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OPEN Determination of in vivo biological activities of Dodonaea viscosa flowers against CCL₄ toxicity in albino mice with bioactive compound detection

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Dodonaea viscosa L.Jacq. is an evergreen shrub and native to Asia, Africa, and Australia. It has been used as traditional medicine in different countries. The foremost objective of the current study was to discover the protective potential of D. viscosa flowers Methanol (DVM) and Chloroform (DVC) extracts against CCL₄ induced toxicity in mice. This study was intended to identify phytochemicals through HPLC, GCMS, and FT-IR, as well as in vitro antioxidant and in vitro anti-tuberculosis activity. Our comprehensive findings indicate that Dodonaea viscosa is valuable and widespread herbal medicine through therapeutic potentials for curing various ailments. Dodonaeaviscosa flowersare found to have a protective effect against oxidative stress produced by CCL₄ in the liver, kidney, and spleen. The intake of DV extracts restored the level of hepatic enzymes (ALP, AST ALT, and Direct bilirubin), hematological parameters (RBCs, WBCs, and Platelets), total protein, and liver antioxidant enzymes (SOD, GPx, and CAT) after a decline in levels by CCL₄. Histopathological results discovered the defensive effect of 300 mg/kg of DVM extract against CCL₄ induced damage, thus having an improved protective effect compared to DVC and control. As a result of metabolite screening, the total flavonoids and total phenolics were present in abundance. A phytochemical investigation by HPLC identified gallic acid, epicatechin, cumeric acid, flavonoids, while GCMS estimated oleic acid (Octadecenoic acid) (C18H34O2), Stearic acid (C18H36O2), Ricinoleic acid (C18H34O3), and Cedrol (C15H26O). DVM extract exhibited resistance against in vitro Mycobacterium tuberculosis strains. So this study proposed that the protective effect of DV against oxidative damage induced in the liver, kidney, and spleen can be correlated to the antioxidant compounds.

Oxidation-reduction reactions fuel different biological processes¹ and produce free radicals such as lipid peroxides that damagecell membrane and alters enzyme activity^{2,3} and induces liver injury, cirrhosis, inflammation, and necrosis⁴.

A foremost cause of toxicity is exposure toalcohol, carbon tetrachloride, x-rays, and many other radiations, which stimulate reactive oxygen species and make different parts of the body susceptible to injury, i.e., liver,

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kidney, spleen, and many others^{5,6} and facilitate damage to the biological molecule and oxidize most of them⁷. CCL₄ has been used in a mouse model for chemically induced hepatotoxicity as a result of oxidative stress. Carbon tetrachloride produces several reactive oxygen species and generates liver injury by cytochrome P450. Synthetic drugs are inadequate and possible to have adverse effects around the world. So to detoxify the xenobiotics, drugs, and infections, natural products have been extensively used in the treatment of various diseases due to the presence of antioxidants^{8,9}. There is a lot of research to uncover the potential of antioxidants to scavenge free radicals and fight against cellular damage, aging, and other chronic diseases^{10,11}. Among antioxidants, phenolics are extensively distributed in herbal medicines, act as anti-inflammatory compounds, and protect the anti-oxidant defense system in the liver^{12,13}. Multi-drug-resistant tuberculosis (MDRTB) is growing very fast in medical science andislooking for a short-term and effective treatment measure¹⁴. Many plant species have always shown potential in ancient times to treat medical disorders with promising therapeutic agents¹⁵. The knowledge about a single chemical component in medicinal plants plays a critical role from the method of extraction to understanding pharmacological assay and possible toxicity¹⁶.

The evergreen shrub*Dodonaea visocosa* (DV) belongs to the family Sapindaceae, which comprises species (2000) and genera (150). This plant is native to Asia, Africa, and Australia and islocally known as vilayati Mehndi in South East Asia¹⁷. The previous studies reported the presence of phytochemicals such as triterpenoids, flavonoids, saponins, tannins, and coumarins in DV¹⁸⁻²⁰. DV has been widely used in traditional medicineto treat inflammation, stomach ulcers, and liver aches^{21,22}. The crude extractof DV revealed antioxidant, anti-microbial, anti-diabetic, gastro-protective activities and also reported hepato-protective activity in mice²³⁻²⁶. There is no detailedstudy reported regarding protective effects of flowers of DV, so taking it into accountpresent study was conducted to validate the potential of therapeutic activity of DV against liver and other organ toxicity. The foremost objective of this study was to discover the protective potential of *D. viscosa* flowers Methanol (DVM) and Chloroform (DVC) extracts against CCL₄ induced toxicity in mice. This study was intended to identify phytochemicals by using HPLC, GCMS, and FT-IR as well as antioxidant anti-tuberculosis activities.

Material and methods

Sample collection and preparation. The collection of *Dodonaea Viscosa* flowers were carried out during March and April from Murree hills, District Rawalpindi, Punjab, Pakistanaccording to the guidelines of institution (Quaid i Azam University, Islamabad) and alsothrough the permission of local community. With the Help of the Department of Botany Quaid-I-Azam University Islamabad, Pakistan, the plant species was detailed identified by Dr. Mushtaq ahmad (Plant taxonomist, QAU) before further processing. Voucher specimen HG052 was submitted to the herbarium of Pakistan (ISL) for future records. Cleaned specimens were subjected to shade drying followed by grinding and sieving. Dried the plant sample in a heating oven (37°C) to eradicate excess moisture for absolute drying, and the pulverized material was prepared for further examination. Five hundred grams of plant powder was dissolved in Methyl alcohol and Chloroform for 5 days, and filtration of extracts was done with Whatman filter paper. Both extracts were evaporated with the help of a rotary evaporator, and crude dried extracts were stored in air-fitted vials for more processing.

In vivo study. Selection and purchase of animals. Fifty male albino mice (body weight 55.2 ± 2.5 g) were purchased from NIH (National Institute for health) Islamabad. The experimental study was approved by the National Veterinary Lab Islamabad ethical committee adhering to the institution's guidelines and also in compliance with ARRIVE guidelines. All the protocols having animal study were in acquiescence with the guidelines of the ethics committee. The mice were housed under controlled conditions and had free access to mouse chow (Feed Mills, Islamabad) and water ad libitum. Animals were cautiously monitored and kept up in standard house conditions.

