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Implication of O₂ dynamics for both N₂O and CH₄ emissions from soil during biological soil disinfestation

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Soil O₂ dynamics have significant influences on greenhouse gas emissions during soil management practice. In this study, we deployed O₂-specific planar optodes to visualize spatiotemporal distribution of O₂ in soils treated with biological soil disinfestation (BSD). This study aimed to reveal the role of anoxia development on emissions of N₂O and CH₄ from soil amended with crop residues during BSD period. The incorporation of crop residues includes wheat straw only, wheat straw with biochar and early straw incorporation. The anoxia in soil developed very fast within 3 days, while the O₂ in headspace decreased much slower and it became anaerobic after 5 days, which was significantly affected by straw and biochar additions. The N₂O emissions were positively correlated with soil hypoxic fraction. The CH₄ emissions were not significant until the anoxia dominated in both soil and headspace. The co-application of biochar with straw delayed the anoxia development and extended the hypoxic area in soil, resulting in lower emissions of N₂O and CH₄. Those results highlight that the soil O₂ dynamic was the key variable triggering the N₂O and CH₄ productions. Therefore, detailed information of soil O₂ availability could be highly beneficial for optimizing the strategies of organic amendments incorporation in the BSD technique.

The emissions of greenhouse gases (methane and nitrous oxide) from soils have become a global concern, because methane (CH₄) is the largest non-carbon dioxide (CO₂) climate forcing agent¹ and nitrous oxide (N₂O) is the most significant ozone-depleting gas in the atmosphere². Agricultural soils contributed to over 47% and 65% of anthropogenic CH₄ and N₂O emissions respectively^{3,4}. Although the main production processes of CH₄ and N₂O in soils are known to be the microbial driven methanogenesis and nitrification/denitrification^{5,6}, it still remain challenging to pinpoint the key biological mechanisms and the interactions among environmental variables on CH₄ and N₂O production.

One of the crucial factors determining GHG emission is the availability of soil O₂, which regulated the production and consumption of CH₄ and N₂O by influencing the soil redox conditions. Soil O₂ availability is the net result of the O₂ consuming and diffusing processes, which is controlled primarily by the microbial substrates and gas diffusivities, such as the bioavailable carbon (C) and nitrogen (N) together with soil moisture and pore structures. Despite the crucial role of O₂ in regulating the processes of CH₄ and N₂O production, very few studies have quantitatively investigated the effects of O₂ on CH₄ and N₂O emissions in soils and how such effects are affected by the complex interactions between soil and management factors⁷. Particularly, soil management practices could easily alter soil O₂ availability^{8,9}. For instance, biological soil disinfestation (BSD), which is a common practice for controlling soil-borne diseases in a variety of crops^{10,11}, would facilitate fast consumption of O₂, due to the involvement of the incorporating organic C enriched sources (such as crop residues) into the soil. Furthermore, it greatly reduces the O₂ diffusion, as it tarps the soil with a plastic mulching film, then irrigating soil to saturation. The O₂ depletion in soil under the BSD treatment would be mainly determined by the degradability of added organic amendments¹². The soil O₂ could additionally come from the diffusion of the O₂ in the narrow headspace between soil surface and mulching film, which would be depleted as well. The induced O₂ depletion and/or the accumulation of organic acids produced from the C source decomposition during BSD treatment are suppressive or toxic for several soil-borne pests and plant pathogens^{13,14}. The soil environment, temporally shifting from aerobic to anaerobic conditions, was a key factor shaping microbial activities^{14,15} consequently influencing the

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formation of the C and N gaseous products^{16,17}. However, the dynamics of soil O₂ contents during the entire BSD period have not been investigated in detail at all. Currently, there is a critical gap in our understanding of how BSD-induced soil O₂ variation impact GHG emissions. Besides, BSD has been adopted as a reliable remediation method but high N₂O emission would occur in nitrate-riched soils during BSD treatment¹⁸. The application of biochar could possibly mitigate N₂O emissions from agricultural soil through the increase of soil pH¹⁹. However, the biochar's mitigation effects on N₂O emission during BSD process has not been fully investigated yet.

In this study we investigated the dynamics of the soil O₂, the emissions of CH₄ and N₂O during the entire BSD treatment period. The specific objectives were: (1) to investigate the effects of the crop residue incorporation on soil O₂ dynamics during BSD treatment period; (2) to link the BSD induced soil O₂ variation to the emissions of CH₄ and N₂O from soil. The different methods of crop residue incorporation included: wheat straw only, wheat straw with biochar, and early straw incorporation.

Materials and methods

Experimental sites and treatments. The experiment was conducted in a greenhouse at Shunyi experimental station, China Agricultural University, Beijing, China. Four treatments had been established: soil without any amendments was served as control (CON); soil amended with wheat straw (WH), soil amended with wheat straw 5 days before the BSD treatment (WH + 5D), soil amended with wheat straw and biochar (WH + BC). The four treatments were arranged in a randomized complete block design with 3 replicates.

Straw berry was the dominant crops planted in the greenhouse. Soil was collected from the 0–20 cm layer after harvest of catch crop (maize). The soil was freshly passed through a 2-mm mesh sieve. 8.5 kg of soil (dry matter basis) was repacked into a PVC column (diameter of 20 cm and length of 21 cm) at a bulk density of 1.35 g cm⁻³. A nylon mesh (0.2-mm pore size) was used to cover the bottom of the column, assisting the soil repacking process and column transportations. The application rate of wheat straw and biochar was 10 g kg⁻¹ and 5 g kg⁻¹ soil respectively, corresponding to rates of 1.62 × 10⁴ kg ha⁻¹ and 0.81 × 10⁴ kg ha⁻¹ respectively. The added wheat straw was grounded to be less than 2 mm in length. A mixture substrate of rice husks (70%) and cotton seed hulls (30%) was pyrolyzed at 400 °C for 4 h in a sealed oven to produce the biochar in this experiment. The properties of the soil, straw and biochar were shown in Table S1.

All of the repacked soil columns were placed in a shallow water tank for 6 h, conditioned to reach the moisture content of water-holding capacity. A polyethylene film (0.04-mm thick) was used as the mulch film for soil columns, sealed with silicon glass to make it airtight. There were always some gaps or headspaces between the mulching film and soils because of the uneven soil surface in situ. Therefore, we kept the top 1.0 cm in each column as the headspace with a volume of 0.31 L. Then soil columns were transported to the greenhouse and buried there to keep the soil surface level the same as the ground level.

Soil sampling. Soil samples were taken at 10 and 30 days with a core sampler (diameter of 2.0-cm). Additional samples were taken at day 0 for treatment WH + 5D. 10 g of each fresh soil sample was used for gravimetric water content determination (drying at 105 °C for 24 h). Another 10 g soil was extracted with 50 mL 1 M potassium chloride (KCl) solution for 1 h and filtered through Whatman no.1 paper. Extracts were frozen at -20 °C and analyzed later by a Flow Injection Analyser (FIAstar 5000 Analyzer, FOSS, Denmark) for NH₄⁺ and NO₃⁻ analysis. The dissolved organic C (DOC) and dissolved organic N (DON) in soil were measured after mixing soil with deionized water (1:5 soil water ratio), shaking for 1 h and then the mixture was filtered through a 0.45 μm microporous film and determined by TOC/TN analyzer (LiquiTOCII, Elementar, Germany).

Gas sampling. Gas samples were taken from the headspace of each soil column using gas-tight syringes at 0, 0.5, 1, 2, 3, 4, 6, 9, 12, 17, 23 and 30 days for CH₄, CO₂ and N₂O analysis. For each sampling event, 10 mL samples were taken at 0 and 30 min after the start of sampling. All gas samples were analyzed using a gas chromatograph (Clarus 690 GC, PerkinElmer, the United Kingdom) configured with two detectors: an electron capture detector (ECD) and a flame ionization detector (FID). The gas sample was separated by a Porapak QS column (2.0-m). The ECD was set up for N₂O analysis, the temperature of the oven and detector were 50 and 350 °C respectively, and argon was used as the carrier gas. The FID channel was set up for CH₄ analysis. CO₂ was reduced to CH₄ by a methanization module and then was detected by FID as well. The temperature of the oven and detector were 50 and 370 °C respectively and the carrier gas was helium.

The temperature of soil (0–10 cm) and surface air was measured with a multi-channel soil thermometer (SYS-21G, SAIYASI, China). The relative humidity levels in the air of the greenhouse was monitored with a hydro-thermometer (TES-1360, HUANYU, China).

