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Structural independence of hydrogen-bond symmetrisation dynamics at extreme pressure conditions

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The experimental study of hydrogen-bonds and their symmetrization under extreme conditions is predominantly driven by diffraction methods, despite challenges of localising or probing the hydrogen subsystems directly. Until recently, H-bond symmetrization has been addressed in terms of either nuclear quantum effects, spin crossovers or direct structural transitions; often leading to contradictory interpretations when combined. Here, we present high-resolution in-situ ¹H-NMR experiments in diamond anvil cells investigating a range of systems containing linear O-H \cdots O units at pressure ranges of up to 90 GPa covering their respective H-bond symmetrization. We found pronounced minima in the pressure dependence of the NMR resonance line-widths associated with a maximum in hydrogen mobility, precursor to a localisation of hydrogen atoms. These minima, independent of the chemical environment of the O-H \cdots O unit, can be found in a narrow range of oxygen oxygen distances between 2.44 and 2.45 Å, leading to an average critical oxygen-oxygen distance of $\vec{r}_{OO}^{crit} = 2.443(1)$ Å.

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Inderstanding the stability and properties of hydrous minerals, possibly contributing to hydrogen transport to the lower mantle is crucial as key properties of the constituents of Earth's mantle, e.g., melting temperatures, rheology, electrical conductivity and atomic diffusivity^{1–5} can be strongly affected by the presence of even small amounts of hydrogen. In particular, the high-pressure (*P*) phases of H₂O ice, (Al,Fe)OOH and dense hydrous magnesium silicates are important candidates for hosting large amounts of hydrogen⁶. These phases have a common O-H-O sequence, with an asymmetric hydrogen-bond (O-H···O) at low compression, which symmetrizes under increasing *P* (O-H-O).

During the symmetrization, compressional (e.g. bulk modulus) as well as transport properties undergo significant modifications, but although the common O-H…O sequence endorses the conclusion that the phases should show very similar behaviour, significantly distinct characteristics are observed, e.g. (i) the symmetrization *P* has a great variation, e.g. $\simeq 120$ GPa in ice-VII⁷⁻⁹ compared to $\simeq 15$ GPa in δ -AlOOH¹⁰⁻¹², (ii) proton tunnelling dominates the transition in ice-VII^{7,8,13-15}, but is absent in δ -AlOOH¹⁰. (iii) The bulk modulus in ice-VII^{8,16,17} shows a noticeable softening, while only a minor effect is found in δ -AlOOH¹¹.

In order to shed light on the underlying mechanisms it is essential to gain a deeper understanding of the similarities and disparities between different oxide-hydroxide phases with respect to the hydrogen-bond symmetrization. Hydrogen atoms, however, have a very low X-ray cross-section and neutron techniques are typically unavailable at $P \gtrsim 25$ GPa, therefore knowledge about the hydrogen subsystem is widely limited to computational and optical studies (e.g. Raman or Infrared). Advances in high-*P* nuclear magnetic resonance spectroscopy in recent years^{18–20}, allow for a direct investigation of the hydrogen subsystem (Fig. 1) beyond the megabar-range and enables previously unattainable experimental insight.





Here, we present in-situ high- P^{1} H-NMR data on four hydrous mineral phases: (i) dense magnesium silicate phase-D with composition (Mg_{0.88},Fe_{0.12})(Si_{0.9}, Al_{0.1})₂O₆H₂, (ii) pure and (iii) iron-bearing aluminum oxide-hydroxide (δ -(Al_{0.3},Fe_{0.7})OOH) as well as (iv) high-P phases of H₂O (ice-VII/X). Under compression, we find a distinct maximum in hydrogen mobility in all sampled systems at the same critical O···H-O in-bond oxygen-oxygen distance (r_{OO}^{crit}). Being a precursor for H-bond symmetrization and under further compression localisation, this minimum was found to solely depend on the short-range local structure, i.e. r_{OO} is independent of the nature of the further chemical surrounding, marking r_{OO}^{crit} as an unifying parameter of pressure-induced hydrogen-bond symmetrization dynamics.

