

Ferromagnetic quantum critical point induced by dimer-breaking in $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$

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In contrast to classical phase transitions driven by temperature, a quantum critical point (QCP) defines a transition at zero temperature that is driven by non-thermal parameters^{1–3}. In the known quantum critical *d*-electron systems, tuning the electronic bandwidth by means of changing the applied pressure or unit-cell dimensions, or tuning the *d*-state population, is used to drive the criticality^{4–6}. Here we describe how a novel chemical parameter, the breaking of bonds in Ge–Ge dimers that occurs within the intermetallic framework in $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$, results in the appearance of a ferromagnetic (FM) QCP. Although both SrCo_2P_2 and SrCo_2Ge_2 are paramagnetic, weak itinerant ferromagnetism unexpectedly develops during the course of the dimer breaking, and a QCP is observed at the onset of the FM phase. The use of chemical bond breaking as a tuning parameter to induce QCP opens an avenue for designing and studying novel magnetic materials.

$\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ belongs to the layered, tetragonal ThCr_2Si_2 structure type, which has over 600 intermetallic members. This AT_2X_2 structure is formed by stacking covalently bonded transition metal–metalloid T_2X_2 layers, made from edge-sharing TX_4 tetrahedra, with ionic A atoms (see Fig. 1a inset). These compounds have attracted recent interest owing to the rich physics they exhibit, such as non-Fermi-liquid behaviour in YbRh_2Si_2 (ref. 7) and unconventional superconductivity in K-doped BaFe_2As_2 (ref. 8). The existence of a molecule-like X–X dimer that crosses the A atom layer in certain members of this family was noted 20 years ago, providing an explanation for their anomalously low ratio of stacking to in-plane lattice parameters^{9,10} (c/a). This X–X bonding between neighbouring T_2X_2 layers pulls the layers closer together and induces a relaxation of the in-plane lattice dimension, leading to what are known as collapsed tetragonal (cT) cells. In contrast, the absence of an X–X dimer results in uncollapsed tetragonal (ucT) cells. The lack of an interlayer X–X bond substantially weakens the electronic interactions between neighbouring T_2X_2 layers, leading to less three-dimensional (3D)-like electronic structures and Fermi surface topologies for the ucT phases compared with the cT phases^{11–13}. Lattice collapse transitions from ucT to cT phases driven by applied pressure or chemical pressure were not shown to have a great impact on the physical properties of the compounds until recently^{14–18}.

SrCo_2Ge_2 is a typical cT phase, with a c/a ratio of 2.6 due to the presence of a strong interlayer Ge–Ge bond, whereas SrCo_2P_2 , with a c/a ratio of 3.1, is a typical ucT phase. The solid solution of these two compounds should therefore trace the crossover from cT to ucT phases, and, as shown in Fig. 1a, the $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ solid solution shows anomalous *S*-shape changes in both *a* and *c* as a

function of *x*. Although *a* and *c* vary noticeably, the absence of a region of chemical phase separation indicates that the transition from cT to ucT is continuous in nature. Figure 1b, which plots important interatomic distances and bond angles derived from structural refinements, shows that the *S*-shape behaviour of *a* and *c* arises from the breaking of the X–X dimer on substitution of P for Ge, which begins at $x \sim 0.4$ and is completed by $x \sim 0.8$. Therefore, the structural phase diagram consists of a cT phase for $x < 0.4$, a ucT phase for $x > 0.8$ and an intermediate state for $0.4 < x < 0.8$. The X–X bond length within the dimer varies from 2.9 Å for $x = 0$ to 3.4 Å for $x = 1$, crossing from bonded to non-bonded distances^{9,10}. Correspondingly, the Co–Co separation within the T_2X_2 layers decreases as the layers separate as a result of the dimer breaking, increasing the in-plane Co–Co interactions. The size and shape of the CoX_4 tetrahedron (the Co–X distances and Co–X–Co tetrahedral angle in Fig. 1b) also vary considerably in the crossover region, $0.4 \leq x \leq 0.8$, as the weakening of the X–X bond allows the structure to relax to a state dominated by 2D rather than 3D bonding interactions. The crystal structure characterization therefore implies that although the electronic structures should be quite different for the cT SrCo_2Ge_2 and ucT SrCo_2P_2 phases, the properties of the solid solution should vary continuously between the two endmembers.

