

ARTICLE

https://doi.org/10.1038/s41467-021-27564-2

Check for updates

Homo-composition and hetero-structure nanocomposite Pnma Bi₂SeS₂ - Pnnm Bi₂SeS₂ with high thermoelectric performance

OPEN

Bushra Jabar^{1,5}, Fu Li[®]^{1,5™}, Zhuanghao Zheng^{1,5}, Adil Mansoor², Yongbin Zhu³, Chongbin Liang¹, Dongwei Ao¹, Yuexing Chen¹, Guangxing Liang¹, Ping Fan[®]^{1™} & Weishu Liu[®]^{3,4™}

Nanocomposite engineering decouples the transport of phonons and electrons. This usually involves the in-situ formation or ex-situ addition of nanoparticles to a material matrix with hetero-composition and hetero-structure (*heC-heS*) interfaces or hetero-composition and homo-structure (*heC-hoS*) interfaces. Herein, a quasi homo-composition and hetero-structure (*hoC-heS*) nanocomposite consisting of *Pnma* Bi₂SeS₂ - *Pnnm* Bi₂SeS₂ is obtained through a Br dopant-induced phase transition, providing a coherent interface between the *Pnma* matrix and *Pnnm* second phase due to the slight structural difference between the two phases. This *hoC-heS* nanocomposite demonstrates a significant reduction in lattice thermal conductivity (~0.40 W m⁻¹ K⁻¹) and an enhanced power factor (7.39 μ W cm⁻¹ K⁻²). Consequently, a record high figure-of-merit *ZT*_{max} = 1.12 (at 773 K) and a high average figure-of-merit *ZT*_{ave} = 0.72 (in the range of 323-773 K) are achieved. This work provides a general strategy for synergistically tuning electrical and thermal transport properties by designing *hoC-heS* nanocomposites through a dopant-induced phase transition.

¹ Shenzhen Key Laboratory of Advanced Thin Films and Applications, College of Physics and Optoelectronic Engineering, Shenzhen University, 518060 Shenzhen, China. ² Faculty of Materials and Manufacturing, Beijing University of Technology, 100 Peenle Yuan, Chaoyang District, 100124 Beijing, China. ³ Department of Materials Science and Engineering, Southern University of Science and Technology, 518055 Shenzhen, China. ⁴ Guangdong Provincial Key Laboratory of Functional Oxide Materials and Devices, Southern University of Science and Technology, 518055 Shenzhen, China. ⁵These authors contributed equally: Bushra Jabar, Fu Li, Zhuanghao Zheng. ^{Ke}email: lifu@szu.edu.cn; fanping@szu.edu.cn; liuws@sustech.edu.cn

hermoelectric (TE) technology, which directly converts heat into electricity, is a potential solution for securing an affordable green energy source by harvesting large-scale mid-grade waste heat (at/near mid-range temperatures)^{1,2}. The dimensionless figure-of-merit $(ZT = \sigma S^2 T/\kappa)$ and power factor $(PF = \sigma S^2)$ are the material-level performance scales for energy conversion efficiency and output power factor, where S, σ , κ , and T are the Seebeck coefficient, electrical conductivity, thermal conductivity, and temperature, respectively^{3,4}. Generally, ZT can be improved by lowering κ or enhancing *PF*. However, simultaneously improving PF while reducing the value of κ is a significant challenge. Nanocomposite engineering, a vital strategy for decoupling the transport of phonons and electrons, involves the in-situ formation and ex-situ addition of nanoparticles to a material⁵⁻¹⁴. The composition difference between the matrix and second phases can induce mass fluctuation, strains¹⁵⁻¹⁷, or dislocation^{18,19} at their interfaces, which can dramatically scatter phonons and reduce lattice thermal conductivity. A nanocomposite with coherent nanoinclusions (e.g., hetero-composition and homo-structure (*he*C-*ho*S)^{13,18,20,21}, PbTe-AgSbTe₂²²) suffers less charge carrier mobility reduction than one with incoherent nanoinclusions (e.g., hetero-composition and hetero-structure (heC-heS)^{7,10,23}, PbTe-Ag₂Te²⁴). Zhao et al. suggested that the band alignment between the matrix and second phases could play a critical role in minimizing electron scattering²⁵.

Furthermore, extremely low thermal conductivities have been observed in various materials near their phase transition, caused by both pressure-induced and temperature-induced transitions. Near the phase transition point, a material can be considered as a metastable nanocomposite with a homo-composition and a heterostructure. So far, some research has been carried out to investigate the structural changes with the symmetry reduction of specific phase change materials and their effect on TE properties under the influence of pressure or temperature^{26–31}. However, the structural phase transitions and underlying driving force (temperature or pressure) of such materials have not yet been systematically investigated due to their undefined intermediate structures and internal atomic distortions. It is also very difficult to obtain precise regulation in pressure-induced transition, while temperature-induced transition may only be effective for the structural transition of specific materials in a narrow temperature range^{26,31–34}.

Herein, we propose a quasi homo-composition and heterostructure (hoC-heS) nanocomposite composed of Pnma Bi2SeS2 -Pnnm Bi₂SeS₂, which provides a coherent interface between the matrix and second phase due to the slight structural difference between the two phases. Br element is familiarly donor in the Bi₂S₂, Bi₂Se₃, and Bi₂SeS₂^{11,23}. However, it was found that the Br dopant caused the phase transition from Pnma Bi₂SeS₂ to Pnnm Bi₂SeS₂ in our work. And the initial orthorhombic *Pnma* phase also exhibits structural distortion. Due to the significant reduction in lattice thermal conductivity ($\kappa_{\rm I}$) and low impact to carrier mobility (μ) caused by the Br dopant, a record high figure-of-merit $ZT_{max} =$ 1.12 (at 773 K) and a record average $ZT_{ave} = 0.72$ (in the range of 323-773 K) were achieved in the Bi₂SeS₂ family. It should be noted that the ZT value of most reported sulfide compounds is lower than 1.0 due to the low carrier mobility and high lattice thermal conductivity so far, although sulfide compounds with cheaper, lowtoxicity, and high earth abundance elements have been extensively studied in recent years. This work provides a general strategy for designing hoC-heS nanocomposites through a dopant-induced phase transition to enhance TE properties.

