

SCIENTIFIC REPORTS



OPEN

Ag fiber/IZO Composite Electrodes: Improved Chemical and Thermal Stability and Uniform Light Emission in Flexible Organic Light-Emitting Diodes

Junhee Choi¹, Cheol Hwee Park¹, Jin Ho Kwack^{1,2}, Dong Jun Lee¹, Jae Geun Kim¹, Jaemyeong Choi¹, Bong Han Bae¹, Soo Jong Park¹, Enjung Kim¹, Young Wook Park³ & Byeong-Kwon Ju¹

Electrospun metal fiber is a promising flexible transparent electrode owing to its extremely long length and facile fabrication process. However, metal-fiber electrodes have problems with chemical and thermal stability and nonuniform emission in organic light-emitting diode (OLED) at low luminance. In this study, we proposed a Ag fiber/IZO composite electrode with high stability. Ag fiber/IZO composite electrodes exhibited chemical and thermal stability. In addition, it was demonstrated that the OLED with the Ag fiber/IZO composite electrode operated stably, and the uniform emission of the OLED with metal-fiber electrodes improved by using highly conductive IZO film.

With increasing need for flexible optoelectronic devices, various flexible transparent electrodes have been intensively studied. Carbon nanotube, Graphene, conductive polymer, dielectric/metal/dielectric (DMD) and Ag nanowire have been investigated as flexible transparent electrodes^{1–9}. Carbon-based materials and conductive polymer have a great potential due to their transparency and flexibility, but they have a higher sheet resistance than that of Indium Tin Oxide (ITO). DMD multi layers such exhibit the superior optical properties. However, dewetting of the thin metal on the dielectric layer is likely to occur, and this problem leads to a serious increase in sheet resistance. Among the candidates for flexible transparent electrodes that replaced brittle ITO, Ag nanowires are attracting attention owing to their comparable performance with ITO, which has a high transparency (~90%) and low sheet resistance (15 Ω /sq)^{6–9}. However, Ag nanowires inevitably induce surface roughness owing to the junctions between the Ag nanowires when forming a Ag nanowire network. The surface roughness of the Ag nanowire induces leakage current and electrical short circuit in thin film devices.

Recently, electrospun metal fiber has offered attractive approaches due to its facile fabrication process and lower percolation threshold than metal nanowires; therefore, they have been recently applied to various electronic devices^{10–14}. Moreover, junction-free electrospun metal-fiber electrodes have been applied to thin film devices, including organic solar cell and organic light-emitting diodes (OLEDs), because they can control the surface roughness, which is a serious problem of metal nanowires^{15–18}.

However, metal fiber electrodes are vulnerable to post-heat and -chemical treatments that can occur after the electrode fabrication in the device fabrication process. For example, metal-fiber electrodes may be vulnerable to developing and baking processes for pixel define layer formation in the OLED manufacturing processes. In addition, an OLED with a metal-fiber electrode does not emit uniform light at low luminance because an electrical current cannot flow in the void space between the metal fibers and only flows through the metal fiber in the metal-fiber-based electrode. Even when buffer layers such as PEDOT: PSS, metal oxide, and graphene are coated

¹Display and Nanosystem Laboratory, School of Electrical Engineering, Korea University Seoul, 02841, Seoul, Republic of Korea. ²Samsung Display Co., Samsung St. 181, Tangjeong-Myeon, Asan-si, Chungcheongnam-do, 31454, Republic of Korea. ³School of Mechanical and ICT Convergence Engineering, SUN MOON University, Asan-si, Chungcheongnam-do, 31460, Republic of Korea. Correspondence and requests for materials should be addressed to Y.W.P. (email: zerook@sunmoon.ac.kr) or B.-K.J. (email: bkju@korea.ac.kr)

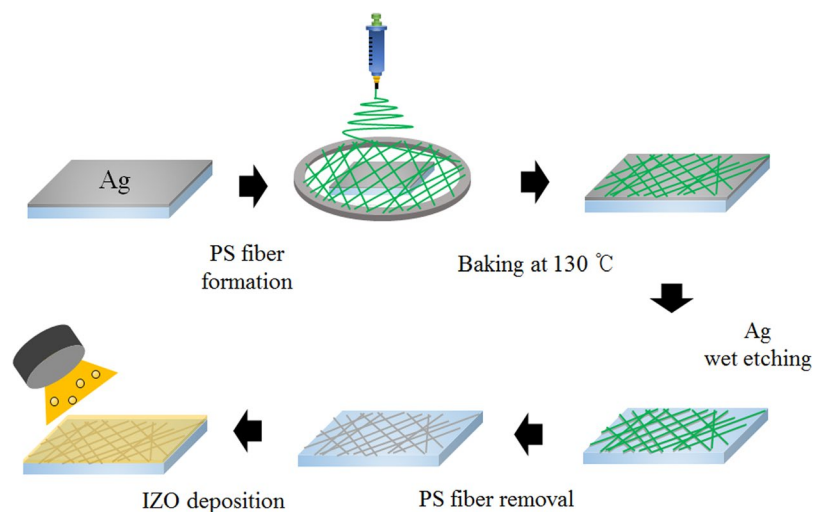


Figure 1. Schematic illustration of the fabrication procedure of the Ag fiber/IZO composite electrodes.

