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# ArCH(OMe)<sub>2</sub> - a Pt<sup>IV</sup>-catalyst originator for diverse annulation catalysis

SUBJECT AREAS:

METHODOLOGY

ORGANOCATALYSIS

CATALYST SYNTHESIS

CROSS-COUPLING REACTIONS

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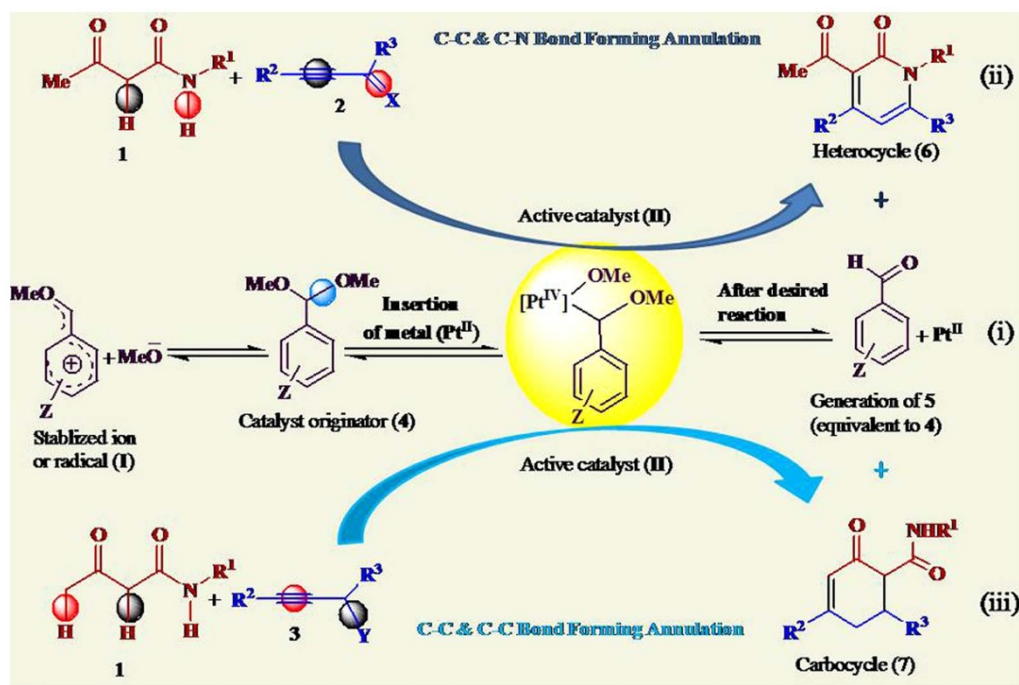
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Received  
3 July 2013Accepted  
30 September 2013Published  
18 October 2013Correspondence and  
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We discover an important property of a small molecule ArCH(OMe)<sub>2</sub> which transforms catalytically inactive Pt<sup>II</sup>Br<sub>2</sub> procatalyst in situ to an powerful catalyst Pt<sup>IV</sup>-species for diverse annulation reaction. The powerful catalytic system enables selective activation of C<sub>2</sub>-H/N-H and C<sub>2</sub>-H/C<sub>4</sub>-H of acetoacetanilide and C=O/C≡C of substituted butyne-1,2-dione for C-C/C-N, C-C/C-C and C-O/C-O bond-forming inter- and intramolecular annulation towards direct syntheses of functionalised 2-pyridones, cyclohexenones and 3(2*H*)-furanones respectively. In contrast to the common ligand, herein highly labile C-OMe bond of ArCH(OMe)<sub>2</sub> is expected to react with PtBr<sub>2</sub> towards generation of the high-valent active catalyst. Unlike catalyst promoter or initiator, the reaction does not occur with PtBr<sub>2</sub> in the absence of ArCH(OMe)<sub>2</sub>. In situ generation of Pt<sup>IV</sup>-species and -OMe fragment of ArCH(OMe)<sub>2</sub> were confirmed from the UV-vis characteristic peaks about 260 nm and trapping of -OMe group respectively. These observations provide new prospects and perspectives in catalysis for innovative catalyst design.

Catalysis<sup>1-12</sup> is a frontier research field in the chemical sciences and their allied branches as catalysis plays a pivotal role in the synthesis of almost every chemical, pharmaceutical, agrochemical and material required for the rapid development of mankind, especially our highly-demanding modern society. A catalyst promoter or initiator<sup>9-12</sup> is a cocatalyst which is crucial for significant improvement of reaction rate and/or selectivity<sup>13</sup> in some catalysis reactions. Herein, we introduce a new concept of catalyst originator using readily available and inexpensive small molecule ArCH(OMe)<sub>2</sub> as the first example. Unlike promoter or initiator, catalyst originator reacts with a completely inactive procatalyst of a reaction and eventually alters oxidation state of the procatalyst which is crucial to empower outstanding catalytic activity for the initiation, execution and completion of the desired transformation with high synthetic efficiency<sup>13</sup>. Development of highly efficient catalytic processes for annulation reactions<sup>14</sup> to natural and unnatural functional molecules is the central focus in modern organic synthesis. Catalytic selective activation<sup>4,5,15-17</sup> and transformation of C-H, N-H, C-O and  $\pi$ -bonds can be utilized for highly selective annulation reaction. Development of new benign and robust strategies is desirable such as discovery of new atom-economical domino reactions<sup>18-20</sup> towards direct synthesis of several target molecules using an efficient catalytic system. Acetoacetanilide (**1**, Figure 1) is a commercially available inexpensive laboratory reagent that has found very limited application in synthetic organic chemistry, including synthesis of the lepidone, martinellid acid derivative<sup>21</sup>, unsymmetrical urea<sup>22</sup> and our recently reported FeCl<sub>3</sub>·6H<sub>2</sub>O-catalysed diastereoselective construction of *trans*-1,2,3,4-tetrahydro-2-pyridones<sup>7</sup>. Furthermore, the three important reactive centres (C<sub>2</sub>-H, C<sub>4</sub>-H and N-H) of acetoacetanilide can be exploited in a selective and cascade fashion for annulation with designed propargyl compounds (**2**, **3** etc.). However, the use of halide precursors (**2** and **3**; X and Y, eq. ii and iii) will generate a considerable amount of byproduct waste in the reaction. Thus, the use of keto (X = O) and alcohol (Y = OH) precursors are highly desirable because the waste can be removed as an environmentally benign water molecule. We envisioned that breaking of a C-O bond in an aromatic aldehyde dimethyl acetal (**4**, eq. I, Figure 1) is possible through formation of a stabilized species **I**. So, ArCH(OMe)<sub>2</sub> may be incorporated into an inactive procatalyst (Pt<sup>II</sup>) by oxidative insertion to the C-O bond, leading to the in situ generation of a stabilised organometallic (**II**) of higher oxidation state (e.g., Pt<sup>IV</sup>) which may be efficiently used in catalysis for development of a robust cyclization process.

