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Received 21 Apr 2015 | Accepted 30 Jul 2015 | Published 11 Sep 2015

DOI: 10.1038/ncomms9216

OPEN

Magnetic fingerprint of individual Fe₄ molecular magnets under compression by a scanning tunnelling microscope

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Single-molecule magnets (SMMs) present a promising avenue to develop spintronic technologies. Addressing individual molecules with electrical leads in SMM-based spintronic devices remains a ubiquitous challenge: interactions with metallic electrodes can drastically modify the SMM's properties by charge transfer or through changes in the molecular structure. Here, we probe electrical transport through individual Fe₄ SMMs using a scanning tunnelling microscope at 0.5 K. Correlation of topographic and spectroscopic information permits identification of the spin excitation fingerprint of intact Fe₄ molecules. Building from this, we find that the exchange coupling strength within the molecule's magnetic core is significantly enhanced. First-principles calculations support the conclusion that this is the result of confinement of the molecule in the two-contact junction formed by the microscope tip and the sample surface.

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Molecular spintronics harnesses magnetic properties of molecules to achieve enhanced functionality in electronic circuits¹. The use of single-molecule magnets (SMMs) with long spin-relaxation times may even enable spin-based quantum computing^{2,3}. Practical incorporation of an SMM into a two-contact device demands strong coupling to the spin subsystem without disrupting the magnetic properties with unwanted electronic or structural modifications^{4–8}. Metal-molecule-metal junctions constructed by electro-migration have demonstrated the coupling between electric current and molecular magnetic moments, as well as the persistence of magnetic anisotropy within the device junction for $[\text{Mn}_{12}\text{O}_{12}(\text{O}_2\text{CR})_{16}(\text{H}_2\text{O})_4]$ (Mn_{12}), $[\text{Fe}_4(\text{L}')_2(\text{dpm})_6]$ and TbPc_2 SMMs^{9–12}. Indirect electronic contact to TbPc_2 using carbon-nanotube and graphene-based devices showed signs of slow magnetization dynamics^{13,14}. However, on metallic surfaces the magnetic bistability of TbPc_2 is quenched^{4–6} and recovers only for significant separation of molecule and surface⁷. Mn_{12} proved to be exceedingly fragile⁸ unless the molecules are isolated by a protective layer¹⁵. In contrast, our focus here is directed to the four exchange coupled Fe atoms of the tetrairon(III) (Fe_4) system which are encased in a robust, rigid, three-dimensional (3D) organic ligand shell. Fe_4 allows significant flexibility in customization of ligands¹⁶ and stable magnetism has been detected in ensembles of both chemically grafted^{17,18} and sublimated^{19,20} Fe_4 derivatives on metallic surfaces.

Here, we use a low-temperature scanning tunnelling microscope (STM) to study the magnetic properties of Fe_4 molecules sublimated onto the surface of a $\text{Cu}_2\text{N}/\text{Cu}(100)$ substrate. The STM tip is used to address individual Fe_4 molecules and probe excitations of the Fe_4 SMM's electron spin by inelastic electron tunnelling spectroscopy (IETS)²¹. Hindering this is the 3D nature of the molecule and its ligand shell. In particular, tip interaction during spectroscopic measurements is extremely strong, commonly leading to molecular fragmentation. We overcome this problem by implementing analysis that correlates independent metrics (topography and IETS spectrum) to

categorize the magnetic fragments on the surface and identify intact molecules. Analysis of the spectrum identified for intact Fe_4 molecules shows that the exchange energy within the magnetic core of the molecule is boosted by a significant margin. From first-principles calculations, we find that the enhanced exchange interaction between the Fe ions can be explained by a small compression of the magnetic core. We attribute this compression to the confinement of the molecule in the two-contact junction formed by the STM tip and the substrate surface.

Results

Fe_4 evaporated on Cu_2N . The Fe_4 derivative used in this work ($[\text{Fe}_4(\text{L})_2(\text{dpm})_6]$, where H_3L is the tripodal ligand $\text{Ph-C}(\text{CH}_2\text{OH})_3$ and Hdpm is dipivaloylmethane; Fig. 1a) permits thermal sublimation of isolated molecules onto a semi-insulating copper nitride (Cu_2N) surface on Cu (100) that has been pre-cooled (see Methods for details)^{19,20}.

