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Efficient MoWO₃/VO₂/MoS₂/Si UV Schottky photodetectors; MoS₂ optimization and monoclinic VO₂ surface modifications

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The distinctive properties of strongly correlated oxides provide a variety of possibilities for modulating the properties of 2D transition metal dichalcogenides semiconductors; which represent a new class of superior optical and optoelectronic interfacing semiconductors. We report a novel approach to scaling-up molybdenum disulfide (MoS₂) by combining the techniques of chemical and physical vapor deposition (CVD and PVD) and interfacing with a thin layer of monoclinic VO₂. MoWO₃/VO₂/MoS₂ photodetectors were manufactured at different sputtering times by depositing molybdenum oxide layers using a PVD technique on p-type silicon substrates followed by a sulphurization process in the CVD chamber. The high quality and the excellent structural and absorption properties of MoWO₃/VO₂/MoS₂/Si with MoS₂ deposited for 60 s enables its use as an efficient UV photodetector. The electronically coupled monoclinic VO₂ layer on MoS₂/Si causes a redshift and intensive MoS₂ Raman peaks. Interestingly, the incorporation of VO₂ dramatically changes the ratio between A-exciton (ground state exciton) and trion photoluminescence intensities of VO₂/(30 s)MoS₂/Si from < 1 to > 1. By increasing the deposition time of MoS₂ from 60 to 180 s, the relative intensity of the B-exciton/A-exciton increases, whereas the lowest ratio at deposition time of 60 s refers to the high quality and low defect densities of the VO₂/(60 s)MoS₂/Si structure. Both the VO₂/(60 s)MoS₂/Si trion and A-exciton peaks have higher intensities compared with (60 s) MoS₂/Si structure. The MoWO₃/VO₂/(60 s) MoS₂/Si photodetector displays the highest photocurrent gain of 1.6, 4.32 × 10⁸ Jones detectivity, and ~ 1.0 × 10¹⁰ quantum efficiency at 365 nm. Moreover, the surface roughness and grains mapping are studied and a low semiconducting-metallic phase transition is observed at ~ 40 °C.

The current experiments of integrating 2D TMDCs into nano-electronic devices such as MoS₂, WS₂, and black phosphorous still have challenges like low carrier mobility and low photoluminescence (PL) efficiency which limit their further applications in optoelectronics. It is believed that in 2D TMDCs, the Van der Waals (VDWs) interlayer forces are weak and little defects can enhance the intrinsic phonon scattering and lead to better electrical conduction. Charge traps in 2D TMDCs are found to have a direct relationship with the carrier mobility and the output resistances of the electronic devices¹. Therefore, in order to enhance the Raman intensity, PL intensity, and the charge carrier mobility of the optoelectronic devices, either by an interfacing functional channel material or a functional substrate that modulates the device output in multiple ways, many functional oxides have been used to offer unique properties such as piezoelectricity, strong polarization, and spin injection.

Chemical vapor deposition (CVD) is widely thought to be the most common method for preparing MoS₂. In which MoO₃ and sulfur powders are utilized with some inert gases through two or three-zone quartz tubes.

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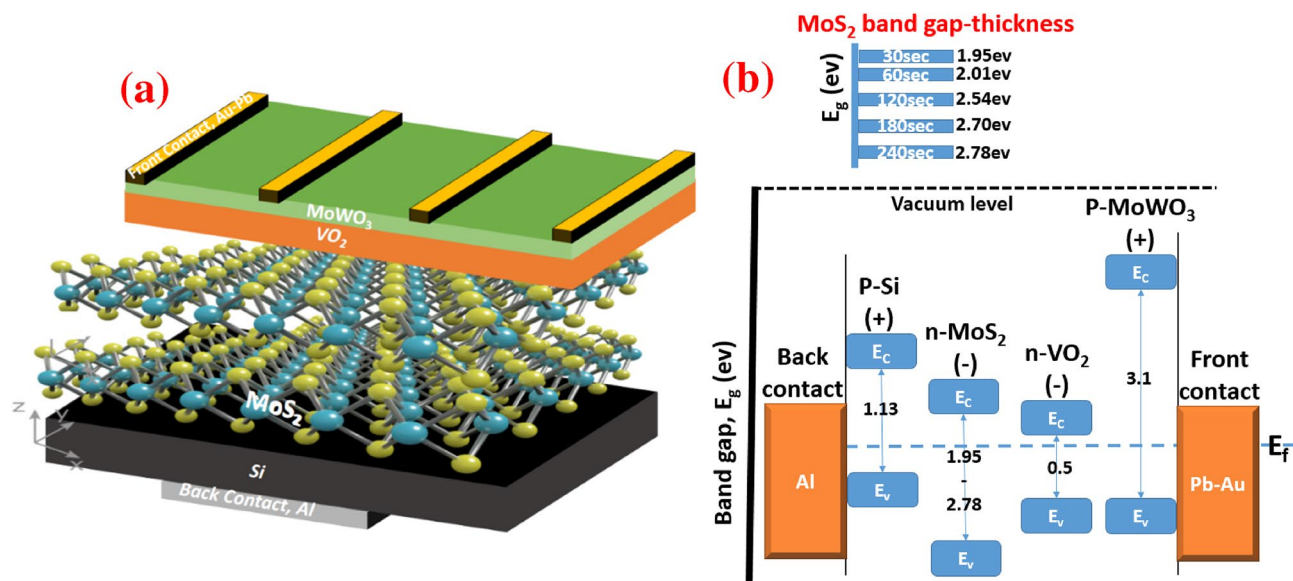


Figure 1. (a) schematic diagram of the proposed MoWO₃/VO₂/MoS₂/Si UV photodetector and (b) its band alignment with different Mo–O sputtering time (30–240 s).

The substrate can be maintained at downstream gas flow to grow MoS₂^{2–4}. Nevertheless, this growth technique has several disadvantages through which non-homogenous, tiny flakes, and 2D MoS₂ in the micro/nanoscale is formed. These disadvantages restrict its application in the industrial optoelectronic devices. On the other hand, methods such as exfoliation approaches, "top-down," have been used to prepare 2D sheets of graphene, MoS₂, etc. This method including different forms such as micromechanical exfoliation⁵, sonication-assisted liquid-exfoliation^{6–8}, shear exfoliation^{9,10} and chemical exfoliation^{11,12}. Micromechanical exfoliation still has some disadvantages such as low quality, small-scale production, and high amount of defects¹³. Another drawback is that the exfoliated MoS₂ must be transferred to a new substratum that handles its scaling and mass production¹⁴. Nonetheless, for the next generation of optoelectronics and quantum computers, the demand for wafer-scale and homogeneous 2D materials such as MoS₂ has increased in recent years. Mainly, these methods including atomic layer deposition (ALD)^{14,15}, pulsed laser deposition (PLD)^{16,17}, thermal evaporation^{18,19}, and magnetron sputtering techniques^{20–22}. Magnetron sputtering was commonly employed at low cost and with easy control in large-scale for commercial manufacturing.

Strongly correlated oxides are a wide range of materials where the associated electronic, magnetic properties, and spin are strongly correlated with each other. The local spin density approximation (LSDA) has been used to determine the energy band structure of many kinds of materials. However, the LSDA fails to describe the electronic structure of some materials in which the interaction among the electrons is strong (e–e interaction) such as strongly correlated electron systems²³. Meanwhile, the dominant role of the Coulomb repulsion forces between the electrons in VO₂ and V₂O₃ systems is opening the insulating gap^{24,25}. However, when strongly correlated oxides interface with 2D TMDCs, multi-functions such as bandgap, charge transfer, energy transfer, and strain can be tuned^{26,27}. Among them, vanadium dioxide (VO₂); an archetypal strongly correlated functional oxide that exhibits a metal–insulator transition (MIT) above room temperature. At ambient pressure, below the transition temperature (T_c), VO₂ has a monoclinic (M1 phase), with space group $P2_1/c$ (#14)²⁸ and lattice constant of $a \approx 5.75\text{\AA}$, $b \approx 4.53\text{\AA}$, $c \approx 5.38\text{\AA}$, $\beta = 122.6^\circ$ ^{29,30}. Above the T_c , VO₂ adopts a tetragonal rutile (R) structure with space group $P4_2/mnm$ (#136) and a lattice constants $a = b \approx 4.55\text{\AA}$, $c \approx 2.85\text{\AA}$ ³⁰.

Due to the superior physical properties at the interface between MoS₂ and VO₂ layer, the manufacture of MoS₂/VO₂ heterostructure for optoelectronic devices has received considerable attention. However, for optical and optoelectronic products, account should be taken of the manufacturing process of homogenous and scaled-up MoS₂. Oliva et al. reported the design of Van der Waals MoS₂/VO₂ photodetector³¹. Nevertheless, in that report, the manufacturing process of MoS₂ includes several steps related to the micromechanical exfoliation process. The photo-excited carriers transfer in MoS₂ and VO₂ was studied, whereas the CVD powder vaporization technique was used for the growth of small flakes of MoS₂³². However, for the easy production of scalable and homogenous MoS₂-based industrial applications, controllability and optimization of deposition time and growth parameters are highly required. Here, we report the preparation, characterization, and optoelectronic application of MoWO₃/VO₂/MoS₂/Si (p–n–n–p) structure towards 365 nm photodetector at room temperature. Different sputtering times of Mo–O have been controlled followed by sulfurization process and interfacing with a monoclinic VO₂ layer. Figure 1a shows a schematic diagram of the proposed MoWO₃/VO₂/MoS₂/Si (p–n–n–p) photodetector. Note that most of the previously reported studies were trying to enhance the Raman, PL, optoelectronic of VO₂–MoS₂ by controlling the device temperature. However, these attempts did not rely for low temperature-phase transition optoelectronic devices. Here, this study focuses on the surface coupling and strain-induced optoelectronic modulation on a few-layer VO₂/MoS₂/Si heterostructure. Figure 1b shows the band alignment of the optimized p–n–n–p-type structure with different Mo–O sputtering times to control the MoS₂

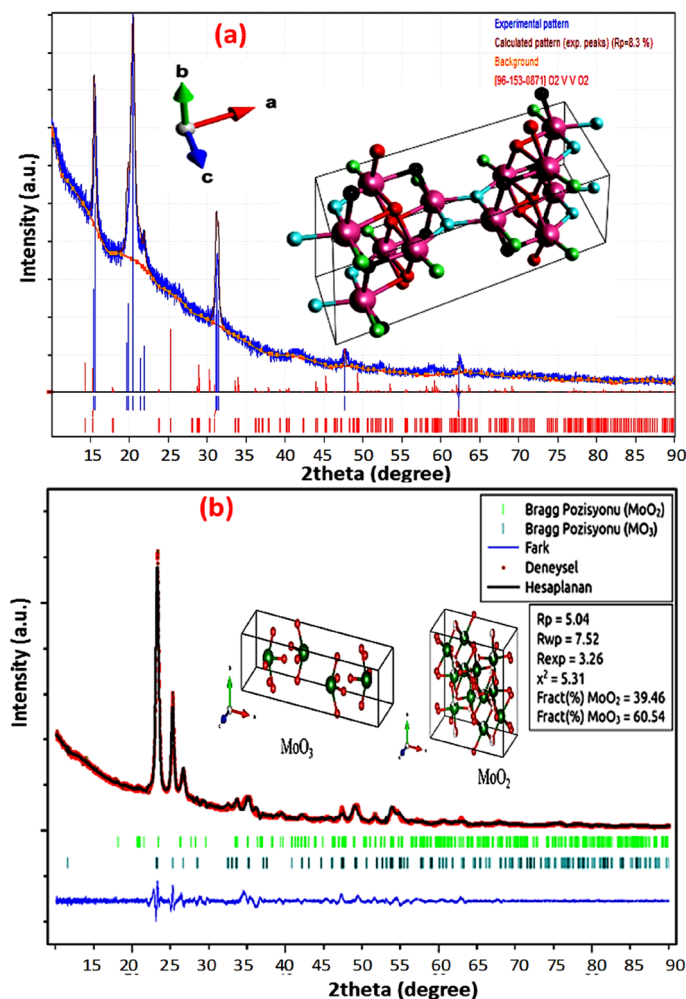


Figure 2. XRD crystal structure of (a) monoclinic VO_2 and (b) Rietveld structure refinement XRD pattern of molybdenum oxide (in our case MoO_2 and MoO_3). The inserted 3D crystal structures are obtained using VESTA 3D visualization program (Model VESTA 3; <https://jip-minerals.org/vesta/en/>).

bandgap³³. The nanostructured thin layer of $\text{Mo}_{0.2}\text{W}_{0.8}\text{O}_3$ was deposited on the surface of VO_2 as a protective and anti-reflection layer³⁴. The optical anti-reflective layer ($\text{Mo}_{0.2}\text{W}_{0.8}\text{O}_3$) is used to improve the responsivity of the photodetector and strongly eliminate the optical interference to minimize its undesired effects. Numerous applications including photodiodes³⁵, image sensor³⁶, and semiconductor photodetectors³⁷ have recorded the uses of the antireflective coating. The structural and morphological characteristics are also studied here. Raman, PL, electrical, optoelectronic characterization of strongly correlated oxide (VO_2), and 2D VDW heterostructure (MoS_2 -Si) will be discussed.