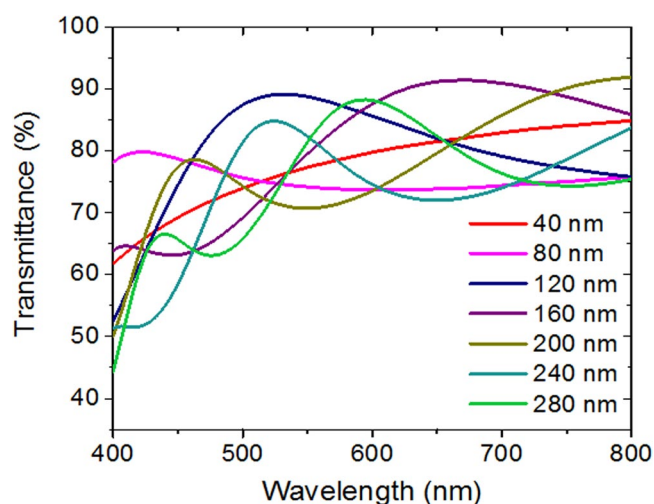


Figure 2. Calculated IZO thin-film transmittances at various thicknesses.

on a metal wire-based electrode^{19–25}, uniform light emission cannot be achieved due to their large resistance difference between the covered layer and metal fibers.

In this study, we fabricated junction-free Ag fiber/indium zinc oxide (IZO) composite electrodes by sputtering IZO on a Ag fiber. By introducing an IZO buffer layer on the Ag fiber, it is possible to obtain chemical and thermal stability. A sputtered IZO film that does not act as a planarization layer can be used as a buffer layer on a junction-free Ag fiber electrode with low surface roughness for stable OLED operation. In addition, the uniform emission of the OLED with a Ag fiber electrode can be achieved by depositing the IZO film, which has a resistance equal to that of the Ag fibers because the Ag fiber/IZO composite electrode has a parallel resistance of two materials—Ag fiber and IZO film.

Results and Discussion

Figure 1 illustrates the entire fabrication process of the Ag fiber/IZO composite electrodes. These electrodes are fabricated by wet-etching of Ag thin films with a polystyrene (PS) fibers mask¹⁵ and subsequent sputtering of an IZO film on the Ag fiber electrodes. First, a polyethylene naphthalate (PEN) substrate is cleaned by UV-O treatment, and a Ag thin film with 40 nm thickness is deposited by thermal evaporation system. PS fibers are formed by the electrospinning process on the deposited Ag thin film, and the PS fibers are annealed on a hot plate at 130 °C for 10 min to improve substrate adhesion. The Ag films not covered with the PS fiber mask are etched by O₂ plasma treatment and subsequent immersion in H₂O₂ solution. The PS fibers are dissolved in the chloroform solution. The IZO thin films are sputtered on the fabricated Ag fiber electrode.

The transmittance of the IZO thin films was calculated to determine the optimum IZO thickness. Figure 2 shows the calculated transmittances by finite-difference time-domain (FDTD) optical simulator. It shows the maximum transmittance peaks depending on the IZO thickness at a specific wavelength due to Fabry–Pérot interference. The IZO thickness was determined to fabricate a green fluorescent OLED with an emission wavelength

