SHORT COMMUNICATIONS

Structure of *Bombyx mori* Silk Fibroin Studied by REDOR NMR Spectroscopy

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It has been confirmed that the solid-state conformation of silk fibroin from Bombyx mori is strongly dependent on sample preparation.¹ It is well known that the silk II form (antiparallel β -sheet), obtained by drying at high temperature or under mechanical stress, is the most stable form. The metastable silk I form is obtained by drying from aqueous solution at room temperature. ¹³C chemical shifts are sensitive to molecular conformation and have been used to show that silk I and silk II are structurally different.¹ Whereas the structure of silk II has been known for some time (for example, see ref 2), the structure of silk I has not been determined due, primarily, to the difficulty in obtaining high quality oriented samples. However, several models have been proposed for the silk I structure. These include the "crankshaft" model by Lotz and Keith² and three models by Asakura and Yamaguchi³ based on ¹³C CP/MAS NMR, X-ray diffraction data and conformational energy considerations.

We are initiating efforts to obtain the backbone conformation of silk I using the

rotational-echo, double-resonance (REDOR) NMR technique to measure internuclear distances between specific ¹³C-¹⁵N spin pairs.⁴⁻⁶ These measured distances will be used to determine the dihedral angles describing the backbone conformation.⁷ A particular advantage of our strategy is that oriented samples are *not* required and simple powder samples can be used. As a demonstration of the technique, we present initial REDOR results on a model compound with the well-characterized silk II conformation.

THEORETICAL BACKGROUND

Rotational-echo, double-resonance NMR is a high-resolution, solid-state MAS NMR technique designed to measure internuclear distances between heteronuclear spin pairs.⁴⁻⁶ REDOR is ideally suited for measuring specific internuclear distances between selectively labeled spin sites. In particular, selective labeling of proteins and peptides with ¹³C and ¹⁵N isotopes and measuring the internuclear separation between the labeled spin pair is

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Figure 1. REDOR pulse sequence. The π -pulse trains use xy-8 phase cycling.

particular attractive due to the low natural abundance of these isotopes.

The internuclear distance, $r_{\rm CN}$, between a ¹³C-¹⁵N spin pair is related to the dipolar coupling constant, *D*, by

$$D = \frac{\gamma_{\rm C} \gamma_{\rm N} \hbar}{2\pi r_{\rm CN}^3} \tag{1}$$

where the γ 's are the respective magnetogyric ratios; *D* is in Hz. Under usual MAS conditions, the heteronuclear dipolar interaction is coherently averaged to zero by the sample rotation. However, the REDOR experiment recovers the dipolar interaction by manipulating the spin coordinates of the dipolar interaction with rotor-synchronous π -pulse trains.⁵

REDOR is performed as a difference experiment. Figure 1 shows the ¹³C-observe REDOR experiment used in this work.¹⁰ This experiment begins with cross-polarization of the ¹³C spins via the protons. After crosspolarization, a strong rf decoupling field removes the proton-carbon dipolar interaction throughout the remainder of the experiment. The cross-polarization preparation is followed by a dipolar evolution period consisting of $N_{\rm e}$ rotor cycles (here $N_c = 10$). A compensated π -pulse train is applied on the ¹³C channel during the dipolar evolution period to refocus isotropic chemical shifts prior to data acquisition, which immediately follows the evolution period. The *full* spectrum is obtained with no π pulses on the ¹⁵N-channel. In this experiment, the dipolar interaction is averaged to



Figure 2. Calculated ratio of the REDOR *difference* signal to the *full*-echo signal.

zero by the sample rotation. However, when a train of π pulses is applied on the ¹⁵N-channel, the ¹³C magnetization is dephased by the ¹³C-¹⁵N dipolar interaction. Two π pulses are applied per rotor cycle to ensure that dipolar dephasing is additive. This dephased signal is called the *reduced* signal. The *difference* signal from the *full* signal. Only ¹³C spins coupled to ¹⁵N spins will appear in the *difference* spectrum.

