

Radionuclide contamination

Nanometre-size products of uranium bioreduction

One strategy that is being pursued to tackle the international problem of actinide contamination of soils, sediments and water is to use microbial activity to 'fix' these radionuclides into an insoluble form that cannot be readily dispersed. Here we show that uraninite (UO_2) particles formed from uranium in sediments by bacterial reduction are typically less than 2 nanometres across and that the small size has important implications for uraninite reactivity and fate. Because these tiny particles may still be transported in an aqueous environment, precipitation of uranium as insoluble uraninite cannot be presumed to immobilize it.

The discovery that bacteria can convert mobile hexavalent uranium, U(VI) , to tetravalent uranium, U(IV) , which precipitates as uraninite¹, has spurred investigations into the *in situ* bioremediation of uranium-contaminated groundwater. These strategies assume that the formation of highly insoluble uraninite² will inhibit the mobilization of uranium.

We used organic substrates to create anaerobic conditions to stimulate growth of native bacteria in complex natural uranium-contaminated sediments and water, taken from the Midnite Mine, Washington, USA. After 1 month, the aqueous uranium concentration decreased from 20 to 0.3 parts per million. Synchrotron-based uranium L-III edge X-ray absorption near-edge spectroscopy (XANES) analysis of the incubated sediment indicated that most U(VI) had been reduced to U(IV) . Selected-area electron-diffraction patterns and crystallographic analysis of two-dimensional transmission electron microscope (TEM) lattice-fringe images confirmed the structure as being that of uraninite (which is similar to that of calcium fluoride).

We also constructed a gene library encoding small-subunit ribosomal RNAs (16S rRNAs) from the incubated sediment³ to identify the bacterium that is likely to be responsible for the conversion of uranium to uraninite. Out of 94 clones, 42 clustered strongly with *Desulfosporosinus* spp. (results not shown), a Gram-positive sulphate-reducing microbe. We confirmed that *Desulfosporosinus* spp. isolated from the sediment was able to reduce U(VI) to U(IV) and to precipitate U(IV) as uranium oxide (manuscript submitted).

Uraninite formed by mixed cultures in the incubated natural sediment, as well as by pure cultures of *Desulfosporosinus* spp., was characterized by high-resolution TEM (Fig. 1a). The uraninite crystals are less than 3 nm in diameter and occur as discrete

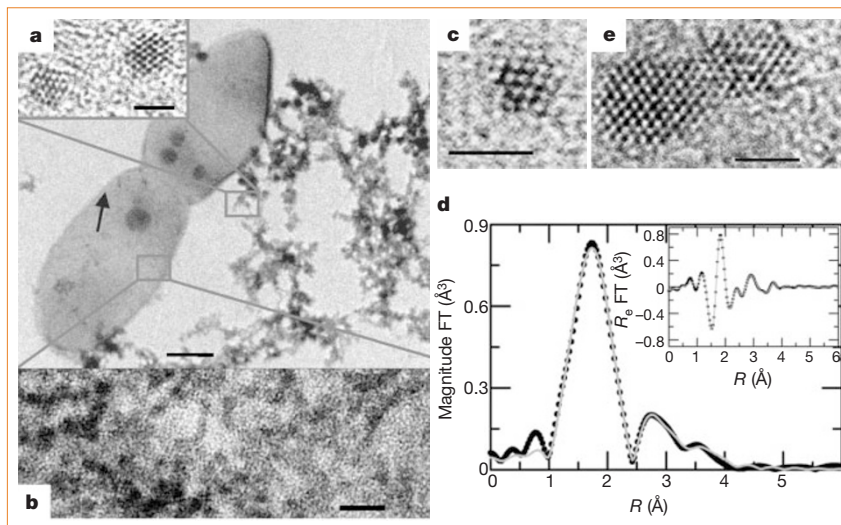


Figure 1 Characterization of bioreduced uraninite (UO_2) nanoparticles. **a**, Transmission electron microscopy (TEM) image of flocculated UO_2 nanoparticles associated with *Desulfosporosinus* spp. bacteria (arrow). Inset, high-resolution TEM image of isolated particles. **b**, TEM image of the *Desulfosporosinus* cell surface coated with UO_2 nanoparticles (size, 1.5–2.5 nm). **c**, A single uraninite particle (about 1.3 nm across). **d**, Magnitude of the Fourier transform (FT) of X-ray-absorption fine-structure (XAFS) data obtained from the sediment, and the best-fit model. R , radial distance (uncorrected for electron phase shift). Inset, real part of the Fourier transform. R_r , real part of Fourier-transformed data. **e**, Growth of uraninite crystals formed by orientated attachment of individual nanoparticles. Scale bars: 0.6 μm (**a**, inset, 2 nm), 6 nm (**b**) and 2 nm (**c**, **e**).

(Fig. 1a, inset) and aggregated particles, and as particle coatings on bacterial cell surfaces (Fig. 1b). Many crystalline particles were less than 1.5 nm in diameter (Fig. 1c), and our images indicated that even smaller particles were present (results not shown).

Extended X-ray-absorption fine-structure (EXAFS) analysis (Fig. 1d) of the incubated sediment revealed that the average uraninite U–U coordination number was 5.6 ± 4 (compared with 12 in bulk uraninite). This confirms that a large proportion of uranium ions in nanoparticles are at the surfaces of the uraninite nanoparticles, as expected for tiny particles. Structure models consistent with the EXAFS-determined coordination number imply an average particle size of about 1.5 nm, which is close to the size of the smallest crystals detected in our high-resolution TEM images. We conclude that bacterial U(VI) reduction produces uraninite particles in a range of sizes that extends down to molecular-scale clusters.

EXAFS data indicate that all U(IV) in the sediment was 8-coordinated by oxygen, as in the bulk uraninite structure. U–U distances (0.380 nm) are 0.007 nm shorter in biogenic uraninite than in bulk uraninite, implying that there may be significant displacement of surface U(IV) ions. The 0.007-nm contraction in U–U spacing corresponds to a surface stress of about 4.82 N m^{-1} , which represents more than a billion-fold increase in the solubility product compared with well-crystallized uraninite⁴ (based on the reaction $\text{UO}_2 + 2\text{H}_2\text{O} = \text{U}^{4+} + 4\text{OH}^-$; the actual value depends on the quantitative relationship between surface stress and surface energy).

The nanometre-scale size of biogenic

uraninite particles is likely to influence both its reactivity and transport. We would expect these UO_2 crystals to be mobile in porous sediments and to re-oxidize rapidly. Flocculation, on the other hand, should reduce the transport potential. TEM images indicate that crystals grow from orientated aggregates of uraninite (Fig. 1e), which may be mediated in some cases by extracellular organic polymers (our unpublished results). To predict the likely environmental fate of UO_2 nanoparticles, it will first be necessary to determine the transport physics of isolated nanoparticles and of nanoparticles adsorbed onto colloidal particles and organics, and to define models that link crystallization with crystal-growth kinetics, flocculation kinetics and fluid flow.

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Competing financial interests: declared none.