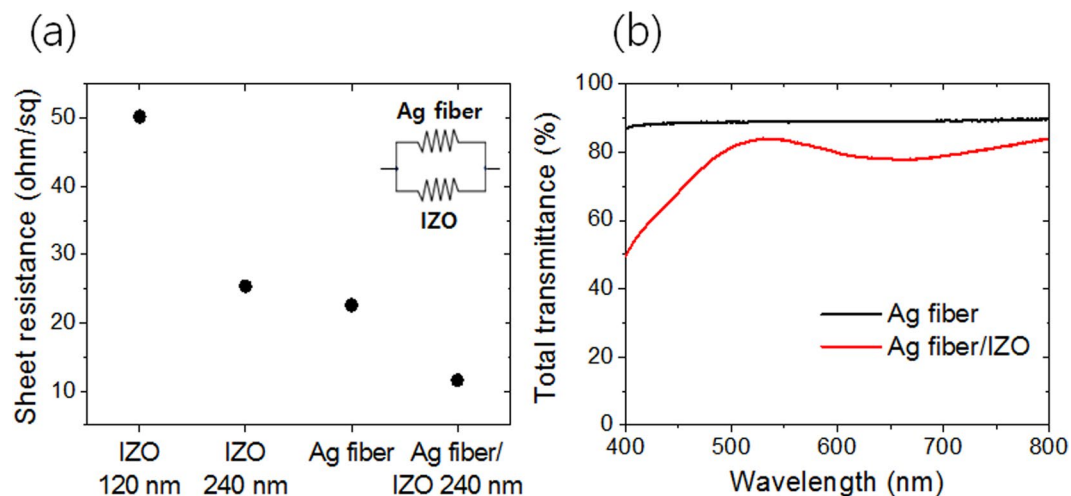


Figure 3. (a) Sheet resistances of IZO, Ag fiber, and Ag fiber/IZO composite electrodes, and (b) total transmittance of Ag fiber and composite electrode of Ag fiber/IZO with 240 nm thickness.

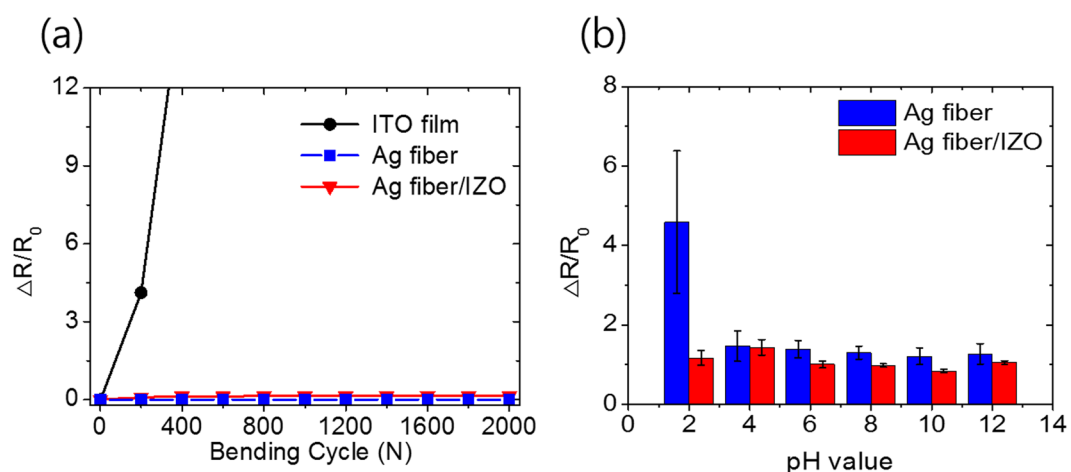


Figure 4. (a) Resistance changes in the cyclic bending test and (b) immersed in different pH solutions for chemical stability.

of 525 nm. When the thicknesses of the IZO film are 120 nm and 240 nm, the peak transmittances are observed at a wavelength of 525 nm.

Figure 3 shows the sheet resistance and total transmittance for the IZO films and Ag fiber electrodes. The sheet resistances of the IZO films with thicknesses 120 nm and 240 nm are 50.2 Ω /sq and 25.4 Ω /sq, respectively. The sheet resistance of the Ag fibers is adjustable by spinning time and Ag thickness¹⁵. To obtain uniform current distribution, the resistances of the Ag fiber and IZO film should be the same because the composite electrode has a parallel current path^{26–29}. The inset in Fig. 3 shows the parallel resistor circuit diagram of the Ag fiber and IZO composite electrode. Therefore, we fabricated an Ag fiber electrode with a resistance of 22.6 Ω /sq, which is almost the same as the sheet resistance the IZO film of 240 nm. The sheet resistance of the Ag fiber/IZO 240 nm composite electrode is 11.6 Ω /sq, which is comparable to the conventional ITO resistance. The composite electrode has better electrode properties than a single IZO film due to Ag fibers having good electrical and optical properties.

Figure 3(b) shows the total transmittances of the Ag fiber and composite electrodes. The transmittances were measured by UV-Vis spectrometer. The baseline is the PEN substrate. The transmittance of the Ag fiber electrode is approximately 90% in a broad-range wavelength. On the other hand, the Ag fiber/IZO composite electrode shows a peak transmittance of 83.6% at 525 nm (green). This can be explained by the Fabry–Perot interference of the IZO thin film as well as the simulation results.

