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Structural and Photoelectric Properties of Epitaxially Grown Vanadium Dioxide Thin Films on c-Plane Sapphire and Titanium Dioxide

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Vanadium dioxide (VO_2) is one of the most extensively studied materials in the strongly correlated electron family capable of sustaining an insulator-to-metal transition. Here we present our studies of high-quality thin films of epitaxially grown VO_2 on $\text{c-Al}_2\text{O}_3(0001)$ and $\text{TiO}_2(001)$ via reactive DC pulsed magnetron sputtering. We present the structural transition probed via Reflection High Energy Electron Diffraction (RHEED) for the first time and we correlate the surface microstructure measurements with simulations before, during, and after the thermally induced transition. We also study the photoelectric conversion of VO_2 on $\text{TiO}_2(001)$ and $\text{c-Al}_2\text{O}_3(0001)$ under 405 nm light and demonstrate up to a 2000% increase in quantum efficiency as the power of the light is varied for VO_2 on $\text{TiO}_2(001)$.

A Mott transition is a fundamental concept that governs the emergence of various electronic phases and physical properties in correlated electron materials, represented for example by the insulator-metal transition (IMT) in vanadium dioxide (VO_2). Vanadium dioxide is one of the most extensively studied materials in the correlated electron family where it is known to undergo a characteristic reversible first order transition from insulator ($T < T_c$) to metal ($T > T_c$) where T_c , $\sim 68^\circ\text{C}$ in bulk, is the temperature at which the transition occurs^{1,2}. This IMT is also accompanied by an associated structural transition from an insulating monoclinic phase to a metallic rutile phase². Thus, it is subject of significant importance to understand the dynamics of electron and lattice systems across such transition in these highly correlated materials³⁻⁵. An interesting proposed aspect of VO_2 in its monoclinic phase is its ability to undergo photoelectric conversion; where holes are formed in the valence band of the substrate and transported to the O_{2p} band of VO_2 then electrons from the $\text{VO}_2 d_{||}$ band move to fill the holes in the lower O_{2p} band⁶. Therefore, a key approach to understand the occurrence of this photoelectric conversion is to investigate the correlations between surface microstructure and photoelectric conversion in thin films deposited on different substrates to explore the potential parameter space for possible applications.

Additionally, it has been argued that the binding between doubly occupied (doublon) and empty (holon) sites plays a key role in the Mott transition in strongly correlated Mott-Hubbard systems and could play a distinct role in photoelectric conversion⁷. In this photo-electric conversion, the application of an electric field to the monoclinic VO_2 phase drives carrier tunneling to create doublon-holon pairs via nonlinear excitation processes such as multiphoton absorption and quantum-tunneling^{8,9}. This pair creation results in an instantaneous insulator to metal transition without direct interaction with the lattice^{8,9}. Allowing for a purely electronic phase transition, however, the thin film surface microstructure and morphology will affect carrier scattering, and the dynamics will be reflected in the ultimate quantum-efficiency⁹⁻¹¹.

Results

Crystal analysis and determination. All films are 30 nm thick and were epitaxially grown via reactive DC magnetron sputtering on $\text{c-Al}_2\text{O}_3$ and TiO_2 as discussed in the experimental section. The initial microstructure characterization of the films was carried out using X-Ray Diffraction (XRD) symmetric scans as shown in

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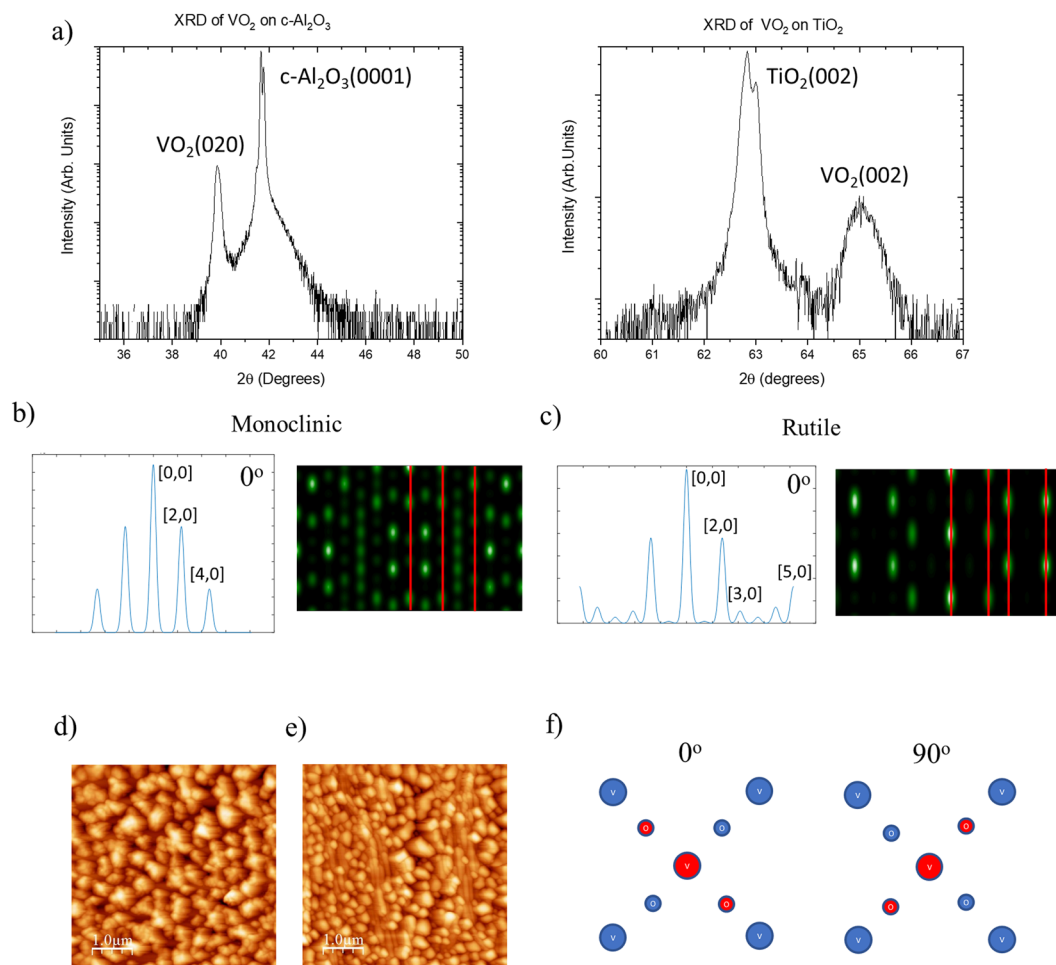


