



Fig. 1

and
$$\sqrt{h}(x_1 + u)/\theta_1 + h(x_2 + v)/\theta_2 \leq 1 \quad (5)$$

Here h is the base of logarithms and v is a parameter, ranging from 0 to 1, measuring the degree of similarity between the physiological actions of the two drugs. When $v = 0$, (3) is the equation for independent action that we obtained before; when $v = 1$, it is the equation for simple similar action.

Fig. 1 shows in plan exemplary regions of the bivariate normal surface defined by (4) and (5), and is comparable to a figure we gave before² to illustrate the regions for independent action. For Fig. 1, $x_1 = -2$, $x_2 = -1$, $\theta_1 = 4.5$, $\theta_2 = 8$ and $h = 10$. The boundaries of integration are the continuous lines marked with the corresponding values of v , namely, 0, $\frac{1}{2}$ and 1. The regions of integration lie to the left of and below the boundaries.

In the special case in which $v = 1$, and $\rho = +1$, (3) becomes

$$h(x_1 - x)/\theta_1 + h(x_2 - x)/\theta_2 = 1 \quad (6)$$

where x is the normal equivalent deviation corresponding to $1 - q$. The equation that we put forward previously for simple similar action was (6) with the restriction that $\theta_1 = \theta_2$. Equation (6) with $\theta_1 \neq \theta_2$ accounts for some data³ on the joint action of certain insecticides closely related chemically.

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P. S. HEWLETT

Pest Infestation Laboratory,
Slough.

R. L. PLACKETT

Department of Applied Mathematics,
University of Liverpool.

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Homoserine in Bleeding Sap of Pea Plants

IN a recent paper¹, it was shown by means of paper chromatography that in bleeding sap of pea plants 5½ weeks old, inoculated with effective strains of *Rhizobium*, some amino-acids and amides occurred. The chief of these compounds proved to be aspartic acid, asparagine, glutamine and a substance giving a colour with ninhydrin, of which the R_F values on nearly all chromatograms corresponded with those of threonine. Only when using phenol/water as solvent was a somewhat higher R_F value found. The other solvents used were: collidine/water; butanol/acetic acid/water; propanol/water².

In an attempt to identify the spot, the following solvents were used: 'Methylcellosolve' (ethylene-glycol monomethylether)/water³; methanol/water/pyridine⁴; *tert.*butanol/methylethylketone/water/diethylamine⁴. Of these three solvents, only the latter produced for the unknown substance an R_F value deviating from that of threonine, as phenol/water did.

By comparing a mixture of known amino-acids (including homoserine) with the substance on the chromatograms, and by adding homoserine to the bleeding sap before running the solvents, the unknown spot was finally identified⁵⁻⁹ as homoserine (an isomer of threonine).

This homoserine was found for the first time in 1953 by A. I. Virtanen and co-workers^{5,6} to be one of the principal free amino-acids (70 per cent ethanol-extract) occurring in pea plants. Later on it was found by Virtanen and co-workers⁷ to occur as such also in other plants (even in non-legumes) although in smaller quantities.

The procedure described above proves that for chromatographical identification of amino-acids in fluids of unknown composition, it is imperative to determine R_F values in several different solvents (five at the least), before deciding upon the identity of a spot.

JANNY A. BAKHUIS

Laboratory of Microbiology,
Agricultural University,
Wageningen,
The Netherlands.

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Abnormal Metamorphosis in Saturniid Moths infected by a Microsporidian

DEVIATIONS from the normal course of metamorphosis in insects may be caused by parasitic infections^{1,2}. In studies on a microsporidian of the genus *Nosema*, found in *Hyalophora (Platysamia) cecropia* and afterwards sub-cultured in *Antheraea polyphemus* and *A. pernyi*³, seventeen of the many infected animals failed to metamorphose normally. The most severely affected were adults of *H. cecropia*. They retained a number of pupal structures. The