SCIENTIFIC REPORTS

OPEN

Received: 9 February 2018 Accepted: 11 June 2018 Published online: 27 June 2018

Metabolites Re-programming and Physiological Changes Induced in *Scenedesmus regularis* under Nitrate Treatment

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Nitrate is required to maintain the growth and metabolism of plant and animals. Nevertheless, in excess amount such as polluted water, its concentration can be harmful to living organisms such as microalgae. Recently, studies on microalgae response towards nutrient fluctuation are usually limited to lipid accumulation for the production of biofuels, disregarding the other potential of microalgae to be used in wastewater treatments and as source of important metabolites. Our study therefore captures the need to investigate overall metabolite changes via NMR spectroscopy approach coupled with multivariate data to understand the complex molecular process under high (4X) and low (1/4X) concentrations of nitrate (NO₃⁻). NMR spectra with the aid of chemometric analysis revealed contrasting metabolites makeup under abundance and limited nitrate treatment. By using NMR technique, 43 types of metabolites and 8 types of fatty acid chains were detected. Nevertheless, only 20 key changes were observed and 16 were down regulated in limited nitrate condition. This paper has demonstrated the feasibility of NMR-based metabolomics approach to study the physiological impact of changing environment such as pollution to the implications for growth and productivity of microalgae population.

Nitrogen is a primary nutrient regulating growth and metabolism in microalgae, often absorbed in the form of nitrate. Nitrogen is important mainly for tissue growth and protein synthesize¹. The ability of microalgae to adapt to varying levels of macronutrients and micronutrients in the environment are proven of an extensive mechanism to holistically reallocate its metabolic components in order to create a response and sustain through the adverse situations^{2,3}.

Although response toward differing levels of nutrients are species-specific, subsequent studies have shown that minor manipulation onto the medium's nutrient content will induce physiological and biochemical changes in carbohydrate, protein and lipid production. Microalgae adapt to varying levels of nutrients via two strategies; luxury consumption (direct uptake in excess without immediate need) and energy mitigating for nutrient supply (directing use of carbon reserves during starvation periods)^{4,5}. However, exactly how microalgae adapt to one of these two strategies is not fully understood yet. Abundance of nitrate may activate luxury uptake and results in the accumulation of amino acid precursors (NO₃⁻ and NH₄⁺) and nitrate reductase^{3,4}. Nitrogen is either assimilated into holding within their cell wall or converted into nitrogen-based reserves such as simple amino acids^{4,6}. Recent molecular studies carried out on *Chlamydomonas reinhardtii* found a family of three protein transporters NRT1 (NPF), NRT2, and NAR1 are responsible for determining substrate specificity and affinity of nitrate uptake in regards to nitrate concentration⁷.

Physiologically, nitrate limitation triggers a stress response in microalgae and are often associated with slow growth rates, decreased production in amino acids and declining enzymatic activity leading to reduce photosynthetic capability^{3,4,8}. Respiratory metabolism becomes secondary as more energy is focussed on reallocation of metabolites designed for storage such as nonphosphorylated polyglucans and lipids⁹.

More notably, nitrogen limitation has been studied for its triacylglycerol (TAG)-enhancing properties and possible formation of economically important secondary metabolites. Many of the microalgae studied reported

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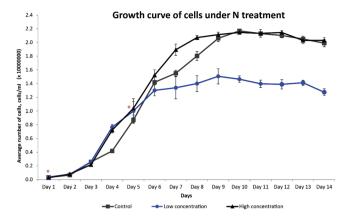


Figure 1. Growth curve of *S. regularis* under different concentrations of NO₃ treatment. Points show average concentration of cells in 1.6 L of liquid culture from each batch of culture. The difference in exponential growth phase of $1/4X \text{ NO}_3$ with the control and $4X \text{ NO}_3$ treatment is noted with asterisks (*).