To investigate mechanical durability, a repetitive bending test was performed. The bending radius was 5 mm, and the bending cycles were 2000 times. Figure 4(a) shows the resistance changes of different electrodes under a repetitive bending test. The resistance of the ITO film increases sharply as the bending cycle increases. On the other hand, the resistances of the Ag fiber electrode and Ag fiber/IZO composite electrode did not increase depending on the bending cycle. It was observed that the Ag fiber/IZO composite electrode had mechanical

Electrode Type	Before Annealing	After Annealing (400 °C)
Ag fiber	22.6	—
Ag fiber/IZO	11.6	15.5

Table 1. Sheet resistance of Ag fiber and Ag fiber/IZO composite electrodes before and after annealing.

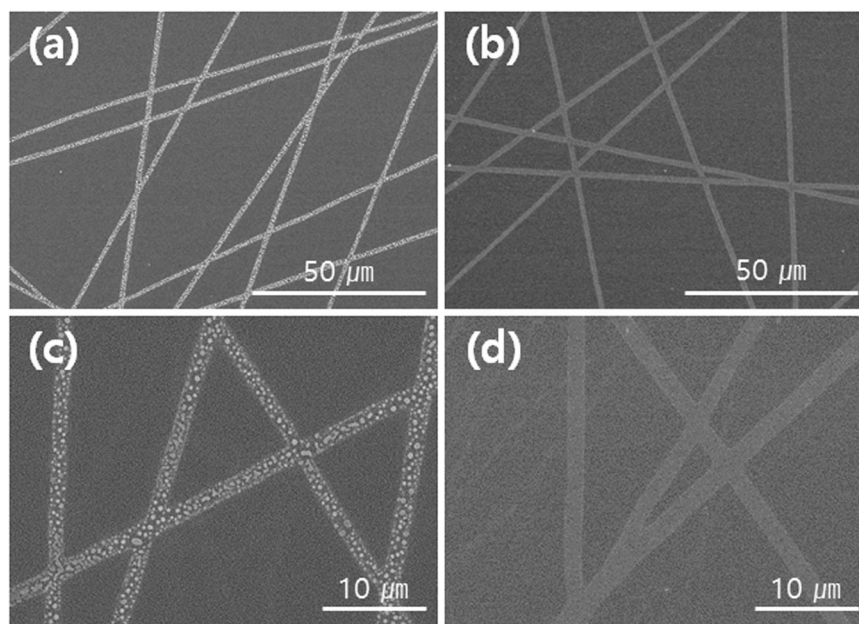


Figure 5. SEM images of the Ag fiber at magnifications of (a) $\times 1000$, (c) $\times 3000$, and Ag fiber/IZO at magnifications of (b) $\times 1000$ (d) $\times 3000$ after thermal annealing.

durability. To investigate the chemical stability, the Ag fiber and Ag fiber/IZO electrodes were immersed in different pH solutions for 20 min. Figure 4(b) shows the resistance change of the electrode depending on different pH solutions. The resistance of the Ag fiber electrode shows no distinct changes while immersing in solutions with pH 4–12. In the case of solutions with pH 2, the resistance change of the Ag fiber electrode is approximately 4.5. On the other hand, the resistance of the Ag fiber/IZO composite electrode hardly increases because the IZO film acts as a protection layer for the chemically vulnerable Ag fiber. Therefore, the Ag fiber/IZO composite electrodes show superior chemical stability.

To investigate the thermal durability of the Ag fiber/IZO composite electrodes, we fabricated the Ag fiber and Ag fiber/IZO composite electrodes on the polyimide substrate, and then, the electrodes were annealed by using the rapid thermal process system at 400 °C for 10 min in air. The sheet resistances of the electrodes were measured before and after annealing, and Table 1 presents the sheet resistances. The surface morphology of the Ag fiber and Ag fiber/IZO composite electrodes are measured by the scanning electron microscope (SEM). Figure 5(a–d) show the SEM images of the Ag fiber and Ag fiber/IZO electrodes after the annealing process. Ag is generally dewetted in the form of particles by the annealing process^{30,31}. In the case of the Ag fiber without the IZO layer, the Ag fiber was changed into Ag islands. (Fig. 5a,c) The network of Ag fibers was broken; thus, the resistance cannot be measured. On the other hand, the Ag fiber/IZO composite electrode shows remarkable stability. (Fig. 5b,d) The Ag fibers were not damaged at all due to the IZO buffer layer. However, the resistance of the composite electrode increased slightly after the heat treatment because the oxygen vacancy decreased by the oxidation of the IZO film, and the resistivity of the IZO film increased^{32,33}.

