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## Triboluminescence dominated by crystallographic orientation

Kuifang Wang<sup>1</sup>, Liran Ma<sup>1</sup>, Xuefeng Xu<sup>2</sup>, Shizhu Wen<sup>1</sup> & Jianbin Luo<sup>1</sup>

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Triboluminescence (TL) is an optical phenomenon that has a long and varied history with broad applications, such as damage detection, X-ray source, and mass health monitoring sensor. So far, the properties and mechanisms of TL remain not completely understood. The TL properties emitted during the sliding contact between  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  surfaces were studied along different crystallographic orientations. In this study, the TL intensity of  $\text{Al}_2\text{O}_3$  was significantly enhanced as  $\text{Al}_2\text{O}_3$  surface was along a particular crystallographic orientation, which is an unconventional phenomenon. TL enhancement of  $\text{Al}_2\text{O}_3$  was not affected by air atmosphere and atomic stocking mode of  $\text{Al}_2\text{O}_3$ . The enhancement mechanism of  $\text{Al}_2\text{O}_3$  may be influenced by the surface state of  $\text{Al}_2\text{O}_3$ . This work provides a new method to control the intensity of TL and novel ideas to elucidate the TL mechanism.

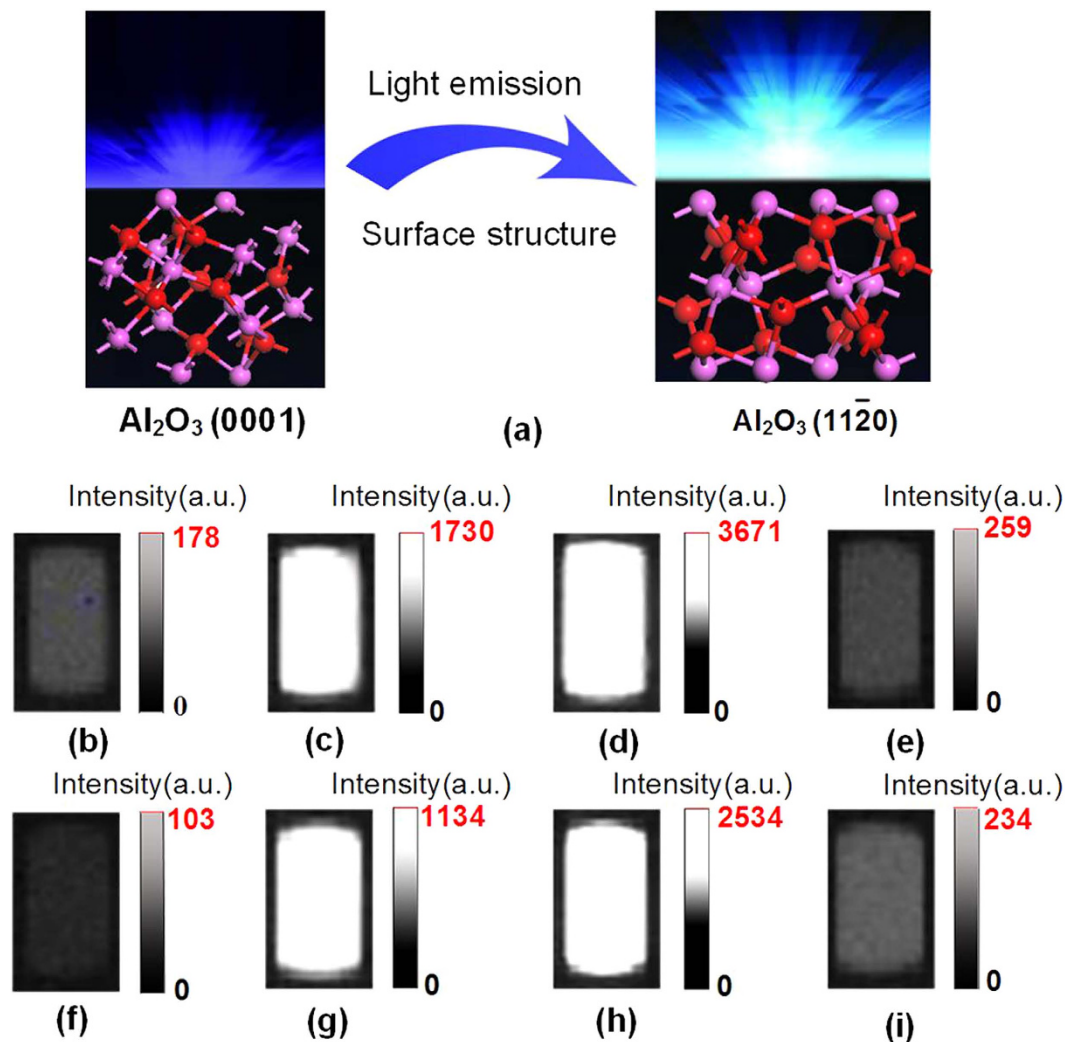
Triboluminescence (TL) is a luminescence phenomenon by solid materials when they are stressed or fractured<sup>1,2</sup>. TL is also regarded as fracture, piezoelectric, and mechanical luminescence. Since the 20th century, this phenomenon has gained increasing attention because of its broad application<sup>3</sup>. TL has been successfully used in mass health monitoring sensor<sup>4</sup>, X-ray source, and damage detector<sup>5</sup>, among others. The triboluminescent properties of crystal materials are mostly investigated because many common crystal materials exhibit TL performance<sup>6</sup>.

TL properties of crystal are often affected by external factors, such as gas atmosphere, gas pressure, temperature, and speed. The photon emission intensity of  $\text{Al}_2\text{O}_3$ ,  $\text{ZrO}_2$ , and  $\text{Si}_3\text{N}_4$  with a diamond stylus decreases with increasing number of carbon atoms in the hydrocarbon molecules<sup>7</sup>, whereas the emission intensity enhances to a maximum value at a particular n-butane gas pressure<sup>8</sup>. The TL intensities of NaCl and LiF doped with Br, Sr, Ca, and Pb decrease with temperature and disappear completely at  $105 \pm 5$  and  $180 \pm 10^\circ\text{C}$ , respectively<sup>9</sup>. Hollerman *et al.*<sup>10</sup> reported that the TL emission of Zn:Mn appears to be a function of speed for collision up to 6 Km/s. The doped impurities can change the TL properties of crystal materials; the TL emission intensity of ZnS:Mn is extremely strong, but TL properties are absent in ZnS<sup>4,9,11</sup>.

