Further Investigation of Equilibrium Modulus of Diblock Copolymer Micellar Lattice

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ABSTRACT: Polystyrene–polybutadiene (PS–SB) and polystyrene–polyisoprene (PS–PI) diblock copolymers form spherical micelles with PS cores and PB/PI corona in a PB/PI-selective solvent, *n*-tetradecae (C14). At high concentrations, the micelles are further arranged on bcc lattices. Structural factors governing the equilibrium modulus G_e of these PS–PB and PS–PI micellar lattices are discussed in this study. It turned out that G_e is primarily determined by the number density of the corona blocks v_{corona} and proportional to v_{corona} while the number density of the bcc lattice cell, D^{-3} with *D* being the lattice spacing (determined from SAXS), has only secondary effects on G_e . The proportionality between G_e and v_{corona} reflects the thermodynamic origin of the lattice elasticity, the entropic elasticity of the corona PB/PI blocks having osmotically correlated conformations. These results suggest a necessity of re-examination of a previous hypothesis that the cell number density primarily determines G_e of bulk copolymer systems forming cubic phases [Kossuth *et al.*, *J. Rheol.*, **43**, 167 (1999)]. Furthermore, the G_e/v_{corona} ratio was found to be moderately different for the PS–PB/C14 and PS–PI/C14 micellar lattices. This difference can be related to a difference in the magnitude of the osmotic correlation for the PB and PI corona blocks due to a slight difference of their solubilities in C14. [DOI 10.1295/polymi.36.430]

KEY WORDS Diblock Copolymer Micellar Lattice / Equilibrium Modulus / Osmotic Requirement / Lattice Stability /

Diblock copolymers form micelles with precipitated cores and solvated corona in selective solvents. These micelles form cubic lattices at high concentrations where the corona blocks of neighboring micelles are overlapping each other.^{1–6} This lattice formation results from a compromise of contradicting thermodynamic requirements, an elastic requirement of randomizing corona conformation and an osmotic requirement of reducing the spatial gradient of the corona concentration.^{1,4–6} Under these requirements, the corona blocks of neighboring micelles are forced to have mutually correlated conformations. The micelles are arranged on cubic lattices so as to satisfy the thermodynamic requirements and minimize this correlation.

The micellar lattice is the thermodynamically stable structure and exhibits equilibrium elasticity.¹⁻⁶ For polystyrene–polybutadiene (PS–PB) copolymers in a PB-selective solvent, *n*-tetradecane (C14), the equilibrium modulus G_e was found to be proportional to the number density ν_{corona} of the corona PB blocks and the proportionality constant (G_e/ν_{corona} ratio) was smaller than the constant k_BT expected for the simplest case of the entropic elasticity of independent corona blocks (k_B = Boltzmann constant and T = absolute temperature).⁴⁻⁶ This proportionality reflects the thermodynamic origin of the lattice. Namely, the osmotically correlated corona blocks sustain the lat-

tice and thus their number density is the primary factor determining G_e . At the same time, respective corona blocks do not behave as an independent entropic unit because of their conformational correlation.⁷ This correlation leads the G_e/v_{corona} ratio smaller than k_BT . Indeed, an experiment demonstrated that the G_e value increases moderately when the solvent quality toward the corona block is mildly reduced to partially screen this correlation.⁸

Thus, the equilibrium elasticity of the micellar lattice in selective solvents is well related to the thermodynamic (osmotic) feature of the corona blocks. However, Kossuth *et al.*⁹ recently reported that G_e of microphase-separated bulk block copolymers forming cubic lattices is primarily determined by the lattice spacing *D* and proportional to the number density of the lattice cell, D^{-3} . Their result, *if applicable to all cubic lattices*, gives us an interpretation that the lattice cell itself is the fundamental entropic unit sustaining the elasticity and the (correlated) corona blocks do not behave as this unit. This interpretation is contradicting to the thermodynamic feature of the micellar lattice (corona-sustained elasticity) confirmed from several experiments.^{2–7}

For this problem, we have examined a relationship between G_e and D^{-3} for a series of PS–PB and polystyrene–polyisoprene (PS–PI) micellar lattices in C14. It turned out that the G_e data of respective series scale

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with ν_{corona} but not with D^{-3} thereby lending support to the molecular picture of the corona-sustained elasticity. (This picture appeared to apply also to bulk copolymers.) Furthermore, the G_e/ν_{corona} ratio was found to be moderately different for the PS–PB and PS–PI micellar lattices, suggesting a difference in the magnitude of the osmotic correlation for the PB and PI corona blocks due to a slight difference of the solubilities of PB and PI in C14. Details of these results are presented in this paper.