Results and discussion

We use the equation of state (EOS) parameters (Fig. 2a) for δ -AlOOH ($V_0 = 56$ Å³, $K_0 = 183$ GPa,K' = 3.7) from²¹, δ -(Al_{0.3},Fe_{0.7})OOH ($V_0 = 63.71$ Å³, $K_0 = 164.7$ GPa, K' = 4.04) and Phase-D ($V_0 = 84.73$ Å³, $K_0 = 162$ GPa, K' = 4), are determined from synchrotron X-ray diffraction (XRD, see Methods section). For H₂O ice-VII/X we use the global EOS by²², based on DFT calculations, which was found to be in excellent agreement with recent dynamical diamond anvil cell XRD experiments⁸. The volume collapse in iron bearing oxide-hydroxide at $P \approx 45$ GPa originates from the S = 5/2 high spin to S = 1/2 low spin transition of ferric iron^{23,24}. Analogous transitions in dense magnesium silicate phase-D are not observable in our XRD data due to the low concentration of iron, but distinct transitions can be found in synchrotron Mössbauer spectroscopy (SMS) (see Methods section).

In order to quantify the evolution of hydrogen NMR signals with respect to the local symmetry under compression, we calculate oxygen-oxygen distances in the H-bonds (r_{OO}), using both diffraction data (for δ -(Al,Fe)OOH²¹ and phase-D) as well as DFT calculations for high-P ice-VII/X²². As shown recently in ref. ²⁰, the hydrogen-bond manifold in δ -(Al_{0.3},Fe_{0.7})OOH can be resolved and the observed signals assigned to different local environments, Q^i (i = 1-6), where the fraction *i* of 6 possible next-nearest-neighbour positions around the hydrogen-bond (Fig. 1) is occupied by ferric iron paramagnetic centres. We deduce the P dependence of r_{OO} in δ -(Al_{0.3},Fe_{0.7})OOH by interpolating between the endmember δ -AlOOH^{10,21} and ϵ -FeOOH²⁵ with respect to the local iron content around the hydrogen nearest-neighbours (Fig. 1). We neglect the volume collapse due to the spin transition in iron bearing phases as is at significantly higher P than the hydrogen-bond symmetrization.

We performed high-resolution ¹H-Lee Goldburg (LG) NMR measurements²⁶ for δ -(Al_{0.3},Fe_{0.7})OOH and Mg_{0.88},Fe_{0.12})(Si_{0.9}, Al_{0.1})₂O₆H₂ phase-D in order to investigate the influence of different next-nearest-neighbour configurations (Figure 1) and recorded ¹H-NMR solid echos for pure δ -AlOOH as well as H₂O ice-VII/X.

The electronic environment of the hydrogen nuclei should, to first approximation, be determined by their nearest-neighbours, i.e. the oxygen atoms involved in the hydrogen-bonds²⁷. Presence of paramagnetic centres among the next-nearest-neighbours, such as in phase-D or δ -(Al_{0.3},Fe_{0.7})OOH, induces additional contributions to NMR resonance shifts and relaxation times²⁸. These should not significantly influence dynamical effects, but instead lead to a modulation of the energy landscape of H-bonds in these systems²⁹ and therefore a separation of the signals in the indirect LG-projection frequency domain F_2 is observable.

Figure 3 a shows a high-resolution ¹H-LG NMR spectrum of δ -(Al_{0.3},Fe_{0.7})OOH at 15.7 GPa. As recently shown²⁰, the 2D-LG spectrum can be separated in six different regions. Under the



Fig. 2 Structural data of investigated systems. a Equation of state (EOS) data of the different hydrogen-bonded phases. We use the EOS for H₂O ice-VII/ X (gray graphs) of from²², based on ab-initio computations and the XRD based EOS of²¹ for δ -AlOOH (green graphs). The other curves are based on third order Birch-Murnagham EOS fits to our diffraction data for iron-bearing δ -AlOOH (brown graphs) and phase-D (blue graphs). δ -(Al_{0.3}, Fe_{0.7})OOH exhibits a 7% volume collapse associated to the electron spin crossover of ferric iron (c.f. Fig. 4a). **b** Oxygen-oxygen distance (r_{OO}) as a function of *P*. Different pressure dependencies of the Q^i (i = 3, 5, 6) (red, brown and yellow graphs respectively) H-bond environments in δ -(Al,Fe)OOH²⁰ were estimated through interpolation of r_{OO} between endmembers δ -AlOOH^{10, 21} and ϵ -FeOOH²⁵ with respect to the local iron content. We only consider the high spin state in iron bearing phases, as the spin transition occurs at significantly higher *P* (-45 GPa) than the hydrogen-bond symmetrization (-15 GPa).



