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## **Crystal Growth of** *Isotactic* **Poly(butene-1) in the Melt. I. Kinetic Roughening**

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Figure 5 on page 230 and Figure 6 on page 231 have to be replaced by the following versions.





Figure 5. Growth rate G versus crystallization temperature T: (▲) sample A (this work), (■) sample B (this work), and (●) data by Icenogle (ref. 18).

Figure 6. Plot of  $\ln G + U/R(T - T_V)$  versus  $1/T \Delta T$ . Symbols are the same as in Figure 5.

From the last line on page 230 to the first paragraph on page 231 has to be replaced by the following version. According to eq 1, the extrapolation to zero of the straight lines in Figure 6 for  $1/T\Delta T$  gives the values of  $G_0 = 9.78 \times 10^3 \,\mu\text{m sec}^{-1}$  for sample A and  $G_0 = 7.29 \times 10^3 \,\mu\text{m sec}^{-1}$  for sample B; from the slope the values of K is obtained to be  $9.04 \times 10^4 \,\text{K}^2$  for sample A and  $9.54 \times 10^4 \,\text{K}^2$  for sample B. Both the values of  $G_0$  and K agree well for samples A and B.