SCIENTIFIC REPORTS

Received: 11 February 2015 Accepted: 19 June 2015 Published: 16 July 2015

OPEN Versatile MoS, Nanosheets in **ITO-Free and Semi-transparent Polymer Power-generating Glass**

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Chemical exfoliated ultra-thin MoS₂ nanosheets (NSs) with well 2D structure were demonstrated for interfacial layers and Aq nanowires composite transparent electrode in polymer solar cells (PSCs). The smooth and uniform n-type and p-type (after the plasma treatment) MoS₂ NSs could improve fill factor of devices and light absorption in active layer. The optimized Ag nanowires-MoS, NSs (AqNW-MoS, NSs) transparent electrode presented a low sheet resistance of 9.8 Ω sq⁻¹, and the corresponding transmittance also exhibited a high value of 93.1% at 550 nm. As a result, ITO-free PSCs based on AqNW-MoS, NSs/n-MoS, NSs cathode and p-MoS, NSs/Aq anode achieved a highest PCE of 8.72%. Furthermore, a high efficiency (6.55%), large area and low cost semi-transparent power-generating glass was obtained, after reducing the thickness of top Ag electrode from 100 nm to 30 nm. To our best knowledge, it is the highest performance for semi-transparent PSCs devices reported up to now. The novel semi-transparent power-generating glass showed good performance and color purity for commercial applications in the near future.

Polymer solar cells (PSCs) are extremely attractive candidates for use in next-generation solar cell technologies because of their mechanical flexibility, light weight, and cost-effective production through solution-based manufacturing processes¹⁻⁸. To reach commercialization and mass production, PSCs must exhibit high performance and special applications⁹⁻¹¹.

A typical PSC is based on the bulk-heterojunction (BHJ) device configuration, which sandwiches a layer of polymer donor and fullerene acceptor blend between a transparent electrode (such as indium tin oxide (ITO)) and an opaque, reflective back electrode (such as aluminum (Al) or argentum (Ag))¹². There are also interfacial layers existing between active layer and electrodes, and the most common anode interfacial layer is poly (3, 4-ethylene dioxythiophene): poly (styrene sulfonic acid) (PEDOT: PSS) or MoO_3 for hole transportation and the ZnO or TiO_x are the most widely used for cathode interfacial modification^{7,12-14}. A great effort is currently being exerted across a variety of research fields in the quest to achieve high-performance devices, for example, the synthesis of novel polymer donors, interfacial morphology control, optimized device structure and processing optimization. These devices have shown high power conversion efficiencies (PCE) approaching to 10%^{4,15–17}.

Unlike traditional bulk solar cells prepared from silicon or polysilicon, organic absorbing materials with confined absorption band can selectively be either transparent or semi-transparent in different regions of the solar spectrum^{11,18}. Accordingly, semi-transparent polymer solar cells (STPSCs), an extension of PSCs which utilize transparent conductive materials as both electrodes, have recently gained much scientific attention and are considered to be the highest priority market for PSCs^{9,19}. The STPSCs offer an extensive spectrum of applications such as power-generating windows for buildings, foldable curtains, mobile terminals and clothes, etc. To date, although many transparent polymer-based organic

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Figure 1. Device configuration of common inverted PSC and MoS₂ nanosheets (NSs) modified ITO-free STPSC.

solar cells have been reported previously^{18,20–24}, the lack of appropriate device structure and suitable materials to use in interfacial layers and electrodes for PSCs has meant that these devices have exhibited either low efficiency (<5%, single junction) or unsatisfactory average visible transmittance (AVT) and color purity for large-scale commercial applications^{10,11}.

Similar to graphene, other two-dimensional (2D) nanomaterials have been receiving great attention in recent years mainly due to their complementary electronic properties when compared with graphene²⁵⁻²⁸. Molybdenum disulfide (MoS₂) is a 2D material consisting of hexagonal sheets of molybdenum (Mo) sandwiched between two hexagonal sheets of sulfur (S). MoS₂ exhibits specific anisotropic mechanical and electrical properties due to the weak bonding between layers^{29,30}. A mobility value of over 200 cm² V⁻¹ s⁻¹ has been reported in a field-effect transistor architecture^{31,32}. It has also been demonstrated that MoS₂ could act as Schottky-barrier active layer or hole transport layer (HTL) in photovoltaic³³⁻³⁷. However, MoS₂ has not been taken full advantages of its 2D plane properties, even though there are few traps on the layer surface due to its lack of dangling bonds.

