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## Nitrogen isotopic signatures and fluxes of N<sub>2</sub>O in response to land-use change on naturally occurring saline–alkaline soil

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The conversion of natural grassland to semi-natural or artificial ecosystems is a large-scale land-use change (LUC) commonly occurring to saline–alkaline land. Conversion of natural to artificial ecosystems, with addition of anthropogenic nitrogen (N) fertilizer, influences N availability in the soil that may result in higher N<sub>2</sub>O emission along with depletion of <sup>15</sup>N, while converting from natural to semi-natural the influence may be small. So, this study assesses the impact of LUC on N<sub>2</sub>O emission and <sup>15</sup>N in N<sub>2</sub>O emitted from naturally occurring saline–alkaline soil when changing from natural grassland (*Phragmites australis*) to semi-natural [*Tamarix chinensis* (Tamarix)] and to cropland (*Gossypium* spp.). The grassland and Tamarix ecosystems were not subject to any management practice, while the cropland received fertilizer and irrigation. Overall, median N<sub>2</sub>O flux was significantly different among the ecosystems with the highest from the cropland (25.3 N<sub>2</sub>O-N μg m<sup>-2</sup> h<sup>-1</sup>), intermediate (8.2 N<sub>2</sub>O-N μg m<sup>-2</sup> h<sup>-1</sup>) from the Tamarix and the lowest (4.0 N<sub>2</sub>O-N μg m<sup>-2</sup> h<sup>-1</sup>) from the grassland ecosystem. The <sup>15</sup>N isotopic signatures in N<sub>2</sub>O emitted from the soil were also significantly affected by the LUC with more depleted from cropland (– 25.3 ‰) and less depleted from grassland (– 0.18 ‰). Our results suggested that the conversion of native saline–alkaline grassland with low N to Tamarix or cropland is likely to result in increased soil N<sub>2</sub>O emission and also contributes significantly to the depletion of the <sup>15</sup>N in atmospheric N<sub>2</sub>O, and the contribution of anthropogenic N addition was found more significant than any other processes.

Nitrous oxide (N<sub>2</sub>O) is a major long-lived anthropogenic greenhouse gas with about 265–298 fold greater potential for global warming in the atmosphere compared to carbon dioxide<sup>1</sup>. It is also an ozone-depleting substance<sup>2</sup>, produced mainly in the soil from nitrification and denitrification processes<sup>3</sup>. Its concentration in the atmosphere has increased to 331 ppb<sup>4</sup> from 270 ppb in the pre-industrial age<sup>5</sup>. This increase of N<sub>2</sub>O in the atmosphere is mainly attributable to rise in anthropogenic nitrogen (N) input to soil<sup>6,7</sup> and this anthropogenic N input to soil increases as more natural ecosystems are converted to croplands.

Soil salinity can influence N<sub>2</sub>O flux in different ways. An increase in salinity in a non-saline soil can increase<sup>8</sup> or have no effect on N<sub>2</sub>O emission<sup>9</sup>. Similarly, on naturally occurring saline soils, both decreases<sup>8</sup> and increases<sup>10</sup> in the N<sub>2</sub>O flux have been found in response to increase in the salinity. These results suggest an ambiguous role of salinity in N<sub>2</sub>O emission. Some meta-analyses<sup>11,12</sup> have reported that alkaline soil emits less N<sub>2</sub>O than neutral or acidic soil. In alkaline soil NH<sub>4</sub> may be converted to NH<sub>3</sub> and volatilize to the atmosphere whereas NH<sub>4</sub> is retained in acid soil, favoring N<sub>2</sub>O formation<sup>13</sup>. N loss from alkaline soil may be high in total, but if much of the N is lost in the form of NH<sub>3</sub> there may be less NH<sub>4</sub> available for nitrification and subsequent denitrification. This evidence suggests that in naturally occurring saline–alkaline soil, the influence of both salinity and alkalinity may significantly affect the N<sub>2</sub>O formation processes. So, quantifying N<sub>2</sub>O flux from the saline–alkaline soil may help to increase knowledge on its contribution to soil–atmosphere exchange of N<sub>2</sub>O.

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Land-use change (LUC) from natural to semi-natural or artificial ecosystems can have different effects on N<sub>2</sub>O emission<sup>14–16</sup>. Specifically, conversion from natural to artificial ecosystems with the addition of N fertilizer significantly increases N<sub>2</sub>O emission while conversion to semi-natural may or may not increase the emission<sup>16–18</sup>. LUC directly impacts on soil physical, chemical and biological properties<sup>19,20</sup>, the main factors affecting N<sub>2</sub>O emission<sup>21,22</sup>. N<sub>2</sub>O emission from soil is reduced when pasture is forested<sup>14</sup>, while conversion of rainforest to pasture or plantation leads to an increase in N<sub>2</sub>O emission<sup>15</sup>. A recent study found that the conversion of a conventional agricultural field to bio-energy crops had no effect on N<sub>2</sub>O emission<sup>23</sup>. Therefore, knowing which LUC practice is appropriate in terms of lower N<sub>2</sub>O emissions, and its implementation could mitigate N<sub>2</sub>O emission to the atmosphere and associated impact of climate change. Moreover, various LUC practices<sup>14,15,23</sup> have different or no effect on N<sub>2</sub>O emission, indicating that LUC is rather an indirect cause of N<sub>2</sub>O emission. The main reason for the differences in N<sub>2</sub>O emission due to LUC is probably the alteration of the controlling factors of N<sub>2</sub>O production and reduction processes in the soil. So, quantifying N<sub>2</sub>O flux from LUC, along with soil physical and chemical parameters, would further enable understanding of the main driving factors for N<sub>2</sub>O production and consumption in the soil.

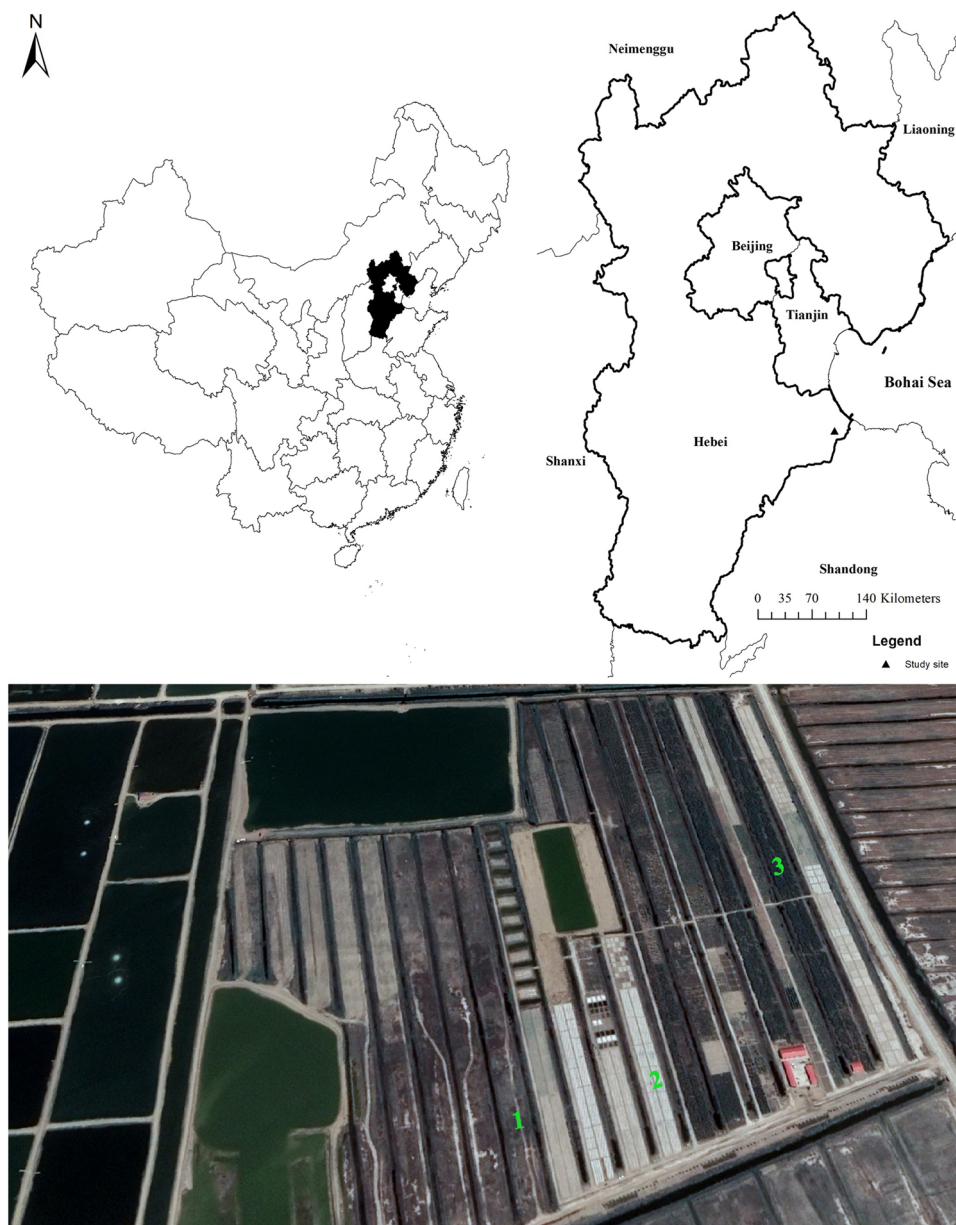
N has two stable isotopes i.e., <sup>14</sup>N and <sup>15</sup>N.  $\delta^{15}\text{N}$  of a sample is the deviation of the samples' <sup>15</sup>N/<sup>14</sup>N from the respective isotope ratio of the reference material<sup>24</sup>. Previously, the <sup>15</sup>N in N<sub>2</sub>O emitted from soil has been used to identify the processes for N<sub>2</sub>O production i.e. nitrification and denitrification; however, using only <sup>15</sup>N values in N<sub>2</sub>O may mislead the interpretation<sup>7</sup> as both the processes generally occur in the soils, possibly in different horizons or niche. The <sup>15</sup>N in N<sub>2</sub>O emitted from soil depends on the <sup>15</sup>N content of the substrates i.e. NH<sub>4</sub> and NO<sub>3</sub>, different microbial community composition, pH, temperature and substrate availability<sup>24,25</sup>. Although it is difficult to predict the sources of N<sub>2</sub>O emission using solely <sup>15</sup>N signatures in the N<sub>2</sub>O, these values could be used to distinguish between N<sub>2</sub>O emitted from natural and artificial ecosystems<sup>25</sup>. N addition in the artificial ecosystems increases the N availability which depletes the <sup>15</sup>N in N<sub>2</sub>O<sup>25</sup>. So, LUC from natural ecosystems to cropland may not only influence the N<sub>2</sub>O fluxes but also the <sup>15</sup>N in the emitted N<sub>2</sub>O as N availability is altered. For example, mean <sup>15</sup>N in N<sub>2</sub>O emitted from natural tropical forest, sub-tropical forest and subarctic soil are – 18.0‰, – 14.3‰, and – 13.0‰, respectively<sup>25–27</sup>; while more depleted after N application i.e., – 37.9‰<sup>28</sup> to – 34.3‰<sup>25</sup> in fertilized soil. The difference of <sup>15</sup>N in N<sub>2</sub>O is useful to distinguish N<sub>2</sub>O emitted between fertilized and natural soils, and it arises from anthropogenic N addition to soil<sup>25,28</sup>. Moreover, application of N fertilizer leads to high concentrations of NO<sub>3</sub> in the soil, resulting in a decrease in N<sub>2</sub>O reduction to N<sub>2</sub> and therefore a higher N<sub>2</sub>O to N<sub>2</sub> ratio from the denitrification process<sup>29</sup>. The reduction of N<sub>2</sub>O to N<sub>2</sub> through denitrification leads to 1–24‰ <sup>15</sup>N enrichment of the remaining N<sub>2</sub>O<sup>30</sup>. So, differences in the capacity to reduce N<sub>2</sub>O to N<sub>2</sub> between various ecosystems may also influence the <sup>15</sup>N in emitted N<sub>2</sub>O.

