

NOTE

Isolation and characterization of spliceostatin B, a new analogue of FR901464, from *Pseudomonas* sp. No. 2663

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FR901464 (Figure 1), a prototype pre-mRNA splicing inhibitor isolated from the culture broth of *Pseudomonas* sp. No. 2663, is a potent but unstable cytotoxic compound.^{1–4} Spliceostatin A (Figure 1) is a methylated and more stable derivative of FR901464, shown to bind noncovalently to the splicing factor 3b subcomplex of the U2 small nuclear ribonucleoprotein particle of mammalian spliceosome, thus inhibiting pre-mRNA splicing and causing pre-mRNA leakage to the cytoplasm.^{5,6} Recently we discovered thailanstatins A, B and C as three new and significantly more stable natural analogs of FR901464 from the culture broth of *Burkholderia thailandensis* MSMB43.⁷ For evaluations of the stability, pre-mRNA splicing inhibitory activity and cytotoxicity of thailanstatins, we used FR901464 as a reference compound, which was freshly purified from the *Pseudomonas* sp. No. 2663 fermentation. During this purification optimization process, another analogous compound of FR901464, named spliceostatin B (1), was discovered. Herein we report the fermentation, isolation, chemical characterization and cytotoxicity of this new natural product.

The FR901464-producing strain *Pseudomonas* sp. No. 2663² was purchased from the National Institute of Bioscience and Human-Technology, Agency of Industrial Science and Technology, Japan. It was routinely activated from a glycerol stock on Luria-Bertani (LB) agar at 30 °C for 2 days as a starting plate. Several colonies from the plate were inoculated into a flask containing 250 ml of LB medium and incubated at 30 °C for 24 h on a rotary shaker (150 r.p.m.) to prepare a seed culture. For fermentation, the seed culture was inoculated at 2% (v/v) to each of the two fermentors (BioFlo IV, 20 l total volume, New Brunswick Scientific, Enfield, CT, USA); each contains 12 l of production medium consisting of soluble starch 1%, glycerin 1%, defatted soybean meal 1%, glucose 0.5%, corn steep liquor 0.5%, (NH₄)₂SO₄ 0.2%, CaCO₃ 0.2%, MgSO₄·7H₂O 0.006% and antifoam-204 0.01% (pH 7.0), slightly modified from the

literature.² The pH of bacterial culture was automatically maintained by the fermentor with 1N HCl or 1N NaOH, and the fermentation was proceeded at 25 °C for 48 h with an agitation of 200 r.p.m. and an air flow-rate of 4 l min⁻¹. The repeated fermentation with two fermentors led to an accumulation of 72 l of fermentation broth.

Fermentation broth was extracted three times with ethyl acetate (3:2, v/v), and the extracts were pooled and concentrated to dryness with a rotary evaporator at 35 °C. The resulting crude extract (25.2 g) was subjected to two steps of silica gel chromatography and one step of octadecyl-silica C₁₈ chromatography on an YFLC AI-580 flash chromatography system (Yamazen, San Bruno, CA, USA), with elution monitored at UV 235 nm (see Supplementary Information: Scheme S1). In the first step, each 5 g of crude extract was suspended in 25 ml of ethyl acetate and mixed with 30 g of silica gel and packed into an injection column (3.0 × 12.5 cm, Catalog No. W830 silica gel, Yamazen), which was mounted atop a silica gel Universal Column (4.8 × 18.5 cm, 200 g silica gel, 40 μm, 60 Å). The column system was sequentially eluted by 1.1 l of each of the following solvents: hexane, hexane:ethyl acetate (3:1, v/v); hexane:ethyl acetate (1:1, v/v); ethyl acetate, ethyl acetate:acetone (1:1, v/v); and acetone, all at a flow rate of 50 ml min⁻¹. The ethyl acetate fraction containing 1 was concentrated to dryness at 35 °C. In the second step, the above resulting residue was suspended in 15 ml of acetone, mixed with 15 g of silica gel and packed into an injection column (2.0 × 6.5 cm, 14 g silica gel), which was mounted atop a silica gel Universal Column (2.6 × 12.0 cm, 40 g silica gel, 40 μm, 60 Å). The column system was sequentially eluted with chloroform, and 1, 2, 4 and 10% of acetone in chloroform, and finally acetone at 18 ml min⁻¹. In the third step, the 1% acetone fraction containing 1 was again concentrated and further fractionated on an octadecyl-silica C₁₈ column system equipped with an injection column (2.0 × 6.5 cm, 14 g gel) and a