Results

Figure 1 shows the structure of the as-fabricated Pnma Bi₂SeS₂ - Pnnm Bi₂SeS₂ quasi hoC-heS nanocomposite induced by the Br dopant. This structure is very different from those of previously



Fig. 1 Schematic representation of the nanocomposition and crystal structure of Bi₂SeS₂. a Summary of previously reported nanocomposites classified by the composition and crystal structure of their nanoinclusion and matrix. Most reported nanocomposites are hetero-composition and homo-structure (*heC-hoS*) or hetero-composition and hetero-structure (*heC-hoS*). **b** Structure and schematic illustration of transport mechanism inside *Pnma* Bi₂SeS₂ - *Pnnm* Bi₂SeS₂ *hoC-heS* nanocomposite. **c-d** Crystal structure of Bi₂SeS₂ *Pnma* phase in the *ac* plane. **e-f** Crystal structure of Bi₂SeS₂ *Pnnm* phase in the *ab* plane.



Fig. 2 Phase structure and lattice parameters. a XRD patterns of $Bi_2Se_{1-x}Br_xS_2$ (x = 0, 0.03, 0.06, 0.09, 0.12, 0.15). **b** Rietveld refinement analysis from XRD patterns of $Bi_2Se_{1-x}Br_xS_2$ (x = 0.12), where obs, calc, bkg, and diff represent the observed, calculated, background, and residual differences between the curves, respectively. **c** *Pnma* and *Pnnm* phase fractions in the sample as a function of Br content. **d** Lattice constant parameters of the *Pnma* phase as a function of Br content. The dashed lines are provided as a guide.

reported nanocomposites with a hetero-composition and homostructure (*heC-hoS*) or hetero-composition and hetero-structure (*heC-heS*) (Fig. 1(a, b)). The matrix phase *Pnma* Bi₂SeS₂ demonstrates an orthorhombic layered structure with each quintuple layer assembled by weak Van der Waals interactions, a widely observed characteristic of polycrystalline Bi₂SeS₂. Five distinct atomic sites (Bi1, Bi2, S1, S2, and Se) are presented in the unit cell due to lattice symmetry operation (S-Bi-S(Se)-Bi-S), in which all the Bi and S (or Se) atoms occupy the 4c Wyckoff positions in the *Pnma* space group (Fig. 1(c))^{9,35–37}. This layered assembly possesses a highly confined S-coordination environment around both the Bi1 and Bi2 cations with three short and four long bonds (Fig. 1(d)). However, a new Bi₂SeS₂ phase crystallized in a *Pnnm* orthorhombic structure can be found (Fig. 1(e, f)).

Figure 2 shows the XRD patterns of the *hoC-heS* nanocomposite with different $Bi_2Se_{1-x}Br_xS_2$ compositions (x = 0, 0.03, 0.06, 0.09, 0.12, and 0.15). The two phases (*Pnma* and *Pnnm*) appear when the Br content is higher than x = 0.01 (Fig. 2(a), Figs. S1 and S2: $Bi_2Se_{1-x}Br_xS_2, x = 0-0.21$). Based on the Reitveld refinement (Figs. 2(b) and S3), the XRD peaks match well with those of orthorhombic *Pnma* (space group: 62) and *Pnnm* (space group: 58). It should be noted that the *Pnma* phase is widely reported for polycrystalline Bi_2SeS_2 and Bi_2S_3 compounds, while the *Pnnm* phase is still a theoretically predicted structure^{38,39}. A noticeable and continuous variation in the *Pnma* crystal structure can be observed with increasing Br content in the normal $Bi_2Se_{1-x}Br_xS_2$ composition. Figure 2(c) shows the molar fraction evaluation of *Pnma* Bi₂SeS₂ and *Pnnm* Bi₂SeS₂. The fraction of *Pnnm* Bi₂SeS₂ increases with increasing Br content while *Pnma* Bi₂SeS₂ shows the opposite trend. For instance, Bi₂Se_{1-x}Br_xS₂ (x = 0.12) consists of 71.3% *Pnma* phase and 28.7% *Pnnm* phase, while Bi₂Se_{1-x}Br_xS₂ (x = 0.21) consists of 44.4% *Pnma* phase and 55.6% *Pnnm* phase.

In addition to the Br dopant-derived phase change, the relative intensity of the (301), (011), (202), (402), (314), and (015) peaks of the Pnma phase significantly change with increasing Br content, indicating continuous lattice distortion before the transition into the Pnnm phase. In addition, it is reported that their surface energies can quantitatively describe the stability of various surfaces⁴⁰. The calculated formation energy along the plane (011) and (301) indicates that the formation energy along the plane (011) has reduced, while it has increased along the plane (301) after Br doping. This means the plane (011) might be easy to form during the preparation due to the low energy after Br doping. Therefore, it is clear that the Bi₂SeS₂ system experiences a twofold structural evolution with increasing Br content. First, the initial orthorhombic Pnma phase exhibits some structural distortion, revealing planar transitions that preserve the unit cell and achieve complete orientation along the (011) plane when $0.09 \le x \le 0.12$. Second, doping-induced structural variations simultaneously promote inter-orthorhombic phase transformation from Pnma to Pnnm. Figure 2(d) shows the Br dopantinduced lattice parameter evolution. In the current Pnma phase, the *b*-axis does not show significant change. However, the *a*- and

c- axes approach each other and overlap without a significant expansion in the unit cell volume (Fig. S4 and Table S1) with increasing Br content from x = 0 to x = 0.015-0.03, indicating the absence of prominent structural disorder in the Pnma phase. When x is further increased from 0.03 to 0.105, the unit cell parameters show similar constant behavior while the unit cell volume significantly expands (Fig. S4). As the Br content increases from x = 0.12 to x = 0.18, the *a*- and *c*- axes move away from each other with a continuous expansion in the unit cell volume (Fig. S4). These discontinuities or deviations in the lattice parameters of the Br-doped Pnma phase Bi₂SeS₂ indicate the existence of lattice distortion. For the Pnnm phase, the lattice parameters of b and c gradually reduce when the Br content x is higher than 0.105 (Fig. S5). However, the decrease is not significant for the samples with low Br content x (x < 0.105), mainly due to the lower dopant content and the slight difference in the ionic radius between Br⁻ (1.96 Å) and Se²⁻ (1.98 Å).