To feed the world's growing population requires an additional 2.7–4.9 Mha of cropland per year on average<sup>31</sup>. Due to limited land resources, natural saline–alkaline areas are being reclaimed for producing food<sup>32</sup>. Agricultural soil alone will contribute about 59% of total global N<sub>2</sub>O emissions by 2030<sup>3</sup> as fertilizer application will need to increase by about 35–60%<sup>33</sup>. Therefore, it is important to quantify, and develop measures to mitigate increases in N<sub>2</sub>O fluxes resulting from the conversion of natural saline–alkaline grassland to cropland. Furthermore, *Tamarix chinensis* (Tamarix), a salt tolerant native species of shrub, is commonly used for the restoration of saline–alkaline soil in coastal areas in China (semi-natural ecosystem)<sup>34</sup>. Local governments have launched a coastal ecological restoration program promoting the planting of Tamarix<sup>35</sup>; however, its effect on N<sub>2</sub>O emission is unknown. Though Zhang et al.<sup>36</sup> reported the differences in N<sub>2</sub>O emission from various natural vegetation in saline–alkaline coastal areas, the impact of LUC from natural to semi-natural or artificial ecosystems on the dynamics of N<sub>2</sub>O emissions from saline–alkaline soil is unknown. Moreover, different plant species have been reported to modify the soil characteristics in varying ways, resulting in significant changes in N<sub>2</sub>O fluxes<sup>37</sup>. Therefore, we hypothesize that: (1) LUC from native saline–alkaline ecosystem (grassland) to semi-natural (Tamarix) may significantly influence N<sub>2</sub>O flux, along with soil environmental variables (soil temperature, soil moisture, ammonium, nitrate), because of the difference in plant species but have no effect on the <sup>15</sup>N in N<sub>2</sub>O emitted from the soil because there is no addition of anthropogenic N and (2) LUC from native saline–alkaline ecosystem (grassland) to artificial (cropland) may influence both N<sub>2</sub>O flux and the <sup>15</sup>N in N<sub>2</sub>O due to anthropogenic N addition and changes in management practices. Therefore, we expect that the <sup>15</sup>N in emitted N<sub>2</sub>O could be used to distinguish N<sub>2</sub>O emitted between unfertilized (natural and semi-natural ecosystems) and fertilized (cropland) ecosystems but not between different unfertilized ecosystems (grassland and Tamarix).

## Methods

**Site description.** The study was carried out from April 2017 to June 2018 at the Haixing experimental station of the Center for Agricultural Resources Research (CARR), Institute of Genetics and Developmental Biology (IGDB), Chinese Academy of Sciences (CAS). This site is located near the Bohai sea in Haixing county (117°33'5" E, 38°09'59"N) of Hebei province, China (Fig. 1). The site has a semi-humid monsoon climate with more than 75% of precipitation occurring during the rainy season, i.e. from July to September. The mean annual precipitation is 582 mm. The groundwater table is at 0.9–1.5 m depth. The soil in this area is classified as solonchak (18.1% clay and 7.8% sand). The salt content in the area ranges from 3 to 20 g kg<sup>-1</sup> soil<sup>38</sup>.

In 2008, the native grassland was converted to Tamarix and cropland with the aim of reclamation of the saline–alkaline soil. The Tamarix stand was left to grow naturally after plantation. For this reason, we consider it as a semi-natural ecosystem. The cropland (artificial ecosystem) has permanent plots 7.25 m × 7.25 m in size, which were left fallow after conversion until 2014. During the fallow period, the cropland plots were irrigated (180 mm per year) around early January with saline groundwater. The irrigated water freezes from January to late February or early March as air temperatures are mostly below 0 °C. The salinity of the irrigated groundwater was 7–27 g l<sup>-1</sup><sup>38</sup>. This practice of irrigation reduces the salinity in the soil and decrease the salt stress on subsequently



**Figure 1.** Map and an aerial photo of the study site. 1, 2 and 3 in the aerial photo represent the positions of the grassland, cropland, and Tamarix, respectively. The map was created using ArcGIS (v10.3.1) (ESRI Inc.) and Google Earth.

planted cotton seedlings<sup>38</sup>. Since 2014, during each March, the cropland has been covered with plastic film until the sowing of the cotton to reduce the evapotranspiration<sup>38</sup>. The cropland received  $400 \text{ kg N ha}^{-1} \text{ year}^{-1}$  applied during May every year ( $200 \text{ kg N ha}^{-1}$  organic fertilizer +  $200 \text{ kg N ha}^{-1}$  diammonium phosphate) before sowing cotton since 2014. During this experimental period, cropland was fertilized on 7th May 2017 and 6th May 2018 and irrigation occurred on 10th Jan 2018. The irrigated water had melted completely by 21st Feb 2018. Other details of the three ecosystems are reported in Table 1.

**Gas sampling.** In each ecosystem, four closed static chambers were randomly placed. The chambers were made of polyvinyl chloride (PVC) and measured  $60 \times 20 \times 40 \text{ cm}$  (L  $\times$  B  $\times$  H) and each chamber contained a fan to homogenize the air. The chambers were fitted with a thermometer and a sampling tube with a three-way stop-cock. Both sampling tube and thermometer were sealed where they passed through the surface of the chamber to prevent leakage. Five 40-ml gas samples were taken for  $\text{N}_2\text{O}$  concentration analysis at 20 min intervals using a glass syringe, while two 160-ml gas samples were taken at 0 and 80 min and stored in glass bottles for  $\delta^{15}\text{N-N}_2\text{O}$  analysis. Gas was sampled between 8:00 AM to 12:00 PM. Sampling was done twice to thrice in a month during March to September (warm season) while once in a month during October to February (cold season).

S. no	Ecosystem	Management practice	Dominant plant species	Soil bulk density (g cm <sup>-3</sup> )	Soil pH	Soil salinity (mS cm <sup>-1</sup> )
1	Grassland	Native grassland, no grazing, no cutting, no fertilizer	Common reed ( <i>Phragmites australis</i> )	1.56 ± 0.04a	8.74 ± 0.07	2.29 ± 0.19
2	Cropland	Converted from the grassland, fertilizer use (organic + chemical), irrigation once a year (saline water)	Cotton ( <i>Gossypium</i> spp., Lumian 28)	1.58 ± 0.01a	8.45 ± 0.06	2.17 ± 0.29
3	Tamarix	Converted from the grassland, no fertilizer use, no cutting, no litter removal	Tamarix ( <i>Tamarix chinensis</i> )	1.38 ± 0.02b	8.58 ± 0.04	2.15 ± 0.10

**Table 1.** Management practices, dominant vegetation and some physical and chemical soil parameters of the three ecosystems. Different letters in the row indicate significant differences (< 0.05), while no letters means no difference.

**N<sub>2</sub>O concentration measurement, flux calculation, and <sup>15</sup>N isotope determination.** The concentration of N<sub>2</sub>O was measured using gas chromatography (Agilent GC-6820, Agilent Technologies Inc., Santa Clara, CA, USA) equipped with <sup>63</sup>Ni electron capture detection (ECD) in the laboratory of CARR, IGDB, CAS, Shijiazhuang, Hebei. The concentrations of N<sub>2</sub>O were calculated based on the measured peak areas relative to the peak areas measured from reference standards which were run twice before and after every fifteen gas samples.

The N<sub>2</sub>O flux was calculated using the following equation from Li et al.<sup>39</sup>.

$$F = M \times V \times A^{-1} \times \Delta C \times \Delta t^{-1} \times 273 \times (273 + T)^{-1} \times P \times (P^0)^{-1} \times 60 \quad (1)$$

where F is the N<sub>2</sub>O flux (μg N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup>), M is the molecular weight of N<sub>2</sub>O-N, V is the volume of the chamber (m<sup>3</sup>), A is the soil surface area occupied by the chamber base (m<sup>2</sup>), ΔC × Δt<sup>-1</sup> is the slope of N<sub>2</sub>O accumulation in the chamber with the time change (10<sup>-6</sup> min<sup>-1</sup>), T is the air temperature (°C) inside the chamber, P is the atmospheric pressure (hPa) on the sampling time and P<sup>0</sup> is standard atmospheric pressure.