~20-40% of increase in total lipid content and a shift towards production of neutral storage fatty acids such as TAG^{1,6,10-12}. Changes in metabolic activities in nitrate and phosphate deplete mediums are also associated with production of reactive oxidative species (ROS) which are believed to trigger the assimilating of nutrients and energy into storage lipids¹³. This is because storage lipids and TAG synthesis provides an electron sink for excess electrons that occur during photo-oxidative stress thus serving as outlets to detoxify the lipid membrane^{14,15}. Production of secondary carotenoid metabolites (asthaxantin, β -carotene, lutein) are associated with the TAG pathway hence some microalgae like Dunaliella sp. and Hematococcus sp. are able to produce these economically important antioxidants under nutrient stress^{12,14}. Nevertheless, the responses toward nutrient stress is species specific, eg Chlorella sorokiniana and Chlorella minutissima showed only a minor increase in lipid accumulation¹⁶, while huge decline of antioxidant activities was detected in *Phaeodactylum tricornutum* and *Chlorella vulgaris*¹². Therefore, in order to capture the metabolic changes on a wider scale, an analytical method that enables examination of individual metabolite developments rather than total extract changes is needed. Nuclear magnetic resonance (NMR) is a reliable and unbiased spectroscopy technique that can capture what happens metabolically on a wide scale¹⁵. Acquisition time for samples is relatively short and hence is time-saving compared to other analytical methods¹⁷. NMR coupled with chemomeric analysis is a powerful tool for discriminating between groups of related samples and identifying regions of spectrum that can be dedicated to further analysis. Nutrient limitation and abundance brings out a myriad of reactions that is of great benefit if explored to its full potential. By knowing the cellular changes under different culture conditions, the full potential use of microalgae for different usage, eg pharmaceutical, bioremediation, or as biofuel production can be explored. By using Scenedesmus regularis or synonym of Pectinodesmus regularis¹⁸ as microalgae model, this study demonstrated some important biochemical developments and changes in appearance, biomass accumulation, chlorophyll content and metabolites reprogramming when exposed to the presence of different nitrate (N) concentration.

Results and Discussion

Biomass and related growth. Algal biomass gained from each treatment was evaluated in order to understand how growth was affected. Approximately one fold decrease in mass gain was observed in nitrate depleting culture when compared against 4X and control NO₃ treatment. The reduction of biomass was correlated with the reduction of growth rate observed in 14 days of treatment (Fig. 1). In nitrate depleting environment, S. regularis was found enter stationary phase mush earlier than in other treatment starting from day 6 (Fig. 1). Nitrate is the most important nutrient depended upon at the initial stage of growth in microalgae thus a lack of NO3 stunted microalgal growth⁷. Besides primary growth, nitrate is utilised as building blocks of genetic materials (DNA and RNA), proteins for cell expansion and pigment building blocks^{1,3}. Therefore, it is not surprise that under limited nitrate condition, when the nitrate was used up after one week of culture, most of the cell division stop and push S. regularis moved into stationary growth phase. Contrarily, when additional of NO₃ was available, carbon fixation occurs and prolong the exponential growth phase, which was in agreement of batch photobioreactors of Scenedesmus and Chlorella cultures¹⁹. Biomass accumulation of the control and 4X treatments in both nutrients did not differ significantly might due to the level of NO₃ abundance were probably not enough to trigger a sudden "bloom" in its numbers as it has been reported that levels of NO_3 have to be over 4.0 mg/L²⁰. Rapidly utilisation of nitrate but not causing bloom phenomenon in high nitrate culture condition may suggest an application in wastewater treatment²¹. Wastewater from industrial, agricultural or urban can serve as growth substrate to support microalgae biomass production²².

