November 2, 1963 No. 4905

phyllized leaf material was wetted with othanol and hydrolysed with 0.33 N potassium hydroxide for periods of 1, 2, 3, 4, 5, 6 and 18 h. Samples in duplicate were passed over columns of 'AG 1-8X' 100-200 mesh chloride form of 'Dowex'; the RNA was determined by absorption at 260 mµ. Yeast RNA purified by similar treatment was used as the standard. Results are given in Table 1.

Table. 1.	. EFFECT	OF EXPOSURE TIME ON HYDROLYSIS OF LEAF TISSUE			
RNA (0.33 N POTASSIUM HYDROXIDE AT 37° C)					
		•			

Sample (A and B)	time (h)	RNA $\mu g/g$ fresh weight	Percentage of recovery
1	1	2,140	62
2	2	2,520	73
3	3	2,550	74
4	4	2,720	79
5	5	2,900	84
ě	ē	3,330	96
7	18	3,450	100

These results indicated that more than 60 per cent of the RNA became acid-soluble nucleotides after 1 h exposure but this material required at least 6 h exposure for complete hydrolysis, based on 18 h as 100 per cent. Since 6 h exposure failed to completely solubilize the RNA, hydrolysis time was lengthened to 8 h as shown in Table 2.

Table 2. EFFECT OF EXPOSURE TIME ON HYDROLYSIS OF LEAF TISSUE RNA (0.33 N POTASSIUM HYDROXIDE AT 37° C). Exposure

Sample	time	RNA $\mu g/g$ fresh weight	Percentage
(A and B)	(h)		of recovery
8	6	3,350	97
9	7	3,600	104·4
10	8	3,500	101·6
11	18	3,440	100

Extension of hydrolysis beyond 7 h appears to be unnecessary. Seven-hour hydrolysis actually gave a somewhat higher reading than the standard 18 h. When repeated in triplicate at 7 and 18 h, the former time gave 101.4 per cent RNA content, relative to the latter.

These investigations indicate that hydrolysis of treated leaf tissue with 0.33 N potassium hydroxide for periods beyond 7 h will not yield additional RNA in the form of acid-soluble nucleotides. However, unlike animal tissues, lesser periods of hydrolysis are insufficient for complete hydrolysis.

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¹ Schmidt, G., and Thannhauser, S. J., J. Biol Chem., **161**, 83 (1945).
 ² Diener, T., and Lashcen, A. M., Proc. Amer. Soc. Hort. Sci., **75**, 195 (1960).
 ³ Kessler, B., and Engelberg, N., Biochim. Biophys. Acta., **55**, 70 (1962).
 ⁴ Fleck, A., and Munro, H. N., Biochim. Biophys. Acta, **55**, 571 (1962).

Amino-acid Configuration of Mycobacillin

MYCOBACILLIN, a new antifungal antibiotic isolated from the culture filtrate of a strain of Bacillus subtilis, is a cyclic peptide containing 13 residues of 7 different amino-acids, the sequence of which has also been worked out^{1-2} . Now the stereotypic nature of the constituent amino-acids is an interesting query. Antibiotics are often characterized by the presence of **D**-amino-acids. Penicillin and antibiotics of actinomycin group contain D-valine3. Actinomycin also contains **D**-alloisoleucine. Likewise, **D**-phenylalanine is the constituent of gramicidine S⁴ and tyrocidines⁵. Extreme examples are, however, polymyxin-B, polymyxin-D and bacitracin A. Polymyxins contain D-phenylalanine, D-leucine, D-serine and D-a.y-diaminobutyric acid, and bacitracin contains p-ornithine, p-glutamic acid, pl-aspartic acid and p-phenylalanine⁶. This communication records the stereotypic nature of each of the constituent

amino-acids of mycobacillin. In essence the method is based on the substrate specificity of D- and L-amino-acid oxidases and L-glutamic acid dehydrogenase.

