

OPEN Activation volume of selected liquid crystals in the density scaling regime

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In this paper, we demonstrate and thoroughly analyze the activation volumetric properties of selected liquid crystals in the nematic and crystalline E phases in comparison with those reported for glass-forming liquids. In the analysis, we have employed and evaluated two entropic models (based on either total or configurational entropies) to describe the longitudinal relaxation times of the liquid crystals in the density scaling regime. In this study, we have also exploited two equations of state: volumetric and activation volumetric ones. As a result, we have established that the activation volumetric properties of the selected liquid crystals are quite opposite to such typical properties of glass-forming materials, i.e., the activation volume decreases and the isothermal bulk modulus increases when a liquid crystal is isothermally compressed. Using the model based on the configurational entropy, we suggest that the increasing pressure dependences of the activation volume in isothermal conditions and the negative curvature of the pressure dependences of isothermal longitudinal relaxation times can be related to the formation of antiparallel doublets in the examined liquid crystals. A similar pressure effect on relaxation dynamics may be also observed for other material groups in case of systems, the molecules of which form some supramolecular structures.

Liquid crystalline (LC) phases are formed by molecules having strongly anisotropic shapes. In the present case the elongated (rod-like) molecular systems are considered. Depending upon the chemical structure of the rigid molecular core and two terminal groups, such substances can form the nematic (N) phase with long-range orientational order of the symmetry molecular axes, and/or the smectic phases, showing besides the nematiclike orientational order, some degree of translational order (Sm A and Sm C with a translational order in one dimension and Sm B, I, F, L with an additional bond-orientational long-range order) as well as several crystalline phases classified as hexagonal in-plane lattice (B, J, G, M) and rectangular in-plane lattice (E, K, H, N)¹.

In the last twenty five years, numerous organic substances forming liquid crystalline (LC) phases were intensively studied under elevated pressure (see reviews refs 2 and 3). Basically, three experimental methods were employed that elucidated different physical properties of the examined substances. Differential thermal analysis (DTA), supported some times by the polarizing optical microscopy (POM) observations, enabled determination of the p-T phase diagrams. Dielectric spectroscopy (DS) measurements yielded information about the molecular dynamics in the LC phase as functions of temperature and pressure²⁻⁶. Pressure-volume-temperature (pVT) studies of selected substances were the basis for a deep analyses of the observed phenomena under different thermodynamic conditions: isobaric, isothermal and isochoric. For example, it was found that the slope of the T(p) clearing line (LC phase - isotropic liquid transition) depends on the details of the molecular structure whereas the contribution of the configuration entropy to the total entropy change depends upon the type of the LC phase³.

A main part of high pressure investigations of LCs has been devoted to study the molecular rotations around the short axes in different LC phases. The corresponding low-frequency relaxation time τ_{\parallel} of this process can be parameterized in terms of activation quantities:

activation enthalpy:

$$H_{act} = R \left(\frac{\partial \ln \tau}{\partial (1/T)} \right)_{p} \tag{1}$$

¹Institute of Physics, University of Silesia in Katowice, ul. Uniwersytecka 4, 40-007 Katowice, Poland. ²Silesian Center for Education and Interdisciplinary Research, ul. 75 Pułku Piechoty 1a, 41-500 Chorzów, Poland. ³Institute of Physics, Jagiellonian University, S. Łojasiewicza 11, 30-348 Kraków, Poland. Correspondence and requests for materials should be addressed to A.G. (email: andrzej.grzybowski@us.edu.pl)

activation energy:

$$E_{act} = R \left(\frac{\partial \ln \tau}{\partial (1/T)} \right)_{V} = H_{act} - T \left(\frac{\partial p}{\partial T} \right)_{V} V_{act}$$
(2)

activation volume:

$$V_{act} = RT \left(\frac{\partial \ln \tau}{\partial p} \right)_{T} \tag{3}$$

It should be noted that the activation quantities were suggested by Williams⁷ a half century ago to investigate the dielectric relaxation in different thermodynamic conditions by adopting the Eyring transition state theory^{8,9} to characterize the thermodynamic evolution of the characteristic timescales (called relaxation times τ) for different dielectric relaxation processes. The relation derived by Eyring for the rate k at which species relax between two nonactivated states via an activated one has been successfully applied by Williams to the dielectric relaxation time $(\tau = 1/k)$. Nevertheless, it is worth mentioning that the considered dielectric relaxation is in general not any relaxation of the order parameter to its equilibrium value near a second-order phase transition, especially in case of the liquid-glass transition, which falls outside the thermodynamic classification of phase transitions due to its kinetic character recently expressed by some of us via a universal equation for the pressure coefficient of the glass transition temperature dT_e/dp^{10} . The concept of the dielectric relaxation time has been defined by Debye for polar liquids as the time required for dipole moments of molecules to revert to a random distribution after removal of the applied field11. In given thermodynamic conditions, the dielectric relaxation process considered by Debye is characterized by a single relaxation time τ_D and a relatively narrow and symmetric dielectric loss peak in the frequency domain. Most dielectric relaxation processes observed experimentally (including those analyzed here for liquid crystal and glass forming systems) exhibit dielectric relaxation loss peaks that are broadened (symmetrically or asymmetrically) in comparison with the Debye dielectric relaxation process and often represented as a superposition of the Debye processes with different relaxation times τ_D at given thermodynamic conditions (T, p). It means that a distribution of microscopic relaxation times is mostly assumed for a non-Debye relaxation process as has been for instance suggested by Powles in context of discussing the effect of a few internal fields on the dielectric relaxation time and the dielectric spectrum¹². Using experimental data, the estimated macroscopic dielectric relaxation time τ of such a non-Debye dielectric relaxation process is usually determined from the frequency f_{max} at the maximum of the measured dielectric relaxation loss peak, i.e., $\tau = 1/2$ $(2\pi f_{max})$. Then, the meaning of the activation quantities defined by Eqs (1-3) can be discussed in terms of a generalized Arrhenius law for different thermodynamic conditions, $\tau(T, p) = \tau_0 \exp[(H_{act} + pV_{act})R^{-1}T^{-1}]$, shown here for the temperature-pressure domain, where H_{act} and V_{act} can depend on thermodynamic conditions⁷. Thus, H_{act} is a macroscopic apparent activation enthalpy reflecting a change in enthalpy, which is needed by a dipolar entity to reorient between two local energy minima at p = const, E_{act} is an analogous macroscopic apparent activation energy at V = const, and V_{act} is a macroscopic apparent activation volume reflecting a free surrounding space needed by a dipolar entity to reorient between two local energy minima at T = const. Therefore, the activation volume is a key parameter that characterizes the isothermal pressure dependences of the dielectric relaxation times, which are analyzed in this paper.

All the activation quantities have been found to be very useful to study the rapid increase in the structural relaxation time τ of glass forming (GF) systems, which is observed when a GF liquid is approaching the glass transition. This phenomenon is called the super-Arrhenius behavior due to increasing values of H_{act} and V_{act} under isobaric cooling and isothermal compression, respectively¹³. For many nematics, smectics, and the crystalline E phase (Cr E) that reveals certain orientational degrees of freedom besides the long-range three-dimensional crystalline order, the activation energy is roughly half the value of the activation enthalpy, whereas the activation volume is ca. 20% of the molar volume. Thus, LC systems show many similarities with the behavior of the structural relaxation times in the glass forming (GF) systems. Unfortunately, the theoretical basis for discussion of the dynamic properties of molecules in LC phases is rather poor, in contrast to GFs. Therefore, our analyses of the thermodynamic aspects of the $\tau_{\parallel}(pVT)$ behaviors have been based on the power density scaling (PDS) law, which was originally formulated for GF systems and successfully verified for many material groups of GFs, but also satisfactorily applied to describe the thermodynamic evolution of the longitudinal relaxation times of various LC systems in different LC phases^{3,6,14-19}. In this context, it is worth noting that there are two significant differences between the LC and GF systems: (i) the range of T and p in which a LC phase is thermodynamically stable is rather limited, in comparison with the supercooled regime of GF systems; (ii) there is no experimental evidence for non-Arrhenius temperature-dependences of τ_{\parallel} within LC phases, whereas such behavior is a common feature of GF substances. In addition, it should be borne in mind that the number of LC materials studied at elevated pressure is rather limited (about 20), whereas hundreds of GF materials have been investigated at high P. It was found that τ_{\parallel} determined for many LCs forming different phases (nematic, smectic A, smectic C, crystalline E, cholesteric) can be successfully scaled to a master curve like in the GF systems as a function of a single scaling variable, ρ^{γ}/T , where ρ is the material density and the scaling exponent γ is a material constant, which may depend on LC phases of a given LC system^{3,6–16,18,19}.