Chlorophyll content and total carotenoids. Chlorophyll pigments are important light harvesting proteins that ultimately vary in concentration under the microalgae's environment conditions. There are evidences shown that nitrate is an important building block for chlorophyll^{12,13}. Chlorophyll pigments in cultures of limited NO_3 were drastically reduced in comparison to control and high NO_3 cultures (Fig. 2a). Due to very low chlorophyll content, the microalgae in nitrate-limited treatment appear pale green with hints of yellow (Fig. 2b). This a)

b)

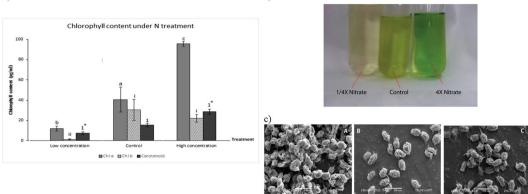


Figure 2. (a) Chlorophyll and carotenoid content of *S. regularis* measured from cultures under NO₃ treatment. Alphabets, roman numerals and numbers represent significant different between chl a, chl b and carotenoid respectively. Asterisks (*) mark significant difference between treatments group. (b) Comparison of colour in medium culture of 1/4X NO₃, control and 4X NO₃. (c) Scanning electron microscope (SEM) images of *S. regularis*; (A) 1/4X NO₃, (B) control and (C) 4X NO₃.

condition known as chlorosis or "bleaching" is characteristic of NO₃ limitation^{12,13,23}. Chloroplasts serve as large reserves of nitrogen sources and has been shown to be degraded in a contractual autophagy mechanism in times of nitrogen starvation so that the remnants of degraded chlorophylls may be recycled to support metabolic needs that require NO₃^{3,4,14,24,25}. The highest chlorophyll content was achieved under 4X NO₃ treatment with total chlorophyll a and b measured at 107.2 µg/ml. Chl a accumulation was often observed coinciding with high rates of cell division²³, similar finding were observed in this study where the highest biomass produced in 4X nitrate treatment. The high Chl content would also coincide with higher photosynthesis activities and production of reactive oxygen species (ROS) as photosynthesis by products. Nevertheless, plants equipped with carotenoids that play photoprotective roles in the photosystem¹². As pointed out by Singh, et al.¹⁴, nutrient stress activates TAG-related secondary carotenoid formation and can be seen as a way to combat high ROS activity that happens during stressed conditions. Therefore, higher Chl a levels coincided with carotenoid levels that were averagely higher than the control in 4X NO₃ treatment. While under limited nitrate condition, photo-oxidative stress occurs with ROS attacking the chlorophyll system but plants fail to trigger more carotenoids to overcome the effect of oxidative stress which would explain no significant increase after being cultured in NO₃ limited conditions^{12,13}. Under limited NO₃ treatment, cells appeared to be clumped together, shrivelled in appearance and spine formation were observed (Fig. 2c). According to²⁶, the length of the spines represent the nitrate starvation experienced in microalgae. Interestingly, cells cultured in nitrate abundant cultures did not produce any signs of clumping and were observed to be more prone to form unicellular cells instead of paired sets of coenobia.

Metabolites detected from aqueous solvent. ¹H-NMR spectroscopy carried out on aqueous extracts returned rich spectra. Figure 3 shows a representative 400 Mhz NMR spectra of a microalgae sample from aqueous extracts of treated sample. The spectra was utilised for the identification of several endogenous metabolites (Table 1). Comparing spectra data from aqueous extracts of treated samples (Fig. 3a and b), some new peaks (trimethylamine, betaine and glycine) were observed in comparison to the control sample. This indicates an elicited response for certain metabolites when exposed to nutrient fluctuations. Choline-2 produces a prominent singlet at 3.19 ppm and is related to resonance from free choline while Choline-1 singlet at 3.20 ppm is a culmination of glycerophosphocholine, phosphorylcholine and minute influences from free choline²⁷. Choline production is an aliphatic monoamine commonly partitioned metabolically as phospholipid- related compounds with the most common being phosphotidylcholine (PC) in microalgae. Ethanolamine produced a visible peak only in NO_3 abundant cultures. Ethanolamine is often related as precursors of betaine-type lipids besides being incorporated as phosphotidylethanolamine (PE) into cell membranes²⁸. Betaine lipids are naturally occurring ether-linked glycerolipids containing a betaine moiety with no phosphate group. Betaine was detected in control and NO₃ abundance treatment but was absent in NO₃ limited cultures (Fig. 3a). The structural capacity of cell membranes in microalgae typically contain diacylglyceryl-3-O-4'-(N,N,N-trimethyl)- homoserine (DGTS), a betaine lipid and PC in interchangeable amounts. It has however, been observed by Boroda, et al.²⁹ that the cell membrane constitution does not distinguish between PC or DGTS as both lipids probably have equivalent functions. Trimethylamine is a constituent for the production of betaine lipids and can be correlated to an increase in ethanolamine production. Glycerol moiety was detected at δ 3.55. Glycerol is an important constituent in regards to both neutral and phospholipid-type lipids and is procured from the direct glycerol pathway or from the fermentation of starch^{14,30}.