Mycobacillin isolated in state of chemical purity was The acid was subjected to complete acid hydrolysis. removed and acid-free hydrolysate was dissolved in water. The amino acid mixture so obtained was incubated with hog kidney p-amino-acid oxidase⁷ and snake venom L-amino-acid oxidase⁸ separately. Another aliquot of the mixture was treated with ox-liver L-glutamic acid dehydrogenase⁹. The amino-acid mixture after incubation was chromatographed to determine if any disappearance of amino-acids has occurred. The spots when obtained were quantitatively assaved.

It appears from Table 1 that leucine is oxidized completely and tyrosine almost completely by L-amino-acid oxidase while they are left practically untouched by D-amino-acid oxidase. L-Configuration of these two amino-acids are thus established. Now L-amino-acid oxidase does not act on L-alanine, L-serine, L-proline and L-aspartic acid¹², which were found to be equally inactive towards D-amino-acid oxidase tested; hence they are of L- variety. Regarding aspartic acid. it may be pointed out that this amino-acid has been repeated as many as five times in the molecule² and that about four-fifths of it was oxidizable by p-amino-acid oxidase. This may be taken to mean that any four out of five aspartic acid residues are of D-isomer. Now for glutamic acid, neither D- nor L- is found to be acted on by either of the oxidases. It is similarly recovered when treated with L- specific crystalline ox-liver glutamic acid dehydrogenase. So glutamic acids present in the molecule are of D-configuration.

Table 1. OXIDATION OF CONSTITUENT AMINO-ACIDS BY AMINO-ACID OXIDASES

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Per cent disappeared after D-amino-acid oxidase treatment	Per cent disappeared after L-amino-acid oxidase treatment	
nil	nil	
nil	96.0	
18.0	89.0	
6-6	nil	
4.0	nil	
76.0	nil	
nil	nil	
	after D-aminò-acid oxidase treatment nil 18:0 6:6 4:0 76:0	

Incubation mixture consists of pyrophosphate buffer pH 8.3 (0.1 M), 1.00 ml., FAD, 0.01 μ M, catalase 100 units, neutralized mycobacillin hydro-lysate and either D- (5 mg) or L- (1 mg) amino-acid oxidase. The reaction was carried out at 37° and after 180 min was terminated by addition of double the volume of ethanol. Amino-acids were then purified on 'Dowex-50 column'' and fractionated by two-dimensional paper chromatography using *n*-butanol: acetic acid : water (4 : 1 : 1) and water-saturated phenol as developing solvents. The amino-acids were estimated by the method of Giri *et al.*¹¹. The estimation of amino-acids in mycobacillin hydrolysate was done similarly after fractionation and it was taken as control.

The stereotypic configuration of the constituent aminoacids of mycobacillin are thus indicated.

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¹ Majumdar, S. K., and Bose, S. K., Nature, 181, 134 (1958).

- ² Majumdar, S. K., and Bose, S. K., Biochem. J., 74, 596 (1960).
- ⁸ Brockmann, H., Angew. Chem., 66, 1 (1954).

- ⁶ Brockmann, H., Angew. Chem., 66, 1 (1954).
 ⁸ Synge, R. L. M., Biochem. J., 39, 363 (1945).
 ⁶ Paladini, A., and Craig, L. C., J. Amer. Chem. Soc., 76, 688 (1954).
 ⁶ Lockhart, I. M., and Abraham, E. P., Biochem. J., 58, 633 (1954).
 ⁷ Burton, K., in Methods in Enzymology, edit. by Colowick, S. P., and Kaplan, N. O., 2, 199 (Academic Press, New York, 1955).
 ⁸ Ratner, S., in Methods in Enzymology, edit. by Colowick, S. P., and Kaplan, N. O., 2, 204 (Academic Press, New York, 1955).
 ⁸ Strecker, H. J., in Methods in Enzymology, edit. by Colowick, S. P., and Kaplan, N. O., 2, 204 (Academic Press, New York, 1955).
 ⁹ Plaisted, P. H., Cont. Rouce Thomson Inst. 19, 221 (1958)

- Plaisted, P. H., Cont. Royce Thompson Inst., 19, 231 (1958).
 Ti Giri, K. V., Radhakrishnan, A. N., and Baidyanathan, C. S., Anal. Chem., 24, 1677 (1942).
- ¹² Bender, A. E., and Krebs, H. A., Biochem. J., 46, 210 (1950).