Figure 1. The experimental XRD, AFM, and simulated RHEED patterns of VO₂ on c-Al₂O₃ and TiO₂. **(a)** The XRD 2θ scan for VO₂ on c-Al₂O₃(0001) and TiO₂(001) where the intensity scale in arbitrary units is log scaled. **(b)** The simulated RHEED pattern of the 0° rotation for the monoclinic phase of VO₂ where the streak intensity patterns are recorded on top and the simulated diagrams are reported on bottom with the in plane lattice planes are recorded for each. **(c)** The simulated RHEED patterns of the 0° rotation for the rutile phase of VO₂ where the placement of the plots is the same as the previous monoclinic phase. **(d)** The AFM image of the surface microstructure of VO₂ on c-Al₂O₃(0001) where the scale is 5.0 μm × 5.0 μm. **(e)** The AFM image of the surface microstructure of VO₂ on TiO₂(001) where the scale is 5.0 μm × 5.0 μm. **(f)** The orientations of the vanadium and oxygen atoms through the 0° and 90° rotations of one unit cell where the first two surface layers of atoms are shown in the c-direction where the red layer is displaced ~1.44 nm below the blue layer.

(Fig. 1a). Upon examination of the 2θ scans we determined that the peak location was consistent with the bulk location for the VO₂ grown on c-Al₂O₃, with a slight degree of strain toward the substrate peak as expected. We note that the main peak location for the VO₂ grown on TiO₂ is consistent with previous reports for growth on this substrate¹². For VO₂ on TiO₂(001), the VO₂ peak location indicated that the film was also strained toward the substrate peak by ~0.5° in comparison to existing data, suggesting that in this case the film contains a more strained monoclinic phase toward the rutile phase than previously reported¹².

We also determined the mosaicity (the degree of crystallite misorientation) for each sample and found a nominal degree of mosaicity ~0.08° for the sample grown on c-Al₂O₃ and ~0.047° for the sample grown on TiO₂.

RHEED structural analysis. Figure 1b,c shows the simulated Reflection High Energy Electron Diffraction (RHEED) patterns and streak diagrams expected for the bulk-like phases of VO₂ through the IMT^{13,14}. As evidenced by (Figs 1b,c,f and S1a–c), upon in-plane rotation of the sample with respect to the electron-beam, the diffraction pattern of the sample changes greatly due to the atomic positions of the vanadium and oxygen atoms on the lattice. The oxygen atoms play a distinct role in these diffraction patterns, as they appear to be the main contributors to streak intensity. Each in-plane rotation shows a large degree of variation in diffraction pattern, with a significant change expected for the monoclinic structure due to the canted angles exhibited by the oxygen atoms¹⁵.

The positions of the vanadium and oxygen atoms have been extensively studied both theoretically and experimentally for both phases across the IMT where space groups for the two phases of VO₂ across the IMT are P4₂/

Atom	X(Å)	Y(Å)	Z(Å)
Vanadium	0	0	0
	2.275	2.275	1.44
Oxygen	1.38775	1.38775	0
	-1.38775	-1.38775	0
	0.88725	3.66275	1.44
	3.66275	0.88725	1.44
	1.38775	1.38775	2.88
	-1.38775	-1.38775	2.88

Table 1. Atomic Positions of VO₂(R).

Atom	X(Å)	Y(Å)	Z(Å)
Vanadium	1.3915	5.2845	0.1345
	-1.3915	7.9945	2.5555
	-1.3915	-5.2845	-0.1345
	1.3915	-2.5745	2.8245
Oxygen 1	0.575	1.1382	1.076
	-0.575	3.8482	1.614
	-0.575	-1.1382	-1.076
	0.575	1.5718	3.766
Oxygen 2	2.2425	3.7398	1.5602
	-2.2425	6.4498	1.1298
	-2.2425	-3.7398	-1.5602
	2.2425	-1.0298	4.2502

Table 2. Atomic Positions of VO₂(M).

mmm(136) for rutile and P2₁/c (14) for monoclinic and for our simulation atom positions were determined from these space groups as shown in Tables 1 and 2¹⁵. The rutile structure possesses greater in-plane symmetry than the monoclinic does. Thus, through each 90° in-plane azimuthal rotation a repetition of patterns will occur with a clear differentiation in diffraction patterns expected along the 45° rotation directions, and with differing patterns between 45° and 135°, depending upon whether the oxygen atoms are in-line with the main axis or rotated 90° off axis.

This in-plane azimuthal dependency is illustrated in (Figs 1f and S2c) in which the first two layers of atoms account for the majority of the diffracted e⁻ beam intensity in the patterns. For the present study, each of the chosen substrates ensures a good lattice match between the square lattice face of the substrates and the square face of the rutile structure for VO₂ (a = 4.55 Å). In the case of c-Al₂O₃ the lattice mismatch is greater (a = 4.785 Å) than that of TiO₂(001) (a = 4.584 Å) however both provide adequate lattice matching to facilitate epitaxial growth of VO₂ as evidenced from the XRD in (Fig. 1a)¹⁵⁻¹⁷.