The physical properties of crystal structure also greatly influence TL. The discussion about TL properties of crystal with different space groups indicated that a non-centrosymmetric crystal structure is necessary but not sufficient for TL in crystal materials<sup>12,13</sup>. Hird<sup>14</sup> reported that the intensity of TL emission during diamond polishing in 'hard' direction is greater than 'soft' direction. The crystallographic orientation of  $\text{Al}_2\text{O}_3$  influences the atomic and electronic structures of alumina surfaces<sup>15</sup>. Brewer *et al.*<sup>16</sup> investigated the fluorescence band at 3.0 eV, which was produced by photoexcitation in high-purity  $\text{Al}_2\text{O}_3$  crystal. The results showed that the emitted light is plane polarized with the maximum intensity that occurs when the electric vector is perpendicular to the c axis of the crystal ( $E \perp c$ ) and with the minimum intensity that occurs when the electric vector is parallel to the c axis ( $E \parallel c$ )<sup>16</sup>. Kurita *et al.*<sup>17</sup> explored detailed atomic structures and electron states of stable and metastable surfaces of three important planes of  $\text{Al}_2\text{O}_3$ , namely, C plane [the (0001) surface], R plane [the (1 $\bar{1}$ 02) surface], and A plane [the (11 $\bar{2}$ 0) surface]. They found that the stoichiometric surfaces of the C plane have the lowest surface energy, followed by the stoichiometric surfaces of the R plane and then the A plane. In other studies, Cs-corrected high-resolution electron microscopy that combines first-principle calculations and image simulations was used to observe and investigate the quantitative and qualitative structures of (11 $\bar{2}$ 0) and (0001) surfaces<sup>18</sup>.  $\text{Al}_2\text{O}_3$  possesses good TL properties, but is not clearly investigated.  $\text{SiO}_2$  is a common crystal material, and the investigation of  $\text{SiO}_2$  of TL properties is very less.

The TL properties of crystal materials in different crystallographic orientations are rarely reported. We explored the TL properties of  $\text{Al}_2\text{O}_3$  along different crystallographic orientations by measuring TL emission during sliding with  $\text{SiO}_2$ . We discovered an unconventional phenomenon that the TL intensity was enhanced several tens of times as  $\text{Al}_2\text{O}_3$  plane was in a particular crystallographic orientation. This work may provide a novel method to control the intensity of TL.

<sup>1</sup>State Key Laboratory of Tribology, Department of Mechanical Engineering, Tsinghua University, Beijing 100084, China. <sup>2</sup>School of Technology, Beijing Forestry University, Beijing 100083, China. Correspondence and requests for materials should be addressed to L.M. (email: maliran@tsinghua.edu.cn)

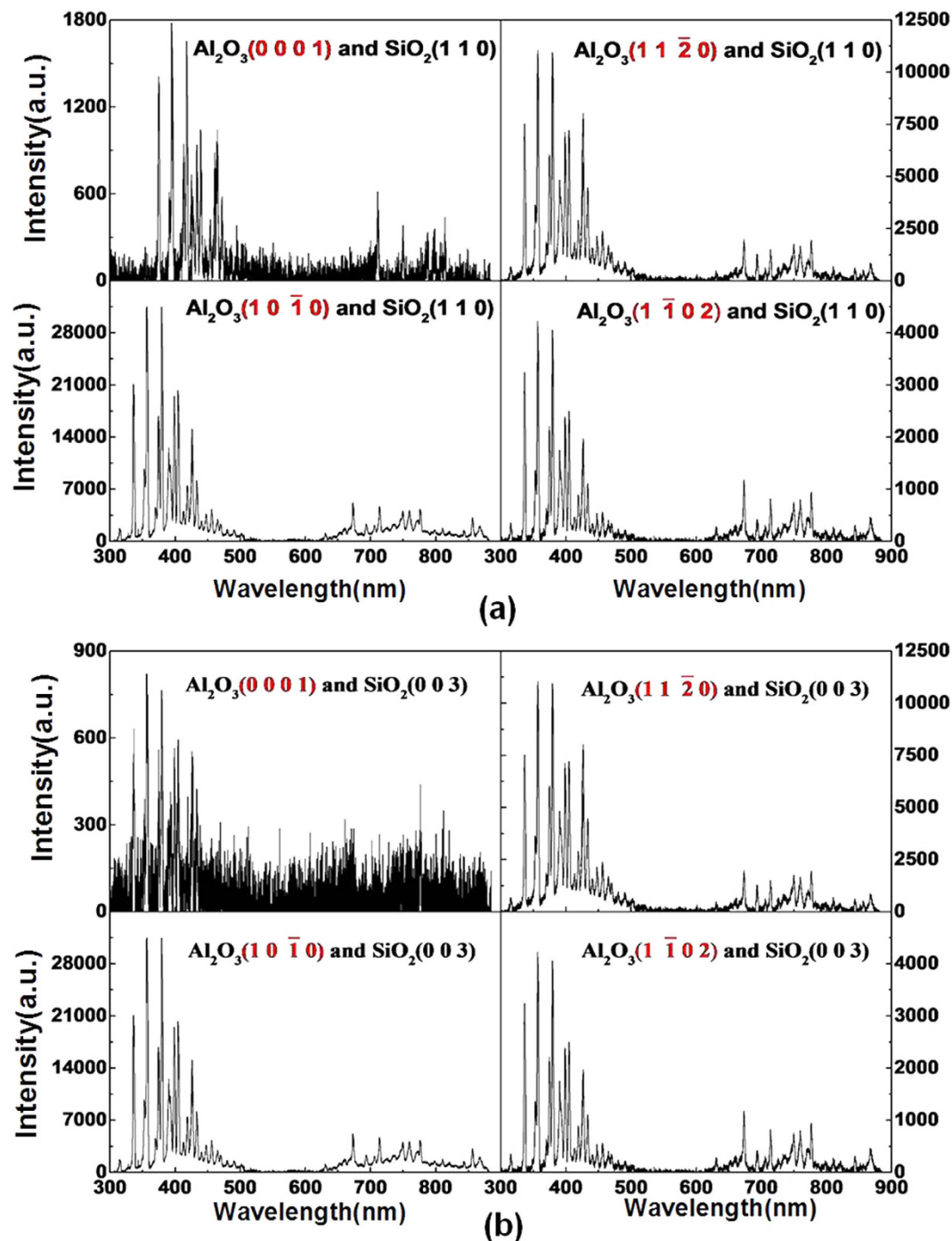


**Figure 1.** (a) Schematic of light emission of  $\text{Al}_2\text{O}_3$  (0001) and ( $11\bar{2}0$ ) surfaces, and photon images emitted under  $F = 10$  N and  $V = 33$  mm/s during the sliding between  $\text{SiO}_2$  (110) and (b)  $\text{Al}_2\text{O}_3$  (0001) with mean intensity  $I_{(b)} = 178$ , (c)  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) with  $I_{(c)} = 1730$ , (d)  $\text{Al}_2\text{O}_3$  ( $10\bar{1}0$ ) with  $I_{(d)} = 3671$ , and (e)  $\text{Al}_2\text{O}_3$  ( $1\bar{1}02$ ) with  $I_{(e)} = 259$ , and during the sliding between  $\text{SiO}_2$  (003) and (f)  $\text{Al}_2\text{O}_3$  (0001) surface with  $I_{(f)} = 103$ , (g)  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) surface with  $I_{(g)} = 1134$ , (h)  $\text{Al}_2\text{O}_3$  ( $10\bar{1}0$ ) surface with  $I_{(h)} = 2534$ , and (i)  $\text{Al}_2\text{O}_3$  ( $1\bar{1}02$ ) surface with  $I_{(i)} = 234$ .

