

Stable-Isotope Dilution Analysis of D- and L-2-Hydroxyglutaric Acid: Application to the Detection and Prenatal Diagnosis of D- and L-2-Hydroxyglutaric Acidemias

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ABSTRACT. A stable-isotope dilution assay has been developed for quantitation of D- and L-2-hydroxyglutaric acids in physiologic fluids. D- and L-2-hydroxyglutaric acids are separated as the O-acetyl-di-(D)-2-butyl esters. The method uses D,L-[3,3,4,4-²H₄]-2-hydroxyglutaric acid as internal standard with ammonia chemical ionization, selected ion monitoring gas chromatography-mass spectrometry. For 13 patients with L-2-hydroxyglutaric aciduria, the concentrations of L-2-hydroxyglutaric acid were urine, 1283 ± 676 mmol/mol creatinine (range, 332–2742; n = 12 patients); plasma, 47 ± 13 μmol/L (range, 27–62; n = 8); cerebrospinal fluid, 62 ± 30 μmol/L (range, 34–100; n = 6). In a child with D-2-hydroxyglutaric aciduria, the levels of D-2-hydroxyglutaric acid were urine, 1565 ± 847 mmol/mol creatinine (range, 729–2668; n = 4); plasma, 61 ± 14 μmol/L (range, 46–73; n = 3); cerebrospinal fluid, 15 and 25 μmol/L (n = 2). Control concentrations of D- and L-2-hydroxyglutaric acids were (D:L): urine (n = 18), 6.0 ± 3.6 mmol/mol creatinine (range, 2.8–17); 6.0 ± 5.4 (range, 1.3–19); plasma (n = 10), 0.7 ± 0.2 μmol/L (range, 0.3–0.9); 0.6 ± 0.2 (range, 0.5–1.0); cerebrospinal fluid (n = 10), 0.1 ± 0.1 μmol/L (range, 0.07–0.3); 0.7 ± 0.6 (range, 0.3–2.3). Investigation of control amniotic fluid (n = 10) revealed the following values (D:L): 1.2 ± 0.4 μmol/L (range, 0.6–1.8); 4.0 ± 0.7 (range, 3.1–5.2), suggesting the feasibility of prenatal diagnosis in families at risk. (*Pediatr Res* 34: 277–280, 1993)

Abbreviations

2-HG, 2-hydroxyglutaric
CSF, cerebrospinal fluid
GC-MS, gas chromatography-mass spectrometry
TMS, trimethylsilyl

2-HG acid is detected in normal human urine and is elevated in the urine of patients with multiple acyl-CoA dehydrogenase deficiency (glutaric aciduria type II) (1). Despite its occurrence in D and L configurations, there has been no systematic investi-

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gation on the concentrations of D- and L-2-HG acids in human physiologic fluids. The separation of D- and L-2-HG acid requires the expensive preparation of chiral derivatives using (D)-2-butanol, which may account for the lack of detailed analyses of the different configurations in humans (2, 3).

Since the description of the index patients with D- and L-2-HG aciduria (2, 3), a number of patients with 2-HG aciduria have been reported. Barth *et al.* (4) reported eight patients with cerebellar dysfunction, extrapyramidal signs, mental regression, and seizure disorders. Magnetic resonance imaging findings suggested a spongiform leukodystrophy. L-2-HG acid concentration was elevated in urine, plasma, and CSF without additional organic aciduria (5). A patient with D-2-HG aciduria manifested a protein-losing enteropathy and normal cognitive development (3). D-2-HG aciduria was reported in another patient who manifested an early-onset, severe seizure disorder with hypotonia and cortical blindness (6). In this patient, D-2-HG acid was increased in urine, plasma, and CSF; conversely, in the patient described by Chalmers *et al.* (3), D-2-HG acid was not elevated in blood, and CSF was not investigated. Haworth *et al.* (7) detected 2-HG aciduria in a growth-retarded newborn female with dysmorphic facial features, hepatosplenomegaly, and tricuspid regurgitation, but there was no indication of which isomer was excreted.

