A pair of sulfur-containing geldanamycin analogs, 19-S-methylgeldanamycin and 4,5-dihydro-19-S-methylgeldanamycin, from Streptomyces hygroscopicus 17997

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Geldanamycin (GDM, Figure 1) is a benzoquinone ansamycin produced by *Streptomyces hygroscopicus*.^{1–3} It is a specific inhibitor of human heat shock protein 90 (Hsp90), which is a molecular chaperone assisting in protein folding, cell signaling and tumor repression, and a potential cellular target for anti-tumor agent. GDM is only used as a promising lead compound for anti-cancer drug development because of its poor water solubility and severe liver toxicity.^{4–6}

Although hundreds of semi-synthetic GDM analogs had been developed by chemical synthesis,^{7–10} a total of only tens of GDM analogs had been created by genetic manipulation of GDM biosynthetic gene cluster,^{11–15} and a few were discovered as natural GDM analogs.^{16–18} There is still an urgent need for novel GDM analog(s) with improved pharmacological profile and lower hepatotoxicity in developing new anti-cancer agent targeting Hsp90.

We are interested in discovering novel natural/biosynthetic GDM analog(s) with improved pharmacological profile(s). In our search for GDM analog(s), we found that *Streptomyces hygroscopicus* 17997, a GDM producer isolated by Institute of Medicinal Biotechnology, Chinese Academy of Medical Sciences,⁶ produced a red compound after the maximal GDM accumulation in fermentation broth. We detected also a similar red compound in the fermenation broth of the *gdmP* (encoding a cytochrome P450 monooxygenase for C4,5 oxidation of the post-polyketide synthase tailoring process in GDM biosynthesis)-disrupted mutant of *S. hygroscopicus* 17997, a 4,5-dihydrogeldanamycin producer.¹⁹ The two red compounds were proved by us to be a pair of sulfur-containing GDM analogs, with chemical structures of 19-*S*-methylgeldanamycin (1) and 4,5-dihydro-19-*S*-methylgeldanamycin (2), respectively, (Figure 1). In this note, we

report the discovery, isolation, structure elucidation and some physicochemical properties of 1 and 2.

Stock-frozen spores of *S. hygroscopicus* 17997 were thawed and spread onto ISPII (0.4% yeast extract, 1.0% malt extract, 0.4% glucose, 1.5% agar power) plates, incubated at 28 °C for 8–10 days for mycelium growth and sporulation, then a slice of the seed culture was picked up and inoculated into a fermentation medium (2% starch, 0.5% cotton seed power, 0.5% glucose, 1.0% cornsteep liquor, 0.5% yeast powder, 0.2% CaCO₃) for shaking (200 r.p.m.) at 28 °C for 120–144 h. The fermentation supernatant was used for subsequent detection and/or isolation of 1.

The fermentation supernatant of S. hygroscopicus 17997, with a culture time of 120-144 h, was extracted with an equal volume of EtOAc. The organic layer was concentrated, then subjected to silica gel TLC for a chromatograph (developed with a mobile phase of EtOAc/ CH₂Cl₂/hexane/MeOH, 9:6:6:1, ν/ν). A red band with a R_f =0.30 (for GDM, $R_f=0.54$) appeared. The red band changed to blue upon spraying 2.0 mol l⁻¹ NaOH (Figure 2), a color reaction for the preliminary discrimination of GDM and its analogs.²⁰ Compound(s) within the red band of silica TLC plate was eluted out by EtOAc, dried and then dissolved in MeOH for LC-ESI(+)-MS analysis. In the LC, a major elution peak appeared, with its UV absorption profile similar to that of GDM (Supplementary Figure S1). In MS¹, the major elution peak contained a principal molecular ion with m/z=629.4 Da ([M+Na]⁺, compound 1), whose fragment ions dispayed a MS² pattern typical of GDM analogs,²¹ (Supplementary Figure S2). On the basis of these data, we believed that 1 was an analog of GDM.

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Figure 1 The chemical structures of GDM, 1 and 2.



Figure 2 Silica gel TLC of EtOAc extracts of fermentation supernatant of Streptomyces hygroscopicus 17997 (A) and its gdmP-disrupted mutant (B).

1 was then purified for chemical structure elucidation. An equal volume of EtOAc was used to extract the fermentation supernatant of S. hygroscopicus 17997, then dried to a crude solid by rotary evaporation at room temperature. After fractionation by silica gel chromatography ($\varphi 2.5 \times 80$ cm), the pool containing 1 (eluted by CH₂Cl₂-MeOH, 9:1, v/v) was dried and then re-dissolved in MeOH for preparative Sephadex LH-20 fractionation (ϕ 1.5×100 cm), obtaining the refined preparation of 1. This refined preparation was used for preparative HPLC (Shimadzu LC-10ATvp, Shimadzu International Trading Co., Ltd, Beijing, China; Agilent ZorBax SB-C18, Agilent Technologies Co. Ltd, Beijing, China, 5 μ m, φ 9.4×250 mm, MeOH/ water, 45:55 (v/v), 1.5 ml min^{-1}), yielding a pure preparation of 1 (purity \geq 96%, calculated by area% of HPLC analysis at 254 nm). From 10.3-l fermentation supernatant, we obtained 10 mg pure preparation of compound 1 as amorphous red powder. It was used for HR-ESI(+)-MS, elemental (sulfur) analysis, ¹H- and ¹³C- NMR analyses and so on.

