This work was supported partly by a grant from the Consejo Nacional de Investigaciones Científicas y Técnicas and partly by a grant (H-3905) from the U.S. Public Health Service.

MANUEL RENE MALINOW ALVARO L. GIMENO* Amanda A. Pellegrino-Iraldi MARTHA A. FERNANDEZ-GIMENO* JORGE L. LACUARA* JAIME A. MOGUILEVSKY*

Institute of Physiology and Institute of Histology,

Medical School.

Buenos Aires.

Aug. 24.

* Fellow of the Consejo Nacional de Investigaciones Científicas y Fellow of the Consejo Nacional de Investigaciones Cientificas y Técnicas.
Pick, R., Stamler, J., Rodbard, S., and Katz, L. N., Circulation, 4, 648 (1951); 6, 276 (1952).
Icorenz, F. W., Vitamins and Hormones, 12, 235 (1954).
Sperry, W., and Webb, M., J. Biol. Chem., 187, 97 (1950).
Malinow, M. R., Fernández, M. A., Gimeno, A. L., and Bur, G. E., Nature, 183, 1262 (1959).
Malinow, M. W., and Pellegrino, A. A., A.M.A. Arch. Pathol., 61, 11 (1956).

(1956)

(1956).
⁶ Fisher, R. A., and Yates, F., 'Statistical Tables for Biological, Agricultural and Medical Research' (Hafner Publishing Co., Inc., fifth edit., New York, 1957).
⁷ Malinow, M. R., and Pellegrino, A. A., A.M.A. Arch. Pathol., 65, 47 (1958). Malinow, M. R., Pellegrino, A. A., and Ramos, E. H., Proc. Soc. Exp. Biol. and Med., 97, 446 (1958).

Relation of Ethanol Metabolism to Several Factors in the Rat

THE most important site for the oxidation of alcohol is the liver. Various experimental approaches, for example, the artificial perfusion of surviving liver with blood containing alcohol, oxidation of alcohol by liver slices and liver brei, diminished alcohol oxidation following either hepatectomy or damage to the liver, and the demonstration that the eviscerated animal metabolizes alcohol very slowly, have all indicated that the liver is of major importance in the metabolism of ethanol¹.

Eggleton², in an extensive study in cats, attempted to find one or more of the important factors responsible for the varying rates of alcohol metabolism between individuals of the same species. It was found that the rate of ethanol metabolism correlated more nearly with liver-weight than with body-weight or surface area. In the present study, instead of using mongrel animals, we have attempted to obtain consistent results by using a pure strain of rat, for example, Sprague-Dawley females. Additionally, an attempt was made to determine the alcohol-oxidizing capacity of liver by modifying the methods of Bücher and Redetski³ and Marshall and Fritz⁴ for use in the coenzometer (Macalaster Bicknell Co.). It is evident in Table 1, summarizing the findings in 17 female rats, that the results obtained with this modified method fail to correlate significantly with the rate of alcohol metabolism where the latter is expressed in terms of the amount of alcohol disappearing from the blood⁵. The finding of Eggleton² in the cat that the rate of metabolism is more closely related to liver-weight than to body-weight is confirmed in the rat.

Table 1. RELATION BETWEEN RATE OF ETHANOL ME (Disappearing Ethanol) AND SEVERAL VARIABLES METABOLISM

The seal markshalters (The seal day	$\text{Mean} \pm 5.D.^*$	Cor. Coel.
appearing) (mgm./kgm./hr.)	249.0 ± 57.9	
Liver oxid. capacity, wet basis (mgm./		
gm. liver/hr.)	2.74 ± 0.40	0.291*
Liver oxid. capacity, N ₂ basis (mgm./		
gm , N_{z}/hr .)	103.6 ± 14.0	0.382*
Liver-weight (gm.)	7.5 ± 0.6	0.485*
Body-weight (gm.)	327.0 ± 29.4	0.082*
Liver-weight : body-weight (per cent)	2.31 ± 0.27	
* Mean + Standard Deviation of Mean	$: *P > \bar{0} \cdot 05.$	

This investigation was supported by Grant B-1446 from the National Institutes of Health, U.S. Public Health Service.

> F. W. KINARD J. C. AULL, jun.

R. E. ULMER Departments of Physiology and Chemistry, Medical College of South Carolina, Charleston,

South Carolina.