After deposition onto the Cu_2N surface, and immediate cooling below 1 K, constant-current topographs reveal molecular objects with a wide variety of morphologies (Fig. 1b). Density functional theory (DFT) computations (see Methods for details) of Fe_4 relaxed on a Cu_2N slab indicate that molecules adsorb with the axis of approximate 3-fold rotational symmetry (pseudo- C_3 axis) canted at a 33° angle from the surface normal, and with a total height of 1.7 nm (Fig. 1a,c). The tallest objects on the surface appear as spheroids of width 2 nm and height between 700 and 800 pm in STM scans (Fig. 1d). Unambiguously identifying these objects as intact Fe_4 SMMs is not possible by topographic measurements alone because the fine, multi-lobed structure is strongly tip-dependent.

Spectroscopic measurements. To corroborate the coarse topographic match we record inelastic electron tunnelling spectra with the tip brought into contact with the molecules. The differential conductance, dI/dV , is measured as a function of bias voltage, V , starting at reduced bias voltage (10 mV) and increased tunnel

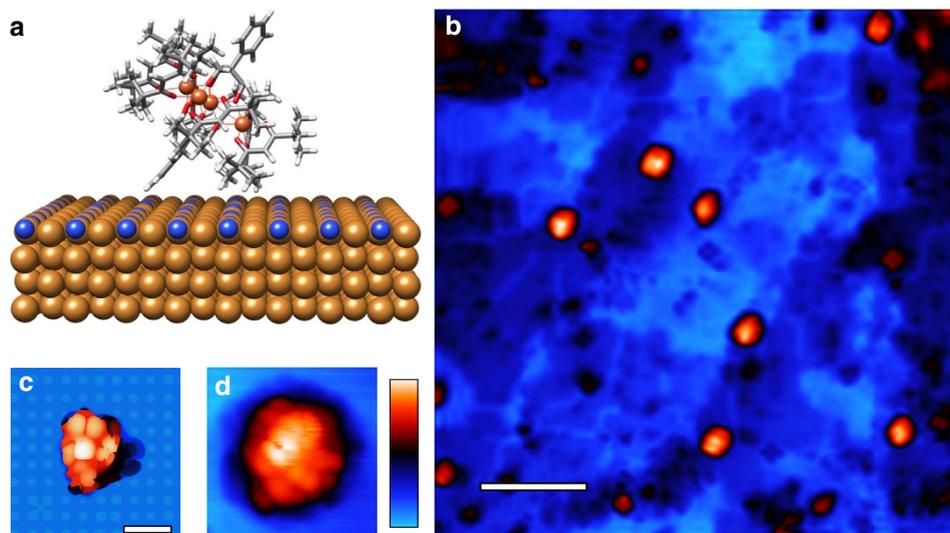


Figure 1 | Fe_4 molecule adsorbed on the Cu_2N surface. (a) $[\text{Fe}_4(\text{L})_2(\text{dpm})_6]$ resting on the Cu_2N surface. Adsorption geometry and molecular structure are computed by density function theory (DFT). The molecule's axis defined by the tripodal ligands is at a 33° angle from the surface normal. Atoms are Fe (orange), O (red), C (grey), H (white), Cu (brown) and N (blue). (b) Overview scanning tunnelling microscope (STM) image of the Cu_2N surface after deposition of molecules (scale bar, 8 nm). A number of Fe_4 molecules are visible as the tallest objects in orange. This image was filtered to remove noise using WSxM software²³. (c) Calculated top view image of relaxed Fe_4 on Cu_2N showing the spatial distribution of the density of states integrated between 0 and +3 eV in energy. (d) STM image of an Fe_4 molecule. It appears as a spheroid of ~ 2 nm diameter with a multi-lobed substructure consistent with the calculated image in (c). The colour scale indicates the topographical height ranging between 0 and 1 nm in (b,d), which were acquired at a tunnel current set-point of 3 pA and bias voltage of 2.3 V. The 1-nm lateral scale bar inset in (c) also applies to (d).

current (5–100 pA; see Methods). Over a molecule at typical scanning conditions (2 V, 3 pA), we estimate the tip-Cu₂N gap to be 1.5 ± 0.1 nm (see Supplementary Fig. 1 and Supplementary Note 1); this matches well with the expected molecule height. Under scanning conditions transient physical tip-molecule interactions occur, confirming that the tip comes into contact with the top of the molecule. Transitioning to spectroscopy conditions moves the tip towards the molecule by 700–800 pm (see Supplementary Fig. 1). Hence, current passes directly through the molecule, as it is sandwiched into a two-contact device formed by the STM junction. Figure 2a shows a representative spectrum acquired on one of the molecules so trapped. It features clear steps in $dI/dV(V)$ at ± 0.5 mV and at ± 7.5 mV that stem from excitations of the molecule's electron spin. We verify the magnetic nature of these excitations by acquiring data on individual molecules at both zero field and under an out-of-plane 9 T field, Fig. 2b. Both excitations shift in energy with magnetic field, qualitatively consistent with Zeeman energies expected for electron spins in a 9 T magnetic field.

