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## Thermal Stability of Hole-Selective Tungsten Oxide: *In Situ* Transmission Electron Microscopy Study

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In this study, the thermal stability of a contact structure featuring hole-selective tungsten oxide ( $\text{WO}_x$ ) and aluminum deposited onto *p*-type crystalline silicon (*c*-Si/ $\text{WO}_x/\text{Al}$ ) was investigated using a combination of transmission line measurements (TLM) and *in situ* transmission electron microscopy (TEM) studies. The TEM images provide insight into why the charge carrier transport and recombination characteristics change as a function of temperature, particularly as the samples are annealed at temperatures above 500 °C. In the as-deposited state, a  $\approx 2$  nm silicon oxide ( $\text{SiO}_x$ ) interlayer forms at the *c*-Si/ $\text{WO}_x$  interface and a  $\approx 2$ –3 nm aluminum oxide ( $\text{AlO}_x$ ) interlayer at the  $\text{WO}_x/\text{Al}$  interface. When annealing above 500 °C, Al diffusion begins, and above 600 °C complete intermixing of the  $\text{SiO}_x$ ,  $\text{WO}_x$ ,  $\text{AlO}_x$  and Al layers occurs. This results in a large drop in the contact resistivity, but is the likely reason surface recombination increases at these high temperatures, since a *c*-Si/Al contact is basically being formed. This work provides some fundamental insight that can help in the development of  $\text{WO}_x$  films as hole-selective rear contacts for *p*-type solar cells. Furthermore, this study demonstrates that *in situ* TEM can provide valuable information about thermal stability of transition metal oxides functioning as carrier-selective contacts in silicon solar cells.

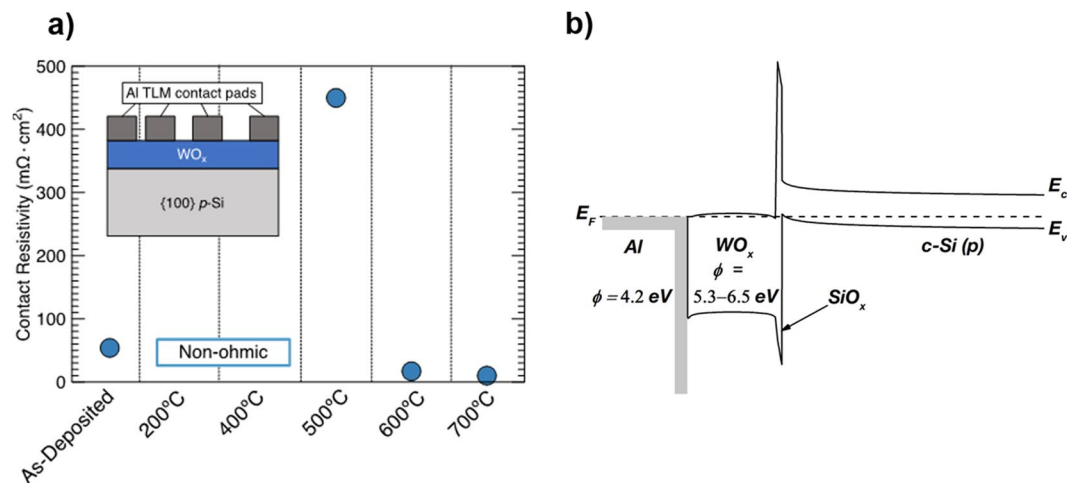
To achieve high conversion efficiencies (>25%) for a single junction crystalline silicon (*c*-Si) solar cell, it is essential to couple low recombination velocities in the bulk and at the Si surface with suitable carrier selectivity at both electron and hole contact regions. These objectives can be achieved by inserting carrier selective contacts at both the front and rear surfaces of a silicon heterojunction cell. They help in passivating the Si surface by reducing surface recombination and they are carrier-selective in nature, meaning that they allow only one type of carrier, i.e., either of electron or holes to pass through it while blocking the other<sup>1–6</sup>.