## Results

**TL in ambient air.** Schematic diagram of light emission is shown in Fig. 1(a), the intensity of light emission of  $\text{Al}_2\text{O}_3$  is greatly enhanced due to crystallographic orientation changed from ( $11\bar{2}0$ ) to (0001). The images of photon emitted in ambient air during the sliding contact between  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}02$ ) surfaces and  $\text{SiO}_2$  (110) under normal force ( $F$ ) of 10 N and relatively shear velocity ( $V$ ) of 33 mm/s are shown in Fig. 1(b–e). Images of (c) and (d) are much brighter than those of (b) and (e). The mean intensities of  $I_{(b)}$ ,  $I_{(c)}$ ,  $I_{(d)}$ , and  $I_{(e)}$  were 249, 1605, 4512, and 222, respectively. The mean intensities of (b) and (e) are much weaker compared with those of (c) and (d).  $I_{(c)}$  is about 10 and 7 times higher than  $I_{(b)}$  and  $I_{(e)}$ , and  $I_{(d)}$  is about 22 and 14 times higher than  $I_{(b)}$  and  $I_{(e)}$ . The images of photons emitted during sliding between  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}02$ ) surfaces and  $\text{SiO}_2$  (003) are also shown in Fig. 1. In this condition, the intensity of TL has the same appearance although crystallographic orientation of  $\text{SiO}_2$  is changed. Thus, the TL intensities of  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) and ( $10\bar{1}0$ ) surfaces are significantly enhanced compared with the TL intensities of  $\text{Al}_2\text{O}_3$  (0001) and ( $1\bar{1}02$ ) surfaces.

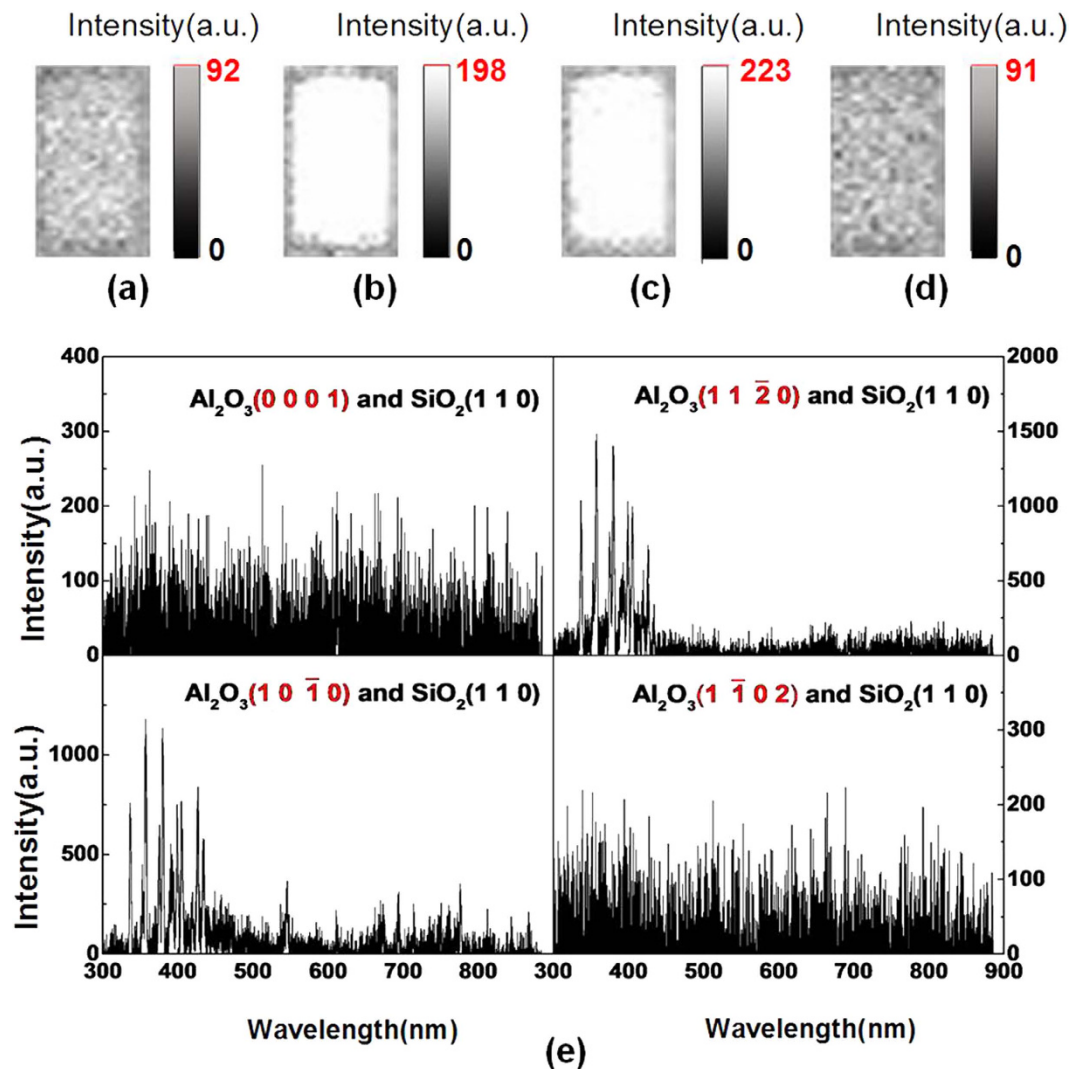
The spectra of photons emitted in ambient air during the sliding contact between  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}02$ ) surfaces and  $\text{SiO}_2$  (110) under  $F = 10$  N and  $V = 33$  mm/s are shown in Fig. 2(a). Many sharp peaks in regions 300–450 nm and 600–900 nm are on spectra, but the spectra intensities of diverse  $\text{Al}_2\text{O}_3$  surfaces are quite different. The maximum spectra intensities of  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) and ( $10\bar{1}0$ ) surfaces are much higher than those of  $\text{Al}_2\text{O}_3$  (0001) and ( $1\bar{1}02$ ) surfaces. The maximum spectrum intensity of  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) is nearly five times of spectra intensity of  $\text{Al}_2\text{O}_3$  (0001) and ( $1\bar{1}02$ ) surfaces, in quantitative terms, while the maximum spectrum intensity of  $\text{Al}_2\text{O}_3$  ( $10\bar{1}0$ ) is more than 10 times than those of  $\text{Al}_2\text{O}_3$  (0001) and ( $1\bar{1}02$ ) surfaces. As the crystallographic orien-



**Figure 2.** The spectra of photons emitted under  $F = 10$  N and  $V = 33$  mm/s in air: (a) during the sliding between (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces of  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  (110), (b) during the sliding contact between (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces of  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  (003).

tation of  $\text{SiO}_2$  is changed from (110) to (003), the spectra intensities of  $\text{Al}_2\text{O}_3$  (11 $\bar{2}$ 0) and (10 $\bar{1}$ 0) surfaces of  $\text{Al}_2\text{O}_3$  are much stronger. Thus, the intensity of emission is enhanced, when  $\text{Al}_2\text{O}_3$  (10 $\bar{1}$ 0) and (11 $\bar{2}$ 0) surfaces are sliding with  $\text{SiO}_2$ .