Fig. 3 ¹**H-NMR spectra of all sample systems. a** High-resolution 2D-Lee Goldburg (LG) spectrum of δ -(Al_{0.3}, Fe_{0.7})OOH at 15.7 GPa showing up to six distinguishable hydrogen signals in the indirect LG projection F_2^{20} . Assuming a stochastic distribution of ferric iron in the sample, a comparison with signal intensities allows for the assignment of the H-bond species Q^i (i = 1-6). Spectra at different pressures were recorded at reduced resolution due to experimental time restrictions and solely signals from Q^3 , Q^5 and Q^6 species were recorded over the full pressure range. The integration limits for 2D signal intensity determination were defined by 35 kHz in the direct time domain F_1 and by the FWHM of each signal in the indirect LG-projection dimension F_2 . The colored regions are guide to the eyes and do not correspond to the integration limits. **b** High-resolution 2D-LG spectra of dense magnesium silicate phase-D, showing a shift towards lower chemical shifts at the HS \rightarrow LS crossover (Fig. 4a). Measurements were conducted at a magnetic field of 7.04 T. ¹H-NMR solid-echos of **c** δ -AlOOH and **d** H₂O ice-VII/X as a function of pressure. Measurements were conducted at a magnetic field of 1.2 T. Figures **a** and **b** are reproduced from ref. ²⁰; licensed under a Creative Commons Attribution (CC BY) license.

assumption of a stochastic distribution of ferric iron, several different H-bond environments should be present in the sample due to a modulation of the local magnetic fields at the hydrogen atoms caused by the paramagnetic centers. We associated these different regions to the different next-nearest neighbour surroundings (Q^i) , by matching the relative size of the regions, i.e. the 2D-signal intensities, with the probability distribution of finding a local surrounding with i of 6 possible positions being occupied by iron atoms. For further details of the signal assignment in δ -(Al_{0.3},Fe_{0.7})OOH (c.f.²⁰). Signals associated to fully Fe^{3+} depleted H-bond environments (Q⁰) lie below the detection limit of our experiments. Interestingly, the signal associated to four ferric iron paramagnetic centres occupying next-nearest neighbour positions (Q^4) was found to be much broader than to the other H-bond environments, possibly due to significant gradients in the local magnetic field stemming from an asymmetric distributions of paramagnetic centres, further hyperfine interactions or signal overlap²⁸. The spectrum shown in Fig. 3a was recorded by oversampling in the indirect Lee-Goldburg dimension F_2 , other spectra were recorded using faster F_2 sampling and therefore reduced resolution due to experimental time restrictions and solely intense signals Q^3 , Q^5 and Q^6 were recorded over the full pressure (see Methods section).

At ambient conditions, the resonance frequencies of these signals were found to be $\omega - \omega_0 = 80,365$ and 506 ppm for Q^3 , Q^5 and Q^6 , respectively (Fig. 4a). The origin of such a pronounced proton resonance shift can reasonably be expected to be

hyperfine interaction of the hydrogen nuclei with localised electron moments of Fe³⁺ ions, which should be modulated at the electron spin crossover around $V/V_0 \approx 0.85$ to 0.80. Indeed, we find a significant reduction in resonance frequency of all three signals at $V/V_0 \leq 0.78$, in excellent agreement with Mössbauer spectroscopy (see Methods section), evidencing the paramagnetic origin of these signals. Additionally, we observe a discontinuity in resonance shift at a relative compression of $V/V_0 = 0.94$ (Fig. 4a) in all Q^i and therefore independent of the local iron content, which we associate with the sub- to super-group phase transition from $P2_1nm$ to Pnnm at $P \simeq 10$ GPa^{10,21,30}.