Traditionally, carrier selective contacts have been realized using doped hydrogenated amorphous silicon (*a*-Si:H), wherein *a*-Si:H(*n*) and *a*-Si:H(*p*) act as electron-selective and hole-selective contacts respectively. These contacts are typically used in combination with a thin *a*-Si:H(*i*) or  $\text{SiO}_2$  passivation layer. However, such cells suffer from certain inherent drawbacks associated with *a*-Si:H such as thermal instability, parasitic photon absorption, complicated deposition processes and high fabrication costs<sup>1,7–11</sup>.

Transition metal oxides have emerged as promising alternatives to doped *a*-Si:H for use as carrier selective contacts in silicon heterojunction cells. Depending on the valence-band and conduction-band offset between the metal oxide and Si, transition metal oxides can be employed as either an electron-selective or a hole-selective contact. For instance, titanium oxide ( $\text{TiO}_2$ ) has a small conduction band offset ( $\Delta E_c = 0.05$  eV), which provides a low barrier for electrons to pass through the  $\text{TiO}_2$  layer and a large valence-band offset ( $\Delta E_v = 2.0$  eV) that results in holes being blocked. Due to this reason,  $\text{TiO}_2$  (<5 nm) has been employed as an electron-selective rear contact for an *n*-type cell with very promising results<sup>1,7,8,12,13</sup>.

On the other hand, if metal oxides have a wide band gap and high work function, the strong work function difference between Si and metal oxide induces a strong upward band bending in the *c*-Si which results in electrons being blocked. Because of this reason, sub-stoichiometric metal oxides such as molybdenum oxide ( $\text{MoO}_x$ ) and tungsten oxide ( $\text{WO}_x$ ) have emerged as promising candidates for use as hole-selective contacts. Because of their wide band gap (>3 eV) and consequently their high transparency, these materials have been employed as

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**Figure 1.** (a) Contact resistivity at various annealing temperatures obtained by TLM and (b) the energy band diagram of the WO<sub>x</sub>/Al contact to *p*-type *c*-Si.

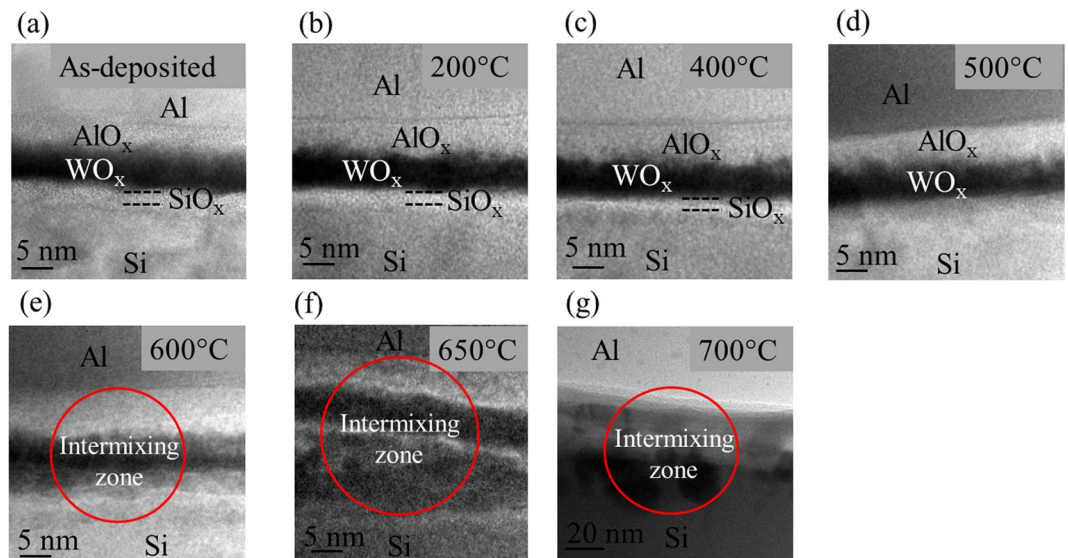
hole-selective front contacts in combination with a transparent conducting oxide (TCO) layer such as hydrogenated indium oxide (IO:H) or indium tin oxide (ITO)<sup>2,14–18</sup>. However, a major limitation of these contact structures is that they are sensitive to low temperature annealing resulting in degradation of device performance. Although, the origin of these losses is yet to be fully understood, it is likely due to a reduction in the work function driven by hydrogenation and/or a lower oxygen concentration resulting in non-ideal hole-selectivity<sup>16,19</sup>.