It should be emphasized that the power law density scaling has been acknowledged as a promising unifying idea in the glass transition physics²⁰ and successfully used to differentiate between physically relevant and irrelevant models of the thermodynamic evolution of the characteristic timescale of molecular dynamics in the GF systems²¹. Since the case of LC systems has shown that the applicability of the density scaling law goes beyond the

GF systems, a comparative study of LC and GF systems in the density scaling regime are highly needed to give a better insight into their universal description.

In this paper, we concentrate our attention on the analysis of the activation volume in the density scaling regime taking into account the theoretical models recently developed and checked for GF systems^{22,23}. Nevertheless, our motivation of this work is two-fold: (i) to thoroughly investigate unusual pressure effects on molecular dynamics of LC systems, which are opposite to those observed for GF systems, and (ii) to verify whether leading entropic models formulated in the power law density scaling limit from Avramov and Adam-Gibbs models, which are based respectively on the total and configurational entropies, can be also employed in the investigations of LC systems similarly to their successful applications to study GF systems. Consequently, we shed a new light on the fervently discussed problem in condensed matter physics and material engineering sciences as to how to properly and most universally model the thermodynamic evolution of the timescale of molecular dynamics in different materials.

Theoretical models

The activation volume defined by Eq. (3) is a fundamental characteristic of materials subjected to squeezing, which naturally informs us about the sensitivity of molecular dynamics of the materials to changes in pressure under their isothermal compression. In general, values of the parameter V_{act} depend on thermodynamic pVT conditions and can be expressed for instance as a temperature-pressure function $V_{act}(T,p)$. However, high pressure experimental data are usually measured too rarely to accurately determine such a function $V_{act}(T,p)$ in a direct way. Therefore, we need to exploit some models to describe the experimental dependence $\tau_{\parallel}(T,p)$, and then reliably evaluate the function $V_{act}(T,p)$.

As already mentioned, both GF and LC systems have been successfully studied in terms of the thermodynamic scaling idea, which provides us an attractive way to relate macroscopic experimental dynamic quantities (such as diffusivity, viscosity, structural relaxation time, segmental relaxation time in polymers, and longitudinal relaxation times in LCs) to an effective intermolecular potential, $U_{eff}(r) = 4\varepsilon (\sigma/r)^{m_{IPL}} - A_t$, which includes a small attractive background A_t and a dominant repulsive inverse power law (IPL) term with $m_{IPL} \approx 3\gamma$, where γ is the scaling exponent of the scaling variable ρ^{γ}/T for a given material. The simple soft-sphere potential discussed to be valid for viscous liquids and disordered solids highly inspired a development in the condensed matter physics. In the last decade, a few models have been suggested ^{23–28} to describe the temperature-density dependences of the dynamic quantities, $Y(T, \rho)$, which obey the power law density scaling law, $Y = f(\rho^{\gamma}/T)$. Among them, the temperature-volume versions of Avramov^{26,27} and MYEGA²³ models have attracted a lot of interest due to their entropic fundamentals, especially that the Avramov model is based on the total system entropy $S^{26,27,29-31}$ while the MYEGA model has been derived^{23,32} from the well-known Adam-Gibbs model³³, which relies on the configurational entropy S_c recently shown²¹ to obey the PDS law with the same value of the scaling exponent γ that scales structural relaxation times of GF systems. Therefore, we have considered both the models in our investigations of the activation volume of LC systems,

T-V Avramov:

$$\tau_{||} = g_A(T, V) = \tau_0 \exp\left[\left(\frac{A}{TV^{\gamma}}\right)^D\right]$$
(4)

T-V MYEGA:

$$\tau_{\parallel} = g_{M}(T, V) = \tau_{0_{M}} \exp \left[\frac{D_{M}}{TV^{\gamma_{M}}} \exp \left(\frac{A_{M}}{TV^{\gamma_{M}}} \right) \right] \tag{5}$$

where the specific volume $V = \rho^{-1}$ and τ_0 , γ , A, D are fitting parameters of the T-V Avramov model and their symbol counterparts with the subscript M are fitting parameters of the T-V MYEGA model, respectively.

Since the PDS law, $Y = f(\rho^{\gamma}/T)$, can be extended to a more general form^{25,34}, $Y = g_{\text{mod}}(\Gamma)$, where the scaling variable, $\Gamma = h(\rho)/T$, can be expressed by a density function $h(\rho)$ which is not limited to the power density function $h(\rho) = \rho^{\gamma}$ with a constant scaling exponent γ , we discuss the activation volume of a dynamic quantity given by the function $Y = g_{\text{mod}}(\Gamma)$. Then, Eq. (3) can be represented as follows, $V_{act} = RT$ ($\partial \ln g_{\text{mod}}(\Gamma)/\partial p)_T$, and separated into two terms $V_{act} = RT$ ($\partial \ln g_{\text{mod}}(\Gamma)/\partial \Gamma$) ($\partial l\Gamma/\partial p)_T$, where $RT(\partial l\Gamma/\partial p)_T$ is independent of the assumed model $Y = g_{\text{mod}}(\Gamma)$ and ($\partial \ln g_{\text{mod}}(\Gamma)/\partial \Gamma$) is the model-dependent derivative. The former term $RT(\partial l\Gamma/\partial p)_T = RT\gamma\Gamma/B_T(p)$, where $B_T(p) = (\partial \ln \rho/\partial p)_T^{-1}$ is the isothermal bulk modulus for volume and the scaling exponent $\gamma = \partial \ln h(\rho)/\partial \ln \rho$ is defined in a general density-dependent form³⁵, which can be reduced to a constant parameter in the PDS law limit, i.e., if $h(\rho) = \rho^{\gamma}$ with $\gamma = const$. Thus, one can derive the following general formula for the activation volume in the general density scaling (GDS) regime

$$V_{act} = RT \left(\partial \ln g_{\text{mod}}(\Gamma) / \partial \Gamma \right) \gamma \Gamma / B_T(p)$$
(6)

If we consider the T-V Avramov and T-V MYEGA models, $\tau_{\parallel} = g_A(\Gamma)$ and $\tau_{\parallel} = g_M(\Gamma)$, given by Eqs (4) and (5), respectively, the model-dependent contributions ($\partial \ln g_{\text{mod}}(\Gamma)/\partial \Gamma$) to Eq. (6) will be as follows

$$(\partial \ln g_A(\Gamma)/\partial \Gamma) = DA^D \Gamma^{D-1}$$
(7)

$$(\partial \ln g_M(\Gamma)/\partial \Gamma) = D_M(1 + A_M \Gamma) \exp(A_M \Gamma)$$
(8)

A few years ago, an equation of state (EOS) was derived 22 to facilitate analyses of the activation volumetric properties, e.g., the isothermal bulk modulus for the activation volume defined by analogy with $B_T(p)$ as $B_{act}(p) = (\partial \ln \rho_{act}/\partial p)_T^{-1}$, where $\rho_{act} = V_{act}^{-1}$. In the activation volumetric EOS,

$$\left(\frac{V_{act}(T, p_0)}{V_{act}(T, p)}\right)^{\gamma_{act}} = 1 + \frac{\gamma_{act}}{B_{act}(T, p_0)}(p - p_0),$$
(9)

two temperature parametrization have been used

$$V_{act}(T, p_0) = F_0 + F_1(T - T_0) + F_2(T - T_0)^2$$
(10)

$$B_{act}(T, p_0) = B_{act}(T_0, p_0) \exp(-g_2(T - T_0))$$
(11)

where $F_0 = V_{act}(p_0, T_0)$, $F_1 = \partial V_{act}(p_0, T)/\partial T|_{T=T_0}$, $F_2 = (1/2)\partial^2 V_{act}(p_0, T)/\partial T^2|_{T=T_0}$, and $g_2 = -\partial \ln B_{act}(p_0, T)/\partial T|_{T=T_0}$. Equation (9) has been well interpreted in terms of the thermodynamic scaling idea²², because this activation volumetric EOS has been patterned on an volumetric EOS earlier derived^{36,37} in the PDS regime,

$$\left(\frac{V(T, p_0)}{V(T, p)}\right)^{\gamma_{EOS}} = 1 + \frac{\gamma_{EOS}}{B_T(p_0)}(p - p_0)$$
(12)

on the basis of another EOS suggested³⁸ for a low compression region. Equation (12) has been later parametrized by the following two temperature functions³⁹,

$$V(T, p_0) = A_0 + A_1(T - T_0) + A_2(T - T_0)^2$$
(13)

$$B_T(p_0) = B_{T_0}(p_0) \exp(-b_2(T - T_0))$$
(14)

where
$$A_0 = V_0(T_0) = \upsilon(p_0, T_0)$$
, $A_1 = \partial V(p_0, T)/\partial T|_{T=T_0}$, and $A_2 = (1/2)\partial^2 V(p_0, T)/\partial T^2|_{T=T_0}$, $b_2 = b_2(p_0) = -\partial \ln B_T(p_0, T)/\partial T|_{T=T_0}$.

