Structure of a New System of CO Bands

In a previous note¹ the discovery of a new system of red-degraded CO bands, placed just before the heads of the violet-degraded Third Positive system, was reported. Inspection and measurement of highdispersion spectrograms now show that the new bands have a structure appropriate to a $\Sigma \rightarrow \Pi$ bandsystem, the lower state being the $a^3\Pi$ state, which is the usual final state of the CO triplet systems. The upper ${}^{3}\Sigma$ levels of the new bands lie about 83705 cm.⁻¹ and 85885 cm.-1 above the CO ground state; they have observable spin-splittings and rotational constants B' = 0.7 to 0.8 cm.⁻¹. The system may tentatively be designated as $f^3\Sigma \rightarrow a^3\Pi$. It may be considered, however, as probably due to transitions from the higher vibrational levels of $a'^{3}\Sigma$ to the lower vibrational levels of $a^{3}\Pi$. For, within the heights of the initial levels $f^{3}\Sigma$ of the new bands occur also the initial levels $b^{3}\Sigma$ of the Third Positive bands, and the latter are strongly perturbed, as has already been established², by vibrational levels of the $a^{\prime 3}\Sigma$ state, for which we have predicted term values and rotational constants of the magnitude now observed for the new bands.

Rotational analysis and details will be published shortly elsewhere. R. SCHMID.

Physical Institute, Royal Hungarian University, Budapest. July 31.

Schmid, R., and Gerö, L., Naturwiss., 25, 90 (1937). ² Schmid, R., and Gerö, L., Z. Phys., 105, 36 (1937) 106, 205 (1937).

L-Emission Bands of Zinc, Copper, Nickel and Cobalt

I HAVE studied the intensity of the L-emission bands of metals from Co (27) to Zn (30). The apparatus used was a vacuum spectrograph using a bent crystal of mica, and the source of the radiation was an X-ray tube.

It is well known that the L-emission bands are emitted when $M_{IV, V}$ -electrons make transitions into the $L_{II, III}$ -levels ($L_{a_1,2}$ and (L_{β_1}). There are also transitions of $N_{\rm I}$ -electrons into the $L_{\rm II, III}$ -levels. For these metals the N_{I} or 4s-electrons are the conduction electrons, and occupy levels which overlap in energy those of the $M_{IV, V}$ or 3*d*-electrons.² Thus the observed bands represent transitions of both 3d- and 4s- electrons. But for copper and zinc, unfortunately, the short wave-length ends of the bands, which are probably due to the 4s-electrons, are completely masked by satellites, which are strongly enhanced by the reorganization of the atom on account of the Auger effect³, L_I-L_{III}.

The position of the short wave-length edge of the L_a -bands may be estimated from the L-absorption spectra, and is given by the wave-length of the slope of the absorption curve near the minimum absorption. For copper, we can deduce from the absorption curve of Sandström⁴ that the edge of the L_a -emission band is given by the arrow marked on Fig. 1. From the comparison of the L-emission and absorption bands it is obvious that the width of the L_a -band is about 5 ev.

Université, Liége. Aug. 3.

¹ Comptes rendus, 204, 1242.

a See Mott and Jones, "The Theory of the Properties of Metals and lloys" (Oxford, 1936), p. 191. Alloys

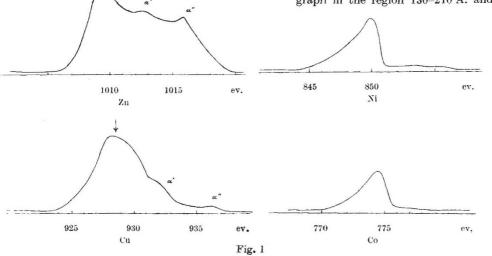
³ See Coster and Kronig, Physica, 2, 13 (1935).

⁴ Sandström, A., Thesis, Uppsala (1935).

M-Emission Bands of Zinc, Copper and Nickel

WE have had the opportunity of seeing Dr. Farineau's letter before it was sent to press. We have recently observed the M-emission bands of nickel, copper and zinc metals, and it seems worth while to add a note, since the two sets of results, especially taken together, allow one to draw certain conclusions about the structure of the 3d- and 4s(conduction)-electron bands in these metals.

Our plates, taken with a concave grating spectrograph in the region 130-210 A. and with an X-ray



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tube run at 3,000 v. as source, give results which, on a scale of energy, are almost identical with Farineau's L - emissionbands. This is good evidence that, apart from transition probability factors, the curves represent features of the density. functions of the combined 3dconduction. levels of the

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To obtain the actual intensity distribution of the emitted radiation, the plate was calibrated by a method already described¹. The curves giving the intensity of the L_a bands against energy (in electronvolts) are shown in Fig. 1.

metals. We also obtain for zinc and copper the peaks marked α' and α'' by him. But, contrary to his conclusion, when we compare the wave-lengths of points on the M-emission bands with the wave-lengths of the *M*-absorption edges which we have determined¹, we