Soil O₂ monitoring. A parallel experiment was conducted in laboratory to assess the dynamics of soil O₂ using O₂-specific planar optodes at 36 °C, which was the average temperature of the soil during BSD treatment. The designed treatments were the same as those in the greenhouse. Soil was incubated in rectangular boxes (height × width × length = 21 cm × 12 cm × 5 cm) suitable for planar optode system. The measuring system of planar optode was adapted from Larsen et al.²⁰. Briefly, Pt(II)-tetrakis (pentafluorophenyl) porphyrin (PtTFPP) was used as an O₂ quenchable luminophore, which was combined with the coumarin dye MACROLEX fluorescence yellow 10GN as an antenna dye. The platinum (II) complex (1% by weight) and the antenna dye (2%) were dissolved in a 10% polystyrene matrix, where toluene was used as the solvent. The mixed solution was then spinning coated onto the 2-mm thick glass inserts (15 × 10 cm), which was fitted into the center of the front window of the acrylic box. A LED assembly was used as the excitation light source. It consisted of seven light-emitting diodes with λ-peak of 447.5 nm (SR-02-R0500, Luxeon Star, Canada). A short-pass filter (475 nm) (Genxu Optics, China) was equipped in front of the excitation light. A long-pass emission filter (Genxu Optics, China)

covered the high-quality prime macro lens (SIGMA 50 mm F2.8 EX DG MACRO) to remove any reflected blue light from the excitation source. The O₂ contents were quantified with the ratio between the red and green luminescence emitted from the indicator and antenna dye respectively. Optode images were taken by a Canon EOS 760D camera, at 30-min intervals for the initial 7 days, and 60-min intervals for the rest of experimental period. The O₂ levels calculated from the sensing area was used to evaluate the temporal patterns of O₂ variations in soil and headspace.

Permeation of CH₄, CO₂ and N₂O through the plastic mulching films. The plastic film used in the BSD was not completely impermeable for CH₄, CO₂ and N₂O. In order to estimate the full fluxes of gases, the permeation of the plastic film for gases was measured by additional laboratory experiments, through a modified method of Nishimura et al.²¹. Briefly, the penetration chamber consists of two 1.0 L cylindrical glass compartments, into which a plastic film is inserted. Round rubber rings on both sides of the film together with silicon glass were used to connect and seal the two compartments tightly. 10 mL of a standard gas including CH₄, CO₂ and N₂O (199 ppmv, 10,100 ppmv and 145 ppmv respectively) was injected into one compartment. After 1, 3, 6 and 12 h, 3.0 mL of the gas from both compartments was drawn respectively with a syringe and injected into the gas chromatograph (Clarus 690 GC, PerkinElmer, the United Kingdom) for gas quantification. Those permeation measurements were carried out in an incubator at 20, 30, 40, 50 and 60 °C under 100% of relative humidity. Three replicates under each temperature were conducted.

Data analysis. The ratio-metric approach was used for the O₂ optode measurement²⁰. The free software ImageJ (<http://rsbweb.nih.gov/ij/>) was used to process the recorded optode images. The O₂ contents in the soil and headspace were calculated from the optode images covering a soil area and headspace area of 10 × 10 cm and 10 × 1 cm respectively, with a unit of % air saturation.

The gas permeation through plastic film was temperature-dependent, which can be considered as an activated energy process and can be well expressed by an Arrhenius-type model, therefore the CH₄, CO₂ and N₂O fluxes to the atmosphere by permeation was then estimated using a modified equation from Stern et al.²²:

$$F = -D(T) \frac{\Delta C}{\Delta x} = F_0 \cdot \exp\left(-\frac{E_a}{RT}\right) \cdot (p_{below} - p_{above}) \quad (1)$$

where F is the gas (CH₄, CO₂ or N₂O) flux through the film ($\mu\text{g m}^{-2} \text{h}^{-1}$), D is the diffusive permeability coefficient ($\text{m}^2 \text{h}^{-1}$) of the film (temperature-dependent), ΔC is the difference of gas concentrations (mg m^{-3}) above and below the film, and Δx is the thickness of the film (m). T is the absolute temperature (K). E_a is the activation energy (J mol^{-1}), R is the universal gas constant ($8.314 \text{ J mol}^{-1} \text{ K}^{-1}$), T is the absolute temperature (K), and F_0 is an Arrhenius constant ($\text{mg m}^{-2} \text{h}^{-1} \text{ atm}^{-1}$). The parameters E_a and F_0 were obtained by the above-mentioned permeation experiment. p_{below} and p_{above} were the partial pressure of the gas below the mulch film and the ambient gas partial pressure respectively. The ambient gas partial pressures above the film were assumed to be stable (because of the large volume in the greenhouse) during the experimental period, which were fixed to 0.17 Pa, 40.53 Pa, 0.04 Pa for CH₄, CO₂ and N₂O respectively, equivalent to 1700 ppbv, 400 ppmv and 400 ppbv for CH₄, CO₂ and N₂O respectively.

Global warming potential (GWP) was calculated to compare the effects of BSD treatment with different crop residue incorporation. GWP in CO₂-e per m² was estimated with a GWP of 28 for CH₄ and 265 for N₂O²³:

$$\text{GWP} = 28 \times R(\text{CH}_4) + R(\text{CO}_2) + 265 \times R(\text{N}_2\text{O}) \quad (2)$$

where $R(\text{CH}_4)$, $R(\text{CO}_2)$ and $R(\text{N}_2\text{O})$ are the cumulative emissions of CH₄, CO₂ and N₂O (mg m^{-2}) during BSD treatment period, respectively.

Statistic analysis. Differences in average soil O₂, headspace O₂, CH₄, CO₂ and N₂O concentrations between different treatments were analyzed by a one-way ANOVA procedure at $p < 0.05$. The Tukey test was used for multiple comparisons. Pearson's bivariate correlation implied the relationships between different measured variables.

Ethics approval and consent to participate. Not applicable.

Consent for publication. Not applicable.

Results

Climate conditions of soil and air in the greenhouse. During the initial 7 days, both temperatures and relative humidity of the air in the greenhouse sharply increased (Fig. 1). The peak temperature was reached at day 12 with 58.0 °C. It was kept in a quite stable condition with 2 degrees of fluctuation. After 10 days, the relative humidity was the highest with 96% until the end of the experimental period. The soil temperature gradually increased from 26 to 40 °C within 10 days, and then was maintained at 36 ± 3 °C.

Available N and C in soil. The BSD treatment had significant effects on N and C availabilities in soil (Fig. 2). Nitrate was consumed dramatically and depleted completely at the end of the BSD treatment. The addition of wheat straw accelerated this depletion process. Particularly, the early application of wheat straw consumed half of the native soil nitrate during the 5 days of aerobic treatment. Biochar addition coupled with wheat straw had the highest nitrate depletion rate during the first 10 days of BSD treatment.

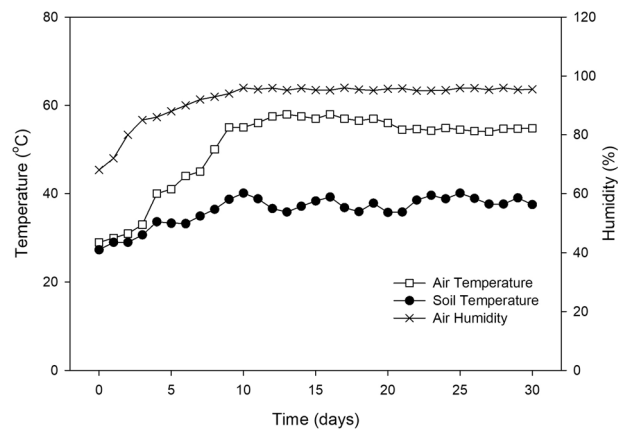


Figure 1. The temperature (soil and air) and relative humidity (air) in the greenhouse during the BSD treatment.

