



BRIEF COMMUNICATION

## Two new secondary metabolites from a mangrove-derived fungus *Cladosporium* sp. JS1-2

Meng Bai<sup>1,2</sup> · Cai-Juan Zheng<sup>1,2</sup> · De-Qing Tang<sup>2</sup> · Fan Zhang<sup>2</sup> · Hai-Yang Wang<sup>2</sup> · Guang-Ying Chen<sup>1,2</sup>

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### Abstract

One new pentenoic acid derivative, named 1,1'-dioxine-2,2'-dipropionic acid (**1**) and a new natural product, named 2-methylacetate-3,5,6-trimethylpyrazine (**2**), along with six known compounds (**3–8**), were obtained from the *Cladosporium* sp. JS1-2, an endophytic fungus isolated from the mangrove *Ceriops tagal* collected in South China Sea. Their structures were elucidated by detailed analysis of comprehensive spectroscopic data, and the structure of **1** was further determined by X-ray diffraction analyses. <sup>13</sup>C NMR chemical shifts of structure **2** was further determined by GIAO based <sup>13</sup>C NMR chemical shifts calculations. Compounds **1–4** and **6** showed growth inhibition activities against newly hatched larvae of *Helicoverpa armigera* Hubner with the IC<sub>50</sub> values ranging from 100 to 150 μg ml<sup>-1</sup>. Compounds **1**, **2**, **4**, **6** and **7** showed moderate antibacterial activities against *Staphylococcus aureus* with the MIC values of 25.0, 12.5, 6.25, 1.25, and 6.25 μg ml<sup>-1</sup>, respectively.

During the last decade, marine-derived fungi have proven to be a prolific source of structurally novel and biologically active natural compounds, which gained considerable attention [1, 2]. Especially the fungus isolated from mangrove could produce structurally novel and bioactive compounds, such as antiosteoporotic citrofulvicin [3], anti-inflammatory chrysogenester [4], antibacterial brocapyrrozin A [5], and antiviral simpterpenoid A [6]. In our search for new bioactive natural products from mangrove-derived fungi in the South China Sea, we have found many bioactive compounds, including cytotoxic indole diterpenes,

chlorinated xanthenes, anthraquinone derivatives, dihydroisocoumarins and isocoumarins [7–11]. In our ongoing search for new bioactive compounds from mangrove-derived fungus, an endophytic fungus *Cladosporium* sp. JS1-2 obtained from the mangrove *Ceriops tagal*, was selected for further research because its EtOAc extract showed insecticidal activity against newly hatched larvae of *Helicoverpa armigera* Hubner. One new pentenoic acid derivative 1,1'-dioxine-2,2'-dipropionic acid (**1**) and a new natural product 2-acetate-3,5,6-trimethylpyrazine (**2**) and six known compounds (**3–8**) were isolated from the EtOAc extract of the fungus (Fig. 1). In this report, we described the isolation, structure elucidation, antimicrobial and insecticidal activities of all compounds.

The fungus *Cladosporium* sp. JS1-2 was isolated from the mangrove plant *C. tagal*, collected from Dongzhai-gang of Hainan Province in China in July, 2016. The strain was deposited in the Key Lab of Tropical Medicinal Resource Chemistry of Ministry of Education, College of Chemistry and Chemical Engineering, Hainan Normal University, Haikou, P.R. China. The fungus was identified according to its morphological characteristics and a molecular biological protocol by 18S rRNA amplification and sequencing of the ITS region. The sequence data have been submitted to GenBank, with an accession number MK234874, and the fungal strain was identified as *Cladosporium* sp.

