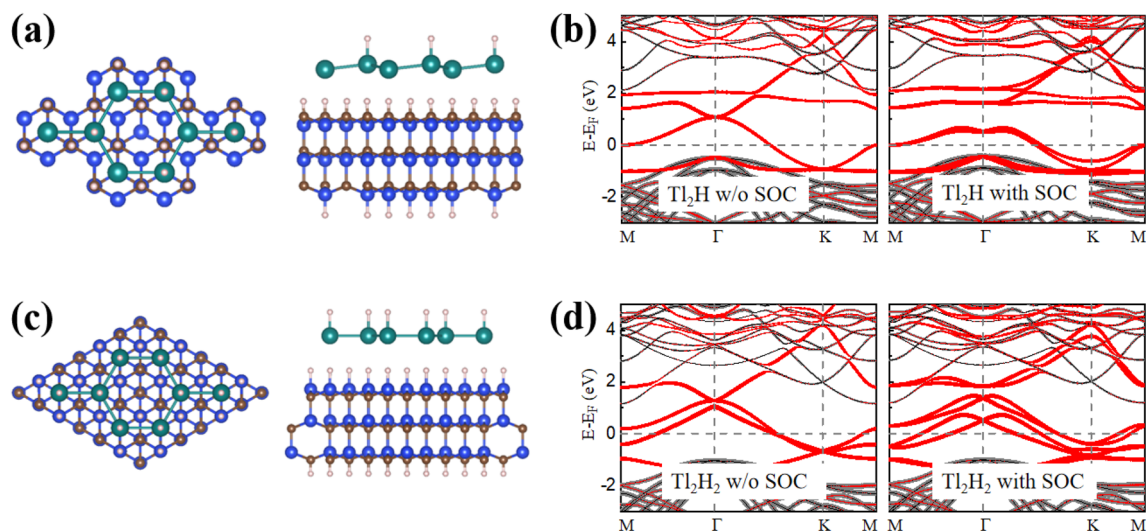


Figure 10. The geometry structures of the (a) Tl_2H and (c) Tl_2H_2 monolayers placed on a threefold symmetric SiC (0001) substrate from the top and side views, respectively. The substrate is modeled as a SiC three-layer, saturated with H atoms in the bottom and top surfaces. The unit cell of the system is in a commensurate ($\sqrt{3} \times \sqrt{3}$) reconstruction of SiC (0001). (b) and (d) give the band structures of the Tl_2H and Tl_2H_2 monolayers on the SiC (0001) substrate without and with SOC, respectively.

	a (Å)	$d_{\text{Tl-Tl}}$ (Å)	$d_{\text{Tl-H}}$ (Å)	h (Å)	E_b (eV)	E_f (eV)	ΔE_g (meV)
Tl_2H	5.24	3.04	1.86	0.29	4.57	-1.62	854
Tl_2H_2	5.28	3.05	1.87	0	5.18	-2.23	334

Table 1. The calculated equilibrium lattice constants (a), Tl–Tl bond lengths ($d_{\text{Tl-Tl}}$), Tl–H bond lengths ($d_{\text{Tl-H}}$), vertical distances between the two Tl atomic planes (h), the binding energies (E_b), the formation energies (E_f), and the global band gaps (ΔE_g) for the monolayer Tl_2H and Tl_2H_2 .

Models and methods

The geometry optimization and electronic structure calculations of the functionalized thallene are performed based on density functional theory (DFT) with the projector augmented wave method as implemented in the Vienna ab initio simulation package (VASP)⁴⁶. The Perdew–Burke–Ernzerhof generalized gradient approximation (GGA–PBE) is adopted for the exchange–correlation functional⁴⁷. The cutoff energy is set as 550 eV for the plane-wave basis and the vacuum space along the c axis is set to about 15 Å to avoid the interactions between the two adjacent slabs. The first Brillouin zone (BZ) is sampled with k meshes of $15 \times 15 \times 1$ by using Gamma-centered Monkhorst–Pack method⁴⁸. The convergence threshold for the total energy is set to 1×10^{-6} eV. All the atoms are allowed to relax until the force on each atom is less than 0.01 eV/Å, which is calculated according to the Hellmann–Feynman theorem. The structural optimization is performed without the symmetry constraint. The maximally localized Wannier functions (MLWFs) are constructed by employing the WANNIER90 code⁴⁹, in which the edge states are calculated with an iterative Green function method^{50,51}. In addition, the screened exchange hybrid density functional by Heyd–Scuseria–Ernzerhof (HSE06)⁵² is also adopted for confirming the electronic structures.

Data availability

The datasets used and/or analysed during the current study available from the corresponding author on reasonable request.

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Author contributions

X.L. and Z.L. performed the theoretical calculations with the assistance of H.B., X.L., Z.L., H.B., and Z.Y. analyzed the results. Z.Y. guided the overall project. X.L. and Z.Y. wrote the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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