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Phase Stability and Compressibility of 3R-MoN₂ at High Pressure

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We report phase stability and compressibility of rhombohedral 3R-MoN₂, a newly discovered layer-structured dinitride, using *in-situ* synchrotron high-pressure x-ray diffraction measurements. The obtained bulk modulus for 3R-MoN₂ is 77 (6) GPa, comparable with that of typical transition-metal disulfides (such as MoS₂). The axial compressibility along *a* axis is more than five times stiffer than that along *c* axis. Such strong elastic anisotropy is mainly attributed to its layered structure with loosely bonded N-Mo-N sandwich interlayers held by weak Van der Waals force. Upon compression up to ~15 GPa, a new hexagonal phase of 2H-MoN₂ occurs, which is irreversibly at ambient conditions. The structural transition mechanism between 3R and 2H phases is tentatively proposed to be associated with the rotation and translation of sandwich interlayers, giving rise to different layer stacking sequences in both phases. At high temperature, the decomposition of 3R-MoN₂ leads to the formation of hexagonal δ-MoN and the onset degassing temperature increases as the pressure increases. In addition, the low-temperature electrical resistivity measurement indicates that 3R-MoN₂ behaves as a semiconductor with an estimated band gap of $E_g \approx 0.5$ eV. 3R-MoN₂ also shows weak antiferromagnetic properties, which probably originates from the occurrence of magnetic zigzag edges in the structure.

Transition-metal (TM) nitrides are a class of technologically important compounds and have attracted considerable attention because they exhibit many unique properties and can be used as hard protective coatings (e.g., TiN and CrN)¹, semiconductors (ScN)², superconductors (e.g., NbN)³, and superior catalysts^{4,5}. Among TM nitrides, molybdenum nitrides (Mo-N) often exhibit interest properties particularly⁴⁻¹⁰. As a typical example, hexagonal δ-MoN holds the highest hardness in the family of metal nitrides with the second highest superconducting temperature (*i.e.*, $T_c \approx 14$ K)¹¹. Thus, the search for new nitrides in the Mo-N system will provide great opportunities for fundamental studies and industrial applications in many fields of science and technology.

However, synthesis of these nitrides is still challenging because the incorporation of nitrogen into the crystalline lattices of transition metals is thermodynamically unfavorable at atmospheric pressure. As a result, most of the reported TM nitrides are poorly crystallized and nitrogen-deficient with molar ratios of N: metal less than unity, which severely limits their use in diverse technological applications. In the binary Mo-N system, three different phases with varying nitrogen concentrations have been reported, referring to ref.⁵ for a thorough overview of this system. In spite of the fact that an oxidation state of Mo as high as +6 occurs in other chemical systems (e.g., MoO₃), the synthesis of nitrogen-rich nitride, MoN₂, is still limited by traditional synthetic routes at ambient pressure.

Thanks to recent advancements in high-pressure techniques, a number of novel nitrogen-rich TM nitrides have recently been synthesized from direct metal-gas (N₂) reactions in a pressure range of 18–50 GPa¹². The new compounds include *Th₃P₄*-type A₃N₄ (A = Zr and Hf)¹³ and noble metal dinitrides (OsN₂, IrN₂, and PtN₂)^{14,15}. Successful high-*P* synthesis of nitrides with higher oxidation states demonstrates that pressure can effectively promote the role of *d*-electrons in chemical bonding with nitrogen. Most recently, we have successfully synthesized a series of novel nitrogen-rich tungsten nitrides (e.g., W₂N₃ and W₃N₄) through a newly formulated solid-state ion-exchange reactions between Na₂XO₄ (X = Cr, Mo and W) and *h*BN at pressures up to 5 GPa, which is in the pressure range of the current technological capability for massive, industrial-scale production^{16,17}. Of particular interest is the discovery of a novel nitrogen-rich nitride, 3R-MoN₂ using this formulated synthesis methodology. Strikingly, the new nitride is explored to adopt a rhombohedral MoS₂-type structure (*i.e.*, a layered structure), which consist typically of one plane of hexagonally packed metal atoms sandwiched by two planes of nitrogen