As shown in Fig. 2, the magnetic properties of $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ do not follow the simple trend implied by the structural variation. Both parent compounds, SrCo_2Ge_2 and SrCo_2P_2 , are Pauli paramagnetic (PM), although the latter compound is Stoner-enhanced. These results are consistent with previous studies^{15,18,19}. A highly unexpected FM state appears, however, for $0.35 \leq x \leq 0.7$ (Fig. 2a and b). As *x* increases from 0 towards 1, the series exhibits spontaneous magnetization, which first appears at 2 K with a small saturated moment ($\mu_{\text{sat}} = 0.02\mu_{\text{B}}/\text{Co}$) for $x = 0.35$. This μ_{sat} increases to a maximum value ($0.1\mu_{\text{B}}/\text{Co}$) at $x = 0.55$ before decreasing again (see Fig. 3). For $x > 0.8$, the FM ground state disappears entirely.

The FM ordering temperature (T_{C}) was determined from the temperature-dependent magnetization and magnetization isotherm (Arrott plot)²⁰ measurements (Fig. 2c). As shown in Fig. 4, T_{C} varies in the same manner as μ_{sat} : the highest T_{C} is 35 K for $x = 0.55$. Figure 2d shows that the series exhibits high-temperature Curie–Weiss behaviour for $x \geq 0.35$. The effective moment (μ_{eff}) increases from $0.64\mu_{\text{B}}/\text{Co}$ for $x = 0.35$ to $1.7\mu_{\text{B}}/\text{Co}$ for $x = 1$. These μ_{eff} values are 10–30 times larger than the corresponding μ_{sat} values, placing these FM compounds solidly in the weak itinerant FM region of the Rhodes–Wohlfarth plot²¹.

To understand how the system evolves from Pauli PM to FM, and then back to enhanced PM with increasing *x*, the samples near

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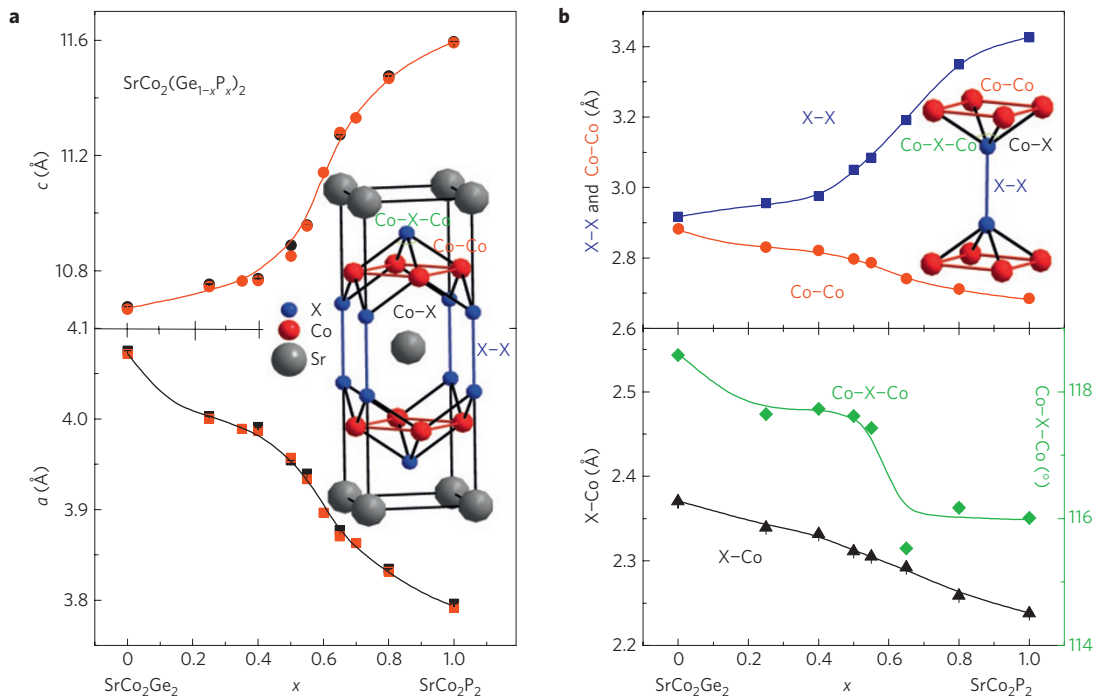