EXPERIMENTAL

Materials

PS–PI and PS–PB diblock copolymers were anionically polymerized with sec-butyllithium in benzene. PI and/or PB blocks were polymerized first, and their living anions were split into several batches. Then, the PS blocks of various lengths were copolymerized in respective batches. For convenience of the molecular characterization, the precursor PI/PB blocks were also recovered.

These copolymer samples were characterized with GPC (CO-8020 and DP-8020, Tosoh Corporation) equipped with refractive index (RI) and ultraviolet absorption (UV) monitors (LS-8000 and UV-8020, Tosoh Corporation) connected in series. The eluent was THF. The PI/PB molecular weights were determined for the precursors through elution volume calibration utilizing previously synthesized/characterized monodisperse linear PI and PB samples^{10–12} as the elution standards. The PS molecular weight was determined from the RI and UV signal intensities of the copolymers.

The systems subjected to viscoelastic and small-angle X-Ray scattering (SAXS) measurements were the copolymer solutions in *n*-tetradecane (C14; Wako Pure Chemical Industries, Ltd.). These solutions were prepared by firstly dissolving prescribed masses of the copolymer and C14 in excess methylene chloride and then allowing methylene chloride to thoroughly evaporate. C14 is a good solvent for PI and PB blocks and a non-solvent for the PS blocks. Thus, the copolymers formed micelles with precipitated PS cores and solvated PB/PI corona in C14, as confirmed from SAXS measurements.

Measurements

For the PS–PI/C14 and PS–PB/C14 micellar solutions of various concentrations, linear viscoelastic measurements were carried out at 25 °C with a laboratory rheometer (ARES; Rheometrics F. E. Ltd). These solutions were charged in the rheometer and disordered at 70–90 °C for 15 min and then quiescently ordered (without being deformed) by cooling to 25 °C.

This process enabled equilibration of the micellar lattice by erasing undesirable shear history during the sample loading. After annealing at 25 °C for 12 h, the storage and loss moduli, G' and G'', were measured under sufficiently small strain (≤ 0.01).

For the micellar solutions, small-angle X-Ray scattering (SAXS) measurements were also conducted at 25 °C with a laboratory goniometer (RINT-2000, Rigaku Corp.) equipped in Professor Kanaya's laboratory in Institute for Chemical Research, Kyoto University. Cu-K α line of the wavelength $\lambda = 0.154$ nm was used. The samples were charged in a cell having a mica window. The SAXS intensity I(q) was measured as a function of the magnitude of the scattering vector $q = 4\pi \sin \theta / \lambda$ (θ = scattering angle). No desmearing correction was made for the I(q) data.

RESULTS AND DISCUSSION

Viscoelastic and SAXS Data

Linear viscoelastic and SAXS measurements were conducted for the PS–PI/C14 and PS–PB/C14 systems of various concentrations at 25 °C. Representative results obtained for the 31.0 wt% PS–PI 11–85/ C14 system are summarized in Figures 1 and 2: Figure 1 shows angular frequency (ω) dependence

Table I. Characteristics of Diblock Copolymer Samples

Code ^a	$10^{-3}M_{\rm S}$	$10^{-3}M_{\rm D}{}^{\rm b}$	$M_{\rm w}/M_{\rm n}$
PS-PI 11-85	10.5	85.0	1.03
PS-PI 19-85	18.7	85.0	1.03
PS-PI 42-85	41.9	85.0	1.03
PS-PB 13-93	12.7	92.8	1.03
PS-PB 23-93	23.2	92.8	1.03

^acode number indicates $10^{-3}M_{\rm S}-10^{-3}M_{\rm D}$. ^bmolecular weight of diene block (PI or PB).



