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Heterogeneous integration of single-crystalline rutile nanomembranes with steep phase transition on silicon substrates

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Unrestricted integration of single-crystal oxide films on arbitrary substrates has been of great interest to exploit emerging phenomena from transition metal oxides for practical applications. Here, we demonstrate the release and transfer of a freestanding single-crystalline rutile oxide nanomembranes to serve as an epitaxial template for heterogeneous integration of correlated oxides on dissimilar substrates. By selective oxidation and dissolution of sacrificial VO₂ buffer layers from TiO₂/VO₂/TiO₂ by H₂O₂, millimeter-size TiO₂ single-crystalline layers are integrated on silicon without any deterioration. After subsequent VO₂ epitaxial growth on the transferred TiO₂ nanomembranes, we create artificial single-crystalline oxide/ Si heterostructures with excellent sharpness of metal-insulator transition ($\Delta \rho / \rho > 10^3$) even in ultrathin (<10 nm) VO₂ films that are not achievable via direct growth on Si. This discovery offers a synthetic strategy to release the new single-crystalline oxide nanomembranes and an integration scheme to exploit emergent functionality from epitaxial oxide heterostructures in mature silicon devices.

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eteroepitaxial growth has been widely used to obtain single-crystal films for developing modern solid-state electronic and photonic devices^{1,2}. In particular, epitaxial oxide heterostructures provide promise for emerging electronics and photonics due to the intriguing phenomena (e.g., metalinsulator transition $(MIT)^{3-6}$ high-electron mobility⁷, ferroelectricity⁸) exhibited by these ionic crystals. Therefore, the stacking of single-crystalline oxide films on dissimilar substrates (e.g., silicon (Si)) will offer ways to integrate the emergent phenomena of oxides with mature electronic and photonic devices⁹⁻¹³. However, the heteroepitaxial growth of oxide films drastically limits the possible material combinations due to the requirement of lattice matching between the epilaver and substrate¹⁴. For instance, direct oxide growth on dissimilar materials typically forms defective or polycrystalline layers near the interface between films and substrates^{15,16}, preventing unrestricted integration of single-crystal oxide films onto any desired substrates, especially on mainstream Si substrates.

Release and transfer of freestanding single-crystal sheets with a nanoscale thickness (i.e., epitaxial lift-off for nanomembrane (NM)) gives the freedom to transfer the released epilayer onto highly mismatched or amorphous substrates, and even allows reusable substrates^{9,17–21}. While crystalline materials that are intrinsically layered (e.g. two-dimensional (2D) materials) are exfoliated spontaneously due to weak van der Waals bonding between layers²², the freestanding NM from three-dimensional (3D) oxide crystals with strong bonding is hindered by the technical challenges of lifting strongly bonded epitaxial films from the oxide substrates. Thus, the technique for releasing the freestanding NM from a host substrate essentially requires the preferential creation of bond breaking from the substrate.

Many techniques have been employed to form freestanding NMs with a mechanically cleavable plane. For example, physical release methods (e.g., laser lift-off²⁰) were originally developed to release epitaxial GaN semiconductor films to break strong bonding. However, these methods are only applicable to the formation of thick semiconductor membranes due to the inevitable structural damage. Moreover, a few monolayers of graphene could be inserted to release single-crystalline oxide membranes from substrates (i.e., "remote" epitaxy). Despite the versatility of this technique, the coalescence of localized nuclei in oxide layers

on the graphene, along with the restriction of oxygen environment during growth, prevents the layer-by-layer growth of ultrathin oxide NM with atomic precision and high quality¹⁰.

By contrast, the freestanding NM was chemically released from the substrate by selective etching of sacrificial layers^{18,19,21}; these chemical lift-off methods are less destructive than physical methods. However, the harsh wet condition with a strong acid or base etchant typically leaves roughening and residue on the host substrates or released membrane after the chemical $etch^{23,24}$. Recently, atomically thin perovskite oxide NM was gently released by dissolving water-soluble Sr₃Al₂O₆ sacrificial layers¹¹, but the moisture-sensitive nature of these layers prevents longtime exposure of the sacrificial layers, which restricts practical application for heterogeneous integration of oxide NM. Moreover, the development of oxide NM has been limited for perovskite structure among chemical lift-off methods so far^{10-12,21,25,26}; to extend the materials spectrum for freestanding oxide NM, a new combination of the sacrificial layer and etchant needs to be developed for the heterogeneous integration of epitaxial oxide NM with other crystal structures on dissimilar substrates¹⁰.