Annual cumulative emission rate was calculated by interpolating the N<sub>2</sub>O flux from four replicate chambers during measured days and the interval between sampling days. While calculating annual emission rate, it was assumed that there was no emission of N<sub>2</sub>O from 10 Jan to 21 Feb, 2018 in the cropland because of the frozen irrigated water on the surface (up to 18 cm thickness which was shrinking when the temperature rising). This assumption might underestimate the annual cumulative emissions. However, for grassland and Tamarix the rate for the whole year was calculated.

The gas samples (160 ml) were passed through a chemical trap [NaOH + Mg(ClO<sub>4</sub>)<sub>2</sub>] (FINNIGAN PRECON) to remove CO<sub>2</sub> and H<sub>2</sub>O using a helium flow of 10–15 ml min<sup>-1</sup>. Using stainless steel trap, the gas sample was passed through liquid nitrogen. After this cryofocusing step, the gas sample passed into a GC (FINNIGAN GC). Finally, the δ<sup>15</sup>N of the N<sub>2</sub>O was measured using an Isotope Ratio Mass Spectrometer (IRMS) (Delta V Plus, Thermo Fisher, Germany). δ<sup>15</sup>N of data reported in this study are in unit of per mill (‰) relative to international standard (atmospheric N<sub>2</sub>). As the N<sub>2</sub>O in the sample represented the isotopic composition of both atmospheric and soil-emitted N<sub>2</sub>O, the following equation from Snider et al.<sup>40</sup> was used to calculate the δ<sup>15</sup>N of soil-emitted N<sub>2</sub>O.

$$\delta^{15}\text{N of soil - emitted N}_2\text{O} = (\delta^{15}\text{N}_{\text{measured}} \times C_{\text{N}_2\text{O}_{\text{measured}}} - \delta^{15}\text{N}_{\text{atmosphere}} \times C_{\text{N}_2\text{O}_{\text{atmosphere}}}) / (C_{\text{N}_2\text{O}_{\text{measured}}} - C_{\text{N}_2\text{O}_{\text{atmosphere}}}) \quad (2)$$

where δ<sup>15</sup>N<sub>measured</sub> and C<sub>N<sub>2</sub>O<sub>measured</sub></sub> are the δ<sup>15</sup>N and concentration of the N<sub>2</sub>O sample at time 80 min after the closure of the chamber, while the δ<sup>15</sup>N<sub>atmosphere</sub> and C<sub>N<sub>2</sub>O<sub>atmosphere</sub></sub> are the δ<sup>15</sup>N and concentration of the sample at time zero (immediately after the closure of the chamber). When the fluxes were lower than 10 N<sub>2</sub>O-N μg m<sup>-2</sup> h<sup>-1</sup>, the <sup>15</sup>N analyses were excluded from the results due to errors introduced with lower fluxes.

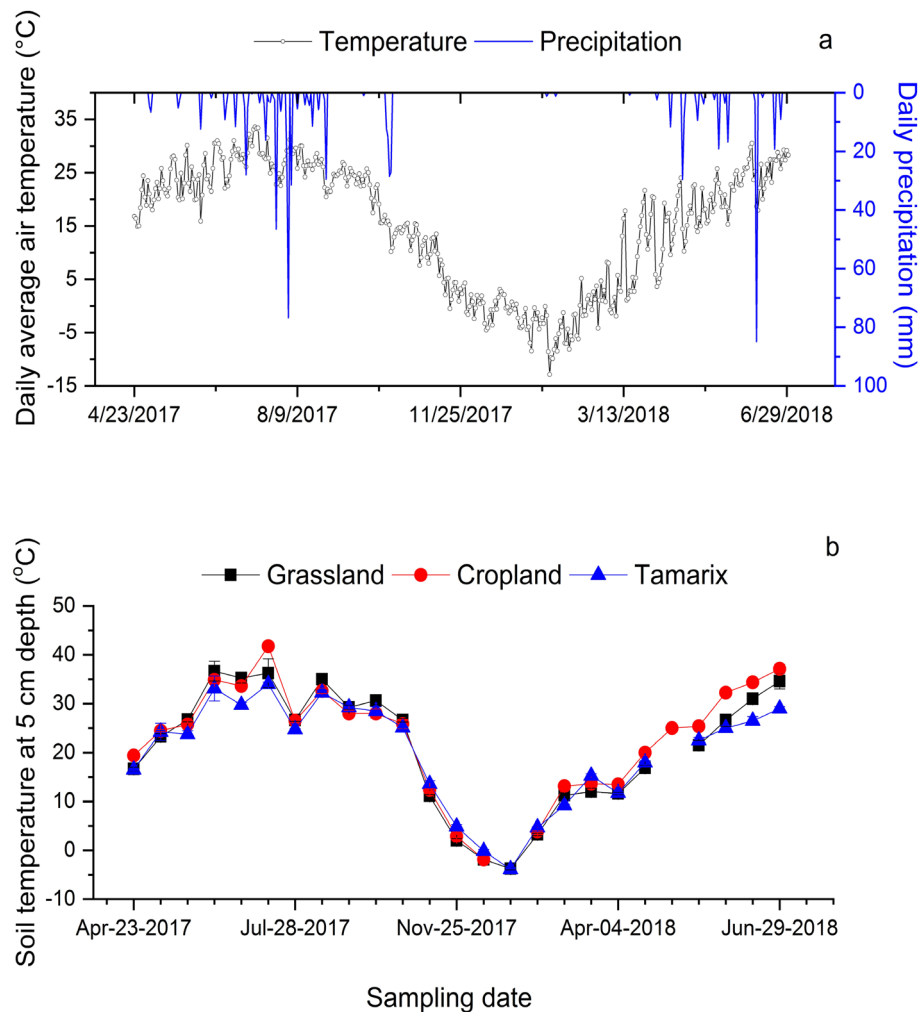
**Measurement of soil parameters.** Soil temperature at 5 cm depth was taken using a thermometer inserted into the soil. Each day after the gas sample collection, soil samples (0–20 cm) were collected from the area nearby the chambers. Thermo-gravimetric technique (oven-drying) method was used to measure the soil moisture content. Water filled pore space (WFPS) was calculated using a formula as stated in Eq. (3):

$$\text{WFPS (\%)} = (\text{SWC} \times \text{BD}) / 1 - (\text{BD} / \text{PD}) \times 100\% \quad (3)$$

where SWC is soil water content (g g<sup>-1</sup>), BD is bulk density (Mg m<sup>-3</sup>), and PD is particle density (2.65 Mg m<sup>-3</sup>).

For soil pH and electrical conductivity (Ec), 10 g of air dried (< 2 mm) soil sample was weighed and mixed with 25 and 50 ml of deionized water, respectively. Then the mixture was mechanically shaken for 1 h. pH was determined in a suspension with a pH meter (METTLER TOLEDO FE20) at 1:2.5 soil–water ratio. Ec was measured using an Ec meter (METTLER TOLEDO SG7) with 1:5 soil–water ratio at room temperature. Soil ammonium (NH<sub>4</sub>-N) and nitrate (NO<sub>3</sub>-N) concentrations were measured using the KCl extraction method. For this, 10 g of fresh soil was mixed with 50 ml of freshly prepared 1 M KCl and the mixture was shaken for one hour, then it was filtered through Whatman 42 filter paper. Then, NH<sub>4</sub>-N and NO<sub>3</sub>-N concentrations of the filtrate were measured by using a Smartchem140 and a UV spectrophotometer, respectively.





**Figure 2.** Daily average air temperature and precipitation at the study site during the study period (a) and soil temperature at 5 cm depth taken at the time of gas sampling (b). Error bars represent mean  $\pm$  standard error (SE) ( $n = 4$ ).

