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OPEN Fluorescence dynamics of the biosynthesized CdSe quantum dots in Candida utilis

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Organisms served as factories of bio-assembly of nanoparticles attracted a lot of attentions due to the safe, economic and environmental-benignity traits, especially the fabrication of the super fluorescence properties quantum dots (QDs). However, information about the developmental dynamics of QDs in living organisms is still lacking. In this work, we synthesized cadmium-selenium (CdSe) QDs in Candida utilis WSH02-08, and then tracked and quantitatively characterized the developmental dynamics (photoactivation, photostable and photobleaching processes) of bio-QDs by translating fluorescence microscopy movies into visual quantitative curve. These findings shed light on the fluorescence properties of the bio-assembled QDs and are expected to accelerate the applications of the synthesized QDs in vivo. It provided a new way to screen bio-QDs and monitor the quality of QDs in vivo.

Quantum dots (QDs) are nano-scale fluorescent materials with high photo-stability, narrow and size-tunable emission spectra, and broad spectral window¹. These amazing optical properties of QDs have motivated intensive research interests from material scientists, chemists and biologists worldwide to explore their massive production and potential applications²⁻⁴. Biosynthesis, with many advantages such as facile operation, low cost and environmental benignity, is considered as a promising approach to realize large-scale production of QDs^{5,6}.

To date, an increasing number of in vivo biosynthesized QDs (bio-QDs) with excellent hydrophilicity and inherent biocompatibility by various types of microorganisms, from single-cellular bacteria⁷⁻⁹ to higher-level animals¹⁰, have been reported. For the practical bioimaging/biomedical applications of bio-QDs, several aspects of fluorescence properties are of particular interests: fluorescence intensity; sensitivity to photoactivation (i.e., the rate of fluorescence enhancement under light excitation)^{11, 12}; and photostability¹³. High photostability means a slow photobleaching (the loss of fluorescence under prolonged illumination) and a long photostable lifetime (here defined as the time span when the fluorescence intensity is no less than 90% of their peak value)^{14, 15}. To evaluate the fluorescence properties, it is usually to extract the QDs from biomass carefully and remove the proteins from bio-QDs, which is time-consuming and not friendly to screen bio-QDs with high quality in industrial production^{16, 17}. Therefore it is of significance to study the fluorescence properties of bio-QDs in vivo, especially the developmental dynamics of bio-QDs in living organisms.

In this work, we applied Candida utilis WSH02-0818, to synthesize cadmium-selenium (CdSe) QDs, and probed and tracked the developmental dynamics of in vivo synthesized CdSe bio-QDs in living cells. The fluorescence intensity, photoactivation rate and index, photostable lifetime, and photobleaching decay time of the obtained CdSe bio-QDs were evaluated. To facilitate real-time monitoring of the fluorescence emission dynamics of the bio-QDs in vivo, we extracted their fluorescence evolution by translating fluorescence video into visual curves as a monitoring tool. This study could enable a better understanding of the fluorescence properties of bio-QDs. It also will provide a new way to screen bio-QDs and monitor the quality of QDs in vivo. This quick feedback may be beneficial for optimizing the parameters for the large-scale biosynthesis of QDs.

Results and Discussion

Bio-assemble of QDs in *Candida utilis***WSH02-08**. In order to synthesize CdSe bio-QDs, the *C. utilis* WSH 02-08 was cultivated with 1 mM sodium selenite (Na₂SeO₃) and 6 mM cadmium chloride (CdCl₂) (Fig. 1a). At the end of the synthesis process, the cells exhibited bright yellow fluorescence observed by fluorescence microscopy (Fig. 1b). The characteristic peaks of CdSe (203 cm⁻¹ and 406 cm⁻¹ assigned to longitudinal optical (LO)

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Figure 1. Characteristics of the CdSe QDs synthesized by *C. utilis* WSH02-08 with 1 mM Na₂SeO₃ and 6 mM CdCl₂. (**a**) Schematic illustration of bio-QDs fabrication process by *Candida utilis* WSH02-08. (**b**) Fluorescence microscopy image of cells exposed to Cd and Se salts. (**c**) *In situ* micro-Raman spectrum of the cells. (**d**) High-resolution transmission electron microscopy (HRTEM) image of the purified QDs. (**e**) The lattice planes spacing of the purified QDs is 0.25 nm. (**f**) X-ray energy-dispersive spectroscopy (EDS) analysis of the purified QDs.

and 2LO phonon of Cd-Se, respectively) were illustrated by *in situ* Raman spectrum. In order to get more detail morphology and structure information of the QDs, the nanoparticles were purified and analyzed by high-resolution transmission electron microscopy (HRTEM) and X-ray energy-dispersive spectroscopy (EDS). The purified QDs yield strong Se, Cd signals and the average diameter is 4.38 ± 1.30 nm (Fig. 1d and f). The HRTEM shows continuous lattice fringes with an interplanar lattice spacing of 0.25 nm, corresponding to the (102) plane of CdSe (Fig. 1e). All these evidences confirm the synthesis of CdSe QDs inside the cells of *C. utilis* WSH02-08.