It is worth noting that the spotty nature of the simulated RHEED pattern (Figs 1b,c and S1a,b) is due to the fact that all surface atoms in the VO₂ structure are not in the same plane, since all surface atoms in the same plane would lead to vertical streaks in the diffraction pattern. In the present case, this spotty pattern is compounded in the experimental RHEED data by the inherent surface roughness that is evidenced in the Atomic Force Microscopy (AFM) images of the actual samples (Fig. 1d,e). We also note that the surface of VO₂ on c-Al₂O₃ has a larger degree of average roughness, ~30 nm, due to the larger spacing between terraces leading to a roughness on the order of the film thickness. In comparison, (Fig. 1e), corresponding to VO₂ on TiO₂(001), exhibits enhanced proximity between terraces, with each terrace sitting directly adjacent to its neighbor. This lack of separation between terraces accounts for a roughness ~13 nm thus yielding a smoother overall film in comparison to VO₂ on c-Al₂O₃(0001). This larger degree of surface roughness in the VO₂ on c-Al₂O₃(0001) may also be a contributor to the lesser photocurrent produced for the VO₂ on c-Al₂O₃(0001) compared to the VO₂ on TiO₂(001) sample due to scattering effects discussed later.

We also present here a comparison study of bulk VO₂ RHEED patterns predicted via simulation and the RHEED patterns recorded from VO₂ grown on c-Al₂O₃. As shown in (Figs 2a and S2a) the experimentally determined RHEED patterns exhibit similar streak configurations and orientations as those determined in the simulated patterns. Figure S2a exhibits numerous cases of pattern matching as well as streak spacing suggesting that the simulations and the observed diffraction patterns for the monoclinic VO₂ phase are in good agreement. We then proceeded to heat the samples up to 100°C to allow for the VO₂ to undergo the full IMT which for this sample occurred at ~55°C. This shift in transition temperature from bulk is likely due to the thin film character of sample, where factors like lattice parameter mismatch between sample and substrate play a key role as previously reported for TiO₂^{12,18}. Once the sample fully transitioned, the RHEED patterns in (Figs 2b and S2b) (right) were compared to the simulations shown in (Figs 2b and S2b) (left). The streak patterns from the rutile phase VO₂