Although, wide band gap and high work function transition metal oxides (e.g., MoO<sub>x</sub>, WO<sub>x</sub>) in combination with TCO (e.g., IO:H, ITO) have been widely investigated as hole-selective front contacts for *n*-type solar cells; their application as potential hole-selective rear contacts are yet to be fully explored. Recently, Lee *et al.* demonstrated that WO<sub>x</sub>/Al can be a potential candidate to be employed as a full area hole-selective rear contact for a *p*-type Si solar cells<sup>20</sup>. In the present work, the thermal stability of a *c*-Si/WO<sub>x</sub>/Al contact structure was investigated while being subjected to annealing temperatures up to 700 °C with the help of *in situ* transmission electron microscopy (TEM). The contact resistivity at various annealing temperatures was measured by transmission line measurements (TLM). The objective is to achieve an in-depth understanding of the mechanisms influencing the stability of WO<sub>x</sub>/Al contacts when subjected to high temperature annealing at the microscopic scale.

The values of contact resistivity measured at various annealing temperatures are shown in Fig. 1(a) and the inset illustrates the schematic of TLM structures employed for contact resistivity measurements. A typical energy band diagram of WO<sub>x</sub>/Al contact to *p*-type *c*-Si is shown in Fig. 1(b)<sup>21</sup>. In the as-deposited state, the electrical behavior of the contact is ohmic and a contact resistivity value of 54 mΩ · cm<sup>2</sup> was obtained. Moreover, it is evident from cross-sectional HRTEM micrographs shown in Fig. 2(a) that even in the as-deposited state, an AlO<sub>x</sub> interlayer is formed at the WO<sub>x</sub>/Al interface. A similar behavior has been previously reported for Al/TiO<sub>2</sub> contacts as well<sup>1,8</sup>. Likewise, in the case of WO<sub>x</sub>/Al, because of the higher oxygen affinity of Al as compared to the oxygen affinity of W, diffusion of oxygen is more likely to occur from WO<sub>x</sub> towards Al resulting in formation of AlO<sub>x</sub> interlayer. This can lead to the creation of defect states such as oxygen vacancies within the WO<sub>x</sub> layer. Moreover, a ≈ 2 nm SiO<sub>x</sub> interlayer is also observed at the *c*-Si/WO<sub>x</sub> interface. This is consistent with observation made by Sachetto *et al.* It is obvious that the formation of a SiO<sub>x</sub> interlayer occurs during deposition of WO<sub>x</sub> by thermal evaporation and can be explained by thermodynamic considerations<sup>22</sup>. An effective carrier lifetime ( $\tau_{\text{eff}}$ ) value of 53 μs at an injection level ( $\Delta n$ ) of 1 · 10<sup>15</sup> cm<sup>-3</sup> has been reported for as-deposited WO<sub>x</sub>/Al<sup>20</sup>. It can be concluded that, in the as-deposited state, although a SiO<sub>x</sub> interlayer forms, this layer provides very little surface passivation, and likely has a reasonably high interface defect density. Additionally, both the work function and film conductivity are sufficiently high to form an ohmic contact with a reasonably low contact resistivity in the as-deposited state.