We conducted other sliding experiments to study further the enhancement mechanism of TL intensity of  $\text{Al}_2\text{O}_3$ . The images and spectra of photons emitted in ambient air during the sliding contact between  $\text{Al}_2\text{O}_3$  (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces and  $\text{SiO}_2$  (110) surface under  $F = 10$  N and  $V = 33$  mm/s by using a wire line are shown in Fig. 3. The wire line that connects the holder and platform could reduce the external electric potential difference between  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  to some extent. The mean intensities of images of  $\text{Al}_2\text{O}_3$  (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces are 92, 198, 223, and 91, respectively. The mean intensities of images of  $\text{Al}_2\text{O}_3$  (11 $\bar{2}$ 0) and



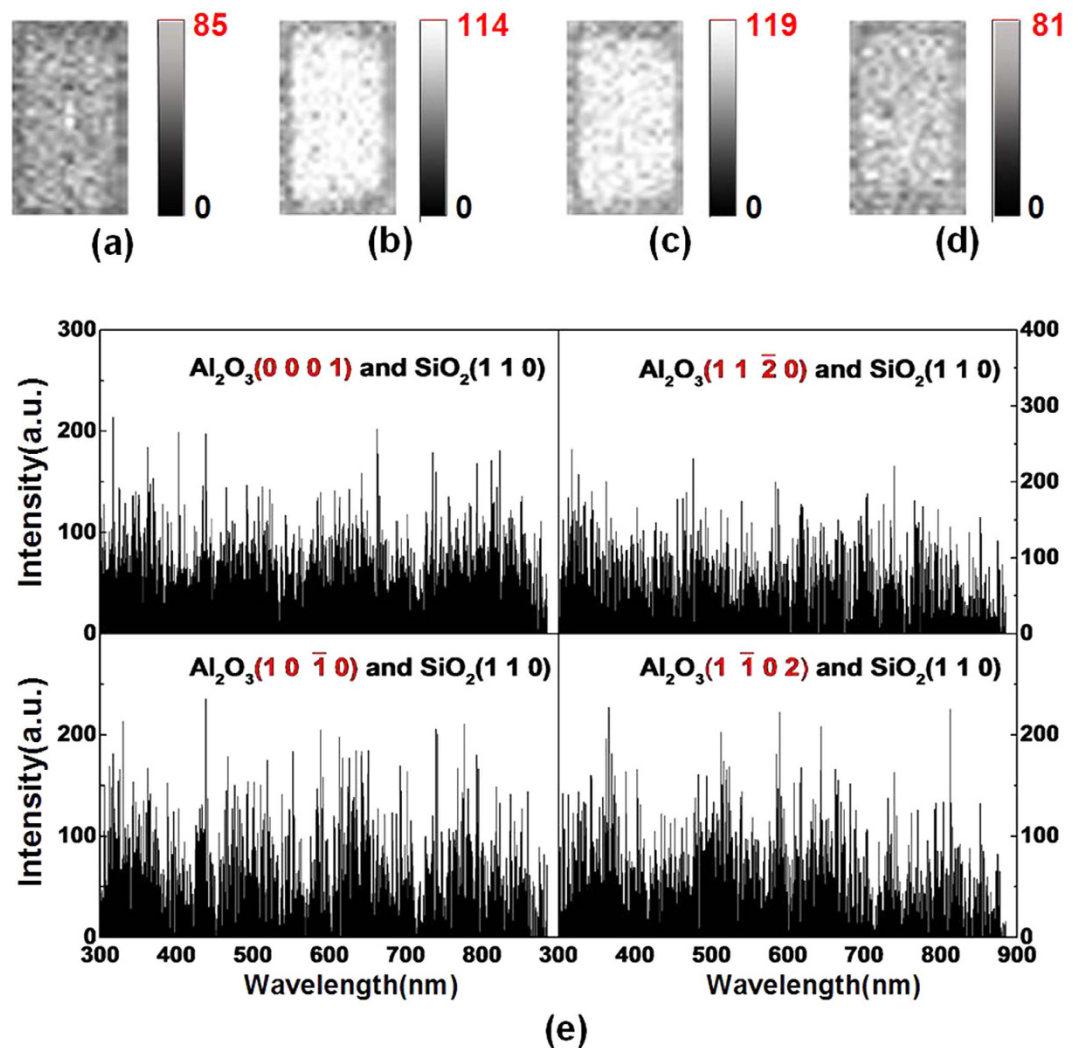
**Figure 3.** Photons under  $F = 10$  N and  $V = 33$  mm/s using a wire line in ambient air during the sliding between SiO<sub>2</sub> (110) and Al<sub>2</sub>O<sub>3</sub> (0001) (a) images with  $I = 92$  and (e) spectra, Al<sub>2</sub>O<sub>3</sub> (1120) (b) images  $I = 198$  and (f) spectra, Al<sub>2</sub>O<sub>3</sub> (1010) (c) images  $I = 223$  and (g) spectra, Al<sub>2</sub>O<sub>3</sub> (1102) (d)  $I = 91$  and (h) spectra.

(1010) surfaces are higher than those of Al<sub>2</sub>O<sub>3</sub> (0001) and (1102) surfaces. The maximum intensity of images of Al<sub>2</sub>O<sub>3</sub> (1010) is nearly three times stronger than those of Al<sub>2</sub>O<sub>3</sub> (0001) and (1102) surfaces. In this condition, the photons emitted in ambient air are extremely weaker through eliminating the effect of external electric potential difference. The spectra of photons of Al<sub>2</sub>O<sub>3</sub> (1120) and (1010) surfaces have sharp peaks as shown in Fig. 3(e), whereas those of Al<sub>2</sub>O<sub>3</sub> (0001) and (1102) surfaces have no peaks. Thus, the TL intensities of Al<sub>2</sub>O<sub>3</sub> (1010) and (1120) surfaces are much higher.