Different molecules, however, exhibit a large variation of the spin excitation energies. It is not, *a priori*, clear if these variations might stem from measurements on partially fragmented molecules or interaction with the electrodes. This uncertainty can be overcome by correlating topographic height with observed spin excitations over a statistically significant population of molecules (>60).

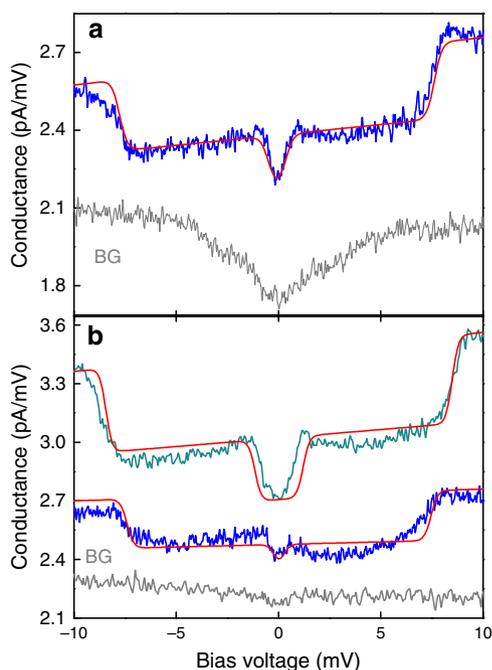


Figure 2 | Conductance spectra of individual molecules. (a) $dI/dV(V)$ spectrum recorded on a molecule with 700 pm topographic height at 0 T magnetic field (blue line, initial current $I_0 = 75$ pA at $V_0 = 10$ mV). A background (BG) spectrum recorded with the same tip on bare Cu₂N is shown in grey. (b) Spectra acquired on a single molecule (800 pm height) without magnetic field (blue line, $I_0 = 25$ pA, $V_0 = 10$ mV) and under a 9 T out-of-plane field (green line, $I_0 = 50$ pA, $V_0 = 15$ mV), background spectrum (BG, grey line). When a magnetic field is applied, the low-energy excitation widens from 0.2 ± 0.2 to 0.8 ± 0.1 mV and the high-energy excitation widens from 7.5 ± 0.1 to 8.5 ± 0.1 mV. Background spectra in (a) and (b) are offset for clarity. Red lines in (a) and (b) indicate spectra computed using the spin Hamiltonian model. The fits yield exchange coupling, $J = 2.93$ meV (23.6 cm⁻¹), and magnetic anisotropy, $D = -52$ μ eV (-0.42 cm⁻¹) for the molecule in (a) and $J = 2.89$ meV (23.3 cm⁻¹), $D = -26$ μ eV (-0.21 cm⁻¹) with a g factor of 2 for the molecule in (b).

Figure 3a shows a 2D histogram of the spin excitation steps detected at zero magnetic field, binned by excitation voltage and topographical height of each molecule, measured following the spectroscopic measurement. Since, spin excitations are symmetric in energy, we plot symmetrized data (for comparison see nonsymmetrized data in Supplementary Fig. 2). Clear groupings of excitations appear correlated with height. In particular a distinct set of peaks is visible for molecules taller than 700 pm (Fig. 3b), corresponding to two spin excitations with characteristic energy $\Delta E_1 = 0.47$ meV and $\Delta E_2 = 7.2$ meV, and distribution standard deviations of 0.14 and 0.7 meV, respectively. The breadths of the observed distributions are larger than the experimental error on excitation energies (0.1 meV), therefore they represent variation in the expression of the molecule in the junction. This emergent spectrum is linked to molecules that feature a coarse topographical match with the computed DFT structure, thus revealing the spin excitation fingerprint of Fe₄ on Cu₂N.

Among shorter molecules measured in Fig. 3a, the peaks are more scattered, supporting the hypothesis that these are disrupted or fragmented molecules. The exact chemical configuration of the fragments remains unknown but the most commonly observed spectra qualitatively match those expected for clusters comprising two and three Fe atoms with coupling parameters similar to intact molecules (see Supplementary Fig. 3 and Supplementary Note 2). The potentially insidious role magnetic fragments may play in the interpretation of experiments on bulky, fragile molecules can thus be mitigated by identifying intact molecules via correlation of independent metrics, such as spectral information and topographic height.