¹H-NMR signals of dense magnesium silicate phase-D (Fig. 3b) show a narrow resonance in the indirect LG-projection frequency domain F_2 , indicating a single, well defined coordination of hydrogen atoms. The observed frequency shift of the signals (Fig. 4a) towards lower ppm values coincides with the electron spin crossover of ferric iron Fe³⁺ in δ -(Al_{0.3},Fe_{0.7})OOH. Therefore, we assume the observed signals to stem from hydrogen atoms predominantly influenced by the presence of Fe³⁺ ferric iron in phase-D. Signals shifted by the presence of Fe²⁺ ferrous iron (see Methods section) were not observed in the chosen spectral range.

One-dimensional solid echo NMR spectra of pure δ -AlOOH (Fig. 3c) at 1.2 T show a single signal at $\omega - \omega_0 \approx 0$ ppm close to the anticipated position of ferric iron depleted Q^i species in the iron bearing oxide-hydroxide analogue, indicating a well defined hydrogen position in agreement with earlier NMR studies under



Fig. 4 Resonance shift and FWHM line widths. a Top panel: Resonance shift found in ¹H-NMR solid echos of H₂O ice VII/X (gray). The signal shift of the resonances follows the dependence previously reported at high magnetic fields¹³: after an initial decrease up to 20 GPa (V/V₀ \approx 0.68), the signals are shifted downfield corresponding to the transition from a high to low barrier hydroge- bond regime. The second jump at 73 GPa (V/V₀ \approx 0.52) signifies the onset of the ice-VII \rightarrow X crossover. Lower panel: Resonance shift of δ -AlOOH (green), δ -(Al,Fe)OOH (red, brown and yellow corresponding to Q^{*i*} (i = 1...3) respectively) and phase-D (blue) as a function of the relative volume. Both δ -AlOOH and δ – (Al_{0.3},Fe_{0.7})OOH undergo a sub to supergroup phase transition from P₁nm to Pnnm which was observed at V/V₀ \approx 0.94 by a sudden increase of the shift by about 5-10 ppm. Between V/V₀ \approx 0.83 – 0.78, both iron bearing δ -(Al_{0.3},Fe_{0.7})OOH as well as phase-D undergo an electron spin crossover, resulting in a partial reduction of the paramagnetic shift interaction as well as a volume collapse observed in Q³, Q⁵ and Q⁶. The red arrows mark the points of maximum hydrogen mobility. Error bars, center of gravity of resonance signals, are within symbol size. **b** Normalized FWHM line widths (same color code as in **a**), i.e. along F1 for the 2D-LG NMR spectra and in the direct frequency domain for the one dimensional spectra, as a function of r_{0.0}. All investigated systems go through a maximum of proton mobility (shaded area in the graph), i.e. motional narrowing of resonance lines and a consequent minimum of FWHM line widths at critical oxygen-oxygen distances between $r_{0.0}^{crit} = 2.44$ and 2.45, with an average value of $\overline{r_{0.0}^{crit}} = 2.443(1)$ Å. Error bars stem from fitting of FWHM linewidths using Voigtian lineshapes.

ambient conditions³¹. Similar to δ -(Al_{0.3},Fe_{0.7})OOH, we associate the discontinuity at $V/V_0 = 0.94$ (Fig. 4a) to the aforementioned sub- to super-group phase transition.

Earlier high-resolution ¹H-LG NMR experiments on δ -AlOOH at similar magnetic fields and pressures¹⁰ indicated that the observed signals are comprised of only a single spectral contribution influenced by a minor chemical shift anisotropy. Influences of smaller pressure gradients might influence the shape of the presented spectra, but are expected to be marginal due to the Argon pressure transmitting medium used to ensure quasihydrostatic conditions. Figure 3d shows representative ¹H-NMR spectra of H₂O ice-VII/X recorded at 1.2 T. We observe two distinct transitions in the P dependence of the resonance frequencies (Fig. 4a): (i) At $V/V_0 = 0.66$ (20 GPa), after a continuous shift of about 40 ppm up-field, the proton signals showed a sudden discontinuity of ≈55 ppm, indicating the transition from the high to low barrier hydrogen-bond regime in ice-VII, where the number of tunnelling protons is increasing^{7,8} and (ii) at $V/V_0 = 0.51$ (73 GPa) a discontinuity of ≈ 20 ppm associated with the beginning of the continuous transition from ice-VII to ice-X. This behaviour is in excellent agreement with our previous study¹³.