Figure 1 | Lattice collapse transition in $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$. **a**, Lattice parameters. Red symbols: laboratory X-ray diffraction (XRD); black symbols: synchrotron XRD. Inset: Unit cell for SrCo_2X_2 . **b**, X-X, Co-Co and X-Co bond lengths, and the Co-X-Co angle, from synchrotron XRD. All the error bars are statistical errors from the GSAS refinement.

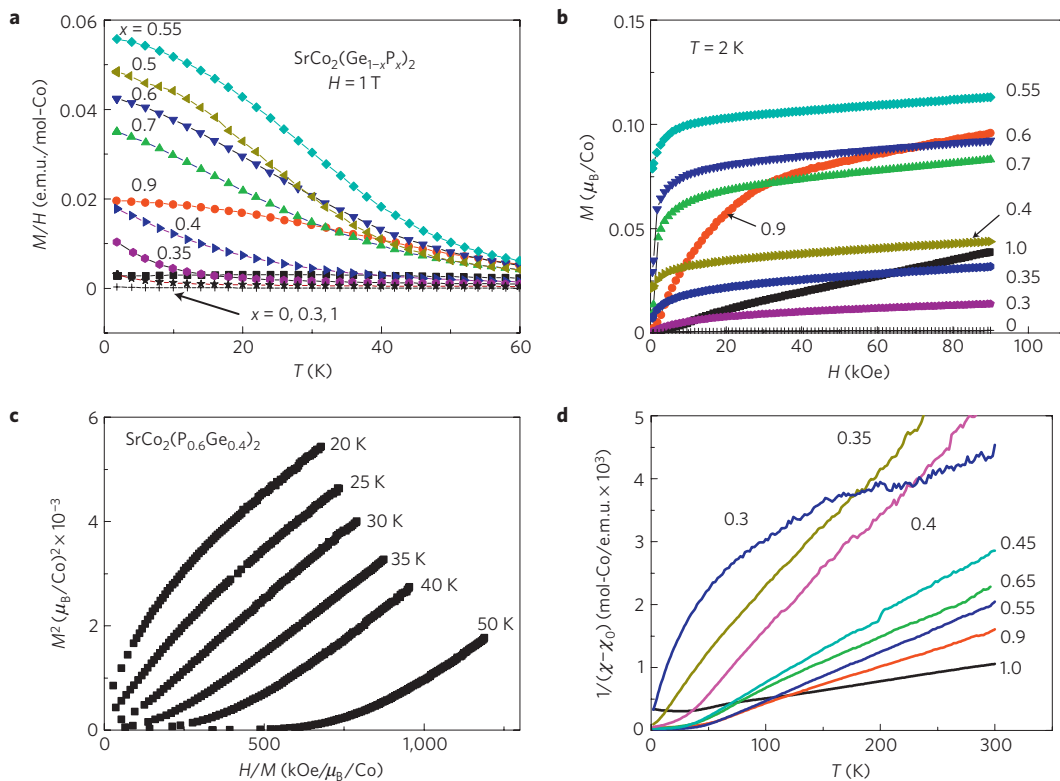


Figure 2 | The magnetic properties of $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$. **a**, M/H versus T . **b**, M versus H at 2 K. **c**, Arrott plot for $x = 0.6$. **d**, Curie-Weiss behaviour for $x \geq 0.35$.

the critical region were measured in more detail. Figure 2a,b shows that the FM ground state develops as x increases from 0.3 to 0.4. The Arrott plot measurements show that T_C for $x = 0.35$ (Fig. 3a) and 0.4 (data not shown) are 2 ± 0.2 K and 12 ± 1 K respectively.