Gratifyingly, we found in situ generated Pt<sup>IV</sup>-species as an efficient catalyst for new organic transformations such as intermolecular annulation reactions (eq. ii, iii). This approach involving a catalysis originator is especially attractive from a synthetic perspective because it performs selective activation, functionalisation and annulation involving C-C and C-N bond formation to valuable 2-pyridone<sup>23-25</sup> (**6**, eq. ii), C-C and C-C coupling to cyclohexenone<sup>26-28</sup> (**7**, eq. iii) or C-O and C-O joining to afford the 3(2*H*)-furanone<sup>29,30</sup> framework, which are available in a number of bioactive natural products, such as the antibiotic tenellin<sup>23</sup>, the antimalarial longirostrerone C<sup>26</sup> and the antibiotic gregatin analogues<sup>29,30</sup>, respectively. We have recently initiated a national research program,



**Figure 1** | Proposed intermolecular annulation strategy by suggested  $\text{Pt}^{\text{IV}}$ -active catalyst with  $\text{ArCH}(\text{OMe})_2$ .

OSDD<sup>31</sup>, towards the easy access of these functionalised molecules through a simple, step-economical and cost-effective process, and these molecules will be developed as inexpensive drugs for millions of patients suffering from deadly malaria<sup>32</sup> and tuberculosis<sup>33</sup>. After completion of a desired reaction, the  $\text{Pt}^{\text{II}}$ -procatalyst will be regenerated for the next cycle (eq. i) along with  $\text{ArCH}(\text{OMe})_2$  equivalent aromatic aldehyde (5). The aldehyde may be converted to 4 through a simple acetal formation.

## Results

**1. Development of the catalyst originator for the intermolecular annulation reaction.** We initiated the domino<sup>18–20</sup> annulation reaction between *N*-(4-bromophenyl)-3-oxo-butylamide (1a:  $\text{R}^1 = 4\text{-BrC}_6\text{H}_4$ ) with 1,3-diphenylpropynone (2a:  $\text{X} = \text{O}$ ;  $\text{R}^2 = \text{R}^3 = \text{C}_6\text{H}_5$ ) in the presence of benzaldehyde dimethyl acetal (4a) and several transition metals ( $\text{NiBr}_2$ ,  $\text{PdCl}_2$ ,  $\text{AuCl}_3$ ,  $\text{RuCl}_3$  etc.) and rare-earth metals ( $\text{La}(\text{OTf})_3$ ,  $\text{CeCl}_3$ ,  $\text{Yb}(\text{OTf})_3$  etc.) as prospective

procatalysts (entry 1, Table 1). Unfortunately, all attempts were unsuccessful, even under heating conditions. Compared to the widespread application of platinum compounds in catalysis<sup>34</sup>,  $\text{PtBr}_2$ -catalysed reactions are limited to only a few. For instance, this catalyst was utilised in the hydroamination of olefin<sup>35</sup>, in intramolecular enyne metathesis<sup>36</sup> and in Markownikoff's hydroarylation of terminal alkynes<sup>37</sup>. Gratifyingly, treatment of  $\text{PtBr}_2$  (3 mol%) at ambient temperature afforded the desired heterocycle 3-acetyl-1-(4-bromophenyl)-4,6-diphenyl-1*H*-pyridin-2-one (6a) with moderate yield (60%, entry 2).  $\text{PtBr}_2$  is insoluble in acetonitrile solvent. Surprisingly, it dissolved in the reaction mixture on addition of  $\text{PhCH}(\text{OMe})_2$  and transformed the colourless reaction mixture into the reddish brown solution. On the other hand, formation of even traces of desired product (6a) was not observed in the absence of the catalyst originator, which was confirmed by monitoring the reaction using TLC, HPLC and NMR spectroscopy. The reaction was optimised (entries 3–6) to improve the yield (86%), reaction rate (9 h) and procatalyst loading (2 mol%, entry

**Table 1** | Optimisation of the intermolecular annulation reaction

Entry	Catalyst (mol%)	Catalysis initiator (4)	Solvent, time (h)	Yield (%)
1	Other catalysts	$\text{PhCH}(\text{OMe})_2$ ( <b>4a</b> )	$\text{CH}_2\text{Cl}_2$ , 24	-
2	$\text{PtBr}_2$ (3)	$\text{PhCH}(\text{OMe})_2$ ( <b>4a</b> )	$\text{CH}_2\text{Cl}_2$ , 24	60
3	$\text{PtBr}_2$ (3)	$\text{PhCH}(\text{OMe})_2$ ( <b>4a</b> )	$\text{MeCN}$ , 8	87
4	$\text{PtBr}_2$ (3)	$\text{PhCH}(\text{OMe})_2$ ( <b>4a</b> )	$\text{DMF}$ , 24	-
<b>5</b>	<b><math>\text{PtBr}_2</math> (2)</b>	<b><math>\text{PhCH}(\text{OMe})_2</math> (<b>4a</b>)</b>	<b><math>\text{MeCN}</math>, 9</b>	<b>86</b>
6	$\text{PtBr}_2$ (1)	$\text{PhCH}(\text{OMe})_2$ ( <b>4a</b> )	$\text{MeCN}$ , 9	34
7	$\text{PtBr}_2$ (2)	$\text{MgSO}_4$ or Molecular sieve	$\text{MeCN}$ , 24	-
8	$\text{PtBr}_2$ (2)	4-MeOC <sub>6</sub> H <sub>4</sub> CH(OMe) <sub>2</sub> ( <b>4b</b> )	$\text{MeCN}$ , 9	82
9	$\text{PtBr}_2$ (2)	4-O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub> CH(OMe) <sub>2</sub> ( <b>4c</b> )	$\text{MeCN}$ , 9	50
10	$\text{PtBr}_2$ (2)	HCH(OMe) <sub>2</sub> ( <b>4d</b> )	$\text{MeCN}$ , 24	-
11	$\text{PtBr}_2$ (2)	Me <sub>2</sub> C(OMe) <sub>2</sub> ( <b>4e</b> )	$\text{MeCN}$ , 24	-