In order to find a common parameter indicating the onset of the hydrogen-bond symmetrization, we calculate the full width at half maximum (FWHM) line-widths as a function of P as well as relative compression (V/V_0) . We analyse the change in line-width as an indicator for changes in proton mobility. The small sample sizes (10 pl) are required for our high-pressure experiments, very long data acquisition times are necessary which prohibits an analysis of other NMR parameters, such as spin-spin or spinlattice relaxation times which would provide an even more detailed picture of hydrogen dynamics. We find a minimum in the ¹H-NMR FWHM line-widths in all signals. These minima occur within a wide compression range, i.e. $V/V_0 \approx 0.92$ to 0.55 (red arrows in Fig. 4a), corresponding to P between 11 to 50 GPa. We furthermore calculate the FWHM as a function of the respective oxygen-oxygen distances (r_{OO}) for all four compounds, resolving different local environments (Qⁱ) in $\delta - (Al_{0.3}, Fe_{0.7})$ OOH (Fig. 4b). Surprisingly, all signals were found to have a continuous transition through this pronounced minimum at almost identical O-O distances $r_{OO}^{crit} = 2.44$ and 2.45, with an mean value of $\bar{r}_{OO}^{crit} = 2.443(1)$ Å. Several possible error sources might lead to a minor modulation of this value, such as insufficient diffraction data above 40 GPa in the electronic low spin state or the interpolation of the r_{OO} evolution of the Q^i species in δ -(Al_{0.3},Fe_{0.7})OOH. Nevertheless, estimating errors based on these effects was found to not significantly alter the position of the observed minima but only the respective width and thus the spread of r_{OO}^{crit} (c.f. inset in Fig. 4b). In absence of magic angle spinning³² or Lee-Goldburg derived experiments²⁰, line narrowing in NMR mainly originates from two distinct mechanisms: (i) local structural symmetrization, leading to sharper resonance line distributions (structural narrowing) and (ii) local diffusive motion of the probed nuclei leading to an averaging of short range electromagnetic interactions (dynamic narrowing)²⁷.

While the first effect is particularly pronounced in quadrupolar NMR as the electric nuclear quadrupole moment of any I > 1/2 nucleus is a sensitive probe of the local electronic environment and charge distribution; such an effect might not be very pronounced for I = 1/2 (e.g. ¹H) nuclei. Contrarily, moderate dynamic averaging of hydrogen NMR resonances is a well established diagnostic tool to identify locally restricted low amplitude motions and has even been employed to determine hydrogen diffusivities in diamond anvil cell based research³³. This minimum indicates the point of maximum hydrogen diffusivity within the H-bond, leading to a striking conclusion. At low

pressures and long $r_{OO} \approx 2.60 - 2.50$ Å, hydrogen mobility continuously increases. At r_{OO}^{crit} , hydrogen mobility reaches a maximum, i.e. minimum in the FWHM line width, as the majority of protons de-localise.

Even more interesting is the fact that the point of maximum hydrogen de-localisation in the oxide-hydroxides and phase-D, possessing a single-well H-bond energy potential is nearly identical to the maximum found in H₂O ice-VII/X characterised by a pronounced double-well character and correlated proton tunnelling^{7,34}. This indicates that the observed effect is in a region of $r_{\rm CO}^{\rm crit}$ where the barrier of the double-well potential is already negligible compared to the protons energy and therefore that the high-*P* regime is independent of the low-*P* symmetry of the H-bond potential and associated tunnelling.

Being a primer for hydrogen-bond symmetrization or, more accurately, for hydrogen localisation as the diffusive motion of protons diminishes at $r_{\rm OO} < r_{\rm Corit}^{\rm crit}$, this hydrogen mobility maximum is spread over a surprisingly wide range of over 40 GPa, ranging from 11 GPa in δ -AlOOH to about 50 GPa in H₂O ice-VII/X. Since the FWHM linewidth dependencies are comparable in all observed samples, despite significantly different charge distributions of the respective environment (ratio of Al and Fe in Q^i , O, etc.) as well as 3D (e.g. H₂O ice-VII/X) and 2D (e.g. δ -AlOOH) hydrogen-bond networks, it is reasonable to assume that this effect is universal for all linear O-H \cdots O hydrogen-bonded materials regardless of their long range atomic structure.