Acute oral toxicity study. An acute toxicity study was done to select suitable doses of plant extracts for animals, as earlier reported by²⁷. The bodyweight of animals was recorded before and after thestudy. Plant extracts were orally given to mice at the dose of 100-300 mg/kg body weight. After the dosage, animals were meticulously observed after 24, 48, and 72 h for the development of any toxicological symptoms. Animals were euthanized on 21 days of the experiment.

Experimental design. Animals of the same age were divided randomly into 10 groups, and each group contains 5 animals.

1	Normal control group	Normal feedwas given without any treatment for 21 days
2	Olive oil group	1 ml of olive oil was given orally with their feed up to 21 days
3	CCL ₄ group (- ve control)	${\rm CCL}_4$ (1 cc/kg b. w) was induced by the intra-peritoneal way and-given normal feed
4	100 mg Methanol group	This group was induced CCL4 and after that 100 mg/kg b. w of methanol extracts was orally given with normal feed
5	200 mg Methanol group	Animals were given ${\rm CCL}_4+200~{\rm mg/kg}$ b. w of methanol extracts with normal feed
6	300 mg Methanol group	Animals were provided ${\rm CCL}_4+300$ mg/kg b. w of methanolic extract
7	100 mg Chloroform group	This group was administered ${\rm CCL_4}$ + 100 mg/kg b. w of chloroform extract

1	Normal control group	Normal feedwas given without any treatment for 21 days
8	200 mg Chloroform group	The group was nourished with 200 mg/kg b. w of chloroform extract after ${\rm CCL}_4$
9	300 mg Chloroform group	After CCL ₄ , 300 mg/kg b. w (chloroform extract) was given
10	Silyamrin group	CCL ₄ + 100 mg/kg body weight of Silymarin (standard drug)

At 20 days of the experiment, mice were kept for fasting for 12 h and animals were anesthetized and euthanized with sodium pentobarbiton. After anesthesia (21 days) whole blood was obtained from the heart by cardiac puncture. To get the serum, place the blood sample tube to clot for 30 minand subjected to centrifugation (3000 rpm for 10 min). Animals were sacrificed through cervical dislocation, and organs were collected and then rinsed using ice-cold saline solution and kept at -20 °C for further analysis. The weight of collected organs from all the groups was recorded. For biochemical analysis, phosphate buffer saline was used to the homogenized liver (one part), then centrifuged at 3000 rpm for 20 min and supernatant was stored at -20 °C. For the histopathological study, the liver, kidney, and spleen were stored in formalin solution (10%).

Analysis of blood samples. The serum biomarkers alanine aminotransferase (ALT), aspartate aminotransferase (AST), alkaline phosphatase (ALP), and bilirubin were examined by using an auto-analyzer with AMS diagnostic kits (Italy). RBC (red blood cells), WBC (White blood cells), and platelets in blood samples were estimated with the method reported by²⁸.

Antioxidant enzymes. Liver homogenates were used to evaluate oxidative defense markers (antioxidant enzymes). Catalase (CAT) and superoxide dismutase (SOD) activity was measured by using the protocol²⁹, Glutathione peroxidase (GPx) was calculated by the method of³⁰, and total protein was evaluated by the method suggested by³¹.

Histopathological study. Mice organs (liver, kidney, and spleen) were removed carefully after sacrifice and preserved in formalin (10%). Test specimen were fixed, dehydrated in alcohol, cleaned in xylene, and then inserted into molten paraffin wax. Paraffin sections were cut into 5 μ m thickness by using a microtome, and obtain tissues were mounted on slides and deparaffinized. Tissue sections were processed to staining with Ehrlich's hematoxylin and eosin counterstained (H&E) and examined under a light microscope^{32,33}.

Determination of total phenolic and total flavonoid content. The total phenolic content of extracts was measured by using Folin–Ciocalteu reagent³⁴. Results were expressed as grams of gallic acid equivalents per 500 g/dry weight. The total flavonoid content of the extracts was measured using colorimetric assay³⁵. Results were expressed by using grams of quercetin equivalents per 500 g/dry weight.

Antioxidant activity. *Chemicals required.* Methyl alcohol, ethyl alcohol, chloroform, DPPH, ABTS, Hydrogen peroxide, EDTA, Formalin, Xylene, Heamotoxylin, Essen, KH_2PO_4 buffer, ALT Alanine aminotransferase, AST Aspartate aminotransferase, ALP alkaline phosphatase, and bilirubin were obtained. Analytical grade solvents and reagents were purchased from local dealers of Sigma Aldrich and Merck.

DPPH assay. This process is used to calculate the scavenging capacity of the sample by the protocol with some modifications³⁶. 2 ml aliquot of DPPH (2,2-diphenyl-1-picrylhydrazyl) was poured into each concentration of plant sample ranged from 20 to 100 μ g/ml. This mixture was incubated at 37 °C for 30 min in darkness. Standard or Positive controls were Ascorbic and Gallic acid. DPPH solution was taken as a negative control. Reading was taken recorded at 517 nm, and results were expressed in ascorbic acid equivalent AAE and gallic acid equivalent GAE. The experiment was done in a triplicate manner, and the inhibition percentage is obtained by the following formula

DPPH % =
$$\left[A^{a} - A^{h}/A^{a}\right] \times 100$$

 A^a —absorbance of reaction mixture exceptfor plant extract. A^h —absorbance of a reaction mixture comprising plant extract. IC 50 (µg /mg) was measured by plotting scavenging percentage against extract concentration.

Iron chelating assay. The Iron Chelating method was described by³⁷. Antioxidant potential of plant extract was assessed through incubation of reaction mixture that comprises of different concentrations of plant sample extracts (20–100 μ g/ml), 2 mM ferrous sulfate (1 ml), and 0.25 mM Ferrozine (1 ml). After stirring, let the mixture stand for10 min, and absorbance was read at 517 nm.

Chelating rate % =
$$\left[A^{a}-A^{h}/A^{a}\right] \times 100$$

 A^a —absorbance of control lacking plant extract. A^h —absorbance of mixture with plant extract. Standard solutions or Positive controls (Ascorbic and Gallic acid) were used to make the calibration curve. IC₅₀ was stated as μ g AAE/mg and GAE/mg.

Hydroxyl radical scavengingassay. Plants extract concentrations 20 to100 μ g/ml were investigated by adding 0.2 M Sodium phosphate buffer (7 pH), 2deoxyribose (10 mM), FeSO₄-EDTA (10 mM), H₂O₂ (10 mM) and 525 μ l of H₂O. Put all the mixture into TCA (2.8%) and TBA (1%) and incubate at 90 °C for color development. Spectrophotometric reading was observed at 520 nm. Standard drugs (Ascorbic and Gallic acids) were taken as Positive control, and results were measured in AAE and GAE μ g/mg³⁸.