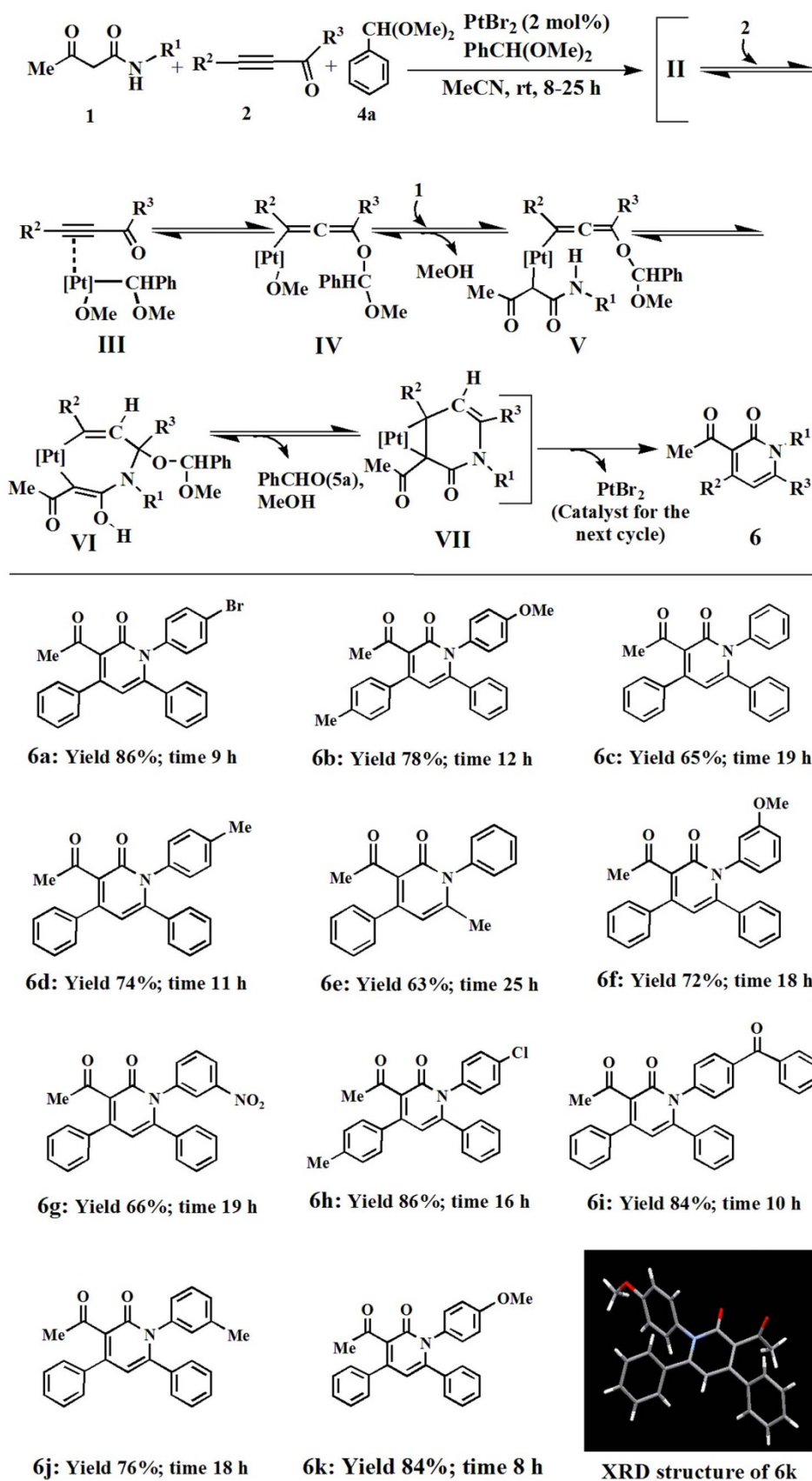


Figure 2 | Bimolecular annulation process to 2-pyridones.



5). However, the role of  $\text{PhCH}(\text{OMe})_2$  is not a desiccant for the reaction which was verified using activated magnesium sulfate and molecular sieve (entry 7). The aromatic dimethyl acetal (**4**) possessing an activated aromatic nucleus (**4b**, entry 8) provided a comparable yield (82%) where as the deactivated one (**4c**, entry 9) drastically reduced the yield (50%). As expected, the reaction did not occur with aliphatic acetal and ketal without assistance from the aromatic ring (entries 10,11) which is essential for breaking the C-O bond and stabilising the active  $\text{Pt}^{\text{IV}}$ -catalyst.

**2. N-C and C-C coupling to functionalized 2-pyridones.** The reported synthesis of 2-pyridones includes multistep strategies, microwave and metathesis protocols, the one-pot Blaise reaction, direct synthesis involving the oxidative annulation of  $\alpha,\beta$ -unsaturated amide and ketone, and our recently reported ring opening of chromone aldehydes with tandem cascade cyclisation<sup>7,38</sup>. Substituted 2-pyridones are potential candidates for antihepatitis B, antitumor, human rhinovirus (HRV) 3C protease (3CP) inhibitor and noncompetitive antagonist related to epilepsy<sup>39,40</sup>. We developed the direct construction of substituted 2-pyridone compounds (**6a-k**, Figure 2) from readily available chemicals acetoacetanilide (**1**) and 1,3-disubstituted propynone (**2**) at ambient temperature under neutral and benign reaction conditions. The substrate scope of this reaction