Figure 3(a-d) shows high-resolution TEM images of the asfabricated *ho*C-*he*S nanocomposite $Bi_2Se_{1-x}Br_xS_2$ (x = 0.12). The existence of the Pnma and Pnnm phases can clearly be seen. The Pnnm phase has a typical irregular oval-like morphology and is coherently embedded inside the *Pnma* phase grains. The inverse fast Fourier transform (IFFT) image shown in Fig. 3(b) displays a defect-free phase boundary between these two phases. In addition, energy dispersive X-ray spectroscopy (EDS) shows that the elements, including Br, are homogeneously distributed in the Pnma and Pnnm phases, with the exception of excess Bi observed in the Pnnm phase (Fig. S6). Figure 3(e, f) shows high-angle annular dark field scanning transmission electron microscopy (HAADF-STEM) images of $Bi_2Se_{1-x}Br_xS_2$ (x = 0, 0.12) along the c-axis. Bi-centric atomic distributions are prominent, with wellarranged atoms along the *c*-axis of the undoped Bi₂SeS₂ (Fig. 3(e)). However, the Br dopant clearly induces atomic disarrangement and lattice distortion with uniform and distinctive elongation or contraction (Fig. 3(f)), with the doped materials displaying local phase structural variations without disturbing the symmetry. By comparing the peak intensity profiles of line 1 (or line 2) to line 3 (or line 4), it can be predicted that the significant contrast in strain causes atomic disarrangement within the frame structure of the orthorhombic Pnma phase due to Br doping. The lattice variations along the c-axis are related to modifications of the bond distances. The calculated bond lengths indicate that the short Bi1- S1(S) and Bi2-S2(S) bonds and the long Bi1-S1(L) and Bi2-S2(L) bonds do not significantly change after doping (Fig. S7). However, the increased Bi1-Se bond length and decreased Bi2-Se bond length (Fig. S7) suggest that Br elongates the orthorhombic layered structure (Fig. 3). This lattice evolution with disordered bond length can lead to lattice anharmonicity, which is an important phonon scattering mechanism in TE materials. Figure 4(a-d) shows that the *Pnnm* phase particles are tensile strained, with the maximum tensile strain reaching 3%. The strains in the ε_{xx} and ε_{yy} directions show continuous contrast across the grain boundary (Fig. 4(a-d)). The rotational strain profile (ε_{rot}) also shows continuous contrast across the phase boundaries (Fig. S8). This means that the strain at the interface is insignificant, suggesting that the corresponding phase boundary between the Pnma and Pnnm phases is defect-free.

Figure 5 show the electrical transport properties of asfabricated *hoC-heS* nanocomposite Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.015, 0.03, 0.06, 0.09, 0.105, 0.12, 0.15, 0.18, and 0.21). The transport properties change with increasing doping content x, and this change can be roughly divided into three parts. The undoped Bi₂SeS₂ exhibits poor electrical conductivity (σ) at room temperature due to its low carrier concentration (*n*) (Fig. 5(a)). However, the room temperature σ value significantly increases from 0.25 × 10⁴ Sm⁻¹ to 6.26 × 10⁴ Sm⁻¹ with a low amount of Br (x = 0.015), corresponding to an enhanced *n* value from 1.28×10^{19} to 46.6×10^{19} cm⁻³ (Fig. 5(b)). Thus, Br is a very effective dopant. However, the room temperature σ decreases from 5.62×10^4 Sm⁻¹ to 2.42×10^4 , 1.49×10^4 , and then 0.59×10^4 Sm⁻¹ as the Br content further increases from x = 0.03to x = 0.06, 0.09, and 0.105 (Fig. 5(e)). For these doped nanocomposite samples, the Pnnm phase appears and increases with increasing Br content but remains lower than 40%. A further increase in Br content (from x = 0.105 to 0.12 and 0.15) results in a slight increase in σ (with slightly enhanced *n* and moderate μ), but σ declines again when $x \ge 0.18$ (Fig. 5(h)). It means that when the *Pnnm* phase fraction is higher than 40% and $x \ge 0.18$, the value of σ deteriorates due to the decrease of μ (Fig. 5(a)). Thenano Pnnm phase would be a new electron scattering center. The temperature-dependent σ exhibits two trends. When $0.015 \le x \le 0.03$, the value of σ decreases monotonically with increasing temperature, acting as a degenerate semiconductor. In contrast, when $0.06 \le x \le 0.21$, the value of σ first decreases and increases with increasing temperature, acting as a thermally activated semiconductor.

Figure 5(c), (f), and (i) shows the temperature-dependent Seebeck coefficient (S) of the as-fabricated hoC-heS nanocomposite $Bi_2Se_{1-x}Br_xS_2$ (*x* = 0, 0.015, 0.03, 0.06, 0.09, 0.105, 0.12, 0.15, 0.18, and 0.21), which presents good consistency with σ . The Br-free Bi₂SeS₂ nanocomposite shows a high room temperature S value of $-300 \ \mu V \ K^{-1}$. The S values decrease at low Br doping concentrations $(0.015 \le x \le 0.09)$ due to the higher *n* values of these nanocomposites compared with the undoped nanocomposite. At high Br doping concentrations $(0.105 \le x \le 0.21)$, the magnitude of S increases to $200 \,\mu\text{V}\,\text{K}^{-1}$, and the bipolar transport shifts to lower temperatures. By using the S and n data and assuming that acoustic phonon scattering is prominent in the Bi₂SeS₂ matrix, a Pisarenko relation (at 300 K) can be obtained for Bi₂Se_{1-x}Br_xS₂ (Fig. S9). The S values of the Br-doped samples deviate from the Pisarenko relation when x > 0.06, suggesting that the electronic density of states N(E) changes due to activation of the conduction bands (CBs) via doping. In addition, the density of state (DOS) effective mass (m_d^*) was obtained using the equations (S2)-(S4) in the Supporting Information, assuming the as-fabricated nanocomposite as quasi-single uniform phase (Fig. S9). A increased m_{d}^{*} from $1.62m_{e}$ (x = 0) to $6.90m_{e}$ (at x = 0.21) with Br content of Bi₂Se_{1-x}Br_xS₂ was observed. The variations in m_{d}^{*} values with Br content results from the band structure change and the interface effect. The enhanced σ and moderate S exhibited by all the doped samples result in a remarkable enhancement in their power factors (PF) across the whole temperature range (Fig. 5(d), (g), and (j)). The largest PF of ~7.39 μ W cm⁻¹ K⁻² at 773 K is achieved by Bi₂Se_{1-x}Br_xS₂ (x = 0.12). This value is seven times higher than that of pure Bi₂SeS₂ and is also the highest value reported thus far among Bi_2SeS_2 and Bi_2S_3 -based materials reported in the literature⁸⁻¹¹.