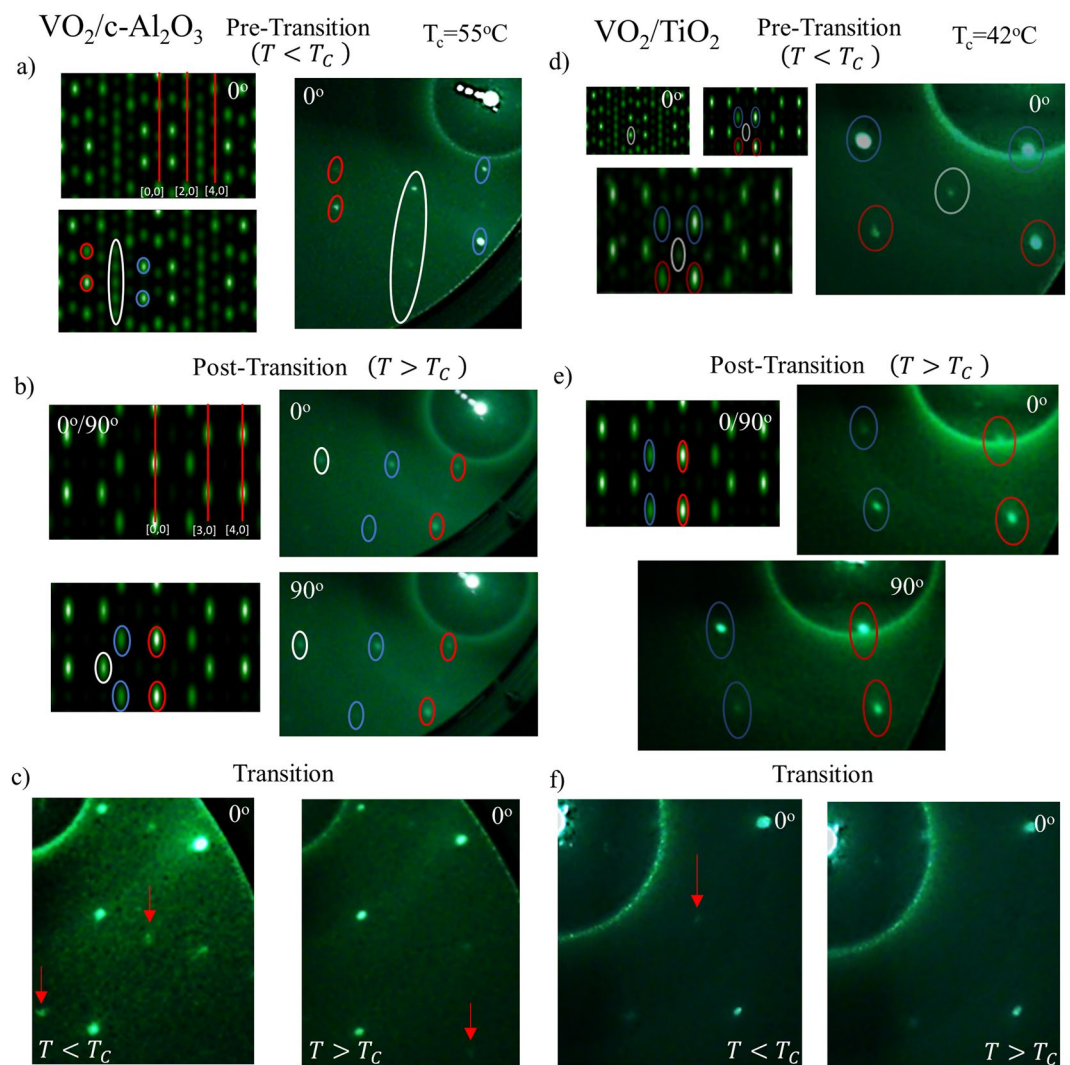


Figure 2. The simulated and experimentally determined RHEED patterns for VO_2 on $c\text{-Al}_2\text{O}_3(0001)$ and VO_2 on $\text{TiO}_2(001)$. For VO_2 on $c\text{-Al}_2\text{O}_3(0001)$: Simulated and experimental RHEED patterns of VO_2 azimuthal rotations and for the two phases. (a) The left two images are simulations of the 0° rotation for the monoclinic phase of VO_2 , the top image denotes the 2-D lattice planes for the streak patterns while the bottom image has streaks highlighted. The right image is the experimentally determined RHEED pattern the highlighted streaks correspond to the simulated image. This plot orientation is maintained for (b) as well where the simulated images are shown at left and experimental images at right. (b) The $0^\circ/90^\circ$ rotation for the rutile phase of VO_2 . The experimentally determined RHEED patterns for the 0° degree rotation through the thermal IMT where the image on the left is prior to transition and the image on the right is post transition. For VO_2 on $\text{TiO}_2(001)$: (f) The top leftmost two images are simulations of the 0° rotation for the monoclinic phase (left) and rutile phase (right) of VO_2 . The bottom left image is a superimposed image of both simulated images. The right image is the experimentally determined RHEED pattern for VO_2 on $\text{TiO}_2(001)$ where the highlighted streaks correspond to the simulated images. (e) The left top image is the simulation and right top and bottom images experimental patterns for VO_2 grown on $\text{TiO}_2(001)$; the 0° rotation is right and 90° is bottom. (f) The experimentally determined RHEED patterns for the 0° degree rotation through the thermal IMT where the image on the left is prior to transition and the image on the right is post transition. (Note: All images have been rotated 45 degrees and the contrast has been increased for ease of streak identification).

exhibit the same configurations as those predicted in the simulation for (Figs 2b and S2b), i.e. experimentally determined patterns are similar to the simulated ones, in combination with the AFM images and the XRD data, further assert that the patterns observed in both the simulations and experiments, are quite close to bulk-like crystalline VO_2 in both the monoclinic and rutile states, and with low mosaicity.

We also examined the RHEED pattern through the transition to investigate the sample evolution in real time. To do so, we heated the sample in 5°C increments from 25°C to 100°C recording the diffraction pattern after each increment. Upon ramping the temperature, we found a marked change in the RHEED pattern at the critical temperature. We observed the disappearance of several peripheral streaks exhibited in the monoclinic phase as well as a strengthening of the intensity of the main streaks that create a repeating “rectangular” pattern as designated

by (Fig. 2c and Supplementary Video S1). We propose that this change in diffraction pattern is likely due to the change in the atomic positions of the oxygen atoms when transitioning from the monoclinic to the rutile phase. During this transition there is a lattice parameter change of $\sim -1.2 \text{ \AA}$ from $a = 5.75 \text{ \AA}$ in the monoclinic phase to $a = 4.55 \text{ \AA}$ in the rutile phase¹⁵.

Finally, we sought to compare our bulk VO₂ simulations to a VO₂ sample grown on TiO₂(001). Upon examination of the VO₂ film prior transition we find in (Fig. 2d) a plausible superposition of the monoclinic and rutile phases with a greater emphasis on the streak locations of the rutile phase. Especially in the case of (Fig. 2d), the non-transitioned locations have a distinctly rutile like pattern that suggests structural strain toward the rutile phase. Thus for (Figs 2d and S2c), we have interlaid the monoclinic and the rutile simulations to make the microstructure more apparent enabling fruitful comparison with experiment. This combined with a 0° rotation asserts that there is microstructural strain toward the rutile phase in the non-transitioned VO₂ grown on TiO₂. We then heated the sample to 100 °C to allow full IMT, which for this sample is $\sim 42 \text{ °C}$ where film substrate strain effects are known to play a key role in lowering the transition temperature^{12,18}.

Once we heated the sample, we compared the simulations with the experimentally determined RHEED patterns. Here we find that the patterns for the 0° and the 90° azimuth positions, observed in (Fig. 2e), have several streaks well matched to the simulations. In comparison to the monoclinic phase, the center streak is missing but the four streaks that make up the “rectangular” shape of the pattern still persist with a slightly larger streak spacing than that of VO₂ on c-Al₂O₃. We also see this in (Fig. S2d) the 45° azimuth position for the diffraction pattern, along with similar streaks seen in the simulation. Also, this pattern now exhibits a slight slant to the locations of the streak that would occupy the same lattice plane. This is likely due to the structural strain on the atomic locations on the lattice, namely the oxygen locations. Slight displacements off the main axis of the lattice could also cause shifts to the pattern, which is the case for the 45° direction. This slight displacement is also visible in the 0° and the 90° azimuth in-plane directions, but the shifts are lesser here since these directions represent less sensitive directions to oxygen displacements.