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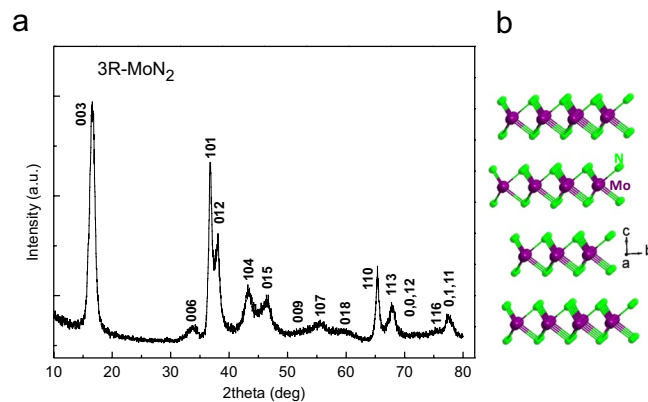


Figure 1. (a) X-ray diffraction pattern of rhombohedral 3R-MoN₂ taken at room temperature. (b) Crystal structure of 3R-MoN₂ characterized by the stacking of N-Mo-N sandwich layers.

atoms. The sandwich layers are vertically stacked and loosely bonded by weak van der Waals forces as suggested by theoretical simulations^{9,18}, similar to that in the TM dichalcogenides¹⁹. Besides, our preliminary experiments indicate that 3R-MoN₂ demonstrates highly catalytic activities for hydrogenation processes, and it may hold great promise as the next-generation catalysts and energy storage materials for a wide range of applications^{7,8}. Regarding physical properties of 3R-MoN₂, to date, it has sparsely been explored, but store exciting physics, especially for the material has the form of an atomic-level thin MoN₂ sheet. Remarkably, recent theoretical calculations indicate that the monolayer MoN₂ may have intriguing structural, electronic, and magnetic properties^{6,9,18}.

For the layer-structured material systems, it often exists a series of different polymorphs such as hexagonal 2H and rhombohedral 3R polytypes as demonstrated in MoS₂²⁰. The only structural difference between 2H and 3R phases is their stacking sequences of close-packed sandwich layers. Because of the weak interlayer interaction, both polymorphs can readily be converted between them through the interlayer rotation coupled with translation at certain high pressure and temperature²⁰. Apparently, it provides an effective protocol to prepare new polytype 2H-MoN₂ by treating the synthesized 3R-MoN₂ under high pressure conditions. To the best of our knowledge, in addition to its electronic and magnetic properties, the phase stability of the newly synthesized 3R-MoN₂ has not yet been investigated which further limits its industrial applications, calling for more experimental data on this material.

With these aims in this work, we present a comprehensive study on 3R-MoN₂ with focus on the phase stability and compressibility using high-*P* synchrotron XRD measurements, leading to the discovery of a new 2H-MoN₂. The elastic, electronic, and magnetic properties of 3R-MoN₂ have also been explored in detail.

Experimental Section

High-purity Na₂MoO₄ (>99.5%, ~50 μm) and *h*BN (>99.9%, ~50 μm) powders in the molar ratio Na₂MoO₄:BN = 1:2 were homogeneously mixed for the synthesis of the nitride. High *P-T* synthesis experiments were performed using a DS 6 × 14 MN cubic press and the detailed experimental procedures have previously been described in refs^{5,21}. The run products were washed with distilled water to remove the byproduct NaBO₂ and unreacted Na₂MoO₄, followed by drying in an oven at 348 K. To obtain phase-pure nitride, a two-step reaction route was adopted, referring to ref.⁵ for more experimental descriptions.