Figure 6 illustrates the evolution of the Br-free and Br-doped Bi₂SeS₂ band structure in both the *Pnma* and *Pnnm* phases. First, Fig. 6(a–d) shows the Br-free and Br-doped Bi₂SeS₂ *Pnma* phase band structures, both of which are direct bandgap semiconductors with conduction band minima (CBM) and valence band maxima (VBM) near the G point along the high symmetric line of the Brillouin zone. However, the CBM and VBM of the Br-doped Bi₂SeS₂ are not as sharp as those of the pure structure, and significant band convergence can be observed due to doping. Moreover, the Fermi level is prominently inclined toward the CB in Br-doped *Pnma* Bi₂SeS₂, confirming the Br is an effective donor. According to the relationship between the electron and hole concentrations and bandgap⁴¹, Br dopant increases the electron concentration but decreases the hole concentration. In short, in the Br-doped Bi₂SeS₂ structure, the CB calculations



Fig. 3 Structural characterization. a TEM micrograph of $Bi_2Se_{1-x}Br_xS_2$ (x = 0.12), where the yellow dotted regions show the *Pnnm* nanophase. **b** HRTEM micrograph of the red dotted square [1] in (**a**). **c** HRTEM micrograph of the red dotted square [2] in (**a**), where the inset is the HRTEM micrograph of region [3]. **d** HRTEM micrograph of the red dotted square [4] in (**c**). **e-f** STEM-HAADF images of undoped and Br-doped Bi_2Se_2 samples showing the orthorhombic *Pnma* structure along the c-direction. (e1)-(e2) Intensity line profiles corresponding to line 1 and line 2 in (**e**). (f1)-(f2) Intensity line profiles corresponding to line 3 and line 4 in (**f**).



Fig. 4 Geometric phase analysis (GPA) of corresponding HRTEM images from Fig. 3 of $Bi_2Se_{1-x}Br_xS_2$ (x = 0.12). GPA strain maps (b) and (d) are related to (a) and (c) and the color bars show the induced strain in the *Pnnm* phase. e-f Corresponding strain maps along the ε_{xx} and ε_{yy} directions for (c).

(with impurity levels in the bandgap) indicate that more free electron carriers are generated once Br is doped into the Bi₂SeS₂ lattice, with Br acting as a donor. The introduced overcharge effect in this Br-doped system, due to defect states, indicates that modifications can be made in the structure to utilize the overcharge. In addition, the partial density of states (DOS) of Bi primarily contributes to the total DOS in the CBM for both the Pnma and Pnnm phases (Fig. 6(c) and (g)). The DOS in the CBM along the y-axis shows that the intensity of peaks in the Pnnm phase Bi₂SeS₂ is reduced, which can be mainly ascribed to lattice distortions (Fig. 6(f)). This results in a reduction in energy. Here, distortions yield an orthorhombic Pnnm Bi2SeS2 structure with a slightly reduced symmetry. Overall, these results indicate that the Br dopant acts as a donor, causing significant structural distortion and the inter-orthorhombic transformation of Pnma to Pnnm. This phase transformation optimizes the electrical transport properties of Bi₂SeS₂. Besides, the calculated formation energy also shows that the formation energy value for Br doping at Se site is much lower than that at S site (Fig. S10(a)). This means Br should prefer to occupy the Se site in Bi_2SeS_2 .

Figure 7 shows the thermal conductivity and ZT values of asfabricated hoC-heS nanocomposite of Pnma Bi2SeS2 - Pnnm Bi_2SeS_2 in a nominal composition of $Bi_2Se_{1-x}Br_xS_2$ (x = 0, 0.09, 0.12, 0.15, 0.21). Other $Bi_2Se_{1-x}Br_xS_2$ with x = 0.015, 0.03, 0.06,0.105, and 0.18 are shown in the Fig. S11. The undoped Bi₂SeS₂ shows a lower κ of 0.72 W m⁻¹ K⁻¹ (Fig. 7(a)). While the value of κ increases with a very low amount of Br (x = 0.015) due to the contribution from carriers (Fig. S11(a)), it gradually decreases to 0.60 W m⁻¹ K⁻¹ with an increase in Br content to x = 0.09 and further reduces to ~0.5 W m⁻¹ K⁻¹ when $x \ge 0.12$. Furthermore, the value of κ shows temperature-independent behavior for the heavily doped samples $(0.09 \le x \le 0.21)$, maintaining a minimum at ~725 K and then slightly shifting to higher values with a further increase in temperature due to the bipolar effect. As shown in Fig. 7(b), the value of $\kappa_{\rm L}$, which can be estimated by subtracting the electronic thermal conductivity $\kappa_{\rm e}$ from κ ($\kappa_{\rm L} = \kappa - \kappa_{\rm e}$, where $\kappa_{\rm L}$ is the lattice thermal conductivity and $\kappa_{\rm e}$ is the electronic thermal conductivity (Fig. S11(b))), is ~0.72 $W\,m^{-1}~K^{-1}$ at room temperature for pure Bi₂SeS₂. The value of κ_L declines with

