SCIENTIFIC CORRESPONDENCE

Chernobyl fallout on Ioannina, Greece

SIR—We have measured the γ -ray spectrum of the fallout from the Chernobyl nuclear reactor accident at Ioannina, located in northwestern Greece, nearly 1,500 km south-west of Chernobyl. We have examined air filters collected every 24 h since 29 April to monitor the γ -radiation activity, using an intrinsic germanium high-resolution detector, shielded to reduce background. The fallout probably reached Ioannina between 1 and 2 May, and the activity peaked on 5 May. After that we noticed a reduction of the air contamination, mainly due to local rains.

Figure 1 shows a typical γ -ray spectrum from one of our filters, the activity of which was measured three times at monthly intervals to estimate the half-lives of the radionuclides present. From a number of spectra, 14 different nuclides were identified by their energy spectra and half-lives and 9 additional peaks in the spectral region between 50 and 800 KeV remain unidentified.

In Table 1 the activity of each of the most prominent radioisotopes is characterized as very strong (VS), strong (S), weak (W) or very weak (VW), and the respective half-life as very long (VL), long (L) or short (S). The ¹³¹I activity is defined 'strong' and its half-life as 'long'. Table 1 also lists the masses of fission decay products generated in the reactor'. Table 2 lists the energies of the unidentified lines.

From our air-filter measurements we conclude that: (1) the relative activities for the most prominent isotopes remained unchanged during most of the period Table 1 Radioisotopes identified from air filters

Isotope	Relative y-activity	Half-life	Mass of fission product (mg MW ⁻¹ day ⁻¹)
95Nb	VW	VL	0.33
⁹⁹ Mo	W	S	107.0
¹⁰³ Ru	W VS	VL	
			65.4
¹⁰⁶ Ru	W	VL	
^{129m} Te	VW	VL	
			15.7
¹³² Te	S	S	
¹³¹ I	S S S S	L	5.86
132I	S	S	
134Cs	S	VL	
136Cs	VW	L	90.4
¹³⁷ Cs	S	VL	
¹⁴⁰ Ba	W	L	38.6
¹⁴⁰ La	W	S	39.8
¹⁴¹ Ce	VW	VL	86.0

between 1 and 10 May; (2) these activities are not what would be expected from the mass of fission products and their volatility. (3) We agree with Pringle *et al.*² that the identified radioisotopes are ²⁰⁵U fission decay products; however, we did not locate the isotopes ⁹⁵Zr, ¹⁴⁴Ce, ¹²⁷Sb, ¹⁰⁵Rh and ¹⁴³Ce. (4) The reactor operation time based on the ratio of ¹³⁷Cs to ¹³⁴Cs activity is 15% higher than that reported by Devell *et al.*³ using the same method. (5) Finally, a trace analysis using X-ray energy spectrometry was inconclusive in locating Sr in our air filters.

The extreme conditions in the reactor core after the accident make it hard to know in what chemical compounds the radioisotopes appear in the radioactive cloud. Unfortunatly, information about core conditions before and during the accident is not yet available to us, and therefore we cannot verify our assumption

Table 2 Unidentified spectral lines			
Energy (keV)	Relative γ -activity	Half-life	
223.4	VW	S	
239.2	VW	S	
257.4	VW	S	
358.4	W	L	
489.7	W	L	
511.6	W	VL	
597.9	W	L	
653.9	W	S	
787.6	VW	S	

that the surprisingly high proportion of non-volatile nuclides (such as ¹⁰⁵Ru) and the absence of volatile nuclides reflect the chemical compound which hosted the nuclides at the time of release to the atmosphere.

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2. Pringle, D.M., Vermeer, W.J. & Alten, K.W. *Nature* 32 569 (1986).

3. Devell, L. et al. Nature 321, 192-193 (1986).

The origins of the Earth's oceans

SIR—Several mechanisms have been proposed for the origin of the Earth's oceans. such as the degassing of crustal rocks over geological time¹ and the formation of impact produced atmosphere, including water, by accretion of planetesimals²³ early during the Earth's formation. The available evidence suggests that sizeable oceans were present early in the Earth's history³ and that the early Solar System

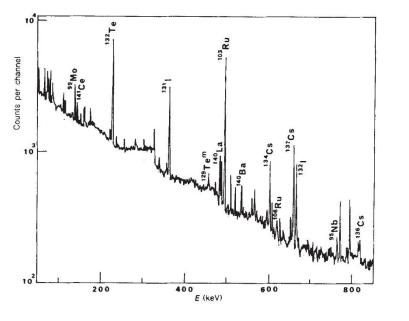


Fig. 1. A typical complex γ -ray spectrum obtained with an intrinsic Ge detector from an air filter.