Figure 3b shows the low-temperature magnetic susceptibility (χ) plotted with respect to $T^{-4/3}$ for $x = 0.3, 0.325$ and 0.35 . Although the χ values are highly sensitive to x , the sample with $x = 0.325$ exhibits very close to the power-law behaviour expected near a FM

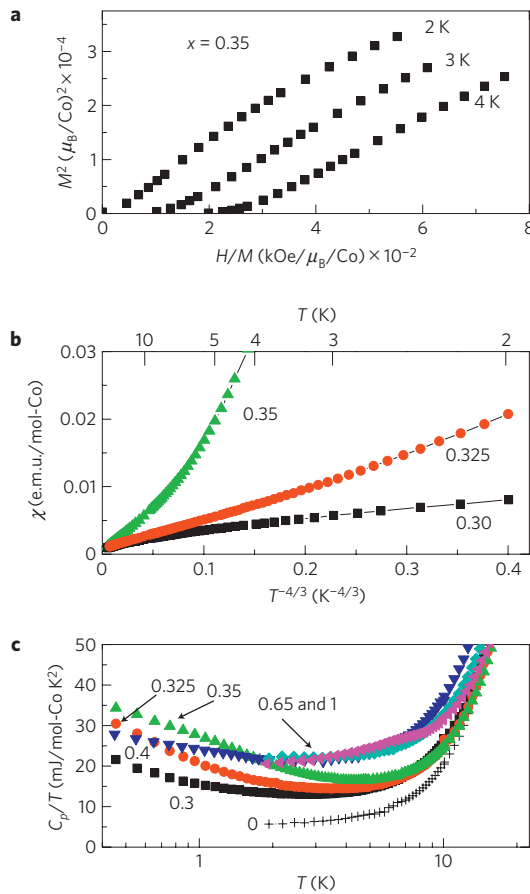


Figure 3 | QCP behaviour near $x = 0.325$. **a**, Arrott plot for $x = 0.35$ showing that T_c is lower than 2 K. **b**, χ versus $T^{-4/3}$ for $x = 0.30, 0.325$ and 0.35 showing a nearly linear dependence at $x = 0.325$. **c**, Low-temperature heat capacity for $x = 0, 0.3, 0.325, 0.35, 0.4, 0.65$ and 1 showing $\log T$ divergence near $x = 0.325$.

QCP (ref. 1). Figure 3c shows the heat-capacity data as C_p/T versus T on semi-logarithmic axes. Initially, C_p/T is a constant at low temperatures. However, at $x = 0.325$ and 0.35 , C_p/T exhibit clear logarithmic behaviour with respect to T , which is characteristic of a system approaching a FM QCP (ref. 22). When $x \geq 0.4$, the logarithmic C_p/T behaviour is suppressed and Fermi-liquid behaviour ($C_p/T = \gamma_0 + \beta T^2$) is recovered. There are no signs of spin-glass behaviour²³ in any of the magnetization or heat capacity data near $x = 0.35$. All these results demonstrate the presence of a FM QCP driven by x near $x = 0.325$ in $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$.

In contrast to the QCP near $x = 0.325$, a simple crossover from the FM state to an enhanced PM state appears for $x \sim 0.8$. Figure 4b shows that T_c maintains a large value (20 ± 2 K) for $x = 0.7$. When $x > 0.7$, the magnetization remains large but the spontaneous magnetization disappears (Fig. 2a,b). Correspondingly, the heat capacity data show Fermi-liquid behaviour with invariant γ_0 for $0.5 \leq x \leq 1$.