Here, we demonstrate that single-crystalline rutile oxide NM with the sharpened MIT can be integrated on the technologically influential, but more challenging, Si substrates by new epitaxial lift-off combination (see Fig. 1 for the process schematic). After the synthesis of an epitaxial TiO2/VO2 heterostructure on TiO2 host substrate (Fig. 1a), the VO₂ sacrificial layer is selectively dissolved in dilute H₂O₂ to release the top TiO₂ film with the mechanical supporting layers (Fig. 1b). Contrary to the previous chemical lift-off using extreme pH solution, dilute aqueous H₂O₂ with mild pH leads to selective etching of epitaxially grown VO2 sacrificial films by phase transformation to two-dimensional layered structure with weak bonding; this selective dissolution of VO₂ layer releases millimeter-scale freestanding TiO₂ NM from the TiO₂/VO₂/TiO₂ heterostructures at room temperature. Then, the single-crystalline TiO₂ NM is transferred onto the Si substrates without any deterioration of crystal quality (Fig. 1c, d). Interestingly, transferred TiO₂ single-crystal NM serves as a template for the heterogeneous integration of single-crystal VO₂ films on Si substrate (Fig. 1e). As a result of the VO₂ film epitaxially grown on TiO₂ NM template, more than three orders of



Fig. 1 Process schematic for single-crystal rutile oxide nanomembranes (NM) on silicon. a Schematic of an epitaxial TiO_2/VO_2 heterostructure on TiO_2 host substrate. **b** The VO₂ layer is dissolved in H_2O_2 to release the top TiO_2 film with the mechanical supporting layer. **c** The freestanding TiO_2 NM is transferred onto the desired substrates (e.g., silicon). **d** By removing the rigid supporting layer, single-crystalline rutile oxide NM is heterogeneously integrated into a silicon substrate. **e** Epitaxial VO₂ film with steep phase transition is grown on the TiO_2 -NM-templated Si substrates.



Fig. 2 Crystal and surface quality of released and transferred TiO₂ **single-crystal nanomembrane (NM). a** Symmetric $2\theta - \omega$ scan. **b** Reciprocal space mapping around (112) reflection of as-grown TiO₂ epitaxial films and TiO₂ single-crystal NM released on supporting layer and transferred on Si substrates. The high-resolution X-ray diffraction measurements show single-crystallinity with uniform out-of-plane orientation and no in-plane rotation in released and transferred TiO₂ NM. **c** EBSD maps of the as-grown TiO₂ epitaxial films on VO₂/TiO₂ substrates (top) and transferred TiO₂ NM on Si substrates (bottom), confirming the single-crystalline out-of-plane orientation. **d** Photograph of 40-nm-thick TiO₂ single-crystal NM released on mechanical support with the identical lateral dimension with TiO₂ host substrates. **e** OM image of transferred TiO₂ NM on the carbon TEM grid. **f** AFM image of as-grown TiO₂ films and transferred TiO₂ NM, which confirms a uniform and flat surface without surface cracks or residues.

magnitude modulation of resistivity ratio $(\Delta \rho / \rho > 10^3)$ and sharpened MIT (i.e., narrow FWHM of Gaussian fitting from d $(\log_{10}(\rho))/dT$ during heating and cooling $(\Delta T_h \text{ and } \Delta T_c \sim 3 \text{ K}))$ are remarkably achieved even in the ultrathin ($\leq 10 \text{ nm}$) VO₂ films on Si substrates, which is not possible using conventional thin film growth.

Results

Synthesis of freestanding single-crystalline TiO₂ nanomembranes. Prior to release and transfer of TiO2 NM, TiO2 (10-70 nm)/VO2 (~17 nm) heterostructures were epitaxially grown on (001)-oriented TiO₂ substrates by pulsed laser deposition (PLD) (Fig. 1a). Symmetrical 2θ - ω scan using synchrotron x-ray scattering on TiO₂/VO₂/TiO₂ detected two (002)_R Bragg reflections, one from the rutile TiO₂ ($2\theta = 49.54^{\circ}$) film and substrate, and one from VO₂ film ($2\theta = 51.75^{\circ}$) (black line in Fig. 2a)²⁷. Due to the close resemblance in crystal structure and small in-plane lattice mismatch of $(001)_{\rm R}$ -VO₂ and (001)-TiO₂ ($f_a \approx 0.86\%$ along a- and b-axis), 17-nm-thick (001)_R-VO₂ films are fully strained by biaxial tensile strain; all heterostructure maintained identical in-plane lattice constant (left figure of Fig. 2b). Thus, the thickness of VO₂ (~17 nm) was determined to create unstrained and defect-free TiO2 epitaxial films regardless of film thickness^{27,28}.