High-*P* angle-dispersive synchrotron x-ray diffraction (XRD) experiments using a diamond-anvil cell (DAC) were performed up to 30 GPa at the HPCAT 16BM-D beamline of the Advanced Photon Source (APS). In each of the high-*P* experiment, the nitride powders with submicron grain size were loaded into the sample hole in a stainless-steel gasket pre-indented to ~30 microns in thickness with neon as the pressure-transmitting medium. A few ruby balls were also loaded into the sample hole to serve as the internal pressure standard. High *P-T* energy-dispersive synchrotron diffraction experiments were performed up to 10 GPa and 1273 K in a large-volume high-*P* apparatus installed at the X17B2 beamline of the National Synchrotron Light Source (NSLS). The experimental details for angle- and energy-dispersive synchrotron measurements have been described previously^{16,22}. The crystal structure was determined from analysis of the x-ray data using the GSAS software²³.

Low-*T* magnetic susceptibility and four-probe resistivity measurements were conducted on a bulk sample sintered at 3.5 GPa and 753 K for 8 hours to measure the electric and magnetic properties. The final bulk sample was 4 mm in diameter and 1 mm in thickness. The density of the sintered sample was measured using the Archimedes method, and the obtained value is within more than 90% of the x-ray determined density.

Results and Discussion

Figure 1a shows a typical XRD pattern of the purified product synthesized at 3.5 GPa and 753 K for 20 hours through a step reaction. The refined lattice parameters $a = 2.854 \text{ \AA}$ and $c = 15.938 \text{ \AA}$ agree well with previously reported values⁵. The crystal structure of the rhombohedral 3R-MoN₂ is shown in Fig. 1b, exhibiting a layered structure similar to that of MoS₂. Figure 2a shows selected high-*P* synchrotron XRD patterns of 3R-MoN₂. During room-temperature compression, a new peak around $2\theta = 13^\circ$ was observed at ~14.8 GPa, and with further increasing pressure its diffraction intensity increases progressively, indicating the formation of a new MoN₂ phase (also see Fig. S1). The two

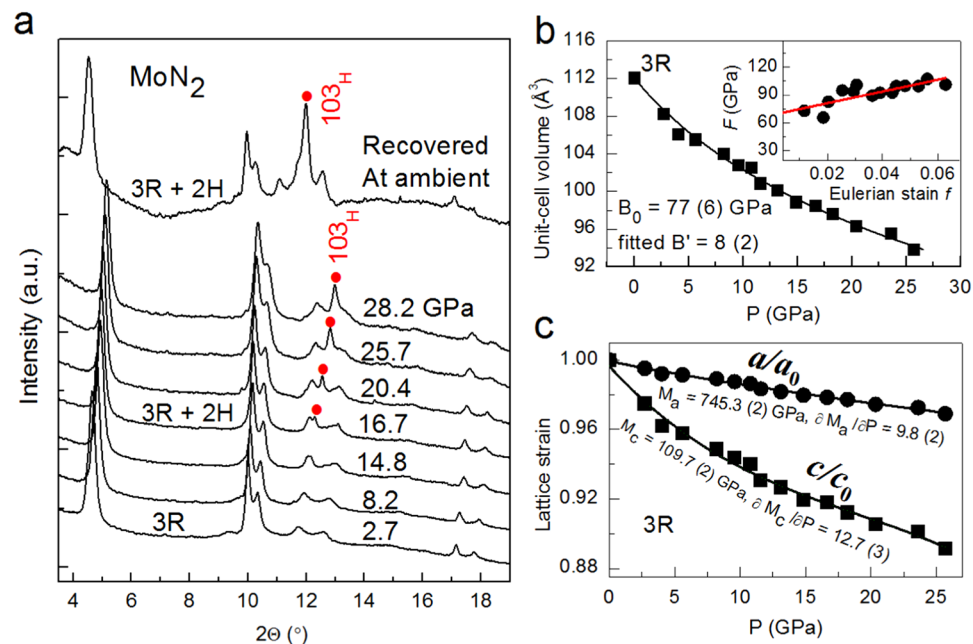


Figure 2. (a) Selected high-*P* synchrotron XRD patterns of 3R-MoN₂ upon room-temperature compression in a DAC. The incident x-ray wavelength (λ) is 0.424603 Å. Red dots denote the new Bragg reflection (103) originated from the hexagonal phase. (b) Pressure-volume data of 3R-MoN₂ fitted to the 3rd Birch-Murnaghan equation of state. The inset shows the normalized pressure (*F*) as a function of Eulerian strain (*f*). (c) Calculated linear elastic moduli along *a*- and *c*-axis for 3R-MoN₂, using the EoSFit program as made by Angel *et al.*³¹. All the error bars in (b,c) are too small to be visible.