Modelling of spin excitations. Magnetic properties of Fe₄ are quantitatively investigated by fitting the spin excitation energies extracted from Fig. 3b to an effective spin Hamiltonian that incorporates the dominant exchange coupling between the central Fe ion and the side ions and net second-order uniaxial anisotropy (equation 1). Less influential contributions from next-nearest neighbour exchange, rhombic anisotropy and higher-order anisotropies are neglected. The energy eigenstates of the simplified Hamiltonian deviate by less than the measurement accuracy from a more complex spin Hamiltonian reported previously¹⁶ (see Supplementary Note 3 and Supplementary Fig. 4).

$$\hat{H} = \sum_{i=1,2,3} J \hat{S}_i \cdot \hat{S}_c + g \mu_B \hat{\mathbf{B}} \cdot \hat{S}_T + D \hat{S}_{T,z}^2 \quad (1)$$

\hat{S}_i and \hat{S}_c denote the spin vector operators for the three side ions and for the central ion respectively, where each ion has a spin 5/2. The total spin operator of the molecule, \hat{S}_T , is given by $\hat{S}_T = \hat{S}_c + \sum_{i=1,2,3} \hat{S}_i$ and $\hat{S}_{T,z}$ is its component along the easy magnetic axis (z). Antiferromagnetic Heisenberg exchange coupling, $J > 0$ (Fig. 3c inset), is mediated by the oxygen bridges that connect the central and side ions and leads to a ferrimagnetic configuration and a $S_T = 5$ ground state. The next higher energy multiplet features $S_T = 4$; other multiplets lie at still higher energies. Uniaxial anisotropy, D , parallel to the tripodal ligands, splits spin states in each multiplet into a parabolic distribution typical of easy-axis ($D < 0$) molecular magnets (Fig. 3c). A simplifying assumption that each multiplet shares the same D value is made (see Supplementary Note 3). Zeeman energy is included for spectra recorded at 9 T field; g is the Landé g factor and μ_B is the Bohr magneton. The out-of-plane magnetic field, \mathbf{B} , is tilted by 33° from z to account for the orientation of the molecule on Cu₂N.

Spin excitations with tunnelling electrons obey the selection rule $\Delta m = \{+1, 0, -1\}$ ($\hbar \Delta m$ is the change in the expectation value, $\hbar m$, of the $\hat{S}_{T,z}$ operator)²². At 0.5 K, only the ground