Then, rigid support layer (e.g., PDMS) was coated on the heterostructure before release to mechanically stabilize the TiO_2 NM and to facilitate the subsequent transfer of the released

 $\rm NM^{10-12,25}$. By simply immersing these TiO₂/VO₂/TiO₂ heterostructures into the dilute (10%) hydrogen peroxide (H₂O₂) solution at room temperature (Fig. 1b), a few millimeter-scale TiO₂ films were successfully released from their substrates by fully dissolving VO₂ sacrificial layers (i.e., denoted as released TiO₂ NM), allowing the growth substrate to be removed. The released single-crystal NM was transferred on any arbitrary substrate (i.e., denoted as transferred TiO₂ NM), including on SiO₂-coated (~80 nm) Si (100) substrates (Fig. 1c). The transferred NM was then heated at 60 °C to form stronger adhesion with substrates and the support layer was slowly detached (Fig. 1d).

Interestingly, the VO₂ sacrificial layer was dissolved selectively and rapidly in the dilute H_2O_2 solution with mild pH (~5.3) (Fig. 1b). The combination of the VO_2 sacrificial layer and H_2O_2 etchant for single-crystal rutile oxide NM in terms of release rate is as effective as that of a Sr₃Al₂O₆ sacrificial layer and H₂O for a single-crystal perovskite oxide NM¹¹. The high etching capability of H₂O₂ on VO₂ was confirmed by direct comparison with that of an HNO3 solution with the same concentration as a wellestablished VO₂ etchant with a strong acid (pH ~ 2)²⁴ (Supplementary Fig. 1): While the residue of VO₂ films remained even after 64 s after VO₂ films were immersed into strong acid HNO₃ solution, the H₂O₂ solution completely removed VO₂ films that were free of any residue on the surface of the TiO₂ substrate within 4 s. Furthermore, repeating H₂O₂ etching preserves the atomically flat surface topography, implying substrate reusability for the production of TiO₂ NM (Supplementary Fig. 2). VO₂ films could be spontaneously oxidized by H_2O_2 and transformed to V_2O_5 and/or water-soluble $V_2O_5 \cdot nH_2O(s)$ gels with a layered van der Waals (vdW) structure along the c-axis^{29,30}; these layered crystals with weak bonding are exfoliated and dispersed in the solution (Supplementary Fig. 3a, b). Since unreacted TiO₂ epitaxial layers cover whole surfaces of VO₂ sacrificial layers during the release process, the release of epitaxial TiO₂ NM by oxidation and dissolution of VO₂ sacrificial layers begins from the edge of the substrate (i.e., side etching); the dissolution (or release) time increases with the lateral size of the TiO₂ substrate (Supplementary Fig. 4).

As a result of effective dissolution of the VO₂ sacrificial layer by H₂O₂, the quality and alignment of epitaxial TiO₂ layers are intact during the release and transfer process as observed in symmetric 20-w scans using synchrotron X-ray scattering (Fig. 2a). While the $(002)_{\rm R}$ peak from the VO₂ sacrificial layer $(2\theta = 51.75^{\circ})$ was clearly removed after the selective etching process (red line in Fig. 2a), the $(002)_{\rm R}$ peak from ~ 40-nm-thick (001) TiO₂ singlecrystal films with thickness oscillations was observed consistently at $2\theta = 49.54^{\circ}$ in released (red line in Fig. 2a) and transferred TiO₂ NM on the Si substrate (blue line in Fig. 2a and Supplementary Fig. 5 with Cu $K_{\alpha 1}$ X-ray radiation). Moreover, the $(002)_R$ peak from TiO₂ NM simultaneously appeared with a (400) peak from the Si substrate ($2\theta = 54.32^{\circ}$) in the transferred TiO_2 NM (blue line in Fig. 2a), which represents that the c-axis of single-crystal TiO₂ NM was aligned with the out-of-plane orientation of the Si substrate. Azimuthal X-ray diffraction (XRD) ϕ scanning of (112)_R crystallographic plane showed identical fourfold symmetry of released and transferred TiO₂ NMs with those of the single-crystal (001) TiO₂ host substrate (Supplementary Fig. 6); this result showed in-plane singlecrystallinity of NMs without any rotated domains. Additionally,

an electron backscatter diffraction (EBSD) map confirmed (001) orientation of rutile crystals over a large area in both as-grown and transferred TiO_2 layers (Fig. 2c).

To obtain more detailed information on the crystal structures of TiO₂ NMs during the release and transfer process, both inplane and out-of-plane lattice parameters were monitored using reciprocal space mapping (RSM) around the (112) reflection of as-grown, released, and transferred 70-nm-thick TiO₂ lavers (Fig. 2b). The RSM data clearly show sharp and intense (112) Bragg reflections and Kiessig fringes from the TiO₂ layer and substrates, and from the VO₂ sacrificial layers in as-grown heterostructures. Since 17-nm-thick VO2 sacrificial layers are coherently grown on the TiO₂ substrates with identical H (i.e., inplane reciprocal space unit), strain-free epitaxial TiO₂ layers with various thickness (10-70 nm) are coherently grown on VO₂/TiO₂ substrates (Fig. 2b and Supplementary Fig. 7b, c). After selective etching of the VO₂ sacrificial layer, H and L in released and transferred TiO₂ NM was identical with those in TiO₂ substrates and films in as-grown heterostructures; these strain-free TiO₂ layers with precisely controlled thickness are released and transferred onto a Si substrate without modification of the lattice parameters and crystallinity (Supplementary Fig. 7a).