Amine based metabolites are nitrogenous compounds and it is not surprising to observe the reduction of amine and amino acids compounds in limited nitrate condition. Chlorophytes usually contain mono, di- and polyamines in varying amounts. These often contribute either to odour, toxins or play ecologically important roles within the microalgae³¹. Amines also exhibit certain biological activity and hold a potential for developing value-added products; spermine as an example involved in nucleic acid synthesis and cell division³¹. Choline,

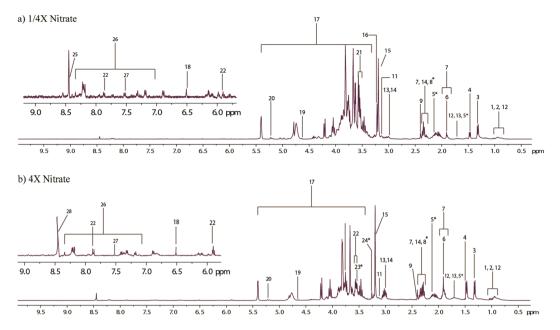


Figure 3. 1H NMR spectra of *S. regularis* extracted using aqueous solvent from (**a**) 1/4X Nitrate and (**b**) 4X Nitrate treatment. Metabolites marked with an asterisk appear only in the treatment sample and not in control. Key for spectra refer to Table 1.

ethanolamine and trimethylamine are common amines often included into cell wall constitution to maintain fluidity and selectivity of the cell membrane.

Carbon source such as alpha- and beta-glucose were also observed in spectra (Fig. 3). Microalgae have adapted to storing carbon in the form of a disaccharide because it is more stable chemically thus allowing the cell to also use sucrose as an osmolyte³². Spectra obtained from 1/4X NO₃ cultures were not as sharp as spectra obtained from other treatment cultures. Broad spectra are indicative of heavy molecular weight components inside the extract sample^{15,33,34}. A possible higher relative concentration of sugar or starch granules (in comparison to other metabolites) may have accumulated within the microalgae cell as microalgae cells have been acknowledged to synthesis starch under nitrate starved conditions^{8,35,36}.

Simple osmolytes (sucrose δ 3.66, betaine δ 3.25 and glycine δ 3.55) identified from the spectra show different intensity peaks among treatments. Two intermediates of the tricarboxylic-cycle (TCA) were also detected from all spectra. Fumarate produced a single singlet at δ 6.51 while succinate produced a singlet at δ 2.40 which can interconvert and produce succinyl-CoA, another TCA intermediate. Glutamate, TCA intermediate, was also detected across all spectra. The TCA cycle is therefore indicative of an aerobic route used by *S. regularis* for procuring its energy currency and subsequent metabolites. Furthermore, metabolites indicative of fermentative pathways were also identified (lactate at δ 1.33; formate at δ 8.45; acetic acid at 1.91).

Metabolites detected from chloroform solvent. Lipids accumulated by microalgae can be divided into polar and non-polar lipids (neutral) lipids (Fig. 4, Table 1). Both types of lipids follow very clear and distinct accumulation strategies and may transform from one to another depending on the needs of cells^{37,38}.