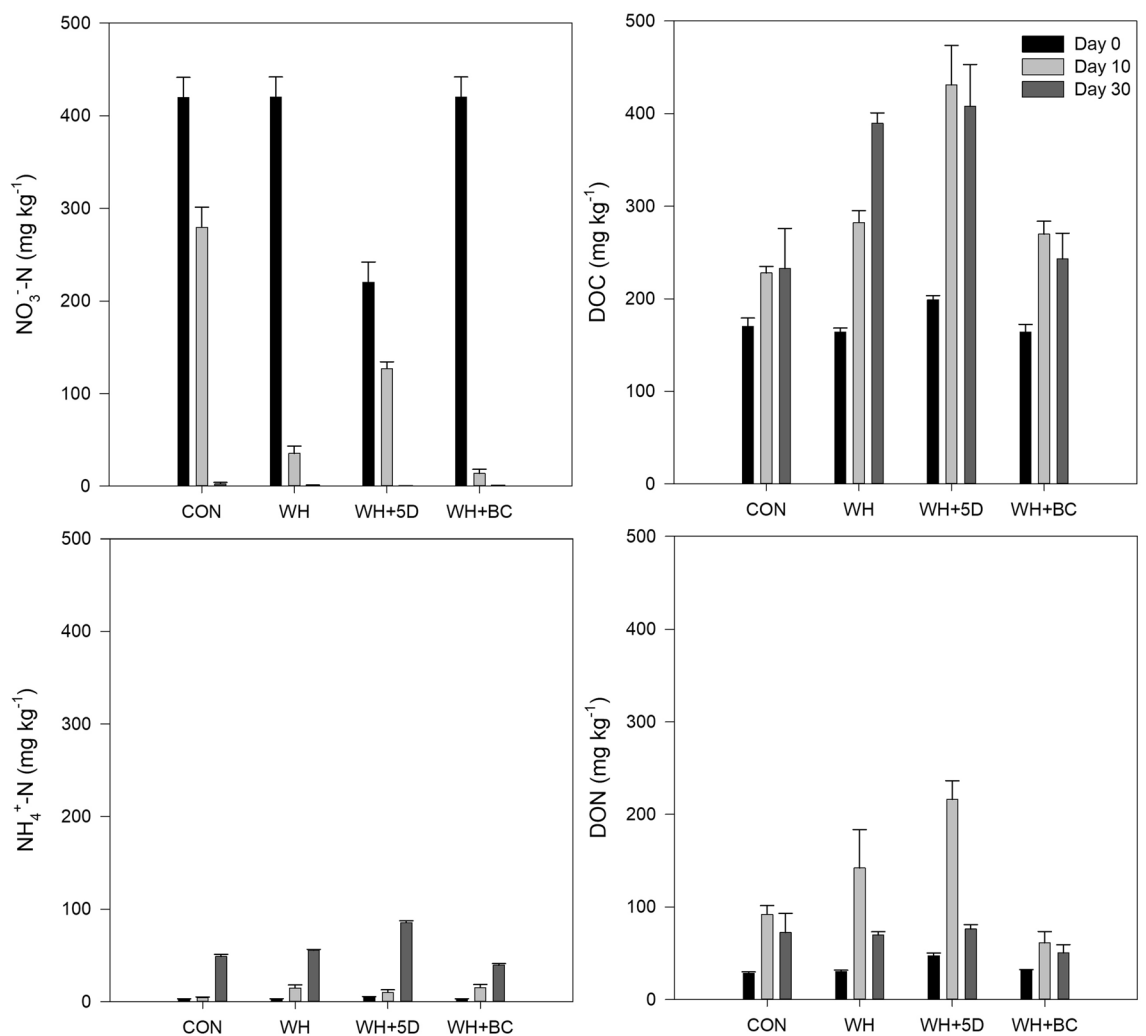


Figure 2. Nitrate, ammonium, DOC and DON contents in soil during the BSD treatment period. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

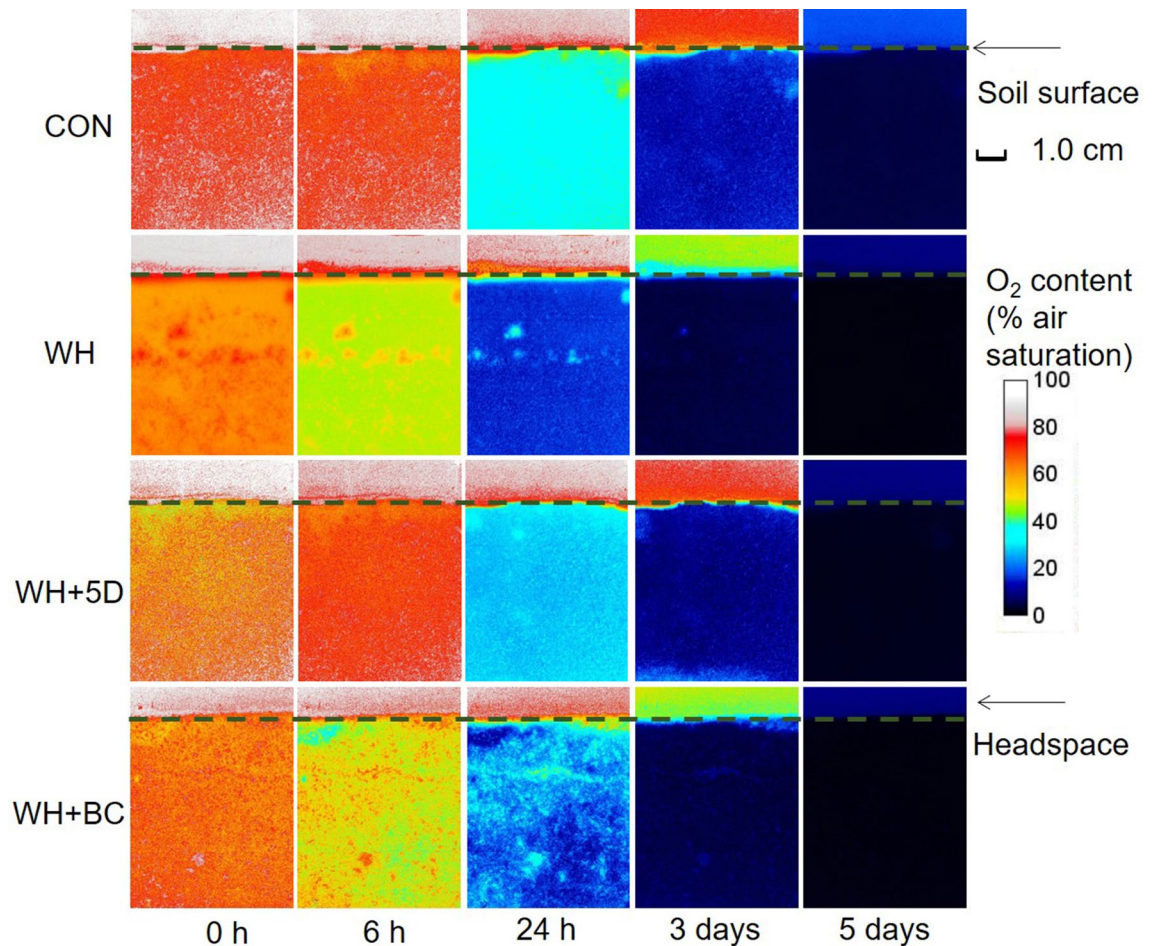


Figure 3. Selected images of O_2 content in soil and headspace after different straw applications. Images are one representative example of the three replicates. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

Contrast to nitrate dynamics, soil ammonium increased gradually, and the treatment of WH + 5D resulted in the highest ammonium content. The addition of wheat straw increased the ammonium accumulation rate during the first 10 days period, which, however, was not significant at the end of BSD treatment. Similarly, the WH + BC followed the same trend as the wheat straw-only addition in terms of ammonium nitrogen changes over the first ten days, but at the end WH + BC had the lowest ammonium nitrogen content among the four treatments. The DOC dynamics was quite similar to the ammonium, the only difference was that DOC accumulation mainly occurred in the first 10 days of BSD treatment.

Soil DON contents increased during the initial 10 days but declined at the end. Biochar addition coupled with wheat straw resulted in the lowest DON production.

Oxygen dynamics in soil and headspace. Generally, the O_2 contents dropped dramatically within 5 days in all treatments. Initially, the soil O_2 contents were 94% air saturation in CON (Fig. 3), slightly higher in straw-amended treatments (around 90% air saturation). This indicated that a small proportion of soil O_2 had already been consumed in straw-amended treatments during the setting up period (about 60 min). After 3 days, the mean O_2 contents in soil decreased sharply to less than 10% air saturation in all treatments. On average, the O_2 decreasing rates increased as the following order: CON < WH + 5D < WH + BC < WH, which indicated the significant influences of amendments on soil O_2 consumption. The O_2 contents in headspace decreased much slower than that in soil. Until 5 days later, the headspace became anoxic (less than 3% air saturation). The crop residue incorporation had similar effects on the O_2 contents in headspace and soil: WH treatment resulted in the highest O_2 consumption rate in headspace.