Furthermore, surfaces of the released and transferred TiO_2 NM are uniform and intact without any defective boundaries³¹. Figure 2d shows that the entire area of TiO_2 films with lateral dimension of millimeter-scale and nanometer thickness was fully released from the VO₂/TiO₂ substrates after this etching process. Optical microscope (OM) images of transferred TiO_2 NM on the carbon TEM grid exhibit crack-free layers with natural wrinkles (Fig. 2e and Supplementary Fig. 8a). Both scanning electron microscope (AFM, Fig. 2f) images confirmed



Fig. 3 STEM analysis of freestanding single-crystal TiO₂ nanomembrane (NM). a HAADF-STEM, b ABF-STEM, and c SADP images from the plane-view observation on 20-nm-thick freestanding TiO₂ NM. The white spots in HAADF-STEM, which are identical to Ti atoms, are consistent with cation sites of $[001]_{R}$ -projected rutile structure. The atomic arrangement of ABF-STEM, visualization of oxygen atoms, was also identical to anion sites of rutile structure; the results indicate that atomic arrangement of freestanding TiO₂ NM was identical to the single-crystal rutile structure without defective features. **d**, **e** are the cross-sectional ABF-STEM images of TiO₂ NM, the quality of which is crystallographically high with perfect registry of Ti (orange color in **e**) and O (red color in **e**) atoms in rutile structure.

uniform and flat surface of both released and transferred TiO_2 NM without surface cracks or residues: the arithmetic average surface roughness of transferred TiO_2 NM ($R_a = 0.06$ nm) was comparable to that of originally grown TiO_2 films on VO_2/TiO_2 substrates ($R_a = 0.04$ nm), representing residue-free VO_2 removal by H_2O_2 etching and subsequent surface cleaning (Fig. 2f and Supplementary Figs. 9, 10).

High-resolution scanning transmission electron microscopy (STEM) confirms the local atomic structure of freestanding single-crystal TiO₂ NM. First, the released TiO₂ NM was transferred onto a TEM grid for STEM observation (Fig. 2e and Supplementary Fig. 11); the high-angle annular dark-field (HAADF) and the annular bright-field (ABF) STEM were applied for the plane-view observation. The HAADF-STEM image shows a square pattern of four titanium atoms (orange, dotted square) without a lattice distortion or defects (Fig. 3a). More interestingly, the ABF-STEM image shows the contrast tails from the titanium atom with contribution by the oxygen atoms and the pattern exactly matched the atomic structure of the rutile TiO₂ with the [001] zone axis (Fig. 3b). The selectedarea diffraction pattern (SADP) provided that the in-plane lattice parameters of TiO2 NM were 0.46 nm (Fig. 3c), which confirms the freestanding TiO₂ film was perfectly transferred. After the TiO₂ NM (~60 nm) was transferred to the SiO₂/Si substrate, cross-sectional STEM observation was performed with the [100] zone axis to visualize the structural coherency of the freestanding oxide NM, as shown in Fig. 3d, e. The low magnification ABF-STEM image shows that the TiO₂ NM is free of defects even after its transfer process from the wide field of view (Fig. 3d); the high magnification ABF-STEM image again verifies the perfect registry of Ti (orange) and oxygen (red) atoms in the rutile structure (Fig. 3e).

Heterogeneous integration of single-crystalline VO₂ films on the TiO₂-NM-templated Si substrates. After transfer to the (SiO₂-coated) Si substrates, the single-crystalline TiO₂ NMs served as templates for the epitaxial growth of high-quality VO₂ thin films on Si substrates (Fig. 1e). As observed in symmetric $2\theta - \omega$ scans, the intense $(002)_{\rm R}$ VO₂ peak appeared at $\sim 2\theta =$ 51.85° along with peaks related to the TiO₂ template (~ $2\theta =$ 49.54°) and Si substrates ($\sim 2\theta = 54.34^\circ$) for the VO₂ films on the TiO₂-templated Si substrates (red line in Fig. 4a), compared to the absence of related peaks for the VO₂ films directly grown on Si substrate (black line in Fig. 4a); this result reveals that latticematched single-crystal templates facilitate the formation of epitaxial VO₂ films on Si substrates. Single-crystallinity of VO₂ films on TiO₂ NM/Si was again verified by identical fourfold symmetry with TiO₂ NM templates in asymmetric ϕ -scans of the (112)_R plane (Fig. 4b). While the out-of-plane strain state of the TiO₂ NM did not change after the growth of VO₂ films, the out-ofplane lattice parameters of VO₂ films (2.838 Å) were reduced compared to those of bulk VO_2 (2.88 Å). Indeed, an RSM near the (112) reflection of the TiO_2 NM confirms that the peaks from the VO₂ films and TiO₂ NM showed identical H (i.e., in-plane reciprocal space unit) (Fig. 4c), which indicates that a 10-nm-thick VO₂ film remains coherently strained to the TiO₂ NM along the in-plane direction^{27,28}.