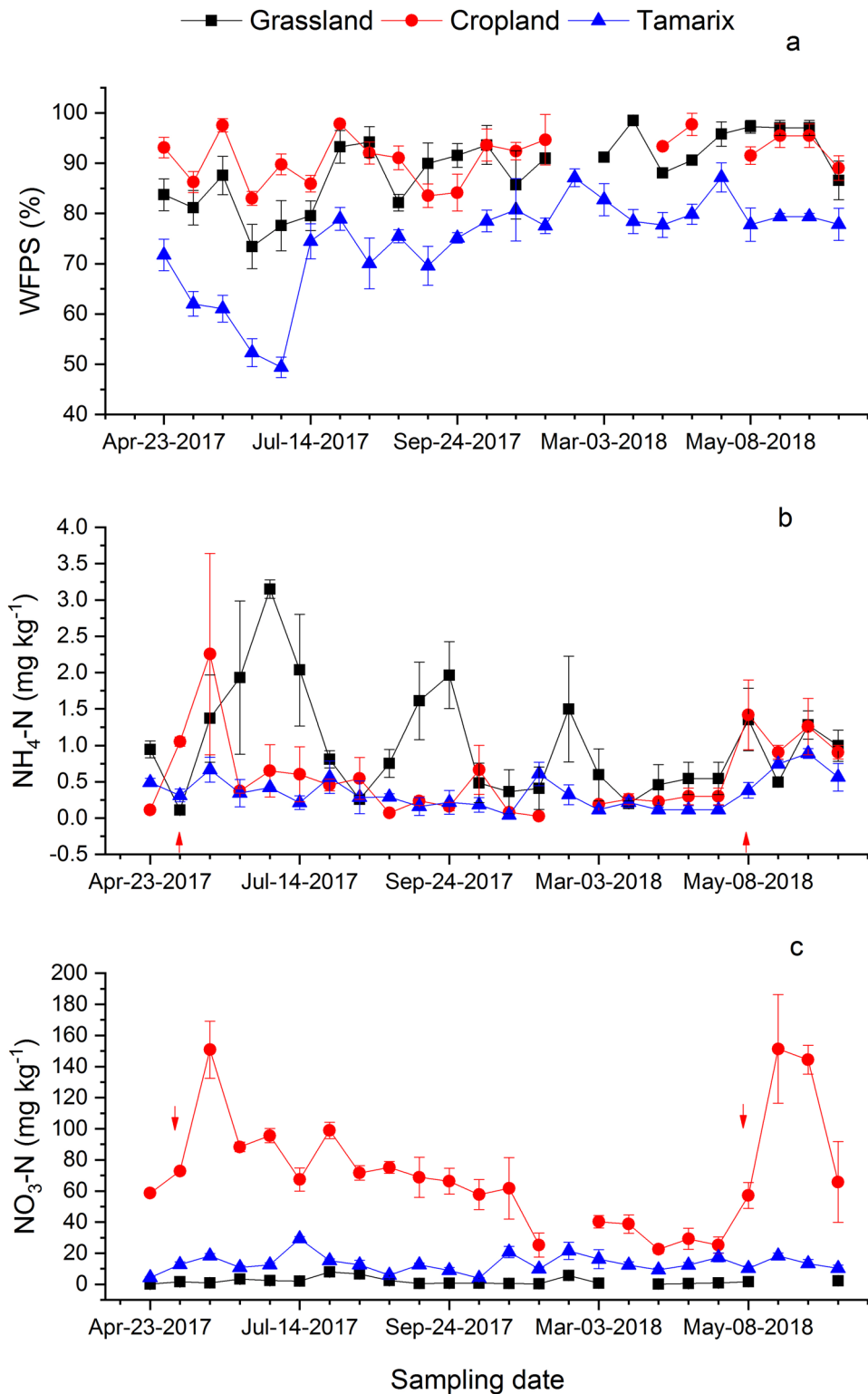
**Statistics.** Data were not normally distributed for all variables. Several possible transformations were tried without success. As our main objectives were to examine differences in  $N_2O$  fluxes and  $^{15}N$  in soil emitted  $N_2O$  in different ecosystems, we conducted the Kruskal Wallis ANOVA (analysis of variance) followed by the Mann Whitney test. The same analysis was used for other measured soil parameters. Similarly, differences in annual cumulative flux between ecosystems were computed through the Kruskal Wallis ANOVA followed by the Mann Whitney test. Spearman correlation analysis was applied to examine the relationships among the measured variables and  $N_2O$  flux. When  $p$  values were less than 0.05 the differences was considered significant. All the figures and statistical analyses were computed in Origin Pro 8 (Origin Lab Ltd., Guangzhou, China).

## Results

**Soil environmental variables.** The pattern of soil temperature was consistent with the air temperature (Fig. 2a,b). Soil temperature at 5 cm soil depth showed a clear and similar seasonal variation (high in summer and low in winter) in all ecosystems. The lowest temperature was  $-4$  °C reported in January while the highest temperature was 42 °C in July. Soil temperature at 5 cm depth at grassland was similar to the cropland and Tamarix. While the Tamarix had significantly ( $p < 0.05$ ) lower soil temperatures than the cropland. The median soil temperature was 24.5 °C, 25.3 °C and 23.5 °C in the grassland, cropland and Tamarix, respectively.

The overall WFPS of the Tamarix was significantly less ( $p < 0.001$ ) than the grassland and cropland. The median value of WFPS in the grassland was 89.6% (ranging from 66.9 to 99.95%), cropland was 90.4% (ranging from 73.32 to 99.97%) and Tamarix was 76.2% (ranging from 44.4 to 97.0%). As water table was around 0.9–1.5 m, normally WFPS exceeded 70% in all ecosystems (Fig. 3a).

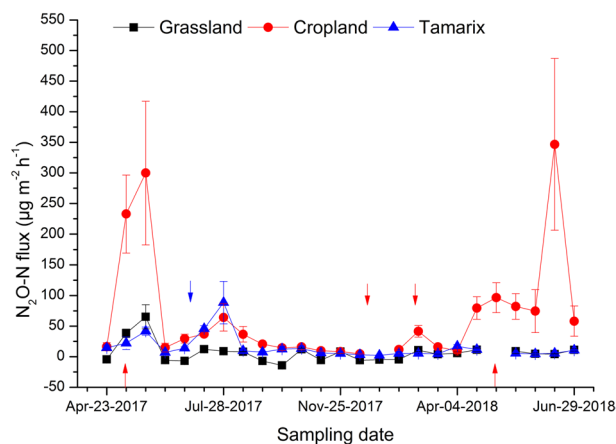
Soil  $NH_4$  was significantly ( $p < 0.01$ ) higher in the grassland compared to the cropland and Tamarix. Overall, median  $NH_4$  concentration in the grassland was  $0.55$  mg  $kg^{-1}$  (ranging from 0.006 to 4.0 mg  $kg^{-1}$ ),  $0.35$  mg  $kg^{-1}$  (ranging 0.006–6.4 mg  $kg^{-1}$ ) in the cropland and  $0.31$  mg  $kg^{-1}$  (ranging from 0.01 to 1.2 mg  $kg^{-1}$ ) in the Tamarix.



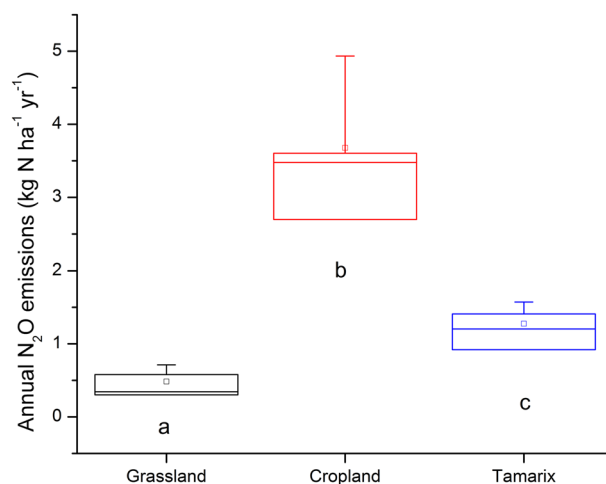
**Figure 3.** Soil water-filled pore space (WFPS) (a), Soil  $\text{NH}_4$  (b), and  $\text{NO}_3$  (c) of the top 20 cm soil. The arrows represent fertilizer application event. Each point represents arithmetic mean of  $n = 1-4 \pm \text{SE}$ .

Grassland and cropland showed higher temporal variation in soil  $\text{NH}_4$  than the Tamarix during the sampling period (Fig. 3b). After fertilization of the cropland, there was a peak in  $\text{NH}_4$  content.

Soil  $\text{NO}_3$  was significantly different ( $p < 0.001$ ) among all three ecosystems. The order of soil  $\text{NO}_3$  was: cropland > Tamarix > grassland. The median concentration of  $\text{NO}_3$  in the grassland was  $1.0 \text{ mg kg}^{-1}$  (ranging  $0.004-14.0 \text{ mg kg}^{-1}$ ),  $65 \text{ mg kg}^{-1}$  ( $6.4-209 \text{ mg kg}^{-1}$ ) in the cropland and  $12.3 \text{ mg kg}^{-1}$  (ranging from 2.6 to



**Figure 4.**  $\text{N}_2\text{O}$  flux from grassland, cropland, and Tamarix. Each point represents the arithmetic mean and standard error of four replicates. Arrows pointing upward indicate fertilization events on the cropland, while those pointing downward indicate the presence of dead pill-bugs in the Tamarix or irrigation and covered by the plastic film in the cropland (left to right). Red color arrows represent specific events in cropland and blue for Tamarix.



**Figure 5.** Box plot for annual  $\text{N}_2\text{O}$  emissions ( $n = 4$ ). Different letters indicate significant difference ( $p < 0.05$ ) and square represents mean values.

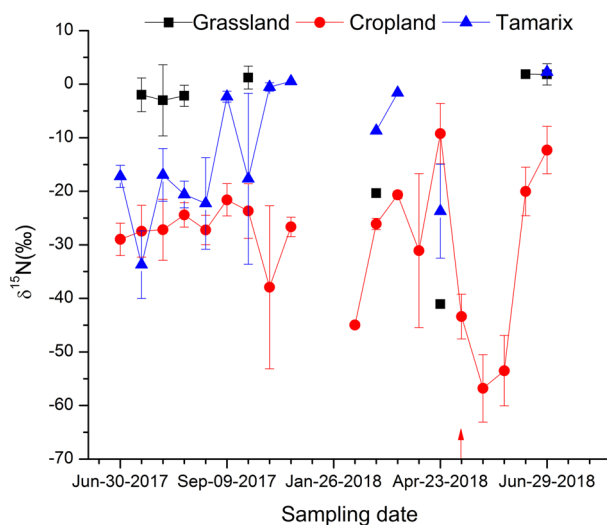
$34.30 \text{ mg kg}^{-1}$ ) in the Tamarix. For some sampling dates,  $\text{NO}_3$  was below the limit of detection in the grassland soil. Fertilizer application in the cropland led to a peak in  $\text{NO}_3$  content in the soil (Fig. 3c).

**$\text{N}_2\text{O}$  fluxes and annual cumulative emission.** Among the 24 sampling occasions, 9 occasions were found negative fluxes in the grassland, but in the cropland and Tamarix there were always positive fluxes (Fig. 4). Overall,  $\text{N}_2\text{O}$  fluxes were significantly different ( $p < 0.001$ ) among the ecosystems. The median  $\text{N}_2\text{O}$  flux was  $4.0 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$  (ranging from  $-22.0$  to  $-1.1$  for negative flux and  $2.8$  to  $117.74 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$  for the positive flux, over the study period),  $25.3 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$  (ranging from  $2.0$  to  $678.04 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$ ) and  $8.2 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$  (ranging from  $0.5$  to  $179.0 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$ ) from the grassland, cropland and Tamarix, respectively. The peak fluxes in the cropland occurred after the application of fertilizer (Fig. 4). In 2017, after fertilization the  $\text{N}_2\text{O}$  peak lasted for two weeks. While in 2018, on the day of fertilization there was a small increase, then the highest peak occurred in the 4th week after fertilization. Results for February 2018 and the 3rd week after fertilization in 2018 are not reported because it was noted that there were unusually high concentrations of  $\text{N}_2\text{O}$  (4 times higher than usual atmospheric concentration) in all samples taken at time zero, which may have led to errors in interpretation of results. For two of the sampling points, high  $\text{N}_2\text{O}$  emissions from Tamarix were observed. This occurred during the decomposition of a large number of pill-bugs that had died at the site (the reason for the pill-bug deaths is unknown).