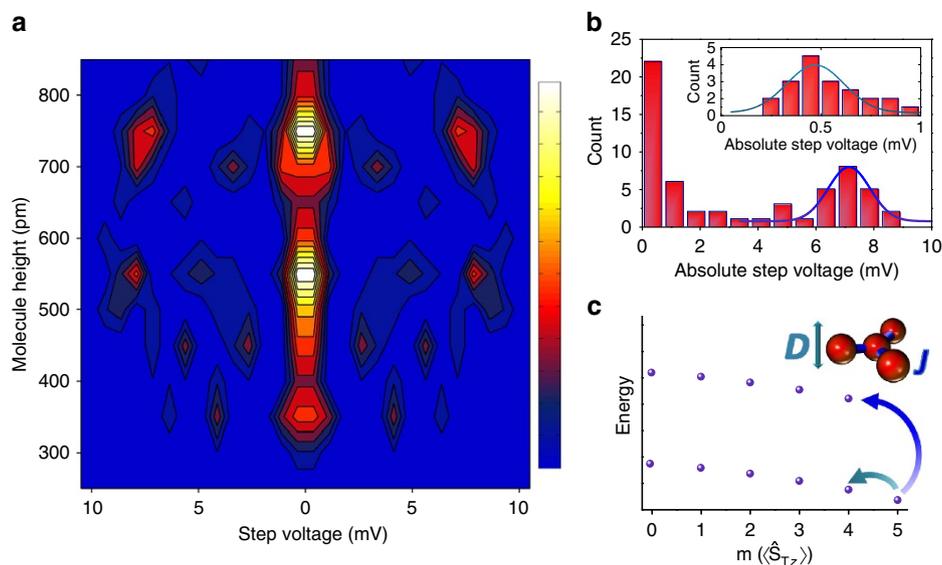


Figure 3 | Spin excitation fingerprint of Fe₄ on Cu₂N. (a) Two-dimensional histogram for an ensemble of molecules (> 60) correlating zero-field spin excitation energies of a molecule with its topographic height. The histogram counts the number of inelastic tunnelling steps observed in $dI/dV(V)$ binned by step voltage and topographic height of the molecule measured after completion of each spectrum. For molecules over 700 pm in height, a dominant spectrum can be identified. For shorter objects large variation indicates significant changes in magnetic structure and possible fragmentation. Bin sizes are 0.75 mV and 50 pm. The colour scale indicates bin count between 1 and 11. Since spin excitations are symmetric with respect to 0 V, the absolute value of the steps is used, and the symmetrized plot is shown. (b) Histogram of all steps measured on molecules > 700 pm in height. Focusing on those molecules reveals the characteristic spectrum for intact Fe₄ molecules, which features two spin excitations: a low-energy excitation at 0.47 mV and a high-energy excitation at 7.2 mV. Both have broad distributions with standard deviations of 0.14 and 0.7 mV, respectively reflecting variations in spin excitation energies for different molecules. (c) Spin state distribution for Fe₄ single-molecule magnets. States are calculated using a model of the magnetic core (inset) incorporating antiferromagnetic exchange coupling of the three outer Fe ions to the central ion (orange balls) with strength J (blue bonds) and easy-axis magnetic anisotropy with strength D applied to the whole molecule (teal arrow). The characteristic excitations found in (b) are consistent with the two lowest energy transitions excitable by inelastic electron tunnelling (curved teal and blue arrows) with J between 2.5 meV (20 cm⁻¹) and 3.1 meV (25 cm⁻¹), and D between $-34 \mu\text{eV}$ (-0.28 cm^{-1}) and $-86 \mu\text{eV}$ (-0.70 cm^{-1}).

doublet with $m = \pm 5$ is occupied. Therefore, the two characteristic excitations found for Fe₄ can be linked to specific spin transitions: ΔE_1 is a low-energy transition within the $S_T = 5$ spin multiplet from $m = \pm 5$ to ± 4 , and ΔE_2 is a transition to the lowest-lying states of the $S_T = 4$ multiplet. Fitting ΔE_1 and ΔE_2 to the above spin Hamiltonian yields a uniaxial anisotropy energy of $D = -60 \pm 26 \mu\text{eV}$ ($-0.48 \pm 0.21 \text{ cm}^{-1}$) and exchange coupling energy of $J = 2.8 \pm 0.3 \text{ meV}$ ($23 \pm 2 \text{ cm}^{-1}$). We note that the standard deviations given here reflect the variation in D and J among the ensemble of molecules.

Discussion

The J value observed for molecules sandwiched in the two-contact junction of the STM is significantly higher than for bulk samples of Fe₄ molecules, where $J = 1.92 \text{ meV}$ (15.5 cm^{-1})¹⁶ for the spin Hamiltonian in equation 1 (see Supplementary Note 3). A scenario with lower J leading to the observed steps in dI/dV would require ΔE_2 to be an excitation into higher-lying multiplets such as $S_T = 6$ but this can be ruled out by the observed magnetic-field-dependent energy shift of ΔE_2 (see Supplementary Note 4). Similarly, a change in the magnetic properties of the molecule due to a change in the molecule's redox state as predicted for other types of molecules²⁴ is unlikely. Changing the redox state would require accessing either the highest occupied or lowest unoccupied molecular orbitals of the Fe₄ ligand shell. These are separated from the Fermi energy by several electronvolts and are not accessible during the IETS measurements performed here. Consequently, there must be some other mechanism that enhances exchange within the molecule.

We explore the influence of the Cu₂N surface on the magnetic properties of the molecule via DFT calculations. Starting with the relaxed geometry computed for Fe₄ on Cu₂N, the magnetic properties may be computed using the broken symmetry approach²⁵ (see Methods for details on DFT calculations). The structural relaxation of the molecule is found to reduce the exchange to 1.50 meV (12.1 cm^{-1}), $\sim 20\%$ below the experimental value for bulk molecular crystals. Inclusion of the electronic effects of the substrate further reduces the exchange to 1.17 meV (9.4 cm^{-1}). The most important information gained here is that the structural changes induced by the surface and the interaction of the substrate electron bath both feature the same qualitative trend. Hence, the influence of the substrate is not responsible for the boosted exchange.