Low magnification ABF- and high magnification HAADF-STEM images (Fig. 4d, e) confirm that the epitaxial growth of 10-nm-thick VO_2 on TiO₂ NM can realize heterogeneous integration of singlecrystal VO_2 films on a Si substrate. In particular, VO_2 and TiO₂ are coherently matched and the interface can be visualized due to the slight contrast difference between VO_2 and TiO₂, as indicated by the arrow (Fig. 4e); the atomic resolution image implicates that the VO_2 layer is tightly constrained from the underlying TiO₂ NM, and thus



Fig. 4 Heterogeneous integration of rutile oxide epitaxial VO₂/TiO₂ heterostructures on Si substrates. a Symmetric $2\theta - \omega$ scan, **b** asymmetric ϕ scan, and **c** reciprocal space mapping around (112) reflection of TiO₂ NM on Si substrates before (blue in **a**, **b**, and left in **c**) and after (red in **a**, **b**, and right in **c**) the growth of VO₂ films, implying single-crystallinity of coherently strained VO₂ films on TiO₂ NM/Si. **d** The cross-sectional ABF-STEM images of the heterogeneous VO₂/TiO₂ NM on SiO₂/Si. The atomic-scale HAADF-STEM image at the area indicated by the yellow square in **d** is shown in **e** (zone axis: [100] in TiO₂).



 TiO_2 NM templates between VO₂ and Si enable the growth of epitaxial VO₂ films with crystallographic perfection free from defects.

To benchmark the quality of the VO2 thin films on the TiO2-NMtemplated SiO₂/Si substrates in terms of MIT, temperaturedependent resistivity was measured for VO2 films depending on the existence of transferred TiO2 NM. Higher resistivity modulation ($\Delta \rho / \rho \sim 3.3 \times 10^3$) was observed across $T_{\rm MI} \sim 296$ K in both 5-nm-thick and 10-nm-thick VO2 on TiO2 NM/SiO2/Si (red line in Fig. 5a and Supplementary Fig. 13b) than in 10-nmthick VO2 on SiO2/Si (Supplementary Fig. 13a) and even in 50-nm-thick VO₂ on SiO₂/Si ($\Delta \rho / \rho \sim 5.2 \times 10^2$ across the T_{MI} ~ 343 K) (black line in Fig. 5a). In addition to the resistivity ratio, other MIT properties $(d(\log_{10}(\rho))/dT, T_h, T_c, \Delta H, \Delta T_h, \Delta T_c)$ were characterized as shown in Fig. 5b³². $d(\log_{10}(\rho))/dT$ was plotted and fitted with a Gaussian function. $T_{\rm h}$ and $T_{\rm c}$ were then defined at the peak position of the Gaussian and the transition width was calculated from the difference ($\triangle H = T_{\rm h} - T_{\rm c}$). The transition sharpness $\triangle T_{\rm h}$ and $\triangle T_c$ was estimated from the full-width at half-maximum of the Gaussian peak. In fact, the transition width and sharpness, as well as Fig. 5 Single-crystalline VO₂ films with steep phase transition on TiO₂-NM-templated Si substrate. a Temperature-dependent resistivity modulation near T_{MIT} in 10-nm-thick VO₂ films on TiO₂ NM/SiO₂/Si (red lines) and 50-nm-thick VO₂ films on SiO₂/Si (black lines). **b** Comparison of various metal-insulator transition properties $(d(\log_{10}(\rho))/dT, T_{h}, T_{c}, \Delta H)$ $\Delta T_{\rm h}$, $\Delta T_{\rm c}$) between 10-nm-thick VO₂ films on TiO₂ NM/SiO₂/Si (red lines) and 50-nm-thick VO2 films on SiO2/Si (black lines). Note that the transition width and sharpness, as well as the resistivity ratio, were substantially improved in VO₂ films with TiO₂ NM, compared to those without TiO₂ NM. **c** Benchmark of resistivity ratio $\Delta \rho / \rho = (\rho_{T_{MIT}-15K} - \rho_{T_{MIT}-15K})$ $\rho_{T_{MIT}+15K})/\rho_{T_{MIT}+15K}$ for VO₂ films on Si substrates across the MIT. For a direct comparison, all films were grown on oxide-coated Si substrates using various growth techniques: (sputtering (black square), ALD (green circle), PLD (blue diamond), and sol-gel (orange hexagon)) reported in the previous literature (Supplementary Table 1). While deteriorated $\Delta \rho / \rho$ was observed in VO₂ films directly grown on Si due to the polycrystallinity of the films and the formation of a defective interfacial layer, epitaxial growth of VO₂ films (5 and 10 nm) guided by transferred TiO₂ NM enables integration of correlated oxides on silicon substrates with the highest modulation of resistivity ratio across the metal-insulator transition (red stars).