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Universal Column (2.0 × 8.0 cm, 14 g gel, 50 μm, 120 Å). After loading the sample, the column system was first eluted with 5% acetonitrile for 3 min, then by a linear gradient of acetonitrile from 5 to 100% in 10 min, and finally by 100% acetonitrile for 14 min, with a flow rate of 20 ml min⁻¹. The fraction containing 1 was eluted at 15.7 min. In the final purification step, acetonitrile solution of 1 was purified with a Varian ProStar HPLC system (210 binary pump and 330 photodiode array detector, Varian, Palo Alto, CA, USA) equipped with an Agilent Prep-C18 column (21.2 × 250 mm, 10 μm, Agilent, Santa Clara, CA, USA) to give 28 mg of pure 1. As shown in Supplementary Figure S1, 1 was present in significant amount in the culture grown in fermentor but barely in that grown in shaking flask, suggesting that production of 1 is heavily culture condition-dependent.

The structure of 1 (Figure 1) was determined using a combination of UV, IR, HR-MS and NMR spectroscopic analyses (Supplementary Figures S2–S14) and was found to be a new compound after a SciFinder database search.

1 was obtained as a white powder and its physicochemical properties are summarized in Table 1. The IR spectra indicate the presence of hydroxyl (3355 cm⁻¹), carboxyl (2978 and 2938 cm⁻¹), carbonyl (1735 cm⁻¹), amide carbonyl (1667 cm⁻¹) and conjugated diene moieties (1635 cm⁻¹). HR ESI-MS revealed a quasi-molecular

ion peak of *m/z* 504.2951 for C₂₈H₄₂NO₇ [M + H]⁺ (calculated 504.2955) and suggested 503 as the MW and C₂₈H₄₁NO₇ as the molecular formula.

The ¹H and ¹³C NMR spectra in combination with ¹H–¹³C HSQC NMR data (Table 2) of 1 exhibited signals of five methyl groups (CH₃-16, CH₃-20, CH₃-21, CH₃-5' and CH₃-2''), six methylene groups (CH₂-2, CH₂-4, CH₂-10, CH₂-13, CH₂-17 and CH₂-19) and 12 methine group (CH-1, CH-5, CH-11, CH-12, CH-14, CH-15, CH-6, CH-7, CH-9, CH-2', CH-3' and CH-4'), as well as the two quaternary carbons (C-3 and C-8) and three carbonyl carbons (C-18, C-1' and C-1'').

The key COSY and HMBC data were shown in Figure 2a. When the ¹H and ¹³C NMR data were compared with those of FR901464,³ the spectra showed overall similarities except for the absence of a methyl group and a hydroxyl group at the C1 and C4 positions, respectively. On the basis of molecular formula and ¹³C chemical shifts, 1 has three more chemical shifts, where two (δ_C 38.7 p.p.m. and δ_C 174.7 p.p.m.) are attached to C1 atom, which was confirmed by the observation of correlations between 1-H and 17-H, and 1-H and 2-H (Table 2) in COSY spectrum. Another methylene group (δ_C 39.4 p.p.m. and δ_H 2.48 p.p.m., and δ_H 2.23 p.p.m.) indicated that this CH₂ is free of hydroxyl groups. Comparing with the methylene group (δ_C 48.0 p.p.m.) at C18 position in FR901464, the highly shifted methylene group (δ_C 111.4 p.p.m.) at C19 position indicated that it was a terminal methylene group with a double bond. According to the molecular formula and the already known NH group, there should be one hydroxyl group in the structure, which was not observable in the ¹H NMR spectrum. The presence of such a hydroxyl group in the terminal carboxyl moiety of 1 was confirmed by a positive bromocresol green visualization reaction (yellow spot on dark blue background) on TLC⁸ (Supplementary Figure S15).

The relative configuration of 1 was determined to be the same as that of FR901464. The geometry of C2' and C6 double bond was proposed as *cis* (*Z*) based on the vicinal coupling constants *J*_{H2'-H3'} = 11.6 Hz and *J*_{H6-H7} = 15.8 Hz (Table 2). The *trans* (*E*) configuration of the double bonds at C8–C9 was indicated by the chemical shift of C20 at 12.6 p.p.m. (<20 p.p.m.),⁹ and the observations of the NOE correlations between 7-H and 9-H and between 6-H and 20-H (Table 2). Seven of the eight chiral carbons of 1 were assigned according to the related NOE correlations, which were divided into three parts. The configurations of the first part C1~C5 on the first tetrahydropyran ring were shown in Figure 2b. The cross peaks between 5-H and 17-H₂ in NOE spectrum suggested a 1, 3-diaxial relationship. In the relative stereostructure of the atoms C11~C15 in the second tetrahydropyran ring (Figure 2b), the observation of signals between 15-H and one of the 13-H₂, and between 15-H and 11-H pointed out the 1, 3-diaxial orientation. The observation of strong NOE correlations between 14-NH and 21-H₃ suggested a 1, 3-diaxial interaction between 14-NH and 12-CH₃.