¹ Jacobsen, E., Pharmacol. Rev., 4, 107 (1952).
 ² Eggleton, M. G., J. Physiol., 98, 239 (1940).
 ³ Bücher, T., and Redetzki, H., Klin. Wchnschr., 29, 615 (1951).
 ⁴ Marshall, E. K., jun, and Fritz, W. F., J. Pharmacol. and Exp. Therap., 109, 431 (1953).
 ⁵ Kinard, F. W., Zemp, J. W., and Aull, J. C. jun., Proc. Soc. Exp. Biol. and Med., 88, 663 (1955).

Effect of Tranquilizing Drugs on Survival from Experimental Diphtheric Intoxication

THE exotoxin of virulent diphtheria bacilli represents an extreme example of a potent biochemical stressor in susceptible mammalian species. Giroud et al.¹ have shown that in animals intoxicated with a lethal dose of diphtheria toxin, the hormone content of the adrenal cortex is diminished by as much as two thirds. Attempts to influence the course of diphtheria toxæmia through the use of cortical hormones or adrenocorticotrophic hormone have met with no success^{2,3,4}, although a decrease in adrenal hæmorrhages has been noted⁵.

In view of the potent pharmacological activity of the tranquilizing drugs in stress states, it seemed of interest to determine the effects of such agents upon experimental diphtheric stress. Since mice and rats are refractory to the toxin, large male guinea pigs were employed.

Using groups of eight or more animals paired by weight, age and source, four major tranquilizers were evaluated. These drugs, representing each of the four chemical classes of tranquilizing drugs were meprobamate, chlorpromazine, reserpine and hydroxyzine. Diphtheria toxin, partially purified from broth, with an LD₅₀ of approximately 3×10^5 /ml., was used as the challenge. A dose of 0.1 ml. undiluted sterile toxin was injected intraperitoneally, and

Table 1. SURVIVAL TIME OF GUINEA PIGS TREATED PRIOR TO CHALLENGE WITH DIPHTHERIA TOXIN

	-			
(A) TRANQUILIZER TREATMENT				
Survival	Significance	Experimental	Survival	Degrees
time as	of difference	treatment	time* of	of
% of	from	(single injection,	Controls \pm	Freedom
$Control \pm$	Control	i.p.)	Standard	
Standard				
Error		(mgm./kgm.)	Error	
140 ± 7	P < 0.001	Meprobamate 400	100 ± 5	14
115 ± 4	P < 0.02	Meprobamate 300	100 ± 3	15
$111 \pm 14^{++}$	P > 0.4	Meprobamate 500	100 ± 4	18
110 ± 4	P > 0.1	Reservine 1.0	100 ± 5	14
103 ± 2	P > 0.4	Meprobamate 200	100 ± 3	15
100 ± 4	P > 0.9	Chlorpromazine 25	100 ± 3	14
96 ± 2	P > 0.3	Hydroxyzine 2.5	100 ± 3	14
94 ± 6	P > 0.3	Hydroxyzine 25	100 ± 3	14
92 ± 10	P > 0.4	Reservine 0.1	100 ± 5	14
89 ± 3	P > 0.05	Reserpine 1.4	100 ± 3	14
87 ± 3	P < 0.01	Reservine 2.5	100 ± 3	18
60 ± 8	P < 0.001	Chlorpromazine 200	100 ± 3	14
(B) MISCELLANEOUS TREATMENT				
106 ± 3	P > 0.1	Chloral Hydrate 400	100 ± 3	14
101 ± 4	P > 0.8	Tubocurarine 0.25	100 ± 3	13
97 ± 3	P > 0.4	Pentobarbital 20	100 ± 3	14
$76\pm17**$	P > 0.1	Hexobarbital 100	100 ± 3	13

* The overall mean survival time for all controls was 13 hr. and the average standard error for all control groups was 25 min. (3 per cent). \dagger The large standard error associated with the dose of 500 mgm./kgm. meprobamate can be attributed to 2 early deaths from the drug (which is close to the D_{s_0} at this dose). If these two deaths are removed from the calculations, survival time for this group was 131 ± 6 per cent of its control, which with 16 degrees of freedom is a significant increase, P < 0.001. P < 0.001

** The large standard error is attributed to the fact that 3 animals died within three hours when given 100 mgm./kgm. hexobarbital. If the calculations are made without these 3 animals, the survival time of the group is 110 ± 6 per cent of its control. With 10 degrees of freedom, this increase is not significant, P>0.1.