The existence of isolated diseases in which D- and L-2-HG acids accumulate suggests the need for an accurate method for quantitation of these metabolites in human physiologic fluids. We have developed a sensitive stable-isotope dilution assay for quantitation of D- and L-2-HG acids using ammonia chemical ionization, combined with GC-MS. Using this procedure, we quantitated the concentrations of D- and L-2-HG acids in urine, plasma, and CSF from controls, 13 patients with L-2-HG aciduria, and one patient with D-2-HG aciduria. To facilitate eventual prenatal diagnosis of these disorders, concentrations of D- and L-2-HG acids were determined in control amniotic fluid. This report describes the procedures and summarizes the results of these investigations.

MATERIALS AND METHODS

Patients. Clinical details on eight of 13 L-2-HG aciduria patients whom we investigated and on the D-2-HG aciduria patient have been reported (4–6, 8). An additional five patients with L-2-HG aciduria have not been reported. The presentation in a 12-year-old boy included mental retardation, seizures, pyramidal signs, brain atrophy, and calcification of the frontal lobe. Clinical evaluation of a 20-year-old male at 6 y revealed mental retardation (IQ 60), cerebellar ataxia, unsteadiness, dysarthria, macrocephaly, scoliosis, seizures, and head tremor. CSF lysine concentration was elevated with normal plasma lysine levels. His sister

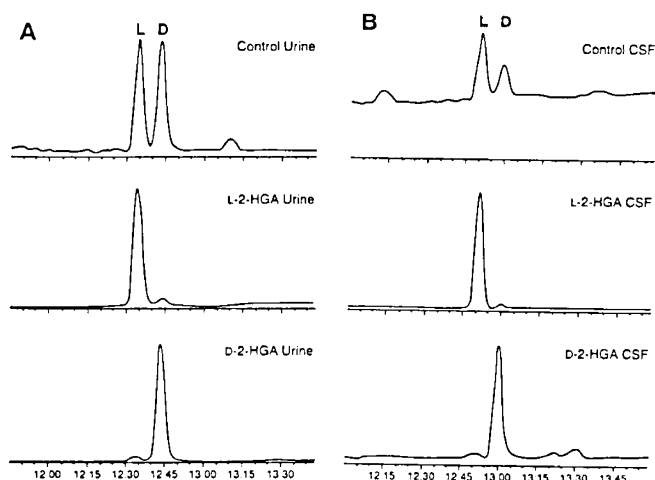


Fig. 1. Mass fragmentograms (m/z 303) of D- and L-2-hydroxyglutaric acids in urine (A) and CSF (B) as the (D and L)-O-acetyl-2-HG acid di-(D)-2-butyl esters. Top, control; middle, a patient with L-2-hydroxyglutaric aciduria; bottom, a patient with D-2-hydroxyglutaric aciduria. Peak heights are normalized to full-scale deflection. For figure clarity, m/z 307 was omitted.

was clinically unaffected. Computed tomography and magnetic resonance imaging revealed ventricular enlargement, cortical atrophy, and subcortical white matter lesions. Two other patients, male and female siblings of Tunisian descent, died at 30 and 6 y of age, respectively (Dr. P. G. Barth, personal communication). Complete clinical details will be presented elsewhere. The remaining patient was a 21-y-old female for whom clinical details were unavailable.

Synthesis of internal standard. D,L-[3,3,4,4- $^2\text{H}_4$]-2-HG acid (zinc salt) was prepared by reduction of commercially available [$^2\text{H}_4$]-2-oxoglutaric acid by reaction with excess zinc. To a stirred suspension of zinc powder (0.12 g) in ice-cold water (1 mL), a solution of [3,3,4,4- $^2\text{H}_4$]-2-oxoglutaric acid (0.15 g, 98.7 atom percent ^2H , Merck Sharp & Dohme, Montreal, Canada) in ice-cold water (0.5 mL) was slowly added. After stirring the cooled mixture for 30 min, a light precipitate was formed. Excess zinc was separated by centrifuging for 2 min at 600 rpm and removal of the turbid water phase. The zinc residue was repeatedly washed with water (1-mL portions) and centrifuged until the water phase was clear. The product was further precipitated by adding three volumes of acetone to the combined water phases. After centrifuging for 5 min at 2000 rpm and removal of the solution, the product was washed twice with acetone (10 mL) and dried under a stream of nitrogen yielding a white powder (0.18 g, 85% yield). The tri-TMS derivative was pure by gas chromatography, D and L isomers were present in equal amounts (for separation conditions, see below), and isotopic purity was >98 atom % ^2H .