 $-\partial \ln B_T(p_0,T)/\partial T|_{T=T_0}$. It should be stressed that Eqs (9) and (12) can be also derived 22,37,40 in the ways independent of the thermodynamic scaling idea. This possibility extends the applicability range of the EOSs to materials, the molecular dynamics of which does not obey the PDS law. One of the alternative derivation ways relies on the assumption that the appropriate isothermal bulk moduli linearly depend on pressure as follows

$$B_{act}(T, p) = B_{act}(T, p_0) + \gamma_{act}(p - p_0)$$
(15)

$$B_T(p) = B_T(p_0) + \gamma_{EOS}(p - p_0)$$
(16)

where $\gamma_{act} = (\partial B_{act}(p, T)/\partial p)_T = const$ and $\gamma_{EOS} = (\partial B_T(p)/\partial p)_T = const$. Moreover, it is worth noting that Eqs (9) and (12) imply Eqs (15) and (16), respectively.

In the case of the supercooled state of GF systems, the reference state (T_0, P_0) for the activation volumetric and volumetric EOSs is usually chosen at the glass transition temperature at ambient pressure. Nevertheless, both the EOSs can be used in other thermodynamic phases, including LC phases, if their activation volumetric and volumetric properties vary in the applicability ranges of the equations, which are defined by Eqs (10), (11), (15) and (13), (14), (16), respectively^{22,39,41-44}. By analogy with GF systems, the reference states (T_0, P_0) have been fixed for examined LC systems at appropriate LC phase transition temperatures at ambient pressure.

As verified in several papers^{22,28,36-40,45-49}, in case of GF systems that belong to different material groups such

As verified in several papers $^{22,28,36-40,45-49}$, in case of GF systems that belong to different material groups such as van der Waals liquids, ionic liquids, polymers, associated liquids, and the Kob-Andersen binary Lennard-Jones (KABLJ) liquid, which is a prototypical model of supercooled liquids, the volumetric and activation volumetric EOSs successfully describe the dependences V(T, P) and $V_{act}(T, P)$, respectively. Moreover, useful relationships have been formulated between some parameters of Eqs (9) and (12) on the additional assumption that the activation volume defined by Eq. (3) is well described in terms of the T-V Avramov model (Eq. (4)), which implies that V_{act} is expressed by Eq. (6) with Eq. (7). As a result, it has been found that the pressure derivative of V_{act} and the value of V_{act} at the reference pressure at a given temperature can be determined with no necessity of the activation volume calculations as follows

$$\gamma_{act} = \frac{\gamma_{EOS}}{\gamma_{EOS} - \gamma D} \tag{17}$$

$$B_{act}(T, p_0) = \frac{B_T(p_0)}{\gamma_{EOS} - \gamma D}$$

$$\tag{18}$$

It is worth noting that the above equations are model-dependent due to the parameter D of the T-V Avramov model. In Supplementary Information (Section S1), we have derived their counterparts, which are based on the T-V MYEGA model (Eq. (5)) and also dependent on the model parameter A_M ,

$$\gamma_{act} = \frac{\gamma_{EOS}}{\gamma_{EOS} - \gamma [1 + A_M \Gamma_0 (2 + A_M \Gamma_0)/(1 + A_M \Gamma_0)]}$$
(19)

$$B_{act}(T, p_0) = \frac{B_T(p_0)}{\gamma_{EOS} - \gamma [1 + A_M \Gamma_0 (2 + A_M \Gamma_0)/(1 + A_M \Gamma_0)]}$$
(20)

where $\Gamma_0 = T^{-1}V^{-\gamma}(T, p_0)$ is constant at T = const. Since one can expect similarly to γ_{EOS} that $\gamma_{act} = (\partial B_{act}(p, T) / \partial p)_T \cong const$ at different temperatures, the value of the parameter Γ_0 should remain unchanged for a given material at least to a good approximation.

Results and Discussion

We have selected to analyze the dielectric and pVT data for six LC substances that exhibit LC phases in relatively broad (p, T) ranges, such as 4-heptyl-4'-cyanobiphenyl (7CB), trans-4-heptyl(4-cyanophenyl)cyclohexane (7PCH), trans-4-n-octyl-(4-cyanophenyl)cyclohexane (8PCH), n-octyloxy-cyanobiphenyl (8OCB), n-hexyl-isothiocyanato-biphenyl (6BT), and n-octyl-isothiocyanato-biphenyl (8BT). All compounds have strong dipole moments directed along the symmetry axes, which are created by the CN or NCS group attached at one terminal position (at the other the alkyl or alkoxy group is attached). For this reason, the dielectric relaxation spectra yield solely the low frequency relaxation time connected with the flip-flop molecular motions in a given LC phase (N for 7CB, 7PCH, 8PCH and 8OCB, Cr E for 6BT and 8BT). It should be added that the compounds with the CN group show a tendency to forming antiparallel doublets in the LC phases⁵⁰ The experimental dielectric and pVT data examined by us herein have been earlier measured and reported for the nematic phase of 7CB^{51,52}, 7PCH^{53,54}, 8PCH^{55,56}, and 8OCB^{18,57} as well as the crystalline E phase of 6BT^{58,59} and 8BT^{60,61}.

From the inspection of the isothermal pressure dependences of the longitudinal dielectric relaxation times τ_{\parallel} of the LC systems subjected to compression, one can spot an intriguing behavior of the isothermal dependences τ_{\parallel} (p), which increase with increasing pressure as those established for the GF systems but reveal an untypical curvature that is negative in contrast to the positive one commonly observed for the GF systems. For comparison, in Fig. 1, we present the pressure dependences of the longitudinal relaxation times $\log_{10}\tau_{\parallel}$ (p) and the structural relaxation times $\log_{10}\tau$ (p), which have been determined from dielectric measurements on one of the investigated LC system (7CB) and a supercooled GF system (phenolphtalein dimethyl ether (PDE) that is a typical van der Waals liquid) in high pressure isothermal conditions as well as the corresponding activation volumes for the dielectric relaxation times. At first glance, one can see that the pressure isothermal dependences V_{act} (p) of PDE increase with increasing pressure, which is the typical behavior of GF systems 22 . However, the isothermal dependences V_{act} (p) of 7CB behave quite opposite, i.e., they decrease with increasing pressure.

We apply the formalism outlined in the previous section to analyze the activation volumetric properties of the selected LC systems, because the activation volume V_{act} defined by Eq. (3) and the isothermal bulk modulus for the activation volume, $B_{act}(p) = -(\partial \ln V_{act}/\partial p)_T^{-1}$, are the macroscopic quantities, which enable us to gain a better insight into the sensitivity of the molecular dynamics of the investigated LC systems to changes in pressure in isothermal conditions. In order to evaluate activation volumes of the investigated LC systems, we first follow the procedure based on the T-V Avramov model (Eq. (4)), which has been earlier successfully applied in ref. 22 to test the activation volumetric EOS (given herein by Eq. (9) with Eqs (11) and (12)) by using experimental data of different GF systems. The procedure relies on the density scaling law, which is obeyed by the longitudinal relaxation times τ_{\parallel} of many LC systems in different LC phases as already mentioned in Introduction. Since dielectric measurements are typically carried out in isobaric and isothermal conditions, we can determine temperature-pressure dependences $\tau_{\parallel}(T,p)$ from such experiments. To transform the dependences $\tau_{\parallel}(T,p)$ to their temperature-volume representation $\tau_{\parallel}(T,V)$, we exploit the experimental pVT data V(T,p), which are parametrized by means of the EOS given by Eq. (10). Then, we can describe the dependence $\tau_{\parallel}(T, V(T, p))$ by using the T-V Avramov model (Eq. (4)), the parameters of which enable us to determine the activation volumes from Eq. (6) with Eq. (7). Since the procedure for fitting the dependences $\tau_{\parallel}(T,p)$ to the T-V Avramov model (Eq. (4)) with V(T, p) taken from the EOS given by Eq. (12) with Eqs (13) and (14) has turned out to be very or even extremely time-consuming due to its very slow convergence at a typical required fit tolerance ($\Delta(\chi^2_{reduced}) < 10^{-9}$) for most of the investigated LC systems, we present details of such an analysis based on the Avramov model in Supplementary Information (Section S2), using 8BT in the crystalline E phase as an example.

