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OPEN Electrical and thermal characterisation of liquid metal thin-film Ga₂O₃–SiO₂ heterostructures

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Heterostructures of Ga₂O₃ with other materials such as Si, SiC or diamond, are a possible way of addressing the low thermal conductivity and lack of p-type doping of Ga₂O₃ for device applications, as well as of improving device reliability. In this work we study the electrical and thermal properties of Ga₂ O₃-SiO₂ heterostructures. Here, thin-film gallium oxide with thickness ranging between 8 and 30 nm was deposited onto a silicon substrate with a thermal oxide by means of oxidised liquid gallium layer delamination. The resulting heterostructure is then characterised by means of X-ray photoelectron spectroscopy and transient thermoreflectance. The thin-film gallium oxide valence band offset with respect to the SiO₂ is measured as 0.1 eV and predicted as -2.3 eV with respect to diamond. The thinfilm's out-of-plane thermal conductivity is determined to be 3 ±0.5 Wm⁻¹ K⁻¹, which is higher than what has been previously measured for other polycrystalline Ga₂O₃ films of comparable thickness.

Gallium oxide is an ultra-wide band gap material (4.8 eV for its β polymorph¹) that has attracted a lot of attention for power electronics in recent years. Its breakdown electric field is predicted to be around 8 MV cm $^{-1.2}$ significantly higher than the 2.6 MV cm⁻¹ and 3.3 MV cm⁻¹ for SiC and GaN respectively, which are established materials for power electronics applications³. Gallium oxide offers the potential for ultra-high voltage power device technology, even exceeding 10 kV. This, along with its high Baliga figure of merit and low cost substrates due to the availability of melt-grown Ga₂O₃, has made gallium oxide an attractive material for power electronic devices for use in various high-voltage applications, including power conversion, electric vehicles, data centres^{4,5}. There has also been significant interest in the fabrication of 2D thin-film gallium oxide for potential 2D materialbased applications, such as gas sensing⁶, water-splitting solar cells⁷ and even wearable electronics⁸.

Thermal transport is one of the main challenges for Ga₂O₃-based devices. The most thermodynamically stable phase of gallium oxide, β -Ga₂O₃, has a relatively low thermal conductivity, which is also anisotropic, ranging between 11 Wm⁻¹ K⁻¹ and 27 Wm⁻¹ K⁻¹ depending on crystallographic direction^{9,10}. To put this in perspective, the relevant values for SiC and GaN are about an order of magnitude higher at 420 $Wm^{-1}K^{-1}$ and 160 $Wm^{-1}K^{-1}$ respectively¹¹. For any potential device, the low thermal conductivity of the semiconductor may lead to device failure under operation due to poor thermal dissipation. A possible solution to this is the integration of Ga_2 O₃ with a high thermal conductivity material/substrate. Numerous approaches have been reported including integration with SiC via wafer bonding, where temperature rise has been predicted to reduce by up to 30% for a bottom side cooling scheme¹². Another issue with Ga_2O_3 is its poor hole mobility, which together with the lack of suitable shallow acceptors, makes Ga₂O₃-based bipolar or p-type devices so far impossible¹³. A p-n junction, however, can be established by integration of n-type Ga₂O₃ with a p-type material, which has been accomplished with p-doped nickel oxide for the purpose of diodes with tuneable electrical and optical properties¹⁴, as well as p-doped GaN for self-powered photodetectors¹⁵. Furthermore, modelling showed that a p-n Ga₂O₃-diamond superjunction would lead to approximately 60% reduction in temperature rise under operation¹⁶. It is necessary to know the band alignment across the heterojunction to design efficient devices of this type. Note that local stoichiometric inhomogeneities in Ga₂O₃ have been shown to affect core and valence states in the material¹⁷. In fact, the valence band offset of Ga₂O₃ with silicon has been reported to vary for different gallium oxide polymorphs, ranging from -2.9 eV (for ϵ -Ga₂O₃) to -3.7 eV (for κ -Ga₂O₃)¹⁸. Because of

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this, we may expect different electronic properties from different amorphous/polycrystalline Ga_2O_3 samples, such as the one examined here.