Derivatization. D- and L-2-HG acids in physiologic fluids were separated as the (D and L)-O-acetyl-2-HG acid di-(D)-2-butyl esters (2). Standards (D- and L-2-HG acids, Sigma Chemical Co., St. Louis, MO) or samples (0.1–0.5 mL, or appropriate dilution of patient specimens) received 20 nmol D,L-[3,3,4,4- $^2\text{H}_4$]-2-HG acid. After volume adjustment to 1 mL, samples were acidified to pH 1–2 with 6 N HCl and extracted three times with 4 mL of ethyl acetate. The solvent fractions were pooled, dried over sodium sulfate, and taken to dryness under a gentle stream of nitrogen. Initially, tri-TMS derivatives of 2-HG acid were made according to established methods (1). Butylation was achieved by addition of 0.5 mL (D)-2-butanol (Fluka, Munich, Germany) and 0.05 mL 12 N HCl to the residue, and the samples were heated for 3 h at 90°C. After cooling, samples were twice extracted with 4 mL of hexane, dried over sodium sulfate, and the solvent evaporated under nitrogen. For acetylation, 0.3 mL of pyridine and 0.3 mL of acetic anhydride were added to the residue with heating for 1 h at 80°C. After evaporation, the residues were

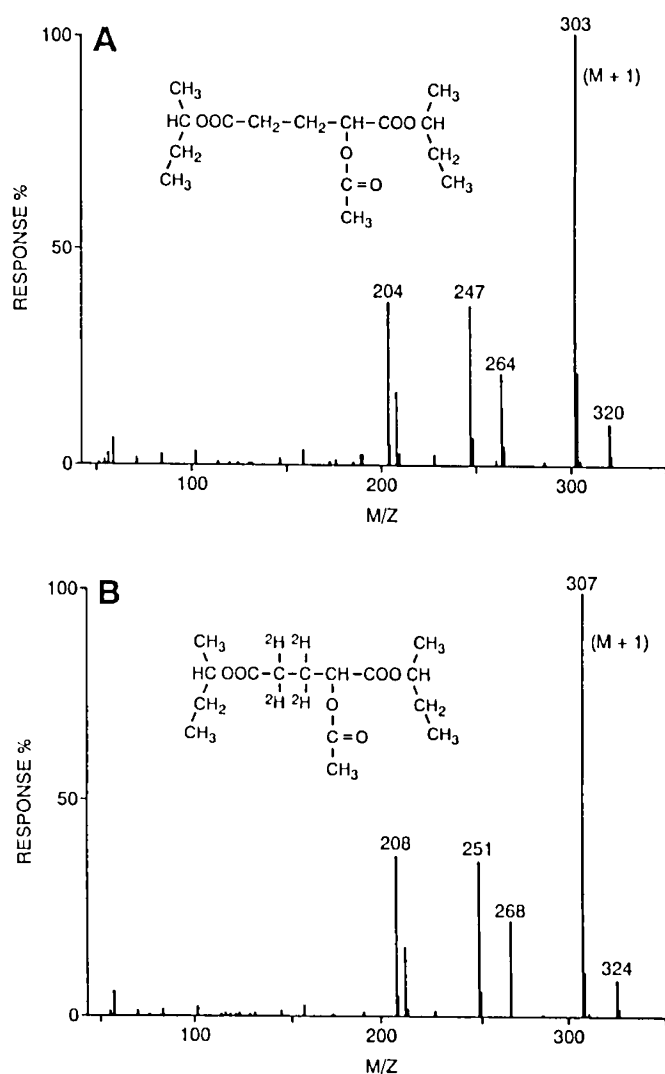


Fig. 2. Ammonia chemical ionization mass spectra of (A) [$^2\text{H}_0$]-2-hydroxyglutaric acid and (B) [$^2\text{H}_4$]-2-hydroxyglutaric acid recorded as the O-acetyl-di-(D)-2-butyl esters. Selected-ion monitoring was performed at m/z 303 and m/z 307.

redissolved in 0.1 mL of chloroform in preparation for GC-MS analysis.