The annual cumulative  $\text{N}_2\text{O}$  emissions were significantly different ( $p < 0.05$ ) among all three ecosystems. The annual cumulative  $\text{N}_2\text{O}$  emissions increased in the order of cropland > Tamarix > grassland. Cropland emitted  $3.5 \text{ kg N}_2\text{O-N ha}^{-1} \text{ year}^{-1}$  (ranging from  $2.7$  to  $3.9 \text{ kg N}_2\text{O-N ha}^{-1} \text{ year}^{-1}$ ) about 1.7 times more than the Tamarix,

Ecosystems/soil parameter	Grassland	Cropland	Tamarix
Soil temperature at 5 cm depth	0.08	0.426*	0.412*
WFPS	0.11	0.15	-0.27*
NH <sub>4</sub>	-0.1	0.50*	0.194
NO <sub>3</sub>	-0.06	0.44*	0.28*

**Table 2.** Spearman correlation analysis between soil environmental variables and N<sub>2</sub>O flux in different ecosystems. “\*” represents significant relationship (p < 0.05).



**Figure 6.** <sup>15</sup>N isotopic signature of soil-emitted N<sub>2</sub>O from studied ecosystems. Each point represents arithmetic mean of 1–4 replicates with standard errors. Arrow represents fertilizer application event in the cropland.

which emitted 1.3 kg N<sub>2</sub>O-N ha<sup>-1</sup> year<sup>-1</sup> (ranging from 0.9 to 1.6 kg N<sub>2</sub>O-N ha<sup>-1</sup> year<sup>-1</sup>), and 7 times more than the grassland (0.5 kg N<sub>2</sub>O-N ha<sup>-1</sup> year<sup>-1</sup>, ranging from 0.3 to 0.7 kg N<sub>2</sub>O-N ha<sup>-1</sup> year<sup>-1</sup>) (Fig. 5).

**Relationship between soil environmental variables and N<sub>2</sub>O flux.** Spearman correlation analysis showed various relationships between N<sub>2</sub>O flux and soil environmental variables measured at three studied ecosystems (Table 2). In grassland, there was no significant relationship between N<sub>2</sub>O flux and any of the measured soil parameters. In the cropland, the analysis showed significant positive correlations of N<sub>2</sub>O flux with soil temperature, NH<sub>4</sub> content, and NO<sub>3</sub> content. There was no significant correlation between N<sub>2</sub>O emission and WFPS in the cropland. Analysis of the Tamarix results showed that there were significant positive correlations of N<sub>2</sub>O flux with soil temperature and NO<sub>3</sub> content, while there was a negative relationship with WFPS.

**<sup>15</sup>N isotopic signature of soil-emitted N<sub>2</sub>O.** There was a significant difference (p < 0.01) in the <sup>15</sup>N isotopic signature of soil-emitted N<sub>2</sub>O between the three ecosystems (Fig. 6). The difference between grassland and Tamarix was at the level of p < 0.01 while between grassland and cropland was at the level of p < 0.001, suggesting N addition has strong effect on depletion of <sup>15</sup>N in N<sub>2</sub>O. N<sub>2</sub>O emitted from cropland was more depleted in <sup>15</sup>N while N<sub>2</sub>O emitted from grassland was less depleted. The median <sup>15</sup>N values in emitted N<sub>2</sub>O were -0.18 ‰ (ranging from -41.0 to 5.8‰, n = 14), -25.3 ‰ (ranging from -68.3 to 4.6 ‰, n = 63) and -13.7 ‰ (ranging from -50.5 to 3.0‰, n = 32) for the grassland, cropland and Tamarix, respectively. Due to problems with the IRMS, results from the beginning of the experiment are not included. In the grassland, due to low and negative fluxes of N<sub>2</sub>O, it was not always possible to calculate <sup>15</sup>N values in soil-emitted N<sub>2</sub>O. Emitted N<sub>2</sub>O was more depleted in <sup>15</sup>N in April in the grassland while in the Tamarix it was during the pill-bug decomposition period. In the cropland, it was just after the application of N fertilizer and this continued for about three weeks after the fertilization, then in the fourth week, when N<sub>2</sub>O emission reached its highest peak, the values returned to the normal range (Figs. 5, 6). There was no significant relationship between measured parameters and <sup>15</sup>N in soil-emitted N<sub>2</sub>O.

## Discussion

At our experimental site, we had a unique opportunity to investigate the impact of land-use change (LUC) from natural to semi-natural and artificial ecosystems on N<sub>2</sub>O flux and its <sup>15</sup>N within the same climatic conditions and soil type. LUC is associated with changes in various land cover types as a result of different management



practices, which then can lead to changes in soil physical, chemical<sup>41</sup> and biological properties<sup>20</sup>. The changes in these soil properties can alter soil greenhouse gas emissions<sup>16,19</sup>. Soil humidity, temperature,  $\text{NH}_4$  content and  $\text{NO}_3$  content are the major soil parameters that influence  $\text{N}_2\text{O}$  emission from soil<sup>21,36,42</sup>. With the change in the land use, it was observed that these soil parameters were significantly influenced at our study site, which may have led to the differences in  $\text{N}_2\text{O}$  flux from the different ecosystems.

In the grassland, no studied soil parameters were significantly correlated to  $\text{N}_2\text{O}$  flux, which may have been due to limited  $\text{NO}_3$  content. The relatively high  $\text{NH}_4$  content and low  $\text{NO}_3$  in grassland soil indicates inhibition of nitrification process, causing low  $\text{N}_2\text{O}$  emissions. The positive correlation between soil temperature and  $\text{N}_2\text{O}$  emission in the cropland and Tamarix, observed in our study is consistent with other studies<sup>36,43</sup> and can be explained by the increase in microbial activity with an increase in temperature<sup>44</sup>. WFPS higher than 80% is favorable for  $\text{N}_2\text{O}$  reduction to  $\text{N}_2$ <sup>22</sup>. Low N content along with higher WFPS and frequent  $\text{N}_2\text{O}$  uptake results reported in the grassland site indicate that denitrification is a dominant process of  $\text{N}_2\text{O}$  emission. Optimum WFPS for  $\text{N}_2\text{O}$  emissions ranges from 60 to 80%<sup>22</sup>, and there have been reports of significant positive to negative or no relationship between WFPS and  $\text{N}_2\text{O}$  emission<sup>45–47</sup>. Increase in soil moisture has a greater effect when dry soil is wetted<sup>48</sup>. So, higher WFPS (around 90%) in grassland and cropland might not be limiting factor controlling  $\text{N}_2\text{O}$  emissions in our study. We only observed significant relationship between WFPS and  $\text{N}_2\text{O}$  flux in Tamarix. The negative relationship might be due to excessive WFPS than that is required for optimum  $\text{N}_2\text{O}$  production<sup>49</sup>.  $\text{NH}_4$  and  $\text{NO}_3$  are the main substrates for nitrification and denitrification<sup>50,51</sup>. Significant positive relationships between  $\text{N}_2\text{O}$  emission and both  $\text{NH}_4$  and  $\text{NO}_3$  have previously been demonstrated<sup>42</sup> indicating that coupled nitrification–denitrification contributes to  $\text{N}_2\text{O}$  formation in the soil<sup>50</sup>. Similarly, in the current study positive relationships were found between  $\text{N}_2\text{O}$  flux and  $\text{NH}_4$  and  $\text{NO}_3$  content in the cropland; however, only with  $\text{NO}_3$  in the Tamarix. It can be difficult to identify the  $\text{N}_2\text{O}$  formation process responsible or the emissions i.e. either nitrification or denitrification, as both processes can occur simultaneously in the soil<sup>50</sup>. The results showing a range of both positive and negative relationships between various soil environmental parameters and  $\text{N}_2\text{O}$  flux indicate that  $\text{N}_2\text{O}$  formation processes have complex interactions with these soil parameters.

Often ecosystems with low N content have a negative flux and low annual  $\text{N}_2\text{O}$  emission. The grassland site in our study was like most natural ecosystems<sup>21,53</sup>, N limited with low atmospheric nitrogen input and densely rooted vegetation and therefore emitted less  $\text{N}_2\text{O}$ <sup>54</sup>. High WFPS with low N content favors denitrification leading to  $\text{N}_2\text{O}$  consumption<sup>53,55</sup>. However, relatively dry ecosystems have also been reported to consume atmospheric  $\text{N}_2\text{O}$ <sup>56–58</sup>; however, the possible mechanisms of  $\text{N}_2\text{O}$  consumption by soil under dry conditions are not well understood<sup>59</sup>.  $\text{N}_2\text{O}$  uptake has been observed at low  $\text{NO}_3$  levels ( $\sim 1 \text{ mg N kg}^{-1}$ ) and  $\text{NH}_4$  content ( $< 2 \text{ mg N kg}^{-1}$ ) levels and high WFPS (90%)<sup>60,61</sup>. The grassland conditions in the current study were similar to these previous findings that may be the reason for  $\text{N}_2\text{O}$  uptake occurring in the grassland in some sampling occasions. It has also been observed that soil under different plant species can have different rates of  $\text{N}_2\text{O}$  reduction<sup>62</sup> and that  $\text{N}_2\text{O}$  consumption rate decreases with increase in soil  $\text{NO}_3$ <sup>63</sup>. In the cropland and Tamarix systems in the present study,  $\text{NO}_3$  content was significantly higher than in the grassland, which might have resulted in a decrease in the reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$ , leading to the higher emission of  $\text{N}_2\text{O}$ . The more depleted  $^{15}\text{N}$  values in soil-emitted  $\text{N}_2\text{O}$  in the cropland and Tamarix compared to the grassland (Fig. 6) is further evidence of a decrease in the reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$  in those systems<sup>29,30</sup>.