We reason that the increase in J must be induced by the confinement of the molecule in the narrow STM junction. Hydrostatic compression has been found to alter the intermolecular exchange in bulk crystals of organic ferromagnets²⁶. In Fe₄, exchange coupling between the central and side ions is predominantly mediated by the oxygen atoms of the tripodal ligands. A distortion of these ligands relative to the plane of the Fe ions provides a direct path to change the Fe-O-Fe angles¹⁶ and, consequently, the exchange coupling strength, J . We therefore consider the effect of distorting this ligand during compression of the molecule. We incorporate a rudimentary representation of this distortion into the DFT framework. Starting with the structure of the molecule, as relaxed on Cu₂N, the upper tripodal ligand is shifted downwards, parallel to the axis of the molecule by 10 pm (Fig. 4). This is equivalent to compressing the molecular core by 2% of its starting breadth. Re-evaluating the

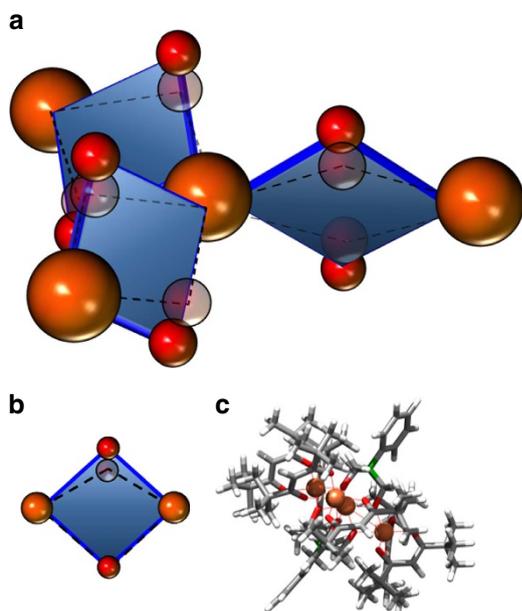


Figure 4 | Structural distortions of Fe₄ in the STM tunnel junction. The strong contact made by the tip induces distortions as the molecule is compressed in the STM junction and enhances super-exchange.

(a) Schematic of the magnetic core of the Fe₄ molecule. Key to the super-exchange coupling within the magnetic core are the Fe-O-Fe bond angles. The blue diamonds represent the planes of these bonds; oxygen atoms are red and iron atoms are orange. Compression of the molecule by the STM tip displaces the O atoms relative to the Fe atoms causing a tilt of the Fe₂O₂ bond planes and a change in the Fe-O-Fe bond angle. The distorted configuration is shown superimposed, semi-transparent with dashed lines. (b) Schematic depicting the small distortion applied for DFT calculation of exchange over a Fe₂O₂ unit. (c) Model of the compressed molecule used in DFT calculations. Atoms are Fe (orange), O (red), C (grey) and H (white). To simulate the compression inside the STM junction, the upper tripodal ligand (highlighted in green) is shifted downwards parallel to the molecule axis by 10 pm, equivalent to a 2% reduction in breadth of the magnetic core. This structural shift nearly doubles the computed exchange coupling.

exchange coupling and comparing to the value computed for the undistorted molecule isolated from the surface, we find that this minimal structural change increases J by a factor of 1.9, to a value of 2.82 meV (22.7 cm⁻¹). The magnitude of the increase is sufficient to explain the boosted exchange interaction found for Fe₄ in the STM junction.

In the experiment, the complexity of the situation is considerably greater. Along with other environmental effects there must be more complex distortions occurring. This is corroborated by the large variation of D and J observed for the ensemble of molecules (Fig. 3b). The 1 σ standard deviation of ΔE_2 relates to a variation in J of 10% (± 0.3 meV). We therefore consider a range of other possible distortions (see Supplementary Fig. 5 and Supplementary Note 5) and find that rotating the tripodal ligand relative to the Fe ions and tilting of the protruding phenyl rings also affects J , albeit significantly less than the simple 2% compression (see Supplementary Tables 1 and 2). Other random contributions to the broad distribution of excitation energies are possible. No selection was made for differing positions on the Cu₂N surface and it was not possible to inspect the surface beneath the molecules for the presence of defects, such as nitrogen vacancies. Consequently, a portion of the random variation may arise from local differences in substrate coupling, although this will be mitigated by the molecule's protective ligand

shell. We additionally expect that the broad distribution found for ΔE_1 reflects the influence of structural distortions on the molecular anisotropy. However, in contrast to the effect on exchange energy, no systematic shift was found experimentally.