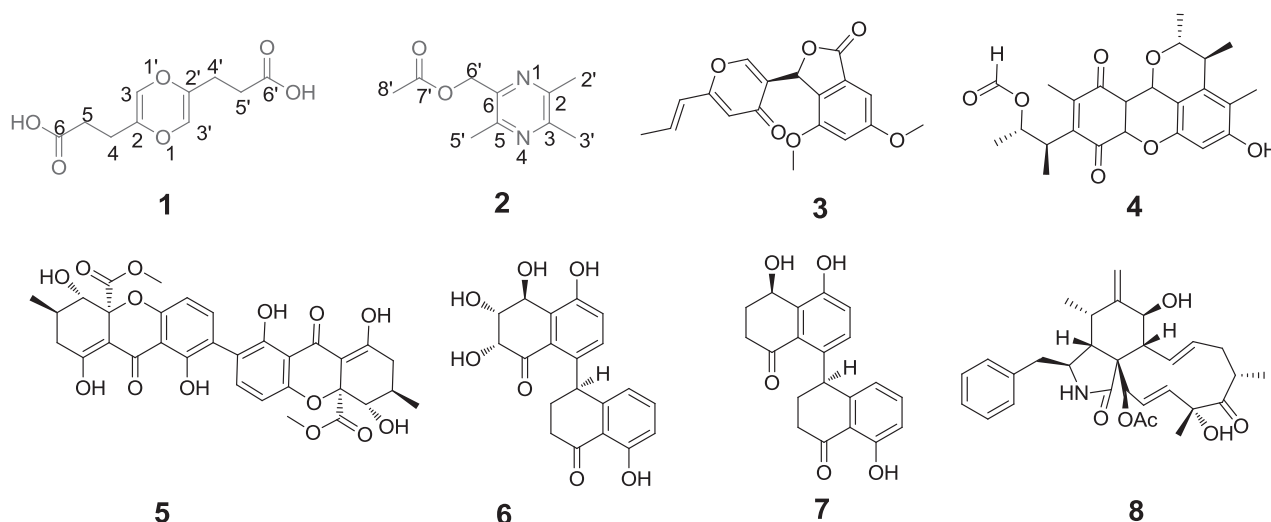
These authors contributed equally: Meng Bai, Cai-Juan Zheng

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✉ Guang-Ying Chen  
chgying123@163.com

<sup>1</sup> Key Laboratory of Tropical Medicinal Resource Chemistry of Ministry of Education, College of Chemistry and Chemical Engineering, Hainan Normal University, Haikou 571158, People's Republic of China

<sup>2</sup> Key Laboratory of Tropical Medicinal Plant Chemistry of Hainan Province, Hainan Normal University, Haikou 571158, People's Republic of China



**Fig. 1** The structure of compounds **1–8**

The fungus *Cladosporium* sp. JS1-2 was cultured in 200 ml of potato dextrose broth at 30 °C on a rotary shaker (120 rpm) for 5 days to prepare the seed culture. Large-scale fermentation was carried out in 200-Erlenmeyer flasks (1000 ml), each containing potato dextrose broth (8 g), and purified water (200 ml), which were soaked overnight before autoclaving at 121 °C for 20 min. These Erlenmeyer flasks were added to 3 ml seed broth after cooling to room temperature, and maintained at room temperature for 30 days in stationary phase.

The fungal cultures was extracted with EtOAc (3 × 10 L, 24 h each), which were filtered through cheesecloth. The extracts were concentrated in vacuo to yield an oily residue (25.3 g), which was subjected to silica gel CC (petroleum ether, EtOAc v/v, gradient 100:0–0:100) to generate six fractions (Fr. A–Fr. E). Fr. C was chromatographed on a silica gel column by stepwise-gradient elution using petroleum ether/ethyl acetate from 100:0 to 0:100. The fractions were further purified with preparative HPLC with an isocratic solvent system of 0.1% formic acid in 30% acetonitrile-water at a flow rate of 2 ml min<sup>-1</sup> to obtain compounds **1** (7 mg) and **4** (3.1 mg). Fr. D was further purified by Sephadex LH20 column (CHCl<sub>3</sub>/MeOH v/v, 1:1) and eluted with preparative HPLC with an isocratic solvent system of 0.1% formic acid in 40% acetonitrile-water at a flow rate of 2 ml min<sup>-1</sup> to obtain compounds **2** (3.4 mg) and **6** (3.7 mg). Like the above method, we obtained **3** (4.0 mg) and **7** (4.4 mg) from Fr. B, and acquired **5** (3.3 mg) and **8** (3.9 mg) from Fr. E.