We examined the RHEED pattern through the transition to see changes due to the IMT. In order to do this, we heated the sample in 5 °C increments from 25 °C to 100 °C recording the diffraction pattern at each increment in agreement with the previous VO₂ on c-Al₂O₃ sample. Upon ramping through the temperature increments, we found a marked change that occurred at the critical temperature. We observed the disappearance of the center streak in the rutile state that was apparent in the non-transitioned phase as demonstrated by (Fig. 2f) and Supplementary Video S2. This change occurs abruptly upon the IMT. As previously described, due to the strong dependence of the streak patterns to the oxygen location and the larger degree of structural strain on the sample, this large pattern adjustment agrees with the expected strained monoclinic structure as previously discussed.

The structural characterization across the transition for these samples provides a suitable framework for subsequent characterization of the photo responsivity of the same samples described below.

Photocurrent analysis. As demonstrated by (Fig. 3a) the VO₂/TiO₂ band structure illustrates that such film substrate heterostructure as a promising candidate for an efficient near-UV to deep UV photo-sensor; which is of interest due to lack of efficient, low power photodetectors in this spectral range. Thus, we examined how the photocurrent produced in these samples under 405 nm illumination is affected through the IMT. Due to the close match between the energy of the incoming photon and the TiO₂ substrate band gap energy, the electron carriers are excited from the O_{2p} band into the conduction band of the TiO₂ substrate. By exciting carriers in the TiO₂ substrate, the holes left behind in the TiO₂ O_{2p} band are then transferred to the VO₂ O_{2p} band leading to carriers from the d_{||} band of VO₂ moving to the O_{2p} band thus inducing a photocurrent⁶. Worth noting is that strain effects on the structure of the VO₂ could result in adjustments to the band structure shown. We anticipate that the d_{||} and d_{||}* bands of the VO₂ could exist closer together energetically resulting in a more metallic electronic structure as the lattice is strained toward the rutile phase.

This photocurrent is only possible in the monoclinic phase as once VO₂ undergoes the phase transition to the rutile phase the band gap collapses and it takes on characteristics of a metal, therefore producing no photocurrent. To this end, we first demonstrated the optical response of VO₂ on TiO₂ as it undergoes the thermal IMT. This optical transition is demonstrated in (Fig. 3b) where a decrease in voltage is correlated to a decrease in resistivity due to the IMT. We then investigated if a photocurrent could be produced under solely illumination of 405 nm light, chosen because its energy ($\sim 3.06 \text{ eV}$) is close to the band gap of TiO₂ is ($\sim 3.0 \text{ eV}$)²⁰ thus providing optimal photon energy to properly excite carriers in the TiO₂ substrate as demonstrated by (Fig. 3a). In (Fig. 3c) we show that the 405 nm laser produces a clear photocurrent, and its magnitude is correlated with the laser power. We also observed a reliable switching of the photocurrent switching through multiple on/off cycles, demonstrating its survivability as demonstrated by (Fig. 3e). We additionally examined the quantum efficiency of the TiO₂ samples under various laser powers in (Fig. 3f). The external quantum efficiency was calculated via

$$EQE = \frac{hc\Delta I}{e\lambda P} \quad (1)$$

Where ΔI is the photoexcited current, P is the total light power irradiated on the VO₂, h is Planck's constant, e is electron charge, and λ is the excitation wavelength. We find that as we decrease the laser power on the surface of the VO₂ the quantum efficiency exhibits a 2000% increase through the laser power reduction as demonstrated by (Fig. 3f). This is potentially due to the carrier excitation in the substrate being inefficient at high powers due to the large influx of photons and the number of available carriers is less than the number of photons available to excite said carriers; as the laser power is decreased the photon to carrier excitation matching (and thus the overall carrier excitation per photon absorbed) is improved, in turn improving the overall quantum efficiency. We also attempted to measure the photocurrent of VO₂ on c-Al₂O₃(0001) upon 405 nm illumination. Since the

