

astronomer Torvald K ohl (*Oversigt over det Kgl. Danske Vidensk. Selsk.'s Forh.*), and it consists of 142 measurements of the lowest observed points of meteors (*limes inferior*). The material, obtained in latitude 56° N. during the years 1875–1917, is very accurate, the mean error of the heights being less than 1 km.

The mean height of the *limes inferior* depends on the mass of air above this limit. Therefore, if the tides of the upper atmosphere have no phase-difference in relation to the tidal forces (which is nearly the case at ground level), the mean height of the lower limits may be expected to vary in the following manner :

$$h = h_0 + l \cos 2t$$

where h_0 is the mean height and l the amplitude of a variation depending on the moon's hour-angle (t). The material has been divided arbitrarily into two parts, and then each of these parts is divided in accordance with the moon's hour-angle, namely :

- (1) flood-tide : $-30^\circ < t < 30^\circ$ and $150^\circ < t < 210^\circ$;
 (2) +0 : $30^\circ < t < 45^\circ$, $135^\circ < t < 150^\circ$, $210^\circ < t < 225^\circ$ and $315^\circ < t < 330^\circ$;
 (3) -0 : $45^\circ < t < 60^\circ$, $120^\circ < t < 135^\circ$, $225^\circ < t < 240^\circ$ and $300^\circ < t < 315^\circ$;
 (4) ebb-tide : $60^\circ < t < 120^\circ$ and $240^\circ < t < 300^\circ$.

The following table gives the mean values of h in kilometres :

Years.	Flood-tide.	+0	-0	Ebb-tide.
1875–1902	87.7	90.6	81.0	79.9
1904–1917	92.7	88.6	78.8	82.8
Whole period	89.3	89.4	79.6	81.6

From these values are found : $h_0 = 85.0 \pm 1.2$ km. ;
 $l = 5.5 \pm 1.9$ km., thus :

$$h = 85.0 + 5.5 \cos 2t(\text{km.}).$$

Assuming from theory that, on an average, the meteors will be extinguished when they have passed through the same mass of air, the expression for h shows that the mass of air situated above 90.5 km. at flood-tide should be equal to the mass of air situated above 79.5 km. at ebb-tide. The corresponding values formerly found by the investigation concerning the aurora were 106 km. and 100 km., respectively. Taking into consideration that the former phenomenon is observed in latitude 56° and refers to heights about 85 km. above the ground, while the latter phenomenon is observed in latitude 70° and refers to heights about 103 km., a fair agreement between the two results will be found.

A further examination of other and if possible more numerous observations of lower limits of the meteors would be of importance for the investigation of the tides of the upper atmosphere. J. EGEDAL.

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The Oxidation of Acetaldehyde by Oxygen.

IN the course of experiments which have been carried out during the last two years on the photochemical oxidation of acetaldehyde, an observation has been made which does not seem to be recorded in the literature. It has been found that when pure liquid acetaldehyde is shaken at ordinary temperature with oxygen in daylight or in the dark, a rapid absorption of gas occurs and a peroxide is formed. 2 c.c. of liquid aldehyde, shaken for one minute with oxygen, absorb more than 1 c.c. of gas, and on adding the liquid to a solution of potassium iodide, iodine is liberated equivalent to about 8 c.c. of n/100 thio-sulphate solution.

The formation of peroxides under these conditions

from other aldehydes has long been known, and B ckstr m (*Medd. K. Vetenskapsakad. Nobel-Inst. 6, No. 16*) has recently investigated the chain mechanism of peroxide (benzoperacid) formation in the case of benzaldehyde. B ckstr m assumed that the photochemical rate of peroxide formation is proportional to the light intensity. Mr. Chatwin, working in this laboratory, has shown that the rate of benzoperacid formation is more nearly proportional to the square root of the light intensity.

We have obtained similar results for acetperacid formation when gaseous acetaldehyde and oxygen are exposed to ultra-violet light. The reaction rate is high, indicating a long chain mechanism, and is approximately proportional to the square root of the light intensity, proportional to the aldehyde concentration, and independent of the oxygen concentration. These facts can only be explained by a mechanism of the following type :

1. $A + h\nu \rightarrow A^+$
2. $A^+ + O_2 \rightarrow AO_2^+$
3. $AO_2^+ + A \rightarrow A^+ + AO_2$
4. $2AO_2^+ \rightarrow ?$

That is to say, the chain mechanism given by reactions 2 and 3 must be assumed to be terminated by the reaction 4 to explain the observed kinetics. This otherwise improbable mechanism is rendered possible by the great length of the reaction chains.

We are now carrying out experiments on the oxidation of acetaldehyde in solutions, and full details of all the measurements will be published when further results are obtained.

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New Phenomena in a Sounding Dust Tube.

I HAVE found the clear photographs of the antinodal ring of dust in a Kundt's tube which were published in NATURE on Nov. 9, p. 724, by Prof. Andrade and Mr. Lewer of special interest inasmuch as, while using a rod excited tube and *Kiesels ure* powder, I observed (*Phil. Mag.*, vol. 7, p. 523, March 1929) an antinodal cloud and stated that "the antinodes are marked almost as definitely as the nodes". As the dust was photographed when the note had ceased, a photograph as detailed as that of Prof. Andrade and Mr. Lewer was not obtained, although a ring-like grouping at the antinode is clearly suggested in Fig. 7 (c') (l.c.).

I have observed that the striae tend to lean over towards the nearest antinode and, in the photograph published in NATURE, it may be seen that this tendency is marked to a minor degree. Also in my paper I pointed out that the figures depend largely on the powder used, and it would be of interest if, in their fuller account, Prof. Andrade and Mr. Lewer state the material they employed and the diameter of their tube.

Like Croft (*J.O.S.A. and R.S.I.*, 14, p. 431, May 1927) I have used a valve oscillator to obtain striae, and I was—before my experiments had to be temporarily suspended some months ago—photographing the figures kinematographically. The results of this investigation I hope to publish in due course, and, in the meantime, I look forward with interest to the contribution Prof. Andrade and Mr. Lewer have promised to make to this fascinating subject.

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 Nov. 19.