increasing temperature until 550 K, then increases due to the bipolar effect. However, for a Br content of x = 0.09, the value of κ_L at room temperature is only $0.52 \text{ W m}^{-1} \text{ K}^{-1}$. This further declines to ~0.40 W m⁻¹ K⁻¹ for x = 0.12. For the Brdoped sample where $x \ge 0.15$, the value of κ_L slightly increases to $\sim 0.47 \text{ W} \text{ m}^{-1} \text{ K}^{-1}$. The contribution of the bipolar effect becomes negligible with increasing Br content (Fig. S11(d)). Because of this, the value of κ_L of the heavily doped samples is temperature-independent (Fig. 7(b)). The theoretical lattice thermal conductivity was calculated via the Debye-Callaway model by combining all the substantial factors (presented in the Supporting Information). In the Bi₂Se_{1-x}Br_xS₂ system, the dominant phonon-scattering mechanisms involve scattering processes from point defects/alloy elements, nanophase, phononphonon Umklapp scattering, boundaries between the Pnma and Pnnm phases, and electron-phonon interactions. Figure 7(b) shows that the calculated value of $\kappa_{\rm L}$ is in good agreement with the experimental results before the onset of the bipolar effect, which is the main reason for the deviation between the calculated $\kappa_{\rm L}$ and the experimental data. Parameter A (Table S3), which is the preset parameter for point defects (including nanophase boundaries) in the expression of total phonon relaxation time, shows a reduction due to doping. However, parameter A increases from 3.48 to 10.4 as x increases from 0.015 to 0.12, which indicates enhanced phonon blocking due to the presence of intensive nanophase boundaries or impurity centers. A further increase in x from 0.15 to 0.21 causes a decrease in the value of A (Table S3). This explains the increase in $\kappa_{\rm L}$ when $x \ge 0.15$. At high values of x, the nanophase boundaries between the Pnma and Pnnm phases agglomerate due to the high density of Pnnm nanoinclusions, which causes a reduction in the number of effective phonon scattering centers and increases the value of κ_L . Therefore, x = 0.12 is the optimum Br dopant level with the appropriate Pnnm phase fraction in this nanocomposite.

Due to the combination of a significantly enhanced *PF* and reduced κ , a record high ZT_{max} of 1.12 (at 773 K) and a record high average ZT_{ave} of 0.72 (in 323–773 K) are achieved for the doped Bi₂Se_{1-x}Br_xS₂ system when x = 0.12 (Fig. 7(c)). The average ZT_{ave} is calculated by integrating the area under the *ZT* curves in



Fig. 5 Electrical transport properies of the fabricated materials. a Variation of carrier concentration (*n*) and mobility (μ) (at 300 K) with Br content. The dashed lines are provided as a guide. **b-d** Temperature dependence of electrical resistivity (σ), Seebeck coefficient (*S*), and power factor (*PF*) of Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.015, 0.03). **e-g** Temperature dependence of electrical resistivity (σ), Seebeck coefficient (*S*), and power factor (*PF*) of Bi₂Se_{1-x}Br_xS₂ (x = 0.06, 0.09, 0.105, 0.12, 0.15). **h-i** Temperature dependence of electrical resistivity (σ), Seebeck coefficient (*S*), and power factor (*PF*) of Bi₂Se_{1-x}Br_xS₂ (x = 0.06, 0.09, 0.105, 0.12, 0.15). **h-i** Temperature dependence of electrical resistivity (σ), Seebeck coefficient (*S*), and power factor (*PF*) of Bi₂Se_{1-x}Br_xS₂ (x = 0.18, 0.21). Error bars were estimated from the repeatability of the experimental result; three measurements were carried out for each material.

Fig. 7(c) according to the following formula,

$$ZT_{ave} = \frac{1}{T_h - T_c} \int_{T_c}^{T_h} ZT \, dT \tag{1}$$

where $T_c = 323$ K and $T_h = 773$ K are the hot-side and cold-side temperatures. These TE transport properties show good reproducibility and thermal stability (Fig. S12). The optimized nano-composite of Bi₂Se_{1-x}Br_xS₂ is composed of 71.3% *Pnma* Bi₂SeS₂ and 28.7% *Pnnm* Bi₂SeS₂. When the *Pnnm* phase fraction increases, the

value of *ZT* decreases (Fig. 7(d)) due to deterioration of electrical conductivity and a slight enhancement of thermal conductivity, demonstrating that a moderate amount of *Pnnm* phase in the nanocomposite is ideal for achieving good TE transport properties. As a comparison, Table S4 and Fig. S13 present the TE properties, including σ , S, κ , and *ZT* value, of some representative Bi₂S₃ related systems. Both the *ZT*_{max} (1.12 at 773 K) and *ZT*_{ave} (0.72 in 323–773 K) values of as-fabricated *hoC-heS* nanocomposite of *Pnma* Bi₂SeS₂ - *Pnnm* Bi₂SeS₂ are much higher than previously reported Bi₂SeS₂ and Bi₂S₃-based materials, and other reported sulfide



Fig. 6 Electronic band structure and density of state (DOS) calculations. a and **c** Electronic band structure and DOS of pure Bi_2SeS_2 with *Pnma* structure. **b** and **d** Electronic band structure and DOS of Br-doped Bi_2SeS_2 with *Pnma* structure. **e** and **g** Electronic band structure and DOS of pure Bi_2SeS_2 with *Pnnm* structure. **f** and **h** Electronic band structure and DOS of Br-doped Bi_2SeS_2 with *Pnnm* structure.