We report here on materials and processes for reliable and cost-efficient processing of ITO-free semi-transparent polymer solar cells by introducing ultra-thin versatile MoS₂ nanosheets as interfacial layers and semi-transparent electrodes with employing a silver nanowires-n-type MoS₂ nanosheets (AgNW-MoS₂ NSs) cathode and thin Ag metal anode. This is the first time to report a novel AgNW-MoS₂ NSs/n-MoS₂ NSs composite cathode can form excellent 2D morphology for interfacial contact and optical distribution in devices, and p-MoS₂ NSs/ thin Ag metal is found to achieve a suitable anode with high conductivity and transparency at the same time. The ultra-thin versatile MoS_2 can effectively avoid aggregation of AgNW electrode and is beneficial for interfacial modification of devices, proposing the MoS₂ can replace the traditional electrode and interface layer in organic optoelectronic materials. Incorporation of the AgNW-MoS₂ NSs/n-MoS₂ NSs cathode and p-MoS₂ NSs/ thin Ag as the hybrid anode not only enhances fill factor of devices but also improves the light absorption in the BHJ layer and the transmittance of the devices simultaneously as shown in Fig. 1. After optimization of the devices, combination of material selection and device engineering facilitates the fabrication of ITO-free PSCs and STPSCs with good average performance of 8.00% (best value of 8.72%) and 6.02% (best value of 6.55%), respectively, based on poly[4,8-bis-(2-ethyl-hexyl-thiophene-5-yl)-benzo[1,2-b:4,5-b]dithiophene-2,6-diyl]-alt-[2-(2-ethyl-hexanoyl)-thieno[3,4-b]thiophen-4,6-diyl] (PBDTTT-C-T)³⁸. (6,6)-phenyl-C₇₀ butyric acid methyl ester (PC70BM) (PBDTTT-C-T:PC70BM) system. Furthermore, these good chrominance large-scale STPSCs module based on the novel device structure can light a LED chip under natural light illuminated, which shows great application possibility in power-generating windows for building integrated photovoltaic (BIPV).

Methods

Sample Preparation. Chemically exfoliated Molybdenum sulfide (MoS₂) nanosheets (NSs) were synthesized through Li intercalation. 0.8 g raw MoS₂ powders (Sigma-Aldrich) were immersed in 24 mL n-butyllithium (n-BuLi, 2.5 M hexane solution) solution. The mixture was stirred at room temperature for 48 h under an Ar atmosphere and then allowed to settle for hours. The mixture was centrifuged several times to remove LiOH and unexfoliated material. The obtained MoS₂ NSs were n-type materials, for the p-type MoS₂ NSs were then prepared after 30 min oxygen plasma treatment (PDC-32G-2 PLASMA

CLEANERM, high level, 18W). The sample was diluted with isopropyl alcohol to ~0.5 mg mL⁻¹ for device fabrication.

For the post-treatments, the AgNWs-MoS₂ ink was manufactured by blending the Ag NWs dispersion (diam. \times L 60 nm \times 10 µm, 0.5% isopropyl alcohol suspension Sigma-Aldrich) and n-type MoS₂ isopropyl alcohol solution (1:1,V:V). Preparation of the Ag NWs and AgNWs-MoS₂ Films: The Ag NW films were prepared using the spin-coating process and were formed on precleaned glass substrates that were attached to a supporting glass substrate. The as-received dispersion containing Ag NWs and AgNWs-MoS₂ were spin coated for 40 s at speeds ranging from 500–4000 rpm. The dispersion was sonicated for 30 min and shaken well before being the spin-coating process. The formed electrode films were annealed at \approx 120 °C for 10 min in a glove box filled with nitrogen (the detail process was shown in support information in Supplementary Fig. S1).

Device Fabrication. The polymer solar cells were fabricated on Glass/ITO, Glass/AgNW or Glass/Ag NW-MoS₂ electrodes. For the electron transport layer, the sol-gel derived ZnO film using the reported method³⁹ or 40 nm thickness of n-type MoS₂ was placed on the top of electrode. The thickness of ZnO film is approximately 30 nm, determined by a profilometer (Alpha-Step-IQ). Subsequently, the modified samples were transferred to the nitrogen-filled glove-box. The D-A copolymer-containing photo-active layer was prepared by spin coating (900 rpm) the dichlorobenzene solution of D-A copolymer (PBDTTT-C-T) and PC₇₀BM (1:1.5 w/w, polymer concentration of 10 mg mL⁻¹) with 3% volume ratio of DIO additive on the modified electrode. The thickness of the photoactive layer was about 100 nm. Then, the MoO₃ precursor or p-type MoS₂ solution with optimal thickness was spin-cast on top of the polymer:fullerene composite layer. Finally, the device was pumped down in vacuum (<10⁻⁷ torr; 1 torr~133 Pa), and a~100 or 30 nm thick Ag electrode was deposited on top. The deposited Ag electrode area defined the active area of the devices as 0.06 or 0.65 cm² (all the areas were tested with an aperture).