However, to establish the activation volumes from Eq. (6) with Eq. (8), we need to find the values of the parameters of the MYEGA model by fitting the dependence $\tau_{\parallel}(T, V(T, p))$ to Eq. (5). Using also 8BT in the crystalline E phase as an example, we illustrate steps of the analysis in Fig. 2. The dielectric isotherms of 8BT (Fig. 2(a)) measured as a function $\tau_{\parallel}(T, p)$ have been transformed to its T-V representation (not shown herein) by using the volumetric EOS (Eq. (12) with Eqs (13) and (14)) fitted to the pVT experimental data for 8BT (Fig. 2(b)), making the assumption that the reference state (T_0, p_0) in the volumetric EOS is fixed at the crystal – Cr E transition temperature T_0 = 301.5 K at ambient pressure p_0 = 0.1 MPa. Then, we can directly describe the dependence $\tau_{\parallel}(T, p)$ as shown in Fig. 2(a) by using Eq. (5) with the specific volume V(T, p) expressed by Eq. (12) with Eqs (13) and (14), where the values of the EOS parameters are found by fitting pVT measurement data to this EOS (see Fig. 2(b) and Table 1). Then, we can successfully use the value of the scaling exponent, γ = 4.59 ± 0.03, evaluated by fitting to the MYEGA model (Eq. (5)) to perform the density scaling of the longitudinal relaxation times of 8BT (see Fig. 2(c)). Subsequently, the isothermal dependences V_{act} can be established for 8BT from Eq. (6) with Eq. (8) at temperatures and pressures (T, p) at which the dependences $\tau_{\parallel}(T, p)$ have been determined. As shown in Fig. 2(d), we have also fitted the isothermal dependences V_{act} based on the MYEGA model for 8BT to the activation volumetric EOS (Eq. (9) with Eqs (10) and (11)), taking the same reference state (T_0 , T_0 , T_0 , as that

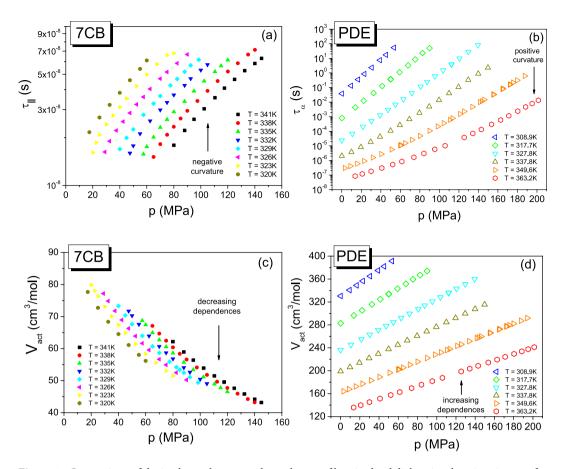


Figure 1. Comparison of the isothermal pressure dependences of longitudinal dielectric relaxation times τ_{\parallel} for a LC system 7CB (a) and structural relaxation times τ_{α} for a typical GL system PDE (b) as well as the isothermal pressure dependences of the corresponding activation volumes V_{act} for 7CB (c) and PDE (d). The values of V_{act} for PDE have been reported in ref. 22, whereas the values of V_{act} for 7CB have been calculated herein in the way suggested in ref. 22.

assumed in the previous volumetric analysis performed for this LC system in terms of the volumetric EOS. The same reference state has been used for 8BT in the Cr E phase in the analysis based on the Avramov model (see Supplementary Information). The values of the parameters of Eq. (5) are collected in Table 2 for all tested LC systems, whereas the values of the parameters of the activation volumetric EOS determined on the basis of Eq. (5) are listed in Table 3. However, the values of the parameters of Eq. (4) and the activation volumetric EOS determined on the basis of Eq. (4) are collected in Supplementary Information (in Tables S1 and S2, respectively).

From the comparative studies, it should be noted that the scaling exponents established by using MYEGA and Avramov models, respectively γ_M and γ , are the same for a given LC system studied in the Cr E phase and they are mutually in accord within their determination uncertainties for a given LC system in the nematic phase (see Table 2 in the main part and Table S1 in Supplementary Information), which lead to the density scaling of the longitudinal relaxation times of each tested LC system to a good approximation (see Fig. 2(c) for 8BT in the Cr E phase as an example). In addition, one can observe an increase in the value of the scaling exponent with increasing the molecular size of the LC systems, which suggests that the role of repulsive interactions in the molecular dynamics grows if the LC system molecular size is larger. After that general conclusion, we need to note that the MYEGA model has been found by us to be definitely faster convergent at the same fit tolerance as that assumed in case of the Avramov model. We have established that the MYEGA model characterized by the double exponential form also better reflects the negative curvature of the isothermal dependences $\tau_{||}(p)$ of the examined LC systems (see details of the analysis in Section S3 in Supplementary Information). Thus, we need to look deeper into what determines the activation volume in both the Avramov and MYEGA models.

The analysis presented in Fig. S3 in Supplementary Information suggests that the activation volumes evaluated for a given material from the Avramov and MYEGA model can be in general different, because the curvatures established from these models for the dependence $\log_{10} T_{\parallel}(\Gamma)$ can slightly differ for this material (as shown in case of the tested LC systems in the nematic phase). To verify this preliminary conjecture, it is convenient to examine the analytical general equation (Eq. (6)) derived for the activation volume in the GDS regime and its representations based on Eqs (4) and (5), which lead to the special cases of Eq. (6) that involve Eqs (7) and (8), respectively. Since the model-dependent and model-independent terms are well separated in the general equation (Eq. (6)), we can analyze both the contributions to the activation volume in case of the Avramov and MYEGA models. The model-independent term, $RT\gamma\Gamma/B_T(p)$, combines volumetric and effective short-range intermolecular potential

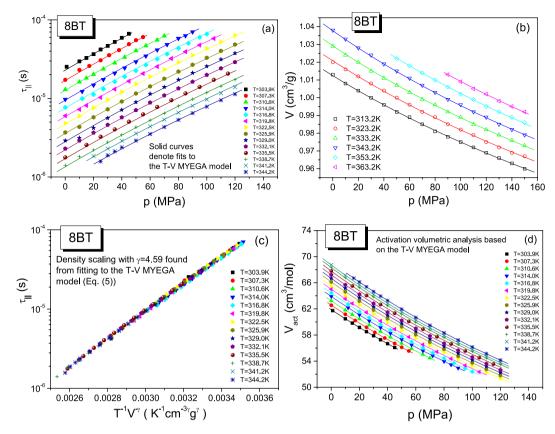


Figure 2. The analysis of the activation volumes of 8BT in terms of the T-V MYEGA model. (a) Pressure dependences of experimental longitudinal relaxation times along different isotherms and their fits to Eq. (5) (see Table 2 for the values of its parameters) with the specific volume expressed as the function V(T, p) by the volumetric EOS given by Eq. (12) with Eqs (13) and (14). (b) Pressure dependences of specific volumes measured along different isotherms and their fits to Eq. (10) (see Table 1 for the values of its parameters). (c) Plot of the master plot according to the power law density scaling law obeyed by longitudinal relaxation times. (d) Pressure dependences of the activation volumes evaluated from Eq. (6) with Eq. (8) and their fits to to the activation volumetric EOS given by Eq. (9) with Eqs (10) and (11) (see Table 3 for the values of its parameters).

LC system (phase)	T ₀ (K)	A ₀ (cm ³ /g)	A ₁ /10 ⁻⁴ (cm ³ /g/K)	$A_2/10^{-6}$ (cm ³ /g/K ²)	B(T ₀ , p ₀) (MPa)	$_2/10^{-3} \ (\mathrm{K}^{-1})$	Yeos	AdjR ² coeff.
6BT (Cr E)	250.75	0.9038 ± 0.0001	3.62 ± 0.03	0.58 ± 0.03	2983 ± 10	2.62 ± 0.06	3.97 ± 0.08	0.9986
8BT (Cr E)	301.5	1.0027 ± 0.0003	8.31 ± 0.11	0.22 ± 0.02	2384 ± 22	3.74 ± 0.23	7.82 ± 0.27	0.9997
7CB (N)	302.0	0.9983 ± 0.0005	9.57 ± 0.43	10.2 ± 2.2	1704 ± 60	18.5 ± 2.9	13.99 ± 0.64	0.9992
7PCH (N)	303.3	1.0493 ± 0.0002	9.23 ± 0.12	3.21 ± 0.35	1755 ± 16	9.71 ± 0.45	10.50 ± 0.16	0.9999
8PCH (N)	309.13	1.0609 ± 0.0007	10.5 ± 0.5	4.16 ± 0.03	1592 ± 65	12.2 ± 2.7	12.03 ± 0.74	0.9986
8OCB (N)	342.0	1.0052 ± 0.0001	9.28 ± 0.07	2.33 ± 0.14	1731±6	9.76 ± 0.28	11.37 ± 0.10	0.9996

Table 1. Values of the fitting parameters of the volumetric EOS found by fitting the measured specific volumes to Eq. (12) with Eqs (13) and (14). The reference state is fixed at the phase transition temperature T_0 at ambient pressure $p_0 = 0.1$ MPa.

properties, involving the isothermal bulk modulus for the specific volume $B_T(p)$ and the scaling exponent γ , which is a material constant in the PDS regime that enables us to describe the molecular dynamics in terms of the scaling quantity $\Gamma = T^{-1}V^{-\gamma}$. As already mentioned, the values of scaling exponents obtained in the Avramov and MYEGA models for each material are almost the same. Thus, the activation volumes evaluated using these models would have been also nearly the same for a given material if the model-dependent term had quantitatively contributed to Eq. (6) on the same level in case of both the models for the material. From Eqs (7) and (8), one can see that the model-dependent contributions to Eq. (6) are some functions of the scaling variable Γ , which are parametrized by the parameters A and D as well as A_M and D_M of the Avramov and MYEGA models, respectively. It confirms that the activation volume values evaluated for a given material by using these models in the PDS regime (Eqs (4) and (5)) in general depend not only on the density scaling exponent and volumetric properties for this material.

