Table 1

SiO ₂ Al ₂ O ₃ Fe ₂ O ₂ FeO MgO CaO I Na ₂ O K ₃ O H ₂ O ⁺ H ₂ O ⁺ H ₂ O ⁺ H ₂ O ⁻ TiO ₂ MnO	$\begin{array}{c} A1 \\ 97.58 \\ 1.54 \\ 0.11 \\ 0.23 \\ trace \\ 0.38 \\ 0.34 \\ none \\ 0.10 \\ \hline 0.21 \\ trace \end{array}$	A2 96·16 1·31 0·02 0·09 0·03 trace 0·56 0·30 0·51 0·08	<i>B</i> 1 86·34 7·82 0·63 2·08 0·92 0·05 0·15 0·87 0·43 0·03 0·52 nil	$\begin{array}{c} B2 \\ 81 \cdot 25 \\ 9 \cdot 03 \\ 0 \cdot 63 \\ 0 \cdot 40 \\ 2 \cdot 48 \\ trace \\ 0 \cdot 25 \\ 1 \cdot 82 \\ 2 \cdot 81 \\ 1 \cdot 09 \\ 0 \cdot 08 \\ trace \end{array}$	$\begin{array}{c} C1 \\ 80 \cdot 52 \\ 9 \cdot 44 \\ \hline \\ 1 \cdot 98 \\ 1 \cdot 73 \\ 1 \cdot 84 \\ 0 \cdot 52 \\ 3 \cdot 15 \\ 0 \cdot 11 \\ 0 \cdot 05 \\ 0 \cdot 72 \\ 0 \cdot 09 \end{array}$	$\begin{array}{c} 02\\ 80.55\\ 9.87\\ 2.13\\ \hline \\ 0.81\\ 2.21\\ 0.42\\ 2.43\\ 1.06\\ \hline \\ \\ -\end{array}$	$\begin{array}{c} D1 \\ 77 \cdot 76 \\ 13 \cdot 30 \\ 0 \cdot 37 \\ 3 \cdot 36 \\ 1 \cdot 19 \\ 0 \cdot 04 \\ 1 \cdot 41 \\ 1 \cdot 97 \\ 0 \cdot 02 \\ 0 \cdot 76 \\ 0 \cdot 01 \end{array}$	$\begin{array}{c} D2\\ 76\cdot71\\ 12\cdot81\\ 0\cdot80\\ 1\cdot72\\ 0\cdot40\\ 1\cdot90\\ 1\cdot76\\ 1\cdot48\\ 0\cdot77\\ 0\cdot33\\ 0\cdot14\\ 0\cdot27\end{array}$	$\begin{array}{c} E1 \\ 72 \cdot 39 \\ 13 \cdot 12 \\ 0 \cdot 42 \\ 4 \cdot 43 \\ 1 \cdot 87 \\ 3 \cdot 17 \\ 1 \cdot 54 \\ 1 \cdot 92 \\ 0 \cdot 11 \\ 0 \cdot 02 \\ 0 \cdot 76 \\ 0 \cdot 05 \end{array}$	$\begin{array}{c} E2\\72\cdot 40\\12\cdot 56\\\text{none}\\4\cdot 03\\1\cdot 87\\2\cdot 09\\2\cdot 14\\1\cdot 95\\1\cdot 25\\0\cdot 15\\0\cdot 15\\0\cdot 77\\0\cdot 50\end{array}$	$\begin{array}{c} F1 \\ 72 \cdot 40 \\ 12 \cdot 68 \\ 0 \cdot 23 \\ 3 \cdot 59 \\ 2 \cdot 34 \\ 2 \cdot 75 \\ 1 \cdot 68 \\ 3 \cdot 16 \\ 0 \cdot 43 \\ \hline 0 \cdot 74 \\ 0 \cdot 06 \end{array}$	$\begin{array}{c} F2 \\ 71 \cdot 62 \\ 12 \cdot 81 \\ 1 \cdot 64 \\ 1 \cdot 76 \\ 1 \cdot 08 \\ 3 \cdot 14 \\ 2 \cdot 73 \\ 3 \cdot 02 \\ 1 \cdot 93 \\ \\ \\ \\ \\ \\ \\ \\$	$\begin{array}{c} G1 \\ 68{\cdot}60 \\ 15{\cdot}80 \\ 0{\cdot}18 \\ 6{\cdot}46 \\ 2{\cdot}88 \\ 1{\cdot}40 \\ 2{\cdot}35 \\ 1{\cdot}92 \\ - \\ - \\ 0{\cdot}80 \\ 0{\cdot}06 \end{array}$	G2 68 90 13 ·39 3 ·45 1 ·39 4 ·10 3 ·31 2 ·07 2 ·88 0 ·90
	100.49	99.06	99.95	100.24	100.36	99.48	100.19	99.66	99.91	99.88	100.20	99.73	100.45	100.59