Power-generating glass Fabrication. The power-generating glass was fabricated from eight 0.65 cm^2 (effective area) **Device E**. All the devices were weld together, and then encapsulating by UV curing glue. The LED chip was accessed in the circuit.

Characterization. X-ray diffraction (XRD) patterns of the MoS_2 nanoparticles were carried out on a Bruker D8 Focus X-ray diffractometer operating at 30 kV and 20 mA with a copper target ($\lambda = 1.54$ Å) and at a scanning rate of 1 o/min. Sheet resistances of electrodes were measured by using a four point probe setup with a source measurement unit (Keithley 2400). Current-voltage (J-V) characteristics were characterized using Keithley 2400. The currents were measured in the dark and under 100 mW·cm⁻² simulated AM 1.5 G irradiation (Abet Solar Simulator Sun2000). All the measurements were performed under ambient atmosphere at room temperature. The scan range is form 0V to 1V, and 6.7 mV for each step. All the J-V curves are based on 150 points. The incident photo-to-electron conversion efficiency spectrum (IPCE) were detected under monochromatic illumination (Oriel Cornerstone 260 1/4 m monochromator equipped with Oriel 70613NS QTH lamp), and the calibration of the incident light was performed with a monocrystalline silicon diode. Transmittance spectra were analyzed by UV-vis spectroscopy (Perkin Elmer Lambda 750). The morphologies of films were investigated by atomic force microscopy (AFM) using a Digital Instrumental Nanoscope 31 operated in the tapping mode. The thicknesses of all the layers were measured by surface profilometer (Alpha-Step-IQ). XPS studies were performed on a Thermo-VG Scientific ESCALAB 250 photoelectron spectrometer using a monochromated AlKa (1,486.6 eV) X-ray source. All recorded peaks were corrected for electrostatic effects by setting the C-C component of the C 1s peak to 284.8 eV. The base pressure in the XPS analysis chamber was 2×10^{-9} mbar. For the UPS measurements, He I (21.22 eV) radiation line from a discharge lamp was used, with an experimental resolution of 0.15 eV. All the UPS measurements of the onset of photoemission for determining the work function were done using standard procedures with a -5V bias applied to the sample. Raman spectroscopy was performed using an InVia Raman Microscope system (Renishaw, Inc.), with an Ar+ ion laser operating at 613 nm and 1.2 mW. All the simulation were calculated via Matlab program, the Matlab files were provided from McGehee group, Center for Advanced Molecular Photovoltaics, Stanford University. For the evaluation of the color rendering indices (CRIs), the experimental transmission of each device is folded with the AM1.5 spectrum to obtain the perceived transmission under solar illumination. The resulting data is coupled with the CIE 1931 2° standard observer color matching functions to obtain the corresponding xyY points.

Results

Properties of ultrathin MoS₂ NSs. In order to obtain ultrathin MoS₂ NSs, the raw MoS₂ powders were exfoliated by chemical lithium intercalation–exfoliation method^{28,40}. Generally, the synthetic procedure involves two steps: n-BuLi can intercalate bulk MoS₂ with lithium atoms and form LixMoS₂ (0 < x < 1) in the gram scale or more, at room temperature. And then LixMoS₂ can react with H₂O rapidly and release large amounts of hydrogen gas, which can push the MoS₂ layers and cause them to separate from each other and form homogeneous ultrathin MoS₂ NSs. The powder X-ray diffraction (XRD) patterns of raw and exfoliated MoS₂ nanosheets (Supplementary Fig. S2, Supporting Information), after chemical exfoliation, show that only the peaks of (002) and (103) plane remain which confirms



Figure 2. Tapping-mode atomic force microscopy (AFM) images of films (a) raw MoS_2 film height image, (b) exfoliated MoS_2 NSs film height image, (scan range: $5\mu m \times 5\mu m$).

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that the MoS2 nanosheets were successfully striped. And the disappearance of other peaks could proves ultrathin MoS₂ nanosheets tightly lie on the substrate with preferred orientation³⁵. The morphology and structure of the raw MoS₂ and exfoliated MoS₂ nanosheets were studied by Transmission electron microscope (TEM), High Resolution Transmission Electron Microscopy (HRTEM), and scanning electron microscope (SEM) characterizations. As shown in Supplementary Fig. S3a-d and S4, bending and folding of the material can be clearly observed from the basal plane of the MoS₂ NSs, which indicates that the MoS₂ NSs are very thin. The HRTEM image Supplementary Fig. S2e shows the typical hexagonal single crystal structure of MoS₂ with a distance of 0.278 nm for the (100) Mo atoms⁴¹. The selected area electron diffraction (SAED) pattern in (Supplementary Fig. S3f) implies that the MoS₂ NSs have a single crystal structure with hexagonal symmetry⁴².