Scavenging activity =
$$\left[1 - A^h/A^a\right] \times 100$$

where A^a—absorbance of the mixture (without plant sample extracts) and A^h—absorbance of a mixture containing plant sample.

ABTS (2,2-azinobis [3-ethylbenzothiazoline-6-sulfonate]) radical cation decolorization assay. Plant extracts were analyzed through the enhanced ABTS+radical cation scavenging capacity by some modification^{39,40}. ABTS+mixture was prepared by adding 3 mM ABTS (2,2-azinobis [3-ethylbenzothiazoline-6-sulfonate]) and potassium persulfate (2.5 mM). Leave the solution in the dark for 12 h. To measure ABTS+activity, ABTS+solution (3 ml) was taken with different concentrations of plant extract (20 to100 μ g/ml). Optical density was measured at 734 nm. Standard drug used was Ascorbic and Gallic acids.

Percent Scavenging potential =
$$\left[A^{a} - A^{h}/A^{a}\right] \times 100$$

A^a—absorbance of control; A^h—absorbance of plant extract.

Reducing power assay. FRAP (Ferric ion reducing power) was determined by the method that involved the mixing of each plant sample concentrations $(20-100) \mu g/ml$) with phosphate buffer (0.2 M) and potassium ferricyanide (0.1%). Allow the mixture to incubate in a water bath for 20 min. Subsequently, add trichloroacetic acid (10%), and the mixture was centrifuged at 3000 rpm for 10 min. The supernatant was mixed in distilled water (2 ml) andferric chloride (0.01%) and set it down for incubation. Blank and samples were interpreted at 700 nm. Standard compounds, i.e., Ascorbic and Gallic acid, were utilized as a positive control. Results were quantified as AAE and GAE ($\mu g/mg$)^{41,42}.

Hydrogen peroxide scavenging activity (H_2O_2). Hydrogen peroxide scavenging activity was described according to the protocol of¹³. Reaction mixture comprised of H_2O_2 solution 4 Mm (prepared in phosphate buffer) with different plant concentrations (20–100) µg/ml) followed by incubation for 10 min at room temperature. The reading was observed at 230 nm against a blank solution comprising phosphate buffer with 7.4 pH. Gallic acid and Ascorbic acid were used as standard and expressed in GAE and AAE (µg/mg).

Scavenging activity
$$\% = \left[A^{a} - A^{h}/A^{a}\right] \times 100$$

A^a absorbance of H₂O₂ and A^h is the absorbance of mixture with plant extract.

Superoxide assay. The activity was determined by NBT reduction as per the method of Beauchamp and Fridovich, 1971^{44} . PMS (phenazine methosulfate) and NADH (nicotinamide adenine dinucleotide) systems produce superoxide radicals that condense nitro blue tetrazolium (NBT) to purple formazan. Add up 50 mM Phosphate buffer, 0.73 mM NADH, 20 mM PMS and 0.5 Mm NBT in numerous concentrations of sample (20 µg/ml) and incubate them for 20 min. Optical density was documented at 560 nm against blank to determine generated formazan. The positive control was Ascorbic acid and Gallic acid. Inhibition concentration was obtained from the formula:

Scavenging percentage =
$$\left[1 - A^{h}/A^{a}\right] \times 100$$

where A^a—absorbance except plant concentrate and A^h—absorbance of mixture with plant distillate.

Analysis of plant with high-performance liquid chromatography (HPLC). Crude extracts analysis was performed using a Shimadzu HPLC (high-performance liquid chromatography) system, Tokyo, Japan, equipped with a C18 column (250 mm \times 4.5 mm, 5 m) used for the separation at the flow rate of 1 ml/min. The column temperature was sustained at 40°C followed by gradient pump and UV/Visible detector. HPLC grade methanol was used for extraction of the crude plant to prepare the tested sample. Before injection, the filtration of samples was done using a 0.2 µm PTFE filter, and the injection volume was 10 µl. The compounds were eluted using a gradient elution of mobile phases A and B (Acetonitrile and 0.1% phosphoric acid; 36:64). Separation steps are as follows: 0 min-5% B, 15 min-15% B, 15 min-45% B, 5 min-90%B, and Conditioning cycle for 5 min along with the analysis of the following initial conditions. The UV–Vis detection was recorded at 280–285 nm at a current rate of 1 ml/min per 20 min retention time. Quercetin was used as standard, and all data were done in triplicates.

GC–MS. Gas chromatography and mass spectrometry analysis. GC–MS system QP2010 model (Shimadzu^{*}) equipped with Mass selective Detector and split-a split-less system of injection. The instrument was fitted with capillary column RTx-5MS (cross bond 5% diphenyl—95% dimethylpolysiloxane) with 30 m×0.25 mm with

Treatments	Liver weight (g)	Kidney weight (g)	Spleen weight (g)	Rise (%) in body weight
Normal	5.60 ± 1.21	0.48 ± 0.1	2.080 ± 0.19	16±2.9
Olive oil control	5.82 ± 0.94	0.51 ± 0.08	2.15±0.2	20±3.10
CCL ₄	3.96 ± 1.18	0.34 ± 0.12	3.34 ± 0.5	7.41 ± 2.05
Silymarin + CCL ₄	6.02 ± 0.749	0.5 ± 0.17	2.150 ± 0.31	22.27±0.09
DV (100 mg/kg) + CCL ₄	4.016 ± 0.83	0.25 ± 0.045	3.11±0.77	6.03 ± 0.67
DV (200 mg/kg) + CCL ₄	4.67 ± 0.71	0.33 ± 0.01	2.85±0.3	10.060 ± 1.04
DV (300 mg/kg) + CCL ₄	5.105 ± 0.51	0.41 ± 0.1	2.49 ± 0.5	15.52 ± 1.75

Table 1. Protective effect of methanol extract of *Dodonaea Viscosa* flowers. Results were expressed in triplicatemanner with mean \pm SD.

 $0.25 \,\mu$ m film thickness. At the rate of 1.2 ml/min, helium was being used as carrier gas. The temperature program of the column was started at 150 °C (1 min) then programmed at 4 °C/min to 150 °C (10 min). The temperature of the injector was 275 °C while the detector was at 250 °C. 0.2 μ l volume was injected in split mode. A split ratio was 1:50, and the mass spectra were operated electron ionization at 70 eV in Selected Ion Monitoring (SIM) mode were maintained. The run time of the machine was 40 min. The relative percentage of the plant extract compounds was expressed in percentage with normalization of peak area.