Compound **1** was obtained as colorless crystals and had a molecular formula of C<sub>10</sub>H<sub>12</sub>O<sub>6</sub> as determined by the HR-ESI-MS at *m/z* 251.0529 [M + Na]<sup>+</sup> (calcd. for 251.0526), indicating five degrees of unsaturation. The <sup>13</sup>C NMR and DEPT 135 spectra data (Table 1) exhibited 5 carbon signals,

**Table 1** <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR (100 MHz) data of compounds **1** and **2**

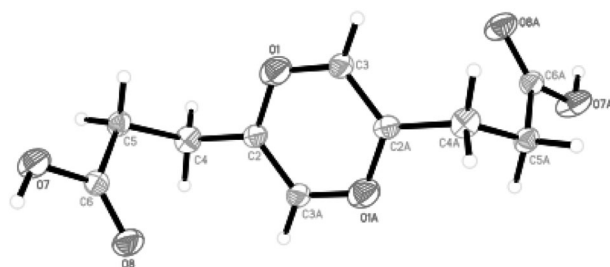
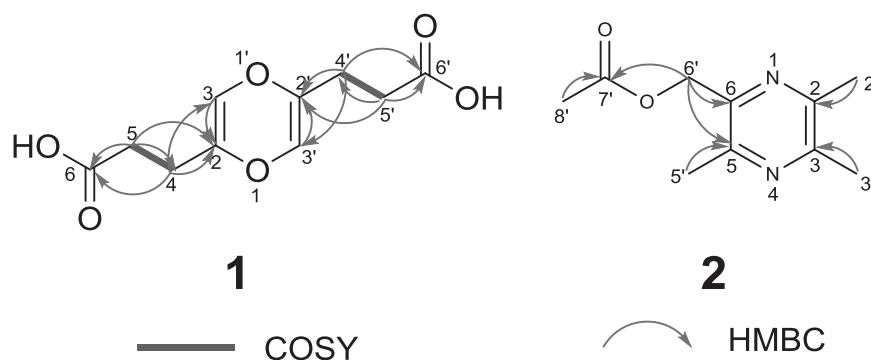
| Position | 1 <sup>a</sup>        |                          | 2 <sup>b</sup>        |                          |
|----------|-----------------------|--------------------------|-----------------------|--------------------------|
|          | δ <sub>C</sub> , type | δ <sub>H</sub> (J in Hz) | δ <sub>C</sub> , type | δ <sub>H</sub> (J in Hz) |
| 1        |                       |                          |                       |                          |
| 2        | 153.0, C              |                          | 148.4, C              |                          |
| 3        | 143.3, CH             | 8.45, s                  | 150.8, C              |                          |
| 4        | 29.1, CH <sub>2</sub> | 2.95, t (7.6)            |                       |                          |
| 5        | 32.2, CH <sub>2</sub> | 2.67, t (7.6)            | 144.6, C              |                          |
| 6        | 173.7, C              |                          | 148.5, C              |                          |
| 1'       |                       |                          |                       |                          |
| 2'       | 153.0, C              |                          | 20.0, CH <sub>3</sub> | 2.44, s                  |
| 3'       | 143.3, CH             | 8.45, s                  | 21.2, CH <sub>3</sub> | 2.42, s                  |
| 4'       | 29.1, CH <sub>2</sub> | 2.95, t (7.6)            |                       |                          |
| 5'       | 32.2, CH <sub>2</sub> | 2.67, t (7.6)            | 21.0, CH <sub>3</sub> | 2.43, s                  |
| 6'       | 173.7, C              |                          | 64.4, CH <sub>2</sub> | 5.12, s                  |
| 7'       |                       |                          | 170.1, C              |                          |
| 8'       |                       |                          | 20.5, CH <sub>3</sub> | 2.06, s                  |

<sup>a</sup>DMSO-*d*<sub>6</sub>

<sup>b</sup>CD<sub>3</sub>OD

including one carbonyl carbon (δ<sub>C</sub> 173.3), one double bond (δ<sub>C</sub> 153.0 and 143.3) and two methylene carbons (δ<sub>C</sub> 32.2 and 29.1). Since one carbonyl group and one double bond accounted for two out of five degrees of unsaturation, implying that **1** was a dimer, and the remaining one degree of unsaturation were assumed for the presence of one ring system in **1**. The <sup>1</sup>H NMR spectrum of **1** (Table 1) in DMSO-*d*<sub>6</sub> showed one downfield proton signal at δ<sub>H</sub> 8.45 (1 H, s), two methylene proton signals at δ<sub>H</sub> 2.95 (2 H, t, *J* = 7.6 Hz), and 2.67 (2 H, t, *J* = 7.6 Hz). The <sup>1</sup>H-<sup>1</sup>H COSY correlation of H-4 and H-5, together with the HMBC