The cytotoxicity of 1 was evaluated in three human cancer cell lines, HCT-116, MDA-MB-235 and H232A, with the MTT method.¹⁰ 1 exhibited cytotoxic activity against these three cell lines with GI₅₀ (the half-maximum growth inhibitory concentration) values of 1,152 ± 0.16 nM, 916.6 ± 1.20 nM and 893.6 ± 1.64 nM, respectively. FR901464 was reported to have GI₅₀ values of 1.8 nM, 1.3 nM, 0.6 nM, 1.0 nM and 3.3 nM against MCF-7, A549, HCT-116, SW480 and P388 cell lines, respectively.¹ Apart from potentially differential cytotoxicities against different cell lines, 1 appeared to be a much weaker cytotoxic compound than FR901464.

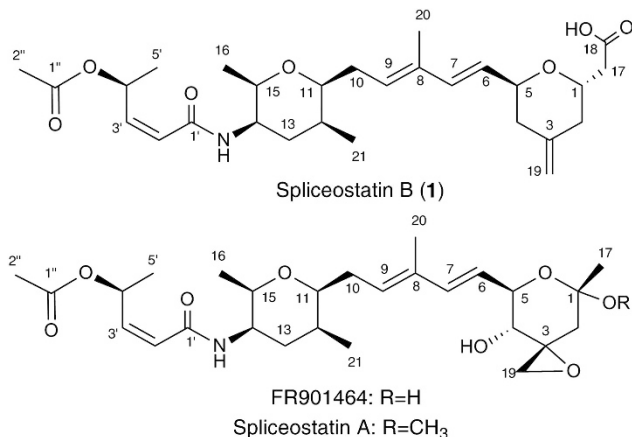


Figure 1 Structures of 1, FR901464 and spliceostatin A.

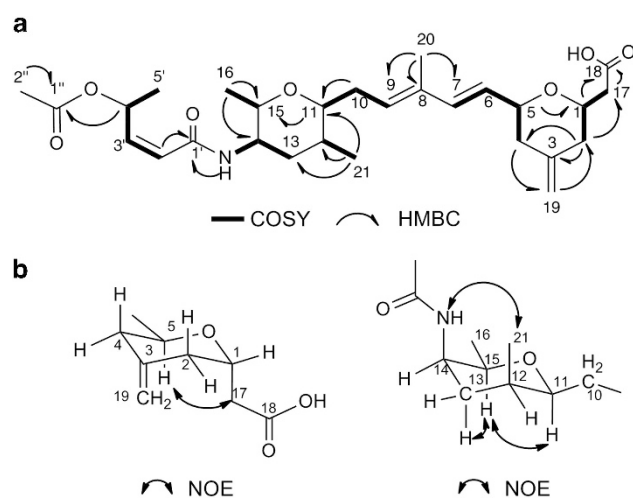
Table 1 Physicochemical properties of 1

Appearance	White amorphous powder
Elemental composition reported by HR ESI-MS [M + H] ⁺	C ₂₈ H ₄₂ NO ₇
HR ESI-MS found	504.2951 [M + H] ⁺ ; 526.2767 [M + Na] ⁺ ; 542.2556 [M + K] ⁺
Calculated	504.2955 [M + H] ⁺
m.p. (°C)	77–83
UV λ _{max} ^{Acetonitrile} (nm; log ε)	236 (4.0)
IR (KBr) ν _{max} (cm ⁻¹)	3355, 2978, 2938, 1735, 1667, 1635, 1526, 1371, 1317, 1244, 1166, 1121, 1051
[α] _D ²⁵	+4.0 (CH ₂ Cl ₂ , c 0.1)
Solubility	DMSO, acetonitrile, acetone, water, CHCl ₃ , CH ₂ Cl ₂

Abbreviation: DMSO, dimethyl sulfoxide.