In this work we investigate the electrical and thermal properties of a Ga_2O_3 -based heterointerface, realised through deposition of thin films of Ga_2O_3 onto silicon with thermal oxide. The deposition method used is based on the exfoliation of thin-film gallium oxide from liquid gallium - a recently proposed technique to realise 2D metal oxides¹⁹. Silicon has a thermal conductivity of about 130 Wm⁻¹ K⁻¹²⁰, significantly higher than Ga_2O_3 and so is a potential material for thermal management of Ga_2O_3 -based devices. We obtain values for the valence band offset of the deposited gallium oxide with SiO₂, its out-of-plane thermal conductivity and thermal boundary resistance to the silicon substrate. This data can also be used predictively to assess the thin-film gallium oxide viability for use in tandem with other high thermal conductivity substrates, such as diamond.

Methods

Pure gallium has a melting point slightly above room temperature - at 29°C. When exposed to air, the surface of the liquid metal is spontaneously oxidised due to a low Gibbs Free Energy for the formation of $Ga_2O_3^{21}$. This passivating oxide layer is up to a few nanometres thick and with a large chemical potential gradient at the interface between the liquid core and oxide layer²². Because of this van der Waals forces have been found sufficient to detach this oxide layer from the bulk and adhere it to a separate substrate²³. A gallium pellet is taken and heated on a hot plate to 50°C, i.e. above its melting temperature. A pipette tip is then used to pick up a liquid gallium droplet, which is in turn placed on a glass slide, kept in liquid form on the hot plate. The thin film of Ga_2O_3 or oxide skin is then put in contact with a B-doped Si substrate with thermal oxide, resulting in large area transfer of Ga_2O_3 film, as illustrated in Fig. 1a. Excess gallium is cleaned off by rinsing the sample in heated ethanol. The sample is then annealed in oxygen at 250 ° C for 1 h. This step has been suggested to aid in stabilising the stoichimometry of the deposited Ga_2O_3 film¹⁹. A microscope image of the layer post annealing is shown in Fig. 1b; the boundary between the substrate and thin-film oxide is evident. An Agilent 5420 Atomic force microscope (AFM) was used in tapping mode, confirming that the layers prepared had a thickness ranging from 8 to 30 nm, shown in Fig. 1c. The extracted profile for a thicker sample is visibly uneven. This is likely due to an overlapping of several oxidation layers.

In order to measure the out-of-plane thermal conductivity of the Ga₂O₃, transient thermoreflectance (TTR) was used. This technique uses a nanosecond laser heating pulse and a continuous wave probe laser to measure the transient thermal response²⁴. A frequency tripled 10 ns 355 nm Nd:YAG pump laser with a 30 kHz repetition



Figure 1. (a) Schematic of the exfoliation method - a liquid gallium droplet is isolated and its passivation oxide is directly transferred onto a substrate. (b) Microscope image of the Ga_2O_3 film deposited on thermally oxidised Si substrate after annealing at 250°C for 1 hour. (c) Atomic Force Microscopy linescans taken across thin-film oxide to substrate edges in two different areas.

rate and a spot diameter of 85 μ m was used to heat up the sample surface and a 532 nm probe laser with spot size of about 2 μ m was used to measure the induced transient reflectivity change. 10 nm of chromium and 100 nm of gold were thermally evaporated on the sample surface prior to the measurement, acting as a transducer. More details about the experimental TTR setup used here are given in Yuan et al.²⁵. It should be noted that the probing depth of this TTR setup does not exceed 10 microns, and has lower sensitivity for layers under 100 nm²⁶. Because of this, for the purpose of said measurements we choose to investigate the deposition area with the largest thickness (30 nm). Thermoreflectance transients were recorded for the Ga₂O₃, as well as on the bare SiO₂ as a reference, and an analytical model outlined in Yuan et al. was used to fit the thermal conductivities across different layers onto the data²⁵.

