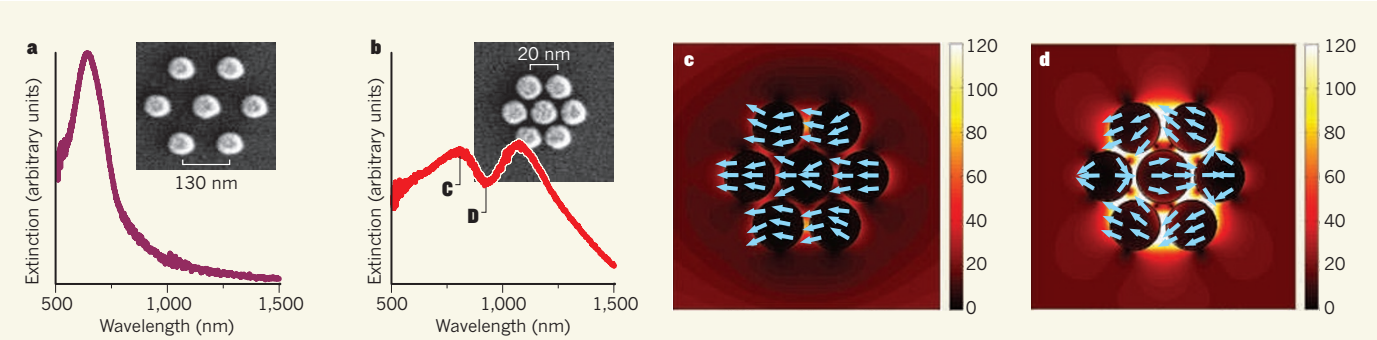

Figure 2 | The nanoclusters studied by Hentschel and colleagues[1]. The authors studied the optical response of clusters comprising seven gold nanoparticles and found that, for interparticle separations of less than 60 nanometres, the pronounced spectral dip of a plasmonic Fano resonance emerges. **a**, **b**, Absorption spectra for interparticle separation of 130 nm (**a**) and 20 nm (**b**); insets display images of the clusters taken with a scanning electron microscope. The Fano dip is marked in **b** with a D. **c**, **d**, Simulated local optical fields and corresponding electrical currents (blue arrows) at the spectral wavelengths marked C and D in **b**. The vertical bars indicate the colour scale for the magnitude of the optical field (relative to the field to which the clusters are subjected). The optical field at the Fano dip's wavelength is characterized by bright 'hot spots' — compare **d** with **c**, in which such an effect is absent. (Figure adapted from ref. 1.)

Fortunately, in addition to these 'bright' nanoplasmonic resonances, there are also 'dark' ones[5], which by themselves are not very prominent in optical spectra. However, if a bright and a dark resonance coexist in a certain spectral range — which is not unlikely, because the bright resonances span relatively wide wavelength ranges — then their optical fields interfere. This interference significantly enhances the manifestation of the dark resonance: it acquires strength from the bright resonance and shows up as an asymmetrical peak-and-dip profile characteristic of a Fano resonance[6]. An important, albeit counter-intuitive, property of dark resonances is that, at exactly the frequency of the Fano dip, the hot spots of the nanolocalized optical fields in the nanosystem are strongest. This is because, at this frequency, the nanosystem emits minimal light and, consequently, does not wastefully deplete the energy of the plasmon oscillations.

In their study, Hentschel *et al.*[1] use state-of-the-art nanofabrication techniques to demonstrate the possibility of switching such a plasmonic Fano resonance on and off. They manufactured symmetrical clusters of gold nanoparticles and investigated the clusters' optical response as a function of the interparticle separation (Fig. 2). They found that, for interparticle separations smaller than 60 nanometres, the distinct spectral profile of a Fano resonance — two uneven peaks separated by a pronounced dip — emerges. What's more, on removal of the clusters' central nanoparticle, the Fano profile disappears.

Thus, the authors' study offers a means to circumvent the light-scattering problem that plagues the use of the bright nanoplasmonic resonances of relatively large nanosystems: by switching on the plasmonic Fano resonance in the dense symmetrical assemblies of nanoparticles, the internal electrical current driving the light scattering and absorption is drastically suppressed. Earlier studies[7,8] demonstrated other ways to control plasmonic Fano resonances, and to get around the light-scattering (radiative loss) problem: in the optical spectral region, by breaking the symmetry of the nanostructure[7], and in the gigahertz regime, through the use of special, artificially engineered structures known as metamaterials[8]. An alternative to these and Hentschel and colleagues' approach may be not to fight the dragon of the radiative loss with the plasmonic Fano resonances but to avoid it altogether — by using smaller nanoparticles (of less than 20 nm) that scatter little light and so do not suffer from radiative loss and the subsequent broadening of the resonances.

What applications are there, then, for nanoplasmonic Fano resonances? One straightforward application is the sensing of ultrasmall amounts of chemical or biological substances. One approach to this is based on the fact that, when chemical or biological molecules bind to a plasmonic nanosystem, the enhanced optical fields of the nanosystem's hot spots polarize them. This causes a shift in the frequency of the nanosystem's nanoplasmonic resonance. Because nanoplasmonic Fano resonances are sharp, and the corresponding hot spots strong, this shift is relatively large and can be easily detected, thus revealing the presence of such molecules.

Another possible application is the widely used spectroscopic technique of surface-enhanced Raman scattering (SERS)[9,10], which relies on plasmonic hot spots to produce spectroscopic fingerprints that identify chemical structure, and changes in it, in minute amounts of molecules. Nanoplasmonic Fano resonances cause very bright hot spots (Fig. 2d) and, as a result, could be used to improve the sensitivity of SERS-based methods and devices. They could also be exploited in techniques such as surface-enhanced spectroscopy of fluorescence[11,12] and absorption[13].

Last, but not least, is the prospect of using the resonances to build a spaser (surface plasmon amplification by stimulated emission of radiation)[14–18], a nanometre-sized nanoplasmonic source of coherent, localized optical fields in which surface plasmons play the part that photons take in conventional lasers. ■

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CORRECTION

In the News & Views article "Nanotechnology: Holes with an edge" by Hagan Bayley (*Nature* **467,** 164–165; 2010), "6-mercaptohexanoic acid", mentioned in the fifth paragraph, should have read "16-mercaptohexadecanoic acid".