MoN₂ phases coexist up to the highest experimental pressure of 30 GPa. After the release of the pressure, this phase was recovered at ambient conditions. As will be discussed below, this phase is referred to 2H-MoN₂. Shown in Fig. 2b is the pressure-volume data of 3R-MoN₂ fitted to the 3rd order Birch-Murnaghan equation of state. In the inset, the normalized pressure (*F*) is plotted vs. the Eulerian strain (*f*). The obtained bulk modulus, *B*₀, is 77 (6) GPa with *B*' = 8 (2), indicating that MoN₂ is slightly stiffer than MoS₂ (*B*₀ ≈ 53 GPa and *B*' ≈ 9)²⁴. This difference is likely due to the enhanced cation-anion bonding in the nitride, because compared with sulfur, the nitrogen is more favorable for the formation of strong covalent bonding states with transition metals. In addition, as shown in Fig. 2c, the *c*-axis is substantially more compressible than the *a*-axis, a behavior that is usually expected in the layer-structured materials. Also noted is that the elastic compressibility of the newly formed 2H phase would be similar due to the structural similarity of both phases as will be discussed below.

For the newly emerged phase at high pressure, as mentioned above it is well known that the layer-structured MoS₂ often exists in two different polymorphs: rhombohedral 3R and hexagonal 2H phases; the major difference between them is in the layer stacking sequence²⁰. Accordingly, the 2H-MoS₂-type structure with the space group *P*6₃/*mmc* was used for analyzing the XRD data of this new MoN₂ phase. As shown in Fig. 3a, the structure refinement shows that the calculated XRD profile is in excellent agreement with the observed data taken at 20.4 GPa with the coexistence of both rhombohedral and hexagonal polytypes. It is noted that the large background of high-*P* XRD patterns should originate from the sample crystallinity, because the synthesis of high-quality MoN₂ sample is still challenging as described in our previous report, referring to ref.⁵. This hexagonal phase is therefore referred to as 2H-MoN₂, and the refined structural parameters are listed in Table 1. The phase transition between 2H- and 3R-MoN₂ is presumably associated with the rotation and translation of N-Mo-N sandwich layers as previously reported in MoS₂²⁰. Because of their slight structural difference, it is challenging to distinguish between the 2H and 3R phases using the TEM techniques (Fig. S3). It is found that the pressure-induced phase transition from the rhombohedral 3R-MoN₂ phase to a hexagonal 2H-MoN₂ structure is irreversible, as referred from the experimental and theoretical x-ray diffraction patterns of MoN₂ at ambient conditions in Fig. 3b. Also noted is that the density of high-*P* 2H phase is anomalously lower than that of 3R-MoN₂ at 20.4 GPa as listed in Table 1. This is not unexpected because the transition from 3R to 2H is often kinetically difficult for achieving a complete conversion; as a result, the final 2H phase would involve a large fraction of the layer stacking disorder and strain in the lattice, which will lead to a significant lattice expansion, hence the reduced the density. For 2H-MoN₂, compared with simulated XRD pattern (see Fig. 3b), a slight peak shift of the 102 and 105 suggests a severe strain or stress that may build up between the interlayer at relatively low pressure. This is because the 2H phase is metastable phase and trends to transform into 3R phase upon decompression, involving a certain degree of the N-Mo-N interlayer rotation and translation. Compared with the 105 reflection, the 102 peak has a large shift relative to the calculated, probably because of the different stress as induced by interlayer stacking disorder involving different number of layers. The similar phenomenon has been observed in refs.^{5,20}. As depicted in Fig. 3c, the only crystal structural difference between of 3R- and 2H-MoN₂ is the interlayer stacking mode. In fact, for most known layer structured materials, they often adopt a common structure of hexagonal 2H polytype (AB|AB|...) with space