**TL in vacuum.** To explore better the TL properties of Al<sub>2</sub>O<sub>3</sub>, we conducted subsequent experiments in vacuum instead of ambient air. Figure 4 shows the images and spectra of photons emitted during the sliding contact between Al<sub>2</sub>O<sub>3</sub> (0001), (1120), (1010), and (1102) surfaces and SiO<sub>2</sub> (110) in vacuum under  $F = 10$  N and  $V = 33$  mm/s. The spectra of Al<sub>2</sub>O<sub>3</sub> in vacuum which are composed of instrument noise, have no characteristic peaks due to very lower photons intensity. The mean intensities of photon images of Al<sub>2</sub>O<sub>3</sub> (1120) and (1010) surfaces are higher than those of Al<sub>2</sub>O<sub>3</sub> (0001) and (1102) surfaces. The mean intensity of Al<sub>2</sub>O<sub>3</sub> (1120) is nearly two times stronger than those of Al<sub>2</sub>O<sub>3</sub> (1102) and (1010) surfaces.

The spectra of photons emitted in vacuum have no peaks because of extremely low light intensity as shown in Fig. 4(e). In this condition, the TL properties of Al<sub>2</sub>O<sub>3</sub> are also affected by crystallographic orientation, and the TL intensities of Al<sub>2</sub>O<sub>3</sub> (1120) and (1010) surfaces are enhanced.

**Friction and abrasion.** The friction coefficients of the sliding contact between Al<sub>2</sub>O<sub>3</sub> (0001), (1120), (1010), and (1102) surfaces and SiO<sub>2</sub> (110) are measured using a Universal Micro-Tribotester (UMT-3; Bruker, America), where the values of friction coefficients are 0.179, 0.258, 0.261, and 0.329. The friction coefficient of Al<sub>2</sub>O<sub>3</sub> (0001) surface is lowest and that of Al<sub>2</sub>O<sub>3</sub> (1102) surface is highest. The friction coefficients of Al<sub>2</sub>O<sub>3</sub> (1120) and (1010)



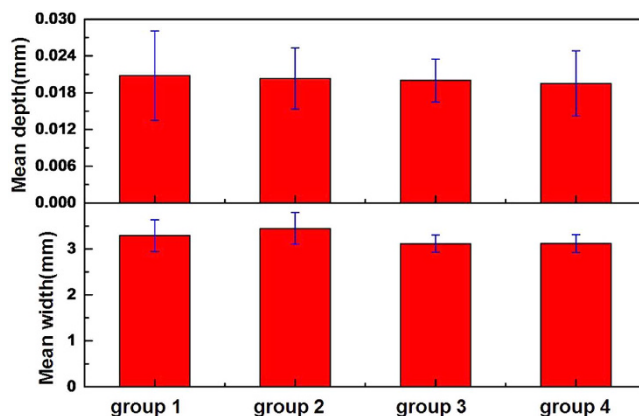
**Figure 4.** Photons under  $F = 10$  N and  $V = 33$  mm/s in vacuum during the sliding between  $\text{SiO}_2$  (110) and  $\text{Al}_2\text{O}_3$  (0001) (a) images with  $I = 35$  and (e) spectra,  $\text{Al}_2\text{O}_3$  ( $1\bar{1}0\bar{2}$ ) (b) images  $I = 114$  and (f) spectra,  $\text{Al}_2\text{O}_3$  ( $10\bar{1}0$ ) (c) images  $I = 119$  and (g) spectra,  $\text{Al}_2\text{O}_3$  ( $1\bar{1}0\bar{2}$ ) (d)  $I = 81$  and (h) spectra.

surfaces are much closer. The results prove that the TL intensity has no evident linear relationship with friction coefficient as the TL intensities of  $\text{Al}_2\text{O}_3$  ( $11\bar{2}0$ ) and ( $10\bar{1}0$ ) surfaces are much higher.

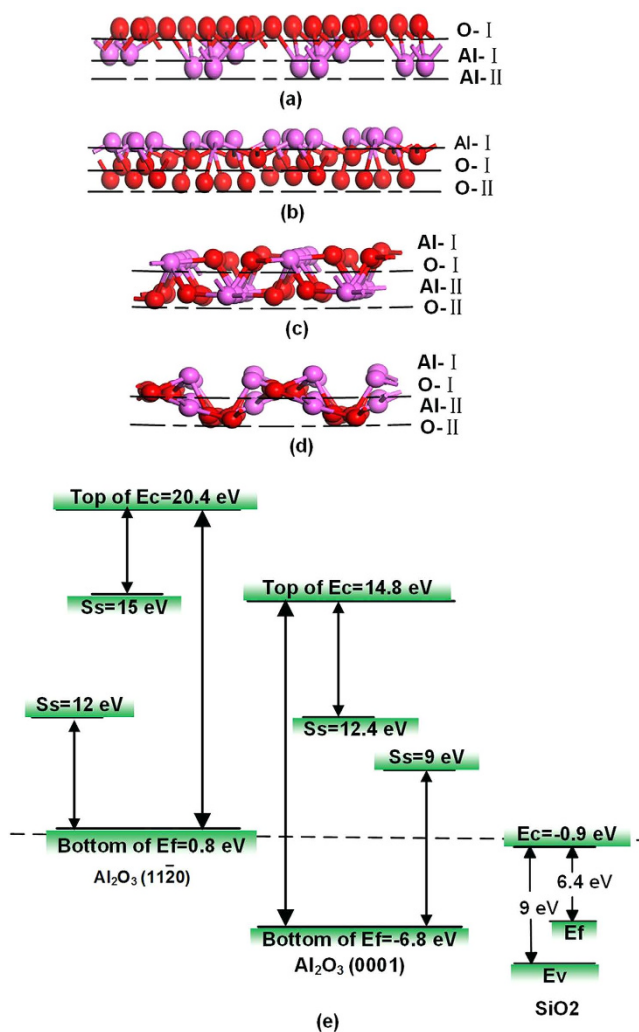
The depth and width of grinding cracks on  $\text{SiO}_2$  (110) surfaces, which are sliding with  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}0\bar{2}$ ) surfaces, are measured using a Talysurf (5P-120; Taylor Hobson, England). Figure 5 is the mean depth and width of the grinding crack of  $\text{SiO}_2$  (110) by sliding with  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}0\bar{2}$ ) surfaces. The values of mean depth of four groups are close to 0.02 mm, whereas those of  $\text{SiO}_2$  (110) surfaces have no distinct differences. The results stated that abrasions of  $\text{SiO}_2$  (110) surfaces sliding with  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}0\bar{2}$ ) surfaces have no apparent variations. Thus, the TL intensity of  $\text{Al}_2\text{O}_3$  is not influenced by wear of  $\text{SiO}_2$ , and enhancement mechanism of TL properties of  $\text{Al}_2\text{O}_3$  is dominated by crystallographic orientation.