Results and discussion

Structural and composition. The crystal structure of vanadium oxide and molybdenum oxide was calculated in detail to understand the structural analysis of the deposited films. Figure 2a shows the XRD pattern of the deposited VO_2 thin film. XRD result of VO_2 shows a monoclinic phase with a JCPDS card number of [#96-153-0871] with a space group of $C12/m_1(12)$. The inserted crystal structure and visualizing 3D data are obtained using VESTA crystal software. Different diffraction peaks are observed at $2\theta = 15.45^\circ$, 20.49° , 31.28° , 47.69° , and 62.32° . The unit cell parameters are $a = 12.03000 \text{ \AA}$, $b = 3.69300 \text{ \AA}$, $c = 6.42000 \text{ \AA}$ while the angle is $\beta = 106.100^\circ$. However, the Wyckoff position of metal atoms are (2a): (0, 0, 0) and (4f): (0.25, 0.25, 0.25) and oxygen atoms are (4i): (0.175, 0, 0.25), (0.175, 0.5, 0.25) and (8j): (0.075, 0.25, 0.75). The calculated amount of O:V in the VO_2 compound indicating that more oxygen vacancies. Oxygen deficient in vanadium oxide (VO_{2-x}) has reported many times to stabilize the metallic state of VO_2 ³⁸, decrease the semiconductor–metal phase transition (SMT)³⁹ as reported here and to narrow the bandgap in the monoclinic phase⁴⁰. Oxygen vacancies are acting as an electron donor with n-type conductivity which can change the electron orbital occupancy, band structures, and contribute to high photocurrent gain when staking with MoS_2 ⁴¹.

The preparation of molybdenum oxide was achieved using the Mo target at sputtering temperature of 400°C . Figure 2b shows the XRD Rietveld refinement of Mo–O bonding in molybdenum oxide. The calculated R-factors were found to be $R_{\text{wp}} = 7.52$, $R_{\text{exp}} = 3.26$, and $\chi = 25.31$. The Refinement analysis shows that MoO_3 and MoO_2

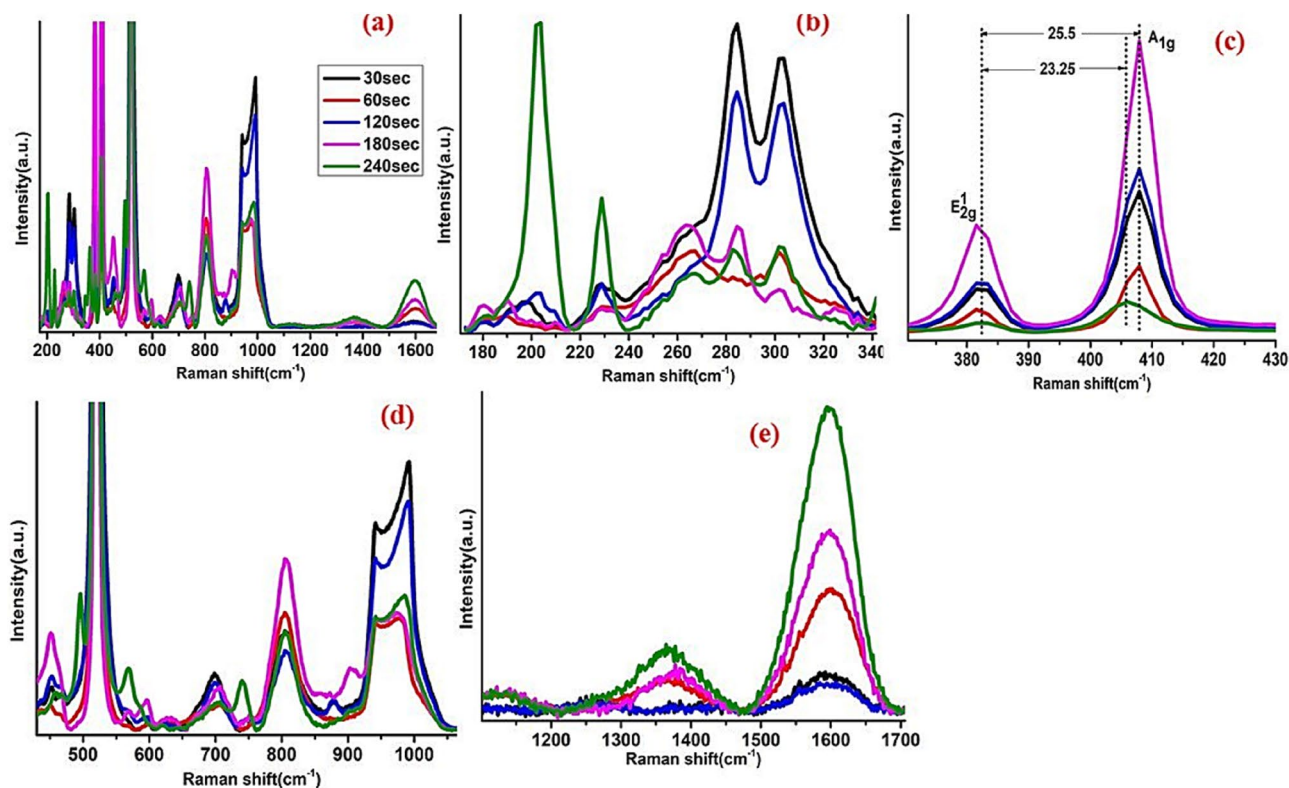


Figure 3. Raman spectra of the deposited multilayer MoWO₃/VO₂/MoS₂ on p-type Si substrate; (a) the full range (200–1700 cm⁻¹) Raman spectra of the deposited films; (b–e) magnified ranges of the full range spectra.

Deposition time	30 s		60 s		120 s		180 s		240 s	
VO ₂ layer	With	Without	With	Without	With	Without	With	Without	With	Without
Position of E _{2g} ¹ (cm ⁻¹)	382.02	384.91	382.16	385.26	382.49	385.93	381.51	385.51	382.51	385.39
Position of A _{1g} (cm ⁻¹)	407.73	410.03	407.61	410.22	407.80	410.39	409.86	410.80	406.14	409.79
Position difference (E _{2g} ¹ - A _{1g}) (cm ⁻¹)	25.71	25.12	25.44	24.96	25.31	24.46	28.31	25.28	23.63	24.40
Intensity of E _{2g} ¹	110,852.8	22,177.5	204,858.9	39,921.1	231,297.2	71,156.9	503,416.9	191,107.0	45,413.2	120,485.6
Intensity of A _{1g}	313,824.8	60,847.9	664,071.8	113,963.2	762,847.3	180,168.1	1,364,933.3	465,300.5	144,965.8	240,574.3
Intensity difference (E _{2g} ¹ - A _{1g})	202,972.0	38,670.4	459,212.9	74,042.1	531,550.1	109,011.2	861,516.4	274,193.5	99,552.6	120,088.7

Table 1. The Raman peaks positions and intensities of VO₂/MoS₂/Si and MoS₂/Si structures.

crystal phases are contributed. The distribution of atoms in unit cells of MoO₂ and MoO₃ was plotted with the VESTA program and shown in Fig. 2b. The amounts of Mo in MoO₂ and MoO₃ were calculated as 1.837 and 1, respectively. While the amounts of O₂ in MoO₂ and MoO₃ were 3.836 and 3.012, respectively. The calculation of the crystal structure was showed that 9.46% as MoO₂ phase and 54% as MoO₃ phase. Moreover, the unit cell parameters of the MoO₂ structure are a = 9.788, b = 8.604, c = 4.714 Å. While for MoO₃ structure are a = 15.309, b = 3.719, c = 3.976 Å.

Raman characterization. The physics behind interfacing structures such as 2D semiconductors and correlated oxides should receive high attention. The importance of these structures can be highlighted by controlling the band alignment of the 2D materials such as MoS₂. Moreover, controlling the carrier mobility, coupling, and strain effect (as reported in the current work)⁴². Raman spectra of the deposited multilayer structure MoWO₃/VO₂/MoS₂ on p-type Si substrate are depicted in Fig. 3. The full range (200–1700 cm⁻¹) Raman spectra of the deposited structures are shown in Fig. 3a and the magnified ranges of the full range spectra are shown in Fig. 3b–e. The Raman peaks of VO₂ at RT are shown in Fig. 3a,b,d which confirm its monoclinic phase. Whereas, Fig. 3c shows the two characteristic peaks of MoS₂ of E_{2g}¹ (385 cm⁻¹) and A_{1g} (405 cm⁻¹) originate from their in-plane and out-of-plane phonon characteristics⁴³. The peaks positions, intensities, distance between E_{2g}¹ and A_{1g} position at different sputtering time of MoS₂ are summarized in Table 1 and Fig. 4.

Many authors tried to study the effect of VO₂ in contact with transition metal dichalcogenides semiconductors (TMDs) layers such as MoS₂ and WS₂^{31,32}. While in our current study, we tried to study the matching behavior

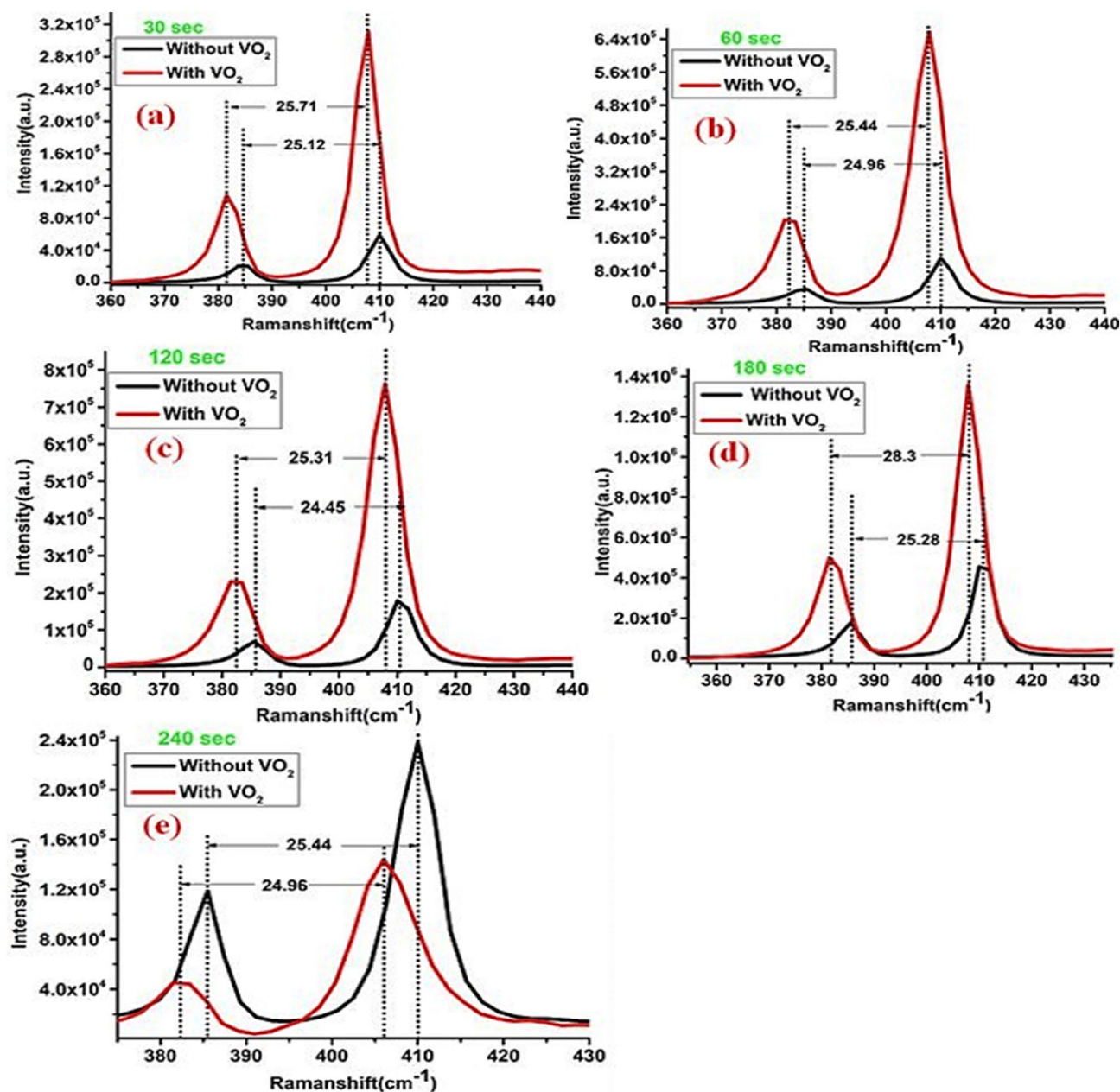


Figure 4. Raman peaks of MoS₂/Si fabricated at different sputtering times in the presence and absence of VO₂ layer on the surface.

of 50 nm thin VO₂ layer onto the surface of MoS₂/Si heterojunction. Additionally, the current work showed that different thicknesses of Mo-O structure have a direct effect on Raman, PL, electrical and optoelectronic characterization of MoS₂ peaks (position and intensity). It is seen from Raman spectra that with incorporated the monoclinic VO₂ layer, there is a redshift and an increase in the intensity peaks as in Fig. 4a–d. These observations showed that MoS₂ and VO₂ were electronically coupled, similar results observed before³². The observed shift in Raman modes may be attributed to the presence of compressive strain induced by implementing the VO₂ layer, similar results observed when interfacing MoS₂ with PMN-PT⁴⁴. Generally, compression stress exerted on VDWS structure decreases the lattice constant^{44,45} and consequently increases the film crystallinity and photocurrent as reported here.