GC-MS. Gas chromatographic separation was achieved on a CP Sil 88 fused silica capillary column (Chrompack, Middelburg, The Netherlands), 25 m \times 0.32 mm inner diameter, film thickness 0.2 μm , which was directly inserted into the ion source of the mass spectrometer. Helium was used as carrier gas with a flow rate of 30 cm/s. The column oven was maintained for 3 min at 120°C, then programmed to 175°C at a rate of 4°C/min and finally programmed to 220°C at a rate of 20°C/min. The samples were introduced using a solid injector (glass falling needle) operated at 300°C.

GC-MS analysis was performed on a Kratos MS-80 instrument (Kratos Ltd., Manchester, UK) in the positive chemical ionization mode. The chemical ionization reagent gas was ammonia. Ionization was initiated with 50 eV electrons with an emission current of 1.5 mA. The source temperature was 230°C. Selected ion monitoring for quantitation was performed by recording the ions at m/z 303 for 2-HG acid and 307 for the [$^2\text{H}_4$]-2-HG acid analogue. The dwell time was 50 ms, and the resolution was 1000 (10% valley). All areas of the mass-spectrometric peaks were integrated using the computer with operator selection of the baseline points. Aqueous standard solutions containing 20 nmol of the internal standard D,L-[$^2\text{H}_4$]-2-HG acid and variable amounts (1–30 nmol) of D- or L-2-HG acid were prepared and

Table 1. D- and L-2-HG acid concentrations in physiologic fluids from controls and patients determined by stable-isotope dilution assay as O-acetyl-Diz(D)-2-butyl esters*

Subject	Body fluid	D Isomer	L Isomer
Controls	Urine (<i>n</i> = 18) (age 10 d–45 y)	6.0 ± 3.6 (2.8–17)	6.0 ± 5.4 (1.3–19)
	Plasma (<i>n</i> = 10) (age 1–60 y)	0.7 ± 0.2 (0.3–0.9)	0.6 ± 0.2 (0.5–1.0)
	CSF (<i>n</i> = 10) (age 2 mo–41 y)	0.1 ± 0.1 (0.07–0.3)	0.7 ± 0.6 (0.3–2.3)
	Amniotic fluid (<i>n</i> = 10)	1.2 ± 0.4 (0.6–1.8)	4.0 ± 0.7 (3.1–5.2)
L-2-HG aciduria patients	Urine (<i>n</i> = 9) (age 9–39 y)	12 ± 5.9 (4.1–22)	1283 ± 676 (332–2742)
	Plasma (<i>n</i> = 8)	0.8 ± 0.4 (0.3–1.3)	47 ± 13 (27–62)
	CSF (<i>n</i> = 6)	0.4 ± 0.2 (0.2–0.6)	62 ± 30 (34–100)
D-2-HG aciduria patient†	Urine (age 3 mo)	1565 ± 847 (729–2668)	12 ± 4.4 (7.5–16)
	Plasma	61 ± 14 (46–73)	1.0 ± 0.7 (0.3–1.7)
	CSF	15, 25	0.4, 0.6

* Amniotic fluid specimens were obtained at 16–18 wk gestation. Units: urine, mmol/mol creatinine; plasma, CSF, and amniotic fluid, μmol/L. Data are presented as mean ± 1 SD, with range in parentheses; *n* values represent the number of individuals studied.

† Four samples of urine, three samples of plasma, and two CSF samples were available.