revealed that the electron-donating substituent on the aryl group of acetoacetanilide derivatives significantly improved the reaction rate and yield relative to the precursor with an electron-withdrawing substituent (**6g**) or no-substituent (**6c** and **6e**). Aromatic and aliphatic substituents (**6e**) were tolerated in this highly regioselective bimolecular reaction. The annulation reaction was very selective to non-terminal precursor **2** as it was unsuccessful with terminal propynone precursors (**2**,  $\text{R}^2 = \text{H}$ ). Exact mechanism of the reaction is unknown to us. Herein,  $\text{Pt}^{\text{IV}}$ -active catalyst (**II**) is expected to bind with the triple bond (**III**) and carbonyl oxygen (**IV**) of **2** and subsequently N-C and C-C coupled cyclization (**V-VII**) with acetoacetanilide (**1**) in a cascade fashion led to removal of  $\text{PhCHO}$  (**5a**) and  $\text{MeOH}$ . The putative fused-intermediate **VII** immediately underwent reductive elimination to produce the desired heterocycle (**6**) along with the regenerated  $\text{PtBr}_2$  precatalyst.

**3. Dual C-C coupling to functionalized cyclohexenones.** In recent studies, cyclohexenone compounds were utilised as versatile synthons for the total synthesis of the antimalarial (+)-artemisinin<sup>41</sup>, the antibiotic platencin<sup>42</sup>, the important skeleton of the antimicrobial alkaloid (+)-2-oxo-agelasidine **C**<sup>43</sup> and for biosynthetic intermediates<sup>44</sup>. However, reports on synthesis of functionalised cyclohexenone are very limited<sup>45,46</sup>. The selective activation of  $\text{C}_2\text{-H}$  and  $\text{C}_4\text{-H}$

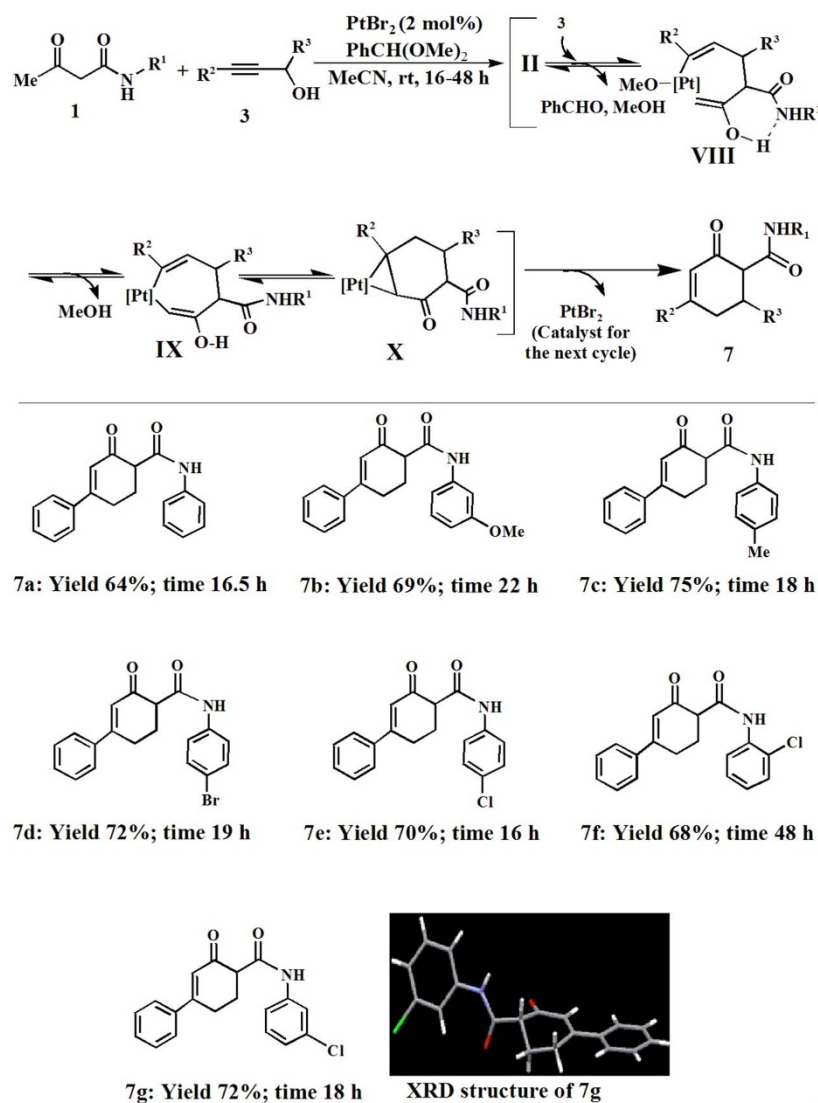


Figure 3 | Regioselective cascade annulation to cyclohexenones.



of acetoacetanilide derivatives and double C-C coupling with an appropriate propargyl moiety (eq. iii, Figure 1) led to the formation of functionalised cyclohexenone (7). However, this is a challenging task because C-C/C-N coupling is a more facile process (Figure 2) in the presence of a -NH- group. To overcome this challenge, we changed the precursor alkyne ketone (2) to its reduced form (3: Y = OH). We hypothesized that first C-C coupling (Figure 3) between 3 and 1 (C<sub>2</sub>-H) occurred with II by replacing the hydroxyl group of 3 (VIII) which expelled PhCHO (5a) and methanol. Second C-C bond formation (X) with CH<sub>3</sub>CO- group proceeded through formation of intermediate IX. The cyclic intermediate X subsequently underwent reductive elimination of the PtBr<sub>2</sub> procatalyst with formation of the desired product 2-oxo-4-phenylcyclohex-3-enecarboxylic acid aryl amides (7). Interestingly, the annulation reaction to compound 7 occurred under similar reaction conditions (entry 5, Table 1) and was very selective, and the corresponding N-C coupling 2-pyridones (6) were not observed in the post-reaction mixture. Herein, no product was observed using 1,3-disubstituted propargyl alcohol precursor (R<sup>3</sup> = Ph, VIII), which might be due to the considerable steric hindrance and/or electronic repulsion that appeared during reaction of propargyl alcohol (3) and acetoacetanilide with Pt<sup>IV</sup>-active catalyst (II, Figure 1). The reaction was also very selective to non-terminal propargyl alcohol because the reaction was completely blocked when investigated using terminal alkynyl alcohols, such as propargyl alcohol.