Compounds identification. GC mass spectrum interpretation was conducted by employing the database of the National Institute Standard and Technology (NIST). The compound name, molecular weight, and structure of the test materials were determined. The percentage (%) of each compound was calculated by comparison of the average area to total area. The spectrum of unknown constituents was compared with the version 2005, software, and Turbo mass 5.2. The aim was to discover the individual compound or group of compounds that might show its current commercial and traditional roles^{45,46}.

Fourier transform infrared spectrophotometer analysis. The plant extract was analyzed for infrared spectrum analysis by FT-IR (Fourier transform Infrared) spectroscopy Shimadzu machine, IR affinity 1, Japan. At first, loaded samples were grounded by KBr (1:100 w/w) with scan range (400–4000 cm) and 4 cm⁻¹ resolution. Samples components were subjected to structural characterization and indicated functional groups with chemical bond types⁴⁷.

Anti-tuberculosis activity. The anti-mycobacterial activity of *Dodonaea viscosa* L. was measured using the REMA method⁴⁸. *Mycobacterium tuberculosis* strains bug 206 and bug 1972 and H37Rv were grown in 7H9 broth. Sample stock solutions were diluted using DMSO to get the final concentration ranging from 0.98 to 250 µg/mL. Rifampicin was used as positive control drug ranges 0.004 to 1 µg/mL. Add the bacterial culture $(5 \times 10^5 \text{ CFU/mL})$ to each well of the 96-well plate and incubate at 37 °C. Viability was tested using resazurin, and the color change and fluorescence were examined in plates by using SPECTRAfluor Plus microfluorimeter (TECAN). Experiments were performed in triplicates. The lowest concentration resulting in 90% growth inhibition of M. tuberculosis. The MIC was defined as the lowest concentration results in the inhibition of 90% growth of M. tuberculosis.

Statistical analysis. All the data were obtained in a triplicate manner, and results are presented as mean \pm standard deviation. One-way ANOVA was used for the processing of results. Statistical analysis (Mean, standard deviation, probability, and Pearson coefficient correlation) was obtained via statistical software (Prism pad 7). The level of significance was considered at < 0.05.

Results

In vivo study. Acute toxicity and effect on weight. According to the results, acute toxicity manifested significant noticeable signs on the mice's body weight. The observed change was shown in Tables 1 and 2. In the present study, DV flowers with methanol and chloroform extracts found no devastating effect on mice. No mortality was found at the highest dose of 2000 mg/kg as it is considered the highest dose by OECD guidelines for any acute toxicity assay. Three plant concentrations 100 mg/kg, 200 mg/kg and 300 mg/kg were selected for the study.

Liver enzymes. DV flower extracts were tested for protective effect on liver enzymes in albino mice. Group 1 and 2 showed normal enzymes level with no treatment. As a result of CCL_4 treatment in (Group 3) slight to elevated changes were observed after 1, 2, 3, and 21 days in liver enzymes while compared with other groups. (Group 4–9) DVM showed a more restorative effect on liver enzymes than chloroform extract, and group 10 indicated hepatoprotective effect by Silymarineffect (Table 3).

Hematological parameters. Induction of CCL₄ resulted in a decline in RBCs, WBCs and plateletslevel. DV methanol and chloroform extract significantly increase the level of parameters (Table 4). DV methanol depicted greater RBCs and WBCs values closer to the normal control with significant results (p < 0.05).

Treatments	Liver weight (g)	Kidneyweight (g)	Spleen weight (g)	Rise (%) in body weight (g)
Normal	5.60 ± 1.21	0.48 ± 0.05	2.080 ± 0.1	16±2.9
Olive oil control	5.82 ± 0.94	0.51 ± 0.02	2.15 ± 0.5	20±3.10
CCL ₄	3.96 ± 1.18	0.34 ± 0.08	3.34 ± 0.2	7.41±2.05
Silymarin + CCL_4	6.02 ± 0.749	0.5 ± 0.1	2.150 ± 0.16	22.27±0.09
DV (100 mg/kg) + CCL ₄	4.15 ± 0.92	0.24±0.03	3.06±0.8	7.22 ± 0.92
DV (200 mg/kg) + CCL ₄	4.62±1.23	0.28 ± 0.14	2.76±0.5	10.24±2.31
DV (300 mg/kg) + CCL ₄	5.16 ± 0.92	0.36±0.02	2.3±0.33	12.94±2.08

Table 2. Protective effect of chloroform extract of *Dodonaea Viscosa* flowers. All the values are obtained in triplicate (mean ± standard deviation).

	ALT	AST	ALP	Direct bilirubin
Normal contro	38±0.83	80.2 ± 4.1	110±9.2	0.2 ± 0.001
Olive oil control	40±5.2	63±2	175 ± 20	0.32 ± 0.06
CCL ₄ control 1st day (24 h)	74.5±7.7	90±11	208 ± 5.1	1.0 ± 0.2
CCL ₄ control 2nd day (48 h)	85±12	95±3.5	225±11	1.25 ± 0.04
CCL ₄ control 3rd day (72 h)	89±14	109 ± 9.2	247 ± 2	1.7 ± 0.45
CCL ₄ control 21 day	120±9	127.8 ± 8.6	294±15	1.91 ± 0.08
Silymarin drug	66.4±7.2	76.1±12	185 ± 31	0.53 ± 0.09
DVM 100 mg + CCL ₄	62.5±8	104 ± 20	264.8±11	0.7 ± 0.08
DVM 200 mg + CCL ₄	61.8 ± 4.5	88±11	236±14	0.6 ± 0.01
DVM 300 mg + CCL ₄	53±2	74.5±6	197±9.8	0.56 ± 0.02
DVC 100 mg+CCL ₄	77±5	126 ± 7.1	357±12.4	1.1 ± 0.2
DVC 200 mg+CCL ₄	72±3	104.3 ± 5	283.6±3.2	0.98 ± 0.05
DVC 300 mg+CCL ₄	69±6	96±4.3	229±11	0.81 ± 0.07

Table 3. Estimation of liver enzymes from mice blood serum. Results are taken in a triplicate way with $mean \pm SD$. Level of significance at < 0.05.</td>