Microalgae primarily store lipid since formation of lipid requires powerful reducing cofactors and ATP³⁹. These lipids can then be used to either stabilize cell due to its hydrophobic nature or broken down to release energy when required. The signals produced by the protons linked on double bonds from polyunsaturated fatty acids (PUFA) culminate in a multiplet signal at δ 5.4. The allylic chain protons give rise to signals at δ 2.06 and δ 2.81 while protons located near the carboxylic end of the chain produce signals at δ 1.61 and δ 2.34. Protons situated along the long saturated carbon chain gives a very obvious multiplet at δ 1.25. It is interesting to note that fatty acid chains ending with a methyl- end attached with a double bond at an omega position gives rise to signals at δ 0.98 while non-omega methyl end protons culminate in a signal at δ 0.88. Recent studies have designated signals at the multiplet δ 2.81 to correspond to double bonds of three types of PUFA; docosahexaenoic acid, linoleic acid, α -linolenic acid⁴⁰. The corresponding signals, marked 16a, b and c, were also observed in this study at: (i) δ 2.77 for linoleic acid (C 18:2); (ii) δ 2.81 for α -linolenic acid (C 18:3); (iii) δ 2.84 for Docosahexaenoic acid (C 22:6/DHA). Signals related to carotenoid were found in the region of δ 6.0– δ 6.7. According to Sobolev, *et al.*⁴¹, salient signals appearing in the region between δ 6.0– δ 6.7 can be attributed to a fragment of conjugated trans double bonds, represented chemically as -(CH₃)-C=CH-CH-CH-C-(CH₃)-, common to all carotenoids. The protons (in bold) give rise to signals of multiplets at δ 6.14, δ 6.64 and δ 6.35 in the literature. A structure related to chlorophyll, i.e. phaeophytin, produces two prominent singlets at δ -1.43 and δ -1.61 in a solvent of $CDCL_3^{41}$. Two prominent singlets were also observed at δ –1.41 and δ –1.61 in this study thus proving presence of chlorophyll-related molecules in the non-aqueous mixture. Carotenoids are important pigments of the photosynthetic apparatus and also act as powerful antioxidant to relief the cell from excess photo-oxidation⁴².