**4. Dual C-O coupling to substituted 3(2H)-furanones.** The 3(2H)-furanone (8) compound is a human tyrosinase inhibitor<sup>47</sup>, the aroma component of soy sauce<sup>48</sup> and is used as a valuable synthon for the total synthesis of the renal cancer cell lines inhibitor (-)-Englerin A<sup>49</sup>. The widespread application of the heterocyclic scaffold and the availability of only a few synthetic methods in the literature

prompted us to establish an easy synthetic method. Gratifyingly, double O-C coupling intramolecular cyclization of non-terminal butyne-1,2-dione (5) with the PhCH(OMe)<sub>2</sub> was observed under the similar reaction conditions to afford 3(2H)-furanone (8, Figure 4). In fact, this developed method is the second approach for the direct synthesis of 2-alkoxy-3(2H)-furanones following the AuCl<sub>3</sub>-catalysed annulation of 3-oxo-butyne analogues reported by Liu<sup>50</sup> and coworkers. From a synthetic perspective, this reaction is straightforward, high yielding (85–95%), tolerant to double bonds (8d) and aromatic rings, and can also directly synthesise a complex compound bearing three heterocycles (8e). It is proposed that the intramolecular cyclisation occurred involving formation of C=O and triple bond-coordinated Pt<sup>IV</sup>-activated intermediate XII which subsequently transformed to 8 by the migration of -OMe, as well as the formation of O-C bond involving the conversion of C≡C to C=C. Interestingly, the Pt<sup>IV</sup>-species was so selective for binding to precursor 5 that even in presence of acetoacetanilide (1), it did not form the corresponding 2-pyridone (6) by annulation with the R<sup>2</sup>-C≡C-CO- moiety (Figure 2).

## Discussion

We have demonstrated that aromatic aldehyde dimethyl acetal is a keen catalyst originator for transforming the catalytically inactive transition metal procatalyst PtBr<sub>2</sub> to an active catalyst Pt<sup>IV</sup>-species for both inter- and intramolecular annulation to several valuable heterocycles and carbocycles. Organic ligands greatly influence the catalytic activity of a metallic compounds<sup>5</sup>. The annulation reaction was examined using different bidentate ligands and conventional additives, such as ethylene glycol dimethyl ether, dimethyl tartarate, 1,2-diphenyl ethylene glycol, catechol, DPPE, COD, α-pinene, norbornadiene, etc., and all attempts were unsuccessful to provide compound 6. These observations strongly supported that the role of ArCH(OMe)<sub>2</sub> was not simple as a conventional ligand in the remarkable catalysis

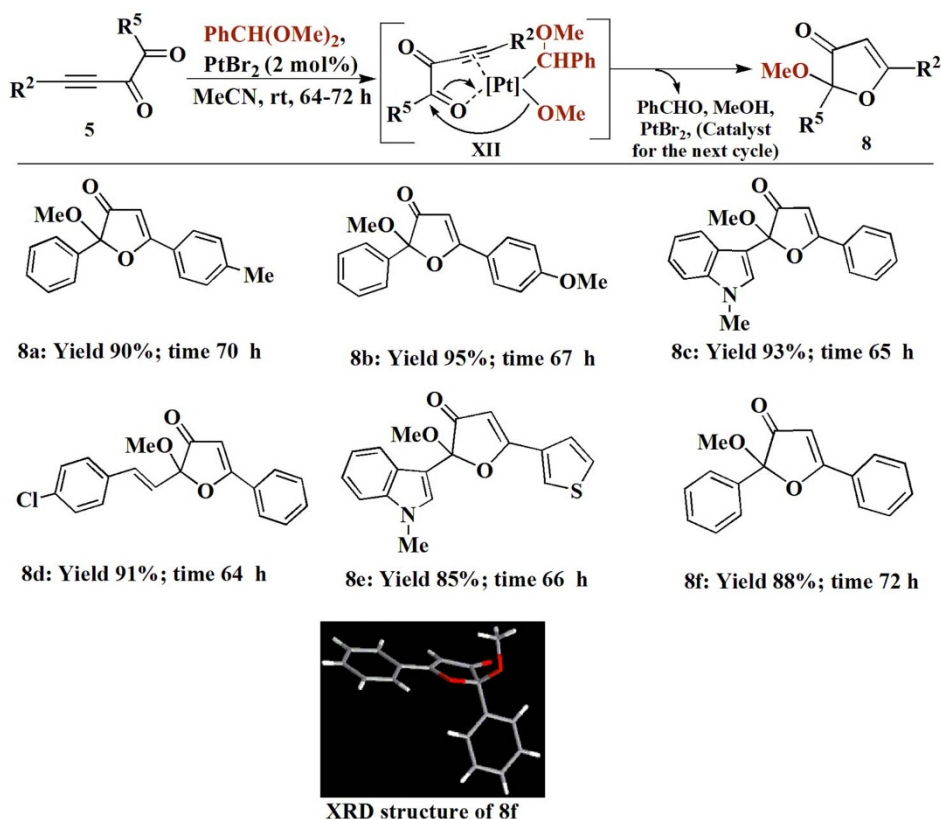


Figure 4 | Synthesis of 3(2H)-furanone with trapping of the OMe group.