When the sample is annealed to 200 °C no significant change is observed (Fig. 2(b) and video 2(b)). However, TLM results show that contact structure becomes non-ohmic. One possible explanation for the non-ohmic transport is a reduction in the work function upon annealing in ambient air resulting in the formation of a barrier to hole collection. Additionally, if oxygen is diffusing from WO<sub>x</sub> to the SiO<sub>x</sub> interlayer, the SiO<sub>x</sub> interlayer may become more insulating<sup>22</sup>. This, however, requires further investigation. With respect to surface passivation of the *c*-Si/WO<sub>x</sub>/Al contact stack, it has been previously reported that  $\tau_{\text{eff}}$  drops from 53 μs to 16 μs ( $\Delta n = 1 \cdot 10^{15}$  cm<sup>-3</sup>) when annealed at 200 °C<sup>20</sup>.

When the annealing temperature is increased to 400 °C, no apparent structural change is observed in the contact stack and the electrical behavior remains non-ohmic (Fig. 2(c) and video 2(c)). However, previous work on *c*-Si/WO<sub>x</sub>/Al reported that  $\tau_{\text{eff}}$  actually increases upon annealing at 400 °C based on photoluminescence images of asymmetrical test structure, although it is unclear whether this increase was due to improved passivation of the SiO<sub>x</sub> interlayer at the WO<sub>x</sub> (i.e., rear) side of the device, or due to an improvement in the surface passivation at the front side of the device<sup>20</sup>. Yang *et al.* have reported that in case of SiO<sub>2</sub>/TiO<sub>2</sub>/Al rear contacts, annealing at 350 °C is essential to activate the surface passivation of SiO<sub>2</sub><sup>7,8</sup>. Therefore, it is possible that when the annealing temperature is increased to 400 °C, it activates the passivation mechanism of SiO<sub>x</sub> which results in increase in  $\tau_{\text{eff}}$ .



**Figure 2.** Cross-sectional HRTEM micrograph of *c*-Si/WO<sub>x</sub>/Al structure obtained by *in situ* TEM studies (a) as-deposited, (b) 200 °C, (c) 400 °C, (d) 500 °C, (e) 600 °C, (f) 650 °C, (g) 700 °C.

When the annealing temperature is further increased to 500 °C, interfaces become diffused, indicating that significant phase changes may begin to occur at 500 °C (Fig. 2(d) and video 2(d)). Furthermore, the TLM studies revealed that the behavior of the contact structure changes from non-ohmic to ohmic, although the contact resistivity value obtained was significantly higher at 450 mΩ·cm<sup>2</sup>. It has been previously reported that crystallization of WO<sub>x</sub> occurs at 400 °C<sup>20</sup>, although no evidence of crystallization was found during *in situ* TEM observation. It is likely that WO<sub>x</sub> begins to crystallize at 400 °C and continues when further annealed to 500 °C. However, short annealing times (≈10 minutes) are insufficient for the WO<sub>x</sub> to undergo complete crystallization, even at 500 °C<sup>20</sup>. Therefore, ohmic behavior upon annealing to 500 °C with a relatively higher contact resistivity can be attributed to partial crystallization of WO<sub>x</sub>.

As already discussed, when annealing temperature is increased beyond 500 °C, it is likely that the phase changes begin to occur in WO<sub>x</sub>/Al. Furthermore, at 600 °C, intermixing takes place between SiO<sub>x</sub>, WO<sub>x</sub>, AlO<sub>x</sub> and Al layers (Fig. 2(e) and video 2(e)). At 650 °C, the layers become indistinguishable (Fig. 2(f) and video 2(f)). At 700 °C, intermixing between Si, WO<sub>x</sub> and Al appears to be near completion and Al is in direct contact with Si (Fig. 2(g)). This can be explained in terms of thermodynamic considerations. It is well known that the melting point of Al is 660 °C, and in the range of 100–150 °C below the solvus line, Al begins to lose its integrity and diffusion begins. This explains why diffused interfaces were observed at a temperature of 500 °C. Al is the faster diffusing species than W because of the much smaller atomic radius. And as the annealing temperature is increased beyond 500 °C, Al atoms diffuse towards Si substrate with an exponentially increasing rate of diffusion with temperature. Eventually, at 650 °C, which is near the melting point of pure Al, significant intermixing occurs. Finally, at 700 °C, the intermixing process nears completion and the Al layer comes in direct contact with the Si. Therefore, much lower contact resistivity values of 17 mΩ·cm<sup>2</sup> and 10 mΩ·cm<sup>2</sup> were obtained at 600 °C and 700 °C, respectively. This implies that drastic drop in contact resistivity upon annealing at 600 °C and beyond is most likely due to direct contact between Al and Si. However, this is an adverse effect on carrier selectivity and surface passivation, which is consistent with lower  $\tau_{\text{eff}}$  values reported at 600 °C<sup>20</sup>, and basically negates the whole purpose of using WO<sub>x</sub>.