The dipolar coupling constant and, consequently, the internuclear separation is obtained by comparing the measured ratio of the *difference* signal, ΔS , to the *full* signal, S_0 , to the calculated ratio. Figure 2 shows the calculated ratio of $\Delta S/S_0$ as a function of the product of the number of rotor cycles during the evolution period, N_c , the dipolar coupling constant, D, and the rotor period, T_r .⁵

EXPERIMENTAL

The ¹³C-¹⁵N labeled hexapeptide used in this work is shown in Figure 3. The ¹³C and ¹⁵N labels are shown in the formula as C and N, respectively. To inhibit the side reaction, the OH group of the side chain of serine residue is changed to an OBzl group. The N- and Cterminus groups were synthesized with Bocand -OBzl, respectively. A sample was also prepared without any isotopic labeling. This natural-abundance sample and the labeled

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Figure 3. The structure of the labeled hexapeptide used in REDOR experiment (top) and ¹³C CP/MAS spectrum of natural-abundance hexapeptide together with the assignment.

sample were recrystallized together in a ratio of 1 part labeled sample to 24 parts unlabeled sample to minimize intermolecular dipolar interactions. The measured ¹³C-¹⁵N distance will place limits on the possible values of the ϕ , ψ dihedral angles of the Ala₅ residue.⁷

The REDOR experiments were performed at a ¹³C frequency of 38.07 MHz at Florida State University on a home-built spectrometer. The 230 mg sample was centered in a zirconia rotor with Kel-F spacers. Sample spinning was done at 3205 Hz and $N_c = 34$. Cross-polarization and all carbon and nitrogen pulses were performed with an rf field strength of 38 KHz. Proton decoupling was performed at an rf field strength of 110 KHz. The rotor-synchronous, compensated π -pulse trains used the *xy*-8 phase scheme to minimize off-resonance effects.⁸⁻¹⁰

RESULTS AND DISCUSSION

Figure 3 shows the ¹³C CP/MAS NMR spectrum of natural-abundance hexapeptide with peak assignments. The C_{β} chemical shift of the alanine residue (*ca.* 20.0 ppm) suggests that this sample takes on the silk II form (antiparallel β -sheet).¹ In a previous study, the alanine torsion angles ϕ and ψ in the silk II form were reported as -142° and 145° , respectively.² Using these torsion angles, the



Figure 4. ${}^{13}C$ REDOR spectra of the carbonyl carbon region of the [${}^{13}C{-}^{15}N$] labeled hexapeptide. (bottom) echo spectrum collected in the absence of all ${}^{15}N$ pulses and (top) the REDOR difference spectrum.

distance between the ¹³C and ¹⁵N labels in the selectively labeled sample is 4.55 Å.

Figure 4 shows the ¹³C-observe REDOR spectra of the carbonyl carbon region of the diluted hexapeptide sample. Figure 4 (bottom) shows the full spectrum collected in the absence of ¹⁵N pulses and Figure 4 (top) shows the difference spectrum. In order to obtain $\Delta S/S_0$, several correlation must be made. First, natural-abundance contributions to the carbonyl carbon resonance from the unlabeled carbonyl carbon positions are taken into account to obtain S_0 . Second, corrections to ΔS from two sources must be made. The first correction to ΔS is to account for the coupling between the labeled carbonyl carbon and natural-abundance ¹⁵N spins located closer to the ¹³C label than the specific ¹⁵N label. The second correction to ΔS accounts for other natural-abundance carbonyl ¹³C spins coupled

to natural-abundance ¹⁵N spins. Both corrections are done analytically assuming standard peptide geometry and bond lengths.⁷

After corrections, the experimentally determined ratio $\Delta S/S_0$ is 0.159. Reference to Figure 2 provides a ¹³C-¹⁵N distance of 4.32 Å (0.1A). This internuclear separation is in good agreement with 4.55 Å distance predicted for silk II conformation. These results show that REDOR will be an effective tool in characterizing silk-like materials.

Further REDOR experiments with different labeling schemes are planned to probe the conformation of model compounds of silk I.

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