Table 2 ^{13}C and ^1H NMR spectroscopic data for **1** (CDCl_3 , 298 K)

Position	$\delta_{\text{C}}^{\text{a}}$ (p.p.m.)	HSQC	$\delta_{\text{H}}^{\text{b}}$ (mult, J/Hz)	COSY	HMBC	NOE
1	69.0	CH	4.35 (qui, 5.7)	2, 17	18	2, 4, 6, 17
2	39.1	CH_2	2.11 (q, 6.6)		3, 4, 17, 19	1, 19
3	140.6	C				
4	39.4	CH_2	2.48 (overlap), 2.23 (overlap)		3, 5, 6, 19	5, 19
5	73.2	CH	4.42 (q, 5.2)	4, 6	1, 3, 6, 7	4, 6, 7, 17
6	125.8	CH	5.62 (dd, 15.8, 6.1)	5, 7	4, 5, 8	1, 4, 5, 7, 20
7	137.1	CH	6.26 (d, 15.84)	6	5, 8, 9, 20	5, 6, 9, 10, 20
8	134.8	C				
9	128.5	CH	5.44 (t, 6.9)	10	7, 10, 11, 21	7, 10, 11, 12, 20
10	31.8	CH_2	2.39 (overlap); 2.24 (overlap)		8, 9, 11, 12	9, 11
11	81.2	CH	3.57 (qt, 7.1, 5.0)	10	15, 21	9, 10, 12, 13
12	28.9	CH	1.79 (overlap)	21		10, 11, 13, 21
13	35.8	CH_2	1.99 (s); 1.94 (overlap)		11, 14, 15	11, 14, 15, 21
14	47.0	CH	3.95 (d, 7.1)	13, NH		NH, 13, 16
15	76.4	CH	3.72 (qd, 6.3, 2.0)	16	16	13, 16, 11, 14
16	17.7	CH_3	1.18 (d, 6.4)	15	14, 15	NH, 14, 15
17	38.7	CH_2	2.71 (dd, 15.1, 8.1); 2.50 (overlap)		1, 2, 18	1, 5
18	174.7	$\text{C}=\text{O}$				
19	111.4	CH_2	4.85 (d, 10.4)		2	2, 4
20	12.6	CH_3	1.74 (s)		7, 8, 9	6, 10
21	14.8	CH_3	1.02 (d, 7.3)	12	11, 12, 13	NH, 10, 12, 13
1'	165.1	$\text{C}=\text{O}$				
2'	122.5	CH	5.80 (d, 11.6)	3'	1', 3'	NH
3'	143.8	CH	5.91 (dd, 11.5, 5.8)	2', 4'	1', 2'	5'
4'	69.1	CH	6.31 (qui, 6.62)	3', 5'	1''	2', 3', 5', 21
5'	20.0	CH_3	1.40 (d, 6.5)	4'	3'	3', 4'
1''	170.4	$\text{C}=\text{O}$				
2''	21.3	CH_3	2.06 (s)	9, 11	1''	
		NH^{c}	6.46 (d, 8.80)	14	14, 15, 1'	2', 14, 16, 21
		OH^{d}				

^aThe ^{13}C NMR was measured at 125 MHz.^bThe ^1H NMR was measured at 500 MHz.^cThe 15N-HSQC was measured at 50 MHz.^dThe hydroxyl group was not observed in NMR spectra.**Figure 2** Key COSY and HMBC correlations of **1** (a), and key NOE correlations of **1** fragments (b).

1 differs structurally from FR901464 at four points (Figure 1): substitution of an epoxide group at C3 position with a terminal methylene moiety, presence of a carboxyl moiety at C17 position, and

absence of two hydroxyl groups at C1 and C4 positions, respectively. It was reported that a loss of C4 hydrogen bond donor decreases the cytotoxicity of meayamycin B (a synthetic FR901464 analog) about fourfold.¹¹ The importance of the C3 epoxide moiety for bioactivity has been documented in two independent studies. First, substitution of the C3 epoxide moiety in FR901464 by a terminal methylene group decreases the cytotoxicity about fivefold.⁵ Second, a non-epoxide analog of meayamycin that still contains the oxygen atom at the C3 position completely lost the activity, with IC_{50} values changed from 0.02 nM to >10 000 nM.¹¹ Although there is no concrete evidence about the influence of the carboxyl moiety at C17 position on the bioactivity of **1**, our studies on thailanstatins suggested that this moiety is critical to compound stability under weak alkaline conditions.⁷ Collectively, the absence of an epoxide moiety at C3 position and a hydroxyl group at C4 position of **1** most likely contributes to its weak cytotoxicity, which, in turn, supports the importance of those functionalities in FR901464 class of natural or synthetic compounds.

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Supplementary Information accompanies the paper on The Journal of Antibiotics website (<http://www.nature.com/ja>)