Supplementary Fig. S5 exhibits the transmission of the raw MoS_2 and exfoliated MoS_2 nanosheets films which were dispersed in isopropyl alcohol (IPA). In the range of the solar spectrum, the raw MoS_2 and exfoliated MoS_2 nanosheets films are transparent with relatively high transmission (>80%). The transmission of exfoliated MoS_2 nanosheets film is slightly higher than the raw MoS_2 one, probably due to the thinner film of exfoliated MoS_2 nanosheets layer. In addition, Fig. 2 demonstrates a quite difference of 2D morphology bettween raw MoS_2 and exfoliated MoS_2 nanosheets films. The exfoliated MoS_2 nanosheets exhibit a much more smooth and homogenous morphology (Rms: 0.91) than raw MoS_2 film (Rms: 2.57), which could enhance the interfacial contact and favor the better morphology, resulting in improved performance for organic optoelectronic devices^{12,13}.

The electric and optical properties of AgNW-MoS₂ electrodes. The MoS₂ NSs were firstly introduced to adding in AgNW ink for transparent electrode. After optimization of the MoS₂ NSs component in AgNW ink, the optical and electrical properties of the AgNW and AgNW-MoS₂ electrodes fabricated by different spin speeds are summarized in Table 1 and Fig. 3a and Supplementary Fig. S6. The optical and electrical parameters of pristine AgNW electrode are basically identical with the previous reports. It is well known that a competitively transparent electrode must has a transparency of at least 90% and a sheet resistance of less than 10 Ω/sq^{43} , therefore only AgNW ink is hard to satisfy the two requirements at the same time^{44,45}. Delightfully, the modified AgNW-MoS₂ loading shows a relative low sheet resistance of 9.8 Ω sq⁻¹, and the corresponding transmittance of the electrode also exhibits a high value of 93.1% at 550 nm. It is obvious to found that the transmittance of the electrodes via different speed after the addition of MoS_2 NSs has a significant increase, mainly attributed to its high refractive index of MoS_2 NSs and a well 2D plane structure. Moreover, the environmental stability of AgNW and AgNW-MoS₂ electrodes (500 rpm spin-coating speed) were tested under 50 °C and 70 relative humidity (RH) conditions for 7 days, as shown in Supplementary Fig. S7. AgNW electrode is easy to oxidation under the atmospheric environment^{46,47}, nevertheless, the incorporation of MoS₂ NSs in AgNW electrodes show eminent oxidation resistance and moisture absorption stability.

The pristine MoS₂ NSs are n type semi-conductor materials^{34,48}, several work has been reported that MoS₂ could be changed as a p type semi-conductor materials with a relative high work function after UV-ozone plasma treatment^{34,37,48}. Thus, the properties of MoS₂ NSs before or after the UV-ozone plasma treatment were also investigated. Supplementary Fig. S8 is the X-ray photoelectron spectroscopy (XPS) profiles of n-MoS₂ NSs (w/o plasma treatment) and p-MoS₂ NSs (with plasma treatment). The Mo 3d

Electrode	Spin speed (rpm)	Thickness (nm)	Sheet resistance $(\Omega \text{ sq}^{-1})$	Transmittance (%, at 550 nm)
AgNW	500	85 ± 8	9.7 ± 0.6	84.5
	1000	61 ± 8	13.5 ± 0.5	89.6
	2000	37 ± 5	19.3 ± 1.7	90.1
	4000	25 ± 5	32.8 ± 2.1	90.8
AgNW-MoS ₂	500	71 ± 8	9.8 ± 0.5	93.1
	1000	58 ± 7	10.4 ± 0.5	93.0
	2000	32 ± 5	17.3 ± 1.1	93.1
	4000	22 ± 5	28.6 ± 2.2	93.4

Table 1. Summary of the key parameters of the AgNW and AgNW-MoS₂ electrodes fabricated at different spin speeds.

spectra of pristine MoS₂ NSs exhibit strong Mo⁴⁺3d_{5/2} and Mo⁴⁺3d_{3/2} bands at 228.5 eV and 231.8 eV, in agreement with the others work for n-MoS₂ NSs^{35,36}. However, the two strong peaks have a notable shift to 235.5 eV and 232.3 eV, respectively, which is consistent with the spectra of MoO₃⁴⁹. Therefore, it is proved that n-MoS₂ NSs can be successfully oxidized to p-type materials after plasma treatment³⁵.