Figure 1. Linear viscoelastic behavior of the 31.0 wt% PS–PI 11–85/C14 micellar lattice system at 25 °C.



Figure 2. SAXS profile of the 31.0 wt% PS–PI 11–85/C14 micellar lattice system at 25 °C.

of the storage and loss moduli G' and G'', and Figure 2 shows the *q* dependence of the SAXS intensity I(q).

Judging from a small volume fraction (= 0.036) of the PS block in the PS-PI 11-85/C14 system, we expect that the PS-PI 11-85 chains form spherical micelles with precipitated PS cores and solvated PI corona. This expectation is confirmed in Figure 2 where the scattering peaks in a range of $q < 0.4 \text{ nm}^{-1}$ are assigned to inter-domain interference in the micellar lattice. In addition, the scattering maximum at $q \cong$ $0.6 \,\mathrm{mn^{-1}}$ is attributable to the intra-domain scattering from the PS cores of the radius $r_{\rm S} = 5.765/q = 9 \pm$ 0.5 nm. This $r_{\rm S}$ value is a little larger than the PS domain radius expected in bulk systems ($\cong 8 \text{ nm}$),⁶ suggesting that the PS block of $M_{\rm S} = 10.5 \times 10^3$ is moderately swollen and plasticized with C14. Indeed, a thermal analysis for a little shorter PS block ($M_{\rm S} =$ 7.2×10^3) revealed the swelling of this block with C14 that decreased the glass transition temperature T_{g} to 30 °C.¹³ At the same time, it should be emphasized that our PS block is longer and should have $T_{\rm g} > 30\,^{\circ}{\rm C}$ to behave as a rigid (glassy) anchor for the PI corona at our experimental temperature, 25 °C. This was the case for all PS-PI and PS-PB copolymers examined in this study.

The arrows in Figure 2 indicate the relative location of the inter-domain scattering peaks expected for a body-centered cubic (bcc) lattice of the spherical micelles, $q_1:q_2:q_3 = 1:\sqrt{2}:\sqrt{3}$. Although the second-order peak is not well resolved, the relative location of the first- and third-order peaks coincides with that for the bcc lattice to confirm the bcc lattice formation of the spherical PS-PI 11-85 micelles. This lattice formation was confirmed for all PS-PI/C14 and PS- PB/C14 systems examined.

In Figure 1, we note that G' and G'' of 31.0 wt% PS–PI 11–85/C14 system are close to each other in magnitude and exhibit considerable ω dependence in a range of $\omega > 1 \text{ s}^{-1}$. This behavior is indicative of a fast relaxation process (attributable to motion of the PI blocks). However, at $\omega < 1 \text{ s}^{-1}$, G' is quite insensitive to ω and G'' is much smaller than G'. This low- ω behavior clearly demonstrates the equilibrium elasticity of the micellar lattice formed in the system. Similar behavior was observed for all systems examined.

Structural Factors Determining Equilibrium Modulus

For all PS-PI/C14 and PS-PB systems, the elastic behavior was observed at low ω and the equilibrium modulus $G_{\rm e}$ was evaluated as the G' at the lowest ω examined (= 0.01 s^{-1}). Figures 3 shows double-logarithmic plots of the normalized modulus $G_{\rm e}/k_{\rm B}T$ against the number density of the corona blocks v_{corona} (calculated from the copolymer concentration and molecular weight). The data obtained in this study are shown with large unfilled symbols. For comparison, the plots are also shown for previously examined PS-PB/C14^{1,2,7,14} and PS-PI/C14^{13,15,16} micellar lattice systems (small unfilled symbols) and for bulk diblock copolymers forming cubic lattices reported by Kossuth et al.⁹ (filled symbols), poly(ethylene propylene)-poly(ethyl ethylene) (PEP-PEE), polyethylenepoly(ethyl ethylene) (PE-PEE), PI-PS, and poly-(ethylene oxide)–poly(ethyl ethylene) (PEO-PEE).