The levels of soil O_2 contents could be divided into three categories based on physiological considerations: oxic conditions where the O_2 content is $> 2.00 \text{ mg L}^{-1}$, hypoxic with an O_2 content between 0.14 and 2.00 mg L^{-1} , and anoxic conditions with O_2 content $< 0.14 \text{ mg L}^{-1}$. Briefly, an O_2 content below 2.00 mg L^{-1} influences many bio-organisms in terms of behaviour, growth and reproduction²⁴; The reduction of elements such as N and Fe, would dominate with anaerobic metabolic process under the conditions of O_2 content below 0.14 mg L^{-1} , with further O_2 decline ($< 0.14 \text{ mg L}^{-1}$), methanogenesis may occur as well²⁶. Using the obtained optode images, the

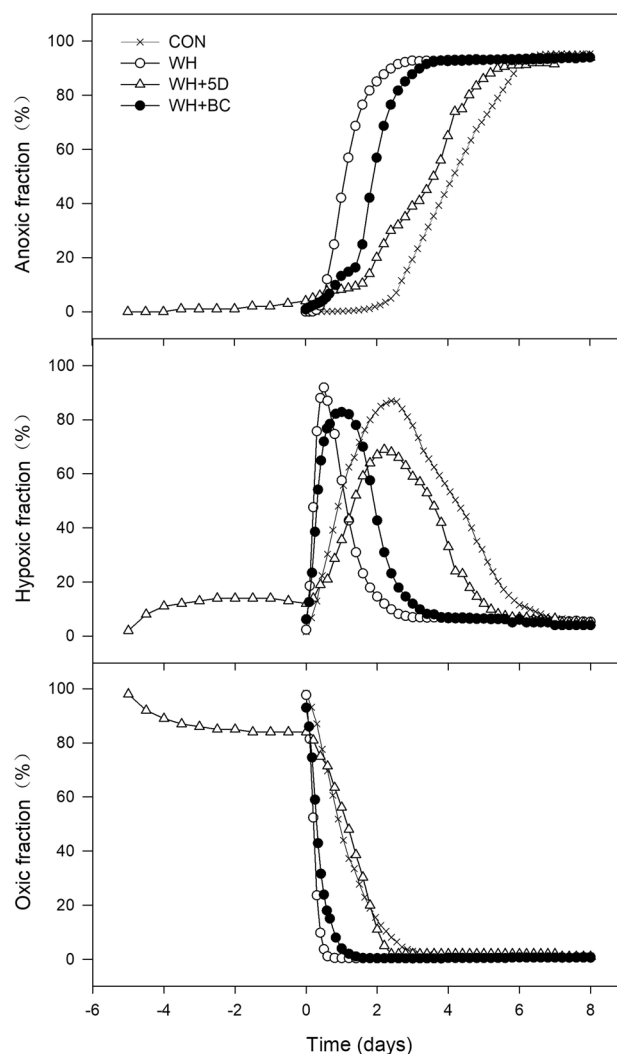


Figure 4. The area fractions of anoxic, hypoxic and oxic conditions (<0.14 , 0.14 – 2.00 , and >2.00 mg O_2 L^{-1} respectively) in the soil during the BSD treatment period. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

fractions of anoxic, hypoxic and oxic conditions in all treatments could be calculated. Anoxic area developed the slowest in CON during the entire BSD period (Fig. 4). The WH treatment exhibited the most rapid development of anoxia. The expansion of anoxic area in WH + BC was slightly slower than in WH + 5D initially, however, after 20 h, it became much greater in WH + BC than WH + 5D. The peak fractions of hypoxic area occurred the earliest in WH (0.5 day), followed with WH + BC (1 day) and WH + 5D (2 days). The decline of the oxic fraction in WH was faster than in others and remained lower during the BSD period. Before BSD process, the oxic fraction decreased by 20% in WH + 5D during the initial 5 days of incubation. Once BSD process began, the declining rate of oxic fraction in WH + 5D increased extensively.

The permeation of CH_4 , CO_2 and N_2O fluxes through mulch film. As the temperature increased from 20 °C to 60 °C, the gas permeabilities of the mulch film increased exponentially (Fig. 5), which was fitted well with Arrhenius models (Table S2). N_2O was characterized by the highest gas transmission, which was 9.4 times and 1.3 times as high as that of CH_4 and CO_2 respectively. The estimated activation energy was the highest from N_2O as well. The CH_4 , CO_2 and N_2O fluxes by permeation through mulch film were not negligible. In the following sections, the emission rates of CH_4 , CO_2 and N_2O from all treatments at every sampling time were corrected with the permeance of respective gases. In addition, by the trapezoidal integration of the estimated flux data during the experiment, we calculated the cumulative emission of each gas by permeation through the mulch film, which accounted for 2.5 – 2.7%, 12.3 – 14.2% and 7.4 – 15.0% of total cumulative emissions of CH_4 , CO_2 and N_2O respectively (Fig. 7).

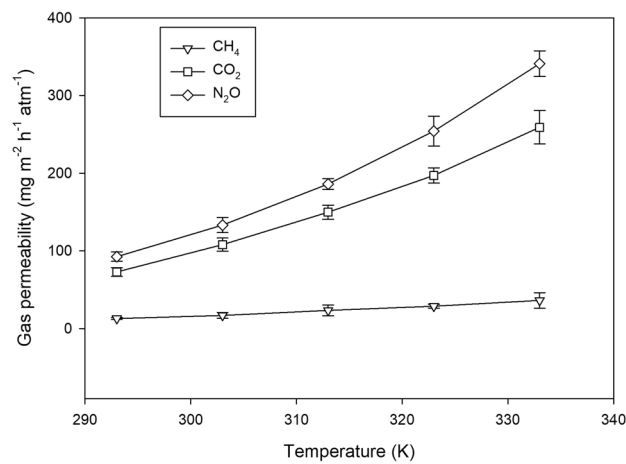


Figure 5. Gas permeability of plastic mulch films as affected by temperature.

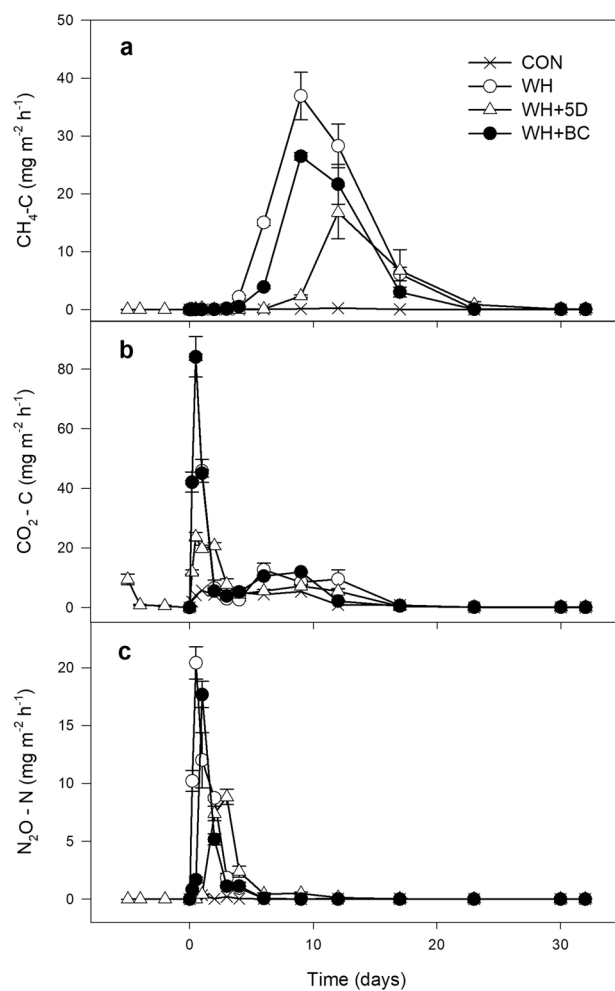


Figure 6. Dynamics of CH₄, CO₂ and N₂O emissions during BSD treatment period. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