**Fig. 2**  $^1\text{H}$ - $^1\text{H}$  COSY (in blue lines) and HMBC (in red lines) correlations of **1–2**

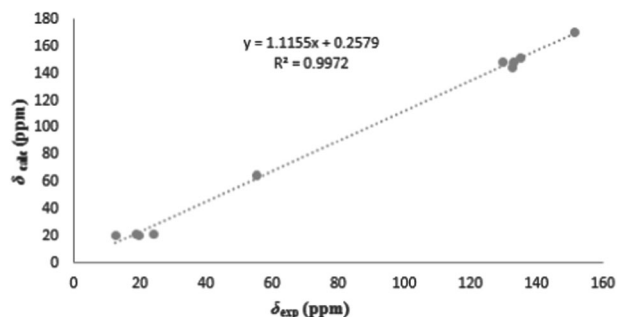


**Fig. 3** X-ray structure of Compound **1**

correlations (Fig. 2) from H-3 to C-2; from H-4 to C-2, C-3, C-5, and C-6; from H-5 to C-2, C-4, and C-6, indicated that **1** comprised two (*Z*)-5-hydroxypent-4-enoic acid units and it was a symmetrical dimer. The structure of **1** was also determined by X-ray diffraction analyses (Fig. 3). Hence, compound **1** was determined as 1,1'-dioxine-2,2'-dipropionic acid.

Compound **2** was isolated as yellow powder and gave a HR-ESI-MS ion peak at  $m/z$  195.1133  $[\text{M} + \text{H}]^+$  (calcd for 195.1128), corresponding to a molecular formula of  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}_2$  with five degrees of unsaturation. The  $^{13}\text{C}$  NMR and DEPT 135 spectra (Table 1) data exhibited 10 carbon signals, including one ester carbonyl carbon ( $\delta_{\text{C}}$  170.1), two double bonds ( $\delta_{\text{C}}$  150.8, 148.5, 148.4, and 144.6), one oxygenated methylene carbon ( $\delta_{\text{C}}$  64.4), and four methyl carbons ( $\delta_{\text{C}}$  21.2, 21.0, 20.5, and 20.0). The  $^1\text{H}$  NMR spectrum of **2** (Table 1) showed one methylene proton signal at  $\delta_{\text{H}}$  5.12 (2 H, s, H-6'), four methyl protons at  $\delta_{\text{H}}$  2.44 (3 H, s, H-2'), 2.43 (3 H, s, H-5'), 2.42 (3 H, s, H-3'), and 2.06 (3 H, s, H-8'). The HMBC correlations (Fig. 2) from H-8' to C-7'; from H-6' to C-5, C-6, and C-7'; from H-2' to C-2; from H-3' to C-3; from H-5' to C-5, primary ascertained the structure of compound **2**.

According to the NMR spectra of compound **2**, we were uncertain the location of C-2 ( $\delta_{\text{C}}$  148.4) and C-3 ( $\delta_{\text{C}}$  150.8). To further verify the structure, a calculation of the  $^{13}\text{C}$  NMR chemical shifts of structure **2** at the B3LYP/6-31G (d)//B3LYP/6-31G (d, p) level with the PCM model in DMSO was obtained [12], and the calculated chemical shifts agreed well with the experimental data (Fig. 4), of which the Fig. 4



**Fig. 4** Regression analysis of experimental versus calculated  $^{13}\text{C}$  NMR chemical shifts of compound **2** [at the B3LYP/6-31G(d)//B3LYP/6-31G(d, p)level]

showed a correlation coefficient ( $R^2$ ) of 0.9972, indicating that the calculation of the  $^{13}\text{C}$  NMR chemical shifts of structure **2** was suitable. Finally **2** was determined as 2,3,5-trimethyl-6-methylacetate-pyrazine. Interestingly, compound **2** was firstly reported as a synthetic intermediate for the synthesis of piperlongumine-ligustrazine hybrids [13] and **2** was isolated from a natural source for the first time.

The structures of known compounds **3–8** were identified by comparison of their  $^1\text{H}/^{13}\text{C}$  NMR spectra with those in the literature as vermistatin (**3**) [14], citrinin H1 (**4**) [15], secalononic acid D (**5**) [16], ladosporol E (**6**) [17], cladosporol C (**7**) [18], and cytochalasin D (**8**) [19].