As summarized in Fig. 4, the physical properties of $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ are strongly correlated with its variation in structure. When x increases, both the zero-temperature susceptibility ($\chi_{T=0}$) and γ_0 increase, but the QCP occurs exactly when the X–X dimer is beginning to break (Fig. 4b). Furthermore, the FM ground state appears only during the breaking of the dimer. When the dimer is fully broken, the FM state disappears and $\chi_{T=0}$ decreases with increasing x , whereas γ_0 is invariant. The phase diagram in Fig. 4b clearly demonstrates that the QCP and ferromagnetism are not induced by a simple electron doping

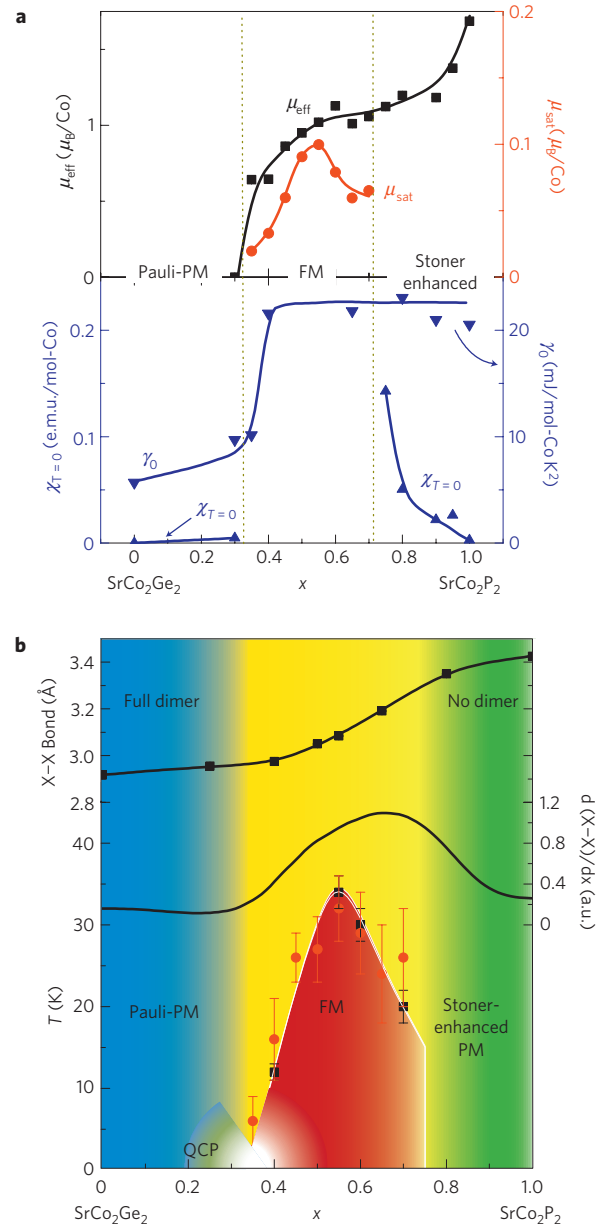


Figure 4 | Summary and phase diagram for $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$. **a**, Physical properties for $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$. Upper panel: effective moment and saturated moment; lower panel: zero-temperature susceptibility ($\chi_{T=0}$) and Sommerfeld coefficient (γ_0). **b**, The structural and magnetic phase diagram for $\text{SrCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$. Black symbols: T_c determined by the Arrott plot; the error bars are the temperature interval of the measured isotherms. Red symbols: T_c determined by assuming $M(T_c) = 20 \pm 10\% M(2\text{ K})$.

effect, but rather by the breaking of the dimer. This result is consistent with continuing studies on $\text{CaCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ and $\text{BaCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ (refs 24,25). The X–X dimer is fully intact for the whole $\text{CaCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ series, and the magnetic properties vary from non-magnetic to antiferromagnetic with no unexpected FM state in the intermediate composition region. In contrast, the X–X dimer is fully broken throughout the $\text{BaCo}_2(\text{Ge}_{1-x}\text{P}_x)_2$ series because of the large size of the Ba^{2+} ions, and only non-magnetic ground states are observed.