It is known that the intensity of Raman peaks is referring to high crystallinity effects. In most cases, Raman scattering is sensitive to the degree of crystallinity in a sample. Typically a crystalline material yields a spectrum with very sharp, intense Raman peaks, whilst an amorphous material will show less intense Raman peaks^{46–48}. Table 1 and Fig. 4 show the Raman intensities of the corresponding thin films. These results show high attention to the applications of enhancing the Raman signal/intensity. The difference between peak positions of E_{2g}^1 and A_{1g} in the presence and absence of VO₂ layer are deposited in Table 1. It shows a decrease with increasing the MoS₂ layer thickness from 30 to 240 s for both MoS₂/Si and VO₂/MoS₂/Si structures, which may contribute to

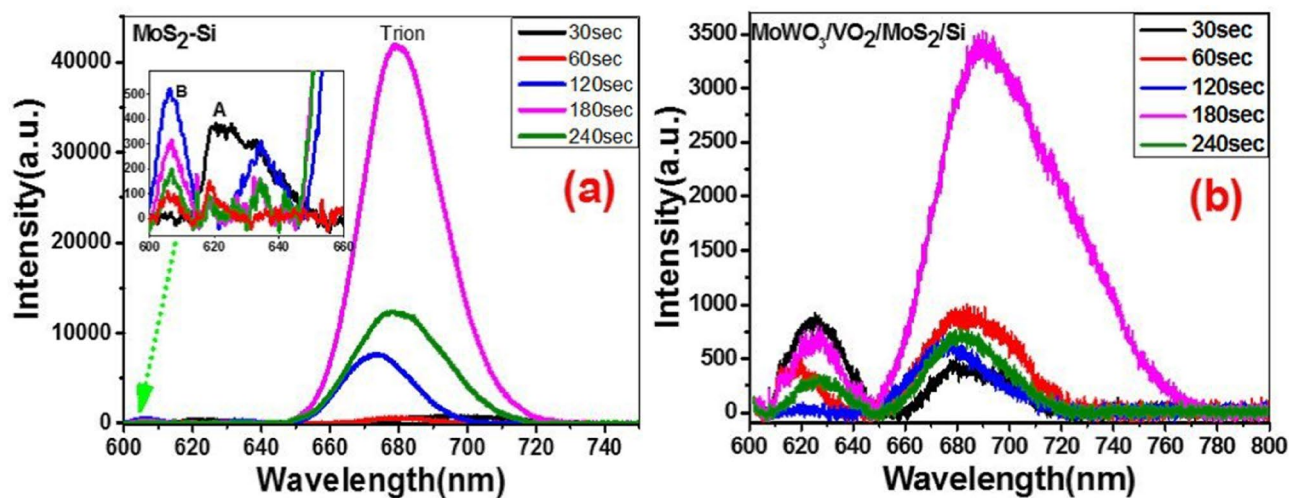


Figure 5. PL spectra of (a) MoS₂-Si heterojunction and (b) MoWO₃/VO₂/MoS₂-Si photodetectors.

a decrease in the film layers and enhance the band gap⁴⁹ as seen from Table 1. However, the film deposited at 180 s is out of this base. Meanwhile, the differences between the intensities of the peaks show an enhancement in the peak intensity with increasing the MoS₂ film thickness which may contribute to high crystallinity effects. In counter, the film deposited at 240 s shows a decrease in the intensity. The highest intensity was observed for VO₂/MoS₂/Si with 180 s, whereas the lowest intensity was attributed to 240 s film. These results are concluded that the MoS₂ sputtering time of 180 s is optimized for VO₂ and MoS₂ optical coupling. Consequently, the observed results may highlight the importance of incorporating strongly correlated oxide through 2D VDWs MoS₂ structure to control film crystallinity, surface-enhanced Raman spectroscopy (SERS) of MoS₂ for better signal detection and spatial resolution^{50–52}, optical coupling^{53,54}, Plasmonic local-field enhancement⁵³ and optoelectronic behavior⁵¹.

For more details about intensity distribution versus position, Raman mappings of E_{2g}¹, A_{1g} and Si peaks of MoS₂/Si at different sputtering time of Mo-O (30, 60, 120, 180, and 240 s) are investigated at 385, 410, and 520 cm⁻¹, respectively, and shown in Fig. S3 (a, b, c, d, and e). The Raman mapping was realized at large area (1.6 mm × 1 mm) with 320 × 200 data point using Leica microscope at 5 × magnification to provide an evidence about the ability to scale up the MoS₂ thin films. The wavelength, power and integration time of the used laser were 532 nm, 3 mW and 1 s, respectively. This analysis illustrated that 30, 60, and 120 s samples show high homogeneity over the full scale (1.6 × 1 mm²). Although, 180 s sample shows regular agglomerations of nanoparticles for both E_{2g}¹ and A_{1g} positions, the 240 s sample shows irregular clusters. The scaling up of MoS₂ can therefore be demonstrated until sputtering time < 180 min.

Photoluminescence measurements (PL). The PL spectrum of MoS₂ is strongly dependent on the number of layers of 2D VDWs structures. In other words, a strong PL peak may observe in single-layer MoS₂ or WS₂ film and decreasing with increasing the number of layers⁵⁵. Trion is defined as a quasi-particle that can potentially carry out more information and data than electrons for which make them useful towards different applications such as optoelectronics and quantum computing⁵⁶. Trions are consisting of three charged particles bound together by very weak bonding energy that makes them quickly fall apart⁵⁷. It is known that the dominated peak in Figs. 5 and 6 attributed to the recombination of the photogenerated electron-hole pair, whereas the observed weaker peak at lower wavelength may be attributed to the valance band splitting due to the presence of strong spin-orbit coupling of MoS₂⁵⁸. In the literature review, the definition of trion and exciton peak is dominated by their locations and the trion peak is located at lower energy than exciton peak^{59–61}. Figure 5a and b shows the PL spectra of MoS₂/Si and MoWO₃/VO₂/MoS₂/Si structures at RT. The MoS₂/Si has two peaks at ~679 and ~620 nm which corresponding to components from trion and A-exciton⁶². However, trion and A-exciton positions have changed by controlling the deposition time of MoS₂ as seen in Table 2. Also, this table shows the position of the characteristic peaks (A-exciton, B-exciton (higher spin-orbit splitting state), and trion) in the presence and absence of the VO₂ layer. Generally, it is seen that the trion peak shifted to lower energy by implementing the VO₂ layer onto the surface of the MoS₂/Si structure as seen in Table 2.

The A-exciton peak has a higher intensity in the case of the VO₂/MoS₂/Si structure than in the case of MoS₂/Si structure for 30 and 60 s deposition time of MoS₂ layer, Fig. 6a and b. The trion peak is enhanced and shifted to longer wavelengths by increasing the deposition time to 60 s. The increase in PL intensity refers to an enhancement in light emission efficiency and increases the density of states of the photo carriers by modifying the band structure⁴⁴. At 30 s, the A-exciton peak intensity is higher for the VO₂/MoS₂/Si structure than the MoS₂/Si structure. i.e., the incorporation of VO₂ dramatically changes the ratio between photoluminescence intensities of A-exciton and trion from < 1 to > 1 for VO₂/(30 s)MoS₂/Si structure. However, the opposite case is observed for the trion. Meanwhile, by incorporating the VO₂ layer, a blue shift in the trion peak is observed, while a redshift is observed for the A-excitons. The observed peak position of PL that shifted towards lower energy (redshift) attributed to the non-radiative electron-hole recombination effect. However, by increasing the deposition time to 60 s, the A-exciton peak have higher intensity compared with the trions peak in the case of VO₂/MoS₂/Si

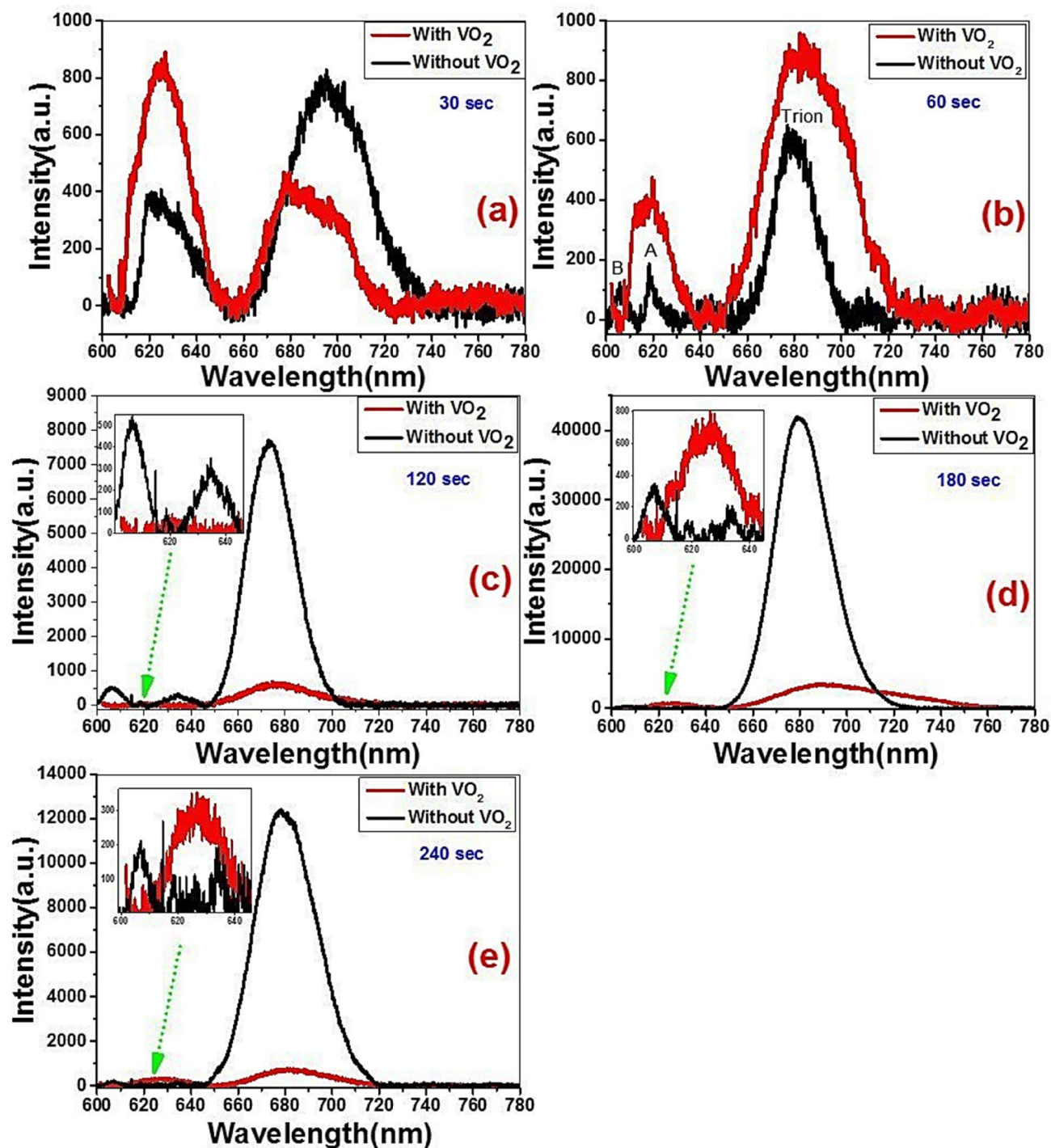


Figure 6. PL spectra of MoS₂/Si fabricated at different sputtering times without and with VO₂ supporting layer on the surface.

structure than MoS₂/Si structure as seen in Fig. 6b. By increasing the deposition time of MoS₂ from 60 to 180 s, the B-exciton/A-exciton relative intensity increases, whereas the lowest ratio at 60 s deposition time refers to the high quality and the low defects densities of VO₂/(60 s) MoS₂/Si structure. Moreover, a redshift was observed for the trion peak and a slight blue shift when incorporating the VO₂ layer. The increase in PL intensity refers to an enhancement in light emission efficiency and increases the density of states of the carriers by modifying the band structure and consequently enhance the radiative recombination of carriers, similar results were observed in a compressively strained trilayer MoS₂ sheet^{44,63,64}. This result shows that strong coupling between VO₂ and MoS₂ at 60 s Mo-O deposition time was observed at room temperature. While the PL intensity with MoS₂/VO₂ structure has only enhanced with increasing the film temperature^{31,32,65}, while our reported results show a dramatic enhancement in the PL intensity at RT by incorporating VO₂ layer on the surface of MoS₂/Si structure.

MoO ₃ deposition time	30 s		60 s		120 s		180 s		240 s	
VO ₂ layer	With	Without	With	Without	With	Without	With	Without	With	Without
Peak position (nm)										
Trion	678.65	696.06 Blue shift	682.47	679.32 Red shift	676.29	673.08 Red shift	690.40	679.54 Red shift	681.32	678.78 Red shift
A-exciton	625.08	622.21 Red shift	617.20	618.11 Blue shift	602.82	634.7 Blue shift	624.97	633.50 Blue shift	626.15	635.12 Blue shift
B-exciton				606.80		605.69		606.41		606.69
Intensity										
Trion	425.54	782.63	939.08	609.64	621.29	7554.96	3383.10	41,642.54	712.12	12,328.14
A-exciton	955.40	378.21	401.69	136.24	64.85	295.21	658.62	134.53	292.37	145.70
B-exciton				88.73		514.81		300.94		209.37

Table 2. The PL peaks positions and intensities of VO₂/MoS₂/Si and MoS₂/Si structures.