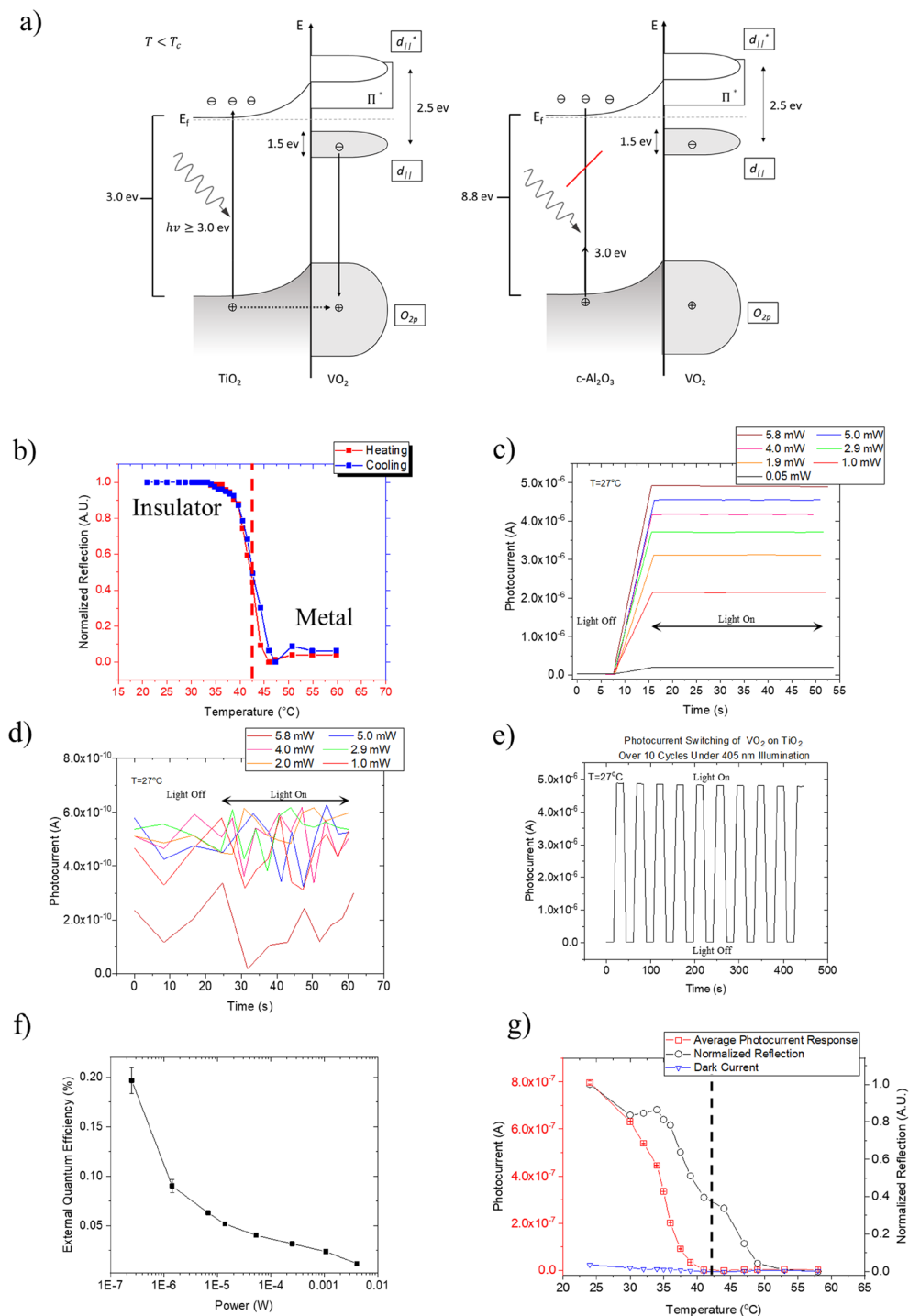


Figure 3. The electronic structure of VO₂ on TiO₂ and c-Al₂O₃ as well as studies of the photoelectric properties of VO₂ on TiO₂. **(a)** Schematic band diagram of the hole transfer mechanism for VO₂ on TiO₂ and c-Al₂O₃ where the photon energy is demonstrated for the VO₂ on TiO₂ as being sufficient to excite carriers in the TiO₂ layer and insufficient in the case of VO₂ on c-Al₂O₃. Where the hole transfer is designated by the segmented line and the carrier movement is designated by the solid lines^{6,19}. **(b)** Reflectivity measurements with a 405 nm diode laser illuminating VO₂ deposited on TiO₂(001) as the sample underwent a thermally induced IMT via heating and subsequent cooling where the dashed line indicated transition temperature. **(c)** The photocurrent switching of VO₂ on TiO₂(001) upon solely 405 nm illumination as the laser power was varied. The light was held on for 40 seconds post switching. **(d)** The same photocurrent switching measurement of VO₂ on c-Al₂O₃(0001) where the 405 nm laser power was varied. The light was held on for 40 seconds post switching. **(e)** The photocurrent switching cycle for VO₂ on TiO₂(001) upon 405 nm illumination switching over 10 cycles. **(f)** External quantum efficiency measurement for VO₂ on TiO₂ as varied with laser power. **(g)** The average photocurrent and optical response of VO₂ on TiO₂(001) as the sample was thermally ramped through the optical transition where the 405 nm laser power was 1 mW through the thermal ramping where the dashed line indicates transition temperature.

405 nm light's photon energy is insufficient to excite carriers in $c\text{-Al}_2\text{O}_3(0001)$ across its ~ 8.8 eV²¹ band gap, we see limited to no photocurrent in the VO_2 film as shown in (Fig. 3d). We in fact found that the produced photocurrent was 4 orders of magnitude smaller and on the order of the dark current, the small current that exists in photosensitive devices when in absence of photons, leading to the lack of defined features in the corresponding graph as opposed to VO_2 on $\text{TiO}_2(001)$.

Finally, we examined the photocurrent produced upon heating the sample and correlated the transition with the photocurrent. As demonstrated in (Fig. 3g), we observe that as the sample is heated, a marked change in the photocurrent produced is noted, approaching zero until it eventually reaches zero. This reduction in photocurrent is due to the IMT where the collapse of the band gap results in VO_2 becoming metallic preventing the production of a photocurrent as previously described. Here our reflected optical signal tracks the phase change optically through the transition, where a reduction in optical response is correlated with the phase change from monoclinic to rutile. However, we note that the photocurrent reduces to zero before the optical transition is complete. This finding would suggest that the migration of carriers in VO_2 is greatly reduced before the band gap is fully collapsed. One possible explanation follows from earlier descriptions of the nucleation of the metallic phase upon heating where localized puddles of the metallic phase in the VO_2 sample become so prevalent such that hole migration from the substrate to the film is substantially impeded preventing the production of photocurrent before the film is fully transitioned to the metallic phase^{1,4,22}.

Thus, local "puddle" formation results in loss of effective carrier movement before the sample has fully transitioned such that a global surface current is no longer possible reducing the photocurrent to zero although the surface of the sample is not fully phase transitioned. This "puddling" across the thermally induced IMT in VO_2 has been extensively studied and reported^{1,4,22}.

Discussion

We have shown that epitaxially grown VO_2 on both $c\text{-Al}_2\text{O}_3$ and $\text{TiO}_2(001)$ by pulsed DC sputtering exhibit good crystal character and a low degree of mosaicity. They also exhibit a characteristic surface morphology with terracing plateaus due to strain. We have been able to accurately compare simulations of VO_2 RHEED patterns consistent with our experimental data on the films that we have studied. We have also presented a first of its kind RHEED temperature transition analysis for VO_2 in which we were able to analyze and conceptualize the structural phase transition of VO_2 on both $c\text{-Al}_2\text{O}_3$ and $\text{TiO}_2(001)$, evidencing the dynamical changes that VO_2 undergoes through its critical transition temperature.