Furthermore, it has been argued that H-bond symmetrization or hydrogen localisation might be a primer to electron spin crossovers in iron bearing hydrous minerals like oxidehydroxides^{25,35}. Contrary, our NMR data clearly demonstrates that electron spin transitions and pressure induced hydrogen localisation are independent physical phenomena.

In this study, we have shown that the onset of hydrogen-bond symmetrization and therefore proton localisation as a consequence of the modulation of the H-bond energy potential is a phenomenon inherent to a wide variety of materials ranging from high-*P* H₂O ices to (iron-bearing) oxide-hydroxides. The observed samples were found to follow identical scaling behaviour in the O-H···O oxygen-oxygen distance r_{OO} , solely dependent on the short range atomic and electronic structure. We identified an average critical oxygen-oxygen distance of $\bar{r}_{OO}^{crit} = 2.443(1)$ Å, where the delocalisation of hydrogen atoms in H-bonds shows a pronounced maximum.

Methods

Sample preparation and characterisation

H₂O *ice-VII/X* & δ -AlOOH. The sample preparation and measurement of the H₂O ice-VII/X and δ -AlOOH data is in detail described in refs. ³⁶ and^{10,21}.

δ-(*Al*_{0.3},*Fe*_{0.7})*OOH*. A mixture of regeant grade FeOOH and Al(OH)₃ with a molar ratio of 7:3 was used as starting material. Single crystals of *ε*-(Fe,Al)OOH were synthesized using a 1000-ton Kawai-type multi-anvil high-pressure apparatus at Bayerisches Geoinstitut, University of Bayreuth, Germany. Tungsten carbide (WC) anvils with 4-mm truncated edge lengths were used to compress the sample in combination with a 5 wt% Cr₂O₃-doped MgO octahedral pressure medium with a 10 mm edge length. A cylindrical LaCrO₃ heater was set at the center of the pressure medium. The starting material was packed in a platinum capsule, which was welded shut. The sample capsule was inserted in an MgO capsule and put at the central part of the heater. Sample temperature was monitored at the central part of the outer surface of the *Pt* capsule using a W – 3%Re/W – 25%Re thermocouple, neglecting pressure effects on electromotive force of the thermocouple.

The sample was compressed to a desired press load at room temperature and then heated to a target temperature of 1200 °C at a rate of 100 °C/min. After keeping this temperature for 180 min, the sample was quenched by turning off electrical power and slowly decompressing to ambient pressure for 15 h.

Recovered single crystals with dimensions up to 200 μ m were selected based on the absence of twinning and sharp optical extinction using a polarizing microscope. The crystals were mounted on a glass fibre and X-ray diffraction (XRD) data collected using a Huber Eulerian cradle single-crystal diffractometer driven by the SINGLE software³⁷. The diffractometer was equipped with a Mo K α X-ray source and operated at 50 kV and 40 mA. Crystals with a half-width of the diffraction peaks less than 0.1° were further analysed in terms of their cell parameters with the vector least-squares method³⁸. The effect of crystal offsets and diffractometer aberrations for each crystal was eliminated using the eight-position centring method³⁹. The crystal had the space group of *P*2₁*nm* and lattice parameters of a = 2.9573(1) Å, b = 4.3884(1) Å, c = 4.8873(2), V = 63.426(4) Å³.

Chemical compositions of the samples were measured using an electron microprobe analyzer (EPMA) with wavelength-dispersive spectrometers (JEOL, JXA-8200) operated at 15 kV and 10 nA, for 20 s on the peaks of Al and Fe and 10 sec on the background. Synthetic hematite and corundum were used as standard material for Fe and Al, respectively. The composition of the dark-red crystals was expressed as (Al_{0.305(9)}, Fe_{0.695(9)})OOH, assuming that the cation number of hydrogen is one.