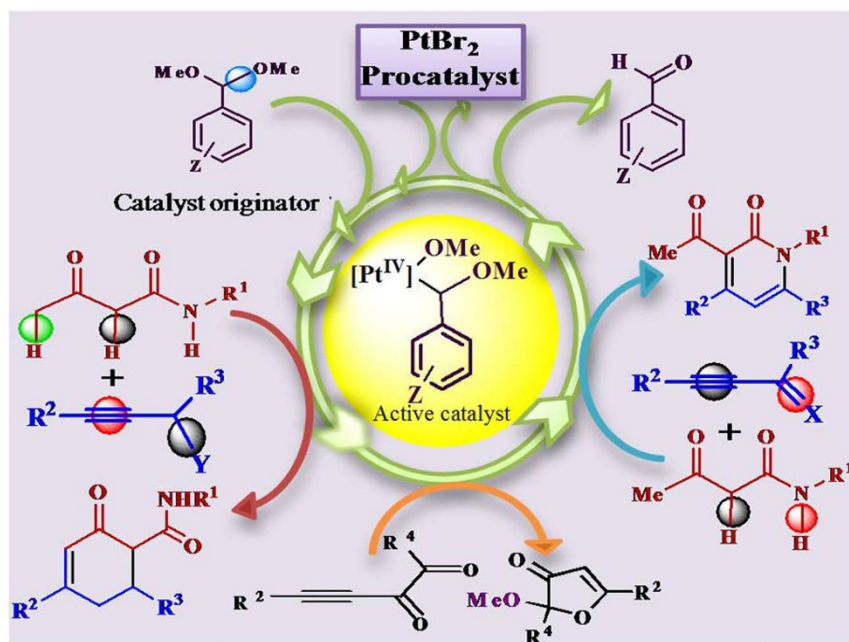


Figure 5 | Catalyst originator mediated diverse annulation reactions.

approach. To the best of our knowledge, there is no such example in the literature of using a catalyst originator to a procatalyst that displays outstanding catalytic activity. To understand the involvement of  $\text{Pt}^{\text{IV}}$ -species in this process, reactions were performed separately with  $\text{Pt}^{\text{IV}}$ -compounds such as  $\text{PtCl}_4$ ,  $\text{H}_2\text{PtCl}_6$ ,  $\text{PtCl}_4\text{-4a}$  and  $\text{H}_2\text{PtCl}_6\text{-4a}$ . Unfortunately, the formation of the desired product (**6a**) was not observed in these reaction mixtures. UV-vis spectroscopic analyses of  $\text{PtBr}_2$  and  $\text{PhCH}(\text{OMe})_2$  in acetonitrile and also the reaction mixture (entry 5, Table 1) revealed the presence of  $\text{Pt}^{\text{IV}}$ -species through the characteristic absorbance<sup>51</sup> band that appeared at 259.84 and 260.02 nm respectively (supplementary information). These observations clearly indicated that only **4a**-modified  $\text{Pt}^{\text{IV}}$  was the active catalyst for the robust annulation processes. The formation of  $\text{PhCHO}$  (**5a**) was observed due to reaction of catalyst originator  $\text{PhCH}(\text{OMe})_2$  and in situ generated water during the course of the annulation process. Compound **5a** was isolated from the post-reaction mixture and characterized. The formation of  $\text{MeO-}/\text{MeOH}$  was also verified by trapping it in an intramolecular cyclisation of disubstituted but-3-yne-1,2-diketone (**5**) to afford 2-methoxy-3(2H)-furanone derivative (**8**, Figure 4). Interestingly, the activated  $\text{Pt}^{\text{IV}}$ -species was so selective for binding to precursor **3** (Figure 3) and **5** (Figure 4) that even in presence of acetoacetanilide (**1**), it did not form the corresponding 2-pyridone (**6**). This inexpensive and readily available organic catalyst originator installed novel catalytic power to procatalyst  $\text{PtBr}_2$  for activation of C-H, N-H, C-O and  $\pi$ -bonds towards selective C-C/C-N, C-C/C-C and C-O/C-O coupled annulation to achieve the direct synthesis of ubiquitous carbocycles and heterocycles. The robust synthetic protocol was developed under benign and neutral reaction conditions to obtain highly functionalised 2-pyridones (**6**), cyclohexenones (**7**) and 3(2H)-furanones (**8**) in excellent yield and with very low catalytic loading using the common laboratory reagent acetoacetanilide, non-halogenated precursor  $\alpha$ -ketoalkynes and propargyl alcohol, and environmentally safe water was generated as a byproduct (Figure 5). The transformation of the catalyst originator to its corresponding aldehyde, trapping of the fragmented -OMe group and the in situ generation of the  $\text{Pt}^{\text{IV}}$ -active catalyst were experimentally confirmed. We anticipate that this new concept in catalysis will find immense application in synthetic organic chemistry towards innovative catalyst design, the development of new catalyst originator,

procatalysts, and novel reactions; and the newly designed catalysts will dominate as a work-horse in the facile synthesis of novel functional molecules with high synthetic efficiency.

## Methods

A solution of  $\beta$ -ketoanilide (**1**, 1.0 mmol), benzaldehyde dimethyl acetal (**4a**, 152 mg, 1.0 mmol) and 1,3-disubstituted 1-propynone (**2**, 1.0 mmol) in acetonitrile (10 mL) was stirred at 0°C. Platinum (II) bromide (**7** mg, 2 mol%) was added, and the mixture was stirred at ambient temperature. The progress of the reaction was monitored by TLC, and the reaction was complete within 8–25 h depending on the use of the substrates. The post-reaction mixture was concentrated in a rotary evaporator under reduced pressure at room temperature and the residue was dissolved in ethyl acetate (50 mL). The organic layer was washed with distilled water ( $3 \times 10$  mL), dried over activated sodium sulphate and concentrated in a rotary evaporator. Thus, the reaction with *N*-(4-bromophenyl)-3-oxo-butyramide (**1a**, 256 mg, 1.0 mmol) and 1,3-diphenylpropynone (**2a**, 206 mg, 1.0 mmol) afforded 3-acetyl-1-(4-bromophenyl)-4,6-diphenyl-1*H*-pyridin-2-one (**6a**) in a yield of 86% (380 mg, 0.86 mmol) after purification by column chromatography on basic alumina (100–200 mesh) with 8% ethyl acetate in hexane as an eluent. All of the new 2-pyridone compounds (**6a–k**) were characterised using NMR ( $^1\text{H}$ ,  $^{13}\text{C}$  and DEPT), FT-IR and mass (HR-MS) spectroscopy and single crystal XRD analyses. Functionalised cyclohexenones (**7**) and 2-methoxy-3(2H)-furanones (**8**) were also synthesised under similar reaction conditions and fully characterised (supplementary information). The structures of all the new compounds (**6–8**) were elucidated by performing NMR, FT-IR and ESI-MS spectroscopic measurements and single crystal XRD-analyses<sup>52</sup> of **6k**, **7g** and **8f**.