LC system (phase)	τ ₀ (s)	$A_{M} \ (Kcm^{3\gamma}/g^{\gamma})$	$D_{\rm M}/10^4$ (Kcm ³ $^{\gamma}/{\rm g}^{\gamma}$)	γ_{M}	AdjR ² coeff.
6BT (Cr E)	$(3.25 \pm 0.48) \times 10^{-15}$	-55 ± 4	0.78 ± 0.03	2.74 ± 0.03	0.9966
8BT (Cr E)	$(2.60 \pm 0.22) \times 10^{-12}$	-66 ± 3	0.61 ± 0.06	4.59 ± 0.03	0.9990
7CB (N)	$(6.49 \pm 0.40) \times 10^{-25}$	-209 ± 16	2.35 ± 0.45	3.42 ± 0.05	0.9941
7PCH (N)	$(3.59 \pm 0.41) \times 10^{-19}$	-204 ± 8	1.70 ± 0.10	3.96 ± 0.02	0.9968
8PCH (N)	$(9.92 \pm 0.11) \times 10^{-30}$	-294 ± 5	4.15 ± 0.24	3.53 ± 0.02	0.9961
8OCB (N)	$(2.29 \pm 0.27) \times 10^{-19}$	-196 ± 6	1.53 ± 0.09	4.13 ± 0.03	0.9983

Table 2. Values of the fitting parameters of the MYEGA model (Eq. (5)) established for longitudinal relaxation times of the examined LC systems.

LC system (phase)	F ₀ (cm ³ /mol)	F ₁ (cm ³ /mol/K)	F ₂ /10 ⁻⁴ (cm ³ /mol/K ²)	$\begin{array}{c} B_{act}(T_0, p_0) \\ (MPa) \end{array}$	$g_2/10^{-3} \ (K^{-1})$	Yact	γ_{act} from Eq. (19)
6BT (Cr E)	42.97 ± 0.11	0.1871 ± 0.0003	0.00 ± 0.15	998.9 ± 5.6	0.214 ± 0.067	0.981 ± 0.023	0.982
8BT (Cr E)	61.38 ± 0.01	0.2090 ± 0.0002	-6.11 ± 0.05	432.92 ± 0.21	0.409 ± 0.014	1.109 ± 0.004	1.105
7CB (N)	61.27 ± 0.50	2.138 ± 0.056	26.1 ± 1.8	73.69 ± 0.36	4.42 ± 0.38	0.518 ± 0.007	0.512
7PCH (N)	66.59 ± 0.06	1.315 ± 0.007	35.7 ± 2.2	121.22 ± 0.25	1.89 ± 0.13	0.59 ± 0.01	0.59
8PCH (N)	66.41 ± 0.18	2.386 ± 0.017	24.1 ± 3.1	67.72 ± 0.20	0	0.363 ± 0.003	0.367
8OCB (N)	74.98 ± 0.07	1.392 ± 0.007	21.5 ± 1.2	113.75 ± 0.17	0	0.459 ± 0.002	0.459

Table 3. Values of the fitting parameters of the activation volumetric EOS by fitting the activation volumes evaluated using the MYEGA model to Eq. (9) with Eqs (10) and (11). The reference state is fixed at the phase transition temperature T_0 (shown in Table 1) at ambient pressure $p_0 = 0.1$ MPa. The fitted value of γ_{act} is compared with that calculated from Eq. (19) for each material.

We have evaluated and compared the activation volumes evaluated by using the T-V MYEGA and Avramov models for each examined LC system. As an example, we have presented the plots of the dependences V_{act}(p) for 8BT in Fig. 2(d) in the main part and S1(d) in Supplementary Information, respectively. From these figures and other such data not shown herein, we can claim that both the models yield slightly different values of the activation volume despite almost the same values of the scaling exponent $\gamma_{\rm M}$ and γ for a given material. The differences are related to the model-dependent contributions (Eqs (8) and (7)) to the activation volume defined in the GDS regime by Eq. (6). Thus, it is worth discussing the other parameters of Eqs (8) and (7) than the scaling exponent, i.e., the parameters A_M and D_M as well as A and D, respectively. First of all, it should be stressed that the fitted values of the parameters found for the investigated LC systems (see Table 2 for the MYEGA model parameters A_M and D_M and Table S1 in Supplementary Information for the Avramov model parameters A and D) show significant differences in comparison with those earlier widely reported^{22,23,26,27,62,63} for many GF systems. Striking differences have been established in case of the parameters A and D of the Avramov model. The values of the former are extremely large, especially for the LC systems investigated in the nematic phase, whereas the values of the parameter D ranges between 0 and 1, which have never been observed yet in other material groups and requires considering its physical meaning for the LC materials. Since the parameter D of the Avramov model has been interpreted to be inversely proportional to the number of available pathways for local motions of a molecule or polymer segment^{26,29} (as also mentioned in Section S4 in Supplementary Information), we need to doubt the possibility that there are sufficiently many such pathways in the LC systems to cause the value of the parameter D < 1. On the other hand, the values of the parameter D_M of the MYEGA model are at most an order of magnitude greater than those reported for the GF systems, whereas the values of the parameter A_M are surprisingly negative for each tested LC system.

Taking into account the fundamentals of the MYEGA model^{32,23} (which considers the network constraints using a two-state pattern according to that they are either intact or broken with the energy difference represented by the parameter A_M), we can suggest that the negative values of this parameter obtained for the investigated LC systems can be related to forming antiparallel doublets in these materials as already mentioned. The doublet formation energy can be considered similarly to the dimerization energy that is usually negative as a difference between the energy of a dimer, which is an energetically favorable state, and the sum of the energy of two monomers. Similarly to the interaction between molecules forming dimers⁶⁴, the negative energy of the intermolecular interaction causing two molecules to form an antiparallel doublet is expected to be accompanied by the negative entropy of this interaction. The T-V MYEGA model is capable to satisfactorily describe such a case, because the configurational entropy originally considered in this model for GF systems can be adopted to represent the negative entropy of the interaction between two LC molecules forming the antiparallel doublet. This is because the entropy of such interactions can be reinterpreted within the T-V MYEGA model for LC systems in terms of Eq. (S21) earlier postulated²³ for GF liquids in the way outlined in Section S4 in Supplementary Information. In the MYEGA model, the number of floppy modes has been considered as the number of broken constraints^{65,66}. If we assume that the intact constraints result in forming the antiparallel doublets, the breakage of such a constraint leading to two unconstrained states (molecules) requires providing a sufficient energy to break the interaction between the molecules forming the antiparallel doublet. Thus, the floppy mode considered as a broken

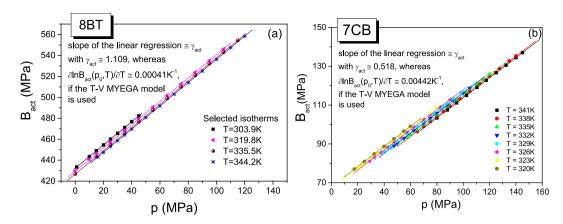


Figure 3. Plot of the pressure dependences of the isothermal bulk moduli for the activation volumes evaluated in terms of the T-V MYEGA model for 8BT in the Cr E phase and 7CB in the nematic phase (panels (**a**) and (**b**), respectively). The solid lines denote the linear dependences that are in accord with Eq. (11), which complies with the activation volumetric EOS given by Eq. (9) with Eqs (10) and (11) (see Table 3 for the values of its parameters).