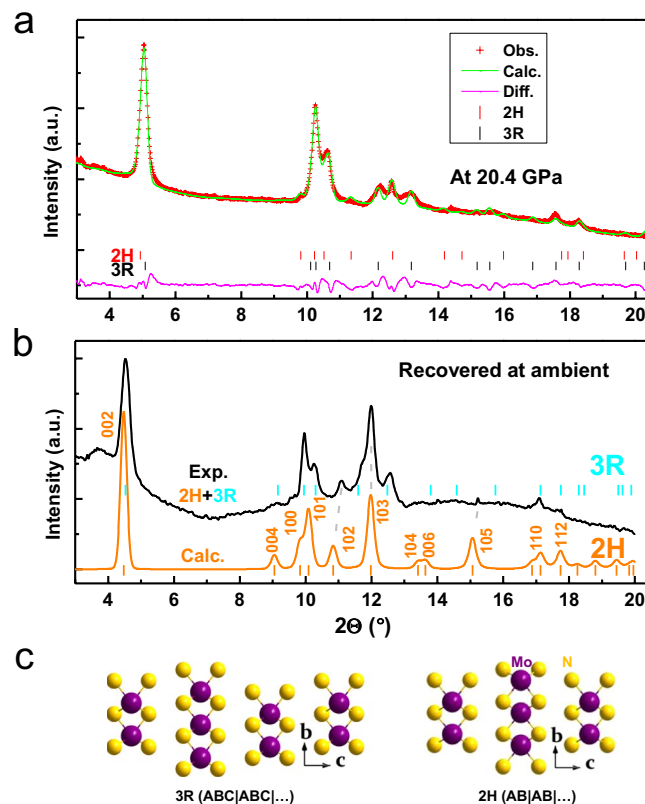


Figure 3. (a) Refined XRD pattern for 3R- and 2H-MoN₂ at 20.4 GPa. (b) XRD pattern of the recovered sample with coexistence of 3R and 2H phases. In (a,b), the incident x-ray wavelength (λ) is 0.424603 Å. (c) Comparison of crystal structures between 3R and 2H polymorphs.

	3R-MoN ₂		2H-MoN ₂
<i>P-T</i> conditions	Ambient	20.4 GPa, 300 K	20.4 GPa, 300 K
Formula	Mo ₃ N ₆		Mo ₂ N ₄
System	rhombohedral		hexagonal
Space group	<i>R</i> 3m (No. 160)		<i>P</i> 6 ₃ / <i>mmc</i> (No. 194)
Cell parameters [Å]	$a = b = 2.854$ (1) $c = 15.938$ (2)	$a = b = 2.776$ (3) $c = 14.380$ (3)	$a = b = 2.757$ (3) $c = 9.910$ (2)
Cell volume [Å ³]	112.42 (4)	95.968 (3)	65.235 (2)
Density [g•cm ⁻³]	5.492 (2)	6.433 (3)	6.310 (2)
Mo Wyckoff site	Mo1: 3a, (0, 0, 0.000) (1)		Mo1: 2c, (1/3, 2/3, 1/4) ^c
N Wyckoff sites	N1: 3a, (0, 0, 0.258) (1) N2: 3a, (0, 0, 0.402) (2)		N1: 4f, (1/3, 2/3, 5/8) ^d
$d_{\text{Mo-N}}$ [Å]	1.976, 2.037		
$D_{\text{interlayer}}$ [Å] ^a	5.313 (1)		
$D_{\text{layer distance}}$ [Å] ^b	3.017 (1)		
R_p , wR_p [%] (XRD) ^e	2.1, 3.4	1.7, 2.3	1.7, 2.3
R_p , wR_p [%] (NPD)	3.2, 5.3		
Refs	ref. ⁴	This study	This study