Overall, it emerged that although WO<sub>x</sub>/Al is a potential candidate to be employed as a hole-selective rear contact on a *p*-type cell, certain issues remain. To activate the passivation mechanism of SiO<sub>x</sub>, annealing samples at 400 °C may be sufficient, but WO<sub>x</sub>/Al becomes non-ohmic at that temperature. This can possibly be overcome by annealing WO<sub>x</sub>/Al for longer times to allow sufficient time for crystallization of WO<sub>x</sub>, which can make it more conductive, but care must be taken to not reach the point of intermixing resulting in essentially a *c*-Si/Al contact.

In summary, some fundamental insight that can help in the development of WO<sub>x</sub> films as hole-selective rear contacts for *p*-type solar cells has been provided in this study. Furthermore, this work has successfully demonstrated that *in situ* TEM can provide valuable information about thermal stability of transition metal oxides employed as carrier-selective contacts in silicon solar cells.

## Experimental Section

WO<sub>x</sub> thin films (<10 nm) were deposited on 5–10 Ω·cm *p*-type FZ {100} *c*-Si wafers under vacuum by thermal evaporation using a powder WO<sub>3</sub> source. During the evaporation process, heating of the *c*-Si substrate was minimized and remained close to ambient temperature. This was followed by evaporation of 500 nm of Al to form the metal contact. For the contact resistivity measurements, TLM structures were fabricated, previously shown as an inset in Fig. 1(a). Dark I-V measurements were taken for the TLM contact pairs at different spacings using a Keithley 2400 Sourcemeater. This was done on samples exposed to different post-deposition annealing

temperatures. This data was then used to extract the contact resistivity of the  $\text{WO}_x$  contact stack, and the data was corrected for current spreading due to the absence of an emitter using the extended TLM<sup>23</sup>.

For TEM studies, cross-sectional TEM specimens were prepared by the focused ion beam (FIB) milling technique using FEI 200 TEM FIB. Specimen lift-out was done *in situ* and attached to Cu grid. To monitor morphological changes occurring at elevated temperatures, the sample was analyzed using TEM and was heated *in situ*. The *in situ* TEM experiments were performed using a Gatan heating holder (Model 652) in combination with a FEI Tecnai F30 under operating voltage of 300 kV. The samples were annealed *in situ* up to a temperature of 700 °C with a heating rate of 50 °C/min and an annealing time for each temperature of 10 min. However, a major limitation of using a Gatan heating holder (652) is that it is not compatible with energy dispersive x-ray spectroscopy (EDX) systems. This is because the TEM specimen is surrounded by a furnace without any direct line of sight from the sample to the EDX detector due to which it is unable to detect characteristic x-rays essential for determination of chemical composition.

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## Author Contributions

H.A.: experimental design, manuscript preparation, *in situ* TEM studies; S.K.: manuscript preparation, *in situ* TEM studies; G.G.: manuscript preparation, TLM studies; J.B.: manuscript preparation, fabrication; A.J.: experimental design, manuscript preparation; A.K.: experimental design, manuscript preparation; K.D.: experimental design, manuscript preparation.

## Additional Information

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