High resolution X-ray photoelectron spectroscopy (XPS) was used to measure the valence band alignment of the Ga₂O₃ film to the substrate using a monochromatic Al k α (h ν = 1486.7 eV) excitation source with a pass energy of 50 eV. Information about the energetics of core levels and valence band maxima were extracted. The valence band offset of Ga₂O₃ with respect to SiO₂ is given as²⁸

$$\Delta E_V = \left(E_{Ga3d}^{Ga_2O_3} - E_V^{Ga_2O_3} \right) - \left(E_{Si2p}^{SiO_2} - E_V^{SiO_2} \right) - \left(E_{Ga3d}^{Ga_2O_3 - SiO_2} - E_{Si2p}^{Ga_2O_3 - SiO_2} \right), \tag{1}$$

where $E_V^{Ga_2O_3}$ and $E_V^{SiO_2}$ denote the valence band energies for the two materials - Ga₂O₃ and SiO₂, respectively, $E_{Si2p}^{SiO_2}$ and $E_{Ga3d}^{Ga_2O_3}$ denote the energies of the core levels Si 2p and Ga 3d in the spectra taken solely from SiO₂ and Ga₂O₃, respectively, while $E_{Ga3d}^{Ga_2O_3-SiO_2}$ and $E_{Si2p}^{Ga_2O_3-SiO_2}$ denote the the energies of the two core levels as measured across the Ga₂O₃-SiO₂ interface. XPS has a low probing depth that rarely exceeds few tens of nanometres, however, due to the thin-film nature of the gallium oxide deposition, any measurement taken from the Ga₂O₃ film is expected to probe through the interface and into the SiO₂ layer. Because of this, values for $E_V^{Ga_2O_3}$ and $E_{Ga3d}^{Ga_2O_3}$ cannot be reliably determined from our data, and a standard value for the term $E_{Ga3d}^{Ga_2O_3}$ - $E_V^{Ga_2O_3}$ =17 eV is used instead²⁷. For measurements, an area on the sample with gallium oxide deposition of 8 nm thickness was chosen, from which $E_{Ga3d}^{Ga_2O_3-SiO_2}$ and $E_{Si2p}^{Ga_2O_3-SiO_2}$ data was extracted. An area on the sample without deposition was also chosen for the estimation of $E_V^{SiO_2}$ and $E_{Si2p}^{Ga_2O_3-SiO_{229}}$. The peak positions for Ga 3d and Si 2p were estimated from the data via Gaussian fitting.

Results and discussion

The recorded XPS spectra from the SiO₂ and deposited Ga₂O₃ on SiO₂ can be seen in Fig. 2a in red and blue respectively. Both data sets were rigid-shifted by 2.3 eV so that the Si 2p peak from the SiO₂ ($E_{Si2p}^{SiO_2}$) spectrum is apparent at 103.3 eV as is standard³⁰. The Si 2p peak is also visible in the Ga₂O₃-SiO₂ spectrum, although is slightly obscured by several overlapping Ga 3p peaks. On the Ga₂O₃-SiO₂ spectrum the Ga 3d peak is apparent at 21.5 eV with the additional peak at about 25 eV being related to oxidation - a characteristic feature of a Ga₂O₃ spectrum³¹. These peaks are also visible in the spectrum obtained from the SiO₂, though with significantly lower intensity, likely appearing due to residual traces of gallium from the deposition. The SiO₂ XPS spectrum in the close vicinity of the valence band maximum is seen in Fig. 2b. The value for the valence band energy is taken as the intercept of two linear fits around the points of steepest increase, determined as 4.4 eV.

The obtained 0.12 eV valence band offset of Ga_2O_3 with respect to SiO_2 is shown schematically in Fig. 2c. The binding energy values used for calculating said offset using (1) from the experimental data are given in Table 1. For the purpose of visualisation and conduction band offset discussion, we are taking a band gap value for the thin film deposited here equal to 4.9 eV (equal to the band gap of standard β -Ga₂O₃). This is consistent with high resolution transmission electron microscopy (HRTEM) characterisation on films deposited under identical conditions, identified as polycrystalline β -Ga₂O₃²³. We also note that β -Ga₂O₃ is the thermodynamically most stable gallium oxide polymorph, with the second most stable - κ -Ga₂O₃ also having a band gap of 4.9 eV¹⁸. Taking the silicon oxide band gap as 8.9 eV³², that results in a conduction band offset of -4.0 eV for our thin film Ga₂O₃ with respect to SiO₂. Comparing to reported values in the literature, considering a valence band offset of 4.4 eV between Si to SiO₂³³, the here obtained valence band offset of Ga₂O₃ to Si ΔE_V would be -4.3 eV, with a conduction band offset of 0.5 eV. For comparison, a value of -3.5 eV was reported for the valence band offset in a PLD β -Ga₂O₃-Si interface (with a conduction band offset of -0.2 eV), showing a significant difference between the pure β phase and the film deposited here³⁴. The change of sign between the two conduction band offsets implies that while a β -Ga₂O₃-Si junction has type I alignment, the Ga₂O₃ film deposited in this work would have a type II alignment to silicon. Figure 2c also shows predicted band alignment of the deposited thinfilm Ga2O3 to GaN, SiC, Al2O3 and diamond, based on the measured band alignment of GaN with respect to SiO_2 and SiC^{35} , GaN with respect to $Al_2O_3^{36}$, and GaN with respect to diamond³⁷. We thus estimate the valence band offset of the thin-film Ga_2O_3 to diamond as -2.3 eV, with a predicted conduction band offset of -2.85 eV. This alignment provides significant energetic barriers for minority carriers across a potential n-type Ga₂O₃ to p-type diamond heterojunction—about 0.8 eV higher than in PLD β -Ga₂O₃. This also correlates to a higher breakdown field in a potential Schottky barrier diode, such as the one proposed by Mishra et al., using a Ga₂O₃ -Al₂O₃-diamond superjunction¹⁶.