Table 1. Summary of structural parameters for MoN₂ phases determined by analyses of x-ray and neutron diffraction data. ^{a,b} $D_{\text{interlayer}}$ is the distances between the two nearest neighboring Mo planes, and $D_{\text{layer distance}}$ corresponds to the distance between the N planes. ^{c,d}Proposed atomic positions for 2H-MoN₂, which cannot be refined accurately using the current high-*P* XRD data. ^e R_p and wR_p represent the profile residual and the weighted profile R-factor of refined XRD patterns.

group of *P*6₃/*mmc* (No. 194), including TM dichalcogenides TMX₂ (TM = Nb, Mo, Ta, and W; X = S and Se). In contrast, the 3R rhombohedral polytype (ABC|ABC|...) is a high-temperature phase with a space group of *R*3m (No. 160)^{25,26}. Therefore, 3R-MoN₂ may be a high-*T* phase with a space group of *R*3m (No. 160), compared with 2H-MoN₂ with a space group of *P*6₃/*mmc* (No. 194)^{18,25}.

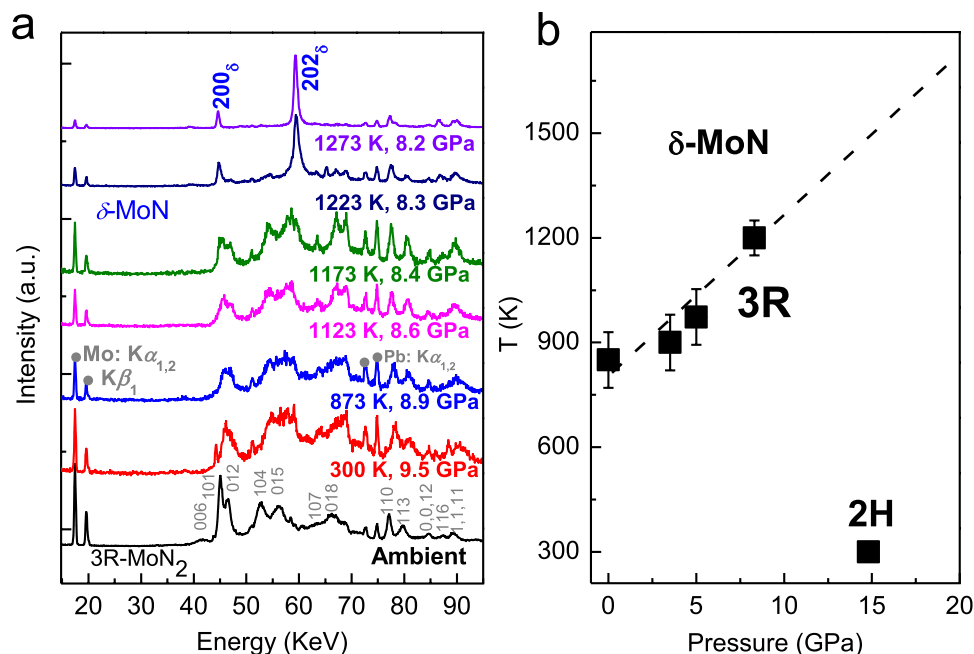
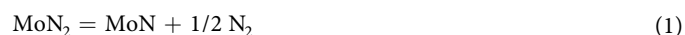


Figure 4. (a) *In situ* high P-T energy-dispersive synchrotron XRD measurements. Fluorescence peaks of Mo: K α_{β} and Pb: K α are denoted by grey solid dots, which originate from MoN₂ and lead shielding of the detector, respectively. (b) Thermal stability of 3R as a function of pressure. δ -MoN forms through the degassing of 3R-MoN₂ at high temperature.