All compounds were evaluated for their antibacterial activities against five terrestrial pathogenic bacteria, including *S. aureus* (ATCC 27154), *Staphylococcus albus* (ATCC 8799), *B. cereus* (ATCC 11778), *Escherichia coli* (ATCC 25922), and *Micrococcus luteus* (ATCC 10240) by the microplate assay method [20]. The result (Table 2) showed that **1**, **2**, **4**, **6** and **7** showed antibacterial activities against *S. aureus* with the MIC values of 25.0, 12.5, 6.25, 1.56 and 6.25  $\mu\text{g ml}^{-1}$ , respectively. Ciprofloxacin was used as positive control with the MIC value of 0.39  $\mu\text{g ml}^{-1}$ .

Compounds **1–8** were also evaluated for growth inhibition activity against newly hatched larvae of *H. armigera* Hubner [21]. Compounds **1–4** and **6** showed growth inhibition activities against newly hatched larvae of *H. armigera* Hubner with the  $\text{IC}_{50}$  values of 150, 100, 150, 100 and

**Table 2** Antibacterial activity for compounds 1–8

| Compound                   | MIC ( $\mu\text{g ml}^{-1}$ ) |                |                 |                  |                  |
|----------------------------|-------------------------------|----------------|-----------------|------------------|------------------|
|                            | <i>S. aureus</i>              | <i>E. coli</i> | <i>S. albus</i> | <i>B. cereus</i> | <i>M. luteus</i> |
| <b>1</b>                   | 25.0                          | 25.0           | >25.0           | 12.5             | >25.0            |
| <b>2</b>                   | 12.5                          | >25.0          | >25.0           | >25.0            | >25.0            |
| <b>3</b>                   | 25.0                          | >25.0          | >25.0           | 25               | >25.0            |
| <b>4</b>                   | 6.25                          | 12.5           | >25.0           | 12.5             | >25.0            |
| <b>5</b>                   | >25                           | >25.0          | >25.0           | >25.0            | >25.0            |
| <b>6</b>                   | 1.56                          | >25.0          | >25.0           | >25.0            | 12.5             |
| <b>7</b>                   | 6.25                          | >25.0          | >25.0           | >25.0            | 12.5             |
| <b>8</b>                   | 25.0                          | >25.0          | >25.0           | >25.0            | >25.0            |
| Ciprofloxacin <sup>a</sup> | 0.39                          | 0.19           | 0.39            | 0.39             | 0.39             |

<sup>a</sup>Ciprofloxacin was used as a positive control

150  $\mu\text{g ml}^{-1}$ , respectively. Azadirachtin was used as positive control with the  $\text{IC}_{50}$  value of 25  $\mu\text{g ml}^{-1}$ .

**Compound 1:** colorless crystals. mp. 108–110 °C; IR (KBr)  $\nu_{\text{max}}$  3536, 1750, 1315  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR see Table 1; HR-ESI-MS  $m/z$  251.0529  $[\text{M} + \text{Na}]^+$  (calcd. for  $\text{C}_{10}\text{H}_{12}\text{O}_6\text{Na}^+$ , 251.0526).

**Compound 2:** yellow powder. IR (KBr)  $\nu_{\text{max}}$  1598, 1445, 873  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR see Table 1; HR-ESI-MS  $m/z$  195.1133  $[\text{M} + \text{H}]^+$  (calcd. for  $\text{C}_{10}\text{H}_{15}\text{N}_2\text{O}_2^+$ , 195.1128).

Crystal data for **1**:  $\text{C}_{10}\text{H}_{12}\text{O}_6$ , Mr = 600.60, monoclinic,  $a = 5.6567$  (3) Å,  $b = 13.2069$  (5) Å,  $c = 8.0189$  (5) Å,  $\alpha = 90^\circ$ ,  $\beta = 90^\circ$ ,  $\gamma = 90^\circ$ ,  $V = 523.64$  (5) Å<sup>3</sup>, space group  $P2_12_12_1$ ,  $Z = 2$ ,  $D_x = 1.447$   $\text{mg/mm}^3$ ,  $\mu$  (Cu  $K\alpha$ ) = 0.816  $\text{mm}^{-1}$ , and  $F(000) = 240$ . Independent reflections: 924 ( $R_{\text{int}} = 0.0619$ ). The final  $R_1$  values were 0.0566,  $wR_2 = 0.1185$  ( $I > 2\sigma(I)$ ).

In summary, we isolated and characterized one new pentenoic acid derivative **1** and a new natural product **2**, along with other six known compounds (**3–8**) from the fungus *Cladosporium* sp. JS1-2. Compounds **1** and **2** showed insecticidal activities against *H. armigera* Hubner and antimicrobial activities against *S. aureus*.

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## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest.

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