Groups	RBCs	WBCs	Platelets
Normal group	4.80 ± 0.06	5.02 ± 0.47	240 ± 6.9
Olive oil control	4.93 ± 0.2	5.50 ± 0.008	252 ± 10.1
CCL ₄ control 1st day (24 h)	3.05 ± 0.081	3.68 ± 0.03	170 ± 5.03
CCL ₄ control 2nd day (48 h)	2.73 ± 0.03	3.25 ± 0.15	155.2 ± 0.25
CCL ₄ control 3rd day (72 h)	2.40 ± 0.009	2.913 ± 0.092	140.01 ± 0.17
CCL ₄ control 21 day	1.88 ± 0.005	2.351 ± 0.11	100.59 ± 2.9
Silymarin drug	3.97±0.12	4.802 ± 0.060	227.3 ± 4.12
DVM 100 mg + CCL ₄	3.7±0.016	4.67 ± 0.04	240.3 ± 0.15
DVM 200 mg+CCL ₄	4.181 ± 0.004	5.002 ± 0.12	247.09 ± 0.21
DVM 300 mg + CCL ₄	4.62 ± 0.21	5.56 ± 0.01	256.6±0.82
DVC 100 mg+CCL ₄	2.612 ± 0.093	3.205 ± 0.83	174.1 ± 0.26
DVC 200 mg+CCL ₄	3.05 ± 0.02	3.84 ± 0.131	198.3 ± 0.03
DVC 300 mg+CCL ₄	3.73 ± 0.1	4.291 ± 0.20	224.20 ± 0.19

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Table 4. Hematological parameters of different groups of mice.

Antioxidant enzymes. The results showed that the normal level of enzymes getsaltered due to CCL_4 administration. After carbon tetrachloride induction, hightoxicity reduced the antioxidant enzymes. Silymarin drug (positive control) and plant extracts revealed positive results. Results obtained in a triplicate manner along with coefficient variation were < 0.05 (Table 5).

Histopathology. Morphological changes in organ (Liver, kidney, and spleen) tissues were investigated by histopathological microscopy. CCL_4 intoxication damaged the normal architecture of cells, and after 21 days of DV treatment, the cell structure of mice tissues was integrated. Several concentrations of plant extracts exhibited a

Groups	CAT (m mol/min/mg protein)	SOD (U SOD/mg protein)	GPx (µmol/min/mg protein)	Protein tissue
Normal control	8.2±2	10.15 ± 1.4	32.10±3	3.15 ± 0.4
Olive oil	7.9 ± 0.5	10.23±2	34.7±4.5	3.2 ± 0.25
CCL ₄ 1st day (24 h)	5.94 ± 1.3	6.89 ± 0.7	22.40±3.7	1.2 ± 0.11
CCL ₄ 2nd day (48 h)	5.28 ± 1.7	6.24±1	20.47±1.9	1.03 ± 0.2
CCL ₄ 3rd day (72 h)	5.03 ± 0.43	5.98 ± 1.4	18.16±2	0.98 ± 0.3
CCL ₄ 21 day	4.75±1.9	5.75 ± 0.6	17.83±3	0.75 ± 0.08
$DVM100\ mg+CCL_4$	6.8±0.2	7.79±1	19.4±1.2	1.58 ± 0.05
$\rm DVM200\ mg+CCL_4$	7.3 ± 0.8	8.58±2.3	23.0±3.5	2.1 ± 0.4
$DVM300 mg + CCL_4$	7.8±1.03	10.7±0.5	29.4±2	2.8 ± 0.08
DVC100 mg+CCL ₄	5.9 ± 1.5	5.86±1.09	20.1±3	1.02 ± 0.5
DVC200 mg+CCL ₄	6.2±2	6.9±1.5	23.4±0.7	1.71 ± 0.9
DVC300 mg+CCL ₄	6.57 ± 1	8.02±1	25.8±1	2.1 ± 0.71
Silymarin + CCL_4	8.11±1.2	11.62±0.6	33.7±1.5	3.22 ± 0.8

Table 5. Effects of *Dodonaea Viscosa* flowers on Antioxidant enzymes and total proteins. All the results were obtained in triplicate with mean and standard deviation.

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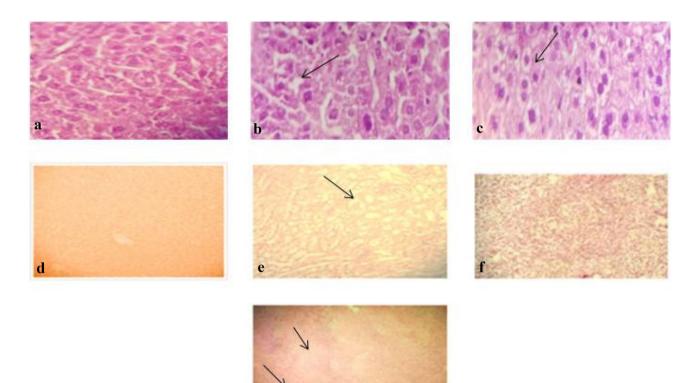
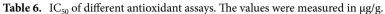


Figure 1. "**a**" showed the normal mice liver tissues, followed by "**b**" (arrow) showed the degeneration of hepatocytes. "**c**" presents recovery of degenerated hepatocytescaused by CCL_4 induction. "**d**" CCL_4 treated cellular structure of kidney. "**e**" DVM treated kidneyshowed slight recovery of renal tubules. "**f**" presents a normal spleen, while "**g**" arrow depicts normal lymphoid masses and red pulp in DVC-treated spleen.

protective effect on tissues by attenuating injuries, while methanol extract presented more supportive evidence on the cellular organization than DVC extract (Fig. 1).

The normal liver architecture depicts central vein, hepatic sinusoids as well as portal veins with normal appearance. Fibrosis and vascular irregularities, for instance, liver sinusoids alteration and central vein congestion, were seen in carbon tetrachloride mice. Renal histology revealed normal features like intact glomerular, tubular structure, bowman's space and capillary tufts. Treated spleen showed white pulp containing normal lymphoid masses followed by extremely vascular red pulp, which was similar to normal mice histology.