HR-ESI(+)-MS indicated that 1 had an accurate mass of 629.25688 ($[M+Na]^+$). The elemental analysis showed 1 containing ~ 5.16% sulfur, suggesting that 1 has a sulfur atom in its molecule (cald.: 5.28%, without Na⁺). Therefore, the molecular formula of 1 was deduced to be $C_{30}H_{42}O_9N_2S$ (exact mass, $[M+Na]^+$ 629.25087).

The ¹H- and ¹³C-NMR spectra of 1 showed strong similarities to those of GDM, but an additional methyl signal (H: δ 2.58, C: δ 16.9)

appeared. Considering its chemical shift ($\delta_{\rm C}$ 16.9) and a sulfur atom in 1, the additional methyl signal was deduced to be as a -SCH₃ group. The HMBC data suggested correlations between the H of -SCH₃ and aromatic quaternary C-19 (δ 130.7), indicating that the -SCH₃ was connected to the benzoquinone ring through C-19. Therefore, the chemical structure of 1 was determined to be 19-S-methylgeldanamycin. The NMR chemical shifts of 1 were assigned completely from HSQC, COSY and HMBC (Table 1, Figure 3).

The discovery, isolation and purification of **2** from the *gdmP*disrupted mutant of *S. hygroscopicus* 17997 was made or carried out in a very similar way as that of **1** from *S. hygroscopicus* 17997. From 43-l fermentation supernatant of the *gdmP*-disrupted mutant, we obtained 68 mg pure preparation of **2** (purity \ge 98%, calculated by area% of HPLC analysis at 254 nm).

2 was also an amorphous red powder, with an accurate mass of 631.26481 ($[M+Na]^+$) by HR-ESI(+)-MS. **2** contained 5.00% sulfur by elemental analysis, suggesting also one sulfur atom in the molecule (cald.: 5.26%, without Na⁺). Therefore, the molecular formula of **2** was deduced to be $C_{30}H_{44}N_2O_9S$ (exact mass, $[M+Na]^+$, 631.26652), which is two hydrogen atoms more than that of **1**. As **2** came from the *gdmP*-disrupted mutant of *S hygroscopicus* 17997, a 4,5-dihydrogelda-namycin producer, the chemical structure of **2** was regarded as, most probably, 4,5-dihydro-19-*S*-methylgeldanamycin.

The ¹H- and ¹³C-NMR spectra of 2 showed strong similarities to those of 1, except for C-4,5 and their hydrogens. In the ¹³C-NMR, two additional carbon signals of CH2- at $\delta_{\rm C}$ 24.3 and $\delta_{\rm C}$ 31.4 appeared in

2, whereas the carbon signals of CH-4 ($\delta_{\rm C}$ 130.0) and CH-5 ($\delta_{\rm C}$ 131.7)

Table 1	NMR	data of	1 and	2	in	CD ₃ OD
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		1		2	
Position	δ_{C}	δ_H	$\delta_{\mathcal{C}}$	δ_H	
1	175.9	_	176.3	_	
2	140.1	—	135.7	—	
3	125.4	6.39 (1H)	135.1	5.70 (1H)	
4	130.0	6.41 (1H)	24.3	1.99 (1H)	
				2.25 (1H)	
5	131.7	5.26 (1H)	31.4	1.17 (1H)	
				1.21 (1H)	
6	76.8	3.99 (1H)	81.0	3.06 (1H)	
7	82.8	4.89 (1H)	82.1	4.93 (1H)	
8	130.0	—	130.8	—	
9	136.0	5.20 (H)	134.1	5.12 (1H)	
10	36.5	2.24 (1H)	36.2	2.29 (1H)	
11	73.4	3.54 (1H)	73.9	3.59 (1H)	
12	81.2	2.87 (1H)	81.4	3.21 (1H)	
13	32.3	0.72 (1H)	33.2	0.80 (1H)	
		1.57 (1H)		1.66 (1H)	
14	30.4	2.08 (1H)	30.6	2.05 (1H)	
15	30.6	2.43 (1H)	30.4	2.38 (1H)	
		2.56 (1H)		2.64 (1H)	
16	130.4	—	130.4	—	
17	158.9	—	159.2	—	
18	180.7	—	180.7	—	
19	130.7	—	132.0	—	
20	136.0	—	130.8	—	
21	182.9	—	183.0	—	
22 (2-CH ₃)	14.3	1.89 (3H)	13.7	1.72 (3H)	
23 (6-0CH ₃)	56.7	3.09 (3H)	56.9	3.29 (3H)	
24 (7-0C0NH ₂)	159.1	—	159.0	—	
25 (8-CH ₃)	13.0	1.37 (3H)	12.7	1.40 (3H)	
26 (10-CH ₃)	18.8	0.94 (3H)	18.2	0.98 (3H)	
27 (12-0CH ₃)	56.6	3.25 (3H)	59.4	3.39 (3H)	
28 (14-CH ₃)	19.5	0.57 (3H)	18.5	0.53 (3H)	
29 (17-0CH ₃)	61.7	3.98 (3H)	61.6	3.97 (3H)	
30 (19-SCH ₃)	16.9	2.58 (3H)	16.9	2.51 (3H)	