Figure 3. Plots of the normalized equilibrium modulus G_e/k_BT of various copolymer lattice systems against the corona number density ν_{corona} . The dotted line indicates a relationship for the simplest case of the entropic elasticity of independent corona blocks, $G_e^{\circ}/k_BT = \nu_{corona}$.



Figure 4. Plots of the normalized equilibrium modulus G_e/k_BT of various copolymer lattice systems against the number density D^{-3} of the bcc lattice cell. The dotted line indicates a relationship, $G_e/k_BT = D^{-3}$. This relationship is to be satisfied if each cell behaves as the independent entropic unit sustaining the lattice elasticity.

The corona concentration is in a range of $0.08 \le c_{\rm corona}/{\rm g \, cm^{-3}} \le 0.25$ for all PS–PB/C14 and PS–PI/C14 lattice systems examined here.

The spacing D of the bcc lattice formed in the above systems is evaluated from the magnitude of the scattering vector at the first-order peak, q_1 , as

$$D = \sqrt{\frac{3}{2}} \frac{2\pi}{q_1} \tag{1}$$

For the lattice systems for which both of the G_e and D data are available, Figure 4 shows the plots of G_e/k_BT against the number density of the bcc lattice cell, D^{-3} . The symbols are the same as in Figure 3. [Kossuth *et al.*⁹ made plots of G_e/RT (R = gas constant) against $2\pi/q_1$. The format of the plots in Figure 4 is equivalent to their format.]

In Figure 3, we first note that the G_e/k_BT data of all systems are smaller than the normalized modulus expected for the simplest case of the entropic elasticity of independent corona blocks, $G_e^{\circ}/k_BT = v_{corona}$ (shown with the dotted line). This result is naturally related to the thermodynamic origin of the lattice: The corona blocks are conformationally correlated under the osmotic requirement and thus the effective number density of the independent entropic unit (a set of correlated corona blocks) is smaller than v_{corona} thereby giving $G_e/k_BT < v_{corona}$.

This statement can be re-cast in the thermodynamic expression of G_e of micellar lattices derived by Hashimoto *et al.*,¹⁴

$$\frac{G_{\rm e}}{k_{\rm B}T} = \frac{\nu_{\rm corona}}{\beta}$$

with

$$\frac{1}{\beta} = \frac{1}{\gamma^2} \ln\left(\frac{P_0}{P}\right) \quad \text{for small } \gamma \tag{2}$$

where P_0 is the probability density of the osmotically constrained corona blocks in the undeformed state and P is the probability density of these blocks under a small strain γ . (Since the $\ln(P_0/P)$ term does change with the direction of shear, this term should be proportional to γ^2 for small γ .) The parameter β appearing in Eq 2 is equivalent to the number of the corona blocks per independent entropic unit explained above. This β would increase from unity (for the case of independent Gaussian blocks) with increasing magnitude of the osmotic constraint. However, for respective series of the PS-PI/C14 and PS-PB/C14 micellar lattices in the rather limited range of $0.08 \le c_{\rm corona}/{\rm g\,cm^{-3}} \le$ 0.25, β appears to be roughly constant to give the proportionality between $G_{\rm e}/k_{\rm B}T$ and $\nu_{\rm corona}$ within experimental scatter (Figure 3).

Nevertheless, β would be determined by the osmotic compressibility for the solvated corona blocks⁸ and may have different values in the systems comprising of chemically different blocks/solvents. Concerning this difference, we note that the $G_{\rm e}/k_{\rm B}T$ value for a given v_{corona} value is somewhat larger for the PS-PI/C14 systems than for the PS–PB/C14 systems; cf. Figure 3. This result reminds us a previous observation⁸ that $G_{\rm e}$ of PS–PI lattices moderately increases (by a factor up to 60%) on replacement of a small fraction (7%) of the solvent C14 by a non-solvent, diethyl phthalate: This replacement resulted in a slight reduction of the solubility for the PI corona, as characterized by a small decrease (6%) of the intrinsic viscosity of a high-*M* homo-PI ($M = 180 \times 10^3$) on the replacement. Judging from this sensitivity of G_e to the osmotic environment for the corona (determined by the solubility), we speculate that the moderate difference in the G_e/k_BT values for the PS-PI/C14 and PS-PB/C14 systems (Figure 3) may reflect a slight difference of the solubilities of PI and PB blocks in C14. (As judged from a difference of the Mark-Houwink–Sakurada exponents α in hexane/heptane chemically similar to C14, $\alpha = 0.58$ for PI/hexane solutions¹⁷ and $\alpha = 0.64$ for PB/heptane¹⁸ solutions (both values obtained from five fractions), C14 appears to be slightly poorer for PI than for PB thereby giving somewhat larger $G_{\rm e}$ for PS–PI.) A further study is desirable for this delicate solvent effect.