Table 2. Total 2-HG acid concentration in physiologic fluids from controls and patients determined by stable-isotope dilution assay as Tri-TMS esters*

Subject	Body fluid	2-HG acid
Controls	Urine (<i>n</i> = 31) (age newborn–30 y)	14.2 ± 11.3 (1.6–40)
	Plasma (<i>n</i> = 8) (age 3 mo–53 y)	1.9 ± 0.4 (1.5–2.5)
	CSF (<i>n</i> = 5) (age 3 mo–4 y)	1.9 ± 0.2 (1.8–2.4)
	Amniotic fluid (<i>n</i> = 15)	4.2 ± 1.4 (1.8–6.5)
L-2-HG aciduria patients	Urine (<i>n</i> = 6) (age 9–39 y)	1594 ± 920 (810–4210)
	Plasma (<i>n</i> = 5)	35 ± 11 (16–46)
	CSF (<i>n</i> = 3)	38 ± 14 (23–49)

* Amniotic fluid specimens were obtained at 16–18 wk gestation. Units: urine, mmol/mol creatinine; plasma, CSF, and amniotic fluid, μmol/L. Data are presented as mean ± 1 SD, with range in parentheses; *n* values represent the number of individuals studied. Body fluids from the D-2-HG aciduria patient were not investigated as the tri-TMS esters.

carried through the entire procedure to establish calibration curves. Linear regression analysis was used to calculate metabolite concentrations in physiologic fluids.

RESULTS

D- and L-2-HG acid, as the O-acetyl-di-(D)-2-butyl esters, could be distinguished on gas chromatographic analysis with near-baseline separation, and patients with D- or L-2-HG aciduria could readily be identified (Fig. 1). Calibration curves yielded straight lines that went through the origin ($r > 0.995$, $n = 10$). Selected ion monitoring at m/z 303 and 307 for the $[M + H]^+$ for 2H_0 and 2H_4 analogues, respectively, allowed accurate quantitation of the concentration of each metabolite in the different body fluids investigated (Fig. 2). The absence of fragments at m/z 303–306 (Fig. 2B) verified that sample preparation and heating in the injector at 300°C did not lead to deuterium exchange. In control body fluids (Table 1), urine and plasma

contained nearly equal amounts of D- and L-2-HG acids, whereas the ratio of L- to D-2-HG acid was higher in CSF.

For L-2-HG aciduria patients, there was a relatively consistent increase of L-2-HG acid in comparison to controls for all physiologic fluids studied, from 80–200-fold, without significant increase in excretion of the D isomer. In patients with L-2-HG aciduria, we detected nearly equal concentrations of the L-isomer in plasma and in CSF, whereas Barth *et al.* (4) suggested an increased ratio of the L isomer between CSF and plasma. In physiologic fluids obtained from first-degree relatives of the two sets of male siblings with L-2-HG aciduria whom we investigated, normal concentrations of D- and L-2-HG acids were detected (data not shown). In a patient with D-2-HG aciduria, the range of increase of D-2-HG acid in physiologic fluids in comparison to control (*i.e.* comparison of mean to mean) was approximately 90–260-fold, with the largest increase in the urine. As for the L-2-HG aciduria patients, no increase in the D or L isomer of 2-HG acid was detected in urine from parents of the D-2-HG aciduria patient. To establish reference values for the potential prenatal diagnosis of both D- and L-2-HG aciduria, control amniotic fluid was investigated for the concentration of D- and L-2-HG acids. In 10 amniotic fluids, we detected D isomer, 1.2 ± 0.4 μmol/L (mean ± 1 SD, range 0.6–1.8) and L isomer, 4.0 ± 0.7 μmol/L (range, 3.1–5.2).

In preliminary investigations, we attempted to separate and quantify D- and L-2-HG acids as the tri-TMS derivatives using the tetradeuterated internal standard. Although accurate quantitation was achieved (Table 2), separation of the D and L isomers was not possible. Comparison of values generated using the two derivatives (Tables 1 and 2) demonstrates relatively good agreement between the two for controls and patients with L-2-HG aciduria. One notable difference is the somewhat higher estimation of total 2-HG acid concentration in control CSF using the tri-TMS derivative as compared with the O-acetyl-di-(D)-2-butyl ester analogues (Tables 1 and 2). As the tri-TMS ester derivatives, the amniotic fluid concentration of 2-HG acid ($n = 15$) was 4.2 ± 1.4 μmol/L (range, 1.8–6.5), which was comparable to individual values obtained for D- and L-2-HG acids analyzed as the O-acetyl-di-(D)-2-butyl esters. This indicates that with the stable-isotope dilution method using tetradeuterated internal standard, it should be possible to perform prenatal diagnosis for either D- or L-2-HG aciduria using tri-TMS derivatives.