¹H- and ¹³C-NMR spectra were obtained at 500 and 100 MHz, respectively, on INOVA-501 with TMS as internal standard, and measured in CD₃OD at room temperature.



Figure 3 Key long-range correlations of 1 and 2 in HMBC ($^{1}H \rightarrow ^{13}C$).

in 1 missed in 2. In the ¹H-NMR, two pairs of CH₂- hydrogen signals $(\delta_{\rm H} 1.99/2.25, \delta_{\rm H} 1.17/1.21)$ appeared in **2**, whereas the two hydrogen signals of CH-4 ($\delta_{\rm H}$ 6.41) and CH-5 ($\delta_{\rm H}$ 5.26) in 1 disappeared in 2. Therefore, the chemical structure of 2 was determined to be 4,5dihydro-19-S-methylgeldanamycin. The NMR chemical shifts of 2 were assigned completely from HSQC, COSY and HMBC (Table 1).

We are not clear how the -SCH₃ group is added onto GDM or 4,5dihydrogeldanamycin. Okabe T²² reported that naphthomycin A could react in vitro with a variety of -SH-containing compounds by replacing the -Cl with a -SR (including -SCH₃) at C-30. Some hindered amines could also undergo addition reaction in vitro at C19- of GDM nonenzymatically.^{7,8} Several sulfur-containing ansamycins, such as thiazinotrienomycin, trierixin, 3-methylthiorifamycin SV and thiazorifamycins, naphthomycins I and J, and awamycin²³⁻²⁸ had been isolated from Micromonospora sp. or Streptomyces sp. Of them, trierixin, 3-methylthiorifamycin SV, and awamycin had a -SCH₃ group in their molecule. Up to now, the production mechanisms of these sulfurcontaining groups of ansamycins remain unclear. Chemical and/or enzymatic reaction(s) may have taken place in their formation processes.

Cysyk RL²⁹ reported that GDM could react chemically (that is, nonenzymatically) with glutathione (GSH), forming a GSH adduct, in which the thiol group of GSH is substituted in the 19-position of the benzoquinone ring of GDM. It is well known that GSH had an important role in the detoxification of reactive drugs (such as alkylating agents) and reactive metabolites formed by hepatic drugmetabolizing enzymes. Thus, blocking of the C-19 of GDM by a -SCH₃ group may disrupt its reaction with GSH (and/or, possibly, the cysteines of some proteins within human cells), therefore reducing its toxicity to cells with lower GSH concentrations. Hence, 1 or 2 may show a different toxicological profile to GDM.

In a review about Hsp90 inhibitors, Janin YL³⁰ suggested that thioether at C19- of GDM was a possible choice of GDM derivatization. We performed a molecular docking of GDM, 1 or 2 with human Hsp90 by SYBYL (Tripos, a Certara Company, St Louis, MO, USA), which gave CScore of 8.22, 7.65 or 7.77 for GDM, 1 or 2, suggesting that both 1 and 2 may still keep fairly high affinities for Hsp90 (CScore is an algorithm to estimate the binding affinity of a given protein-

Table 2 Solubility and cytotoxicity of 1 and 2

Compound	Solubility ($\mu g m l^{-1}$)	IC ₅₀ (μ mol I $^{-1}$)	
GDM	2.0	0.062	
1	2019	19	
2	2132	23	



ligand complex, with a known three-dimensional structure). Therefore, we performed a preliminary test for the cytotoxicity of 1 or 2 against HepG2 cancer cell line, assayed by sulforhodamine B assay.³¹ Both 1 and 2 displayed decreased, but still very strong, cytotoxicities against HepG2 cancer cells (Table 2), but a comprehensive evaluation against different cancer cell lines is needed to appraise the cytotoxicity of 1 or 2 in the future.

We assayed the aqueous solubility of 1 or 2 in sodium chloride injection solution, following the method of Li *et al.*³¹ Compared with GDM, both 1 and 2 showed approximately 1000-fold increases (Table 2), suggesting that 1 and 2 should have significant improvements in bioavailability and pharmacokinetic properties. Besides, both 1 and 2 showed better light stability than that of GDM (Supplementary Figure S3), hence, they may have the potentials as anti-cancer drug candidates for further studies.

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