It is noteworthy that the structures of compounds containing more than one alkaline-earth ion, barium and calcium, or strontium and magnesium, for example, conform more closely to the ideal cubic symmetry than do the structures of compounds containing a single alkaline-earth ion. It appears that when the alkaline-earth atomic positions are occupied by two kinds of atoms of appreciably different ionic size, this allows packing of the other ions without much distortion, and the lattice as a whole can assume an undeformed cubic ayumetry. mmetry. It is hoped to publish these observations in greater detail at a later

H. P. ROOKSBY. E. G. STEWARD.

Research Laboratories, General Electric Co., Ltd., Wembley. Jan. 22.

<sup>1</sup> Bragg, W. L., "Atomic Structure of Minerals".

<sup>2</sup> Rooksby, H. P., Nature, 155, 484 (1945).

## Artificial Condensation Nuclei and Relative Humidity

Artificial Condensation Nuclei and Relative Humidity
WORKING at Heidelberg, during 1942-44, I have investigated the
change in size of artificially produced condensation nuclei with variation of relative humidity, by measuring the diminution of conductivity
of air through ion-adsorbing suspensoids in a chamber ionized by
gamma rays. A differential compensation method was used, the principle of which was first indicated by Greinacher in Switzerland in 1922.
The results enabled a new examination to be made of the theory of
the growth of solution nuclei with relative humidity, commonly
associated with the name of the Norwegian scientific worker Koehler.
I have been able to show that, contrary to the accepted view, substances of different hygroscopic power, namely, strongly hygroscopic, less hygroscopic and non-hygroscopic particles, show very
different behaviour. Compared with the extreme importance of
the individual chemosorptional hygroscopicity, other characteristics do not seem to influence considerably the growth of the nuclei
with relative humidity. Salt nuclei generally showed no tendency
to crystallize even at low humidities.

Details of this work will be published as soon as possible.

NAVINKUMAR DALAL.

Dantestr. 25, Heidelberg.

### Neutron and Negative Proton

Neutron and Negative Proton

About a year ago, I proposed a method of detecting the decay of the neutron¹. Recently, it has been reported that Kapitza has, in the course of the analysis of cosmic ray spectra, discovered the negative proton. Whether this news is true or not, it is quite reasonable to suppose that a free neutron has also the probability of being  $\beta$ + radioactive\*, thus producing a negative proton. This process would, of course, occur only if the mass of the negative proton is smaller than that of the neutron. To detect the existence of this negative proton, one may apply the method similar to that proposed in my previous note¹ and then search for the  $\beta$ + rays. If the negative proton is the anti-particle of the ordinary proton, the annihilation process might be expected to occur with the emission of one or two photons of energy about 10° eV. These photons will, on encountering the wall of the containing vessel, produce electron showers of a considerable number of particles which may easily be detected by using a counter-controlled cloud chamber arranged in usual ways. These showers can very easily be distinguished from the showers of cosmic rays since we know that the latter are mostly in downward directions.

Department of Physics, National University of Chekiang, Meitan, Kweichow. Dec. 3.

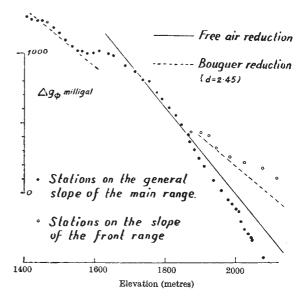
<sup>1</sup> Wang, K. C., Nature, 155, 574 (1945).

\* The suggestion that possibly a free neutron might be  $\beta+$  radioactive was made in a speculative paper by Gamow (*Phys. Rev.*, 45, 728; 1934).—Editors.

#### Gravitational Data from the Nan Shan Range

Gravitational Data from the Nan Shan Range STUDYING the gravitational data obtained in the front of the Nan Shan Range, China, interesting facts pertaining to the crustal structure of the earth are revealed. The gravimetric survey was not primarily carried out for this academic purpose. The geodetic results are nevertheless interesting. The variation of the gravitational acceleration was measured by static gravity meters with a sensitivity of  $0\cdot 1$  milligal, and the elevation of the stations was determined by a transit to within a few feet.

The gravity variation of long linear traverses, after reduction to one common latitude  $(\Delta r_{\rm g})$ , is plotted separately against the elevation of the stations. Then we find that the isostatic compensation and the free air reduction of the elevation fit well in front of the main range. Beyond a sharp boundary, however, the picture changes. Usually the Bouguer reduction becomes the controlling factor. In the accompanying figure is shown a typical example. The gravity anomaly and the elevation along a traverse across the fanning slope of the mountain down to the extensive expanse of the desert below are plotted in solid dots. In this particular case the sharp transition occurs at thirty miles from the thrust zone of the main range. Therefore the crust of the earth seems here to be blocky in nature.



Our data also confirm the current conception that minor land features like the front range, which may be thirty to seventy miles in base and more than 1,500 ft. above the surrounding country, is not even slightly compensated in isostasy. In the same figure, the circles represent the data of a side traverse up the slope of such a front range. It appears that this kind of sensitive criterion can be usefully employed in geographical and physiographical classifications.

Kan-su Petroleum Administration, Su-chow, Kan-su, China. Dec. 14.

# Action of Tyrocidin and Detergents in Liberating Amino-Acids from Bacterial Cells

Action of Tyrocidin and Detergents in Liberating Amino-Acids from Bacterial Cells

The estimation of amino-acids by the amino-acid decarboxylase technique¹ provides a means for the investigation of the amino-acid content of the internal environment of bacterial cells. If 1 ml. of a thick washed suspension of S. feealis cells grown in casein digest medium is placed along with 1.5 ml. phosphate buffer pH 6-0 in the main cup of a Warburg manometer and 0.5 ml. lysine decarboxylase preparation tipped in from a side-bulb in the usual way, there is a small output of carbon dioxide of magnitude depending upon the degree of washing of the cells and corresponding to the amount of lysine carried down with and adsorbed on the surface of the cells (see curve labelled 'intact cells'). If a further aliquot of cell suspension is treated in such a way as to disintegrate the cells, either by mechanical shaking with glass beads' or by immersion in a boiling water bath for 10 min., before the addition of the enzyme, the amount of lysine assayed increases markedly (curve labelled 'boiled cells'). The difference in the quantities of lysine measured with the intact cells and with the disintegrated cells corresponds to the lysine liberated from the internal environment by disruption of the cell-wall, through which the enzyme cannot pass. Investigation of the amount of such lysine contained in the internal environment of the cells shows that its concentration depends upon that existing in the external environment, the ratio of the internal to external concentration varying from 2-3 for external concentrations of the order 0.02 M to approximately 20 for external concentrations of the order 0.02 M to approximately 20 for external concentrations of the order 0.02 M to approximately 20 for external concentration of glucose, etc. When cells containing a high internal environment, but that the passage of glutamic acid across the cell-membrane requires energy which can be supplied by the simultaneous fermentation of sysine and glutamic acid a