Despite the broad distribution of parameters, individual Fe₄ SMMs retain their qualitative magnetic character when incorporated into a prototypical two-contact device formed by an STM tunnel junction. Ascertaining this is made possible by performing an exhaustive survey of molecular objects. Correlation of spin excitation spectra with topographic height is a critical step in the study of bulky, polynuclear SMMs. By using this approach, we find that Fe₄ molecules in the STM junction feature increased exchange coupling strength that can be accounted for by tip-induced structural distortions. While demonstrating that Fe₄ may be incorporated into a prototypical device, this work also addresses key challenges in combining electronic transport devices with SMMs. Not only can the molecule's bulky ligand shell aid in the preservation the molecule's magnetic properties, but it can also participate in strong mechanical interactions with the electrodes. Confronting and overcoming these challenges facilitate the rational design of SMMs and creates the opportunity for identification of novel effects, such as the possibility of tuning intra-molecular properties by mechanical motion.

Methods

Experiment. Observations were made with an ultrahigh-vacuum low-temperature STM (Unisoku USM1300) equipped with a ³He cryostat operated at 0.5 K, and with a magnetic field of up to 9 T applied perpendicular to the sample surface. The copper crystal was cleaned using cycles of Ar sputtering (1 kV, pressure = 5×10^{-6} mbar) and annealing ($T = 870$ K) using an electron-beam heater built into the sample holder. The final annealing cycle used a lower temperature of 770 K. The copper nitride monolayer was formed by three minutes of nitrogen sputtering (1 kV, pressure = 5×10^{-6} mbar) followed by annealing for 8 min at 640 K. Platinum iridium (90:10) tips were used to perform all STM measurements. Tips were prepared by 5 min of Ar sputter cleaning (1 kV, pressure = 5×10^{-6} mbar) followed by heating using 10 s pulses of electron beam bombardment and a gentle dipping into the Cu crystal.

Fe₄ molecules, synthesized and isolated in crystalline form¹⁶, were mechanically ground into a powder and sublimated from a home-built quartz Knudsen cell maintained at a constant temperature of 483 K. Deposition times ranged between 10 and 30 s. The Cu₂N/Cu substrate was pre-cooled to 4 K in the STM cryostat, then moved rapidly to the deposition chamber and back to the cryostat; during this procedure the sample temperature reached an estimated maximum of 100 K.

Acquisition of spin excitation spectra. Inelastic electron tunnelling spectra were acquired by positioning the tip over candidate molecules at typical scanning conditions (3 pA, 2.0–2.5 V). Subsequently, the bias voltage was reduced slowly to 10–15 mV, then the tunnel current set-point was increased to 5–100 pA. Following this procedure, the feedback loop was disengaged, and the sample bias voltage was swept. A small modulation voltage (100–300 μ V) at 691 Hz was added to permit direct lock-in detection of differential conductance, $dI/dV(V)$.

Calculated spin excitation spectra. The conductance spectra were calculated from the eigenstates and eigenenergies of the effective spin Hamiltonian (equation 1) using a perturbative treatment of inelastic electron-spin scattering²⁷. In this approach, a spin excitation between state $|2\rangle$ and $|1\rangle$ with ΔE energy appears as steps in $dI/dV(V)$ at $+\Delta E/e$ and $-\Delta E/e$ voltage. The height of the conductance steps is proportional to the transition matrix element $\sum_j |(1|\hat{\sigma} \cdot \hat{S}_j|2)|^2$, where $\hat{\sigma}$ is the spin vector operator of the tunnelling electron and \hat{S}_j are the spin vector operators of the Fe atoms in the Fe₄ molecule. The distribution of scattering strength between the four Fe atoms influences the relative height of the spin excitations steps. For simplicity, we assumed that scattering occurs with equal probability at any of the side atoms. Other ratios are possible but do not influence the voltage position of the conductance steps.

The calculated conductance spectra were used to fit experimental data by iterative computation of the spectrum while changing the exchange (J) and anisotropy (D) parameters in the spin Hamiltonian describing the Fe₄ molecule (equation 1). Broadening of the conductance steps by finite temperature and bias modulation was taken into account. This led to step width of 400–600 μ V. A linear slope was added to the conductance spectra following the fitting. Calculations that account for the out-of-plane magnetic field used a 9 T magnetic field applied at 33° from the molecular easy axis.

DFT calculations. The relaxed structure of the Fe₄ molecule adsorbed on the Cu₂N surface was computed within the DFT framework with the Cp2k program package^{28,29} using the Dudarev simplified version³⁰ of the DFT + U approach^{31,32} together with the PBEsol functional³³.