To demonstrate the possibility of using Ag fiber/IZO composite electrode with a flexible OLED, we fabricated a flexible fluorescent OLED with the Ag fiber/IZO composite electrode. Figure 6(a) presents the J-V-L characteristics of the fabricated OLED. The OLED with the Ag fiber/IZO composite electrode showed stable operation without leakage current, and the turn-on voltage was 3 V. Figure 6(b) exhibits the external quantum efficiency of the OLED with the Ag fiber/IZO composite electrode. The maximum external quantum efficiency is 1.39%. It is the general efficiency of fluorescent NPB/Alq3 based OLEDs^{34–37}.

Figure 7 shows the photographs of the OLEDs with different buffer layers and different luminances. Figure 7(a–c) shows the OLEDs with the PEDOT:PSS buffer layer on the Ag fibers, and Fig. 7(d–f) shows the OLED with the IZO buffer layer on the Ag fiber at luminances of 20 cd/m², 40 cd/m², and 60 cd/m², respectively. When the PEDOT:PSS buffer layer is applied, the resistance of PEDOT:PSS is larger than that of the Ag fiber. Therefore, it was observed that nonuniform emission occurs at low luminances of 20 cd/m² and 40 cd/m². When the luminance was 60 cd/m², the light emission became uniform. On the other hand, when the IZO buffer layer

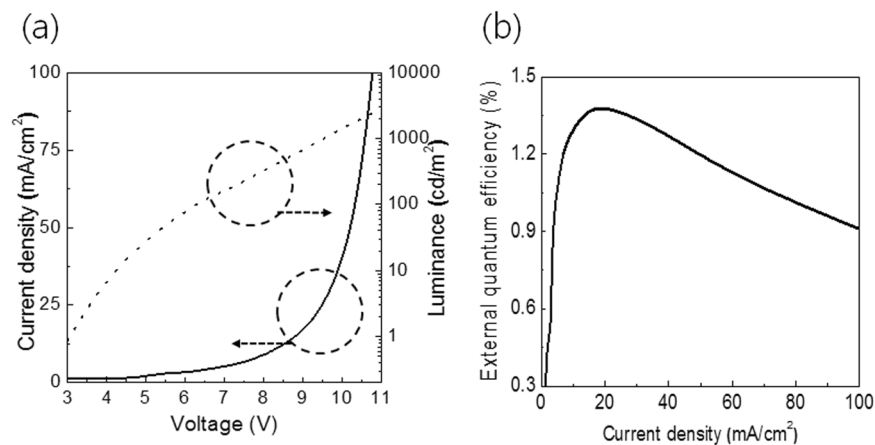


Figure 6. EL characteristics of OLED with the Ag fiber/IZO composite electrodes. (a) J-V-L characteristics, (b) external quantum efficiency.