We also determined the photocurrent switching of VO_2 on $\text{TiO}_2(001)$ and $c\text{-Al}_2\text{O}_3(0001)$ at 405 nm and saw a stark reduction in photocurrent for VO_2 grown on $c\text{-Al}_2\text{O}_3(0001)$. This reduction in photocurrent is likely a product of the high resistivity and large substrate band gap as well as VO_2 films with much rougher surface, where roughness also contributes to scattering effects resulting in lower carrier efficiency thus reducing the photocurrent. We have also demonstrated a marked increase in the quantum efficiency of the VO_2 on TiO_2 with decreasing laser power seeing as large as a 2000% difference in quantum efficiency from 4.6 mW to 250 nW laser power. Finally, we have demonstrated how the IMT influences the produced photocurrent under illumination determining that as the VO_2 transitions through the IMT to the metallic state there is a significant reduction of produced photocurrent, reaching zero value once the global state of the surface is metallic hindering production of photocurrent for VO_2 .

Methods

Sample growth. Epitaxial VO_2 films were sputtered-deposited on one side polished (1-sp) $c\text{-Al}_2\text{O}_3(0001)$ and 1-sp $\text{TiO}_2(001)$ substrates in an ultrahigh vacuum (UHV) deposition system with a base pressure $\sim 3.0 \times 10^{-7}$ Torr range. The films were deposited via reactive D.C. pulsed magnetron sputtering with a vanadium target of 99.95% purity in a 90% Ar and 10% O_2 environment at 550 °C. Prior to deposition the substrates were annealed *in-situ* for 30 minutes at 600 °C to de-gas and recrystallize the top-most surface layers on each substrate. The target voltage was periodically pulsed to a positive charge to eject the excess insulating oxide layers that usually poison the surface of the target during sputter deposition. Growth rates were determined via X-Ray Reflectivity (XRR) in calibration samples. The films were evaluated via Reflection High-Energy Electron Diffraction (RHEED), Atomic Force Microscopy (AFM), X-Ray Diffraction (XRD), and Van der Pauw 4-point probe (VDP) measurement.

Rheed, AFM, XRD, and photocurrent characterization. RHEED was used to determine the crystallographic structure of the film surface. This technique utilizes glancing incident angle geometry for an electron beam on the probed surface allowing the beam to sample mainly the top-most layers of the film^{13,14}. Thus, RHEED enables qualitative and quantitative descriptions of the in-plane surface microstructure. The space groups for the two phases of VO_2 across the IMT are $P4_2/mmm(136)$ for rutile with lattice spacing $a = 4.55$ Å, $c = 2.88$ Å, and $P2_1/c(14)$ for monoclinic with lattice spacing $a = 5.75$ Å, $b = 5.42$ Å, $c = 5.38$ Å (we note that the drastic change in c lattice spacing is due to a doubling of the unit cell along the c direction)¹⁵. This information allows comparisons of various experimental diffraction patterns to predicted models for both phases. This technique also allows us to map the crystallographic changes through the thermally induced IMT.

The surface morphology of the samples was characterized via *ex-situ* AFM. The AFM images were collected using a Nanotec Cervantes AFM instrument. The AFM was operated in non-contact mode with an AppNano ACTA AFM tip with nominal tip diameter ~ 10 nm. The software WSxM from Nanotec was used for analysis of the AFM images²³. All images were flattened in the WSxM program and the images were processed via the root mean squared (RMS) analysis tool in the program. The RMS roughness gives an average estimate of the surface roughness of the film, and this quantitative description can be paired to the qualitative description obtained from RHEED.

The microstructure of the films throughout their full thickness and out of plane crystallographic information was determined via XRD. The instrument used was a four-circle diffractometer with a quasi-monochromatic CuK_{α} ($\lambda = 1.5406 \text{ \AA}$) beam. This technique was used to determine the out of plane lattice parameter, grain size²⁴, and mosaicity of the films.

The electrical and quantum efficiency information was determined with an MMR Technology Variable Temperature Microprobe System (VTMP) and a 405 nm diode laser for the VO_2 grown on TiO_2 as well as a 1520 nm laser for the VO_2 grown on $c\text{-Al}_2\text{O}_3$. The sheet resistance was determined via the Van der Pauw method²⁵. The samples were illuminated with a 405 nm laser and the produced photocurrent was measured. The samples were then thermally ramped through the IMT and both the photocurrent, using the VTMP, and the optical response, using a 405 nm photodetector, were recorded simultaneously for each sample.

RHEED simulations. The simulations for this experiment were carried out using the RHEEDsim MATLAB program²⁶. This software utilizes the atomic positions of the atoms in VO_2 films as well as TiO_2 and $c\text{-Al}_2\text{O}_3$ substrates determined from the space groups of the respective molecules. The program utilizes a kinematic approach for single scattering events, and then approximates the Ewald sphere to a planar surface to compute streak intensities. Additionally, intensity modulation extends to within the z direction²⁶. The form factors for atomic scattering were determined via NIST standards for X-ray form factor, attenuation, and scattering tables²⁷. This simulations yield realistic representations for comparison to experimentally determined RHEED diffraction patterns.