On the other hand, trion peak quenching was found in Fig. 6c,d,e; with increasing the Mo-O layer from 120, 180, and 240 s when VO₂ is deposited on MoS₂/Si structure. We thought that the quenching of PL spectra in Fig. 6c,d,e may be owing to the fact that MoS₂ is an n-type with a close Fermi level to the conduction band. However, the deposited VO₂ layer at thicker Mo-O layer (120, 180, and 240 s) may shift Fermi level to the mid-band gap by drawback the electron coupling of VO₂ and MoS₂^{66,67}. Similar results observed using back-gating with SiO₂/Si⁶⁶, dopants molecule like F4-TCNQ, metal-centered Phthalocyanine molecules on the surface of monolayered TMD^{55,68}. It is interesting to note that in all Fig. 6a–e, A-excitons have higher intensities when incorporated VO₂ layer. This result draws high attention for enhancing A-exciton peak intensity and raises strong spin-orbit coupling by incorporating the monoclinic VO₂ thin layer. Moreover, the B-exciton peak was observed in the MoS₂/Si structure at 606 nm (2.04 eV), but it did not appear in the VO₂/MoS₂/Si structure as seen in the insets of Fig. 6b–e. It is known that the PL spectra of MoS₂, surprisingly, increases with decreasing layer thickness⁶². However, the origin of PL spectrum in MoS₂ arises from the direct excitonic electronic transitions which shows higher radiative recombination rate than nanocrystals⁶⁹. Therefore, the enhanced photoluminescence with increasing the deposition time of Mo-O has to be attributed to a dramatically slower electronic relaxation factor κ_{relax} as in Eq. 1, suggesting a substantial change in electronic structure of MoS₂ when going from the short to longer deposition time of Mo-O as seen in Fig. 6.

$$\eta_{Lum} \sim \frac{\kappa_{rad}}{(\kappa_{rad} + \kappa_{defect} + \kappa_{relax})} \quad (1)$$

where κ_{rad} , κ_{defect} , κ_{relax} , and η_{Lum} are representing the rates of radiative recombination, defect trapping, and electron relaxation, Luminescence quantum efficiency within the conduction and valence bands, respectively.

It is concluded that by depositing the VO₂ layer on MoS₂/Si structure, both trion and exciton peaks get shifted as seen in Table 2. It is implemented that the presence of the VO₂ layer on the surface of the MoS₂/Si structure results in a redshift through trion peaks, while a blue shift for A-exciton. The peak position of PL for trion is shifted towards lower energy due to the occurred non-radiative electron-hole recombination. However, 30 s sample is out of this rule with a blue and redshift in the trion and A-exciton peak on the VO₂ layer, respectively.

Surface topography and grain boundary mapping. AFM has been used to investigate the surface topography, roughness, grain, and grain boundary mapping. Surface topography and parameters such as average roughness R_a (nm), root mean square or standard deviation of the height value R_q (nm), height different or peak-to-valley (R_{pv}), ten-point height (R_z), skewness (R_{sk}) and kurtosis (R_{ku}) as well as fractal and grain analysis were inspected by the XEI software. Figure 7a,d,g,j,m shows a 2D surface topography, Fig. 7b,e,h,k,n shows the 3D visualization, and Fig. 7c,f,i,l,o shows the grain boundary mapping of the prepared MoWO₃/VO₂/MoS₂/Si thin film with sputtering time of 30, 60, 120, 180, and 240 s, respectively. The films that deposited at short deposition time show higher uniformity, while with increasing the deposition time a small clusters of different sizes less than 100 nm have been observed. The average roughness values of the prepared thin films have summarized in Table 3 and show that 30 s and 180 s thin films have the lowest and highest R_a value of 3.28 and 48.0 nm, respectively. It seems that with increasing the deposition time of Mo-O, the accumulated nanoparticles show bigger sizes, consequently higher roughness factors. The calculation of the grain and grain boundaries of interfacing thin films are important parameters that provided information about the nature of interfaces between two layers. Figure 7c,f,i,l,o shows the grain size and grain boundary distribution maps of the prepared thin films. The SEM images of VO₂, MoO₃, Mo_{0.2}W_{0.8}O₃, and MoS₂/Si thin films are presented in Figures S1 and S2 and discussed in supplementary information. Large scale MoS₂ thin films have been studied in our previous work by combining CVD and sputtering technique⁷⁰. SEM images (Fig. S2, supplementary information) are provided to illustrate a homogenous and approximately uniform nanoparticles distribution. The SEM images in Fig. S2 (supplementary information) are provided to illustrate the homogeneity and approximate uniform surface distribution of nanoparticles, particularly for sputtering time < 180 s. Also, the AFM images of the MoWO₃/VO₂/MoS₂-Si thin films, Fig. 7, show almost uniformly distributed grains for sputtering time up to 120 s. Besides the presented

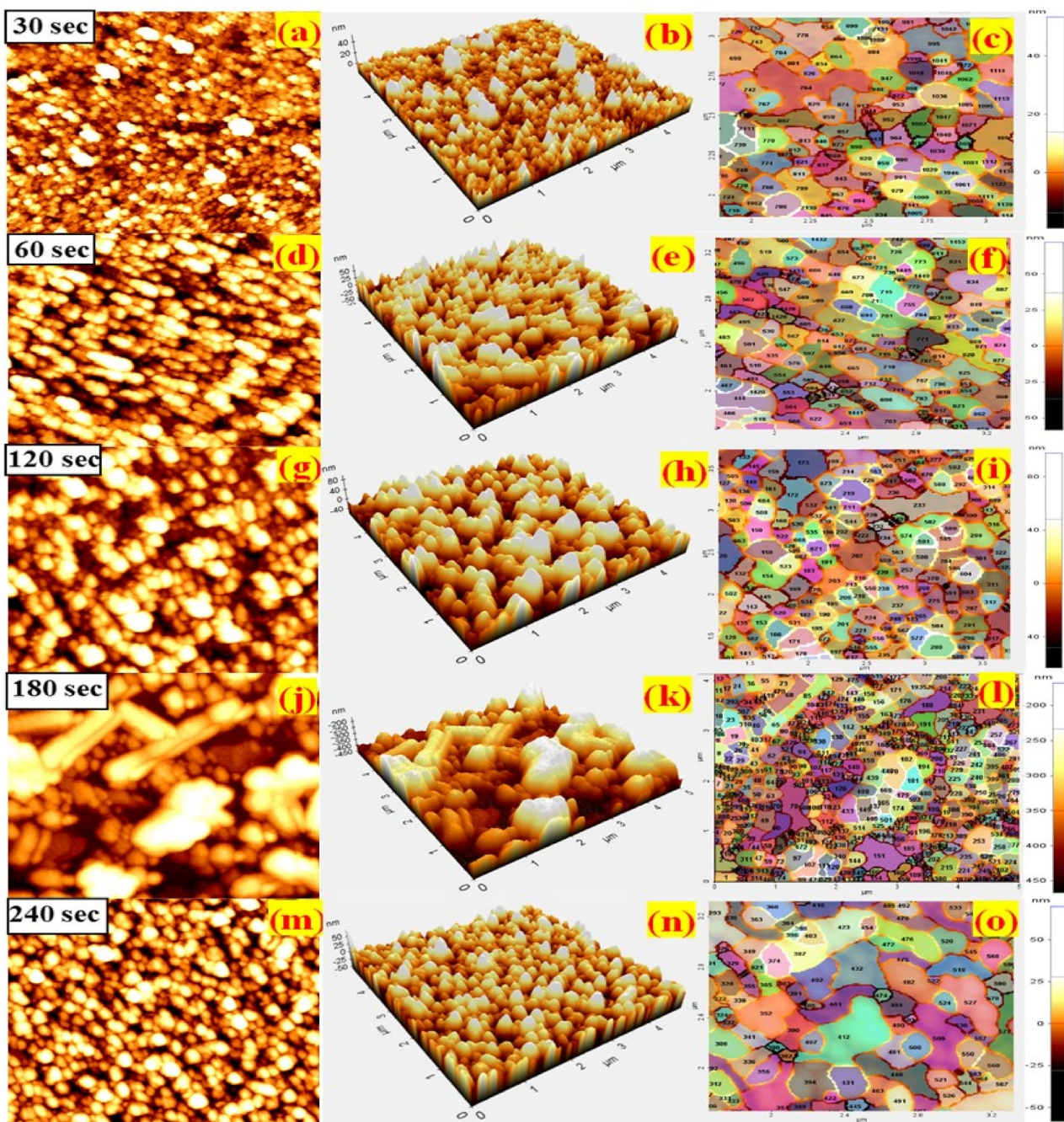


Figure 7. 2D AFM images (a, d, g, j, and m), 3D AFM images (b, e, h, k, and n), and the grain/grain boundary mapping (c, f, i, l, and o) of the prepared MoWO₃/VO₂/MoS₂-Si thin film with sputtering time of 30, 60, 120, 180, and 240 s, respectively. XEI software was used for image analysis and processing. [XEI PSIA, Version 1.5, <https://parksystems.com/102-products/park-xe-bio>].

Parameters	30 s	60 s	120 s	180 s	240 s
R _{pm} (nm)	26.364	76.913	91.878	226.585	73.849
R _q (nm)	4.458	15.987	22.246	55.048	12.934
R _s (nm)	3.283	13.460	18.430	48.080	10.688
R _z (nm)	22.126	59.141	74.573	161.619	48.946
R _{sk}	-1.143	0.026	-0.196	-0.172	-0.091
R _{ku}	5.362	2.333	2.143	1.973	2.895

Table 3. The roughness parameters of MoWO₃/VO₂/MoS₂/Si photodetector.

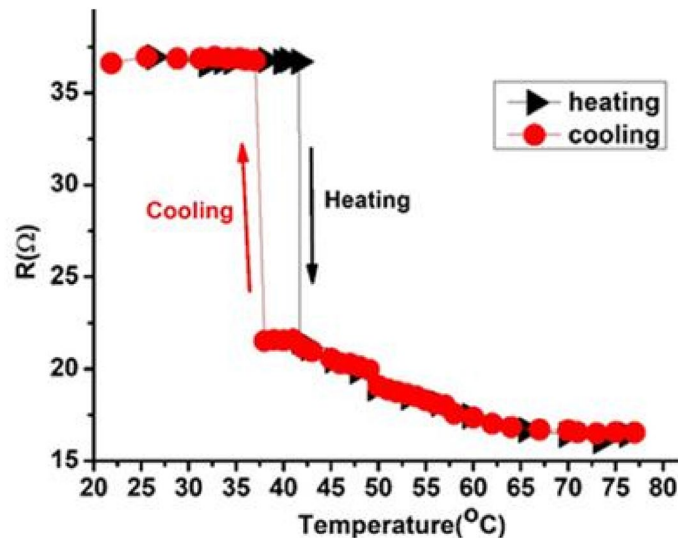


Figure 8. Semiconductor–metal phase transition of 50 nm thin VO₂ layer from room temperature to 78 °C.

grain/grain boundaries mapping in Fig. 7, the scanning area of the AFM images was selected to be 5 μm × 5 μm to provide evidence about the ability to scale up the MoS₂ thin films.

Temperature-resistance measurement (T-R). The phase transition of the prepared VO₂ thin film has been performed using a four-probe measurement system connected to a heating stage ranging from RT to 78 °C. We investigated the influence of the 50 nm VO₂ thin layer on Raman, PL, and optoelectronic measurements of a few-layers MoS₂ with different sputtering times of Mo-O layer. So, the electrical semiconductor–metal phase change of VO₂ has been tested as depicted in Fig. 8. In our case, the VO₂ phase transition temperature was calculated to be 40 °C by controlling the sputtering condition (high vacuum and long-time annealing temperature) which may affect the lattice-strain and oxygen vacancy concentrations of VO₂^{71–73}. The reason for the low semiconductor-metallic phase transition may owing to the high concentration of oxygen vacancies. Oxygen deficient in vanadium oxide (VO_{2–δ}) has reported many times to stabilize its metallic state¹⁰, decrease the semiconductor–metal transition (SMT)¹¹ and narrowing the bandgap of the monoclinic VO₂ phase¹². Oxygen vacancies are electron donors with n-type conductivity which can change the electron orbital occupancy and band structures and contribute to the high photocurrent generation when staking by MoS₂¹³. To check these coupling effects between these layer, UV-optoelectronic measurements have been carried out.

Electric and optoelectronic properties. This section discusses the electric characterization of MoS₂/Si heterostructures before and after depositing the VO₂ layer under dark and UV conditions. In order to investigate the I–V and photoresponse of the prepared devices, we measured the I–V curve under dark and upon UV light illumination by applying a sweep voltage from –5 to +5 V for different sputtering times of Mo-O as shown in Fig. 9. This figure shows the electrical and optoelectronic properties of MoWO₃/VO₂/MoS₂/Si thin film with different thicknesses of the MoS₂ layer. Figure 9a–e shows the current–voltage (I–V) curves under dark and UV illumination at MoS₂ deposition time of 30, 60, 120, 180, and 240 s, respectively. The observed photocurrent in this figure is larger than that reported for previously proposed MoS photodetector with lateral contacts arrangement³¹. The back Al and front Pd–Au contacts may play important role in that because these contacts allow the vertical electron transport in the heterostructure photodetector besides the lateral electron transportation and consequently 2D conductivity measurements^{31,37}. The vertical electron transport offers a high density of active edge sites³⁷. Also, our optimized heterojunction photodetector does not have a high-resistance layer like the SiO₂ layer that was previously used and obstruct the vertical electron transport³¹. The back Al contact has been used for better collecting signals. Al metal makes Ohmic contact type with p-Si, which is also observed when probed on two contact pads on the same side, however, noble metals such as Ag, Au, etc. make Schottky contact with p-Si. On the other hand, the Au–Pd front contact was built in the anti-reflection Mo_{0.2}W_{0.8}O₃ layer, in which the formed Schottky barrier height and width could be controlled by the current passing through the metal–semiconductor contact. Under dark conditions and as predicted by Basyooni et al., a non-linear I–V curve was obtained indicating that a good double-Schottky contact behavior was formed between the front Pd–Au contact and the film surface³⁴. The position of asymmetric metal contacts can provide an integrated potential gradient assigned to the work function difference of asymmetric electrodes as previously stated for various applications, such as gas sensors³⁴ and photodetector⁷⁴, which leads to enhanced device performance as reported here. For instance, Casalino et al. used an asymmetric Al–Si–Cu (metal–semiconductor–metal) structure-based Si photodetector⁷⁵. Moreover, several studies used asymmetric metal contacts for the photodetector application to control the dark current^{70,74}.