1. Rothenberg, G. *Catalysis* (Wiley/VHC, Weinheim, 2008).
2. Berzelius, J. J. Årsberättelsen om framsteg i fysik och kemi. *Royal Swedish Academy of Sciences* (1835).
3. Peplow, M. *Catalysis: The accelerator. Nature* **495**, S10–S11 (2013).
4. Davies, H. M. L. & Manning, J. R. Catalytic C-H functionalization by metal carbenoid and nitrenoid insertion. *Nature* **451**, 417–424 (2008).
5. Hartwig, J. F. Carbon-heteroatom bond formation catalysed by organometallic complexes. *Nature* **455**, 314–322 (2008).
6. Dong, G., Teo, P., Wickens, Z. K. & Grubbs, R. H. Primary alcohols from terminal olefins: formal anti-Markovnikov hydration via triple relay catalysis. *Science* **333**, 1609–1612 (2011).
7. Sengupta, T., Gayen, K. S., Pandit, P. & Maiti, D. K.  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ -catalyzed intermolecular-cascade cyclization of acetoacetanilide: aldehyde-tuned synthesis to valuable 2-pyridone analogues. *Chem. Eur. J.* **18**, 1905–1909 (2012).
8. Krautwald, S., Sarlah, D., Schafroth, M. A. & Carreira, E. M. Enantio- and diastereodivergent dual catalysis:  $\alpha$ -allylation of branched aldehydes. *Science* **340**, 1065–1068 (2013).
9. Jackson, S. D., Munro, S., Colman, P. & Lennon, D. Effect of toluene- $d_8$  on the hydrogenation of 1,3-hexadiene over a Pd/silica catalysts promoter and poison. *Langmuir* **16**, 6519–6526 (2000).