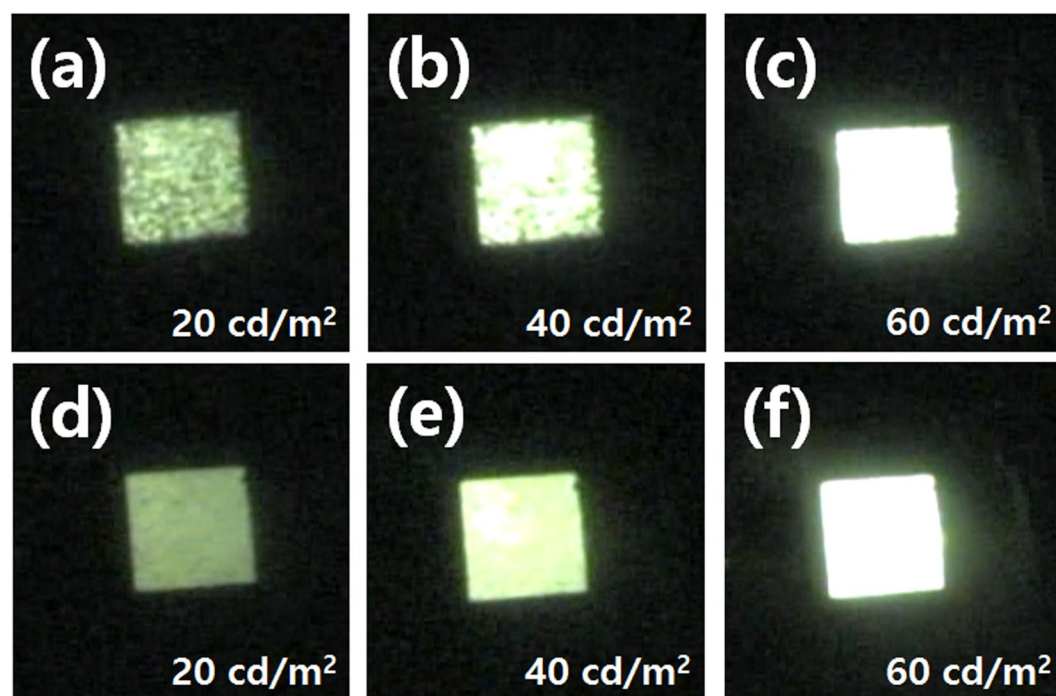


Figure 7. Photographs of the OLED emissions with the Ag fiber electrode under (a,b,c) PEDOT:PSS buffer layer and (d,e,f) IZO at different luminances.

was applied, it was observed that the uniform emission was maintained at a low luminance of 20 cd/m² by controlling the IZO film resistance to be the same as the resistance of the Ag fibers.

In summary, we demonstrated a Ag fiber/IZO composite electrode with chemical and thermal stability and observed uniform emission of an OLED with the Ag fiber/IZO composite electrode. FDTD optical simulation was performed to observe the optimal IZO thickness. A sputtered IZO buffer layer protected the Ag fibers from chemical and thermal damages and supported OLED uniform emission by equaling the resistance to that of the Ag fiber, which was possible because the IZO films had high conductivity.

Methods

Preparation of Ag Fiber/IZO Composite Electrodes. PEN substrates were cleaned by UV-O treatment for 10 min. Ag thin film with 40 nm thickness were deposited on PEN substrate by a thermal evaporator. Polystyrene powder with molecular weight 192,000 g/mol was purchased from Sigma-Aldrich Korea Ltd. Polystyrene solution is prepared by mixing 400 mg of polystyrene into 2 ml of acetone and 1 ml of dimethylformamide. The polystyrene fiber were deposited on Ag thin films by electrospinning process at 6 kV with an injection rate of 20 $\mu\text{L min}^{-1}$. The electrospun polystyrene fibers were annealed using a hot plate at 130 °C for 10 min to

promote the adhesion between the fibers and substrate. After heat treatment, the O₂ plasma was treated to oxidize Ag thin film not covered by the Ppolystyrene fibers at 100 W for 5 min under a pressure of 500 mTorr. The samples were immersed in H₂O₂ solution for 10 s to remove the oxidized Ag film. Next, the samples were immersed in chloroform for 5 min to remove the polystyrene fibers. Finally, the indium zinc oxide films were deposited on fabricated Ag fiber electrode.

Fabrication of Flexible OLEDs. The organic layers and cathode were deposited using a thermal evaporator (Digital Optics Vacuum Co., Ltd.) under high vacuum (10⁻⁶ Torr). To compare the uniform light emission, the IZO and PEDOT:PSS buffer layers were deposited. The PEDOT:PSS (Clevios PH1000 from Heraeus) buffer layers were deposited by spin-coating on Ag fiber electrodes at 1000 rpm for 30 s, then was baked at at 150 °C for 10 min. The fluorescent green OLED consisted of 5 nm hexaazatriphenylenehexacarbonitrile (HATCN)/60 nm N, N'-bis (naphthalen-1-yl)-N, N'-bis (phenyl)-benzidine (NPB)/80-nm-thick tris (8-hydroxy-quinolino) aluminum (Alq3). The HATCN, NPB, and Alq3 acted as hole injection layer, hole transport layer, and emission layer, respectively. A 0.7-nm-thick lithium fluoride (LiF) and 100-nm-thick Al was deposited as cathode.

Characterization. The transmittances were measured by a Cary 5000 (Varian/Agilent) UV–visible spectrometer. The sheet resistances were analyzed by a standard 4-point measurement. The mechanical durability test was performed using a bending tester (Z-Tec. Co., Inc.) To investigate the chemical stability, the pH solutions were prepared by mixing an HCl and NaOH solution with Di water. To observe the thermal stability, the surface morphology of the Ag fiber and the composite electrode were measured using a field emission scanning electron microscope (S-4800, Hitachi, Ltd.). The electroluminescence characteristics of the OLEDs were measured using a source-measurement unit (Model 237, Keithley Instruments, Inc.) and a spectroradiometer (PR-670 SpectraScan, Photo Research, Inc.).