DISCUSSION

This method for separation and quantitation of D- and L-2-HG acid in physiologic fluids is sensitive and accurate and for the first time provides accurate values, including control values. The use of stable-isotope dilution assay should facilitate the prenatal diagnosis of both disorders. Although L-2-HG aciduria appears to be a defined Mendelian disorder, additional patients with D-2-HG aciduria will need to be characterized to define the phenotype and the inheritance of that disorder. It is of interest that the original patient with D-2-HG aciduria (3) had no neurologic abnormalities. This raises the possibility that D-2-HG acid is unrelated to neurologic findings in the patient with D-2-HG aciduria whom we investigated. However, D-2-HG acid was not increased in blood, and CSF was not studied in the patient reported by Chalmers *et al.* (3). In the D-2-HG aciduria patient whom we investigated, the increase of CSF D-2-HG acid, in comparison to the control mean value, was elevated 200-fold, raising the possibility that D-2-HG acid may be at least partially produced in neural tissue.

This is the first study in which body fluids from most available patients with D- and L-2-HG aciduria have been studied in one laboratory using a uniform, standardized method. Little is known about the role of D- and L-2-HG acids in mammalian metabolism. In patients with multiple acyl-CoA dehydrogenase deficiency, 2-HG acid is a characteristic and highly elevated urinary metabolite, and the isomer predominantly excreted has the D configuration (9). This raises the possibility that one source of D-2-HG acid may be the mitochondrial electron transport chain. For D-2-HG acid, two mammalian pathways have been described that metabolize this compound (6). In one D-2-HG acid is converted to 2-ketoglutaric acid by D-2-HG acid dehydrogenase in the pathway of 5-aminolevulinic acid metabolism. In the second pathway, D-2-HG acid is converted to 2-ketoglutaric acid by a transhydrogenase that uses succinic semialdehyde as hydrogen acceptor (10). This reaction is nicotinamide-independent and has not been described in human tissues. The possibility that one of these reactions is abnormal in our patient with D-2-HG aciduria has not yet been explored. In those inborn errors of metabolism in which 5-aminolevulinic acid and succinic semialdehyde accumulate, including acute intermittent porphyria, tyrosinemia type I, and succinic semialdehyde dehydrogenase deficiency, 2-HG acid concentration is not elevated in urine.

Other pathways involved in the metabolism of D-2-HG acid are mainly bacterial (6), and little is known about the precursors of D-2-HG acid. Similarly, even less is known about the precursors and metabolites of L-2-HG acid (4). In the eight patients described, Barth *et al.* (4) found elevations in plasma and CSF L-lysine concentrations, but oral loading with L-lysine in three L-2-HG aciduria patients did not result in enhanced L-2-HG acid excretion. Plasma lysine concentration was normal in the patient with D-2-HG aciduria. In both disorders, pipecolic acid concen-

trations in physiologic fluids was normal, arguing against disordered lysine metabolism.

Eventually, it will be necessary to undertake studies of the metabolism of D- and L-2-HG acids in animal systems, either *in vivo* or in isolated organs, tissue homogenates or both. To this end, the use of stable isotope precursors, as described in this report, will be of value in determining metabolic routes. The report and characterization of additional patients, with accurate screening of body fluids for organic and amino acids, may provide insights into the physiologic precursors of both D- and L-2-HG acids. At present, there is no information on the concentration of D- and L-2-HG acids in mammalian tissues, and the ratio of isomers in neural and non-neural tissues is undefined. Investigation of the quantities of D- and L-2-HG acids in different mammalian tissues will provide a basis from which to decide which tissue may be optimal for metabolic studies. Studies on the normal metabolism of D- and L-2-HG acids may provide insights into the primary defect in patients with specific D- and L-2-HG aciduria.

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