Fluorescence dynamics of the bio-QDs. To examine the fluorescence properties of the CdSe QDs, we performed *in vivo* long term recording of fluorescence dynamics of the bio-QDs using fluorescence microscope. Interestingly, the fluorescence intensity of the bio-QDs in *C. utilis* WSH02-08 declined after 30 min illumination, suggesting a photobleaching process. To short the recording time, the bio-QDs synthesized at a lower Cd precursor concentration (i.e., 1 mM Na₂SeO₃ and 2 mM CdCl₂) were selected here for visualization of the fluorescence evolution over time (Fig. 2). Dim fluorescence appeared at 1.8 seconds after the illustration and kept strengthening till reaching a plateau. After about 13.8 seconds, the fluorescence started to decrease.

To quantitatively describe the fluorescence dynamic changes of the bio-QDs synthesized in *C. utilis* WSH02-08, we plotted the average fluorescence intensity of the whole image over time using image J software¹⁹. Four parameters of the fluorescence dynamics, i.e., the photoactication index and photoactivation rate, photostable lifetime and photobleaching decay time, were evaluated. The photostable lifetime of the bio-QDs was 12.4 seconds (Fig. 3a), while the photoactivation rate increased initially to 6.314 within 3 seconds and then decreased (Fig. 3b). Photoactivation process fit well with the stretched-exponential law ($I = I_0 - Aexp - [t/\tau]^{\beta}$). In this function, I is the fluorescence intensity, I_0 is the highest value of the fluorescence intensity, τ is the photoactivation half time, in which the fluorescence intensity reaches the value of $I_0/2$. A is a scaling coefficient, we called photoactivation index here and β is a stretching parameter. The best-fit results are presented in Fig. S1 and Table S1. The photobleaching process of the bio-QDs exhibites exponential decay rule and the photobleaching data were well fitted by the bi-exponential form of " $a_1exp(-t/\tau_1) + a_2exp(-t/\tau_2)$ ", where τ_1 represents the fast decay and τ_2 represents slow decay, and a_1 and a_2 are the amplitudes of the exponentials²⁰. The photobleaching consisted of fast decay and slow decay (Fig. 3c). The estimated fast decay time (the dominant parameter to judge the bleaching rate) was 35.57 seconds (Fig. 4a).

Tuning the fluorescence properties of the bio-QDs. The fluorescence properties of the above bio-QDs synthesized by *C. utilis* WSH02-08 at different $CdCl_2$ concentrations varied substantially, suggesting a considerable impact of cultivation conditions on the bio-QDs synthesis process and a feasibility of regulating the bio-QDs properties by adjusting the precursor concentration.

To validate this hypothesis, we comparatively evaluated the photoactivation index and rates, photostable lifetimes and photobleaching decay times of the bio-QDs in the WSH02-08, which were synthesized at 9 different



Figure 2. Time series fluorescence microscopy images of the cells to show the fluorescence evolution of the bio-QDs *in vivo*. The cells were grown in medium with 1 mM Na₂SeO₃ and 2 mM CdCl₂. The time interval between two images was 1.2 second. Video was taken from a 150- μ m² area where there were at least 80 cells.



Figure 3. Fluorescence curves of the bio-QDs in cells over time. (a) Variation of intracellular mean fluorescence intensity that shows the processes of the photoactivation and photobleaching, and photostable lifetime; (b) The derivative of non-linear function of the photoactivation process reveals the photoactivation rate; (c) Best-fit bi-exponential form of the photobleaching curve. PA, PL and PB represent photoactivation, photostable and photobleaching process, respectively.

combinations of precursor concentrations. As expected, substantial differences in the fluorescence intensities and dynamics of the bio-QDs were observed for the 9 tested groups (Fig. 5a). The Se1Cd6 group exhibited the maximum fluorescence intensity (162.6), while the Se5Cd2 group showed the minimum (28.5). In addition, the Se1Cd6 group also had the longest photostable lifetime, the fastest photoactivation rate and the maximum photoactivation index (Fig. 5b and S1). Here, Se1Cd6 represents the precursor concentration is 1 mM Na₂SeO₃ and 6 mM CdCl₂, while Se5Cd2 represents the precursor concentration is 5 mM Na₂SeO₃ and 2 mM CdCl₂. From the fitted curves, the average values of τ_1 under different Se content synthesis conditions are summarized in Fig. 4. The fastest decay time was observed for the Se5Cd2 group, indicating that a higher Se content could accelerate



Figure 4. Photobleaching process exposed to different selenium contents, and the bi-exponentials $(I = a_1 \exp[-t/\tau_1] + a_2 \exp[-t/\tau_2])$ fitted line. The red line represents the fitted data and the dark line represents experiment values. (a) Se1Cd2, (b) Se2Cd2, (c) Se3Cd2, (d) Se4Cd2, (e) Se5Cd2.