10. Campos, K. R. *et al.* Controlled semihydrogenation of aminoalkynes using ethylenediamine as a poison of Lindlar's catalyst. *J. Org. Chem.* **66**, 3634–3635 (2001).
11. Dronavajjala, K. D. *et al.* A simple technique to grow polymer brushes using in situ surface ligation of an organometallic initiator. *J. Am. Chem. Soc.* **128**, 13040–13041 (2006).
12. Xiaofei, L., Haoxi, J., Guiming, L. & Minhua, Z. Investigation of Cu-based catalyst for direct synthesis of ethyl acetate from ethanol: improvement of thermal stability of Cu–Cr–Zr composite oxide catalyst by addition of Mn promoter. *Ind. Eng. Chem. Res.* **51**, 8974–8978 (2012).
13. Dondoni, A. & Massi, A. Design and synthesis of new classes of heterocyclic C-glycoconjugates and carbon-linked sugar and heterocyclic amino acids by asymmetric multicomponent reactions (AMCRs). *Acc. Chem. Res.* **39**, 451–463 (2006).
14. Ma, S. *Handbook of Cyclization Reactions*, (Wiley/VHC, 2009).
15. Ritleng, V., Sirlin, C. & Pfeffer, M. Ru-, Rh-, and Pd-catalyzed C–C bond formation involving C–H activation and addition on unsaturated substrates: reactions and mechanistic aspects. *Chem. Rev.* **102**, 1731–1769 (2002).
16. Saima, Y., Khamurai, S., Gayen, K. S., Pandit, P. & Maiti, D. K. Efficient catalytic cyclizations of three and two imine assemblies: direct access to tetrahydroimidazo[1,5-c]imidazol-7-ones and imidazoles. *Chem. Commun.* **48**, 6601–6603 (2012).
17. Arockiam, P. B., Bruneau, C. & Dixneuf, P. H. Ruthenium(II)-catalyzed C–H bond activation and functionalization. *Chem. Rev.* **112**, 5879–5918 (2012).
18. Tietze, L. F., Brasche, G. & Gericke, K. *Domino reactions in organic synthesis* (Wiley-VCH: Weinheim, Germany, 2006).
19. Dhara, D. *et al.* CeCl<sub>3</sub>·7H<sub>2</sub>O catalyzed C–C and C–N bond-forming cascade cyclization with subsequent side-chain functionalization and rearrangement: A domino approach to pentasubstituted pyrroles analogues. *J. Org. Chem.* **77**, 10441–10449 (2012).
20. Pellissier, H. Stereocontrolled domino reactions. *Chem. Rev.* **113**, 442–524 (2013).
21. Zhang, Z., Zhang, Q., Yan, Z. & Liu, Q. One-step synthesis of the tricyclic core of martinellidic acid from 2-(cyanomethyl)-3-oxo-N-arylbutanamides. *J. Org. Chem.* **72**, 9808–9810 (2007).
22. Wei, Y. *et al.* Acetoacetanilides as masked isocyanates: facile and efficient synthesis of unsymmetrically substituted ureas. *Org. Lett.* **12**, 4220–4223 (2010).
23. Halo, L. M. *et al.* Late stage oxidations during the biosynthesis of the 2-pyridone tenellin in the entomopathogenic fungus *Beauveria bassiana*. *J. Am. Chem. Soc.* **130**, 17988–17996 (2008).
24. Peng, X. *et al.* Cerebrosides and 2-pyridone alkaloids from the halotolerant fungus penicillium chrysogenum grown in a hypersaline Medium. *J. Nat. Prod.* **74**, 1298–1302 (2011).
25. Kumarihamy, M. *et al.* Antiprotozoal and antimicrobial compounds from the plant pathogen septoria pistaciarum. *J. Nat. Prod.* **75**, 883–889 (2012).
26. Panthama, N., Kanokmedhakul, S., Kanokmedhakul, K. & Soyong, K. Cytotoxic and antimalarial azaphilones from chaetomium longirostre. *J. Nat. Prod.* **74**, 2395–2399 (2011).
27. Yang, F. *et al.* Antimicrobial metabolites from the Paracel Islands sponge agelas mauritiana. *J. Nat. Prod.* **75**, 774–778 (2012).
28. Long, C. *et al.* Proteasome inhibitors from *neoboutonia melleri*. *J. Nat. Prod.* **2012**, 75, 34–47.
29. Taylor, S. L., Peterson, R. E. & Gray, L. E. Isolation of gregatin A from phialophora gregata by preparative high-pressure liquid chromatography. *Appl. Environ. Microbiol.* **50**, 1328–1329 (1985).
30. Burghart-Stoll, H. & Brückner, R. A serendipitous synthesis of (+)-gregatin B, second structure revisions of the aspertetrinins, gregatins, and graminin A, structure revision of the penicilliois. *Org. Lett.* **13**, 2730–2733 (2011).
31. Open Source Drug Discovery (OSDD). <http://www.osdd.net>. (Date of access 24.09.2013).
32. Peplow, M. Malaria drug made in yeast causes market ferment. *Nature* **494**, 160–161 (2013).
33. Zumla, A., Hafner, R., Lienhardt, C., Hoelscher, M. & Nunn, A. Advancing the development of tuberculosis therapy. *Nature Reviews Drug Discovery* **11**, 171–172 (2012).
34. Fürstner, A. & Davies, P. W. Catalytic carbophilic activation: catalysis by platinum and gold  $\pi$  acids. *Angew. Chem. Int. Ed.* **46**, 3410–3449 (2007).
35. B  thegnies, A., Daran, J.-C. & Poli, R. Platinum-catalyzed hydroamination of ethylene: study of the catalyst decomposition mechanism. *Organometallics* **32**, 673–681 (2013).
36. Bajracharya, G. B., Nakamura, I. & Yamamoto, Y. PtBr<sub>2</sub>-catalyzed consecutive enyne metathesis-aromatization of 1-(1-methoxy-but-3-enyl)-2-(1-alkynyl)benzenes: dual role of the Pt catalyst. *J. Org. Chem.* **70**, 892–897 (2005).
37. Patil, N. T., Kavthe, R. D., Raut, V. S. & Reddy, V. V. N. Platinum-catalyzed formal Markownikoff's hydroamination/hydroarylation cascade of terminal alkynes assisted by tethered hydroxyl groups. *J. Org. Chem.* **74**, 6315–6318 (2009).
38. Fujii, M., Nishimura, T., Koshiba, T., Yokoshima, S. & Fukuyama, T. 2-Pyridone synthesis using 2-(phenylsulfanyl)acetamide. *Org. Lett.* **15**, 232–234 (2013).
39. Lv, Z. *et al.* Design, synthesis, and antihepatitis B virus activities of novel 2-pyridone derivatives. *J. Med. Chem.* **53**, 660–668 (2010).
40. Hibi, S. *et al.* Discovery of 2-(2-oxo-1-phenyl-5-pyridin-2-yl)-1,2-dihydropyridin-3-yl)benzotrile (perampanel): a novel, noncompetitive  $\alpha$ -amino-3-hydroxy-5-methyl-4-isoxazolepropanoic acid (AMPA) receptor antagonist. *J. Med. Chem.* **55**, 10584–10600 (2012).
41. Zhu, C. & Cook, S. P. A concise synthesis of (+)-Artemisinin. *J. Am. Chem. Soc.* **134**, 13577–13579 (2012).
42. Yoshimitsu, T., Nojima, S., Hashimoto, M. & Tanaka, T. Total synthesis of ( $\pm$ )-Platencin. *Org. Lett.* **13**, 3698–3701 (2011).
43. Yang, F. *et al.* Antimicrobial metabolites from the paracel islands sponge agelas mauritiana. *J. Nat. Prod.* **75**, 774–778 (2012).
44. McGlinchey, R. P., Nett, M. & Moore, B. S. Unraveling the biosynthesis of the sporolide cyclohexenone building block. *J. Am. Chem. Soc.* **130**, 2406–2407 (2008).
45. Bolze, P., Dickmeiss, G. & J  rgensen, K. A. Organocatalytic asymmetric synthesis of 5-(trialkylsilyl)cyclohex-2-enones and the transformation into useful building blocks. *Org. Lett.* **10**, 3753–3756 (2008).
46. Inokoishi, Y., Sasakura, N., Nakano, K., Ichikawa, Y. & Kotsuki, H. A new powerful strategy for the organocatalytic asymmetric construction of a quaternary carbon stereogenic center. *Org. Lett.* **12**, 1616–1619 (2010).
47. Okombi, S., Rival, D., Bonnet, S., Mariotte, A.-M., Perrier, E. & Boumendjel, A. Discovery of benzylidenebenzofuran-3(2H)-one (aurones) as inhibitors of tyrosinase derived from human melanocytes. *J. Med. Chem.* **49**, 329–333 (2006).
48. Steinhaus, P. & Schieberle, P. Characterization of the key aroma compounds in soy sauce using approaches of molecular sensory science. *J. Agric. Food Chem.* **55**, 6262–6269 (2007).
49. Li, Z., Nakashige, M. & Chain, W. J. A brief synthesis of (–)-Englerin. *J. Am. Chem. Soc.* **133**, 6553–6556 (2011).
50. Liu, Y. *et al.* Gold-catalyzed highly efficient access to 3(2H)-furanones from 2-oxo-3-butynoates and related compounds. *Org. Lett.* **8**, 3445–3448 (2006).
51. Georgieva, M. & Andonovski, B. Determination of platinum(IV) by UV spectrophotometry. *Anal. Bioanal. Chem.* **375**, 836–839 (2003).

## Acknowledgements

The authors gratefully acknowledge financial supports of DST (SR/S1/OC-05/2012, SR/S4/GC-05/2012 and SR/NM/NS-29/2010) and CRNN, India and research fellowships from CSIR, India.

## Author contributions

Major work was done by S.G. All authors (S.K., K.S.G. & D.K.M.) were contributed in UV-vis study and characterizing the molecules, manuscript preparation, discussion on chemistry involved, and commenting on the manuscript.

## Additional information

Supplementary information accompanies this paper at <http://www.nature.com/scientificreports>

**Competing financial interests:** The authors declare no competing financial interests.

**How to cite this article:** Ghosh, S., Khamarui, S., Gayen, K.S. & Maiti, D.K. ArCH(OMe)<sub>2</sub>-a Pt<sup>IV</sup>-catalyst originator for diverse annulation catalysis. *Sci. Rep.* **3**, 2987; DOI:10.1038/srep02987 (2013).



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