macroneurones. Hassler⁴ judged that his "thalamic internuncial cells" in the human dorsomedial nucleus, indistinguishable in his Fig. 14 from my "microneurones", suffer equal retrograde degeneration with the "specific" neurones after substantial cortical removals, but relatively less after very restricted lesions; and surmised that they project 'unspecifically' to the cortex. It may be that the tempo of degeneration is higher after wide cortical damage than after standard leucotomy; but in my judgment at least the majority of microneurones are wholly intrathalamic 'interneurones', susceptible to disuse atrophy rather than to retrograde degeneration, and I would still predict that they will be found to constitute structural essentials to the brain's attention-focusing mechanism, as well as to its electroencephalographic alpha-rhythm¹.

TURNER MCLARDY

The Laboratory,

St. Andrew's Hospital, Northampton.

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Reversal of Solubility Characteristics of 'Luxol' Dye-phospholipid Complexes

'LUXOL' fast blue G (Du Pont), the diarylguanidine salt of a sulphonated azo dye, forms complexes stoichiometrically with phospholipids¹. These complexes have been found to be insoluble in a varying pattern in the lower alcohols. For this study phospholipid-dye complexes were prepared with the same dye and phosphatidyl choline, phosphatidyl serine, phosphatidyl ethanolamine, phosphatidyl inositol, or sphingomyelin. The complexes were produced by bringing together the dye and phospholipid in pyridine solution. After evaporation of the solvent, excess uncomplexed dye or phospholipid could be removed by the alcohol in which the complex was insoluble. The solubility of the complexes in methanol, ethanol, isopropanol and isobutanol was determined. The results are summarized in Table 1. It was found that the complexes were soluble to the extent of at least 0.1 per cent or else were completely insoluble in the alcohols When alcoholic solutions of dye and phosphoused. lipid were mixed, precipitates were formed in those alcohols in which the particular complex had been found to be insoluble. The absorption maximum for the dye alone in the alcohols used is $597 \text{ m}\mu$. When alcoholic solutions of the dye and phospholipid were mixed, the absorption maximum dropped to 570 mµ before precipitation took place. If the precipitate was then centrifuged and the deposit redissolved in methanol or an alcohol in which it is soluble the absorption maximum was found to return to 597 m μ . This can be taken as evidence of dissociation where complexes are marked soluble in Table 1.

With the foregoing in vitro data as a guide, we tried to devise a method for the differentiation of phospholipids in tissue sections. Only partial success was achieved. In sections of brain tissue some differential staining of the phospholipids present was accomplished when solu-

tions of the dye in the various alcohols were used. However, in other tissues some non-phospholipid material was stained. Tryptophan, present in the necrotic areas in livers of experimental animals, stained intensely with the dye in isopropanol solution. In methanol the dye stained neither phospholipid nor tryptophan but collagen and elastin were stained selectively. Frozen unfixed tissues were employed with similar results.

The dye 'Luxol' fast blue G is not a specific histochemical reagent for phospholipids, but its staining affinities for certain tissue components vary with the solvent used. This may explain some of the anomalous staining of phospholipids found in the literature. Recently, Wolman² found that the affected neurones in three cases of Neimann-Pick's disease were not stained by 'Luxol' blue; whereas Baker's procedure for phospho-Current 'Luxol' blue techniques lipids was positive. utilize ethanol as the dye solvent, and from reference to Table 1, it will be seen that sphingomyelin does not form insoluble complexes with 'Luxol' fast blue G in this alcohol. As an increase in sphingomyelin storage can be expected in Neimann-Pick's disease, the in vitro data furnish a possible explanation for the non-staining of the affected neurones by 'Luxol' blue dissolved in ethanol.

T. N. SALTHOUSE

Pathology Department,

The Squibb Institute for Medical Research, New Brunswick, New Jersey.

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RADIOBIOLOGY

Effect of Multiple Injections of Calcium Compounds on the Survival of X-irradiated Rats

A MODERATE increase in the survival of X-irradiated rats has been observed when either parathyroid extract or ethylenediamine tetraacetate (EDTA) was injected before or after mid-lethal doses of X-rays¹⁻³. The action of these substances has been attributed to bone resorption, which was thought to protect the adjacent radiosensitive bone marrow by exposing it to an increased concentration of newly-released calcium. In the work recorded here, we have tested this view directly by the injection of calcium compounds.

Adult males of a hooded strain of laboratory rat, weighing between 270-340 g, were used. Calcium was administered intraperitoneally as the acetate (0.12 M), lactate (0.12 M), or gluconate (0.067 M). The sodium salt of each of these compounds was also tested and administered in twice the molar concentration so that an equivalent amount of anion could be given in the same volume of fluid. All compounds were dissolved in a Krebs-Ringer salt solution⁴ without phosphate buffer. The radiation source was a Philips constant-potential X-ray machine operated at 300 kV and 10 m.amp (half-value layer, 1.14 mm copper), with a target-object distance of 80 cm and a dose-rate in air of approximately 34 r./min. All rats received a whole body dose of 740 r. The animals were exposed 12 at a time in a revolving, partitioned, 'Lucite' cage. Each group of 12 consisted of 4 control animals (injected with Krebs-Ringer solution only), 4 animals injected with the sodium salt, and 4 animals with the calcium salt of the compound under investigation.

Table 1. SOLUBILITY OF 'LUXOL' DYE-PHOSPHOLIPID COMPLEXES

Dye or dye complex 'Luxol' fast blue G	Water*	Methanol Soluble	Ethanol Soluble	Isopropanol Soluble	Isobutanol Soluble	Pyridine Soluble
Phosphatidyl choline—dye complex Phosphatidyl serine—dye complex Phosphatidyl inositol—dye complex Phosphatidyl ethanolamine—dye complex Sphingomyelin—dye complex	Soluble	Soluble	Insoluble	Insoluble	Insoluble	Soluble
	Soluble Soluble	Soluble Soluble	Soluble Soluble	Insoluble Insoluble	Insoluble Soluble	Soluble Soluble

* pH 7.0