Overall,  $\text{N}_2\text{O}$  flux in the grassland was low ( $4.0 \text{ N}_2\text{O-N } \mu\text{g m}^{-2} \text{ h}^{-1}$ ) with an annual cumulative emission of  $0.5 \text{ kg N}_2\text{O-N ha}^{-1} \text{ year}^{-1}$ . These findings are similar to those observed in other studies on natural grassland under different climatic conditions on non-saline soils<sup>54,59,64–66</sup>. However, compared to a saline grassland with the same dominant vegetation<sup>36</sup> the flux rate in the current study was low. This was possibly due to the low  $\text{NO}_3$  and  $\text{NH}_4$  content. When natural grasslands with low N content are converted to cropland, the addition of a large amount of N fertilizer may potentially contribute to high  $\text{N}_2\text{O}$  emissions<sup>65</sup>. Consistent with this, the cropland in the current study emitted about 7 times more  $\text{N}_2\text{O}$  than the grassland. The annual  $\text{N}_2\text{O}$  emission rate was similar to the IPCC default emission factor, i.e. 1% of applied N fertilizer is emitted as  $\text{N}_2\text{O}$  in the agricultural fields<sup>67</sup>. The observed  $\text{N}_2\text{O}$  emission from our cropland was lower than that from non-saline–alkaline soils in the same climatic area under application of the same amount of fertilizer<sup>68</sup>. Similarly, the  $\text{N}_2\text{O}$  flux from some non-saline–alkaline soils, receiving a similar rate of fertilizer, was three times higher than from the cropland in our study<sup>43</sup>. A saline–alkaline sunflower field, receiving  $300 \text{ kg N ha}^{-1} \text{ year}^{-1}$ , emitted  $9.8 \text{ kg N ha}^{-1} \text{ year}^{-1}$ <sup>10</sup>, which is 3.8 times higher than the emission rate from the cropland in the current study, which had  $400 \text{ kg N ha}^{-1} \text{ year}^{-1}$  applied. The Tamarix ecosystem emitted 2.6 times more  $\text{N}_2\text{O}$  than the native grassland. This increase can be attributed to the higher  $\text{NO}_3$  content. The increase in  $\text{NO}_3$  content could also be linked to a lower reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$  in the Tamarix system because high  $\text{NO}_3$  inhibits  $\text{N}_2\text{O}$  reduction<sup>69</sup>. Conversion of grassland to tree plantations has a contrasting (increased to no influence) effect on  $\text{N}_2\text{O}$  emission<sup>17,18</sup>. Overall, our results support our hypothesis that conversion of native grassland to cropland or Tamarix ecosystems would lead to change in soil environmental variables and an increase in  $\text{N}_2\text{O}$  emission.

When compared with studies involving similar land use or land-use change (Supplement Information S1) our results from the respective ecosystems are within the ranges reported in the literature. This result suggests that saline–alkaline soils may not always have a higher potential for  $\text{N}_2\text{O}$  emission, as hypothesized by Ghosh et al.<sup>70</sup> and Yang et al.<sup>10</sup>. For the cropland the fertilizer application rate was higher than other studies in the literature (Supplement Information S1), this is likely to have led to the higher rate of  $\text{N}_2\text{O}$  emission from the cropland. In saline–alkaline soil,  $\text{NH}_4$  can be converted to  $\text{NH}_3$  and lost to the atmosphere, which may decrease the probability of  $\text{N}_2\text{O}$  formation due to nitrification<sup>13</sup>. Two meta-analyses<sup>11,12</sup> reported that alkaline soils emit less  $\text{N}_2\text{O}$  compared to natural and acidic soils. Furthermore, high salinity inhibits both nitrification and denitrification processes<sup>8,9</sup>. These negative effects of both salinity and alkalinity on  $\text{N}_2\text{O}$  production processes and emissions further suggest that saline–alkaline soil may not emit more  $\text{N}_2\text{O}$ .

It is evident from previous research<sup>25,28</sup> that there may be differences in the <sup>15</sup>N in soil-emitted N<sub>2</sub>O between fertilized and unfertilized ecosystems. Therefore, significant differences were expected in the <sup>15</sup>N isotopic signatures in soil-emitted N<sub>2</sub>O between the unfertilized ecosystems (grassland and Tamarix) and the fertilized cropland. As there was no anthropogenic N input in grassland and Tamarix, our expectation was <sup>15</sup>N in N<sub>2</sub>O would be similar in these two ecosystems. However, differences were observed among all three ecosystems. The <sup>15</sup>N in N<sub>2</sub>O emitted from the grassland, cropland, and Tamarix were all within the range reported by other studies<sup>25,26,28,62,71</sup>. As we can see from Fig. 6 that temporal variability of <sup>15</sup>N in soil-emitted N<sub>2</sub>O was highest in cropland, indicating that N cycling process in the cropland is relatively open. The more depleted <sup>15</sup>N in N<sub>2</sub>O emitted from the cropland implies that N availability can be considered enhanced (due to the high rate of N fertilizer) in the ecosystem<sup>25</sup>. When nitrogen availability is enhanced, the N<sub>2</sub>O production process favors larger <sup>15</sup>N fractionation, leading to more depleted <sup>15</sup>N in N<sub>2</sub>O from the soil<sup>25,72</sup>. This phenomenon can lead to difference in the <sup>15</sup>N in N<sub>2</sub>O emitted from the cropland compared to the grassland and Tamarix, as observed in this study. After application of fertilizer the cropland could be considered to have unlimited N availability so the N<sub>2</sub>O emitted was strongly depleted in <sup>15</sup>N, indicating the production of N<sub>2</sub>O, either by nitrification or denitrification, favored larger <sup>15</sup>N fractionation rather than shift from denitrification to nitrification<sup>25,28,71,72</sup>. Although <sup>15</sup>N values in soil-emitted N<sub>2</sub>O can sometimes be used to predict sources of N<sub>2</sub>O when combined measurements of <sup>15</sup>N values in substrates for N<sub>2</sub>O production<sup>28</sup> and molecular analysis of N<sub>2</sub>O producing organisms<sup>40</sup>, with data from this trial was not possible to estimate relative contributions of nitrification and denitrification. Moreover, more powerful tools like <sup>15</sup>N site preference (SP) is a good indicator of production pathways<sup>24,73,74</sup>, which was not used in this study, making difficult to generalize dominant process of N<sub>2</sub>O production in different ecosystems.

Contrarily to our hypothesis, there was a difference between <sup>15</sup>N in soil-emitted N<sub>2</sub>O within unfertilized (grassland and tamarix) ecosystems. The reason for differences in the <sup>15</sup>N in N<sub>2</sub>O between the grassland and Tamarix may be a difference in N<sub>2</sub>O reduction capability. It is likely that N<sub>2</sub>O reduction in the grassland (as evidenced by N<sub>2</sub>O consumption) enriched the <sup>15</sup>N in N<sub>2</sub>O, so when it was emitted to the atmosphere it was less depleted than N<sub>2</sub>O emitted from soil in which reduction has not occurred<sup>29,75</sup>. A possible reason for the reduction of N<sub>2</sub>O being favored in the grassland soil may be the low concentrations of NO<sub>3</sub><sup>-</sup><sup>69</sup> and high WFPS<sup>22</sup>. For this reason reduction of N<sub>2</sub>O to N<sub>2</sub> might be more prominent in the grassland compared to the Tamarix. However, it could be a possibility that gross N<sub>2</sub>O consumption may be masked by higher rates of N<sub>2</sub>O production<sup>76</sup> in the cropland and Tamarix. The <sup>15</sup>N isotope content of the substrates (NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup>) for N<sub>2</sub>O production were not measured in the current study, which could have provided more insight into the reason for the observed differences between the ecosystems. The <sup>15</sup>N differences in the emitted N<sub>2</sub>O between ecosystems could also be due to variation in the microbial community composition in the soils<sup>77</sup>. Several factors favor complete denitrification, such as differences in microbial community composition (denitrifiers), presence of denitrification enzymes, high soil water content, high soil pH, a low rate of O<sub>2</sub> diffusion and presence of labile carbon<sup>55</sup>. So differences in those factors should not be ruled out as causes for the differences in the <sup>15</sup>N content in emitted N<sub>2</sub>O between the ecosystems.

The <sup>15</sup>N content in atmospheric N<sub>2</sub>O has been decreasing since the preindustrial age<sup>78</sup>; however, atmospheric N<sub>2</sub>O concentration is increasing<sup>5</sup>. This decrease in the <sup>15</sup>N in N<sub>2</sub>O has been considered to be a result of an increase in the use of chemical fertilizer<sup>5,28</sup>. Moreover, global decline in the N<sub>2</sub>O reduction process relative to production might also contribute to the decrease in the <sup>15</sup>N<sup>29</sup>. Our results indicate that the conversion of natural ecosystems to cropland with the addition of anthropogenic N would greatly contribute to the depletion of the <sup>15</sup>N in atmospheric N<sub>2</sub>O by emitting more depleted <sup>15</sup>N in N<sub>2</sub>O along with higher N<sub>2</sub>O emission rate, which was according to our hypothesis. Moreover, if ecosystems with more reduction capability (such as grassland) are converted to Tamarix that have less reduction capability (assumed due to the absence of measured atmospheric N<sub>2</sub>O consumption in our study), this would also play a role in the depletion of <sup>15</sup>N in atmospheric N<sub>2</sub>O. Overall, it can be concluded that the addition of anthropogenic N to cropland would contribute more to deplete <sup>15</sup>N in atmospheric N<sub>2</sub>O than any other processes.