As previously reported, chemical pressure drives P–P dimer breaking in $\text{Sr}_{1-x}\text{Ca}_x\text{Co}_2\text{P}_2$, which is associated with a transition from a magnetic cT phase ($x = 1$) to a non-magnetic ucT phase ($x = 0$). A simple explanation for the magnetic transition is

based on the Zintl concept^{9,15}; for example, the singly bonded P–P dimer in cT phases has the electron formal count (P–P)^{4–}, whereas the non-bonded P in ucT phases has the formal electron count P₂^{6–}. Charge redistribution therefore occurs in the Co₂P₂ layers during the lattice collapse transition, and changes in the magnetic ground state may be anticipated. A magnetic to non-magnetic transition is observed in CaFe₂As₂ when applied pressure induces a lattice collapse¹⁷. In contrast, SrCo₂(Ge_{1–x}P_x)₂ shows different behaviour: with both endmembers non-magnetic, a FM ground state arises during the dimer breaking that destabilizes the collapsed phase. Within the Zintl concept, the substitution of P for Ge in this system does not induce a formal electron count change in the *d*-state because the extra charge contributed by P doping is consumed through the breaking of the dimer. Therefore the unexpected ferromagnetism cannot be understood as simply being a result of a valence change of the Co.

We speculate that a possible explanation may be found by consideration of differences in the band structure between bonded and non-bonded phases²⁶. With the presence of a covalent Ge–Ge interlayer bond, the chemical potential for SrCo₂Ge₂ must be located between the bonding orbitals (σ) and antibonding orbitals (σ^*) of the Ge–Ge dimer. In contrast, with the lack of an interlayer P–P bond, both the σ and σ^* orbitals are fully occupied in SrCo₂P₂. The σ^* band must therefore straddle the chemical potential during the continuous process of dimer breaking in SrCo₂(Ge_{1–x}P_x)₂. A higher density of states at the Fermi level and strong electron correlations therefore might occur when the σ^* orbitals and the Co 3*d* orbitals, which are possibly hybridized with it, are at the chemical potential at intermediate compositions, surpassing the Stoner criterion and driving the material into a band FM state. Further theoretical treatment is needed to validate this proposed mechanism for the unexpected correlation of dimer breaking and the onset of ferromagnetism at a QCP in this system. Finally, we believe that the presence or absence of X–X bonding in ThCr₂Si₂-type phases, not generally taken as an influence on structure–property relationships in this large family of compounds, requires more careful consideration.

Methods

Polycrystalline samples were prepared from 1.05Sr, 2xCoP and 2–2xCoGe, as described in ref. 18, at 950–1,000 °C for two days. CoP was prepared as in ref. 27 and CoGe was made from arc-melted CoGe_{1.05}. All the samples were characterized by laboratory X-ray powder diffraction with Cu K α radiation (D8 focus, Bruker). Selected samples with *x* = 0, 0.25, 0.4, 0.5, 0.55, 0.65, 0.8 and 1.0 were measured by synchrotron X-ray powder diffraction at room temperature using beam line 11-BM at Argonne National Laboratory. We analysed the structure using the program GSAS with EXPGUI (refs 28,29). As the refined *x* were the same as the nominal *x* values within $\pm 2\%$ error, the nominal *x* values were used for all samples. All physical property characterizations were carried out on a Quantum Design Physical Property Measurement System with a ³He option and a Quantum Design Magnetic Property Measurement System.

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Author contributions

P.J. and S.J. synthesized the materials. M.R.S. and B.H.T. performed the synchrotron XRD. S.J., J.G.C. and N.P.O. performed the thermodynamic measurements. S.J., P.J. and R.J.C. analysed the data. S.J. and R.J.C. wrote the paper. R.J.C. designed the study.

Additional information

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