The work function (*WF*) of the n and p type MoS_2 NSs was determined via UPS (Supplementary Fig. S9) and Kelvin probe (Supplementary Fig. S10). The *WF* of the n-MoS₂ NSs (pristine MoS₂ NSs) is found to be 4.3 eV, while that of the AgNW-MoS₂ electrode is 4.1 eV, which is even lower than Ag electrode (4.3 eV), meaning more efficient electron collection for cathode¹². As expected, after plasma treatment, the oxygen doping at the surface of the MoS₂ NSs film and making the *WF* of p-MoS₂ NSs upshifts to 5.0 eV, which is nearly to the *WF* of MoO₃ (5.3 eV). The substantial increased work function is close to the HOMO level of PBDTTT-C-T, as illustrated in Fig. 3b, which can avoid the hole from capturing and gathering at the interface to form a recombination center, leading to more effective hole selection and transportation for PSCs¹².

Photovoltaic performance and characterization. Besides suitable energy alignment in the devices, excellent interfacial morphology is also critical for high performance PSCs¹³. The formation of the AgNW and AgNW-MoS₂ films can also be verified by optical microscope (OM) and AFM images displayed in Supplementary Fig. S11 and Fig. 4a,b. Enormous aggregation of AgNW can be observed from large-scale OM and micro-scale AFM observation, while the AgNW-MoS₂ film develops more smooth morphology with no apparent aggregation in Supplementary Fig. S11. Intriguingly, the AFM image of AgNW-MoS₂ film demonstrates highly oriented Ag nanowires arrange uniformly in MoS₂ NSs background, with a more smooth surface than AgNW one (RMS:1.68 vs 21.62,). These oriented Ag nanowires can provide a convenient pathway for electrons transportation and collection in AgNW-MoS₂ electrodes, beneficial to the high conductive electrode. In addition, the interpenetrating network structure of AgNW and MoS₂ NSs also render a great contribution to the homogeneous surface of AgNW-MoS₂ film and high transmittance. The Ag composite electrodes could also have a great influence on active layer, as shown in Supplementary Fig. S12. Supplementary Fig. S12a shows the morphology of BHJ active layer which based on AgNW electrode with a rough surface (RMS: 6.46). However, the AgNW-MoS₂ film can provide a 2D substrate for deposition of BHJ active layer with a reduced RMS of 1.21. The different BHJ active layers and p-MoS₂ NSs have a great effect on the morphology of top Ag electrode as well. The AFM height and three-dimensional images of 30 nm Ag film deposited on bare BHJ layer (Fig. 4c,e), display a horrifically rough surface with a RMS of 20.07, leading to the poor device performance (discuss later). In contrast, the Ag film covered on p-MoS₂ layer/BHJ exhibits an uniform and smooth surface morphology in Fig. 4d,f, which means p-MoS₂ NSs can act as a soft mattress and provide an ingenious interfacial contact between BHJ layer and Ag electrode.

To determine the function of the versatile MoS_2 NSs on the optoelectronics, the PSCs based on: (PBDTTT-C-T) :(6,6)-phenyl-C70 butyric acid methyl ester (PC₇₀BM), Glass/Cathode/ETL/ PBDTTT-C-T:PC₇₀BM/HTL/Ag) were fabricated with the solution-processed 2D n-MoS₂ NSs films as ETL and solution-processed 2D p-MoS₂ NSs as HTL. The current density-voltage (*J*–*V*) characteristics of inverted cells with various buffer layers under AM 1.5G irradiation at 100 mW·cm⁻² are shown in Fig. 5a, and the related electrical parameters are summarized in Table 2. The **Device A** (Glass/ITO/ZnO/BHJ/ MoO₃/Ag(100 nm)) as the control device delivers a PCE of 7.62% with a short-circuit current density (*J*_{sc}) of 15.97 mA·cm⁻², an open circuit voltage (*V*_{oc}) of 0.76 V and a fill factor (*FF*) of 0.63. After the PSCs are modified by n-MoS₂ and p-MoS₂ interfacial layer, the *FF* of the **Device B** (Glass/ITO/n-MoS2/ BHJ/p-MoS2/Ag (100 nm)) increases to 0.70 together with improved PCE and *J*_{sc}. Since the *FF* of PSCs is normally determined by the interfacial contact and morphology of the devices, the increase in *FF* can be attributed to the smoother interfacial contact between electrodes and active layer after introducing the 2D MoS₂ NSs layers. Moreover, the relative high conductive MoS₂ layers also tend to form favorable