 $(Mg_{0.88}$ - $Fe_{0.12})(Si_{0.9}$ - $Al_{0.1})_2O_6H_2$ phase-D. A starting material was prepared as a powdered mixture of regent grade chemicals of Mg(OH)2 (45.11 wt.%), SiO₂ (38.02 wt%), Al(OH)3 (5.48 wt.%) and ⁵7Fe enriched Fe₂O₃ (11.38 wt.%). Single crystals of hydrous phase-D were synthesized using a 1200-ton Kawai-type multi-anvil high-pressure apparatus at Bayerisches Geoinstitut, University of Bayreuth, Germany . Tungsten carbide anvils with 3 mm truncated edge lengths were used in combination with a 5 wt% Cr₂O₃-doped MgO octahedral pressure medium with a 7-mm edge length. A cylindrical LaCrO₃ heater was located at the center of the pressure medium. The starting material was packed in a welded platinum capsule. The sample capsule was inserted in an MgO capsule and put at the central part of the heater. Sample temperature was monitored at the central part of the outer surface of the Pt capsule using a W-3%Re/W-25%Re thermocouple. Pressure effect on electromotive force of the thermocouple was ignored.

The sample was compressed to a desired press load at room temperature and then heated to a target temperature of 1100 °C at a rate of 100 °C/min. After keeping this temperature for 240 min, the sample was quenched by turning electrical power off and slowly decompressed to ambient pressure for 15 h. Chemical composition of the recovered hydrous phase-D was determined using an electron microprobe analyzer (EPMA) with wavelength-dispersive spectrometers (JEOL, JXA-8200) operated at 15 kV and 10 nA, for 20 s on the peaks of Al and Fe and 10 s on the background. Natural enstatite for Mg and Si, synthetic hematite for Fe and corundum for Al were used as standard material.

Mössbauer spectroscopy. Mössbauer absorption spectra were collected at ambient temperature in a diamond anvil cell at the Nuclear Resonance beamline (ID18) at the European Synchrotron Radiation Facility (Grenoble) using Synchrotron Moessbauer (SMS) Source spectroscopy⁴⁰. The experiment was conducted in transmission geometry and folded spectra contained 512 channels. The line width of the SMS was determined before and after collection of each pressure point by measuring the reference single line absorber (K₂Mg₅₇Fe(CN)₆). The Mössbauer spectra were fitted using MossA software⁴¹ with the full transmission integral assuming a Lorentzian-squared line shape of the SMS. Isomer shift values are referred to that of α -Fe at 300 K.

⁵⁷Fe Mössbauer spectra of phase-D measured at ambient pressure and temperature show broadened asymmetric paramagnetic doublets, indicating a superposition of several subspectra. The spectra were interpreted as a mixture of two quadrupole doublets. The value of the isomer shift $\delta = 0.36(1)$ mm/s of one subspectrum corresponds to high-spin (HS) ions Fe³⁺(d5, S = 5/2) located in oxygen octahedra. Sub-spectra showing high values of isomer shifts $\delta = 1.22(2)$ mm/s are characteristics of high-spin ferrous iron Fe²⁺(d6, S = 2)⁴². The high quadrupole splittings Δ of the doublets 0.81(5) mm/s and 2.19(3) mm/s showed that iron ions are located in crystal positions with a strong electric-field gradient, characteristic to a Al/Mg substitution in the phase-D crystal lattice.

Upon compression, SMS spectra change drastically: above 25 GPa, spectra can no longer be described by only two doublets. Additional sub-spectra correspond to low-spin ferrous iron, indicating the onset of a HS \rightarrow LS spin crossover. The ferric to ferrous iron ratio was constant throughout the whole compression range.

Above 36 GPa, SMS signatures of HS ferric iron are lost and new sub-spectra with characteristic hyperfine parameters of LS Fe³⁺ appear indicating another long range spin crossover. Isomer shifts LS Fe³⁺ are lower than in the high-spin state and quadrupole splitting was found amplified due enhanced electric field gradients, associated with the non-uniform distribution of uncompensated $3d - t_{2g}$ valence electrons. The pressure dependence of hyperfine parameters show no anomalous changes in the local environment of the iron ions.