antiparallel doublet involves two unconstrained states (molecules) characterized by the same energy of the breakage of the antiparallel doublet. This simple picture shows that the degeneracy of the floppy mode can be quantified by $\Omega = 1/2$ in terms of the unconstrained state, and then $\ln \Omega$ is negative, similarly as that evaluated in some cases of recently reported predictions based on the MYEGA model for supercooled borate and silicate liquids⁶⁶. Thus, in case of examined LC systems, the preexponential factor dependent on $\ln\Omega$ is also negative in Eqs (S21) and (S22) presented in Section S4 in Supplementary Information. Then, Eq. (S22) results in an increase in the interaction entropy (in the domain of negative values) related to forming the antiparallel doublets by molecules of the examined LC systems due to an increase in the interaction energy H_M with increasing density in isothermal conditions, which is predicted by Eq. (S18) earlier postulated²³ for GF liquids as mentioned in Section S4 in Supplementary Information. However, in case of the tested LC systems, the latter behavior is expected in the domain of negative values following the negative value of the fitting parameter A_M of the T-V MYEGA model (Eq. (4)), because the parameter A_M involves the reference value of the interaction energy $H_M(V_r)$ according to Eq. (\$20) shown in Section \$4 in Supplementary Information. In this way, the interaction entropy increases with increasing the interaction energy, which occurs in the domains of negative values for both the quantities, similarly as often reported on the dependence of entropy on energy for interactions leading to a dimer formation⁶⁷. It is worth noting that the negative values of the interaction entropy are compensated by the negative value of the parameter B_M in the Adam-Gibbs equation³³ underlying the MYEGA model^{23,32}, $\log_{10} \tau_{||} = \log_{10} \tau_{0_M} + B_M T^{-1} S_c^{-1}$. As discussed in Section S4 in Supplementary Information, the parameter B_M depends on the critical configurational entropy s_*^* . In case of the examined LC systems, the value of the critical entropy s_*^* is expected to be negative, because it can be interpreted as the limiting value of the interaction entropy related to the minimal value of the interaction energy, which is required to form antiparallel doublets. The reinterpreted fundamentals of the MYEGA model rationalize our hypothesis that the negative value of the parameter A_M of the MYEGA model as well as the negative curvature of the dependences $\tau_{\parallel}(p)$ can reflect the formation of antiparallel doublets in the examined LC systems.

To gain a better insight into the activation volumetric properties of the LC systems, we have described the pressure dependences of the activation volumes by the activation volumetric EOS. The dependences V_{act}(p) have been very successfully fitted to the activation volumetric EOS independent of the model (MYEGA or Avramov) used to evaluate the values of V_{act} (see Fig. 2(d) in the main part and S1(d) in Supplementary Information as examples). Nevertheless, the differences in the values of the activation volumes determined from the MYEGA and Avramov models have been also reflected in the values of the activation volumetric EOS parameters (see Table 3 in the main part and Table S2 in Supplementary Information). It should be noted that the differences in the activation volumes considerably affect the isothermal bulk moduli for the activation volumes (see Fig. 3 in the main part and Fig. S2 in Supplementary Information as examples), and consequently the different activation volumetric properties can be better demonstrated by using the dependences B_{act}(p) than V_{act}(p). We first need to stress that the values of B_{act} are positive. Such an activation behavior results from the decreasing pressure dependence of the activation volume of the tested LC systems, which is quite opposite to the activation volumetric behavior of the GF systems (for which the dependences Vact(p) increase and the dependences Bact(p) decreases with increasing pressure, and the values of B_{act}(p) are negative). Thus, the activation volumetric behavior is qualitatively in accord with the typical pressure behavior of the specific volume V and the isothermal bulk modulus B_T for the volume. Moreover, all the dependences B_{act}(p) increase with increasing pressure in the linear manner according to Eq. (11), which complies with Eq. (9) that very well describes the dependences $V_{act}(p)$ determined from both the MYEGA and Avramov models. Nevertheless, the values of the exponent γ_{act} (which is also the slope of the linear dependence Bact(p)) are less in case of the activation volumes determined from the MYEGA model, and they can be even over 2 times smaller (see Table 3 in the main part and Table S2 in Supplementary Information) than those established for the activation volumes evaluated from the Avramov model. The different slopes of the linear isothermal dependences $B_{act}(p)$ can be seen for 8BT and 7CB as examples in Fig. 3 in the main part and Fig. S2 in Supplementary Information, depending on the model used to evaluate the activation volumes. However, the most striking observation can be made from Fig. 3. This is an extremely low sensitivity of the activation volumetric prosperities to changes in temperature at a given pressure in case of the activation volumes determined from the MYEGA model. For this reason, in Fig. 3(a), only some chosen dependences $B_{act}(p)$ have been shown in order to avoid making this panel of Fig. 3 obscure. This very weak temperature effect on the isothermal bulk modulus for the activation volume at a given pressure is reflected in small values of the parameter $g_2 = -\partial \ln B_{act}(p_0, T)/\partial T\Big|_{T=T_0}$ of Eq. (11), which can even reduce to 0 as established for 8PCH and 8OCB in their nematic phase. One can expect that this surprising activation volumetric behavior is also related to some networking of LC systems via the formation of antiparallel doublets.

Finally, we have tested the relations given by Eqs (17) and (19) by using the experimental data for all the studied LC systems and the results of their analyses by employing the T-V Avramov and MYEGA models (Eqs (4) and (5)) as well as the activation volumetric and volumetric EOSs (Eqs (9-11) and (12-14)). We have established a very good agreement between the calculated values of γ_{act} from Eq. (17) and those found by fitting the isothermal dependences $V_{act}(p)$ to the activation volumetric EOS if the values of $V_{act}(p)$ have been evaluated from the Avramov model (Eq. (6) with Eq. (7)). Since the relations given by Eqs (17) and (19) are model-dependent, it has not been able to apply Eq. (17) to satisfactorily predict values of the exponent γ_{act} obtained from fitting the dependences $V_{act}(p)$ to the activation volumetric EOS if the values of $V_{act}(p)$ have been determined from the MYEGA model. Therefore, we have employed Eq. (19) in this case. As already mentioned in the previous section, Eq. (19) is an approximate equation and requires remaining a value of the scaling quantity Γ_0 nearly unchanged for a given material. As a consequence, we have needed to find an appropriate value of Γ_0 for each LC material prior to calculating its exponent γ_{act} from Eq. (15). To successfully predict the value of γ_{act} from Eq. (19), which are in accord with that obtained by fitting the dependences Vact(p) determined from the MYEGA model to the activation volumetric EOS, we have used the values of Γ_0 close to the upper experimental limit of the scaling quantity Γ (the experimental ranges of Γ are shown in Fig. S3 in Supplementary Information) in case of the LC systems investigated in the nematic phase, and the values of Γ_0 about 50% larger than the upper experimental limits of the scaling quantity Γ for 6BT and 8BT in the Cr E phase. It should be noted that the T-V MYEGA model has been applied to evaluate the activation volumes for the first time to our best knowledge, and Eq. (19) is the first attempt made to interrelate the exponents γ , γ_{EOS} , and γ_{act} , which requires further studies to enhance its prediction capability.

Summary and Conclusions

In this paper, we have thoroughly investigated the activation volumetric properties of the selected LC systems in different phases (N and Cr E), which obey the power law density scaling law for the longitudinal dielectric relaxation times. Inspired by the negative curvatures of the pressure dependences of the isothermal longitudinal relaxation times $\tau_{\parallel}(p)$, which is quite opposite to those typically obtained for the structural relaxation times of many GF systems, we have compared the activation volumetric properties of LC and GF systems in the PDS regime. For this purpose, we have applied the formalism based on the temperature-volume versions of the Avramov and MYEGA models (Eqs (4) and (5)), which has been also developed by us herein within the framework of the MYEGA model. Based on the general equation (Eq. (6)) for the activation volume in the general density scaling regime, we have been able to examine the model-independent and model-dependent contributions to V_{act} . We have also exploited the activation volumetric and volumetric equations of state (Eqs (9–11) and (12–14)) recently derived for the viscous liquids and well interpreted in the PDS regime. The volumetric EOS has enabled us to transform between the sets of thermodynamic variables (T, p) and (T, V), and the activation volumetric EOS has considerably facilitated the analysis of the activation volumetric properties of the tested LC systems.