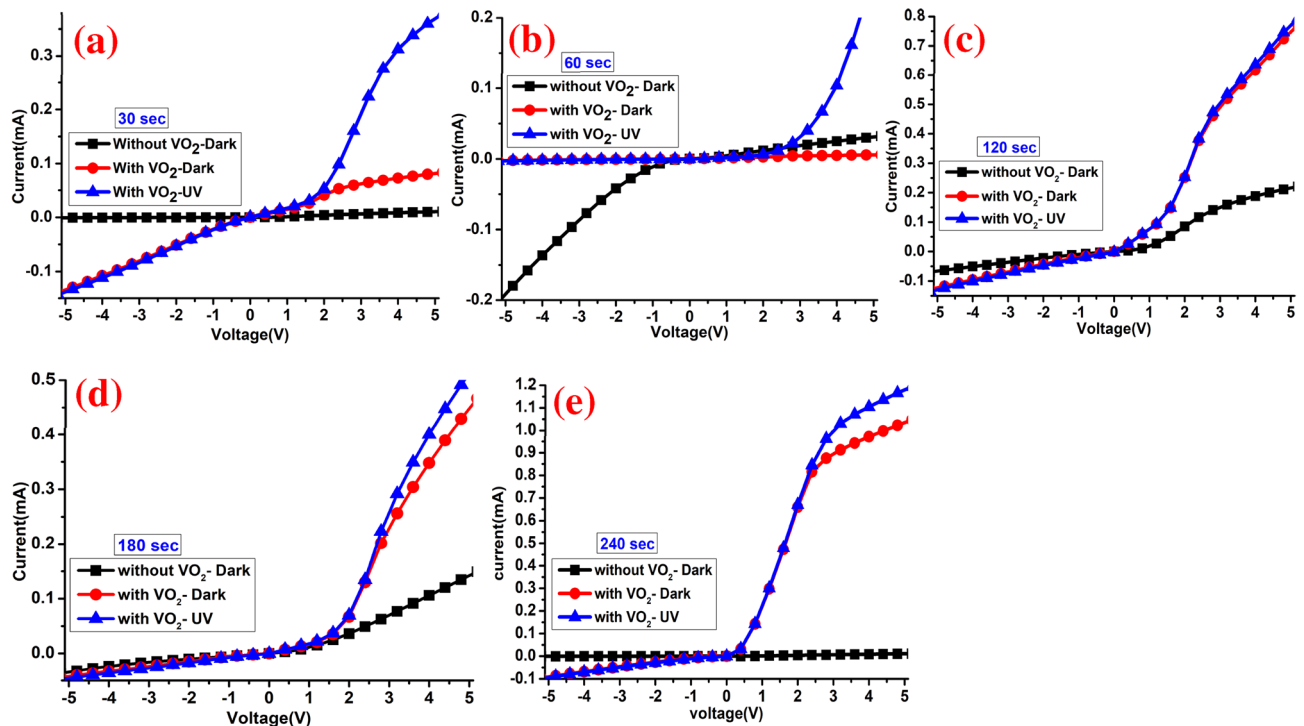


Figure 9. The linear I–V characteristics of the MoWO₃/VO₂/MoS₂/Si device in dark and under UV illumination with MoS₂ sputtering time of (a) 30, (b) 60, (c) 120, (d) 180, and (e) 240 s.

Figure S4 shows the linear and semi-logarithmic scale current–voltage characteristics of MoS₂/Si device without VO₂ layer under the dark condition with different sputtering times of MoS₂ layer; 30, 60, 120, 180, and 240 s. The positive part shows an increase in the associated dark current with increasing the sputtering time from 30 to 60 s. While at 120 s, a jump in the forward dark current is observed due to the related folding-effects in MoS₂. Folding effect decreases the interlayer coupling and enhances the photoluminescence emission yield of A- and B-exciton peaks as seen in Fig. 5. Whereas instability measurements were observed at the negative bias part. The large increase in the negative dark current for MoS₂ (60 s) may be attributed to the release of charges that were trapped on MoS₂'s surface at the interface trap sites (oxygen sites). The highest reverse dark current, which suggests the lowest potential barrier, was observed at 60 s. This may be ascribed to the values of the optical band gaps as shown in our previous study, whereas the 60 s MoS₂-Si thin film displayed optical band gaps of 1.75 and 2.01 eV³³. By increasing the sputtering time the reverse dark current decreases and almost becomes identical for sputtering time \geq 180 s, as shown in Fig. S4(a,b).

Figure 10a–e shows the log-current curves under dark and UV illumination of 30, 60, 120, 180, and 240 s conditions, respectively. It is clearly seen that UV illumination shifts the logarithmic I–V curve towards the negative voltage region. This behavior may address the induced strain effects from the VO₂ layer or unidirectional charge transport mechanism from the top to bottom layer due to the different electron concentrations⁷⁶. Interestingly, it seems that the VO₂ layer enhances the positive and negative current. Meanwhile, the dark current obtained after depositing the VO₂ layer on the surface of MoS₂/Si is about 2–3 folds' improvement over pure MoS₂/Si device for 30 and 60 s samples, as in Fig. 10a,b. The observed higher value of photocurrent under UV illumination is also attributed to enhancement through the band-to-band excitation in the VO₂/MoS₂/Si region. Moreover, carrier recombination and tunneling across the device junction may be addressed as a reason for the enhancement I–V under UV illumination. Nevertheless, fast response and recovery times, high responsivity, high reliability, and low signal-to-noise ratio are important characteristics for detector applications^{77,78}, which is discussed below in detail.

Transient response. Figure 11a–e shows the ON–OFF time-resolved photoresponses of MoWO₃/VO₂/MoS₂/Si devices with varying MoS₂ sputtering time (30, 60, 120, 180, and 240 s) in dark and under UV illumination. These optoelectronic transient/dynamic curves were measured with a switching time of 5 s, a bias voltage of 1 V, and UV illumination of 365 nm. Also, the optical modulations (photocurrent-time characteristics@ 1 V) under dark and UV illuminations of Mo_{0.2}W_{0.8}O₃/VO₂/MoS₂/Si devices are presented in Figure S5 (supplementary information). Note that the ON/OFF and OFF/ON transitions of the UV light source are repeated many times for each 5 s at a bias voltage of 1 V. The time-resolved photoresponse curves in Fig. 11 show different sputtering time dependence. The curves at 30 and 120 s show increasing on/off behaviors, whereas the curves at 60, 180, and 240 s show stable on/off behaviors. The increasing on/off behaviors in Fig. 11a,c may come from some organic trap states that accumulated during the CVD sulfurization process. While, the stable on/off behaviors in Fig. 11b,d,e can be attributed to the high stability, high quality, and low density of the defects.

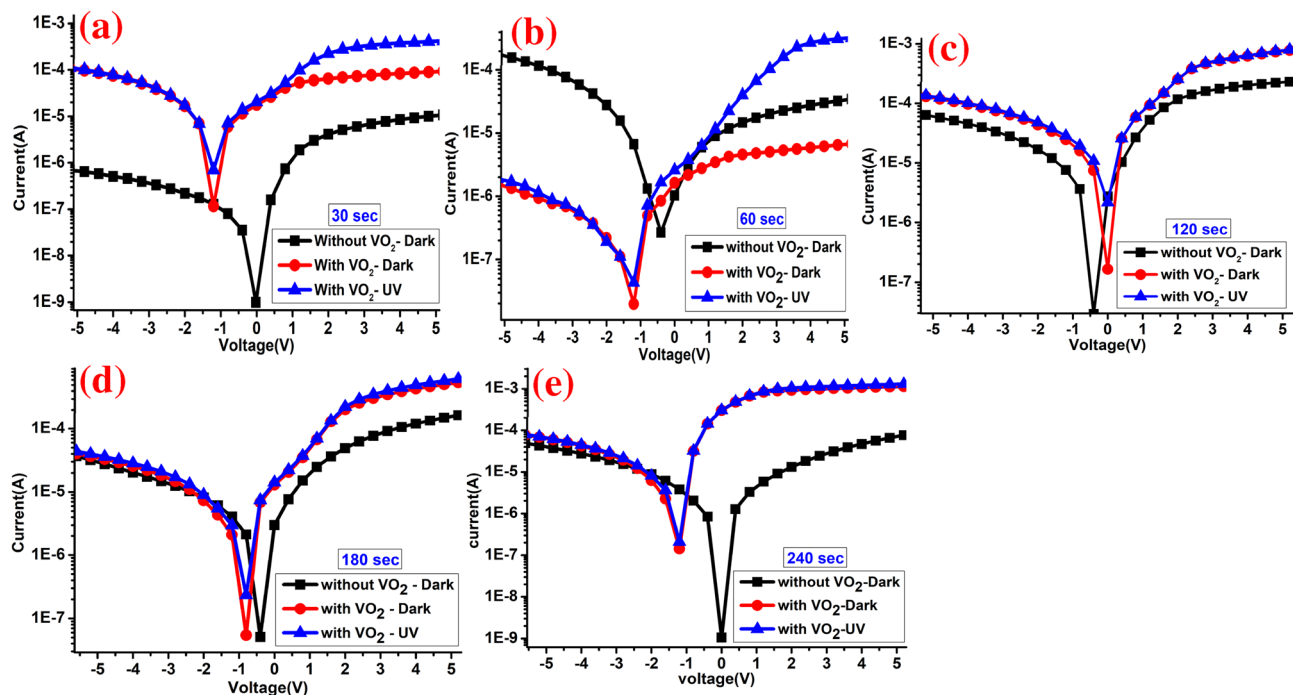


Figure 10. The semi-logarithmic scale I–V of MoWO₃/VO₂/MoS₂/Si device under dark and UV illumination with MoS₂ sputtering time of (a) 30, (b) 60, (c) 120, (d) 180, and (e) 240 s.

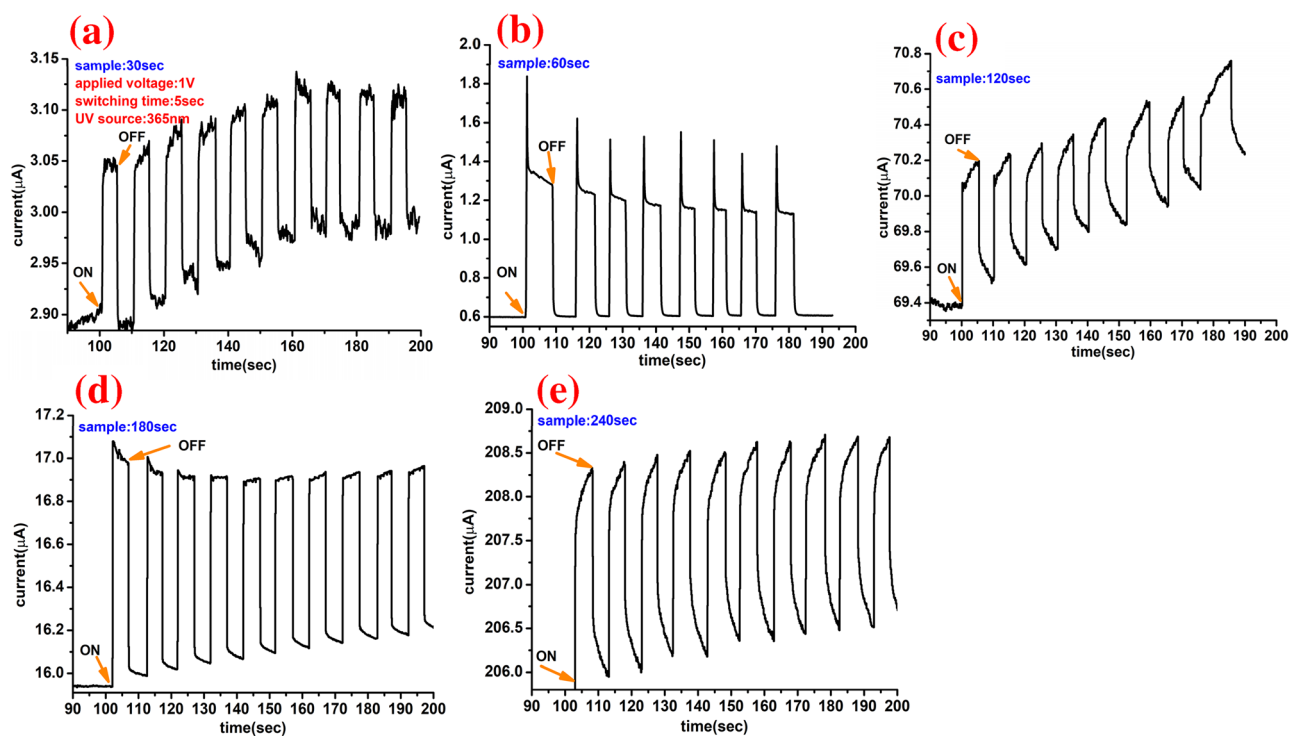


Figure 11. ON/OFF time-resolved photoresponse of the MoWO₃/VO₂/MoS₂/Si device with MoS₂ sputtering time of (a) 30, (b) 60, (c) 120, (d) 180, and (e) 240 s. All measurements were carried out at a bias voltage of 1 V, UV illumination of 365 nm, and switching time of 5 s.