## Discussion

The  $\text{Al}_2\text{O}_3$  crystal is a hexagonal crystal, and the side views of atom arrangement of  $\text{Al}_2\text{O}_3$  (0001), ( $11\bar{2}0$ ), ( $10\bar{1}0$ ), and ( $1\bar{1}0\bar{2}$ ) surfaces are shown in Fig. 6(a–d), respectively. In the hexagonal unit cell, the atoms are stacked along the [0001] direction in a sequence of an oxygen layer and Al double layers:  $-\text{AlAlO}_3-\text{AlAlO}_3-\text{R}$  (Fig. 6(a)). The atoms along the  $[11\bar{2}0]$  direction are a layer unit that consists of five atomic layers: an O layer constructed of one O atom in the  $1 \times 1$  unit, an O layer constructed of two O atoms, an Al layer constructed of four Al atoms, and an O layer constructed of one O atom<sup>17</sup>. This  $-\text{O}-\text{O}_2-\text{Al}_4-\text{O}_2-\text{O}-$  repeating layer unit has no dipole moment along the  $[11\bar{2}0]$  direction<sup>17</sup>. The atoms are stacked along the  $[10\bar{1}0]$  direction in a sequence of an O layer and an Al layer:  $-\text{O}-\text{Al}-\text{O}-\text{Al}-\text{R}$  (Fig. 5(b))<sup>18</sup>. The  $1 \times 1$  lateral unit of the ( $1\bar{1}0\bar{2}$ ) surface is a layer unit in a sequence of an Al layer constructed of two Al atoms and an O layer constructed of two O atoms. The  $-\text{Al}_2-\text{O}_2-\text{Al}_2-\text{O}_2-$  repeating layer unit has no dipole moment along the direction perpendicular to the ( $1\bar{1}0\bar{2}$ ) surface<sup>18</sup>. The atomic stacking mode of

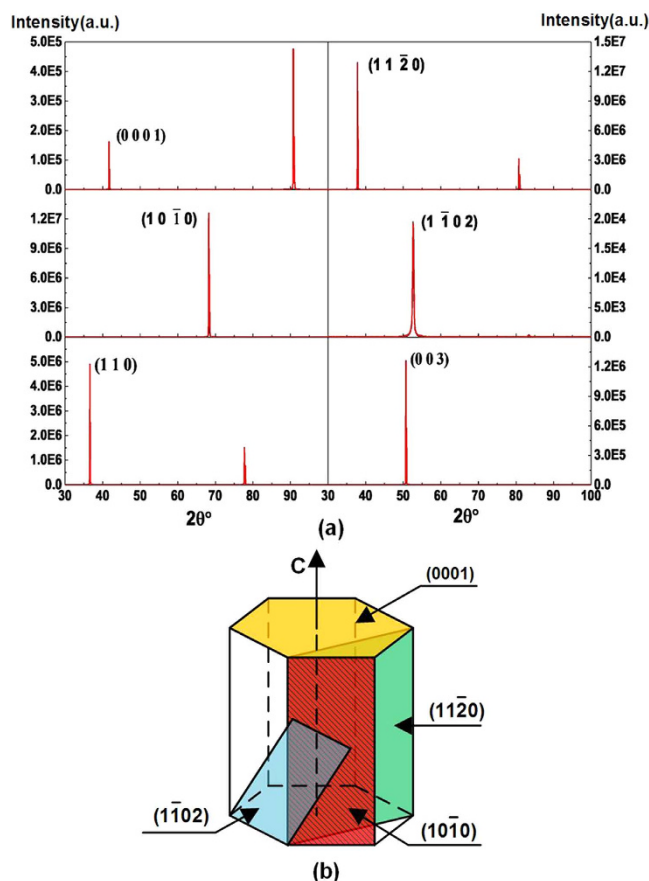


**Figure 5.** Mean depth and width of grinding crack of SiO<sub>2</sub> (110). Group 1 is SiO<sub>2</sub> (110) sliding with Al<sub>2</sub>O<sub>3</sub> (0001), group 2 is SiO<sub>2</sub> (110) sliding with Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0), group 3 is SiO<sub>2</sub> (110) sliding with Al<sub>2</sub>O<sub>3</sub> (10 $\bar{1}$ 0), group 4 is SiO<sub>2</sub> (110) sliding with Al<sub>2</sub>O<sub>3</sub> (1 $\bar{1}$ 02).

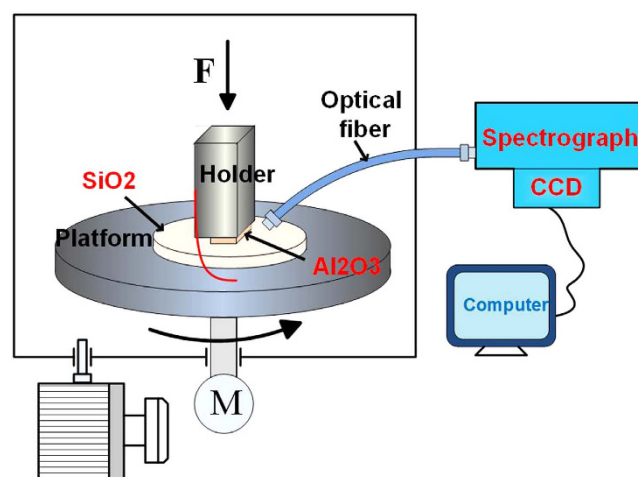


**Figure 6.** Side view of atom arrangement of Al<sub>2</sub>O<sub>3</sub> surfaces: (a) (0001), (b) (11 $\bar{2}$ 0), (c) (10 $\bar{1}$ 0), (d) (1 $\bar{1}$ 02); and (e): energy levels of Al<sub>2</sub>O<sub>3</sub> and SiO<sub>2</sub>.

Al<sub>2</sub>O<sub>3</sub> is extremely complicated along different crystallographic orientations, where (10 $\bar{1}$ 0) and (11 $\bar{2}$ 0) Al<sub>2</sub>O<sub>3</sub> surfaces have no similar stacking mode. No regular discipline exists to indicate that the enhancement of TL intensities of (10 $\bar{1}$ 0) and (11 $\bar{2}$ 0) Al<sub>2</sub>O<sub>3</sub> surfaces is related to atomic stacking mode.



**Figure 7.** (a) Single-crystal X-ray diffractometer results of  $\text{Al}_2\text{O}_3$  surfaces with miller indices (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), (1 $\bar{1}$ 02), and  $\text{SiO}_2$  surfaces with miller indices (110) and (003). (b)  $\text{Al}_2\text{O}_3$  planes in hexagonal unit cell.



