

same configurational change, since after titration with acid at 38.5°C., both the 6-keto and 6-amino groups ionize without hysteresis.

These results show that ribonucleic acid in solution may be present in one of at least two configurations depending on ionic strength, pH, and temperature. The transition from one configuration to another may be impeded as shown by the hysteresis in the titration cycle. These observations could be accounted for if rotations about the linkages of the sugar phosphate back-bone were sterically hindered. It is possible that one configuration may be stabilized by sequences of intra-molecular bonds, although the sedimentation velocity and intrinsic viscosity of *E. coli* (and also tobacco mosaic virus ribonucleic acid) are consistent with a randomly coiled configuration. Further experiments are required to elucidate the configurations indicated above and to determine the extent to which they may reflect the *in vivo* structure⁸ found for ribonucleic acid in nucleoproteins.

This research was supported in part by a U.S. Public Health Service research grant RG-5217.

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¹ Eisenberg, H., and Littauer, U. Z., *Bull. Research Council of Israel*, **7A**, 115 (1958).

² Littauer, U. Z., and Eisenberg, H., *Biochim. Biophys. Acta*, **32**, 320 (1959).

³ Laskov, R., Margolash, E., Littauer, U. Z., and Eisenberg, H., *Biochim. Biophys. Acta*, **33**, 247 (1959).

⁴ Conway, B. E., *J. Polymer Sci.*, **18**, 265 (1955).

⁵ Cox, R. A., Ph.D. Thesis, Birmingham (1955).

⁶ Cox, R. A., Jones, A. S., Marsh, G. E., and Peacocke, A. R., *Biochim. Biophys. Acta*, **21**, 576 (1956).

⁷ Gierer, A., *Z. Naturf.*, **13b**, 477 (1958).

⁸ Franklin, R. E., Klug, A., Finch, J. T., and Holmes, K. C., *Disc. Farad. Soc.*, **25** (1958).

The acids were extracted from the plant tissue and determined by titration after separation by partition chromatography on a column of silica gel according to methods already described⁴. Recovery of glyceric acid under those conditions is practically quantitative. It is however poorly separated from shikimic acid. The two acids are however readily separated by paper chromatography² and distinguished by the characteristic colour reaction⁵ given with sodium nitroprusside and piperazine after oxidation with periodate. No shikimic acid could be detected in these extracts.

The identity of the D-glyceric acid was established by isolation as the crystalline calcium salt after being separated from other acids by partition chromatography on silica gel followed by ion-exchange chromatography on 'Dowex 1' (acetate form)⁶. The calcium salt had $[\alpha]_D^{18} + 12.8^\circ$ (c, 4, water) and its infrared spectrum was identical with that of an authentic sample of calcium D-glycerate prepared by resolution⁷ of DL-glyceric acid obtained by the oxidation of glycerol⁸.

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¹ Balansard, J., and Arnoux, M., *Med. Tropicale*, **11**, 872 (1951).

² Isherwood, F. A., Chen, Y. T., and Mapson, L. W., *Biochem. J.*, **56**, 15 (1954).

³ Palmer, J. K., *Science*, **123**, 415 (1956).

⁴ DeKock, P. C., and Morrison, R. I., *Biochem. J.*, **70**, 277 (1958).

⁵ Cartwright, R. A., and Roberts, E. A. H., *Chem. and Ind.*, 230 (1955).

⁶ Palmer, J. K., *Conn. Agric. Exp. Sta. Bull.*, 589 (1955).

⁷ Anderson, E., *J. Amer. Chem. Soc.*, **42**, 413 (1920).

⁸ Schering-Kahlbaum, D. R. Patent, 605, 307 (1933).

Glyceric Acid in Broad Bean (*Vicia faba* L.)

In recent years there have been one or two reports of the occurrence of free glyceric acid in higher plants. Balansard¹ identified the acid as the diuretic principle in the pods of broad bean (*Vicia faba* L.) but gave no indication of the amount present other than could be inferred from the diuretic effect. Isherwood, Chen and Mapson² isolated D-glyceric acid from cress seedlings; they found that it was present in amounts ranging from 5 to 50 m. equiv./kgm. of fresh weight in seedlings cultured for 5 days at 20°C. in the dark on 0.04 M sodium bicarbonate solution. Palmer³ isolated D-glyceric acid from tobacco leaves (*Nicotiana tabacum* var. Connecticut) grown in the shade and estimated the quantity present to be of the order of 5-15m. equiv./kgm. (fresh weight).

We have recently determined the levels of D-glyceric acid in leaves and other parts of broad bean plants grown under various conditions and the results (Table 1) show that it is one of the major organic acids accumulated by the plant.

Table 1. PRINCIPLE ORGANIC ACIDS IN *Vicia faba* L.
(m.equiv./kgm. fresh weight)

Sample	Description	Origin	Malic	Citric	Glyceric
1a	leaves	Field	6.6	2.4	28.7
b	stems	Field	13.7	15.6	33.4
2a	leaves	Water culture	12.8	48.0	23.5
b	stems	Water culture	31.4	27.5	34.0
3a	roots	Pot grown	1.3	0.45	1.10
b	leaves (young)	" "	10.8	36.7	44.9
c	leaves (old)	" "	8.5	73.8	24.4
d	stems	" "	13.5	19.7	34.7
e	flowers	" "	16.4	3.7	6.6
f	Pods	" "			

Partial Identification of Lysins and Agglutinins in Lymphomatous Mouse Tissue

LYSINS and agglutinins have been extracted from normal mouse tissue, from mammary carcinomas of female C3H mice, and recently from human leukaemic cells¹, and from the lymphomatous glands of AKR mice². The lytic materials have been tentatively termed 'soap-like' and 'lysolecithin-like', supposedly bound to proteins^{3,4}, but in reality the nature of both the lysins and the proteins is still unknown. This communication is concerned with the last point.

The methods of 'pre-incubation' and of extraction with organic solvents have been largely abandoned because they probably involve the splitting of complexes. Instead, lymphomatous tissue is removed from the AKR mouse, placed in saline in the proportion of 1 gm. of tissue to 3 ml. of saline, and immediately homogenized for 5 minutes in a 'VirTis' homogenizer at 23,000 r.p.m. Gross particles are immediately removed by slow centrifugation. Examination of the supernatant fluid with phase contrast shows innumerable myelin forms and tiny fragments. The supernatant fluid of the homogenate, after the throwing down of the gross particles, is diluted in powers of 2 with Michaelis buffer at pH 8.5. Washed mouse or human red cells are added and both inhibitors (in the less diluted homogenates) and lysins are observed within 3 hours at 37°C.

Identification of the lysins. The homogenate, after the removal of the gross particles, is placed on a strip of fat-free filter paper so that it spreads over about 1 cm. The paper is dried at 56°C.; saturated rhodamine B in benzene and 1 per cent uranyl acetate are added to the paper (a) in the region to