We have established that the negative curvature of the dependences $\tau_{\parallel}(p)$ for all the examined LC systems results in the unusual activation volumetric properties that are quite opposite to those widely reported for many GF systems. Among the activation properties of the LC systems, the special attention should be paid to a few of them: (i) the decrease in the activation volume with increasing pressure in isothermal conditions, (ii) the increase in the isothermal bulk modulus for the activation volume with increasing pressure, B_{act}(p), which satisfies the linear dependence (Eq. (15)) that complies with the activation volumetric EOS (Eq. (9)) well describing the dependences $V_{act}(p)$ in the tested experimental range, (iii) the positive values of B_{act} , which are a consequence of the increasing dependence Vac(p). All the activation volumetric properties are qualitatively in accord with those typically reported for the specific volume and the isothermal bulk modulus for the specific volume of the LC and GF systems, whereas it cannot be said about the typical activation volumetric properties of GF systems. The analysis based on the activation volumetric EOS has additionally revealed a weak sensitivity of the activation properties (for instance Bact) of the examined LC systems to changes in temperature. The comparative studies of the activation volume for the selected LC systems in terms of the T-V Avramov and MYEGA models lead to the conclusion that the MYEGA model is able to more efficiently describe the dependences $\tau_{\parallel}(p)$ characterized by the negative curvature and the fitted values of its parameters are more reasonable from the viewpoint of the physical meaning of the model parameters. Within the framework of the MYEGA model, we have achieved a very interesting result, which is the unexpectedly negative value of the parameter A_M in Eq. (5) for each tested LC system. The energetic interpretation of this MYEGA model parameter (based on the two-state model of constraints that can be either intact or broken) allows us to suggest that the negative value of the parameter A_M is related to the energy barrier for the formation of antiparallel doublets, which most likely occurs in the case of examined LC systems. This finding can be also regarded as an indirect explanation for the negative curvature of the pressure dependences of isothermal longitudinal relaxation times observed for the LC systems as well as its consequence that is the decrease in activation volumes with increasing pressure. Thus, the MYEGA model seems to be a useful tool to explore properties of liquid crystals, which is worthy of confirmation in further investigations of other LC systems. The observed advantage of the MYEGA model over the Avramov model in analyzing relaxation dynamics of selected LC systems is an important outcome of our study, which is a relevant clue in search of the most universal description of the thermodynamic evolution of the timescale of molecular dynamics in different materials.

It should be emphasized that the pressure effect on relaxation dynamics, which is observed in case of LC systems, has an extraordinary character that enhances our understanding of relaxation behaviors, which are possible in complex materials subjected to various thermodynamic conditions. One can expect that representatives of other material groups may also behave at elevated pressure similarly to the examined LC systems, if their molecules form some special supramolecular structures. However, this suggestion should be verified in further studying.

References

- de Jeu, W. H., Ostrovski, B. I. & Shalaginov, A. N. Structure and fluctuations of smectic membranes. Phys. Mod. Phys. 75, 181–235 (2003).
- 2. Urban, S. & Würflinger, A. Dielectric properties of liquid crystals under high pressure in *Advances in Chemical Physics* (eds. Prigogine, I. & Rice, S. A.), vol. 98, pp. 143–216, (John Wiley & Sons, New York, 1997).
- 3. Urban, S. High Pressure Investigations of Liquid Crystals in *Handbook of Liquid Crystals* (eds. Goodby, J. W. et al.), vol. 2, ch. 15, pp. 443–486 (Wiley-VCH, Weinheim 2014).
- 4. Urban S. & Würflinger, A. Molecular rotations in liquid crystals as seen by the dielectric spectroscopy method In *Relaxation Phenomena* (eds. Haase, W. & Wróbel, S.), ch. 4.2, pp. 181–204 (Springer, 2003).
- 5. Urban S. & Würflinger, A. Influence of pressure on the dielectric properties of liquid crystals In *Nonlinear Dielectric Phenomena in Complex Liquids* (eds. Rzoska, S. J. & Zhelezny, V. P.), pp. 211–220 (Kluwer Academic Publishers, 2004).
- Urban A. & Roland, C. M. Low frequency relaxation in liquid crystals in relations to structural relaxation in glass-formers. J. Non-Cryst. Solids 357, 740–745 (2011).
- 7. Williams, G. Complex Dielectric Constant of Dipolar Compounds as a Function of Temperature, Pressure and Frequency. *Trans. Faraday Soc.* **60**, 1548–1555 (1964).
- 8. Eyring, H. The Activated Complex in Chemical Reactions. J. Chem. Phys. 3, 107-115 (1935).
- 9. Guggenheim, E. A. Thermodynamics of an Activated Complex. Trans. Faraday Soc. 33, 607-614 (1937).
- Koperwas, K., Grzybowski, A., Tripathy, S. N., Masiewicz, E. & Paluch, M. Thermodynamic consequences of the kinetic nature of the glass transition. Sci. Rep. 5, 17782-1–17782-8 (2015).
- 11. Debye, P. Polar Molecules, p. 84 (Chemical Catalog Co., New York 1929, reprinted by Dover, New York).
- 12. Powles, J. G. Dielectric Relaxation and the Internal Field. J. Chem. Phys. 21, 633-637 (1953).
- 13. Floudas, G., Paluch, M., Grzybowski, A. & Ngai, K. The Glass "Transition" in *Molecular Dynamics of Glass-Forming Systems: Effects of Pressure*. (Series: Advances in Dielectrics, Series ed. Friedrich Kremer), ch. 1, pp. 1–37 (Springer-Verlag, Berlin Heidelberg 2011).
- 14. Urban S. & Würflinger, A. Thermodynamical scaling of the low frequency relaxation time in liquid crystals. *Phys. Rev. E* 72, 021707-1-021707-4 (2005).
- 15. Fragiadakis, D. M. et al. The Phase Diagram and Dynamics of 5*CB. J. Phys. Chem. B 115, 6437-6444 (2011).
- 16. Urban, S. Thermodynamic scaling of the viscosity and the longitudinal relaxation time for three nematic PCHs. *Liq. Cryst.* **38**, 1147–1152 (2011).
- 17. Roland, C. M. et al. Volumetric, dielectric, calorimetric and X-ray studies of smectogenic 10PBO8 at atmospheric and elevated pressures. Liq. Cryst. 39, 993–1001 (2012).
- 18. Roland, C. M., Fragiadakis, D., Roy, D. & Urban, S. Volumetric study of n-octyloxy-cyanobiphenyl (8OCB). *Liq. Cryst.* 41, 9-14 (2014).
- 19. Roy, D. et al. Phase Behavior and Dynamics of a Cholesteric Liquid Crystal. J. Chem. Phys. 140, 074502-1-074502-6 (2014).
- 20. Angell, C. A. & Klein, I. S. Glass physics: Prigogine and Defay say relax. Nature Phys. 7, 750-751 (2011).
- 21. Masiewicz, E. *et al.* Adam-Gibbs model in the density scaling regime and its implications for the configurational entropy scaling. *Sci. Rep.* **5**, 13998-1–13998-13 (2015).
- 22. Grzybowski, A., Koperwas, K., Swiety-Pospiech, A., Grzybowska, K. & Paluch, M. Activation volume in the density scaling regime: Equation of state and its test by using experimental and simulation data. *Phys. Rev. B* 87, 054105-1–054105-15 (2013).
- 23. Masiewicz, E., Grzybowski, A., Sokolov, A. P. & Paluch, M. Temperature-Volume Entropic Model for Viscosities and Structural Relaxation Times of Glass Formers. *J. Phys. Chem. Lett.* **3**, 2643–2648 (2012).
- Casalini, R. & Roland, C. M. Scaling of the supercooled dynamics and its relation to the pressure dependences of the dynamic crossover and the fragility of glass formers. *Phys. Rev. B* 71, 014210-1-014210-10 (2005).
- 25. Alba-Simionesco, C. & Tarjus, G. Temperature versus density effects in glassforming liquids and polymers: A scaling hypothesis and its consequences. *J. Non-Cryst. Solids* **352**, 4888–4894 (2006).
- Casalini, R., Mohanty, U. & Roland, C. M. Thermodynamic interpretation of the scaling of the dynamics of supercooled liquids. J. Chem. Phys. 125, 014505-1-014505-9 (2006).
- 27. Grzybowski, A., Paluch, M., Grzybowska, K. & Haracz, S. Communication: Relationships between Intermolecular potential, thermodynamics, and dynamic scaling in viscous systems. *J. Chem. Phys.* 133, 161101-1–161101-4 (2010).
- 28. Floudas, G., Paluch, M., Grzybowski, A. & Ngai, K. Origin of Glass Formation In *Molecular Dynamics of Glass-Forming Systems: Effects of Pressure.* (Series: Advances in Dielectrics, Series ed. Friedrich Kremer), ch. 2, pp. 39–65 (Springer-Verlag, Berlin Heidelberg 2011).
- 29. Avramov, I. & Milchev, A. Effect of Disorder on Diffusion and Viscosity in Condensed Systems. *J. Non-Cryst. Solids* **104**, 253–260 (1988)
- 30. Avramov, I. Pressure dependence of viscosity of glassforming melts. J. Non-Cryst. Solids 262, 258-263 (2000).
- 31. Grzybowska, K., Grzybowski, A., Pawlus, S., Pionteck, J. & Paluch, M. Role of entropy in the thermodynamic evolution of the time scale of molecular dynamics near the glass transition. *Phys. Rev. E* 91, 062305-1–062305-13 (2015).
- 32. Mauro, J. C., Yue, Y. Z., Ellison, A. J., Gupta, P. K. & Allan, D. C. Viscosity of glass-forming liquids. Proc. Natl. Acad. Sci. USA. 106, 19780–19784 (2009).
- 33. Adam, G. & Gibbs, J. H. On the Temperature Dependence of Cooperative Relaxation Properties in Glass-Forming Liquids. *J. Chem. Phys.* 43, 139–146 (1965).
- 34. Bøhling, L. et al. Scaling of viscous dynamics in simple liquids: theory, simulation and experiment. New J. Phys. 14, 113035-1-113035-12 (2012).
- Ingebrigtsen, T. S., Bøhling, L., Schrøder, T. B. & Dyre, J. C. Communication: Thermodynamics of condensed matter with strong pressure-energy correlations. J. Chem. Phys. 136, 061102-1–061102-4 (2012).
- Grzybowski, A., Paluch, M. & Grzybowska, K. Consequences of an Equation of State in the Thermodynamic Scaling Regime. J. Phys. Chem. B 113, 7419–7422 (2009).