Now, we turn our attention to the relationship between G_e and D^{-3} . Since D for a given copolymer decreases with an increase of the copolymer concentration, a general trend in the G_e/k_BT vs. D^{-3} plots (Figure 4) is similar to that in the G_e/k_BT vs. ν_{corona} plots (Figure 3). However, a considerably large scatter is observed in the $G_{\rm e}/k_{\rm B}T$ vs. D^{-3} plots for respective series of the PS-PI/C14 and PS-PB/C14 systems; see the unfilled circles, for example. This scatter suggests that the bcc lattice cell is *not* the independent entropic unit sustaining the lattice elasticity. This point can be most clearly noted for the two pip-circles indicating the G_e data for the 16.6 wt % PS–PI 19–85/C14 and 20.0 wt % PS-PI 42-85/C14 systems having quite different *D* values but identical v_{corona} value (= 7.8 × 10^{23} m⁻³): *G*_e hardly changes with *D*⁻³ if v_{corona} is kept constant. Similar behavior is noted for the two pip-squares showing Ge of previously examined PS-PB/C14 systems having quite different D^{-3} values and fairly close v_{corona} values (= 2.2×10^{24} and $3.2 \times 10^{24} \,\mathrm{m}^{-3}$).

If the bcc lattice cell is the independent entropic unit, each cell should sustain the modulus of the order of $k_{\rm B}T$ and thus $G_{\rm e}/k_{\rm B}T$ should be close to the number density of the cell, D^{-3} . However, Figure 4 demonstrates that the $G_{\rm e}/k_{\rm B}T$ data of all systems are significantly larger (by a factor > 30) than the modulus expected for this case, $G_{\rm e}/k_{\rm B}T = D^{-3}$ (dotted line). This fact means that each cell contains many (more than 30) entropic units. This is the case not only for the solvated PS–PB and PS–PI micellar lattices but also for the bulk copolymers examined by Kossuth *et al.*⁹

The above results allow us to conclude, from both qualitative and quantitative points of view, that the bcc lattice cell is not the independent entropic unit sustaining the lattice elasticity. The cell contains many (more than 30) entropic units. Thus, a set of the conformationally correlated β corona blocks should behave as this unit, as concluded from the results shown in Figure 3.

Finally, we should point out that the delicate solvent effect on the number β (seen for the PS–PI/C14 and PS–PB/14 systems) has not been described quantitatively, although the role of the solubility can be explained qualitatively. This description requires us to formulate the probability densities *P* and *P*₀ (Eq 2) as functions of the osmotic compressibility and other thermodynamic parameters. This formulation is considered to be an interesting subject of future work.

CONCLUSIONS

For the PS–PB/C14 and PS–PI/C14 micellar lattices as well as the lattices in bulk diblock copolymers, the equilibrium modulus G_e was found to be primarily determined by the number density of the corona blocks v_{corona} . For respective series of the micellar lat-

tice systems, G_e was proportional to v_{corona} but smaller, in magnitude, than that expected for the simplest case of the entropic elasticity of the corona blocks, $G_e^{\circ} = v_{corona}k_BT$. In contrast, no proportionality was found between G_e° and the number density of the bcc lattice cell, suggesting that the cell size has only secondary effects on G_e . These results indicate that a set of osmotically correlated corona blocks behaves as the independent entropic unit sustaining the lattice elasticity and each cell contains many entropic units. This conclusion holds not only for the solvated micellar lattices but also for the lattices in bulk copolymer systems.

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