References

1. Wang, L. *et al.* Distinct Length Scales in the VO_2 Metal-Insulator Transition Revealed by Bi-chromatic Optical Probing. *Adv. Opt. Mater.* **2**, 30–33 (2014).
2. Morin, F. J. Oxides which show a metal-to-insulator transition at the neel temperature. *Phys. Rev. Lett.* **3**, 34–36 (1959).
3. Zylbersztejn, A. & Mott, N. F. Metal-insulator transition in vanadium dioxide. *Phys. Rev. B* **11**, 4383–4395 (1975).
4. Qazilbash, M. M. *et al.* Electrostatic modification of infrared response in gated structures based on VO_2 . *Appl. Phys. Lett.* **92**, 241906 (2008).
5. Chae, B. G., Kim, H. T., Youn, D. H. & Kang, K. Y. Abrupt metal-insulator transition observed in VO_2 thin films induced by a switching voltage pulse. *Phys. B Condens. Matter* **369**, 76–80 (2005).
6. Muraoka, Y., Yamauchi, T., Ueda, Y. & Hiroi, Z. Efficient photocarrier injection in a transition metal oxide heterostructure. *J. Phys. Condens. Matter* **14**, L757–L763 (2002).
7. Zhou, S., Wang, Y. & Wang, Z. Doublon-holon binding, Mott transition, and fractionalized antiferromagnet in the Hubbard model. <https://doi.org/10.1103/PhysRevB.89.195119> (2013).
8. Oka, T. & Aoki, H. Ground-state decay rate for the Zener breakdown in band and mott insulators. *Phys. Rev. Lett.* **95**, 137601 (2005).
9. Oka, T. Nonlinear doublon production in a Mott insulator: Landau-Dykhne method applied to an integrable model. *Phys. Rev. B - Condens. Matter Mater. Phys.* **86**, 075148 (2012).
10. Tellier, C. R. & Tosser, A. J. Thermoelectric power of metallic films in the Mayadas-Shatzkes model. *Thin Solid Films* **41**, 161–166 (1977).
11. Tellier, C. R. A theoretical description of grain boundary electron scattering by an effective mean free path. *Thin Solid Films* **51**, 311–317 (1978).
12. Muraoka, Y. & Hiroi, Z. Metal-insulator transition of VO_2 thin films grown on TiO_2 (001) and (110) substrates. *Appl. Phys. Lett.* **80**, 583–585 (2002).
13. Mahan, J. E., Geib, K. M., Robinson, G. Y. & Long, R. G. A review of the geometrical fundamentals of reflection high-energy electron diffraction with application to silicon surfaces. *J. Vac. Sci. Technol. A Vacuum, Surfaces, Film.* **8**, 3692–3700 (1990).
14. Droopad, R., Williams, R. L. & Parker, S. D. RHEED intensity oscillations observed during the MBE growth of InSb (100). *Semicond. Sci. Technol.* **4**, 111–113 (1989).
15. Leroux, C., Nihoul, G. & Van Tendeloo, G. From $\text{VO}_2(\text{B})$ to $\text{VO}_2(\text{R})$: Theoretical structures of VO_2 polymorphs and *in situ* electron microscopy. *Phys. Rev. B* **57**, 5111–5121 (1998).
16. Gražulis, S. *et al.* Crystallography Open Database (COD): an open-access collection of crystal structures and platform for world-wide collaboration. *Nucleic Acids Res.* **40**, D420–D427 (2012).
17. Gražulis, S. *et al.* Crystallography Open Database - An open-access collection of crystal structures. *J. Appl. Crystallogr.* **42**, 726–729 (2009).
18. Radue, E. *et al.* Substrate-induced microstructure effects on the dynamics of the photo-induced metal-insulator transition in VO_2 thin films. *J. Opt.* **17**, 025503 (2015).
19. Shin, S. *et al.* Vacuum-ultraviolet reflectance and photoemission study of the metal-insulator phase transitions in VO_2 , V_6O_{13} , and V_2O_3 . *Phys. Rev. B* **41**, 4993–5009 (1990).
20. Pascual, J., Camassel, J. & Mathieu, H. Resolved quadrupolar transition in TiO_2 . *Phys. Rev. Lett.* **39**, 1490–1493 (1977).
21. Robertson, J. Band offsets of wide-band-gap oxides and implications for future electronic devices. *J. Vac. Sci. Technol. B Microelectron. Nanom. Struct.* **18**, 1785 (2000).
22. Qazilbash, M. M. *et al.* Mott Transition in VO_2 Revealed by Infrared Spectroscopy and Nano-Imaging. *Science* (80-.). **318**, 1750–1753 (2007).
23. Horcas, I. *et al.* WsXM: A software for scanning probe microscopy and a tool for nanotechnology. *Rev. Sci. Instrum.* **78**, 013705 (2007).
24. Patterson, A. L. The scherrer formula for X-ray particle size determination. *Phys. Rev.* **56**, 978–982 (1939).
25. van der Pauw, L. J. A Method of Measuring The Resistivity And Hall Coefficient On Lamellae Of Arbitrary Shape. *Phillips Tech. Rev.* **20**, 220–224 (1958).
26. Wang, K. & Smith, A. R. Efficient kinematical simulation of reflection high-energy electron diffraction streak patterns for crystal surfaces. 1–9 (2011).
27. Chantler, C. T. Detailed tabulation of atomic form factors, photoelectric absorption and scattering cross section, and mass attenuation coefficients in the vicinity of absorption edges in the soft X-ray ($Z = 30\text{--}36$, $Z = 60\text{--}89$, $E = 0.1\text{--}10 \text{ keV}$) – addressing convergence iss. *J. Synchrotron Radiat.* **8**, 1124–1124 (2001).

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Author Contributions

J.A.C., S.E.M., D.B.B. and M.R.B. contributed to the experiments. J.A.C. analyzed the data. J.A.C. prepared the figures in the manuscript. J.A.C. wrote the main manuscript. I.N. and R.A.L. supervised the project. All authors reviewed the manuscript.

Additional Information

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