References

1. Yu, L., Shearer, C. & Shapter, J. Recent development of carbon nanotube transparent conductive films. *Chemical reviews*. **116**, 13413–13453 (2016).
2. Rana, K., Singh, J. & Ahn, J. H. A graphene-based transparent electrode for use in flexible optoelectronic devices. *Journal of Materials Chemistry C*. **2**, 2646–2656 (2014).
3. Sun, K. *et al.* Review on application of PEDOTs and PEDOT: PSS in energy conversion and storage devices. *Journal of Materials Science: Materials in Electronics* **26**, 4438–4462 (2015).
4. Yang, D. Y., Lee, S. M., Jang, W. J. & Choi, K. C. Flexible organic light-emitting diodes with ZnS/Ag/ZnO/Ag/WO₃ multilayer electrode as a transparent anode. *Organic Electronics* **15**, 2468–2475 (2014).
5. Makha, M. *et al.* MoO₃/Ag/MoO₃ anode in organic photovoltaic cells: Influence of the presence of a CuI buffer layer between the anode and the electron donor. *Applied Physics Letters* **101**, 233307 (2012).
6. Chu, H.-C. *et al.* Spray-deposited large-area copper nanowire transparent conductive electrodes and their uses for touch screen applications. *ACS applied materials & interfaces*. **8**, 13009–13017 (2016).
7. Hwang, B. *et al.* Highly Flexible and Transparent Ag Nanowire Electrode Encapsulated with Ultra-Thin Al₂O₃: Thermal, Ambient, and Mechanical Stabilities. *Scientific Reports*. **7**, 41336 (2017).
8. Mardiansyah, D. *et al.* Effect of temperature on the oxidation of Cu nanowires and development of an easy to produce, oxidation-resistant transparent conducting electrode using a PEDOT: PSScoating. *Scientific reports*. **8**, 10639 (2018).
9. Yun, H. J. *et al.* Silver nanowire-IZO-conducting polymer hybrids for flexible and transparent conductive electrodes for organic light-emitting diodes. *Scientific reports*. **6**, 34150 (2016).
10. An, B. W. *et al.* Stretchable, transparent electrodes as wearable heaters using nanotrough networks of metallic glasses with superior mechanical properties and thermal stability. *Nano letters*. **16**, 471–478 (2015).
11. An, B. W. *et al.* Stretchable and transparent electrodes using hybrid structures of graphene–metal nanotrough networks with high performances and ultimate uniformity. *Nano letters*. **14**, 6322–6328 (2014).
12. An, S. *et al.* Facile processes for producing robust, transparent, conductive platinum nanofiber mats. *Nanoscale*. **9**, 6076–6084 (2017).
13. Park, J. *et al.* Flexible transparent conductive films with high performance and reliability using hybrid structures of continuous metal nanofiber networks for flexible optoelectronics. *ACS applied materials & interfaces*. **9**, 20299–20305 (2017).
14. Wu, H. *et al.* Electrospun metal nanofiber webs as high-performance transparent electrode. *Nano letters*. **10**, 4242–4248 (2010).
15. Choi, J. *et al.* Junction-Free Electrospun Ag Fiber Electrodes for Flexible Organic Light-Emitting Diodes. *Small*. **14**, 1702567 (2018).
16. He, T., Xie, A., Reneker, D. H. & Zhu, Y. A. A tough and high-performance transparent electrode from a scalable and transfer-free method. *ACS nano*. **8**, 4782–4789 (2014).
17. Hong, K. *et al.* Continuous 1D-Metallic Microfibers Web for Flexible Organic Solar Cells. *ACS applied materials & interfaces*. **7**, 27397–27404 (2015).
18. Yang, J., Bao, C., Zhu, K., Yu, T. & Xu, Q. High-Performance Transparent Conducting Metal Network Electrodes for Perovskite Photodetectors. *ACS applied materials & interfaces*. **10**, 1996–2003 (2018).
19. Chen, D. *et al.* Thermally stable silver nanowire–polyimide transparent electrode based on atomic layer deposition of zinc oxide on silver nanowires. *Advanced Functional Materials*. **25**, 7512–7520 (2015).
20. Chen, S. *et al.* Neutral-pH PEDOT: PSS as over-coating layer for stable silver nanowire flexible transparent conductive films. *Organic Electronics*. **15**, 3654–3659 (2014).
21. Chung, C.-H., Song, T.-B., Bob, B., Zhu, R. & Yang, Y. Solution-processed flexible transparent conductors composed of silver nanowire networks embedded in indium tin oxide nanoparticle matrices. *Nano Research*. **5**, 805–814 (2012).
22. Huang, Q. *et al.* Highly Flexible and Transparent Film Heaters Based on Polyimide Films Embedded with Silver Nanowires. *RSC Advances*. **5**, 45836–45842 (2015).
23. Huang, Q. *et al.* Highly thermostable, flexible, transparent, and conductive films on polyimide substrate with an AZO/AgNW/AZO structure. *ACS applied materials & interfaces*. **7**, 4299–4305 (2015).
24. Kim, A., Won, Y., Woo, K., Kim, C.-H. & Moon, J. Highly transparent low resistance ZnO/Ag nanowire/ZnO composite electrode for thin film solar cells. *ACS nano*. **7**, 1081–1091 (2013).
25. Scheideler, W. J. *et al.* A robust, gravure-printed, silver nanowire/metal oxide hybrid electrode for high-throughput patterned transparent conductors. *Journal of Materials Chemistry C*. **4**, 3248–3255 (2016).
26. Dhar, A. & Alford, T. High quality transparent TiO₂/Ag/TiO₂ composite electrode films deposited on flexible substrate at room temperature by sputtering. *APL materials*. **1**, 012102 (2013).
27. Han, J. H. *et al.* Highly conductive transparent and flexible electrodes including double-stacked thin metal films for transparent flexible electronics. *ACS applied materials & interfaces*. **9**, 16343–16350 (2017).