Fig. 7 Thermal transport properties and *Z***T values. a** Temperature dependence of the total thermal conductivity (κ) of Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.09, 0.12, 0.15, 0.21). Error bars were estimated from the repeatability of the experimental result; three measurements were carried out for each material. **b** Experimental (symbols) and theoretically derived (dotted lines) lattice thermal conductivity (κ_L) of Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.09, 0.12, 0.15, 0.21). **c** *ZT* values of Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.09, 0.12, 0.15, 0.21). **c** *ZT* values of Bi₂Se_{1-x}Br_xS₂ (x = 0, 0.09, 0.12, 0.15, 0.21). **d** *ZT* values as a function of Br content, with the bars showing the *Pnma* and *Pnnm* phase percentages inside the doped systems. **e-f** Comparison of *ZT*_{ave} and *ZT*_{max} values of Br-doped Bi₂Se₂ (this work) with other typical sulfide-based materials reported in the literature^{9-11,23,37,42-48}.

compounds (Fig. 7(e,f))^{8–11,24,37,42–48}. Our work shows that the *hoC-heS* nanocomposite of *Pnma* Bi₂SeS₂ - *Pnnm* Bi₂SeS₂ is a promising TE material. The *hoC-heS* nanocomposite through a dopant-induced phase transition is an effective strategy to decouple the transport of electrons and phonons and boost the TE figure-of-merit.

Discussion

This work experimentally showed that the quasi hoC-heS nanocomposite Pnma Bi₂SeS₂ - Pnnm Bi₂SeS₂, whose phase composition was induced by a Br dopant, demonstrated excellent TE properties. The properties of this hoC-heS nanocomposite are very different from those of previously reported nanocomposites with heC-hoS or heC-heS interfaces. Br served as an effective carrier donor and also induced a partial Bi₂SeS₂ phase transition from Pnma to Pnnm, forming the hoC-heS nanocomposite. The observed Bi–Se elongated bond length clearly predicted the doping-dependent local structural disorder in the doped orthorhombic Bi₂SeS₂ system. The coherent interface between the Pnnmnanoprecipitates and Pnma matrix resulted in strongly enhanced phonon scattering and only slightly impacted the transport of electrons. As a result, a high *PF* of more than 7.30 μ W cm⁻¹ K⁻² at 773 K was obtained, inducing a record high *ZT_{max}* of 1.12 and a record high *ZT_{ave}* of 0.72 (at 323–773 K) in the optimal Br-doped *Pnma* Bi₂SeS₂ - *Pnnm* Bi₂SeS₂ nanocomposite (nominal composition: Bi₂Se_{0.88}Br_{0.12}S₂). This work provides a general strategy for enhancing TE properties by designing *hoC-he*S nanocomposites through a dopant-induced phase transition.

Methods

Polycrystalline $Bi_2Se_{1-x}Br_xS_2$ (where x = 0, 0.015, 0.03, 0.06, 0.09, 0.0105, 0.12, 0.15, 0.18, or 0.21) powders were prepared by melting stoichiometric amounts of highpurity elements (>99.99%) packed under vacuum in glass ampoules at 1173 K for 10 h, followed by annealing at 773 K for 48 h. The obtained ingots were pulverized into fine micron-sized powders by hand grinding. The powders were then compacted at 773 K by spark plasma sintering for 10 min in vacuum under a pressure of 60 MPa.

X-ray diffractometry (XRD) (Riguku, Japan) analysis was performed on all specimens with Cu K\alpha radiation, a wavelength of $\lambda = 1.5406$ Å, and a scanning speed of 4°/min. Rietveld refinement analyses were carried out by using the Generalized Structural Analysis System (GSAS-II) program. Field emission scanning electron microscopy (FE-SEM) (Ultra 55, Zeiss) was utilized to perform microstructure analysis of the freshly fractured surfaces of specimens. High-

ARTICLE

resolution transmission electron microscopy (HRTEM) (JEOL-F2010, acceleration voltage of 200 kV) was employed to characterize the microstructures. A ZEM-3 apparatus (ULVAC-Riko) was used to measure the Seebeck coefficient and electrical resistivity in a helium atmosphere from 300 to 550 K. A laser flash method with a commercial system (Netzsch, LFA-427) was used to measure the thermal diffusivity (D) in the identical direction (in-plane) in electrical resistivity measurements to avoid overvaluing ZT. Differential scanning calorimetry (DSC) (Netzsch, DSC404-C) was employed to determine specific heat (Cp). Density (d) was determined through the Archimedes method (for all Bi₂Se_{1-x}Br_xS₂ samples, the relative density ranges from 98 to 96% as x increases from 0 to 0.21). The thermal conductivity (κ) of the specimens was obtained by using the equation $\kappa = DdC_p$. Hall coefficients $(r_{\rm H})$ were evaluated through a physical properties measurement system (PPMS, Quantum design) (Fig. S14). The carrier concentration (n) was calculated by $n=1/(er_{\rm H})$, where e is the electronic charge, $r_{\rm H}$ is the Hall coefficient. The carrier mobility (μ) was calculated by $\mu = r_{\rm H}/\rho$, where ρ is the electrical resistivity, which was measured from a ZEM-3 apparatus (ZEM-3, ULVAC-Riko). Sound velocity measurements were performed at room temperature on all samples. Longitudinal and transverse sound velocities were determined by using a pulsereceiver (Olympus-NDT) equipped with an oscilloscope (Keysight). The accuracies of the resistivity, Seebeck coefficient, and thermal conductivity measurements are approximately $\pm 2\%$, $\pm 5\%$, and $\pm 5\%$, respectively. The uncertainty of ZT is about ±10%.

DFT calculations were performed using a pseudo-potential projector augmented wave method with the Perdew-Burke-Ernzerh of generalized gradient approximation exchange-correlation potential as implemented in the Vienna Atomic Simulation Package. A $2 \times 2 \times 1$ supercell containing 80 atoms was constructed with its Brillouin zone sampled with a $1 \times 3 \times 2$ k-point mesh. The plane wave cutoff energy was set to 350 eV and the atomic coordinates were relaxed until the total energy converged to 10^{-5} eV. The calculated bandgap of pristine Bi₂SeS₂ (Bi₃₂Se₁₆S₃₂) is 1.14 eV, in good agreement with previously reported values. For Br doping at Se sites, the doped system was obtained by replacing one Se atom with one Br atom (for instance, Bi₃₂Se₁₅Br₁S₃₂) as the object of calculations. This corresponded to a doping concentration of ~6.25%. The difference in doping concentration of computational experiments is due to the limitation of computational capacity, meaning the lower doping levels reported in the experimental scheme are unable to be computationally investigated. Therefore, these calculation results can only be discussed in a qualitative sense.