the resistivity ratio, were significantly improved in the VO2 films with TiO₂ NM (i.e., 8.70 and 3.3 K for $\triangle H$ and $\triangle T_c$, respectively) compared to those without TiO₂ NM (i.e., 17.94 and 14.54 K for $\triangle H$ and ΔT_c , respectively). The sharpened phase transition is attributed to the single-crystalline nature of the VO2 films perfectly aligned with the underlying single-crystal TiO2 NM templates. Moreover, the transferred single-crystal TiO2 NM forms coherently tensile-strained VO_2 films; this epitaxial strain leads to a T_{MI} shift close to room temperature in VO2 films on TiO2 NM/SiO2/Si compared to relaxed VO_2 films on SiO₂/Si substrates^{27,28,32}. Along with steep MIT under temperature, these single-crystalline VO2 films on TiO2 NM/Si show high endurance during thermal and electrical cycling. Both thermally induced MIT and electrically induced MIT $(I_{on}/I_{off} > 10^3)$ were consistently observed without any drift during the multiple cycles of thermal switching (Supplementary Fig. 14) and electrical switching (Supplementary Fig. 15), respectively.

Discussion

It should be noted that the integration of VO₂ films with excellent electrical properties on a Si substrate has proved challenging due to fundamental limitations. A representative set of data of MIT properties from VO₂ thin films on Si (or oxide-coated Si or buffered Si) substrates was compiled to allow for a direct comparison with those from our films (Fig. 5c, Supplementary Table 1, and Supplementary Fig. 16). Despite previous attempts to directly grow VO₂ thin films on Si substrates using various deposition techniques, $\Delta \rho / \rho$ across the $T_{\rm MI}$ deteriorated due to the formation of polycrystallinity of the films. Notably, a further reduction in $\Delta \rho / \rho$ was found to occur with decreasing film thickness, likely due to the formation of defective interfacial layers (e.g., by thermodynamic reaction of VO_2 with Si (or SiO₂)); the cation-to-anion stoichiometry was not maintained in the VO₂ films near the interface^{33,34}. If this compositional variation (i.e., off-stoichiometry) contains a significant fraction of the films, the resistivity ratio and transition sharpness will significantly degrade in the regime of an ultrathin thickness (<40 nm), as observed in our 10-nm-thick VO2 films on SiO2/Si and Al2O3-buffered Si (Supplementary Figs. 13, 16); steeper transition cannot be engineered simply by direct deposition of thin VO₂ films on Si substrates due to the existence of interfacial layers and the substantial density of defects.

However, the TiO₂ template NM on Si substrates permits epitaxy with VO₂ thin films due to its identical crystal structure (P4₂/mnm) and small lattice mismatch ($f_a \sim 0.86\%$). The transferred NM could play a crucial role as a seed layer for epitaxy, and artificially allows the first demonstration on heterogeneous integration of singlecrystalline VO₂ on Si substrates, which are commonly employed in electronic devices. Furthermore, the TiO₂ NM is likely to prevent subsequent reaction between VO₂ and the Si substrate, and allows excellent control over the V-oxidation states without any extended defects^{33,34}. Thus, epitaxial growth of ultrathin VO₂ films guided by transferred TiO₂ NM enables integration of correlated oxides with unprecedented modulation of the resistivity ratio ($\Delta \rho / \rho > 10^3$) across the MIT in the regime of an ultrathin thickness (5–10 nm) on Si substrates (see red stars in Fig. 5c).

In summary, heterogeneous integration of a freestanding singlecrystalline rutile oxide NM was achieved by exploiting selective oxidation and dissolution of isostructural sacrificial layers. Despite a mild pH condition under a dilute H₂O₂ solution at room temperature, VO₂ sacrificial films are spontaneously oxidized by H₂O₂ and transformed to layered crystals, which in turn could be exfoliated and dispersed in solution to release and transfer the millimeter-scale TiO₂ NM with controlled thickness on oxide-coated Si substrates. Owing to the nearly perfect single-crystallinity of the transferred rutile TiO₂ NM, this lattice-matched single-crystal template permits heterogeneous integration of epitaxial VO₂ films on Si substrates; exceptional MIT characteristics in terms of transition sharpness ($\Delta T_h \sim$ 2.85 K and $\Delta T_c \sim$ 3.3 K) and resistivity ratio ($\Delta \rho / \rho > 10^3$) were realized in ultrathin VO₂ films integrated on Si substrates, benefitting from the superior quality of epitaxial oxide films.