**Figure 8.** Schematic diagram of the sliding experiment setup for observation of the images and spectra of photons.

The results of rough measurements of surface charge of  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  by using faraday cup<sup>19</sup> showed that  $\text{Al}_2\text{O}_3$  surface was negatively charged and  $\text{SiO}_2$  was positively charged. Tribocharging mechanism is that electrons transferred from a surface with a low work function to a mating surface with a high work function<sup>20</sup>. In surface state theory, charge is exchanged between surface states in proportion to the difference between the effective or surface work functions of the two materials<sup>21</sup>. Surface potential difference is the fermi level difference between

original surfaces as well as the work function difference<sup>22</sup>. In equation (1),  $V_C$  is the surface potential difference,  $\phi_1, \phi_2$  are the work functions of two surfaces respectively<sup>23</sup>.

$$V_C = \frac{\phi_1 - \phi_2}{e} \quad (1)$$

SiO<sub>2</sub> surface has a lower work function than Al<sub>2</sub>O<sub>3</sub> surface<sup>24,25</sup>. Thus, electrons are transferred from SiO<sub>2</sub> to Al<sub>2</sub>O<sub>3</sub>, resulting in the former being positive and the latter being negative, then electric field between SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> contacting surfaces is formed<sup>26</sup>. Electrons of ambient air molecules in electric field will be excited from ground level to the excited levels, then fall down to the lower or ground level, photons are emitted<sup>26</sup>. The sharp peaks of spectra are caused by the electrical breakdown of ambient gas<sup>27</sup>. The spectra peaks of photons in the region 300–450 nm are assigned to C<sup>3</sup>π → B<sup>3</sup>π electron transitions of N<sub>2</sub><sup>28</sup>. Other sharp peaks of spectra are mainly due to the B<sup>3</sup>π → A<sup>3</sup>Σ electron transitions of N<sub>2</sub> and the b<sup>1</sup>Σ<sub>g</sub><sup>+</sup> → X<sup>3</sup>Σ<sub>g</sub><sup>-</sup> electron transition in O<sub>2</sub><sup>27</sup>. The pressure of vacuum chamber is between 1 to 10 Pa during the sliding experiment. Air molecules are much smaller under this condition. Thus, Photons emitted during sliding between SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> in vacuum is very few, the spectra of Al<sub>2</sub>O<sub>3</sub> in vacuum have no peaks.

As shown in Fig. 6(e), the band gap of SiO<sub>2</sub> is 9 eV, and the conduction band (Ec) is -0.9 eV<sup>29,30</sup>. The surface states (S<sub>s</sub>) of Al<sub>2</sub>O<sub>3</sub> (0001) and Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0) are 9 and 12 eV, respectively. The bottom of Fermi level of Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0) is 0.8 eV<sup>31,32</sup>. Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0) has two surface-state bands at 12 and 15 eV. The bottom of the Fermi level of Al<sub>2</sub>O<sub>3</sub> (0001) is -6.8 eV, which is much lower than that of (11 $\bar{2}$ 0). Al<sub>2</sub>O<sub>3</sub> (0001) has two surface-state bands at 9 and 12.5 eV. The surface states of Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0) are much higher than those of Al<sub>2</sub>O<sub>3</sub> (0001). Thus, enhancement mechanism of Al<sub>2</sub>O<sub>3</sub> may be related to the energy levels of Al<sub>2</sub>O<sub>3</sub>, while much higher energy levels of Al<sub>2</sub>O<sub>3</sub> (11 $\bar{2}$ 0) surface result in increasing TL emission.

The enhancement mechanism of TL properties of Al<sub>2</sub>O<sub>3</sub> may be influenced by the surface state of Al<sub>2</sub>O<sub>3</sub>. The enhancement mechanisms of Al<sub>2</sub>O<sub>3</sub> still need further exploration, and our work may provide a novel method to control the TL intensity.

## Methods

**Materials.** SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> crystals with trigonal and hexagonal crystal structures, respectively, were used in the sliding experiment. The two types of crystal planes of SiO<sub>2</sub> were (110) surface by X cut and (003) surface by Z cut. The results of single-crystal X-ray diffractometer of SiO<sub>2</sub> surfaces are shown in Fig. 7(a). Four surface planes of Al<sub>2</sub>O<sub>3</sub> crystal, including C plane (0001), A plane (11 $\bar{2}$ 0), M plane (10 $\bar{1}$ 0), and R plane (1 $\bar{1}$ 02)<sup>33</sup>, were used as shown in Fig. 7(b). (11 $\bar{2}$ 0) and (10 $\bar{1}$ 0) planes are parallel to axis, and (0001) plane is perpendicular to axis, (1 $\bar{1}$ 02) plane is crossed with C axis. The dielectric constant of sapphire at 298 K in 10<sup>3</sup>–10<sup>9</sup> Hz interval is  $\parallel C = 11.5$ ,  $\perp C = 9.3$ <sup>34</sup>. Dielectric constant of four planes of Al<sub>2</sub>O<sub>3</sub> have little difference. The SiO<sub>2</sub> with width of 3 mm and thickness of 2 mm and the Al<sub>2</sub>O<sub>3</sub> with diameter of 30 mm and thickness of 2 mm were purchased from Shanghai Daheng Optics & Fine Mechanics Co. Ltd. The Vickers hardness of (110) and (003) surfaces of SiO<sub>2</sub> are 1257 and 1167, respectively, and those of Al<sub>2</sub>O<sub>3</sub> (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces are 2060, 2119, 2076, and 2329, respectively. The roughness of (110) and (003) surfaces of SiO<sub>2</sub> are 1.6 and 1.44 nm, respectively, and those of Al<sub>2</sub>O<sub>3</sub> (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces are 5.43, 5.21, 5.18, and 5.51 nm. In our experiments, Al<sub>2</sub>O<sub>3</sub> (0001), (11 $\bar{2}$ 0), (10 $\bar{1}$ 0), and (1 $\bar{1}$ 02) surfaces would be slid with SiO<sub>2</sub> (110) and (003) surfaces, respectively.

**Experimental setup.** The schematic of the experiment setup used to observe the images and spectra of photons during sliding between SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> is shown in Fig. 8. Optical fiber was used to gather light and then transmitted the light to a spectrograph (SP2500; Princeton Instruments, America). The images and spectra of photons were obtained with the spectrograph and CCD. The spectra of photons ranged from 300 nm to 900 nm, and the image of photons reflected the overall intensity of light.

**Experimental condition.** SiO<sub>2</sub> was adhered to the rotating platform along with the motor, and the Al<sub>2</sub>O<sub>3</sub> wafer was fixed on a holder under a normal force of 10 N, as shown in Fig. 8. The bottom surface of Al<sub>2</sub>O<sub>3</sub> wafer was sliding over the top surface of SiO<sub>2</sub> wafer. The integration time (T) of CCD camera was 10 min, and the relative shear velocity (V) between Al<sub>2</sub>O<sub>3</sub> and SiO<sub>2</sub> wafer was 33 mm/s. The sliding experiment was performed in ambient air and vacuum. The vacuum pressure was between 1 and 10 Pa, and air humidity was nearly 10%. The red line in Fig. 8 is a removable wire line. The wire line connected the holder and the platform to reduce the influence of the external electrical potential difference. Each test was run three times, and mean was obtained to remove any discrepancies. The mean intensity of photon images is calculated by summing values of bright zone then dividing numbers of pixel points.