28. Kang, S. *et al.* Characteristics of an oxide/metal/oxide transparent conducting electrode fabricated with an intermediate Cu–Mo metal composite layer for application in efficient CIGS solar cell. *RSC Advances* **7**, 48113–48119 (2017).
29. Lee, M.-S. *et al.* High-performance, transparent, and stretchable electrodes using graphene–metal nanowire hybrid structures. *Nano letters*. **13**, 2814–2821 (2013).
30. Azeredo, B. *et al.* Silicon nanowires with controlled sidewall profile and roughness fabricated by thin-film dewetting and metal-assisted chemical etching. *Nanotechnology*. **24**, 225305 (2013).
31. Liu, R., Zhang, F., Con, C., Cui, B. & Sun, B. Lithography-free fabrication of silicon nanowire and nanohole arrays by metal-assisted chemical etching. *Nanoscale research letters*. **8**, 155 (2013).
32. Barquinha, P., Goncalves, G., Pereira, L., Martins, R. & Fortunato, E. Effect of annealing temperature on the properties of IZO films and IZO based transparent TFTs. *Thin Solid Films*. **515**, 8450–8454 (2007).
33. Gonçalves, G. *et al.* Influence of post-annealing temperature on the properties exhibited by ITO, IZO and GZO thin films. *Thin solid films*. **515**, 8562–8566 (2007).
34. To, B. D., Yu, C. C., Ho, J. R., Kan, H. C. & Hsu, C. C. Enhancing light extraction efficiency of organic light-emitting diodes by embedding tungsten trioxide islands or network structure pattern-transferred from a self-assembled deliquesce cesium chloride mask. *Organic Electronics*. **53**, 160–164 (2018).
35. Kim, K. H. & Park, S. Y. Enhancing light-extraction efficiency of OLEDs with high- and low-refractive-index organic–inorganic hybrid materials. *Organic Electronics*. **36**, 103–112 (2016).
36. Lim, T. B., Cho, K. H., Kim, Y. H. & Jeong, Y. C. Enhanced light extraction efficiency of OLEDs with quasiperiodic diffraction grating layer. *Optics express*. **24**, 17950–17959 (2016).
37. Lee, S. M., Cho, Y., Kim, D. Y., Chae, J. S. & Choi, K. C. Enhanced Light Extraction from Mechanically Flexible, Nanostructured Organic Light-Emitting Diodes with Plasmonic Nanomesh Electrodes. *Advanced Optical Materials*. **3**, 1240–1247 (2015).

Acknowledgements

This research was supported by the National Research Foundation of Korea (NRF) grant funded by the Korea Government (MSIP) (No. 2016R1A2B4014073) and the Ministry of Education (No. NRF-2017R1D1A1B03036520), the Industry Technology R&D program of MOTIE/KEIT [10048317, Development of red and blue OLEDs with external quantum efficiencies over 20% using delayed fluorescent materials], and the Brain Korea 21 Plus Project in 2018.

Author Contributions

J. Choi (Junhee Choi) designed the experiment, performed the fabrication and analyzed the results. C.H. Park and J.H.K. wack fabricated the OLED device. D.J. Lee, J.G. Kim and B.H. Bae fabricated the composite electrode. D.J. Lee performed the mechanical bending test. J. Choi (Jaemyeong Choi) performed the thermal stability test. S. Park and E. Kim performed the chemical stability test. Y.W. Park and B.-K. Ju supervised the project. J. Choi (Junhee Choi), Y.W. Park and B.-K. Ju prepared the manuscript.

Additional Information

Competing Interests: The authors declare no competing interests.

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2019