The calculations followed methodology developed for SMMs on surfaces³⁴ and used successfully for Fe₄ molecules on Au surfaces¹⁹. The dispersion correction term (D3) was added to the energy³⁵. The norm-conserving Goedecker-Teter-Hutter (GTH) pseudopotentials³⁶ were used with GTH double- ζ polarized molecularly optimized basis sets for all atomic species³⁷. The energy cutoff applied to the plane-wave basis sets was set to 500 Ry, in agreement with other studies on Cu₂N³⁸. The values of the parameter U , 4.1 eV on the Fe $3d$ and 3.0 eV on the O $2p$ orbitals were chosen following other calculations involving Fe₄ molecules³⁹. The convergence criterion for the self-consistent field method (SCF) energy was 1×10^{-6} Hartree. A threshold of 1×10^{-3} Hartree Bohr⁻¹ for the atomic forces is considered sufficient for a reliable optimization. Optimization runs used a 0.17 eV smearing of the occupational numbers around the Fermi level to account for the metallicity of the surface, and to ease the convergence. For calculation of magnetic parameters, using the optimized geometry as a starting point, a U parameter may be added to the $3d$ states of the Cu atoms to ascertain the impact of the substrate on magnetic interactions within the Fe₄ cluster.

The cluster and surface were relaxed separately before optimizing the adsorbed structure. The Cu₂N surface slab consisted of $8 \times 8 \times 2$ copper unit cells (width, breadth and thickness, respectively) and a layer of nitrogen atoms, positioned as starting geometry at 0.3 Å above the topmost copper layer in a $c(2 \times 2)$ manner. Out of the total four atomic copper layers, the bottom two were kept fixed to the Cu bulk positions during the geometry optimization. The experimental copper lattice constant of 3.615 Å was used⁴⁰. The cluster was positioned on the relaxed Cu₂N surface, leaning on two tert-butyl groups and the phenyl ring¹⁹. It was separated from its periodic images laterally by 13 and 14 Å, in x and y respectively, while the vacuum region extended 38 Å above the surface.

The exchange coupling constants in Fe₄ were computed through the broken symmetry approach^{25,41,42} incorporating exchange interactions between nearest and next-nearest neighbours. For comparison to the giant spin Hamiltonian used in the main text, an equivalent nearest neighbour exchange coupling was calculated from the computed values of both the nearest and next-nearest neighbour coupling constants (See Supplementary Note 3 for details). The magnetic properties were evaluated on the optimized geometry with a tighter SCF convergence criterion of 5×10^{-7} Hartree, computing the energies of the determinants |uuuu>, |duuu>, |duuu>, |juudu> and |juudu> (u stands for spin up ($m = 5/2$), d for spin down ($m = -5/2$) on the four Fe ions). The interacting topmost layers of the substrate slab can be approximated as Cu₂N, consequently, $U_{Cu} = 5 \text{ eV}^{43,44}$ was applied to all Cu atoms as reported previously in the literature. To verify the reliability of the computational protocol, the exchange coupling constants were also computed for the geometry determined from X-ray measurements and compared with experimental measurements (see Supplementary Table 1). To decouple the geometrical effects of adsorption from the electronic contribution of the surface, the magnetic couplings were computed for the relaxed molecule in the presence of the Cu₂N surface and with the slab removed. In the distorted molecule compression was incorporated as a rigid shift of the upper tripod ligand, with no further geometric relaxation.

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Acknowledgements

J.A.J.B. acknowledges support from the Alexander von Humboldt foundation and the Natural Sciences and Engineering Research Council of Canada. J.A.J.B., S.Y., S.R.-P. and S.L. acknowledge E. Weckert and H. Dosch (Deutsches Elektronen Synchrotron, Germany) for providing lab space. L.M., S.N., V.L., M.M., A.C., F.T. and R.S. acknowledge the support of ERC through AdG MolNanoMaS (267746) and of Italian MIUR (FIRB project RBAP117RWN).

Author contributions

S.L., R.S. and J.A.J.B. designed the experiment. A.C. synthesized the molecules. L.M., V.L. and M.M. characterized the sublimation of the molecules. J.A.J.B., L.M. and S.L. performed the STM experiments with assistance from S.Y. The spin Hamiltonian was constructed by J.A.J.B., A.C., R.S. and S.L. Analysis of STM data was carried out by J.A.J.B. DFT calculations were performed by S.N. and F.T. All authors contributed to interpretation of the results and the manuscript.

Additional information

Supplementary Information accompanies this paper at <http://www.nature.com/naturecommunications>

Competing financial interests: The authors declare no competing financial interests.

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How to cite this article: Burgess, J. A. J. *et al.* Magnetic fingerprint of individual Fe_4 molecular magnets under compression by a scanning tunnelling microscope. *Nat. Commun.* **6**:8216 doi: 10.1038/ncomms9216 (2015).



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