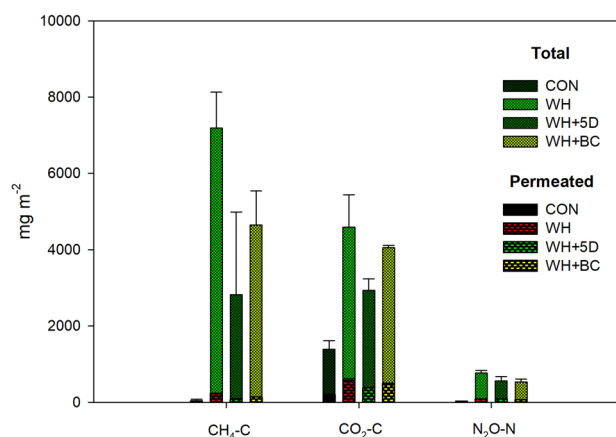


Figure 7. The cumulative emissions (mg m^{-2}) of CH_4 , CO_2 and N_2O from different amendments during BSD treatment period. Bars with dash-filled pattern were the cumulative emissions of gases permeated through the cover film. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

CH₄ emissions. The temporal patterns of CH_4 emissions were significantly different among treatments (Fig. 6a). Generally, CH_4 emissions from all treatments were negligible in the first 3 days, and CON maintained low fluctuating emission rates during the entire experimental period. After 4 days, the emission rates in WH treatment were significantly higher than the other treatments and the highest emission peak was observed after 9 days. The temporal pattern of CH_4 emission from treatment WH + BC was quite similar as that from WH, but with significantly lower emission rates. The WH + 5D treatment induced the latest emission peak (at day 12) as well as the lowest peak emission rate. The correlation analysis revealed that CH_4 emission rates was positively correlated with anoxic fractions of soil (Table S3). The cumulative CH_4 emission during the entire BSD treatment was the largest from the WH treatment, followed by WH + BC and WH + 5D treatment (Fig. 7).

CO₂ emissions. The emission rates of CO_2 increased dramatically at the beginning of the experiment and reached the peaks within 24 h for all treatments (Fig. 6b). The emissions peaked after 12 h in both treatment of WH and WH + BC, followed by a sharp decrease afterwards. The highest emission rate in treatment of WH + 5D was observed after 12 h but followed with significantly slower decreasing rates. Except for day 2 and day 3, CO_2 emission rates from WH + 5D were consistently lower than those of WH and WH + BC during the BSD treatment period. At day 4, the air temperature in the greenhouse changed sharply from 35 °C to 50 °C, consequently, the CO_2 emission rates from all treatments increased slightly. As expected, CO_2 emission rates was observed a negative correlation with soil anoxic fractions, while hypoxic fraction of soil was positively correlated with CO_2 emission rates (Table S3). The cumulative CO_2 emission was the highest from the WH treatment and intermediate with the WH + BC treatment, and the lowest from the WH + 5D treatment (Fig. 7).

N₂O emissions. N_2O emissions were influenced by the different amending methods of wheat residue in the soil (Fig. 6c). The N_2O emission rates increased extensively and reached to the emission peaks within 3 days in straw-amended treatments. The CON had rather low peak emission rate. The earliest and highest emission peak occurred after 12 h from the WH treatment, followed by the WH + BC (peak after 24 h) and WH + 5D treatments (peak after 72 h). The emissions decreased dramatically in treatments of WH and WH + BC after 2 days, while the WH + 5D treatment was observed with relatively longer period of peak emission (about 4 days). Significant correlation (positive) was detected between soil hypoxic fraction and N_2O emission rates (Table S3). The cumulative N_2O emissions for the entire BSD treatment period from the WH treatment was the largest, which was 36.5% and 43.2% higher than that from the WH + BC and WH + 5D treatment respectively (Fig. 7).

Global warming potential. The GWP was the largest in treatment of WH, which was 1.81 and 1.50 times as high as that in treatments of WH + 5D and WH + BC, respectively (Fig. 8). The CO_2 emissions contributed to only 1.8–2.5% of the GWP in all treatments. The contributions of each gas to the GWP varied among treatments. The N_2O emissions (52.5%) contributed slightly more compared to CH_4 (45.5%) in treatments of WH, which was the same pattern in treatment of WH + BC. However, the proportion of CH_4 emissions became much less (29.6%) and N_2O emissions contributed the main part of GWP (68.6%) in treatment of WH + 5D.

Discussions

Effects of crop residue incorporation on O₂ dynamics during BSD process. The depletion of soil O₂ contents was significantly influenced by the application of the BSD treatment. The formation of anaerobic conditions during this treatment had regulating impacts on its efficacy²⁷. Deploying optode technology enables monitoring the dynamics of anoxia development both in the soil and headspace during BSD process. Addition of organic amendments coupled with sufficient irrigation simultaneously enhanced microbial O₂ consumption

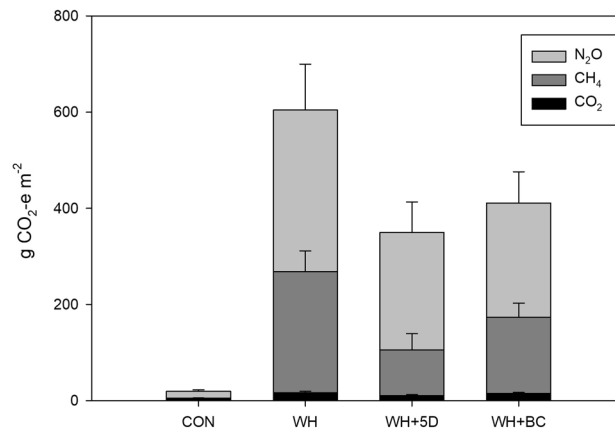


Figure 8. The GWP (g CO₂-e m⁻² soil) from different amendments during BSD treatment period. CON: soil without any amendments; WH: soil amended with wheat straw; WH + 5D: soil amended with wheat straw 5 days before the BSD treatment; WH + BC: soil amended with wheat straw and biochar.

and physical inhibition of O₂ diffusion²⁸. The completely anoxic bulk soil environment developed after 3–5 days. Although the soil was supposed to be tightly covered by the mulching film, there were always some spaces in situ between the film and soil due to the rough soil surface. Such headspace is quite important as a O₂ source in the initial stage of BSD process. The O₂ could diffuse from the headspace air into the soil, and the diffusion rates were dependent on the soil gas diffusivity and the gradients of O₂ contents between headspace and soil. The crop residue addition could possibly improve the gas diffusivity of soil and increase the soil O₂ consumption as well^{29,30}, therefore, it took the longest period (more than 7 days) to reach the anoxic condition in the headspace and soil at the CON treatment. It was worth noting that there were still 1–2% air saturation of O₂ in headspace and 1–4% of hypoxic area in all soils after the O₂ status was relatively stable. The O₂ in the air could possibly permeate through the covered plastic film³¹ and diffuse into the headspace as well as the soils. Such slow O₂ permeation could be one of the sources for the soil O₂. In addition, some tortuous pores in soil may trap some gases which were not accessible to microbes³², and those pores may be maintained with limited level of oxygen contents and kept with slightly hypoxic conditions.

Generally, the main O₂-consuming processes in soil are aerobic respiration and nitrification. Soil aerobic respiration is the mineralization of organic C, usually associated with the decomposition of native or added organic matters in soil. Both ammonium (NH₄⁺) and nitrite (NO₂⁻) oxidation in nitrification would consume O₂. In this study, the soil NH₄⁺ content was very low, and nitrification was not the main contributor to the O₂ consumption. Hence, the anoxia development was more closely linked with the organic matter availability³³. The addition of C-rich straw enhanced the soil microbial respiration and stimulated a quick consumption of O₂ in the bulk soil. Additionally, the water absorption by straw could be another reason that affected O₂ diffusion and caused soil hypoxia²⁹. The biochar addition resulted in slightly slower anoxia development than straw treatment, this could be partly due to the biochar adsorption capability of dissolved organic C³⁴, which may reduce the availability of organic matter, consequently slower down the O₂ consumption. Additionally, the porous biochar amendments may have impacts on O₂ diffusion processes through the influences on the soil pore structure³⁵. Both the pores inside biochar and the pores created between biochar particles and soil particles (or the amended straw) may play significant roles in soil pore network, which are very important for gas transport in soils, therefore the biochar addition may enhance the O₂ diffusions into the soil³⁶. Early straw incorporation could allow sufficient time for the mineralization of easily degradable organic matter³⁷ under oxic conditions before the BSD process. Additionally, the assisted O₂ diffusion in the soil by early straw incorporation was indicated by the limited fractions of anoxic area during the early 5 days before BSD (Fig. 3). Furthermore, the early addition of straw may enhance the formation of soil aggregates³⁸ that would increase physical protection of organic matter once BSD process began. Therefore, the O₂ consumption was relatively slower in the treatment that the straw was incorporated earlier before BSD process.