To study the phase stability at high temperature, we performed *in-situ* energy-dispersive high *P-T* synchrotron XRD measurement using a large volume pressure. As shown in Fig. 4a, the XRD patterns were collected on heating at a constant load of 80 ton. Because of the thermal effect of the sample cell, the corresponding pressure decreases from 9.5 GPa at 300 K to 8.2 GPa at 1273 K. The strongest Bragg reflection (003) of 3R-MoN₂ is located at the low-energy region, and it is thus undetectable by the energy-dispersive XRD measurement. Clearly, δ -MoN is formed at ~1200 (20) K and 8.3 GPa through nitrogen degassing of 3R-MoN₂. The decomposition process is expressed by,



The phase stability of 3R-MoN₂, however, is substantially enhanced with increasing pressure, leading to highly increased N₂ decomposing temperature (see Fig. 4b). At 8.3 GPa, for example, the reaction (1) happens at a higher temperature of ~1200 K based on *in-situ* high *P-T* synchrotron XRD measurement. The phase diagram of MoN₂ is eventually determined and summarized in Fig. 4b. Apparently, at high temperature, the δ -MoN can be formed through nitrogen degassing of 3R-MoN₂, on the basis of our previous study of this material in ref.⁵. However, it is experimentally difficult to determine the phase boundary between 2H and 3R at high temperatures above 15 GPa. Nevertheless, the 3R phase seems thermodynamically more stable than 2H phase at high temperature as shown in Fig. 4b. Further experimental work is warranted to determine the detailed phase boundary between 3R and 2H using the state-of-the-art high-P techniques.

Figure 5 shows the magnetic and electrical resistivity measurements. At first glance, the low-*T* magnetic susceptibility (χ) data plotted in Fig. 5a suggests a weak paramagnetic-like behavior, as also reported in LiXN₂ (*X* = Mo and W)^{27,28}. However, after further analyses, the data can be fitted to the Curie-Weiss law in two temperature ranges, 130–300 K and 2–30 K. The obtained Weiss constants are $T_0 = -283$ (10) K and -6 (2) K, respectively, indicating that there exist two weakly antiferromagnetic (AFM-I and -II) phases for 3R-MoN₂. The corresponding magnetic moments of Mo atom are $\mu_{\text{eff}} = 0.12$ (1) and 0.05 (1) $\mu_B/\text{f.u.}$ Such weak antiferromagnetism in 3R-MoN₂ presumably originates from the magnetic zigzag edges at the grain boundaries as reported for MoS₂²⁹. The detailed magnetic measurements are listed in Table 2. Moreover, it is worthwhile to mention that the atomically-thin MoN₂ layers possess intrinsic high-*T* ferromagnetic properties on the basis of recent *ab-initio* calculations¹⁸, which may be closely associated with the observed magnetism in this work.

As shown in Fig. 5b, the electrical resistivity of 3R-MoN₂ increases dramatically as temperature decreases, which is characteristic of a semiconductor. Based on the typical law for a semiconductor, $\rho(T) = \rho_0 e^{(E_g/2k_B T)}$, a linear fit of the $\ln \rho - 1/T$ data yields a narrow band gap of $E_g = 0.50$ (1) eV, which is comparable to the value of 0.47 eV for PbS³⁰. However, it is substantially smaller than that of MoS₂, an indirect semiconductor with $E_g \approx 1.2$ eV²⁹. In contrast, LiMoN₂ and LiWN₂ both behave as an intrinsic metal; the two ternary nitrides also adopt the 3R symmetry of its parent MoN₂ with intercalated Li layers^{27,28}. The measured semiconductor-like behavior in 3R-MoN₂ may be partially associated with the degrees of crystallinity and defects, such as layer stacking faults. Hence, further resistivity measurements on well-crystallized samples are warranted to clarify this issue.