Figure 3. The optical property of AgNW composite electrodes and energy level alignment of devices. (a) Transmittance of spin-coated AgNW and AgNW-MoS₂ on glass deposited at different spin speeds. (b) the schematic energy diagram of the electrodes and interfacial layers involved in the STPSCs.

interfacial transportation to enhance the charge extraction and reduce the charge combination, leading to the improvement in the J_{sc} . When the ITO cathode was replaced to AgNW, all of the device parameters, e.g. PCE, J_{sc} , V_{oc} and *FF* are substantially reduced in **Device C** (Glass/AgNW/n-MoS₂/BHJ/p-MoS₂/Ag(100 nm)), resulting from the poor transmittance of cathode and reduced light absorption in active layer. Intriguingly, after introducing AgNW-MoS₂ electrode, the PCE of ITO-free device is further improved from 6.39% for **Device C** to 8.00% for **Device D** (Glass/AgNW-MoS2/n-MoS2/BHJ/p-MoS2/Ag(100 nm)), which is even comparable to the control **Device B** based on ITO cathode. The improved efficiency is mainly related to the increased J_{sc} (15.66 mA cm⁻²) and *FF* (0.66), revealing the function of AgNW-MoS₂ electrode and in good agreement with AFM measurement (Fig. 4 and Supplementary Fig. S12).

To fabricate a semi-transparent polymer solar cells (STPSCs), the top Ag electrode directly decreased Ag thickness from 100 nm to 30 nm¹¹, since the control device with 100 nm thick Ag anode was almost opaque (average visible-light transmittance: <0.05). When the thickness of top Ag layer decreases to 30 nm for a thin electrode, the total transmittance of the devices shows an average value of 0.21, as shown in Supplementary Fig. S13. Gratifyingly, the STPSCs based on the structure (Device E: Glass/ AgNW-MoS₂/n-MoS₂/BHJ/p-MoS₂/Ag(30 nm)) emerges an average PCE of 6.02%, with a J_{sc} of 12.66 m⁻², an V_{oc} of 0.75V and a FF of 0.64, when the light was illuminated from bottom side, the PCE is the highest values reported so far for ITO-free semi-transparent PSCs devices. Moreover, the PCE of the Device F with the light injected from the top Ag side also present a PCE of 2.77%. The effectiveness of Device E structure inspires us to apply it for large-scale STPSCs. The J-V curves of the STPSCs which increase the effective area from $0.06 \,\mathrm{cm}^{-2}$ to $0.65 \,\mathrm{cm}^{-2}$ were shown in Fig. 5b and Table 2. the ITO-free inverted device with a structure of Glass/AgNW-MoS $_2$ /n-MoS $_2$ /BHJ (0.65 cm⁻²)/p-MoS $_2$ /Ag (30 nm)) (Device G) still shows a PCE as high as 2.71%, with a $V_{\rm oc}$ of 0.54V, a $J_{\rm sc}$ of 9.38 mA·cm⁻², and an FF of 0.53, and parameters of the device tested from top Ag side also reach a high level (1.20, Device H). The IPCE spectra of different devices based on PBDTTT-C-T:PC70BM are shown in Fig. 5c,d, and the values of J_{sc} via the IPCE measurement well match with the ones from *I*-*V* curves. These results demonstrate the universality of this novel and large-scale STPSCs.

Discussion

To find out the reason for higher efficiency of PSCs caused by introducing the MoS_2 NSs interfacial layer, the simulation of the spatial distribution of the squared optical electric field (normalized to the incoming plane wave) for the devices with ZnO/MoO_3 and $n-MoS_2$ and $p-MoS_2$ were shown in Fig. 6a. The increased J_{sc} of these devices based on MoS_2 interfacial layers can be attributed to high refractive index of $n-MoS_2$ NSs (3.25 from 300 nm to 800 nm) leading to the higher light intensity within the active layer, as shown in Fig. 6a. From the figure we also can see that there is a dramatic improvement of the optical electric field intensity of 500 nm wavelength in active layer of the devices with $n-MoS_2/p-MoS_2$ with respect to the devices with ZnO/MoO_3 . Furthermore, the simulation of the spatial distribution of the squared optical electric field for the **Device C**, **D** and **E** were presented in Fig. 6b, where the figures of 0.36, 0.37 and 0.28 in brackets mean the relative optical electric field intensity of active layer for different devices. It is obviously found that there is no big difference between **Device C** and **D** (0.36 vs 0.37, the relative area ratio). The thickness of the top Ag electrode sharply decreases from 100 nm to 30 nm, while





Figure 4. Tapping-mode AFM images of electrodes (a) AgNW bottom electrode film height image, (b) AgNW- MoS₂ bottom electrode film height image, (scan range: $5\mu m \times 5\mu m$) height images of 30 nm thin Ag top electrode film (c) which based on BHJ layer (d) which based on BHJ/p-MoS₂ layer. Three-dimensional images of 30 nm thin Ag top electrode film (e) which based on BHJ layer (f) which based on BHJ/p-MoS₂ layer, (scan range: $3\mu m \times 3\mu m$).