NMR-DAC preparation. DACs for high-pressure NMR experiments were prepared following a procedure described in e.g^{43,44}. First, rhenium gaskets were indented to the desired thickness, which depends on the size of the diamond anvil culets employed, but usually $\lesssim 25 \mu m$. Sample cavities were drilled using specialized laser drilling equipment. After gasket preparation, the diamond anvils were covered with a layer of 1 μm of copper or gold using chemical vapour deposition. To ensure the electrical insulation of the conductive layers from the rhenium gasket, the latter were coated by a thin layer (~500 nm) of Al_2O_3 using physical vapour deposition. The Lenz lens resonators were shaped from the conductive layer on the diamonds by using focused ion beam milling.

Before the final cell assembly, radio frequency resonators were prepared accordingly to their desired operation frequency. A pair of high inductance solenoid coils (\approx 100 nH) for low frequency experiments of below 100 MHz or a pair of single turn printed circuit board (PCB) plated copper resonators for ¹H-NMR frequencies at high fields were used as driving coil arrangements for the Lenz lens resonators' structure and were placed around each diamond anvil. After sample loading and initial pressurisation, the driving coils were connected to form a Helmholtz coil-like arrangement.

Pressure calibration was performed using the shift of the first derivative of the first order Raman signal of the diamond edge in the center of the culet^{45,46}. All DACs were fixed and connected to home built NMR probes equipped with customized cylindrical trimmer capacitors (dynamic range of ≈ 150 pF) for frequency tuning to the desired resonance frequencies and impedance matching to the spectrometer electronics (50 Ω).

Proton shift referencing was conducted using the 63 Cu resonances of the Lenz lenses themselves as internal references taking into account the additional shielding of B_0 inherent to every DAC. These resonances were cross referenced with standard metallic copper samples at ambient conditions without a DAC. The resulting shift between both 63 Cu-NMR signals are then used as a primer for the NMR signals of the samples under investigation.

Lee-Goldburg decoupling experiments were initially calibrated by quick twodimensional nutations (~56 2D spectra) for different off resonant frequencies of the LG-pulse. Optimal pulse offset frequencies $\omega_{\rm off}$ were found between 25 to 35 kHz. One dimensional LG-spectra were recorded by oversampling in the indirect time domain using previously determined values of $\omega_{\rm off}$ at a pulse power of about 100 mW attenuation relative to the excitation pulse. Two dimensional LG spectra were recorded with identical direct and indirect time domains (usually 2048 points in each dimension), while matching the incrementation of the LG-pulse to the direct time domain dwell time of the spectrometer.

X-ray diffraction. Single crystal X-ray diffraction data for phase-D and δ -(Al_{0.3},Fe_{0.7}) OOH were collected at the beamlines P02.2 (PETRA III, DESY, Hamburg, Germany) and ID15 (ESRF, Grenoble, France). At PETRA III the data were collected using Perkin Elmer XRD1621 flat panel detector and X-rays with the wavelength $\lambda = 0.2887$ Å. For the different experiments at ID15 the beamline was equipped with a MAR55 flat panel or a EIGER2 X 9M CdTe (340 × 370 mm) detector. Before collecting XRD data, the experimental geometry was precisely defined based on freshly collected diffraction data from enstatite (at PETRA III) or a vanadinite callibrant (ESRF). The X-ray energy at ID15 was 30 keV and the wavelength is $\lambda = 0.4133$ Å, while the beamsize was ~10 μ m in FWHM. In all our experiments, at each pressure point data collection was performed in an omega range of least ± 32° with 0.5° step per frame and an exposure time of 1 to 3 s per step. Integration of the reflection intensities and absorption corrections were performed using the CrysAlisPro software. Structure solution and refinement were done in the isotropic approximation using Jana2006⁴⁷ with Superflip⁴⁸ and SHELXT⁴⁹.

Data availability

The data supporting the findings of this study are publicly available from the corresponding author upon request.

Code availability

Python code used for analysing NMR data is available from the corresponding author upon request.

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Author contributions

T.M. designed the experiments, prepared the DACs, conducted NMR experiments and analysed data. F.T. conducted DFT calculations and data analysis. S.K., D.L. and A.A. characterised samples using diffraction and moessbauer spectroscopy. T.I. synthesised samples. T.M., F.T., N.D. and L.D. wrote the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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