Next, we investigate the thermal properties of the deposited thin gallium oxide film. As discussed earlier, 10 nm of Cr and 100 nm of Au were evaporated on the sample surface prior to TTR measurements. A diagram of the layers for the two areas thermoreflective transients were recorded for can be seen in Fig. 3a. Values for the out-of-plane thermal conductivity, heat capacity and density of the individual layers are presented in Table 2. The thermal conductivities used for gold and SiO₂ are reduced with respect to their bulk values due to their thin-film nature^{38,39}. The thermal conductivity for the silicon is also reduced from its pure bulk literature value due to the effects of doping⁴⁰. A sensitivity analysis⁴¹ of the thermoreflectance transient trace with respect to the thermal



Figure 2. XPS energy spectra recorded from the (**a**) SiO_2 and Ga_2O_3 film on SiO_2 ; a zoom into the valence band region for the SiO_2 is shown separately in (**b**), where the intersect of dashed lines is used to identify the valence band maximum (VBM); (**c**) shows a diagram of the band alignment of the Ga_2O_3 film to the SiO_2 and extended to other materials. A band gap of 4.9 eV is assumed for our Ga_2O_3 film to determine conduction band offsets.

	Ga ₂ O ₃	SiO ₂	Ga ₂ O ₃ /SiO ₂
Ga 3d	20.25 eV*		21.5 eV
Si 2p		103.3 eV	103.5 eV
VBM	3.23 eV*	4.4 eV	

 Table 1. Table of biding energies used for valence band offset determination. *Values taken from Huan et al.²⁷.

conductivities of the individual layers was carried out and is shown in Fig. 3b. It decouples the contributions from each layer to the overall data and indicates their relative weighting when summed up into the full transient thermal response. We note that the sensitivity to the thermal conductivity of the Ga₂O₃ is fairly low, which would imply a larger uncertainty in the fitting. On the other hand, we observe a high sensitivity to the thin Cr adhesion layer. Its thermal conductivity is first determined from fitting to the data from the bare thermal oxide on the Si substrate as κ_{Cr} =0.14±0.005 Wm⁻¹K⁻¹, equivalent to a TBR of 7.1±0.2 m² KGW⁻¹. The normalised transients trace measured on the thin gallium oxide film with its fit as determined by the model is shown in Fig. 3c. With the remaining values for the layers' thermal conductivities set (including that of the Cr layer ascertained from the dataset without any Ga₂O₃ deposition), the out-of-plane thermal conductivity of the Ga₂O₃ film is obtained as 3 ± 0.5 Wm⁻¹K⁻¹. Taking into account the non-uniform nature of the deposition thickness, we further estimate the thermal conductivity of the film to vary between approximately 1.7 Wm⁻¹K⁻¹ and 4.8 Wm⁻¹K⁻¹ for thicknesses between 20 and 40 nm, respectively. This is in line with theoretical predictions for the thermal conductivity of crystalline β -Ga₂O₃ thin films (with expected values up to 4 Wm⁻¹K⁻¹ for films of about 30 nm thickness), although lower due to its polycrystalline nature⁴².