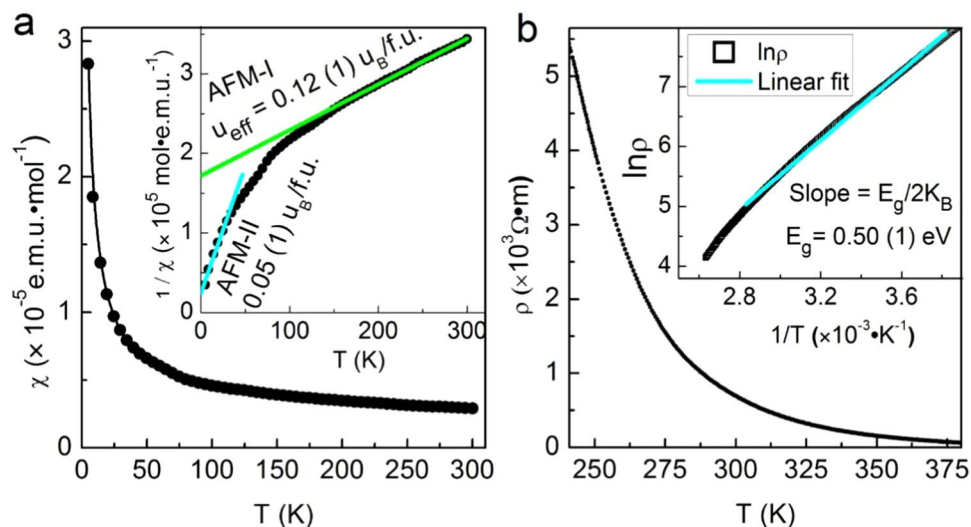


Figure 5. (a) Magnetic susceptibility as a function of temperature. The measurements were performed in an external magnetic field of $H = 1000$ Oe. The inset shows the inverse susceptibility, $1/\chi$, against T . It seems both AFM phases coexist in the 30–130 K temperature range. (b) Four-point probe resistivity as a function of temperature. The measurements were conducted on a well-sintered bulk sample. The inset shows a linear fit of $\ln \rho$ vs. $1/T$ based on the expression $\rho(T) = \rho_0 e^{(E_g/2k_B T)}$, which yields a band gap of $E_g = 0.50$ (1) eV.

	AFM-I	AFM-II
Temperatures [K]	130–300	2–30
Curies constant, C [10^{-3} emu·mol $^{-1}$ ·K]	1.69 (1)	0.29 (1)
Weiss constant, T_θ [K]	−283 (10)	−6 (2)
Magnetic moment, μ_{eff} [μ_B /f.u.]	0.12 (1)	0.05 (1)

Table 2. Magnetic parameters of 3R-MoN $_2$ derived from the magnetic susceptibility measurements. The magnetic susceptibility data were fitted to the Curie-Weiss law $\chi = C/(T - T_\theta)$ in two different temperature ranges of 130–300 K and 2–30 K.

Conclusions

In summary, the structural stability and compressibility of a newly discovered layer-structured rhombohedral 3R-MoN $_2$ have been studied using high-P compression measurements. A recoverable 2H-MoN $_2$, isotopic with hexagonal MoS $_2$, is also discovered via high-pressure processing of 3R-MoN $_2$, probably involving the interlayer rotation and translation as reported in MoS $_2$. Because of their structural similarity, the obtained bulk modulus and axial compressibility for 3R-MoN $_2$ are comparable to those of MoS $_2$. The obtained the bulk modulus for 3R-MoN $_2$ is $B = 77(6)$ GPa and the axial compressibility along c axis is much softer than that along a axis, confirming that the sandwich interlayers are loosely bonded by Van der Waals force. At a high temperature exceeding ~ 873 K, 3R-MoN $_2$ transforms into hexagonal δ -MoN through the degassing of N $_2$ and this disassociation temperature increases as pressure increases. Besides, 3R-MoN $_2$ is weakly antiferromagnetic, may resulting from the occurrence of magnetic zigzag edges in the structure. Moreover, the nitride behaves as a semiconductor with $E_g = 0.50(1)$ eV.

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Author Contributions

S.W. and Y.Z. designed research; X.Z. and D.M. performed high-pressure synthesis; X.Y. did low-temperature magnetic and electrical resistivity measurements; J.Z. and X.Z. performed high P-T synchrotron XRD measurement. X.Z. prepared Figures 1–5 and wrote the main manuscript text. All authors reviewed the manuscript.

Additional Information

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