Linking O₂ dynamics to GHG emissions. The dynamics of greenhouse gas emissions during the BSD process have been rarely reported. Some studies measured gaseous emission on the top of covered film¹⁷, which could only indicate the gas fluxes by permeation through the mulch film, and may underestimate the production and emissions of those gases. The present study has, for the first time, monitored the dynamics of direct emissions of CO₂, N₂O and CH₄ during the entire BSD process, and has estimated the significant contribution (2–15%) of CO₂, N₂O and CH₄ fluxes by permeation through the mulch film. The high risk of N₂O and CH₄ fluxes were the major concern for the BSD application in vegetable soils.

N₂O emissions In such high nitrate accumulated soils, denitrification is not limited by the N substrate, and is considered as the major process for N₂O production in all treatments once the hypoxic fraction prevailed in soil (Fig. 4). Right after BSD process was initiated, straw application enhanced the anoxia development, which, coupled with high nitrate contents in soil, were favorable conditions for the denitrification process^{39,40}, mainly due to the unlimited N substrate and extra C supply (electron donors as energy source)⁴¹. Particularly, the soil

O₂ contents decreased to a certain level and enhanced N₂O emissions from the straw amended treatments. Soil O₂ availability in all treatments was identified as the main factor responsible for the initiation of both N₂O and CH₄ emissions. Zhu et al.⁴² reported that N₂O emission accumulated significantly for all soils tested at hypoxic fraction reaching 30% of the soil. In their study, N₂O emissions were low when hypoxic fraction was higher than 50% or less than 10%. Since those studies were performed using soils amended with animal manure, which had varied degradability of C and N, the dynamics of anoxia development in soils differed with different amendments and soil properties. Interestingly, the peak emissions of N₂O and peak fractions of hypoxic area were synergized quite well in all treatments at the present study. After the hypoxic area decreased and anoxia developed further, the N₂O emissions declined, probably due to the further reduction of N₂O to N₂ induced by more strict anaerobic conditions.

During the initial period of BSD process, both N₂O and CO₂ fluxes increased sharply with the straw incorporation, showing a significant relationship between respiration and denitrification rates^{43,44}. The emission peaks of N₂O from straw amended treatments lasted for 1–3 days, and the peak emission rates were much higher than that of other studies¹⁷. This difference was partly due to much higher contents of NO₃⁻ and application rates of straw in the present study. A significant proportion of the straw would be degraded in well-aerated soil within 5 days²⁹, thus the earlier incorporation of straw decreased the available C, furthermore, it enhanced the reduction of soil nitrate (Fig. 2), therefore reduce the availability of substrate for denitrification and N₂O formation as well. Aside from the availability of nitrate and carbon sources, other factors modulating microbial N₂O production may include soil temperature, soil moisture and pH^{45,46}. Soil temperature directly impacts microorganism activity, soil aeration, substrate availability and redistribution. Previous studies reported that N₂O emissions were synchronized with surface soil temperature¹⁷. The increases of temperature in this study would possibly boost N₂O emission rates and cumulative emissions from the soil with relatively high water content⁴⁷.

Biochar's mitigating effects on N₂O emissions was still significant under the extreme conditions: high temperature (36 °C) and high water content (close to saturation) in anaerobic soils. There are several possible explanations. One of the mechanisms may relate to the changes of O₂ availability in the soil, caused by increased gas diffusivity after amending porous biochar^{48,49}. It is also partially due to that biochar reduced both DOC and DON availability for microorganisms (Fig. 2), which depends on the net effects of C sorption by biochar^{50,51} and C release from soil amendments⁵². Finally, pH changes and the toxic effects of compounds formed during pyrolysis could affect microbial activity after biochar addition¹⁹.

CH₄ emissions Variations of biochemical and microbial parameters can cause different CH₄ fluxes by influencing CH₄ production and oxidation processes⁵³. One of the key parameters is the soil O₂ availability⁵⁴. Strictly anaerobic conditions are required for methanogens to produce CH₄, due to that the key enzymes in methanogenesis are likely inactivated by the presence of O₂, additionally, methanogens are poor competitors for shared substrates in soils⁵⁵. Therefore, the CH₄ emissions were not significant until the anoxic fractions of the soil dominated. The hypoxia in agricultural soils usually favored the methanotrophs, which is the major contributor for CH₄ oxidation⁵⁶, thus CH₄ emission rates increased dramatically once the hypoxic fraction declined to the minimum.

Wheat straw incorporation significantly stimulated CH₄ emission, which agrees with previous studies^{57,58}. Soil CH₄ emissions are primarily regulated by the availability of C substrates in soil, and wheat straw as an organic material to soil could provide the available C sources to support the growth of methanogenic populations under anaerobic conditions⁵⁹, especially in the later stage of BSD process with very limited soil O₂ availability.

The incorporation time of straw had significant impacts on CH₄ productions and the following emissions as well. As reported by other studies^{58,60}, CH₄ emissions were correlated positively with dissolved organic carbon (DOC), as methanogens are often limited by DOC⁶¹. In the treatment of WH + 5D, the straw was incorporated 5 days earlier in the soil. Part of DOC contents would be consumed by microbial activities under oxic conditions, resulting in decreased DOC, consequently there were lower CH₄ emissions from this treatment. The co-application of biochar significantly reduced CH₄ emissions by 35.4% compared to the treatment of straw addition during the entire BSD process (Fig. 5a). Several studies had reported that the addition of biochar may attenuate the methanogenic activity and improve methanotrophic gene abundance and potential activity^{62–64}. Additionally, the oxygen diffusions into the soil may be possibly enhanced by the biochar addition³⁶, therefore the biochar amendment could make the soil favorable for methanotrophs but unfavorable for methanogens, consequently led to the declined CH₄ emissions under BSD conditions.

Conclusions

The work demonstrated that straw incorporation strategies determined the soil O₂ dynamics during the BSD treatment. Such amendment-dependent soil O₂ variations had close linkage with greenhouse gas emissions from vegetable soils. The soil amendment enriched in C-source coupled with the anaerobic conditions created during the BSD treatment may facilitate denitrification and methanogenesis, thereby increasing N₂O and CH₄ emissions. Incorporating the straw 5 days before the BSD process could reduce the C availability in soil by oxic degradation and decrease the content of available N by immobilization, therefore, both N₂O and CH₄ emissions could be mitigated. The co-application of biochar with straw could reduce the emissions of N₂O and CH₄, possibly by extending the hypoxic fractions of soil during BSD process. The results from this study could offer new insights for developing sensitive approaches using soil O₂ dynamics to predict or mitigate CH₄ and N₂O emission during the application of BSD technique.

Data availability

The datasets used and/or analysed during the current study are available from the corresponding author on reasonable request. All data generated or analysed during this study are included in this published article [and its supplementary information files].