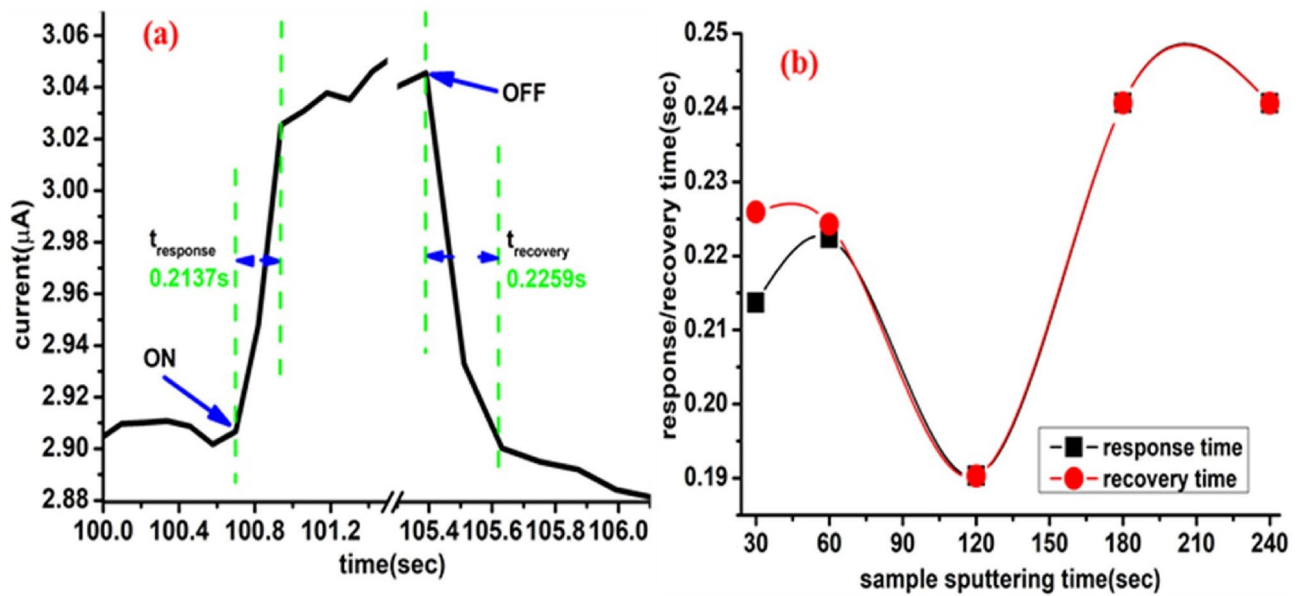


Figure 12. (a) Enlarged view of a single current–time photoresponse cycle for MoWO₃/VO₂/MoS₂ (30 s)/Si UV photodetector to identify the response and recovery times and (b) the estimated response and recovery times as a function of the sputtering time.

The response/raise time was measured when the light source turned on, while the recovery/decay time was measured when the light turned off as shown in Fig. 12a. The response and recovery times have been estimated from the ON/OFF dynamic photoresponses at different sputtering times, Fig. 11. The combined sputtering and CVD deposition process of MoWO₃/VO₂/MoS₂/Si UV photodetector device shows symmetrical response and recovery time which not exceed 0.25 s using the selected wavelength (365 nm) and bias (1 V), as shown in Fig. 12b. Consequently, our proposed photodetector is considered more efficient than the previously tested photodetector by Ang et al.⁷⁸

The fast response and recovery speed, indicating that electron–hole pairs could be effectively generated and separated in the proposed structure under UV illumination at room temperature. It is important to note that the fabricated device using MoS₂(120 s) shows the fastest response/recovery times (0.19 s at 1 V) among the studied devices as shown in Fig. 12b. The fast response/recovery time at 120 s can indicate the fast and stable generation and separation of the electron–hole pairs. Unlike Dhyani and Das who reported rapid response for the silicon–MoS₂ photodetector@ 580 nm and 3 V bias⁷⁹, our measurements are performed without any external series resistance. Nevertheless, it is known that higher applied bias voltage can generate more photocurrent and consequently decrease the response and recovery time. A clear high photocurrent can be observed in the ON state at 1 V which makes the gate voltage lowers the potential barriers at the contacts, resulting in highly efficient photogenerated carrier extraction and thus increased photocurrent at a low applied voltage (1 V). The reason behind this is that the gate voltage can affect the height of the Schottky barrier between the metal contact and film surface and thus shift the Fermi level^{80,81}. It seems that our designed photodetector did not require high bias voltage which makes it more applicable for low power photodetector technology.

Photocurrent gain (P_g) and photoresponsivity (R_i). The induced photocurrent I_{ph} is given by $I_{ph} = I_{Light} - I_{Dark}$, where I_{ph} increases with increasing the applied voltage and the light power⁸². Photocurrent gain (P_g) can be defined and determined by $P_g = (I_{photo} - I_{dark})/I_{dark}$, where I_{photo} and I_{dark} are photocurrent and dark current respectively⁷⁷.

Also, the detector responsivity (R_i) can be expressed as $R_i = \Delta I / (A \times P)$, where ΔI is the difference between the photocurrent and dark current, A is the illuminated area, and P is the UV light power. Figure 13a shows the photocurrent and photocurrent gain of the tested samples under a 365 nm UV illumination source. Figure 13b demonstrates the responsivity in A/W of MoS₂/Si and MoWO₃/VO₂/MoS₂/Si photodetectors with different MoS₂ sputtering time (30–240 s). In both photodetectors, the responsivity values reflect nearly linear increases by increasing the sputtering time to 180 s. MoWO₃/VO₂/MoS₂/Si UV detector responsivity shows high values ranging from 0.3 to 4.7 A/W, which corresponds to 30 to 180 s MoS₂ sputtering time. The optimized value at 180 s is higher than that reported by Li et al. (2.4 mW cm⁻²)⁸³. The enhanced photoresponsivity by interfacing the VO₂ layer may be owing to the film strain include stresses arising from the different thermal expansion coefficients of the VO₂ and MoS₂/Si film due to a high deposition/sulphurization temperatures of ~400/650 °C and growth stresses arising from crystal structure changes after deposition. Nevertheless, more efficient light absorption involving more *e*–*h* pairs generation, resulting in higher mobility and more detection capability.

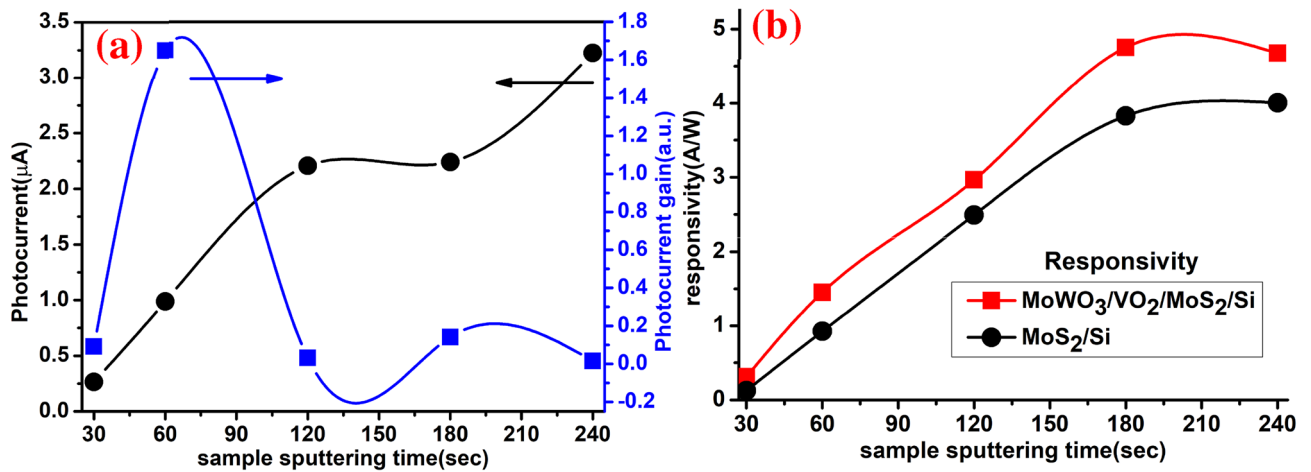


Figure 13. (a) the photocurrent and photocurrent gain of MoWO₃/VO₂/MoS₂/Si photodetector and (b) responsivity for both MoS₂/Si and MoWO₃/VO₂/MoS₂/Si photodetectors as a function of the MoS₂ sputtering time.

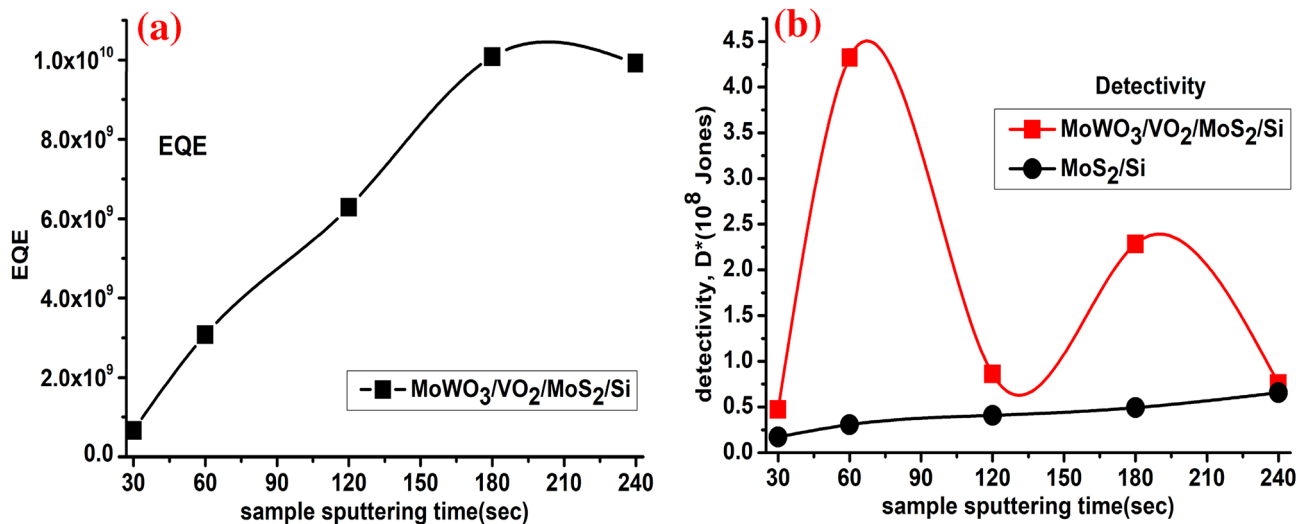


Figure 14. (a) The EQE of MoWO₃/VO₂/MoS₂/Si photodetector as a function of the MoS₂ sputtering time and (b) the detectivity of both MoS₂/Si and MoWO₃/VO₂/MoS₂/Si photodetectors as a function of the MoS₂ sputtering time.

External quantum efficiency (EQE) and detectivity (D*). In order to produce the photocurrent I_{ph} , the fraction of the extracted free charge carriers to the photo flux ϕ_{in} collected at a given energy E_{ph} is called the External Quantum Efficiency, defined by⁸⁴ as $EQE = \frac{hcR_p}{e\lambda}$, where h is the Planck's constant ($\sim 4.135 \times 10^{-15}$ eV s), e is the elementary electron charge ($\sim 1.602 \times 10^{-19}$ C), c is the light velocity ($\sim 3 \times 10^8$ m/s), and λ is the excitation wavelength (365 nm). The EQE values as a function of the prepared samples are plotted in Fig. 14a, where EQE varies from 6.6×10^8 to $\sim 1.0 \times 10^{10}$ at 365 nm which considered higher than the mesoscopic multilayer MoS₂ as reported before⁸⁵. Another important figure of merit of a photodetector is the detectable signal⁸², referred by the specific detectivity measured in Jones, which given by $D^* = \frac{(AB)^{0.5} R_p}{i_n} (\text{cmHz}^{\frac{1}{2}} \text{W}^{-1})$, where A is the effective area of the d in cm², B is the bandwidth, and i_n is the measured noise current. If the shot noise from the dark current is the main noise source, the specific detectivity can be simplified as $D^* = \frac{R_p A^{0.5}}{(2eI_{dark})^{0.5}}$, Where e is the charge of an elementary electron⁸⁶. The calculated D^* for MoS₂/Si and MoWO₃/VO₂/MoS₂/Si is depicted in Fig. 14b. D^* of MoS₂/Si device shows a progressive increase with sputtering time from 30 to 240 s. For MoS₂/Si device, the maximum D^* was 0.4×10^8 Jones. While the maximum D^* for the MoWO₃/VO₂/(60 s)MoS₂/Si is $\sim 4.3 \times 10^8$ Jones at RT and applied voltage of 1 V. In contrast the MoWO₃/VO₂/MoS₂/Si with a 30 s deposition time shows D^* of 0.47×10^8 Jones.