Our strategy to release and transfer a freestanding epitaxial rutile oxide NM for a MIT will provide an unprecedented platform using emergent phenomena in epitaxial oxide heterostructures to be integrated with current state-of-the-art Si-based technology (e.g., integrated electronics^{3,6} and photonics^{13,35} on Si). For example, our epitaxial oxide NM with steep phase transition could deliver a hybrid optical modulator with a high extinction ratio, low loss, and high modulation speed integrated on Si waveguides^{13,35,36}. Moreover, our approach for single-crystalline oxide NMs with high thermal stability are generally applicable for heterogeneous integration of high-quality oxide NM with any rutile crystal structures on Si substrates (e.g., epitaxial growth of metallic RuO₂ single-crystalline NMs on transferred TiO₂ NM/Si or direct transfer of metallic RuO₂ single-crystal NMs on Si, Supplementary Figs. 17-20); our study enables us to extend the materials spectrum for freestanding single-crystalline rutile oxide NM by developing a new combination of sacrificial layer and etchant. It also offers a unique opportunity for new types of artificial heterostructures by stacking multi-functional oxide NMs with other 2D/3D single-crystalline NMs (e.g., III-V, exfoliated 2D layered materials, complex oxides)^{10,17} or by controlling twisted angle between two misaligned sheets of oxide NMs (similar to twisted bilayer graphene heterostructure with atomic and electronic reconstruction)37,38 for novel interfacial physics and a new generation of emergent devices.

Methods

Epitaxial film growth. The epitaxial VO₂ thin films (~17 nm) were grown on (001) TiO₂ single-crystal substrates, followed by the growth of TiO₂ films (10–70 nm) by PLD. First, (001) TiO₂ single-crystal substrates (Shinkosha CO, LTD) with lateral size of up to 5 mm × 5 mm were loaded into the PLD chamber, which was then evacuated to a base pressure of ~1 × 10⁻⁶ Torr. The rotating V₂O₅ and TiO₂ targets were then ablated by focusing a KrF excimer laser (Coherent Complex Pro 102 F, $\lambda = 248$ nm) with a fluence of 1 J/cm² and a repetition rate of 1 Hz. The growth of VO₂ films was performed at fixed P_{O2} = 15 mTorr and T_g = 300 °C, which were selected to induce a steep MIT near room temperature from coherently tensile-strained VO₂ films. Subsequently, TiO₂ films were grown on VO₂ templates under P_{O2} = 28 mTorr and T_g = 300 °C. After the growth of heterostructure, the samples were cooled to room temperature at a rate of 20 °C/min. The epitaxial VO₂ film (5–10 nm) was grown on the

transferred TiO_ NM/Si with an identical growth condition of that of VO_ sacrificial layers on the (001) TiO_ substrate.

Release and transfer of freestanding TiO, NMs. To release the freestanding TiO2 NM from the VO2/TiO2 substrates, the surface of the grown heterostructure was adhered onto a rigid supporting layer (e.g., polydimethylsiloxane, thermal release tape (Haeun Chemtec, RP70N5)). The structure was immersed in a dilute H2O2 solution until the sacrificial VO2 layer was completely dissolved. To transfer the released TiO₂ NM to other substrates, TiO₂ NM was placed onto an oxidecoated silicon substrate and exposed to an appropriate temperature. Finally, the freestanding NM remained on the silicon substrate after peeling off the rigid supporting layer slowly. The structure was immersed in a dilute 10% H2O2 solution (i.e., 50 ml of 35% $H_2O_2 + 150$ ml of H_2O) at room temperature until the sacrificial VO₂ layer was completely dissolved, with the freestanding NM left on the rigid supporting layer. After dissolving the VO2 layers in H2O2, H2O residue on the released TiO₂ surface was evaporated in a vacuum desiccator for 10 min. To transfer the released TiO₂ NM to other substrates (such as thermally grown SiO₂coated silicon), the TiO2 NM/rigid supporting layer was attached to a oxide-coated silicon substrate and was heated to 60-80 °C for 10 min. Finally, the freestanding NM remained on the silicon substrate after peeling off the rigid supporting layer slowly.