Figure 5. Fluorescence properties of the bio-QDs in *C. utilis* WSH02-08. (a) Intracellular mean fluorescence intensity trajectories in different synthesized bio-QDs recorded under lamp illumination using fluorescence microscopy; (b) The derivative of non-linear function of the photoactivation process under different synthesis conditions; (c) Best-fit photobleaching decay time constants for the bi-exponential form of the bio-QDs. All the experiments were conducted under the same conditions.

the photobleaching of the bio-QDs in WSH02-08. These results indicate that a higher Cd and a lower Se content would increase the photoactivation rate and photostable time of the bio-QDs, and decrease the photobleacing rate, resulting in better photostable bio-QDs. These results also reflect that such fluorescence dynamic parameters are facile index to screen good fluorescence properties QDs.

Conclusion

In summary, we synthesized bio-QDs using *C. utilis* WSH02-08 and monitored their fluorescence dynamics *in vivo*. The critical parameters to evaluate the bio-QDs, e.g., fluorescence intensity, photostable lifetime, photoactivation and photobleaching decay time, were obtained and quantified. In addition, higher Cd contents are found to favor the formation of the bio-QDs with a higher fluorescence intensity and photoactivation index in *C. utilis* WSH02-08, implying the possibility of using fluorescence dynamics parameters as screen index. Benefited from these results, the preparation of bio-QDs with various compositions (e.g., CdTe, CdS) by different microorganisms can be expected, which may ultimately make the bioimaging/biosensing applications of bio-QDs a real possibility. Therefore, our work provides an opportunity to gain new microscopic insights into the developmental dynamics of bio-QDs and their fluorescence properties in microbial cells, and is expected to accelerate the pace of development of environmentally friendly and cost-effective ways for bio-assembled QDs.

Methods

Biosynthesis of CdSe QDs in *Candida utilis. Candida utilis* WSH02-08 strain was cultured into the medium (30 g/L glucose, 10 g/L ammonium sulfate, 6 g/L yeast extract, 4.5 g/L potassium dihydrogen phosphate, 0.75 g/L magnesium sulfate, 14.7 g/L sodium citrate dehydrate and pH is 5.5) for 36 h at $30 \text{ }^{\circ}\text{C}$ under 200 rpm sharking in aerobic condition. The activated strain was incubated with Na₂SeO₃ and glucose (10 g/L) for another 24 h. Then CdCl₂ was added to finish the QDs fabrication for another 22 h. All the bio-assembled QDs process were conducted at $30 \text{ }^{\circ}\text{C}$ under 200 rpm sharking in aerobic conditions. After the synthesis procedure, we collected the cells by centrifugation ($6000 \text{ g}, 4 \text{ }^{\circ}\text{C}, 5 \text{ min}$) and washed two times with 10 mM Tris-Cl (pH = 7.6).

Fluorescence microscopic observation. To monitor the fluorescence dynamics process of the bio-QDs *in vivo*, the collected cells were observed using a wide field fluorescence microscope (BX-51, Olympus Co., Japan) with a wideband MWU2 filter (Ex 330–385 nm). All images were taken using a water immersion objective $(100 \times)$ under 120 W mercury lamp (X-Cite 120 Q) irradiation. Furthermore, the images and videos were recorded by the DP2-BSW software (Olympus Co., Japan) under identical conditions. The videos were recorded at 15 frame/s with a CCD camera (DP72, Olympus Co., Japan) and the recording duration is limited to 1 minute in order to make sure the viability of the cells.

Raman measurements. In situ Raman spectrum was obtained using a Thermo ScientificTM DXRTMxi spectrometer. The excitation wavelength was 532 nm and using the $100 \times$ objective to record. Silicon wafer was used to automatically calibrate the instrument wavelength and the samples were placed on a quartz plate.

Characterization of the purified QDs. The synthesized bio-QDs were collected and resuspended in 10 mM Tris-Cl (pH = 7.6). To isolate the QDs, the resuspended cells were disrupted by high pressure cell disruption device. The crushed cells were then sonication (2 s with 5 s intervals) for 20 times in ice bath. The suspension was centrifuged at 4 000 g for 10 min to collect the fluorescent supernatant. The resulting supernatant was concentrated and washed using a 50 kDa tubular ultrafiltration membrane (MWCO-10000, Merck Millipore Co., USA). To digest the protein impurity, $100 \,\mu$ g/mL proteinase K was added and treated at 37 °C for 1 h. Then resulting solution was purified by centrifugation (15000 × g, 10 min) and washing by 50 kDa tubular ultrafiltration membrane and dialysis. Finally, the obtained purified QDs were subjected to high-resolution transmission electron microscopy analysis (TEM-JEM-2010F, JEOL, Tokyo, Japan).

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Author Contributions

L.J.T. and H.Q.Y. conceived the idea. L.J.T., N.Q.Z. and T.T.Z. performed the synthetic experiments. L.L.L. and X.Z. analyzed the TEM data. L.J.T., H.Q.Y., X.W.L. and W.W.L. co-wrote the manuscript. All authors discussed the results and commented on the manuscript.

Additional Information

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