Mechanism. Now we turn to the underlying photoresponse mechanism of the VO₂/MoS₂/Si as a UV photodetector device. The photoresponse properties of VO₂/MoS₂/Si heterojunction can be understood from the

energy-band alignment diagram as in Fig. 1b. Due to the free dangling bonds of the surfaces of MoS₂ film, the MoS₂/Si heterojunction can be affected by lattice matching. Clearly, the implanting of VO₂ layer-based UV photodetector was overwhelmingly play an important role in enhancing the Raman signal/intensity, PL intensity, electrical and optoelectronic performance of MoS₂/Si device. Under the zero-bias condition, an insignificant current was observed due to the high depletion layer at the n–p (MoS₂-Si) junction which restricts the movement of the carriers. Under VO₂ interfacing, both positive and negative current increased significantly and the photocurrent I_{ph} of VO₂/MoS₂/Si film is much higher than that in MoS₂-Si, which can be attributed to the more photon absorption on the top of MoWO₃/VO₂ layer and larger photocurrent-gain due to higher carrier mobility^{87,88}. With increasing the amount of Mo-O (deposition time of 30, 60, 120, 180, and 240 s), the induced current is enhanced due to the more electron–hole pair generation by UV light absorption and the applied voltage shifts dramatically towards negative voltage, which indicates the continuous accumulation of electrons in the vertical VO₂-MoS₂ channel as seen from the logarithmic scale current. In the positive voltage region, the MoS₂/Si n–p structure shows that umpteen electrons are accumulated on the MoS₂ band which shifts Fermi level near the conduction band. Because 1 V is able to decrease the depletion width and the barrier height, electrons are able to overcome the barrier height through thermionic emission, resulting in a high flux of photocurrent and more efficient photocurrent extraction. It is interesting to observe that under a shorter sputtering time of Mo-O (30, then 60 s), a significant forward photocurrent was observed which did not observe before in MoS₂/Si structures^{52,89}. Meanwhile, with increasing the Mo-O content, MoO₃ starts to get folded and the reverse photocurrent starts to get highlighted as seen in the semi-logarithmic scale I–V, Fig. 10.

Conclusion

In summary, the next generation of optoelectronic devices integrates the physics of light-matter interaction of 2D materials at nanoscale for light-harvesting applications and these optoelectronic devices can control the light that converts trions, excitons, and photons to electrical signals. Our approach is based on a high vacuum deposition of Mo-O compound at 400 °C, followed by a sulphurization process in a chemical vapor deposition tube. Here we study the interfacing effect of monoclinic VO₂ with MoS₂ film for UV optoelectronic applications. It showed that different thicknesses of the MoS₂ compound have a direct effect on the Raman, PL, electrical, optoelectronics of MoS₂ peaks. A redshift was observed in Raman spectra with a high electronic coupling between VO₂ and MoS₂ for the case of 180 s sputtering time. Photoluminescence measurements showed that the intensity of the trion peak has a higher intensity than the A-exciton peak for MoS₂/Si structure. On the other hand, the opposite case was observed for the VO₂/MoS₂/Si device. Current–voltage, response/recovery time, external quantum efficiency, time-resolved photocurrent, and detectivity, photocurrent gain, photo-responsivity of VO₂/MoS₂/Si device have been demonstrated. It's shown that the increase of the deposition time of MoS₂ from 30 to 240 s enhances the photo-absorption, photo-responsivity, and external quantum efficiency of the MoWO₃/VO₂/MoS₂/Si device due to the associated folding effects of MoO₃. These results show a multiplexed photodetector fabrication technique of high reproducible and scalable process based on CVD and PVD system and draw high attention towards the interfacing effects of strongly correlated oxide films MoS₂ devices.

Materials and methods

Device fabrication. Preparation of MoS₂ layer on p-type Si substrate has prepared through two steps in physical vapor deposition (PVD)—Radio Frequency magnetron sputtering system, followed by chemical vapor deposition (CDV) process. Si substrates were cleaned through many steps; firstly, kept in NH₄OH–H₂O₂ solution diluted with de-ionized (DI) water for 5 min at 75 °C, then rinsed with DI water for 5 min. After that, they left in HF (5%) solution for 5 s, then rinsed in DI water and dried with high purity N₂. Immediately, the cleaned Si-substrates transferred to a 3×10^{-7} Torr RF magnetron sputtering system (VAKSIS Midas 3M1T). In-situ Ar-plasma source has activated for 10 min at a power of 100 W and low pressure of 6×10^{-3} Torr at room temperature to activate the Si surface. Mo-O thin films were grown using a 3-inch. pure molybdenum target (99.9%) utilizing Ar plasma as a carrier gas and O₂ as a reactive gas. The substrate temperature was stabilized at 400 °C for more than 30 min before the deposition process with steps of 100 °C/30 min. The O₂ and Ar flow rates were kept constant, whereas O₂/(Ar + O₂) of ¼ ratio. The deposition was carried out at 5×10^{-3} Torr and 137 W with different sputtering times of 30, 60, 120, 180, and 240 s. The system was kept to cool down normal up to the room temperature (RT), then immediately transferred to the two-zone CVD quartz chamber (MTI-OTF 1200 system) for the sulphurization process. The as-deposited molybdenum oxide (Mo-O) thin films transferred to the center of the CVD furnace and the temperature is raised to 650 °C. Sulphur powder (0.5 g) is put in a ceramic boot with 100sccm high purity Ar source. An external heating belt with a distance of 50 cm to the substrate was used to evaporate the sulphur for 22 min. Then, the system cooled down until RT with the same flow rate of Ar (100 sccm).

After forming the MoS₂ layer, a thin layer of monoclinic VO₂ has grown. The same sputtering system was used with a 190 W deposition power and Ar/O₂ ratio was 41/2.2 sccm, while the deposition time was set to produce 50 nm film thickness. Then the samples were in-situ annealed at 400 °C for 2 h with 50 sccm Ar flow. A protective and anti-reflection thin layer of Mo_{0.2}W_{0.8}O₃ was deposited on the surface of VO₂ as optimized in our previous work³⁴. High vacuum thermal evaporation system was used to deposit aluminum and gold–palladium that used as a back and front contacts, respectively.

Device characterization. The crystal structures were analyzed using Grazing Incidence X-ray diffraction (XRD GNR ADP PRO 2000), with CuKα ($\lambda = 1.5405$ Å) radiation source with a step of 0.01. VO₂ layer was deposited at a high vacuum condition to ensure its high crystallinity and low semiconductor-metallic phase transition. Parameters such as space group, diffraction peaks, angles, Wyckoff position of vanadium (V) and

oxygen (O) atoms, ratio of O:V, and oxygen vacancy concentrations were calculated from refinement analysis. The refinement calculations were done using Match and Fullprof Suite program. Moreover, the refined structures were plotted in a three-dimensional view using 3D visualization VESTA program. The surface morphology was recorded using scanning electron microscopy (SEM) TESCANMAIA3 XMU. Atomic Force Microscopy (AFM) has been used to investigate the surface topography, roughness, and grain mapping. Each sample was characterized by XE-6 AFM (Park Systems Corp., Suwon-Korea) that controlled with XEP software for data acquisition and XEI software for image analysis and processing. AFM images were obtained through a $0.5 \times 0.5 \mu\text{m}$ area (x - y accessible area) at a 0.5 Hz scan rate. Measurements were taken with a non-contact mode using a PPP-NCHR silicon cantilever consisting of tip radius $< 10 \text{ nm}$ and 42 N/m force constant (Nanosensors TM, Neuchâtel-Switzerland). Raman measurements and photoluminescence (PL) spectra were carried out using Renishaw in Via Confocal Raman microscope with a 532 nm laser beam, while an incident laser power of 3 mW was chosen to acquire a single Raman spectrum. The temperature-resistance measurement of monoclinic and high crystalline nanostructure VO_2 thin film has been performed using a four-probe measurement system connected to a heating stage ranging from RT to 100 °C. The electrical and optoelectronics measurements were measured using 2450 Kethley Source -Meter and 365 nm ultraviolet (UV) light lamp for optoelectronic measurements.

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References

- Kim, J. K. *et al.* Trap-mediated electronic transport properties of gate-tunable pentacene/MoS₂ p-n heterojunction diodes. *Sci. Rep.* **6**, 36775 (2016).
- Yu, J., Li, J., Zhang, W. & Chang, H. Synthesis of high quality two-dimensional materials via chemical vapor deposition. *Chem. Sci.* **6**, 6705–6716 (2015).
- Cai, Z., Liu, B., Zou, X. & Cheng, H. M. Chemical vapor deposition growth and applications of two-dimensional materials and their heterostructures. *Chem. Rev.* **118**, 6091–6133 (2018).
- Wang H *et al.* (2012) Large-scale 2D electronics based on single-layer MoS₂ grown by chemical vapor deposition. In *Technical digest—international electron devices meeting, IEDM* (2012). <https://doi.org/10.1109/IEDM.2012.6478980>.
- Sharma, D. *et al.* Transfer characteristics and low-frequency noise in single- and multi-layer MoS₂ field-effect transistors. *Appl. Phys. Lett.* **107**, 162102 (2015).
- Nicolosi, V., Chhowalla, M., Kanatzidis, M. G., Strano, M. S. & Coleman, J. N. Liquid exfoliation of layered materials. *Science* **340**, 1226419 (2013).
- Coleman, J. N. *et al.* ChemInform abstract: two-dimensional nanosheets produced by liquid exfoliation of layered materials. *ChemInform* **42**, 568 (2011).
- Yu, X., Prévot, M. S. & Sivula, K. Multiflake thin film electronic devices of solution processed 2D MoS₂ enabled by sonopolymer assisted exfoliation and surface modification. *Chem. Mater.* **26**, 5892–5899 (2014).
- Paton, K. R. *et al.* Scalable production of large quantities of defect-free few-layer graphene by shear exfoliation in liquids. *Nat. Mater.* **13**, 624–630 (2014).
- Varrla, E. *et al.* Large-scale production of size-controlled MoS₂ nanosheets by shear exfoliation. *Chem. Mater.* **27**, 1129–1139 (2015).
- Knirsch, K. C. *et al.* Basal-plane functionalization of chemically exfoliated molybdenum disulfide by diazonium salts. *ACS Nano* **9**, 6018–6030 (2015).
- Eda, G. *et al.* Photoluminescence from chemically exfoliated MoS₂. *Nano Lett.* **11**, 5111–5116 (2011).
- Yang, X. Y. & Chauhan, N. Photoenergy and Thin Film Materials. Wiley, New York. <https://doi.org/10.1002/9781119580546> (2019).
- Giannazzo, F., Lara Avila, S., Eriksson, J. & Sonde, S. *Integration of 2D Materials for Electronics Applications* (MDPI, Basel, 2019).
- Browning, R. *et al.* Atomic layer deposition of MoS₂ thin films. *Mater. Res. Exp.* **2**, 035006 (2015).
- Loh, T. A. J. & Chua, D. H. C. Growth mechanism of pulsed laser fabricated few-layer MoS₂ on metal substrates. *ACS Appl. Mater. Interfaces* **6**, 15966–15971 (2014).
- Serrao, C. R. *et al.* Highly crystalline MoS₂ thin films grown by pulsed laser deposition. *Appl. Phys. Lett.* **106**, 052101 (2015).
- Ma, X. & Shi, M. Thermal evaporation deposition of few-layer MoS₂ films. *Nano-Micro Lett.* **5**, 135–139 (2013).
- Wu, S. *et al.* Vapor-solid growth of high optical quality MoS₂ monolayers with near-unity valley polarization. *ACS Nano* **7**, 2768–2772 (2013).
- Pacley, S. *et al.* Impact of reduced graphene oxide on MoS₂ grown by sulfurization of sputtered MoO₃ and Mo precursor films. *J. Vac. Sci. Technol. A Vac. Surf. Films* **34**, 041505 (2016).
- Hussain, S. *et al.* Synthesis and characterization of large-area and continuous MoS₂ atomic layers by RF magnetron sputtering. *Nanoscale* **8**, 4340–4347 (2016).
- Matsuura, K. *et al.* Low-carrier-density sputtered MoS₂ film by vapor-phase sulfurization. *J. Electron. Mater.* **47**, 3497–3501 (2018).
- Imai, Y., Solov'yev, I. & Imada, M. Electronic structure of strongly correlated systems emerging from combining path-integral renormalization group with the density-functional approach. *Phys. Rev. Lett.* **95**, 176405 (2005).
- Sass, B. *et al.* Structural and electronic properties of epitaxial V₂O₃ thin films. *J. Phys. Condens. Matter* **16**(1), 77–87 (2004).
- Rozenberg, M. J. *et al.* Optical conductivity in Mott-Hubbard systems. *Phys. Rev. Lett.* **75**, 105–108 (1995).
- Tongay, S. *et al.* Tuning interlayer coupling in large-area heterostructures with CVD-grown MoS₂ and WS₂ monolayers. *Nano Lett.* **14**, 3185–3190 (2014).
- Sun, Y. *et al.* Probing local strain at MX₂-Metal boundaries with surface plasmon-enhanced Raman scattering. *Nano Lett.* **14**, 5329–5334 (2014).
- Goodenough, J. B. The two components of the crystallographic transition in VO₂. *J. Solid State Chem.* **3**, 490–500 (1971).
- Budai, J. D. *et al.* In situ X-ray microdiffraction studies inside individual VO₂ microcrystals. *Acta Mater.* **61**, 2751–2762 (2013).
- McWhan, D. B., Marezio, M., Remeika, J. P. & Dernier, P. D. X-ray diffraction study of metallic VO₂. *Phys. Rev. B* **10**, 490–495 (1974).
- Oliva, N. *et al.* Van der Waals MoS₂/VO₂ heterostructure junction with tunable rectifier behavior and efficient photoresponse. *Sci. Rep.* **7**, 14250 (2017).
- Lin, Y. C. *et al.* Photoluminescence of monolayer transition metal dichalcogenides integrated with VO₂. *J. Phys. Condens. Matter* **28**, 504001 (2016).