Figure 3. (a) Schematics of sample layer structure - with and without Ga_2O_3 deposition. (b) Plot of the fitting model's sensitivity to the layers' thermal conductivities as parameters. (c) Measured and modelled transient thermoreflectance traces for data including the Ga_2O_3 layer. (d) 2D FEM thermal simulation showing the ΔT versus depth below a 4 μ m-length, 1 Wmm⁻¹ heat source in the Ga_2O_3 layer.

Layer	Out-of-plane thermal conductivity [Wm ⁻¹ K ⁻¹]	Heat capacity [Jkg ⁻¹ K ⁻¹]	Density [kgm ⁻³]	Thickness [nm]
Au	200 ³⁸	12945	19300 ⁴⁵	100
Cr	0.14*	448 ⁴⁵	7150 ⁴⁵	10
Ga ₂ O ₃	3*	560 ⁴⁵	5880 ⁴⁵	30
SiO ₂	1.239	1000 ⁴⁵	2370 ⁴⁵	30
Si	8040	70045	2329 ⁴⁵	400,000

 Table 2. Parameters used for the TTR fitting. *Obtained from fitting the experimental data.

This value, however, is twice as high as the thermal conductivity achieved from atomic layer deposition (ALD) of polycrystalline β -Ga₂O₃ film of comparable thickness onto diamond (measured as 1.5 Wm⁻¹ K⁻¹ at 30 nm thin film)⁴³. Previously the thermal conductivities of polycrystalline β -Ga₂O₃ films (grown by open atmosphere annealing of GaN films) have been measured in the range between 0.34 Wm⁻¹ K⁻¹ up to 8.85 Wm⁻¹ K⁻¹ for thicknesses ranging between 12.5 nm and 895 nm respectively, which makes the result presented here on the high end of the spectrum of predicted values⁴⁴.

In Fig. 3d an ANSYS 2D finite element method (FEM) simulation of the steady state temperature rise across the heterojunction is shown, using the standard and measured thermal conductivities and thicknesses given in Table 2. The simulation predicts a temperature rise of approximately 10° C across the SiO₂ layer from a 4 μ m long 1 Wmm⁻¹ heat source within the Ga₂O₃ layer. By comparison, the Δ T across the Ga₂O₃ layer is much smaller. This illustrates that for a typical device heat source (such as in a metal-oxide-semiconductor field-effect transistor (MOSFET)) the Ga₂O₃ layer presents a negligible thermal resistance because it is very thin. Therefore, this is a viable thermal management approach for a thin-channel transistor.

One of the aspects that contributes to the thermal resistance across an interface is the mismatch of the vibrational density of states (VDOS) between the two materials^{46,47}. Among the three considered materials— Ga_2O_3 , SiO₂ and Si, the pair with the largest VDOS overlap is Ga_2O_3/SiO_2 , while the pair with the lowest is $SiO_2/Si^{10,48}$. This suggests that a TBR between Ga_2O_3 and silicon (without SiO_2 as interlayer) could still be low (comparable or lower than between SiO_2 and Si). Of course, this should be subject to future confirmation as the phonon modes primarily responsible for interfacial thermal transport can be unique to the interface in question and are not necessarily represented in the VDOS of the individual bulk materials⁴⁹.

In summary, electrical and thermal properties of thin-film Ga_2O_3 -SiO₂ heterostructure were studied. We reported band offsets and out-of-plane thermal conductivity of thin-film Ga_2O_3 , realized through delamination of thin passivation layers from a liquid gallium droplet onto Si with thermal oxide substrate. The estimated valence band offset of our thin film Ga_2O_3 with respect to SiO₂ is 0.1 eV and the predicted offset with respect to diamond is -2.3 eV, suggesting possibly a non-blocking interface of Ga_2O_3 with SiO₂ and a blocking interface with diamond. Moreover, out-of-plane thermal conductivity of thin-film Ga_2O_3 was found to be around 3 Wm⁻¹ K⁻¹, which is lower than bulk β -Ga₂O₃, although higher than what has previously been achieved for polycrystalline films of comparable thickness.

Data availability

The datasets generated during and/or analysed during the current study are available from the corresponding author on reasonable request.

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Author contributions

A.P. and A.M. prepared samples. X.P.S. experiment was performed by M.C., T.T.R. was performed by D.F. Data analysis was done by A.P., A.M. and J.P. M.K. supervised the work. Main manuscript text was written by A.P. and reviewed by all authors.

Competing interests

The authors declare no competing interests.

Additional information

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