	ABTS assay	Reducing power assay	DPPH assay	Iron chelating assay	Hydrogen peroxide assay	Hydroxyl scavenging Assay	Superoxide assay
D. viscosa	107.1 ± 11.4	75.59 ± 4	54.95 ± 2.1	20.7 ± 1.3	11.37 ± 0.4	19 ± 0.56	111.6±2.1
Ascorbic acid	119±7.9	25.7±2	15.7±3	$29.2 \pm .7$	16.8 ± 2.1	24.5 ± 0.84	116.6±2.8
Gallic acid	229 ± 15	39.2±1	24.7 ± 2	34.8±2	13.1±1	26.2 ± 1	134.2±5.6



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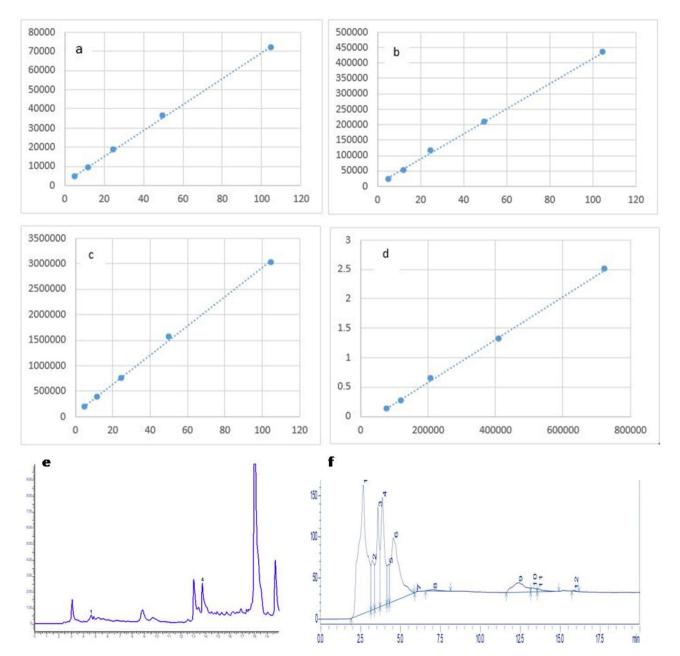


Figure 2. Different peaks on chromatograms showing different compounds. (**a**) Gallic acid standard graph (**b**) epicatechin standard graph (**c**) cumeric acid standard graph (**d**) quercetin standard graph (**e**) peaks showing gallic acid, epicatechin and cumeric acid from DVM (**f**) chromatogram peaks exhibited quercetin from DVM.

Phytochemical analysis and antioxidant assays. Dodonaea viscosa flower possessed total phenolic content 174 ± 4 mg/g dry weight and total flavonoid content 98 ± 7 mg/g, where the results are significantly different p < 0.05. DVM extract was evaluated against different antioxidant activities. The lowest IC₅₀ was found in hydrogen peroxide assay 11.37 ± 0.4 mg/g dry weight and hydroxyl radical scavenging assay 19 ± 0.56 mg/g dry

Compounds	Area	Quantity (mg/kg)
Gallic acid	690,219	196.78
Catechin	0	- 1.04
Epichatechin	101,583	140.76
Cumeric acid	3,374,640	110.85
Quercetin	1,760,828	59.49

 Table 7. HPLC analysis of Dodonaea viscosa flower methanol extract.

Peaks	Compounds name	Formula	Molecular weight (amu)	Area %	Retention time (min)
1	Ethyl fluoride	C ₂ H ₅ F	48	44.66	1.132
2	Isobutyl alcohol	C ₄ H ₁₀ O	74	16.36	1.218
3	Isopentyl alcohol	C ₅ H ₁₂ O	88	16.78	1.350
4	Furanone	C ₅ H ₈ O ₂	100	1.58	1.488
5	Dimethyl Sulfoxonium formylmethylide	C ₄ H ₈ O ₂ S	120	0.34	1.612
6	Isopentyl alcohol	C ₇ H ₁₄ O ₂	130	0.30	1.660
7	Ascorbic acid	C38H68O8	652	4.56	14.900
8	Ascorbic acid	C38H68O8	652	5.56	15.846
9	Octadecenoic acid	C ₁₈ H ₃₄ O ₂	282	2.11	18.858
10	Ricinoleic acid	C ₁₈ H ₃₄ O ₃	298	1.02	19.205
11	Octadecenoic acid	C ₁₈ H ₃₄ O ₂	282	4.30	19.278
12	Stearic acid	C ₁₈ H ₃₆ O ₂	284	0.74	19.697
13	Carboxylic acid	$C_{23}H_{32}O_4$	372	0.40	26.460
14	Cyclopentanone	C15H20O	216	1.08	28.112
15	Cedrol	C ₁₅ H ₂₆ O	222	0.21	28.862

 Table 8. GCMS of *Dodonaea viscosa* flowers compounds. Data were obtained by triplicate readings with mean and standard deviation.

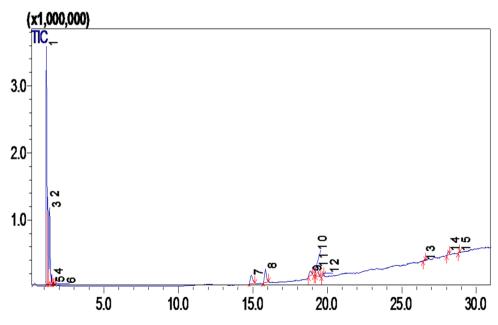


Figure 3. GCMS Chromatogram of Dodonaea viscosa flowers.

Peaks	Wavelength	Bond	Functional group
1	3399.20	O–H stretch, H-bonded	Alcohols, phenols
2	2926.68	O–H stretch	Carboxylic acids
3	2855.03	C–H stretch	Alkanes
4	1712.97	C=O stretch	Carbonyl (general)
5	1651.11	-C=C- stretch	Alkenes
6	1514.15	N–O asymmetric stretch	Nitro compounds
7	1454.94	C–H Bend	Alkanes
9	1265.80	C–H wag (–CH ₂ X)	Alkyl halides
10	1168.38	C–N stretch	Aliphatic amines
11	1079.18	C–N stretch	Aliphatic amines
12	724.89	C–H rock	Alkanes
13	632.98	C–Br stretch	Alkyl halides

Table 9. FT-IR analysis of flowers of Dodonaea Viscosa.