Reporting summary. Further information on research design is available in the Nature Research Reporting Summary linked to this article.

Data availability

All data generated or analyzed during this study are included in the published article and its Supporting Information. The data that support the findings of this study are available from the corresponding author (lifu@szu.edu.cn) upon reasonable request.

Received: 25 August 2021; Accepted: 27 November 2021; Published online: 10 December 2021

References

- Bell, L. E. Cooling, heating, generating power, and recovering waste heat with thermoelectric systems. *Science* 321, 1457–1461 (2008).
- Maciá, E. Thermoelectric materials: advances and applications (Pan Stanford, 2015).
- 3. Rowe, D. Thermoelectric handbook: macro to nano-structured materials (Taylor & Francis Boca Raton.2, 0, 2005).
- Yang, L., Chen, Z. G., Dargusch, M. S. & Zou, J. High performance thermoelectric materials: progress and their applications. *Adv. Energy Mater.* 8, 1701797 (2018).
- Liu, W. S., Yan, X., Chen, G. & Ren, Z. F. Recent advances in thermoelectric nanocomposites. *Nano Energy* 1, 42–56 (2012).
- Li, J.-F., Liu, W. S., Zhao, L.-D. & Zhou, M. High-performance nanostructured thermoelectric materials. NPG Asia Mater. 2, 152–158 (2010).
- Jabar, B. et al. Enhanced thermoelectric performance of n-type SnxBi2Te2.7Se0.3 based composites embedded with in-situ formed SnBi and Te nanoinclusions. *Compos. Part B: Eng.* 197, 108151 (2020).
- Yang, J. et al. Thermoelectric properties of n-type CuxBi2S3 materials fabricated by plasma activated sintering. J. Alloy. Compd. 780, 35–40 (2019).
- Li, L. et al. Thermoelectric property studies on CuxBi2SeS2 with nano-scale precipitates Bi2S3. *Nano Energy* 12, 447–456 (2015).
- ji, W. T. et al. Boosting the thermoelectric performance of n-type Bi2S3 by hierarchical structure manipulation and carrier density optimization. *Nano Energy* 87, 106171 (2021).

- Guo, J. et al. High thermoelectric properties realized in earth-abundant Bi2S3 bulk via carrier modulation and multi-nano-precipitates synergy. *Nano Energy* 78, 105227 (2020).
- Liu, N. M. et al. Improved thermoelectric performance of eco-friendly β-FeSi2–SiGe nanocomposite via synergistic hierarchical structuring, phase percolation, and selective doping. Adv. Funct. Mater. 29, 1903157 (2019).
- Jabar, B. et al. Enhanced power factor and thermoelectric performance for n-type Bi2Te2.7Se0.3 based composites incorporated with 3D topological insulator nanoinclusions. *Nano Energy* 80, 105512 (2021).
- Nunna, R. et al. Ultrahigh thermoelectric performance in Cu2Se-based hybrid materials with highly dispersed molecular CNTs. *Energy Environ. Sci.* 10, 1928–1935 (2017).
- Xu, Y. & Li, G. Modeling and analysis of strain effects on thermoelectric figure of merit in Si/Ge nanocomposites. In: Wang X. & Wang Z.-M., eds. *Nanoscale Thermoelectrics*. 451–495 (Springer International Publishing, 2014).
- Zhou, X. Y. et al. Routes for high-performance thermoelectric materials. Mater. Today 21, 974–988 (2018).
- Zhang, T. et al. Mechanically durable and flexible thermoelectric films from PEDOT:PSS/PVA/Bi0.5Sb1.5Te3 nanocomposites. *Adv. Electron. Mater.* 3, 1600554 (2017).
- Biswas, K. et al. High-performance bulk thermoelectrics with all-scale hierarchical architectures. *Nature* 489, 414–418 (2012).
- Deng, R. G. et al. High thermoelectric performance in Bi0.46Sb1.54Te3 nanostructured with ZnTe. *Energy Environ. Sci.* 11, 1520–1535 (2018).
- Jabar, B. et al. Achieving high thermoelectric performance through constructing coherent interfaces and building interface potential barriers in n-type Bi2Te3/Bi2Te2.7Se0.3 nanocomposites. J. Mater. Chem. A 7, 19120–19129 (2019).
- Zhang, S. N., Zhu, T. J., Yang, S. H., Yu, C. & Zhao, X. B. Improved thermoelectric properties of AgSbTe2 based compounds with nanoscale Ag2Te in-situ precipitates. *J. Alloy. Compd.* **499**, 215–220 (2010).
- Hsu, K. F. et al. Cubic AgPbmSbTe2+m: bulk thermoelectric materials with high figure of merit. *Science* 303, 818–821 (2004).
- Liu, Z. H. et al. Enhanced thermoelectric performance of Bi2S3 by synergistical action of bromine substitution and copper nanoparticles. *Nano Energy* 13, 554–562 (2015).
- Pei, Y., Lensch-Falk, J., Toberer, E. S., Medlin, D. L. & Snyder, G. J. High thermoelectric performance in PbTe due to large nanoscale Ag2Te precipitates and La doping. *Adv. Funct. Mater.* 21, 241–249 (2011).
- Zhao, L.-D. et al. Raising the thermoelectric performance of p-type PbS with endotaxial nanostructuring and valence-band offset engineering using CdS and ZnS. J. Am. Chem. Soc. 134, 16327 (2012).
- Alsaleh, N. M., Shoko, E., Arsalan, M. & Schwingenschlögl, U. Thermoelectric materials under pressure. *Phys. Status Solidi-R.* 12, 1800083 (2018).
- 27. Chang, C. & Zhao, L.-D. Anharmoncity and low thermal conductivity in thermoelectrics. *Mater. Today Phys.* **4**, 50–57 (2018).
- Li, J. et al. Low-symmetry rhombohedral GeTe thermoelectrics. *Joule* 2, 976–987 (2018).
- Loa, I., Husband, R. J., Downie, R. A., Popuri, S. R. & Bos, J. W. G. Structural changes in thermoelectric SnSe at high pressures. *J. Phys. - Condens Mat.* 27, 072202 (2015).
- Ovsyannikov, S. V. et al. Giant improvement of thermoelectric power factor of Bi2Te3 under pressure. J. Appl. Phys. 104, 053713 (2008).
- Zhu, Q. et al. Bottom-up engineering strategies for high-performance thermoelectric materials. *Nano-Micro Lett.* 13, 119 (2021).
- Lu, Q. P., Yu, Y., Ma, Q.-L., Chen, B. & Zhang, H. 2D transition-metaldichalcogenide-nanosheet-based composites for photocatalytic and electrocatalytic hydrogen evolution reactions. *Adv. Mater.* 28, 1917–1933 (2016).
- Manjanath, A., Samanta, A., Pandey, T. & Singh, A. K. Semiconductor to metal transition in bilayer phosphorene under normal compressive strain. *Nanotechnology* 26, 075701 (2015).
- Tolbert, S. H. & Alivisatos, A. High-pressure structural transformations in semiconductor nanocrystals. Annu. Rev. Phys. Chem. 46, 595–626 (1995).
- Łukaszewicz, K., Stępień-Damm, J., Pietraszko, A., Kajokas, A. & Grigas, J. Crystal structure, thermal expansion, dielectric permittivity and phase transitions of Bi2S3. *Pol. J. Chem.* 73, 541–546 (1999).
- Chen, Y. et al. Enhancing the thermoelectric performance of Bi2S3: a promising earth-abundant thermoelectric material. *Front. Phys.* 14, 1–12 (2019).
- Biswas, K., Zhao, L.-D. & Kanatzidis, M. G. Tellurium-free thermoelectric: the anisotropic n-type semiconductor Bi2S3. *Adv. Energy Mater.* 2, 634–638 (2012).
- Liu, W. S. et al. Studies on the Bi2Te3-Bi2Se3-Bi2S3 system for midtemperature thermoelectric energy conversion. *Energy Environ. Sci.* 6, 552-560 (2013).
- Kirklin, S. et al. The open quantum materials database (OQMD): assessing the accuracy of DFT formation energies. NPJ Comput. Mater. 1, 1–15 (2015).

