

Table 1. EFFECT OF PHENYLALANINE ON TRANSPORT OF AMINO-ACIDS

Amino-acid	Concentration difference developed between slices and suspending medium (mM)		Inhibition by L-phenylalanine (per cent)
	When present alone	In the presence of an equimolar amount of L-phenylalanine	
L-Proline	8.48 ± 0.73 (16)	7.71 ± 0.67 (16)	9
L-Histidine	15.6 ± 1.1 (12)	9.0 ± 1.6 (20)	42
L-Arginine	2.70 ± 0.56 (16)	1.47 ± 0.32 (15)	46
L-Ornithine	3.44 ± 0.31 (12)	1.82 ± 0.14 (12)	47
L-Tyrosine	3.88 ± 0.32 (13)	1.15 ± 0.10 (12)	70

Initial concentration of each amino-acid in suspending medium 2 mM. Figures shown for concentration difference are mean and standard deviation with the number of samples in parentheses. Experimental period, 1 hr. 37° C.

acids in brain and other tissues of the body. The considerable interference with tyrosine transport would be of greater significance in those tissues such as brain and intestinal mucosa which transport tyrosine the most actively⁹. With tissues in which there does not appear to be active transport of tyrosine⁹, movement of tyrosine would probably not be affected, as has already been shown with liver¹⁰.

I wish to thank Prof. N. L. Edson for giving me facilities in the Department of Biochemistry, and the Medical Research Council of New Zealand for financial support.

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Stability of Dextran during Prolonged Storage

So far as we are aware, no measurements relating to the stability of clinical dextran solutions kept for several years have been published. Two dextran solutions made in the United States, *A* and *B*, and two made in Great Britain, *C* and *D*, were examined in 1954, and again after storage for 5 yr. at 4° C. Where applicable the methods of testing described in the British Pharmacopoeia¹ were used. The optical rotation of each dextran was determined and used in calculating the concentrations of solutions for the viscosity measurements (Table 1). Dextran *D* had undergone the most change; the changes in the others were relatively small.

The renal excretion of these dextrans was measured in rabbits (Table 2). The greater part of the dextran

Dextran	Optical rotation		Intrinsic viscosity at 25° C.	
	1954	1959	1954	1959
<i>A</i>	195.1	0.249	0.248	
<i>B</i>	197.7	0.254	0.251	
<i>C</i>	201.2	0.394	0.388	
<i>D</i>	198.1	0.362	0.326	

was excreted on the first day, less than 3 per cent on the second and less than 1 per cent on the third day. Two, three or four rabbits were used for each determination and the daily results averaged; Table 2 records the totals for 3 days after injection. The results for dextran *A* suggest a change in molecular composition, but no corroboration of this was found in the other measurements.

Dextran	Percentages of injected dose excreted	
	1954	1959
<i>A</i>	37.2	47.6
<i>B</i>	43.1	39.1
<i>C</i>	13.5	14.6
<i>D</i>	15.9	15.4

The retention of each dextran in the plasma was determined by taking the average of daily estimations in the same groups of rabbits (Table 3). During five years of storage little if any change had occurred in the dextrans affecting their retention in the circulation.

Table 3. PLASMA CONCENTRATION OF DEXTRAN AS PERCENTAGE OF CONCENTRATION 10 MIN. AFTER INJECTION

Dextran	10 min.	Day							
		1	2	3	4	5	6	7	
<i>A</i>	1954	100	28.8	11.7	0.4	0.0	0.0	0.0	0.0
	1959	100	29.6	15.2	2.25	0.8	0.0	0.0	0.0
<i>B</i>	1954	100	29.2	13.8	3.6	0.0	0.0	0.0	0.0
	1959	100	30.0	14.1	2.0	0.6	—	—	0.0
<i>C</i>	1954	100	56.8	34.8	12.3	3.8	0.7	—	0.0
	1959	100	61.0	40.9	18.3	10.0	—	—	0.6
<i>D</i>	1954	100	60.5	40.2	22.8	12.8	5.1	—	0.8
	1959	100	56.5	38.7	22.0	—	—	3.8	1.5

The difference in molecular composition of American and British dextrans is well illustrated by these results. The lower average molecular weight of the American dextrans is associated with shorter retention in the circulation and greater renal excretion. The British dextran, of higher average molecular weight, would induce greater aggregation of red cells *in vitro*² and possibly *in vivo*, too.

From our observations we conclude that during the 5-yr. period there was little, if any, change in the molecular composition of these dextran solutions and none that would be noticeable in clinical use. It is probable that any limitation of the storage 'life' of dextran solution will be imposed by defects in the container and its closure rather than by any instability of dextran solution itself.

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PHARMACOLOGY

Effects of 7-Chloro-2-methylamino-5-phenyl-3H-1,4-benzodiazepin-4-oxide on Mitochondria from Rat Liver and Brain

ALTHOUGH little is known about the relations between *in vivo* drug action and the effects of a drug on metabolic processes *in vitro*, it has been confirmed that psychopharmacological materials in many cases act on oxidative reactions in the mitochondrial respiratory chain. The two different drugs, amytal and chlorpromazine, both, for example, have inhibi-