## Conclusions

Our study showed that LUC from native grassland to Tamarix and cropland on saline–alkaline soil significantly influence soil temperature, soil moisture and NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> contents. The changes in these soil factors, along with the observed correlations between N<sub>2</sub>O fluxes and the soil parameters, could explain the differences in N<sub>2</sub>O flux caused by the LUC. Saline–alkaline soil may not always act as a potentially high source of N<sub>2</sub>O, as our fluxes and annual emissions result are in the usual ranges for the respective ecosystems reported in the literature. The conversion from native grassland to Tamarix ecosystem increased more N<sub>2</sub>O 2.6 times while cropland increased 7 times. The LUC also influenced the <sup>15</sup>N in soil-emitted N<sub>2</sub>O, greatly depleting it in cropland and moderate in Tamarix compared to native grassland. The differences in the <sup>15</sup>N in soil-emitted N<sub>2</sub>O between the fertilized and unfertilized ecosystems could be attributable to anthropogenic N fertilization. The differences in the <sup>15</sup>N in N<sub>2</sub>O between the unfertilized ecosystems (grassland and Tamarix) could be attributable to the N<sub>2</sub>O reduction capacity of native grassland. Our results further suggest that the depletion of the <sup>15</sup>N in atmospheric N<sub>2</sub>O since the pre-industrial age could be highly attributable to anthropogenic N addition and to lesser extent to land-use changes where ecosystems with more N<sub>2</sub>O reduction capacity have been converted to ecosystems with less N<sub>2</sub>O reduction capacity.

## Data availability

The datasets of the current study will be available from the corresponding author on reasonable request.

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## References

- Myhre, G. *et al.* Anthropogenic and natural radiative forcing. In *Climate change 2013: The physical science basis; Working Group I contribution to the fifth assessment report of the Intergovernmental Panel on Climate Change* (ed. Stocker, T.) 659–740 (Cambridge University Press, Cambridge, 2014).
- Ravishankara, A. R., Daniel, J. S. & Portmann, R. W. Nitrous oxide (N<sub>2</sub>O): The dominant ozone-depleting substance emitted in the 21st century. *Science* **326**, 123–125 (2009).
- Thomson, A. J., Giannopoulos, G., Pretty, J., Baggs, E. M. & Richardson, D. J. Biological sources and sinks of nitrous oxide and strategies to mitigate emissions. *Philos. Trans. R. Soc. B Biol. Sci.* **367**(1593), 1157–1168 (2012).
- NoAA, ESLR (2019). <https://www.esrl.noaa.gov/gmd/hats/insitu/cats/conc.php?site=brw&gas=n2o>. Accessed on 01 May 2019.
- Park, S. *et al.* Trends and seasonal cycles in the isotopic composition of nitrous oxide since 1940. *Nat. Geosci.* **5**, 261–265 (2012).
- Davidson, E. A. The contribution of manure and fertilizer nitrogen to atmospheric nitrous oxide since 1860. *Nat. Geosci.* **2**, 659–662 (2009).
- Hu, H., Chen, D. & He, J. Microbial regulation of terrestrial nitrous oxide formation: Understanding the biological pathways for prediction of emission rates. *FEMS Microbiol. Rev.* **39**(5), 729–749 (2015).
- Zhou, M., Butterbach-Bahl, K., Vereecken, H. & Brüggemann, N. A meta-analysis of soil salinization effects on nitrogen pools, cycles and fluxes in coastal ecosystems. *Glob. Chang. Biol.* **23**, 1338–1352 (2017).
- Inubushi, K., Barahona, M. A. & Yamakawa, K. Effects of salts and moisture content on N<sub>2</sub>O emission and nitrogen dynamics in Yellow soil and Andosol in model experiments. *Biol. Fertil. Soils* **29**, 401–407 (1999).
- Yang, W., Yang, M., Wen, H. & Jiao, Y. Global warming potential of CH<sub>4</sub> uptake and N<sub>2</sub>O emissions in saline–alkaline soils. *Atmos. Environ.* **191**, 172–180 (2018).
- Aliyu, G. *et al.* A meta-analysis of soil background N<sub>2</sub>O emissions from croplands in China shows variation among climatic zones. *Agric. Ecosyst. Environ.* **267**, 63–73 (2018).
- Wang, Y. *et al.* Soil pH as the chief modifier for regional nitrous oxide emissions: New evidence and implications for global estimates and mitigation. *Glob. Chang. Biol.* **24**, 617–626 (2018).
- Bouwman, A. F., Boumans, L. J. M. & Batjes, N. H. Estimation of global NH<sub>3</sub> volatilization loss from synthetic fertilizers and animal manure applied to arable lands and grasslands. *Glob. Biogeochem. Cycles* **16**, 81–814 (2002).
- Allen, D. E. *et al.* Nitrous oxide and methane emissions from soil are reduced following afforestation of pasture lands in three contrasting climatic zones. *Soil Res.* **47**, 443 (2009).
- Pendall, E. *et al.* Land use and season affect fluxes of CO<sub>2</sub>, CH<sub>4</sub>, CO, N<sub>2</sub>O, H<sub>2</sub> and isotopic source signatures in Panama: Evidence from nocturnal boundary layer profiles. *Glob. Chang. Biol.* **16**, 2721–2736 (2010).
- Van Lent, J., Hergoualc, H. K. & Verchot, L. V. Reviews and syntheses: Soil N<sub>2</sub>O and NO emissions from land use and land-use change in the tropics and subtropics: A meta-analysis. *Biogeosciences* **12**, 7299–7313 (2015).
- Benanti, G., Saunders, M., Tobin, B. & Osborne, B. Contrasting impacts of afforestation on nitrous oxide and methane emissions. *Agric. For. Meteorol.* **198–199**, 82–93 (2014).
- de Godoi, S. G. *et al.* The conversion of grassland to acacia forest as an effective option for net reduction in greenhouse gas emissions. *J. Environ. Manag.* **169**, 91–102 (2016).
- Zona, D. *et al.* Fluxes of the greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O) above a short-rotation poplar plantation after conversion from agricultural land. *Agric. For. Meteorol.* **169**, 100–110 (2013).
- Li, C., Di, H. J., Cameron, K. C., Podolyan, A. & Zhu, B. Effect of different land use and land use change on ammonia oxidiser abundance and N<sub>2</sub>O emissions. *Soil Biol. Biochem.* **96**, 169–175 (2016).
- Ussiri, D. & Lal, R. *Soil Emission of Nitrous Oxide and Its Mitigation* (Springer, Netherlands, 2013).
- Butterbach-Bahl, K. *et al.* Nitrous oxide emissions from soils: How well do we understand the processes and their controls?. *Philos. Trans. R. Soc. B Biol. Sci.* **368**, 1621. <https://doi.org/10.1098/rstb.2013.0122> (2013).
- Sarauer, J. L. & Coleman, M. D. Converting conventional agriculture to poplar bioenergy crops: Soil greenhouse gas flux. *Scand. J. For. Res.* **33**, 781–792 (2018).
- Denk, T. R. A. *et al.* The nitrogen cycle: A review of isotope effects and isotope modeling approaches. *Soil Biol. Biochem.* **105**, 121–137 (2017).
- Park, S. *et al.* Can N<sub>2</sub>O stable isotopes and isotopomers be useful tools to characterize sources and microbial pathways of N<sub>2</sub>O production and consumption in tropical soils?. *Glob. Biogeochem. Cycles* **25**, 1–16 (2011).
- Koba, K. *et al.* The <sup>15</sup>N natural abundance of the N lost from an N-saturated subtropical forest in southern China. *J. Geophys. Res. Biogeosci.* **117**, G2. <https://doi.org/10.1029/2010JG001615> (2012).
- Gil, J., Pérez, T., Boering, K., Martikainen, P. J. & Biasi, C. Mechanisms responsible for high N<sub>2</sub>O emissions from subarctic permafrost peatlands studied via stable isotope techniques. *Glob. Biogeochem. Cycles* **31**, 172–189 (2017).
- Pérez, T. *et al.* Identifying the agricultural imprint on the global N<sub>2</sub>O budget using stable isotopes. *J. Geophys. Res. Atmos.* **106**, 9869–9878 (2001).
- Conen, F. & Neftel, A. Do increasingly depleted δ<sup>15</sup>N values of atmospheric N<sub>2</sub>O indicate a decline in soil N<sub>2</sub>O reduction?. *Biogeochemistry* **82**(3), 321–326 (2007).
- Wada, E. & Ueda, S. Carbon, nitrogen, and oxygen isotope ratios of CH<sub>4</sub> and N<sub>2</sub>O in soil ecosystems. In *Mass Spectrometry of Soils* (eds Boutton, T. W. & Yamaski, S. I.) 177–203 (Marchel Dekker, New York, 1996).
- Lambin, E. F. & Meyfroidt, P. Global land use change, economic globalization, and the looming land scarcity. *Proc. Natl. Acad. Sci. USA* **108**, 3465–3472 (2011).
- Shi, Z., Wang, R., Huang, M. X. & Landgraf, D. Detection of coastal saline land uses with multi-temporal landsat images in Shangyu City, China. *Environ. Manag.* **30**, 142–150 (2002).
- Manning, M. *et al.* *The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change* (Cambridge University Press, Cambridge, 2007).
- Cui, B., Yang, Q., Zhang, K., Zhao, X. & You, Z. Responses of saltcedar (*Tamarix chinensis*) to water table depth and soil salinity in the Yellow River Delta, China. *Plant Ecol.* **209**, 279–290 (2010).
- Feng, X. *et al.* Spatiotemporal heterogeneity of soil water and salinity after establishment of dense-foliage *Tamarix chinensis* on coastal saline land. *Ecol. Eng.* **121**, 104–113 (2018).
- Zhang, L. L. *et al.* Seasonal dynamics in nitrous oxide emissions under different types of vegetation in saline–alkaline soils of the Yellow River Delta, China and implications for eco-restoring coastal wetland. *Ecol. Eng.* **61**, 82–89 (2013).
- Abalos, D., van Groenigen, J. W. & De Deyn, G. B. What plant functional traits can reduce nitrous oxide emissions from intensively managed grasslands?. *Glob. Chang. Biol.* **24**, 248–258 (2018).
- Ju, Z., Du, Z., Guo, K. & Liu, X. Irrigation with freezing saline water for 6 years alters salt ion distribution within soil aggregates. *J. Soils Sedim.* **19**, 97–105 (2019).
- Li, F. *et al.* Impact of rice–fish/shrimp co-culture on the N<sub>2</sub>O emission and NH<sub>3</sub> volatilization in intensive aquaculture ponds. *Sci. Total Environ.* **655**, 284–291 (2019).