Figure 5. Current (*J*)-voltage (*V*) characteristics and Incident photon-to-current efficiency (IPCE) of cells based on different devices (a) *J*-*V* result for Device A-E (b) *J*-*V* result for Device E-H, (c) IPCE result for Device A-E (d) IPCE result for Device E-H.

Device (BHJ:PBDTTT-C-T:PC ₇₁ BM)	$J_{\rm sc}$ (mA cm ⁻²)	V _{oc} (V)	FF	PCE (%)	Area (cm ⁻²)	AVT ^d (%)
A Glass/ITO/ZnO/BHJ/MoO ₃ /Ag(100 nm)	15.79 ± 0.51	0.76 ± 0.01	0.63 ± 0.03	$7.62 \pm 0.59(8.21)^a$	0.06	< 0.05
B Glass/ITO/n-MoS ₂ /BHJ/p-MoS ₂ /Ag(100 m)	15.96 ± 0.52	0.76 ± 0.01	0.70 ± 0.02	$8.43 \pm 0.62 (9.05)^a$	0.06	< 0.05
C Glass/AgNW/n-MoS ₂ /BHJ/p-MoS ₂ /Ag(100 nm)	14.79 ± 0.55	0.75 ± 0.01	0.57 ± 0.03	$6.39 \pm 0.49 (6.88)^{a}$	0.06	< 0.05
D Glass/AgNW-MoS ₂ /n-MoS ₂ /BHJ/p-MoS ₂ /Ag(100 nm)	$15.66{\pm}0.61$	0.76 ± 0.01	0.67 ± 0.03	$8.0\pm0.72(8.72)^a$	0.06	< 0.05
E Glass/AgNW-MoS ₂ /n-MoS ₂ /BHJ/p-MoS ₂ /Ag(30 nm) (0.06 cm ²)	12.66 ± 0.41	0.75 ± 0.01	0.64 ± 0.03	$6.02\pm0.53(6.55)^a$	0.06	21.18
F Glass/AgNW-MoS ₂ /n-MoS ₂ /BHJ/p-MoS ₂ /Ag(30 m) (0.06 cm ²)	7.54 ± 0.52	0.70 ± 0.01	0.52 ± 0.05	$2.77\pm0.40(3.17)^{a,b}$	0.06	21.64
G Glass/AgNW-MoS ₂ /n-MoS ₂ /BHJ/p-MoS ₂ /Ag(30 nm) (0.65 cm^2)	9.38 ± 0.48	0.54 ± 0.01	0.53 ± 0.05	$2.71 \pm 0.42 (3.13)^{a}$	0.65	20.97
$\label{eq:hardsolution} \textbf{H} \ Glass/AgNW-MoS_2/n-MoS_2/BHJ/p-MoS_2/Ag(30nm) \ (0.65m^2)$	7.43 ± 0.41	0.39 ± 0.02	0.40 ± 0.07	$1.20\pm0.31(1.51)^{a,b}$	0.65	21.11

Table 2. Performance of PSCs (PBDTTT-C-T:PC₇₀**BM) with different devices under the illumination of AM1.5G, 100 mW/cm².** ¹best device PCE. ²the devices were tested from the back side (top Ag side). ³All values represent averages from 12 devices on a single chip, and the areas were tested with an aperture. ⁴average visible-light transmittance from 300 nm to 800 nm.

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the relative optical electric field intensity of active layer slightly reduces to 0.28 for **Device E**. In addition, to gain deeper insight into the performance of the large-scale STPSCs were illuminated under different light intensity, we studied a power law dependence of J_{sc} upon illumination intensity is generally observed in PSCs and can be demonstrated as:



Figure 6. Simulated spatial distribution of the squared optical electric field $|\mathbf{E}|2$ (normalized to the incoming plane wave) for different the devices (a) Device A and Device B for 500 nm, (b) Device C, Device D and Device E for 400 nm, 500 nm and 600 nm with the same y-axes scale. (The insert figures in brackets mean the relative area ratio in active layer. All the simulation were calculated via Matlab program, the Matlab files were provided from McGehee group, Center for Advanced Molecular Photovoltaics, Stanford University⁵³).

$$J_{sc} \propto I^{\alpha}$$
 (1)

where *I* is the light intensity and α is the exponential factor^{50,51}. In Supplementary Fig. S14, the data are plotted on a log–log scale and fitted to a power law using eq 1: $\alpha = 0.955$, higher than the reference device⁵², indicating that bimolecular recombination could be less in the cell with the MoS₂ NSs transport layer. It is also found that the STPSCs based on **Device E** structure behave an excellent performance under weaker illumination, as shown in Supplementary Fig. S14.