9 99.05 99.95 100.24 100.35 99.48 100.19 99.06 99.91 99.88 100.20 89.43 A1, Libyan Desert glass (Univ. Texas Pub. 3945, anal. 65). A2, Granite (marginal variety). Eskdale, England (U.S. Geol. Surv. Prof. Paper 99, 51). B1, Darwin glass (also ZrO₂, 0.11) (Univ. Texas Pub. 3945, anal. 1). B2, Metarhyolite, Shasta Co., Calif. (also S, 0.35; BaO, 0.05) (U.S. Geol. Surv. Prof. Paper 99, 51). C1, Moldavite (Univ. Texas Pub. 3945, anal. 12). C2, Felsite, Co. Waterford, Ireland (U.S. Geol. Surv. Prof. Paper 99, 53). D1, Bediastie (Univ. Texas Pub. 3945, anal. 16). D2, Granite, Settsu, Japan (also P.O., 0.48; ZrO₂, 0.03; S, 0.06) (U.S. Geol. Surv. Prof. Paper 99, 53). E1, Australite (also Nio, 0.06) (Univ. Texas Pub. 3945, anal. 56). E2, Granite, Reefton, New Zealand (also P.O., 0.09; SO., 0.08) (U.S. Geol. Surv. Prof. Paper 99, 331). F1, Indochinite (also P.O., 0.14) (Univ. Texas Pub. 3945, anal. 40). F2, Granite, Stria, Austria (U.S. Geol. Surv. Prof. Paper 99, 103). G1, Ivory Coast tektite (Univ. Texas Pub. 3945, anal. 63). G2, Quartz diorite, Corsica (also P.O., 0.00); (U.S. Geol. Surv. Prof. Paper 99, 331).

must have originated by the fusion of sedimentary rocks is unwarranted.

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Dec. 12.

¹ Urey, H. C., Nature, 182, 1078 (1958).

² Barnes, V. E., Univ. Texas Pub., 3945 (1940). ³ Washington, H. S., U.S. Geol. Surv. Prof. Paper 99 (1917).

Glaciation in the Jammu Hills

MUCH evidence of the last or Pleistocene Ice Age has been observed in various parts of the Central and the Lesser Himalayas. Lieut.-Col. J. L. Grinlinton has referred to the shape of polished and striated or grooved rock surfaces in the Liddar and Sind Valleys of the Kashmir Himalayas¹. Many terminal moraines have also been noticed on the southern flank of the Pirpanjal at altitudes of somewhat less than 8,900 ft.; but no traces of the Pleistocene glaciation have been so far discovered anywhere in the Jammu Hills south of the Pirpanjal.

My colleague, Mr. T. R. Bhatia, and I recently visited Tringla, close to Batote hill station 78 miles from Jammu. Batote is situated at lat. 33.7° and long. 75.20° at an altitude about 5,000 ft. above sealevel. This hill station and the entire country from Udhampur to Peera and leading to Srinagar, Murree and beyond in West Pakistan, to Teetwal in the north-west and for a considerable distance eastwards is covered by the rocks belonging to the Murree series, of Oligocene and Lower Miocene age.

Our attention was attracted in this series by a few outcrops of the compact and hard sandstone which showed an unusual degree of smoothness. A closer examination of these outcrops revealed a number of grooves, roughly parallel to one another, running in a north-south direction and 5–9 in. long. Evidently these were glacial striæ on the polished sandstone dipping roughly southwards at an angle of about 40°, furnishing convincing proofs of their having been produced by glacial action during the Pleistocene Ice Age. The grooves possessed a virtually original freshness and appeared to have been rubbed off from the neighbouring polished surface. In one sandstone projection a fairly large cut with polished sides was seen. This must have been carved

by the silt-charged water when the ice was in the process of melting. Further investigation near the same spot showed a number of large boulders with polished and grooved surfaces. On one of them the grooves were intact, but the polish had completely gone. Similar observations were made about 1 mile to the east towards the Chakou Nala.

In our opinion, the Patni Top area south of Batote, altitude 7,000 ft., served as the gathering ground for the snow and ice during the Pleistocene times and from this snowfield one or more glaciers descended northwards towards Batote. The entire Patni Top area has at present the topography of a big cirque now breached by a couple of small streams.

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¹ Geol. Surv. Ind., Mem. 49, Pt. 2.

Niobium-Zirconium-Thorium-Uranium and Rare-Earth Minerals from the Pegmatites of South Harris, Outer Hebrides

DURING a recent examination of some of the pegmatites in the Lewisian rocks of South Harris a suite of rare minerals was found which may stimulate further research into the numerous, but somewhat neglected, pegmatite occurrences in other parts of the Lewisian of north-west Scotland.

References to the South Harris pegmatites have been made from time to time¹⁻⁴, and Holmes, Shillibeer and Wilson⁵ estimated the age of the potashfelspars from the Chiapaval and Roneval pegmatites by the potassium-argon method and found them to be $1,110 \pm 70$ million years, and $1,140 \pm 70$ million years, respectively.

In the earlier part of the Second World War, the Chiapaval pegmatite near Northton and the Sletteval pegmatites east of Roneval were opened out and quarried for potash-felspar on a considerable scale. These large quarries attracted our interest, and a search for rare minerals was made.

The Chiapaval pegmatite vein, which stretches for a mile, and in places is up to 80 ft. thick, consists of