- Zhang, Y. N. Communication: Surface stability and topological surface states of cleaved Bi2Se3: First-principles studies. J. Chem. Phys. 143, 151101 (2015).
- Liu, W. S. et al. Enhanced thermoelectric properties in CoSb3-xTex alloys prepared by mechanical alloying and spark plasma sintering. *J. Appl. Phys.* 102, 103717 (2007).
- 42. Ge, Z. H. et al. Highly enhanced thermoelectric properties of Bi/Bi2S3 nanocomposites. *ACS Appl. Mater. Inter.* **29**, 4828–4834 (2017).
- Li, F. et al. High thermoelectric properties achieved in environmentally friendly sulfide compound Bi2SeS2 by nanoenginnering. *Nano Energy* 88, 106273 (2020).
- Du, X. et al. Enhanced thermoelectric performance of n-type Bi2S3 with added ZnO for power generation. RSC Adv. 5, 31004–31009 (2015).
- Yang, J. et al. Enhanced the thermoelectric properties of n-type Bi2S3 polycrystalline by iodine doping. J. Alloy. Compd. 728, 351–356 (2017).
- Ruan, M., Li, F., Chen, Y. X., Zheng, Z. H. & Fan, P. Te-free compound Bi2SeS2 as a promising mid-temperature thermoelectric material. *J. Alloy. Compd.* 849, 156677 (2020).
- Guo, J., Ge, Z. H., Qian, F., Lu, D. H. & Feng, J. Achieving high thermoelectric properties of Bi2S3 via InCl3 doping. J. Mater. Sci. 55, 263–273 (2020).
- Guo, Y., Du, X. L., Wang, Y. L. & Yuan, Z. H. Simultaneous enhanced performance of electrical conductivity and Seebeck coefficient in Bi2-xSnxS3 by solvothermal and microwave sintering. *J. Alloy. Compd.* **717**, 177–182 (2017).

Acknowledgements

This work was supported by the National Natural Science Foundation of China (No. 52072248), National Key Research and Development Program of China (No. 2018YFB0703600), Natural Science Foundation of Guangdong Province of China (No. 2011515012128, No. 2018A030313574), Guangdong Innovative and Entrepreneurial Research Team Program (No. 2016ZT06G587), and Natural Science Foundation of SZU (No. 827-000357), as well as Shenzhen Key Projects of Long-Term Academic Support Plan (No. 20200925164021002). F.L. and B.J. also wish to acknowledge the HAADF-STEM assistance provided by the Electron Microscope Center of Shenzhen University. W.L. acknowledges the Tencent Foundation through the XPLORER PRIZE and the Guangdong Provincial Key Laboratory Program (2021B1212040001) of the Department of Science and Technology of Guangdong Province, China.

Author contributions

The paper was prepared though the contribution of all authors. B.J., F.L. and W.L. designed the work. B.L., Z.Z., C.L. and D.A. prepared the nanocomposite and measured

the thermoelectric transport properties. B.L. and Y.C. performed structrural nanocomposite characterization. A.M. and Y.Z. performed simulation. Z.Z. and P.F. planed and supervised the work. F.L. and W.L. wrote the paper. B.L. and G.L. had major input in the wirting of the paper. All the authors edited the paper.

Competing interests

The authors delcare no competing intrerests.

Additional information

Supplementary information The online version contains supplementary material available at https://doi.org/10.1038/s41467-021-27564-2.

Correspondence and requests for materials should be addressed to Fu Li, Ping Fan or Weishu Liu.

Peer review information *Nature Communications* thanks Yung-Kang Kuo and the other anonymous reviewer(s) for their contribution to the peer review of this work. Peer reviewer reports are available.

Reprints and permission information is available at http://www.nature.com/reprints

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/ licenses/by/4.0/.

© The Author(s) 2021