To evaluate the transparency perception of the above STPSCs in human eyes, the tristimulus value (X, Y, Z) and the color coordinates (x, y) were calculated from the transmission spectra while the incident light source was replaced with the AM 1.5 spectrum. The color coordinates of the studied STPSCs with different thickness of Ag on CIE 1931 chromaticity diagram are illustrated in Fig. 7a,b. The corresponding coordinates of the ITO, AM 1.5 point, AgNW-MoS₂ and **Device E** are (0.350, 0.326), (0.359, 0.355),



Figure 7. Color properties of the electrode (a) The representation of color coordinate of the studied different electrodes and **Device E** on CIE 1931xyY chromaticity diagram. (b) the picure of **Device E**. Two digital photographs (c) taken through and (d) not taken through the STPSCs based on **Device E** structure. (The photographs were taken by Xiaotian Hu).

(0.333, 0.347), and (0.313, 0.329), respectively. All these films show very nice transparency color perception that is close to AM 1.5 point. Fig. 7c,d show two digital photographs taken through and not taken through the STPSCs based on **Device E** structure. There is no big aberration between the two pictures, which means it could provide an opportunity for these STPSCs applying in power-generating windows. In order to further to prove the possibility, a power-generating glass was fabricated which is made up of eight 0.65 cm² **Devices E**, as shown in Fig. 8. Fig. 8a delivers the power-generating glass illuminated under 1.5 AM solar condition and the LED chip basically can achieve the rated power. Miraculously, the solar cell glass presents an excellent performance out of doors condition (Test time: October 25th, 2014 PM 13:00 and Environmental conditions: Nanchang City, Jiangxi Province, P.R China, 25 °C Sunshine and 43% HR). The photocurrent of power-generating glass is enough to light the LED chip regardless of illumination direction from the front or back side. From these applied tests, it can be sure to say, this kind of power-generating glass provide a great opportunity for building integrated PV (BIPV).

Conclusions

In summary, we have demonstrated a versatile MoS_2 nanosheets with well 2D structure that were introduced into Ag nanowires transparent electrode, anode and cathode interfacial layers in PSCs. High







Test time:

October 25(th), 2014. PM 13:00

Environmental conditions:

Nanchang City, Jiangxi Province,

P.R. China, 25°C Sunshine and 43% HR.

Figure 8. Power-generating glass images. Power-generating glass was fabricated via combine eight 0.65 cm^2 deivces E, the devices were encapsulated by UV curing glue. To test the power of this semi-transparent solar cell glass, a 5 mW LED chip was put in circuit. (a) the power-generating glass was illuminated under 1.5 AM solar condition (b) the power-generating glass was tested from back side out of doors (c) the power-generating glass was tested from front side out of doors.

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performance ITO-free PSCs based on Glass/AgNW-MoS $_2$ /n-MoS $_2$ /BHJ/p-MoS $_2$ /Ag(100 nm) achieve a high PCE of 8.00%.

Furthermore, a novel STPSC is incorporating the MoS_2 NSs in interfacial layers and AgNW transparent electrode with admirable interfacial contact and superior light-harvesting abilities. The STPSCs exhibited good PCE and AVT with neutral color perception close to that of standard sunlight, which is the highest performance for ITO-free semi-transparent PSCs devices reported up to now. The delightful results demonstrate STPSCs do have great values in power-generating windows of building integrated PV in near future.

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Acknowledgements

This work was financially supported by the National Science Fund for Distinguished Young Scholars (51425304), National Natural Science Foundation of China (51273088, 51263016 and 51473075), and National Basic Research Program of China (973 Program 2014CB260409).

Author Contributions

X.H., L.C. and Y.C. designed the device and experiments. Y.Z., L.H., L.T and B.X. supervised the experimental measurements. X.H. wrote the manuscript together with L.C. and Y.C. All authors contributed to scientific discussions of the manuscript. X.H. and L.C. contributed equally to this work.

Additional Information

Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Hu, X. *et al.* Versatile MoS₂ Nanosheets in ITO-Free and Semi-transparent Polymer Power-generating Glass. *Sci. Rep.* **5**, 12161; doi: 10.1038/srep12161 (2015).

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