- 37. Grzybowski, A., Haracz, S., Paluch, M. & Grzybowska, K. Density Scaling of Supercooled Simple Liquids Near the Glass Transition. *J. Phys. Chem. B* 114, 11544–11551 (2010).
- 38. Bardik, V. Yu. & Shakun, K. S. Investigations of the steepness of a repulsive potential in accordance with the equation of state and light-scattering spectra. *Ukr. J. Phys.* **50**, 404–406 (2005).
- 39. Grzybowski, A., Grzybowska, K., Paluch, M., Swiety, A. & Koperwas, K. Density scaling in viscous systems near the glass transition. *Phys. Rev. E* 83, 041505-1–041505-7 (2011).
- 40. Grzybowski, A., Paluch, M. & Grzybowska, K. Phys. Rev. E 82, 013501 (2010).
- 41. López, E. R. et al. Density scaling of the transport properties of molecular and ionic liquids. J. Chem. Phys. 134, 144507-1-144507-11 (2011).
- 42. Grzybowski, A., Koperwas, K. & Paluch, M. Equation of state in the generalized density scaling regime studied from ambient to ultra-high pressure conditions. *J. Chem. Phys.* **140**, 044502-1–044502-12 (2014).
- 43. Chorążewski, M., Grzybowski, A. & Paluch, M. The complex, non-monotonic thermal response of the volumetric space of simple liquids. *Phys. Chem. Chem. Phys.* **16**, 19900–19908 (2014).
- Chorażewski, M., Grzybowski, A. & Paluch, M. Isobaric Thermal Expansion of Compressed 1,4-Dichlorobutane and 1-Bromo-4chlorobutane: Transitiometric Results and a Novel Application of the General Density Scaling-Based Equation of State. *Ind. Eng. Chem. Res.* 54, 6400–6407 (2015).
- 45. Grzybowski, A., Koperwas, K. & Paluch, M. Scaling of volumetric data in model systems based on the Lennard-Jones potential. *Phys. Rev. E* 86, 031501-1-031501-9 (2012).
- 46. Paluch, M. et al. A Relationship between Intermolecular Potential, Thermodynamics, and Dynamic Scaling for a Supercooled Ionic Liquid. J. Phys. Chem. Lett. 1, 987–992 (2010).
- 47. Wojnarowska, Z. et al. On the scaling behavior of electric conductivity in [C₄mim][NTf₂]. Phys. Chem. Chem. Phys. 16, 20444–20450 (2014).
- Adrjanowicz, K., Grzybowski, A., Grzybowska, K., Pionteck, J. & Paluch, M. Effect of High Pressure on Crystallization Kinetics of van der Waals Liquid: An Experimental and Theoretical Study. Cryst. Growth Des. 14, 2097–2104 (2014).
- Koperwas, K., Grzybowski, A., Grzybowska, K., Wojnarowska, Z. & Paluch, M. In search of correlations between the four-point measure of dynamic heterogeneity and other characteristics of glass-forming liquids under high pressure. J. Non-Cryst. Solids 407, 196–205 (2015).
- 50. Leadbetter, A. J., Richardson, R. M. & Cooling, C. N. The Structure of a Number of Nematogens. J. de Phys. 36, C1-37 (1975).
- 51. Brückert, T., Würflinger, A. & Urban, S. Dielectric Studies on Liquid Crystals under High Pressure. Ber. Bunsenges. Phys. Chem. 97, 1209–1213 (1993).
- 52. Sandmann, M. & Würflinger, A. PVT Measurements on 4'-n-Hexyl-Biphenyl-4-Carbonitrile (6CB) and 4-n-Heptyl-Biphenyl-4-Carbonitrile (7CB) up to 300 MPa. Z. Naturforsch. 53a, 233–238 (1998).
- Brückert, T., Büsing, D., Würflinger, A. & Urban, S. Differential Thermal Analysis (DTA) and Dielectric Studies of 4-(trans-4-Heptyl-Cyclohexyl)-benzonitrile (7PCH) under High Pressure. Z. Naturforsch. 50a, 977–983 (1995).
- 54. Kuss, E.pVT-Data and Tait-Equation for 4-trans-(4-alkyl)-Cyclohexyl-Benzonitriles. Mol. Cryst. Liq. Cryst. 76, 199-210 (1981).
- 55. Urban, S., Büsing, D., Würflinger, A. & Gestblom, B. Dielectric studies of 8PCH at ambient and high pressure. *Liq. Cryst.* 25, 253–261 (1998).
- 56. Urban, S. & Würflinger, A. Dielectric anisotropy in the nematic phase of 8PCH as a function of temperature and pressure. *Liq. Cryst.* **27**, 1119–1122 (2000).
- Brückert, T., Urban, S. & Würflinger, A. Dielectric studies of octyloxy-cyanobiphenyl (8OCB) under high pressure. Ber. Bunsenges. Phys. Chem. 100, 1133–1137 (1996).
- 58. Urban, S. & Würflinger, A. Phase Diagram and Dielectric Relaxation Studies of *n*-Hexyl-isothiocyanato-biphenyl (6BT) in the Smectic E Phase under High Pressure. *Z. Naturforsch.* **57a**, 233–236 (2002).
- 59. Roland, C. M. *et al.* Thermodynamic scaling and the characteristic relaxation time at the phase transition of liquid crystals. *J. Chem. Phys.* **128**, 224506-1–224506-9 (2008).
- 60. Urban, S., Würflinger, A. & Kocot, A. Phase diagram and dielectric relaxation studies of n-octyl-isothiocyanato-biphenyl (8BT) in the crystalline E phase under high pressure. *Liq. Cryst.* **28**, 1331–1336 (2001).
- 61. Würflinger, A. & Urban, S. PVT measurements on n-octyl-isothiocyanato-biphenyl (8BT) at elevated pressures. *Liq. Cryst.* **29**, 799–804 (2002).
- 62. Casalini, R. & Roland, C. M. An equation for the description of volume and temperature dependences of the dynamics of supercooled liquids and polymer melts. J. Non-Cryst. Solids 353, 3936–3939 (2007).
- Lunkenheimer, P., Kastner, S., Köhler, M. & Loidl, A. Temperature development of glassy α-relaxation dynamics determined by broadband dielectric spectroscopy. *Phys. Rev. E* 81, 051504-1-051504-6 (2010).
- 64. da Costa, L. M. et al. Computational Study of the Effect of Dispersion Interactions on the Thermochemistry of Aggregation of Fused Polycyclic Aromatic Hydrocarbons as Model Asphaltene Compounds in Solution. J. Phys. Chem. A 118, 896–908 (2014).
- Gupta, P. K. & Mauro, J. C. Composition dependence of glass transition temperature and fragility. I. A topological model incorporating temperature-dependent constraints. J. Chem. Phys. 130, 094503-1-094503-8 (2009).
- 66. Rodrigues B. P. & Wondraczek L. Floppy mode degeneracy and decoupling of constraint predictions in super-cooled borate and silicate liquids. *Front. Mater.* 1, 32-1–32-7, doi: 10.3389/fmats.2014.00032 (2015).
- 67. Viñals, J. Andrzej Kolinski, A. & Skolnick J. Numerical Study of the Entropy Loss of Dimerization and the Folding Thermodynamics of the GCN4 Leucine Zipper. *Biophys. J.* 83, 2801–2811 (2002).

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Author Contributions

A.G. and S.U. wrote the manuscript. S.U. provided most experimental data for LC systems. M.P. provided experimental data for the representative of GF systems. A.G. performed analytical calculations. A.G. and S.M. carried out analyses of the experimental data, the results of which are reported in the manuscript. All authors discussed and reviewed the manuscript.

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