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References

- Dlugokencky, E. J., Nisbet, E. G., Fisher, R. & Lowry, D. Global atmospheric methane: budget, changes and dangers. *Philos. Trans. R. Soc. Math. Phys. Eng. Sci.* **369**, 2058–2072. <https://doi.org/10.1098/rsta.2010.0341> (2011).
- Ravishankara, A. R., Daniel, J. S. & Portmann, R. W. Nitrous oxide (N₂O): the dominant ozone-depleting substance emitted in the 21st century. *Science* **326**, 123–125. <https://doi.org/10.1126/science.1176985> (2009).
- Solomon, S., Qin, D., Manning, M., Chen, Z., Marquis, M., Averyt, K.B., Tignor, M., Miller, H.L., 2007. Climate change 2007: the Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change. Summary for Policymakers.
- Reay, D. S. *et al.* Global agriculture and nitrous oxide emissions. *Nat. Climate Change* **2**, 410–416. <https://doi.org/10.1038/nclimate1458> (2012).
- Butterbach-Bahl, K., Baggs, E. M., Dannenmann, M., Kiese, R. & Zechmeister-Boltenstern, S. Nitrous oxide emissions from soils: how well do we understand the processes and their controls?. *Philos. Trans. R. Soc. London B Biol. Sci.* **368**, 2177 (2013).
- Oertel, C., Matschullat, J., Zurba, K., Zimmermann, F. & Erasmii, S. Greenhouse gas emissions from soils—a review. *Geochemistry* **76**, 327–352. <https://doi.org/10.1016/j.chemer.2016.04.002> (2016).
- Song, X., Ju, X., Topp, C. F. E. & Rees, R. M. Oxygen regulates nitrous oxide production directly in agricultural soils. *Environ. Sci. Technol.* **53**, 12539–12547. <https://doi.org/10.1021/acs.est.9b03089> (2019).
- Bollmann, A. & Conrad, R. Influence of O₂ availability on NO and N₂O release by nitrification and denitrification in soils. *Glob. Change Biol.* **4**, 387–396. <https://doi.org/10.1046/j.1365-2486.1998.00161.x> (1998).
- Markfoged, R., Nielsen, L. P., Nyord, T., Ottosen, L. D. M. & Revsbech, N. P. Transient N₂O accumulation and emission caused by O₂ depletion in soil after liquid manure injection. *Eur. J. Soil Sci.* **62**, 541–550. <https://doi.org/10.1111/j.1365-2389.2010.01345.x> (2011).
- Shennan, C., Muramoto, J., Lamers, J., Mazzola, M., Roskopf, E.N., Kokalis-Buelle, N., Momma, N., Butler, D.M., Kobara, Y., 2014. Anaerobic soil disinfestation for soil borne disease control in strawberry and vegetable systems: current knowledge and future directions. International Society for Horticultural Science (ISHS), Leuven, Belgium, pp. 165–175. <https://doi.org/10.17660/ActaHortic.2014.1044.20>
- Roskopf, E. N. *et al.* Anaerobic soil disinfestation and soilborne pest management. In *Organic amendments and soil suppressiveness in plant disease management* (eds Meghvansi, M. K. & Varma, A.) 277–305 (Springer, Cham, 2015).
- Christel, W. *et al.* Spatiotemporal dynamics of phosphorus release, oxygen consumption and greenhouse gas emissions after localised soil amendment with organic fertilisers. *Sci. Total Environ.* **554–555**, 119–129. <https://doi.org/10.1016/j.scitotenv.2016.02.152> (2016).
- Oka, Y. Mechanisms of nematode suppression by organic soil amendments—a review. *Appl. Soil. Ecol.* **44**, 101–115. <https://doi.org/10.1016/j.apsoil.2009.11.003> (2010).
- van Agtmaal, M. *et al.* Legacy effects of anaerobic soil disinfestation on soil bacterial community composition and production of pathogen-suppressing volatiles. *Front. Microbiol.* <https://doi.org/10.3389/fmicb.2015.00701> (2015).
- Mowlick, S. *et al.* Development of anaerobic bacterial community consisted of diverse clostridial species during biological soil disinfestation amended with plant biomass. *Soil Sci. Plant Nutr.* **58**, 273–287. <https://doi.org/10.1080/00380768.2012.682045> (2012).
- Achmon, Y. *et al.* Assessment of tomato and wine processing solid wastes as soil amendments for biosolarization. *Waste Manage.* **48**, 156–164. <https://doi.org/10.1016/j.wasman.2015.10.022> (2016).
- Di Gioia, F. *et al.* Anaerobic soil disinfestation impact on soil nutrients dynamics and nitrous oxide emissions in fresh-market tomato. *Agr. Ecosyst. Environ.* **240**, 194–205. <https://doi.org/10.1016/j.agee.2017.02.025> (2017).
- Maeda, M., Kayano, E., Fujiwara, T., Nagare, H. & Akao, S. Nitrous oxide emissions during biological soil disinfestation with different organic matter and plastic mulch films in laboratory-scale tests. *Environ. Technol.* **37**(4), 432–438 (2016).
- Cayuela, M. L. *et al.* Biochar's role in mitigating soil nitrous oxide emissions: a review and meta-analysis. *Agr. Ecosyst. Environ.* **191**, 5–16. <https://doi.org/10.1016/j.agee.2013.10.009> (2014).
- Larsen, M., Borisov, S. M., Grunwald, B., Klimant, I. & Glud, R. N. A simple and inexpensive high resolution color ratiometric planar optode imaging approach: application to oxygen and pH sensing. *Limnol. Oceanogr. Methods* **9**, 348–360. <https://doi.org/10.4319/lom.2011.9.348> (2011).
- Nishimura, S., Komada, M., Takebe, M., Yonemura, S. & Kato, N. Nitrous oxide evolved from soil covered with plastic mulch film in horticultural field. *Biol. Fertil. Soils* **48**, 787–795. <https://doi.org/10.1007/s00374-012-0672-7> (2012).
- Stern, S. A., Krishnakumar, B., Nadakatti, S. M. & Mark, J. E. *Physical properties of polymers handbook* (API Press, 1996).
- IPCC, 2014. Core Writing Team, R.K. Pachauri, L.A. Meyer (Eds.), Climate Change 2014: Synthesis Report. Contribution of Working Groups I, II and III to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change, IPCC, Geneva, Switzerland.
- Levin, L. A. *et al.* Effects of natural and human-induced hypoxia on coastal benthos. *Biogeosciences* **6**, 2063–2098. <https://doi.org/10.5194/bg-6-2063-2009> (2009).
- Naqvi, S. W. A. *et al.* Marine hypoxia/anoxia as a source of CH₄ and N₂O. *Biogeosciences* **7**, 2159–2190. <https://doi.org/10.5194/bg-7-2159-2010> (2010).
- Keiluweit, M., Nico, P. S., Kleber, M. & Fendorf, S. Are oxygen limitations under recognized regulators of organic carbon turnover in upland soils?. *Biogeochemistry* **127**, 157–171. <https://doi.org/10.1007/s10533-015-0180-6> (2016).
- Shrestha, U., Auge, R. M. & Butler, D. M. A meta-analysis of the impact of anaerobic soil disinfestation on pest suppression and yield of horticultural crops. *Front. Plant Sci.* <https://doi.org/10.3389/fpls.2016.01254> (2016).
- Liptzin, D., Silver, W. L. & Detto, M. Temporal dynamics in soil oxygen and greenhouse gases in two humid tropical forests. *Ecosystems* **14**, 171–182. <https://doi.org/10.1007/s10021-010-9402-x> (2011).
- Kravchenko, A. N. *et al.* Hotspots of soil N₂O emission enhanced through water absorption by plant residue. *Nat. Geosci.* **10**, 496–500. <https://doi.org/10.1038/ngeo2963> (2017).
- Kutlu, T., Guber, A. K., Rivers, M. L. & Kravchenko, A. N. Moisture absorption by plant residue in soil. *Geoderma* **316**, 47–55. <https://doi.org/10.1016/j.geoderma.2017.11.043> (2018).
- Stefan, W. *et al.* Mechanisms of oxygen permeation through plastic films and barrier coatings. *J. Phys. D Appl. Phys.* **50**, 425301 (2017).
- Rubol, S., Dutta, T. & Rocchini, D. 2D visualization captures the local heterogeneity of oxidative metabolism across soils from diverse land-use. *Sci. Total Environ.* **572**, 713–723. <https://doi.org/10.1016/j.scitotenv.2016.06.252> (2016).