## References

- Zink, J. I. & Triboluminescence, *J. Accounts Chem Res.* **11**, 289–295 (1978).
- David, O. O. *et al.* Progress in triboluminescence-based smart optical sensor system. *J. Lumin.* **131**, 1407–1418 (2011).
- Chandra, B. P. Kinetics of triboluminescence in crystals. *J. Phys D: Appl Phys.* **10**, 1531–1538 (1977).
- Hollerman, W. A. *et al.* Annealing effects of triboluminescence production on irradiated ZnS:Mn. *J. Surf Coat Technol.* **201**, 8328–8387 (2011).
- Olawale, D. O. *et al.* Getting light through cementitious composites with *in situ* triboluminescent damage sensor. *J. Struct Health Monit.* **13**, 177–189 (2014).
- Sweeting, L. M., Triboluminescence with and without Air. *J. Chem Mater.* **13**, 854–870 (2001).
- Nakayama, K. J. & Hashimoto, H. S. Triboemission of charged particles and photons from wearing ceramic surfaces in various hydrocarbon gases. *J. Wear.* **185**, 183–188 (1995).
- Nakayama, K. J. & Hashimoto, H. S. Triboemission, tribochemical reaction, and friction and wear in ceramics under various n-butane gas pressures. *J. Tribol Int.* **29**, 385–393 (2019).



9. Chandra, B. P., Khokhar, M. S. K., Elyas, M. & Rao, A. M. D. The effect of temperature on the triboluminescence of Ba, Sr, Ca and Pb doped NaCl and LiF single crystals. *J. Phys D: Appl Phys.* **23**, 90–94 (1990).
10. Hollerman, W. A. *et al.* Review of Triboluminescence Impact Research at Projectile Speeds of 1 m/s to 6km/s. *J. Procedia. Eng.* **58**, 392–400 (2013).
11. Chandra, V. K., Chandra, B. P. & Jha, P. Self-recovery of mechanoluminescence in ZnS:Cu and ZnS:Mn phosphors by trapping of drifting charge carriers. *J. Appl Phys Lett.* **103**, 1–5 (2013).
12. Sweeting, L. M. *et al.* Crystal Structure and Triboluminescence 2. 9-Anthracenecarboxylic Acid and Its Esters. *J. Chem Mater.* **9**, 1103–1115 (1997).
13. Sweeting, L. M. & Rheingold, A. L. Crystal Structure and Triboluminescence. 1. 9-Anthryl Carbinols. *J. Phys. Chem.* **92**, 5648–5655 (1988).
14. Hird, J. R., Chakravarty, A. & Walton, A. J. Triboluminescence from diamond. *J. Phys D: Appl Phys.* **40**, 1464–1472 (2007).
15. Fang, C. M. & Groot, R. A. The nature of electron states in AlN and  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>. *J. Phys.: Condens. Matter.* **19**, 1–6 (2007).
16. Brewer, J. D., Jeffries, B. T. & Summers, G. P. Low-temperature fluorescence in sapphire. *J. Phys Rev B.* **22**, 4900–4906 (1980).
17. Kurita, T., Uchida, K. & Oshiyama, A. Atomic and electronic structures of  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> surfaces. *J. Phys Rev B.* **82**, 1–14 (2010).
18. Gautier, M., Fenaud, G., Pham Van, L., Villette, B., Pollak, M., Thomat, N. & Duraud, J. P.  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) Surfaces: Atomic and Electronic Structure. *J AM Ceram Soc.* **77**, 323–334 (1994).
19. Zhang, Y. Y. & Shao, T. M. A method of charge measurement for contact electrification. *J. Electrostat.* **71**, 712–716 (2013).
20. Nakayama, K. J. & Nevshupa, R. A. Effect of dry air pressure on characteristics and patterns of tribomicroplasma. *J. Vacu.* **74**, 11–17 (2004).
21. Castle, P. S. G. Contact Charging Between Insulators. *J. Electrostat.* **40**, 13–20 (1997).
22. Lowell, J. & Rose-innes, A. C. Contact electrification. *J. Adv Phys.* **29**, 947–1023 (1980).
23. Lavielle, L. Electric field effect on the friction of a polyethylene-terpolymer film on a steel substrate. *Wear.* **176**, 89–93 (1994).
24. Fonseca, L. R. C. & Knizhnik, A. A. First-principles calculation of the TiN effective work function on SiO<sub>2</sub> and on HfO<sub>2</sub>. *J. Phys Rev B.* **74**, 1–13 (2006).
25. Guo, J., Ellis, D. E. & Lam, D. J. Electronic structure and energetics of sapphire (0001) and (10 $\bar{1}$ 0) surfaces. *J. Phys Rev B.* **45**, 647–656 (1997).
26. Zink, J. I. Squeezing Light out of Crystals: Triboluminescence. *J. Naturwissenschaften.* **10**, 507–512 (1981).
27. Nakayama, K. J. & Nevahupa, R. A. Characteristics and Pattern of Plasma Generated at Sliding Contact. *J. Tribol.* **125**, 780–787 (125).
28. Hardy, G. E. & Zink, J. I. Triboluminescence and Pressure Dependence of the Photoluminescence of Tetrahedral Manganese(II) Complexes. *J. Inorg Chem.* **12**, 3061–3065 (1976).
29. Markoy, S. From ab initio properties of the Si-SiO<sub>2</sub> interface, to electrical characteristics of metal-oxide-semiconductor devices. *J. Phys Conf Ser.* **242**, 1–4 (2010).
30. Geis, M. W., Gregory, J. A. & Pate, B. B. Capacitance-Voltage Measurements on Metal-SiO<sub>2</sub>-Diamond Structures Fabricated with (100)- and (1 1 1)-Oriented Substrates. *J. Ieee t electron dev.* **38**, 619–626 (1991).
31. Minseok, C., Anderson, J. & Chris, G. V. Native point defects and dangling bonds in  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>. *J. Appl Phys.* **113**, 1–6 (2013).
32. Nagy, L. T. *et al.* Electronic structure of alumina surface. *J. Int J Quantum Chem.* **70**, 341–350 (1998).
33. Dobrovinskaya, E. R., Lytvynov, L. A. & Pishchik, V. *Sapphire: material, manufacturing, applications.* Ch. 2, 76–78 (Spring Science & Business Media, 2009).
34. Dobrovinskaya, E. R., Lytvynov, L. A. & Pishchik, V. *Sapphire: material, manufacturing, applications.* Ch. 2, 115–117 (Spring Science & Business Media, 2009).

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## Author Contributions

L.M., X.X. and S.W. conceived the experiments. K.W. conducted the experiments and processed the experimental data. L.M., X.X. and K.W. analysed the results. All authors reviewed the manuscript.

## Additional Information

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