40. Snider, D., Thompson, K., Wagner-Riddle, C., Spoelstra, J. & Dunfield, K. Molecular techniques and stable isotope ratios at natural abundance give complementary inferences about N<sub>2</sub>O production pathways in an agricultural soil following a rainfall event. *Soil Biol. Biochem.* **88**, 197–213 (2015).
41. Dong, W. H., Zhang, S., Rao, X. & Liu, C.-A. Newly-reclaimed alfalfa forage land improved soil properties comparison to farmland in wheat–maize cropping systems at the margins of oases. *Ecol. Eng.* **94**, 57–64 (2016).
42. Livesley, S. J. *et al.* Soil–atmosphere exchange of greenhouse gases in a *Eucalyptus marginata* woodland, a clover–grass pasture, and *Pinus radiata* and *Eucalyptus globulus* plantations. *Glob. Chang. Biol.* **15**, 425–440 (2009).
43. Lin, S. *et al.* Differences in nitrous oxide fluxes from red soil under different land uses in mid-subtropical China. *Agric. Ecosyst. Environ.* **146**, 168–178 (2012).
44. Song, A. *et al.* Substrate-driven microbial response: A novel mechanism contributes significantly to temperature sensitivity of N<sub>2</sub>O emissions in upland arable soil. *Soil Biol. Biochem.* **118**, 18–26 (2018).
45. Huang, X. *et al.* A flexible Bayesian model for describing temporal variability of N<sub>2</sub>O emissions from an Australian pasture. *Sci. Total Environ.* **454**, 206–210 (2013).
46. Nan, W. *et al.* Characteristics of N<sub>2</sub>O production and transport within soil profiles subjected to different nitrogen application rates in China. *Sci. Total Environ.* **542**, 864–875 (2016).
47. Chaddy, A., Melling, L., Ishikura, K. & Hatano, R. Soil N<sub>2</sub>O emissions under different N rates in an oil palm plantation on tropical peatland. *Agriculture* **9**, 213. <https://doi.org/10.3390/agriculture9100213> (2019).
48. Davidson, E. A. Sources of nitric oxide and nitrous oxide following wetting of dry soil. *Soil Sci. Soc. Am. J.* **56**, 95–102 (1992).
49. Kazuya, N. *et al.* Evaluation of uncertainties in N<sub>2</sub>O and NO fluxes from agricultural soil using a hierarchical bayesian model. *J. Geophys. Res.* **117**, G04008. <https://doi.org/10.1029/2012JG002157> (2012).
50. Sakata, R. *et al.* Effect of soil types and nitrogen fertilizer on nitrous oxide and carbon dioxide emissions in oil palm plantations. *Soil Sci. Plant Nutr.* **61**, 48–60 (2015).
51. Smith, K. A. Changing views of nitrous oxide emissions from agricultural soil: Key controlling processes and assessment at different spatial scales. *Eur. J. Soil Sci.* **68**, 137–155 (2017).
52. Bateman, E. J. & Baggs, E. M. Contributions of nitrification and denitrification to N<sub>2</sub>O emissions from soils at different water-filled pore space. *Biol. Fertil. Soils* **41**, 379–388 (2005).
53. Chapuis-lardy, L., Wrage, N., Metay, A., Chotte, J. L. & Bernoux, M. Soils, a sink for N<sub>2</sub>O? A review. *Glob. Chang. Biol.* **13**, 1–17 (2007).
54. Glatzel, S. & Stahr, K. Methane and nitrous oxide exchange in differently fertilised grassland in southern Germany. *Plant Soil* **231**, 21–35 (2001).
55. Sagar, S. *et al.* Denitrification and N<sub>2</sub>O:N<sub>2</sub> production in temperate grasslands: Processes, measurements, modelling and mitigating negative impacts. *Sci. Total Environ.* **465**, 173–195 (2013).
56. Warneke, S., Schipper, L. A., Bruesewitz, D. A., McDonald, I. & Cameron, S. Rates, controls and potential adverse effects of nitrate removal in a denitrification bed. *Ecol. Eng.* **37**, 511–522 (2011).
57. Wang, Y. *et al.* Depth-dependent greenhouse gas production and consumption in an upland cropping system in northern China. *Geoderma* **319**, 100–112 (2018).
58. Wu, D. *et al.* N<sub>2</sub>O consumption by low-nitrogen soil and its regulation by water and oxygen. *Soil Biol. Biochem.* **60**, 165–172 (2013).
59. Dijkstra, F. A., Morgan, J. A., Follett, R. F. & LeCain, D. R. Climate change reduces the net sink of CH<sub>4</sub> and N<sub>2</sub>O in a semiarid grassland. *Glob. Chang. Biol.* **19**, 1816–1826 (2013).
60. Ryden, J. C. Denitrification loss from a grassland soil in the field receiving different rates of nitrogen as ammonium nitrate. *J. Soil Sci.* **34**, 355–365 (1983).
61. Rosenkranz, P. *et al.* N<sub>2</sub>O, NO and CH<sub>4</sub> exchange, and microbial N turnover over a Mediterranean pine forest soil. *Biogeosciences* **3**(2), 121–133 (2006).
62. Menyailo, O. V. & Hungate, B. A. Stable isotope discrimination during soil denitrification: Production and consumption of nitrous oxide. *Glob. Biochem. Cycle.* **20**, 3. <https://doi.org/10.1029/2005GB002527> (2006).
63. Roobroeck, D., Butterbach-Bahl, K., Brüggemann, N. & Boeckx, P. Dinitrogen and nitrous oxide exchanges from an undrained monolith fen: Short-term responses following nitrate addition. *Eur. J. Soil Sci.* **61**, 662–670 (2010).
64. Kammann, C., Grünhage, L., Müller, C., Jacobi, S. & Jäger, H.-J. Seasonal variability and mitigation options for N<sub>2</sub>O emissions from differently managed grasslands. *Environ. Pollut.* **102**, 179–186 (1998).
65. Yang, X. *et al.* Nitrous oxide emissions from an agro-pastoral ecotone of northern China depending on land uses. *Agric. Ecosyst. Environ.* **213**, 241–251 (2015).
66. Peng, Q., Qi, Y., Dong, Y., Xiao, S. & He, Y. Soil nitrous oxide emissions from a typical semiarid temperate steppe in inner Mongolia: Effects of mineral nitrogen fertilizer levels and forms. *Plant Soil* **342**, 345–357 (2011).
67. Eggleston, S. *et al.* (eds) *IPCC Guidelines for National Greenhouse Gas Inventories* (Institute for Global Environmental Strategies, Hayama, 2006).
68. Qin, S. *et al.* Yield-scaled N<sub>2</sub>O emissions in a winter wheat–summer corn double-cropping system. *Atmos. Environ.* **55**, 240–244 (2012).
69. Blackmer, A. M. & Bremner, J. M. Inhibitory effect of nitrate on reduction of N<sub>2</sub>O to N<sub>2</sub> by soil microorganisms. *Soil Biol. Biochem.* **10**, 187–191 (1978).
70. Ghosh, U., Thapa, R., Desutter, T., He, Y. & Chatterjee, A. Saline-sodic soils: Potential sources of nitrous oxide and carbon dioxide emissions?. *Pedosphere* **27**, 65–75 (2017).
71. Townsend-Small, A. *et al.* Nitrous oxide emissions and isotopic composition in urban and agricultural systems in southern California. *J. Geophys. Res. Biogeosci.* **116**, 1–11 (2011).
72. Mariotti, A. *et al.* Experimental determination of nitrogen kinetic isotope fractionation: Some principles; illustration for the denitrification and nitrification processes. *Plant Soil* **62**, 413–430 (1981).
73. Ibraim, E. *et al.* Attribution of N<sub>2</sub>O sources in a grassland soil with laser spectroscopy based isotopocule analysis. *Biogeosciences* **16**(16), 3247–3266 (2019).
74. Timilsina, A. *et al.* Potential pathway of nitrous oxide formation in plants. *Front. Plant Sci.* <https://doi.org/10.3389/fpls.2020.01177> (2020).
75. Webster, E. A. & Hopkins, D. W. Nitrogen and oxygen isotope ratios of nitrous oxide emitted from soil and produced by nitrifying and denitrifying bacteria. *Biol. Fertil. Soils* **22**, 326–330 (1996).
76. Wrage, N. *et al.* Distinguishing sources of N<sub>2</sub>O in European grasslands by stable isotope analysis. *Rapid Commun. Mass Spectrom.* **18**, 1201–1207 (2004).
77. Sutka, R. L., Ostrom, N. E., Ostrom, P. H., Gandhi, H. & Breznak, J. A. Nitrogen isotopomer site preference of N<sub>2</sub>O produced by *Nitrosomonas europaea* and *Methylococcus capsulatus* Bath. *Rapid Commun. Mass Spectrom.* **17**, 738–745 (2003).
78. Röckmann, T., Kaiser, J. & Brenninkmeijer, C. A. M. The isotopic fingerprint of the pre-industrial and the anthropogenic N<sub>2</sub>O source. *Atmos. Chem. Phys.* **3**, 315–323 (2003).



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## Author contributions

A.T., C.H., Y.W., and W.D. designed the experiment. A.T. performed all the field and laboratory work. A.T. analyzed the data and wrote the manuscript. C.H., Y.W., and W.D. interpreted the data. J.L. and S.L. gave critical comments to improve the manuscript.

## Competing interests

The authors declare no competing interests.

## Additional information

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