33. Keiluweit, M., Gee, K., Denney, A. & Fendorf, S. Anoxic microsites in upland soils dominantly controlled by clay content. *Soil Biol. Biochem.* **118**, 42–50. <https://doi.org/10.1016/j.soilbio.2017.12.002> (2018).
34. Kasozi, G. N., Zimmerman, A. R., Nkedi-Kizza, P. & Gao, B. Catechol and Humic Acid Sorption onto a Range of Laboratory-Produced Black Carbons (Biochars). *Environ. Sci. Technol.* **44**, 6189–6195. <https://doi.org/10.1021/es1014423> (2010).
35. Sun, Z., Arthur, E., de Jonge, L. W., Elsgaard, L. & Moldrup, P. Pore structure characteristics after 2 years of Biochar application to a sandy loam field. *Soil Sci.* **180**, 41–46. <https://doi.org/10.1097/ss.0000000000000111> (2015).
36. Obia, A., Mulder, J., Hale, S. E., Nurida, N. L. & Cornelissen, G. The potential of biochar in improving drainage, aeration and maize yields in heavy clay soils. *PLoS ONE* **13**, e0196794. <https://doi.org/10.1371/journal.pone.0196794> (2018).
37. Htun, Y. M., Tong, Y., Gao, P. & Xiaotang, J. Coupled effects of straw and nitrogen management on N₂O and CH₄ emissions of rainfed agriculture in Northwest China. *Atmos. Environ.* **157**, 156–166. <https://doi.org/10.1016/j.atmosenv.2017.03.014> (2017).
38. Zhao, H. *et al.* Effect of straw return mode on soil aggregation and aggregate carbon content in an annual maize-wheat double cropping system. *Soil Tillage Res.* **175**, 178–186. <https://doi.org/10.1016/j.still.2017.09.012> (2018).
39. Baggs, E. M. Soil microbial sources of nitrous oxide: recent advances in knowledge, emerging challenges and future direction. *Curr. Opin. Environ. Sustain.* **3**, 321–327. <https://doi.org/10.1016/j.cosust.2011.08.011> (2011).
40. Senbayram, M. *et al.* Interaction of straw amendment and soil NO₃⁻ content controls fungal denitrification and denitrification product stoichiometry in a sandy soil. *Soil Biol. Biochem.* **126**, 204–212. <https://doi.org/10.1016/j.soilbio.2018.09.005> (2018).
41. Giles, M. E., Daniell, T. J. & Baggs, E. M. Compound driven differences in N₂ and N₂O emission from soil; the role of substrate use efficiency and the microbial community. *Soil Biol. Biochem.* **106**, 90–98. <https://doi.org/10.1016/j.soilbio.2016.11.028> (2017).
42. Zhu, K., Bruun, S., Larsen, M., Glud, R. N. & Jensen, L. S. Heterogeneity of O₂ dynamics in soil amended with animal manure and implications for greenhouse gas emissions. *Soil Biol. Biochem.* **84**, 96–106. <https://doi.org/10.1016/j.soilbio.2015.02.012> (2015).
43. Miller, M. N. *et al.* Crop residue influence on denitrification, N₂O emissions and denitrifier community abundance in soil. *Soil Biol. Biochem.* **40**, 2553–2562. <https://doi.org/10.1016/j.soilbio.2008.06.024> (2008).
44. Xiao, Y. *et al.* Influence of winter crop residue and nitrogen form on greenhouse gas emissions from acidic paddy soil. *Eur. J. Soil Biol.* **85**, 23–29. <https://doi.org/10.1016/j.ejsobi.2017.10.004> (2018).
45. Schaufler, G. *et al.* Greenhouse gas emissions from European soils under different land use: effects of soil moisture and temperature. *Eur. J. Soil Sci.* **61**, 683–696. <https://doi.org/10.1111/j.1365-2389.2010.01277.x> (2010).
46. Luo, G. J., Kiese, R., Wolf, B. & Butterbach-Bahl, K. Effects of soil temperature and moisture on methane uptake and nitrous oxide emissions across three different ecosystem types. *Biogeosciences* **10**, 3205–3219. <https://doi.org/10.5194/bg-10-3205-2013> (2013).
47. Wang, C. *et al.* Factors related with CH₄ and N₂O emissions from a paddy field: clues for management implications. *PLoS ONE* <https://doi.org/10.1371/journal.pone.0169254> (2017).
48. Singh, B., Singh, B. P. & Cowie, A. L. Characterisation and evaluation of biochars for their application as a soil amendment. *Aust. J. Soil Res.* <https://doi.org/10.1071/SR10058> (2010).
49. Zhang, A. *et al.* Effect of biochar amendment on yield and methane and nitrous oxide emissions from a rice paddy from Tai Lake plain, China. *Agr. Ecosyst. Environ.* **139**, 469–475. <https://doi.org/10.1016/j.agee.2010.09.003> (2010).
50. Miura, A. *et al.* Characteristics of the adsorption of dissolved organic matter by charcoals carbonized at different temperatures. *Japan Agric. Res. Quarterly JARQ* **41**, 211–217. <https://doi.org/10.6090/jarq.41.211> (2007).
51. Liang, B. *et al.* Black carbon affects the cycling of non-black carbon in soil. *Org. Geochem.* **41**, 206–213. <https://doi.org/10.1016/j.orggeochem.2009.09.007> (2010).
52. Major, J., Lehmann, J., Rondon, M. & Goodale, C. Fate of soil-applied black carbon: downward migration, leaching and soil respiration. *Glob. Change Biol.* **16**, 1366–1379. <https://doi.org/10.1111/j.1365-2486.2009.02044.x> (2010).
53. Hernandez, M. E., Beck, D. A. C., Lidstrom, M. E. & Chistoserdova, L. Oxygen availability is a major factor in determining the composition of microbial communities involved in methane oxidation. *PeerJ* <https://doi.org/10.7717/peerj.801> (2015).
54. Shukla, P. N., Pandey, K. D. & Mishra, V. K. Environmental determinants of soil methane oxidation and methanotrophs. *Crit. Rev. Environ. Sci. Technol.* **43**, 1945–2011. <https://doi.org/10.1080/10643389.2012.672053> (2013).
55. Cedervall, P. E., Dey, M., Pearson, A. R., Ragsdale, S. W. & Wilmot, C. M. Structural insight into methyl-coenzyme M reductase chemistry using coenzyme B analogues. *Biochemistry* **49**, 7683–7693. <https://doi.org/10.1021/bi100458d> (2010).
56. Walkiewicz, A., Brzezińska, M. & Bieganski, A. Methanotrophs are favored under hypoxia in ammonium-fertilized soils. *Biol. Fertil. Soils* **54**, 861–870. <https://doi.org/10.1007/s00374-018-1302-9> (2018).
57. Hou, P. *et al.* Methane emissions from rice fields under continuous straw return in the middle-lower reaches of the Yangtze River. *J. Environ. Sci.* **25**, 1874–1881. [https://doi.org/10.1016/S1001-0742\(12\)60273-3](https://doi.org/10.1016/S1001-0742(12)60273-3) (2013).
58. Hu, N. *et al.* Effects of different straw returning modes on greenhouse gas emissions and crop yields in a rice-wheat rotation system. *Agr. Ecosyst. Environ.* **223**, 115–122. <https://doi.org/10.1016/j.agee.2016.02.027> (2016).
59. Sander, B. O., Samson, M. & Buresh, R. J. Methane and nitrous oxide emissions from flooded rice fields as affected by water and straw management between rice crops. *Geoderma* **235–236**, 355–362. <https://doi.org/10.1016/j.geoderma.2014.07.020> (2014).
60. Zhang, L. *et al.* Integrative effects of soil tillage and straw management on crop yields and greenhouse gas emissions in a rice-wheat cropping system. *Eur. J. Agron.* **63**, 47–54. <https://doi.org/10.1016/j.eja.2014.11.005> (2015).
61. Bossio, D. A., Horwath, W. R., Muters, R. G. & van Kessel, C. Methane pool and flux dynamics in a rice field following straw incorporation. *Soil Biol. Biochem.* **31**, 1313–1322. [https://doi.org/10.1016/S0038-0717\(99\)00050-4](https://doi.org/10.1016/S0038-0717(99)00050-4) (1999).
62. Dong, D. *et al.* Responses of methane emissions and rice yield to applications of biochar and straw in a paddy field. *J. Soils Sediments* **13**, 1450–1460. <https://doi.org/10.1007/s11368-013-0732-0> (2013).
63. Han, X. *et al.* Mitigating methane emission from paddy soil with rice-straw biochar amendment under projected climate change. *Sci. Rep.* **6**, 24731. <https://doi.org/10.1038/srep24731> (2016).
64. Yang, S. *et al.* Biochar improved rice yield and mitigated CH₄ and N₂O emissions from paddy field under controlled irrigation in the Taihu Lake Region of China. *Atmos. Environ.* **200**, 69–77. <https://doi.org/10.1016/j.atmosenv.2018.12.003> (2019).

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C.W.: Methodology, Lab analysis, Writing- Original draft preparation. X.M.: Methodology, Lab analysis. G.W.: Supervision, Writing- Reviewing and Editing. G.L.: Conceptualization, Writing- Reviewing and Editing. K.Z.: Conceptualization, Methodology, Writing and Reviewing.

Competing interests

The authors declare no competing interests.

Additional information

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