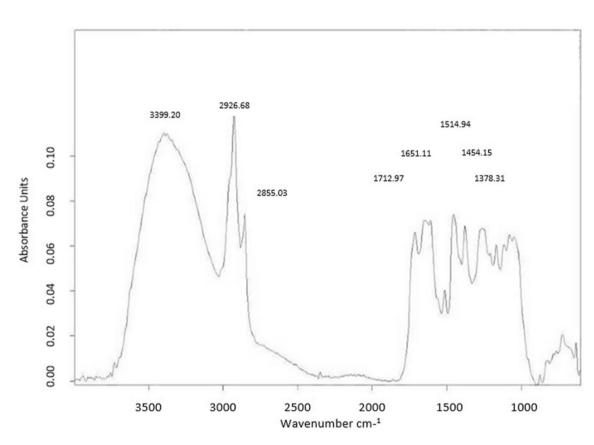


Figure 4. FT-IR analysis of Dodonaea viscosa flowers methanolic extract.

	Mean CFU or	Mean CFU on media		Percentage	Percentage Inhibition		
Isolates	bg1972	H37Rv	Bg206	bg1972	H37Rv	Bg206	
Control	130	140	150	130	140	150	
5 mg/ml extracts	80	40	53	38	71	65	
10 mg/ml	55	4	26	58	97	83	
50 mg/ml	0	0	0	100	100	100	

Table 10. Anti-tuberculosis activity of DV Methanol against different strains. % Inhibition = $Ccn - Ct/Ccn \times 100$, where Ccn = no of colonies in the control media slope, Ct = No of colonies in the Test media slope.

weight, which is greater than Ascorbic and Gallic acid. DVM also showed good activity to quench free radicals in DPPH free radical scavenging assay (Table 6).

High-performance liquid chromatography. For the detection of some important medicinal compounds, HPLC analysis was carried out. Identification of peaks wasmade by comparing the retention time of the *Dodonaea viscosa* flower with standard compounds. The resulting peaks correspond to each compound were proportioned (Fig. 2). HPLC quantification of DVM flower extract identified the presence of gallic acid, epicatechin, cumeric acid, and quercetin compound. Whereas catechin has not been quantified in DV extract (Table 7). Gallic acid was the highest content observed with the quantity of 196.78 mg/kg.

Gas chromatography-mass spectroscopy (GC–MS) analysis. DVM extract is composed of volatile-based organic compounds, mainly fatty acids. Numerous compounds were identified by GC–MS, and the compound list followed by the corresponding GC–MS spectrum was presented in Table 8 (Fig. 3). Among all compounds, the most significant were Ascorbic acid ($C_{38}H_{68}O_8$), Octadecenoic acid ($C_{18}H_{34}O_2$), Ricinoleic acid ($C_{18}H_{34}O_3$), Carboxylic acid ($C_{23}H_{32}O_4$), Stearic acid ($C_{18}H_{36}O_2$), and Cedrol ($C_{15}H_{26}O$).

Fourier transform infrared spectroscopy. The most notable peaks in DVM were observed between 3500 and 2800 nm. A peak at 2926.68 nm belongs to O–H stretch, Carboxylic acids and then at 3399.20 nm correspond to O–H stretch, H-bonded which signifies Alcohols, phenols. Peaks ranging from 2900 to 700 nm belong to C–H stretch, C=O stretch, C–N stretch, and –C=C– stretch (Table 9; Fig. 4).

Anti-tuberculosis assay of *Dodonaea Viscosa* **flowers.** Anti-tuberculosis assay of DVM extract was screened out against *Mycobacterium tuberculosis* 3 strains bg 1972, bg 206 and H37Rv. 5 mg, 10 mg, and 50 mg/ ml concentrations were used, and tuberculosis % inhibition was increased with the rise in concentrations. Plant extract showed resistance against all strains, but the highest activity was found against the H37Rv strain (Table 10). Minimum inhibitory concentration (MIC) was determined at 25 mg against H37Rv and bg 206. At the same time, standard drug (Rifampicin) exhibited MIC at 0.125 mg against H37Rv strain.

Discussion

The present study demonstrated the in vitro and in vivo biological activities of *Dodonaea viscosa* flowers. The liver, kidney, and spleen are important parts of our body and are involved in different pivotal functions. The liver is one of the most important organs and played an important role in the detoxification of toxins^{49,50}. CCl_4 has been used to illuminate some mechanisms regarding various toxicities, i.e., lipid peroxidation, and cause necrosis, fibrosis, and apoptosis of cells².

 CCL_4 is the toxin well known to produce chemical stimulated liver injurywhen it get metabolized into trichloromethyl radical (CCl_3). This radical damages important cellular process by altering lipid metabolism and quantities of protein and then also induce mutations and produce hepatocellular carcinoma (HCC). Moreover, CCl_3 oxygenation resulted in trichloromethylperoxy radicals (CCl_3OO) thatlead to lipid peroxidation, polyunsaturated fatty acidsdestruction and lowered the permeability of the cellular membrane, and cause hepatic damage that is distinguished by fibrosis, cirrhosis, HCC, and Inflammation⁵¹.

Mice weight was reduced with the variation in weight of organs (Liver, Kidney, and Spleen). An increase in body weight owing to *Dodonaea viscosa* exhibited its protective effect. *Dodonaea viscosa* showed innocuous and protective to the mice, as reported earlier⁵². Level of liver enzymes ALT, ALP, AST, and direct bilirubin aimed to determine the sternness of damagedtissue⁵³. Biochemical markers get altered by carbon tetrachloride and then restored by the treatment with plant extracts and standard drug Silymarin, indicating the usage of *Dodonaea viscosa* flowers against liver injury. CCL₄ has been reported to increaseliver enzyme levels in some biochemical studies^{54,55}. When there areunstablechangesobserved in the ALP level, it causes liver diseases⁵⁶. Harmful changes in liver enzymes reflected several conditions likethe development of tissue necrosis, the decline in liver capacity (biosynthetic and catabolic), and alter normal structure of hepatocytes⁵⁷.

Hematological parameters such as RBCs, WBCs, and platelets were also disturbed by CCL_4 administration. Methanol and chloroform extracts of DV flower revealed positive effects on hematological parameters, which specify its suitability for managing blood cell disorders⁵⁸. An endogenous enzyme CAT, SOD, and GPx involved in scavenging free radicals and declining normalenzyme levels indicate hepatic damage⁵⁸. CCL_4 reduces the level of antioxidant enzymes (CAT, SOD, and GPx) and total protein compared with the normal group and confirmsliver injury⁵⁹, whereas the above factors were reinstated after administrating plant extracts. Intoxication of CCL_4 in antioxidant enzymes can be improved using Medicinal plants^{60,61}. Natural products have been investigated for the source of antioxidants that are being used for the hepato-protective activity. For the management of different diseases, flavonoids rich plants have shown protective effects decreasing serum markers with antioxidant and anti-inflammatory processes⁶².