SEM, EBSD, and AFM measurements. OM images were recorded using BX53M (Olympus, Tokyo, Japan) microscope, equipped with an objective MPlanFL N (Olympus) and i-solution IMT cam CCD camera. The SEM and EBSD measurements were made using an XL30SFEG and FEI Helios Nanolab 650 equipped with an EBSD detector, respectively. The EBSD pattern was acquired by an EDAX Hikari EBSD camera while the sample was tilted at 70° and scanned with an electron beam of 25 nA at 20 kV. The measured data were analyzed using TSL OIM Analysis7. The AFM images were obtained using a VEECO Dimension 3100 in tapping mode.

X-ray diffraction (XRD) and X-ray absorption spectroscopy (XAS). Highresolution x-ray scattering was performed by using synchrotron radiation at the 3D XRS ($\lambda \sim 0.12398$ nm, energy ~10 keV at Si (111)) beamline of Pohang Light Source-II (PLS-II, Pohang, Republic of Korea), and using an in-house HRXRD (Bruker Discover 8 X-ray diffractometer) with Cu K_{a1} radiation ($\lambda \sim 0.15406$ nm). The detailed information on in-plane and out-of-plane lattice parameters and strain states of each films and NM was obtained by using both symmetric 20- ω scan and asymmetric RSM around the (112) reflection. X-ray absorption spectroscopy (XAS) was performed using the 2 A MS beamline at PLS-II. The total electron yield mode with an energy resolution of ~0.1 eV was used for measurement at a base pressure of 5 × 10 $^{-10}$ Torr in the analysis chamber by measuring the sample current (I₁) divided by the beam current (I₀) to remove the variation of the beam intensity.

Scanning transmission electron microscope (STEM). Two sample types were prepared for plane-view STEM imaging of transferred TiO₂ NM on a carbon TEM grid and cross-sectional STEM imaging of transferred TiO2 NM and VO2/TiO2 hetero-NM on SiO₂/Si. For the plane-view observation, the TiO₂ film was directly transferred onto the carbon TEM grid; TiO2/VO2/TiO2 epitaxial heterostructure was physically attached to the PDMS with the carbon TEM grid and immersed in a dilute H2O2 solution. After selective oxidation and dissolution of the VO2 sacrificial layer, the single-crystal TiO2 NM was naturally released and attached to the carbon TEM grid (Supplementary Fig. 4). For the cross-sectional observation, the transferred TiO2 NM and VO2/TiO2 hetero-NM on SiO2/Si substrates were prepared by a focused ion beam (FIB) system (Helios G3, FEI), in which the samples were thinned by a Ga ion beam. The atomic structures were observed using a STEM (JEOL ARM 200 F, JEOL Ltd., Japan) with a fifth-order aberration corrector (ASCOR, CEOS GmbH, Heidelberg, Germany); the probe diameter and convergence angle of the beam were ~0.7 Å and ~27 mrad under an acceleration voltage of 200 kV, respectively. The collection semi-angles of the detectors for HAADF imaging were 54-210 mrad, and those for ABF imaging were 8-16 mrad to detect light elements (i.e., oxygen). The obtained STEM images were local difference filtered to reduce background noise (HREM Research Inc., Japan).

Resistivity measurements. The resistivity (ρ) was measured in the van der Pauw geometry during heating and cooling from 260 to 380 K by using a chamber probe station equipped with a Hall measurement system and a temperature control system. The resistivity change over the MIT ($\Delta\rho/\rho$) was defined as $\Delta\rho/\rho = (\rho_{T_{\rm MIT}} - 15K - \rho_{T_{\rm MIT}} + 15K)/\rho_{T_{\rm MIT}} + 15K$. The fitting of the derivative of log₁₀ (ρ) as a function of temperature (K) was performed based on the Gaussian function. $T_{\rm h}$ and $T_{\rm c}$ were then defined at the peak position of the Gaussian and the transition width was calculated from the difference ($\Delta H = T_{\rm h} - T_{\rm c}$). The transition sharpness $\Delta T_{\rm h}$ and $\Delta T_{\rm c}$ was estimated from the full-width at half-maximum of the Gaussian peak.

Data availability

All relevant data within the article are available from the corresponding authors on reasonable request.

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Author contributions

J.S., D.K.L., and Y.P. conceived the idea and designed the study with assistance from C.-B.E.; D.K.L. and Y.P. developed the fabrication process of freestanding nanomembranes, performed the growth of heterostructures, XRD, XAS, SEM, EBSD, AFM, and electrical transport measurement with assistance from J.P. and Y.K.; H.S., G.-Y.K., and S.-Y.C. characterized the nanomembranes by STEM; Y.K. assisted synchrotron spectroscopy measurement; J.S., D.K.L., Y.P., S.-Y.C., and H.S. wrote the manuscript and all authors commented on it; J.S. directed the overall research.

Competing interests

The authors declare the following competing financial interest(s): D.K.L., Y.P., H.S., S.-Y.C., and J.S. are co-inventors on a patent application based on the results of this work filed by Pohang University of Science and Technology.

Additional information

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