33. Karataş, A. & Yilmaz, M. Molybdenum disulfide thin films fabrication from multi-phase molybdenum oxide using magnetron sputtering and CVD systems together. *Superlattices Microstruct.* **143**, 106555 (2020).
34. Basyooni, M. A., Zaki, S. E., Ertugrul, S., Yilmaz, M. & Eker, Y. R. Fast response of CO₂ room temperature gas sensor based on mixed-valence phases in molybdenum and tungsten oxide nanostructured thin films. *Ceram. Int.* **46**, 9839–9853. <https://doi.org/10.1016/j.ceramint.2019.12.259> (2020).
35. US7196314B2—Image sensor and pixel having an anti-reflective coating over the photodiode—Google Patents. <https://patents.google.com/patent/US7196314B2/en>. Accessed on 18th June 2020.
36. US8546742B2—Array of nanowires in a single cavity with anti-reflective coating on substrate—Google Patents. <https://patents.google.com/patent/US8546742B2/en>. Accessed on 2nd July 2020.
37. US6803249B2—Method of making an integrated photodetector in which a silicon nitride layer forms an anti-reflective film and part of multi-layer insulator within transistor structures—Google Patents. <https://patents.google.com/patent/US6803249B2/en>. Accessed on 2nd July 2020.
38. Zhang, J. *et al.* Evolution of structural and electrical properties of oxygen-deficient VO₂ under low temperature heating process. *ACS Appl. Mater. Interfaces* **9**, 27135–27141 (2017).
39. Wang, X., Wang, Z., Zhang, G. & Jiang, J. Insight into electronic and structural reorganizations for defect-induced VO₂ metal-insulator transition. *J. Phys. Chem. Lett.* **8**, 3129–3132 (2017).
40. Chen, L. *et al.* First-principles study of the effect of oxygen vacancy and strain on the phase transition temperature of VO₂. *RSC Adv.* **6**, 86872–86879 (2016).
41. Fan, S., Fan, L., Li, Q., Liu, J. & Ye, B. The identification of defect structures for oxygen pressure dependent VO₂ crystal films. *Appl. Surf. Sci.* **321**, 464–468 (2014).
42. Yuan, Z., Hou, J. & Liu, K. Interfacing 2D semiconductors with functional oxides: fundamentals, properties, and applications. *Crystals* **7**, 265 (2017).
43. Zhang, X. *et al.* Phonon and Raman scattering of two-dimensional transition metal dichalcogenides from monolayer, multilayer to bulk material. *Chem. Soc. Rev.* **44**, 2757–2785 (2015).
44. Hui, Y. Y. *et al.* Exceptional tunability of band energy in a compressively strained trilayer MoS₂ sheet. *ACS Nano* **7**, 7126–7131 (2013).
45. Manzeli, S., Allain, A., Ghadimi, A. & Kis, A. Piezoresistivity and strain-induced band gap tuning in atomically thin MoS₂. *Nano Lett.* **15**, 5330–5335 (2015).
46. Kotula, A. P., Snyder, C. R. & Migler, K. B. Determining conformational order and crystallinity in polycaprolactone via Raman spectroscopy. *Polymer (Guildf)* **117**, 1–10 (2017).
47. Nikolaeva, G. Y. *et al.* Using Raman spectroscopy to determine the structure of copolymers and polymer blends. *J. Phys. Conf. Ser.* **826**, 012002 (2017).
48. Samuel, A. Z. Direct estimation of polymer crystallinity with Raman spectroscopy using ratio of scattering cross-sections estimated from variable temperature measurements. *Spectrochim. Acta Part A Mol. Biomol. Spectrosc.* **224**, 117431 (2020).
49. Sundaram, R. S. *et al.* Electroluminescence in single layer MoS₂. *Nano Lett.* **13**, 1416–1421 (2013).
50. Ghopry, S. A., Alamri, M. A., Goul, R., Sakidja, R. & Wu, J. Z. Extraordinary sensitivity of surface-enhanced Raman spectroscopy of molecules on MoS₂ (WS₂) nanodomains/graphene van der Waals heterostructure substrates. *Adv. Opt. Mater.* **7**, 1801249 (2019).
51. Er, E. *et al.* High-yield preparation of exfoliated 1T-MoS₂ with SERS activity. *Chem. Mater.* **31**, 5725–5734 (2019).
52. Majee, B. P., Mishra, S., Pandey, R. K., Prakash, R. & Mishra, A. K. Multifunctional few-layer MoS₂ for photodetection and surface-enhanced Raman spectroscopy application with ultrasensitive and repeatable detectability. *J. Phys. Chem. C* **123**, 18071–18078 (2019).
53. Jeong, H. Y. *et al.* Optical gain in MoS₂ via coupling with nanostructured substrate: Fabry-Perot interference and plasmonic excitation. *ACS Nano* **10**, 8192–8198 (2016).
54. Hu, J. Q., Shi, X. H., Wu, S. Q., Ho, K. M. & Zhu, Z. Z. Dependence of electronic and optical properties of MoS₂ multilayers on the interlayer coupling and Van Hove singularity. *Nanoscale Res. Lett.* **14**, 288 (2019).
55. Mouri, S., Miyauchi, Y. & Matsuda, K. Tunable photoluminescence of monolayer MoS₂ via chemical doping. *Nano Lett.* **13**, 5944–5948 (2013).
56. Plechinger, G. *et al.* Identification of excitons, trions and biexcitons in single-layer WS₂. *Phys. Status Solidi Rapid Res. Lett.* **9**, 457–461 (2015).
57. Kwon, H. *et al.* Probing trions at chemically tailored trapping defects. *ACS Cent. Sci.* **5**, 1786–1794 (2019).
58. Zhu, Z. Y., Cheng, Y. C. & Schwingenschlögl, U. Giant spin-orbit-induced spin splitting in two-dimensional transition-metal dichalcogenide semiconductors. *Phys. Rev. B Condens. Matter Mater. Phys.* **84**, 153402 (2011).
59. Christopher, J. W., Goldberg, B. B. & Swan, A. K. Long tailed trions in monolayer MoS₂: temperature dependent asymmetry and resulting red-shift of trion photoluminescence spectra. *Sci. Rep.* **7**, 1–8 (2017).
60. Lin, Z. *et al.* Controllable growth of large-size crystalline MoS₂ and resist-free transfer assisted with a Cu thin film. *Sci. Rep.* **5**, 1–10 (2015).
61. Liu, H. & Chi, D. Dispersive growth and laser-induced rippling of large-area single-layer MoS₂ nanosheets by CVD on c-plane sapphire substrate. *Sci. Rep.* **5**, 1–8 (2015).
62. Splendiani, A. *et al.* Emerging photoluminescence in monolayer MoS₂. *Nano Lett.* **10**(4), 1271–1275. <https://doi.org/10.1021/nl903868w> (2010).
63. Bano, A. & Gaur, N. K. Interfacial coupling effect on electron transport in MoS₂/SrTiO₃ heterostructure: an ab-initio study. *Sci. Rep.* **8**, 1–8 (2018).
64. Zhang, J. *et al.* Observation of strong interlayer coupling in MoS₂/WS₂ heterostructures. *Adv. Mater.* **28**, 1950–1956 (2016).
65. Tongay, S. *et al.* Thermally driven crossover from indirect toward direct bandgap in 2D semiconductors: MoSe₂ versus MoS₂. *Nano Lett.* **12**, 5576–5580 (2012).
66. Mak, K. F. *et al.* Tightly bound trions in monolayer MoS₂. *Nat. Mater.* **12**, 207–211 (2013).
67. Peimyo, N. *et al.* Chemically driven tunable light emission of charged and neutral excitons in monolayer WS₂. *ACS Nano* **8**, 11320–11329 (2014).
68. Choi, J., Zhang, H. & Choi, J. H. Modulating optoelectronic properties of two-dimensional transition metal dichalcogenide semiconductors by photoinduced charge transfer. *ACS Nano* **10**, 1671–1680 (2016).
69. Godefroy, S. *et al.* Classification and control of the origin of photoluminescence from Si nanocrystals. *Nat. Nanotechnol.* **3**, 174–178 (2008).
70. Hwang, J. D. *et al.* Suppressing the dark current of metal-semiconductor-metal SiGe/Si heterojunction photodetector by using asymmetric structure. *Thin Solid Films* **515**, 3837–3839 (2007).
71. Xu, H. Y. *et al.* Effects of annealing ambient on oxygen vacancies and phase transition temperature of VO₂ thin films. *RSC Adv.* **6**, 79383–79388 (2016).
72. Wang, L. *et al.* Oxygen vacancy induced room-temperature metal-insulator transition in nickelate films and its potential application in photovoltaics. *ACS Appl. Mater. Interfaces* **8**, 9769–9776 (2016).
73. Zaki, S. E. *et al.* Role of oxygen vacancies in vanadium oxide and oxygen functional groups in graphene oxide for room temperature CO₂ gas sensors. *Sensors Actuators A Phys.* **294**, 17–24 (2019).

74. Jain, S. K. *et al.* Effect of metal contacts on a GaN/Sapphire-based MSM ultraviolet photodetector. *J. Electron. Mater.* **47**, 6086–6090 (2018).
75. Casalino, M., Iodice, M., Sirleto, L., Rendina, I. & Coppola, G. Asymmetric MSM sub-bandgap all-silicon photodetector with low dark current. *Opt. Exp.* **21**, 28072 (2013).
76. Reddy, M. S. P., Kim, B.-J. & Jang, J.-S. Dual detection of ultraviolet and visible lights using a DNA-CTMA/GaN photodiode with electrically different polarity. *Opt. Exp.* **22**, 908 (2014).
77. Chao, C.-H., Weng, W.-J. & Wei, D.-H. Enhanced UV photodetector response and recovery times using a nonpolar ZnO sensing layer. *J. Vac. Sci. Technol. A Vac. Surf. Films* **34**, 02D106 (2016).
78. Ling, Z. P. *et al.* Large-scale two-dimensional MoS₂ photodetectors by magnetron sputtering. *Opt. Exp.* **23**, 13580 (2015).
79. Dhyani, V. & Das, S. High-speed scalable silicon-MoS₂ P-N heterojunction photodetectors. *Sci. Rep.* **7**, 44243 (2017).
80. Das, S., Chen, H. Y., Penumatcha, A. V. & Appenzeller, J. High performance multilayer MoS₂ transistors with scandium contacts. *Nano Lett.* **13**, 100–105 (2013).
81. Yang, Y., Huo, N. & Li, J. Sensitized monolayer MoS₂ phototransistors with ultrahigh responsivity. *J. Mater. Chem. C* **5**, 11614–11619 (2017).
82. Ko, P. J., Abderrahmane, A., Kim, N. H. & Sandhu, A. High-performance near-infrared photodetector based on nano-layered MoSe₂. *Semicond. Sci. Technol.* **32**, 065015 (2017).
83. Zhuo, R. *et al.* High-performance self-powered deep ultraviolet photodetector based on MoS₂/GaN p-n heterojunction. *J. Mater. Chem. C* **6**, 299–303 (2018).
84. Wu, J. M. & Chang, W. E. Ultrahigh responsivity and external quantum efficiency of an ultraviolet-light photodetector based on a single VO₂ microwire. *ACS Appl. Mater. Interfaces* **6**, 14286–14292 (2014).
85. Saenz, G. A., Karapetrov, G., Curtis, J. & Kaul, A. B. Ultra-high photoresponsivity in suspended metal-semiconductor-metal mesoscopic multilayer MoS₂ broadband detector from UV-to-IR with low Schottky barrier contacts. *Sci. Rep.* **8**, 1–11 (2018).
86. Liu, X. *et al.* All-printable band-edge modulated ZnO nanowire photodetectors with ultra-high detectivity. *Nat. Commun.* **5**, 1–9 (2014).
87. Choi, W. *et al.* High-detectivity multilayer MoS₂ phototransistors with spectral response from ultraviolet to infrared. *Adv. Mater.* **24**, 5832–5836 (2012).
88. Dhakal, K. P. *et al.* Confocal absorption spectral imaging of MoS₂: optical transitions depending on the atomic thickness of intrinsic and chemically doped MoS₂. *Nanoscale* **6**, 13028–13035 (2014).
89. Lopez-Sanchez, O. *et al.* Light generation and harvesting in a van der waals heterostructure. *ACS Nano* **8**, 3042–3048 (2014).

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Author contributions

M.A.B. suggested, designed and performed the whole experiments, characterizations, analysis, and writing the manuscript. S.E.Z. performed some experiments, characterization, and contributed for writing the manuscript. Y.R.E. and M.Y. contributed to writing, supervision, discussions and analysis of the data. M.S. characterized and analyzed the data, contributes to the plot of some figures, writing of the manuscript in addition to revision and editing.

Competing interests

The authors declare no competing interests.

Additional information

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