Exposure to CCL_4 leads to liver damage such as necrosis, fibrosis, and central vein alteration. In the kidney, it caused renal fibrosis, glomerular and tubular changes, while in the spleen, deterioration in white and red pulp occurred^{63–65}. As a result of CCL_4 toxicity, the cellular structure and function of the kidney rely on the functional state of the liver⁶⁶. Similar studies reveal that at high doses of extracts, liver, kidney, and spleen seemed nearly normal, with no observable gross morphological and histopathological modifications, supporting present findings⁶⁷. Studies signify the use of *Dodonaea viscosa* against toxicity inan animal model and found to have revealed a protective effect for liver enzymes and attenuated the injury by diminishing the production of reactive oxygen species in hepatocytes⁶⁸.

Methanol solvent was selected for further activities based on the best results in the in-vivo study compared to chloroform. Preliminary screening of secondary metabolites resulted ina significant amount of total phenolic compounds and total flavonoid content. Phenolics and Flavonoids are considered singlet oxygen quenchers, radical scavengers, reducing agents, and hydrogen donors⁶⁹. So, the analysis of the plant's total flavonoids and phenolic compounds is important to measure its antioxidant capacity. The results of the experiment presented strong antioxidant activities of D. viscosa flowers. The highest antioxidant activity of DVM was shown against hydrogen peroxide assay. DVM manifested great radical scavenging ability as follows Hydroxyl radical assay > iron chelating assay > DDPH assay > Reducing power assay > ABTS radical assay > Superoxide assay. In the current study, the reducing capacity of D. viscosa significantly decreases the complex of ferric cyanide to ferrous. The occurrence of antioxidants was determined by evaluating the ability of plant extract to form ferrous by reducing the ferric cyanide complex⁷⁰. Reducing the power of plant compounds specifies its potential antioxidant capacity. High reducing power in a sample has a great ability to donate the electron and free radicals and produce stable elements by accepting donated electrons, which terminates the free radical reaction⁷¹. Hydroxyl radicals are highly reactive free radicals in biological systems, and there are no specific enzymes present in humans to protect against them. Their presence in the human body causes oxidative DNA damage. Therefore, there is a need for a solution to scavenge ROS with natural products having scavenging activity. Due to the high reactivity of OH radicals, the antioxidant activity of scavenging hydroxyl radicals is important^{72,73}. The most commonly used method for the evaluation of antioxidants is the DDPH assay. The quenching of DPPH measurement relies on the discoloration of the purple-colored 2,2-diphenyl-2-picryl-hydroxyl compound by antioxidant. Donor antioxidant decolorized DPPH radical by electron acceptance and measured quantitatively from variations in absorbance⁷⁴. Furthermore, D. viscosa expressed significant radical scavenging activity against ABTS assay with a low value of IC_{50} . All the essays are positive as well as significantly correlated with phenols and flavonoids.

HPLC quantified four compounds in DVM, i.e., gallic acid, epicatechin, quercetin, and cumeric acid. Quercetin is an iso flavonoid and flavonoid content (rutin and quercetin) was identified in the stem of dodonaea viscosa. The remedial aptitudes of *Dodonaea viscosa*are associated using pharmacological effects brought through the synergistic action of numerous constituents, i.e., flavonoids, saponins, di, and triterpenes, along with a combination of phenolics existing in the plant⁷⁵. Flavonoids and diterpenoids are the richest secondary metabolites that werepreviously identified and isolated from Dodonaea⁷⁶. These phenolic and flavonoid compounds revealed anticancer, antiallergic, antibacterial, antiviral, and anti-inflammatory activities⁷⁷. The chemical compounds elucidated by GCMS were Oleic acid (Octadecenoic acid), Ascorbic acid, Ricinoleic acid, Stearic acid, Carboxylic acid, Cyclopentanone, and Cedrol. Fatty acids (Oleic, linoleic and linolenic acids) enriched food showed pleiotropic effects and used for the management of inflammation, hypertension, cardiovascular diseases, hyperlipidemia, reproductive ailments, immune system, and aggregation of platelets^{78,79}. Research studies showed that Oleic acid exerts remedial effects on the human body, such as cancer, anti-inflammatory and autoimmune diseases, anda vital role in wound healing⁸⁰. Ricinoleic acid is a significant unsaturated and hydroxylated fatty acid that depicts antipathogenic activity by deterring bacteria, viruses, mold, and yeast⁸¹. DVM showed very good activity against Tuberculosis strains. Mycobacterium tuberculosis is responsible for tuberculosis, which is among the fatal diseases. Dodonaea Viscosa has been locally used in traditional medicines to treat tuberculosis^{82,83}. Tested plant extract of DV flowers exhibited stronger resistance from all tested strains of Mycobacterium tuberculosis owing to the occurrence of bioactive components among the different concentrations of plant methanol extract that are probably anti-mycobacterial metabolites. Tuberculosis remains accountable for numerous mortalities around the world. During treatment, TB patients require extensive chemical analysis and eventually generate antagonistic effects on patient wellbeing. To diminish the use of unnatural resistant drugs, medicinally important plants contribute to great sureness as a potential reason for bioactive anti-mycobacterial metabolites⁸⁴. A limited distinct species of genus Dodonaea have been extensively examined both chemically and pharmacologically. The most known species of genus Dodonaea is D. viscosa in literature⁸⁵.

Conclusion

Dodonaea viscosa is well-known plant species and widely possesses so many biological activities. Results showed the potential pharmacological effect of *Dodonaea Viscosa* against acute toxicity in albino mice, specifying its use against different diseases, most of all liver diseases. This plant showed significant biological activities such as antioxidant and anti-tuberculosis. The chemical composition of the plant is rich in antioxidant compounds, flavonoids, and phenols, and a rich source of Fatty acids, mainly oleic acid. These compounds could probably protect elevated hepatic enzymes caused by carbon tetrachloride and chronic tuberculosis. These curative effects are linked with the traditional use of this plant against different diseases. This plant might be used to extract promising drugs for the management of liver and multiple organs injury. The active compounds and their action mechanism, pharmacokinetics, toxicology, efficacy, and molecular mechanisms still need to be explored to attain integration into remedial practice.

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Author contributions

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Competing interests

The authors declare no competing interests.

Additional information

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