LipoproteinsCH091 (m)AqueousLoacheTerninal-CH, Terninal-CH, S-CH, o-CH, 095 (b). 12 (m)AqueousValueTerninal-CH, Terninal-CH, S-CH, o-CH, 095 (b). 13 (b). 12 (m). 3.65 (d)AqueousLactateCH, CH133 (d). 14 (q)AqueousJamareCH, S-CH, 3-CH, 3-CH, 1137 (b). 157 (m). 3.60 (b). 3.60 (c)AqueousJysineCH, S-CH, 3-CH, 3-CH, CH137 (b). 157 (m). 2.10 (m). 3.01 (b). 3.00 (b). 3.00 (c)AqueousSpernineHB, 13, (H2, H9, H6, H1)191 (c). 2.31 (b). 3.01 (b). 3.01 (c). 3.	Metabolites	H Group	δH ⁺ (Multiplicity)	Solvent	
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AlamineCH ₂ CH148(d), 376(q)Aqueosi1yine~CH ₂ -CH ₂ , 1-CH ₂ , 2-CH13 (n), 1.2 (n), 3.03 (n)AqueosiSpernineBK, BL(2, BP, BK, BL1)17 (n), 2.2 (n), 3.14 (n)AqueosiActic acidCH ₂ , 0-CH ₂ 19 (q), 2.31 (n)AqueosiGimma-mino-N-buryta2-CH ₂ , 0-CH ₂ 19 (q), 2.31 (n), 3.03 (n)AqueosisGimma-mino-N-buryta2-CH ₂ , 0-CH ₂ 21 (n), 2.33 (n), 3.01 (n)AqueosisGintanati3-CH ₂ , 0-CH ₂ 21 (n), 2.33 (n), 3.01 (n)AqueosisSociatatCH ₂ , 0-CH ₂ 21 (n), 2.33 (n), 3.05 (n)AqueosisChalan-1CH ₂ , CH ₂ NH3.14 (n), 3.85 (n)AqueosisChalan-1CH ₂ , CH ₂ NH3.26 (n), 5.51 (n)AqueosisSociasisCH ₂ , CH ₂ NH3.26 (n), 5.51 (n)AqueosisGylareCH ₂ , CH ₂ NH3.55 (n)AqueosisGylareCH ₂ , CH ₂ O3.55 (n)AqueosisGylareCH ₂ , CH ₂ O3.55 (n)AqueosisGylareCH ₂ , CH ₂ O3.55 (n)AqueosisGylareCH ₁ , CH ₂ O3.55 (n) <td>Valine</td> <td></td> <td>0.98 (d), 1.03 (d), 2.2 (m), 3.63 (d)</td> <td>^</td>	Valine		0.98 (d), 1.03 (d), 2.2 (m), 3.63 (d)	^	
AlamineCHq, CHq, CHq, CHq14 (d), 375 (d)AqueosiLyaineAqueosiSpernineHB, H(2), H9, H5, H1)11 (u), 12 (u), 31 (u) (u), 01, 01, 01, 01, 01, 01, 01, 01, 01, 01	Lactate			^	
lysine?CH ₂ & CH ₂ & CH ₂ 15 (m), 172 (m), 130 (m)AqueousSpermineH8, H3, H2, H9, H0, H11)171 (m), 212 (m), 314 (m)AqueousGarma-amino N-hutyra>CH ₂ or CH ₂ 191 (q), 231 (l), 134 (m)AqueousGarma-amino N-hutyra>CH ₂ or CH ₂ 191 (q), 231 (l), 303 (l)AqueousBatyric acid 4-aminoSCH ₂ or CH ₂ 11 (m), 231 (k), 305 (l)AqueousSuccinat>ZCH ₂ or CH ₂ 240 (s)AqueousChalmartCH ₁ U ₂ or CH ₂ 240 (s)AqueousChoine 1CH ₂ CH ₂ 240 (s)AqueousChoine 2CH ₂ CH ₂ 316 (s), 355 (n)AqueousChoine 2CH ₂ CH ₂ 326 (s), 355 (n)AqueousSucciaeCH ₁ CH ₂ 356 (s), 356 (s), 351 (m), 405 (s), 424 (d), 54 (d)AqueousGlycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 405 (s), 424 (d), 54 (d)AqueousOglycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 405 (s), 424 (d), 54 (d)AqueousOglycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 405 (s), 424 (d), 54 (d)AqueousOglycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 405 (s), 424 (m), 405 (s)AqueousOglycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 406 (s), 424 (m), 405 (s)AqueousOglycralCH1CH ₂ 356 (s), 356 (s), 351 (m), 406 (s), 424 (m), 406 (s)AqueousOglycralH ₂ S46 (s), 752 (d)AqueousAqueousOglycralCH1CH ₂ S46 (s), 752 (m), 400 (s) <t< td=""><td>Alanine</td><td></td><td></td><td>Aqueous</td></t<>	Alanine			Aqueous	
SpermineHS, H3, (H2, H9, H6, H11)1.71 (m), 2.12 (m), 3.14 (m)AqueousActic acidCH,1.91 (a)AqueousActic acidCH,1.91 (a)AqueousBayric acid-sumino1.9 (m), 2.31 (n), 3.03 (n)AqueousBayric acid-sumino2.40 (a)AqueousSuccinateDaCH2, co-CH2.40 (a), 3.77 (n)AqueousSuccinateCH2, Ch2, oc-CH2.40 (a)AqueousChalan-1CH2, co-CH3.14 (n), 3.85 (n)AqueousChalan-2CH2, CH, NH3.20 (a), 5.55 (n), 7.66 (a), 4.00 (c), 4.21 (d), 5.44 (d), 7.85	Lysine	-	*	· ·	
Aretic acidCH,191 (a)AqueousGamma-amino N-burynaJ-CH, or CH,114 (a), 231 (1)AqueousGamma-amino N-burynaJ-CH, or CH,131 (a), 231 (J), 303 (1)AqueousGhtamata3-CH, -, CH, or CH24 (n), 231 (d), 377 (1)AqueousStaccinate24CH;240 (a)AqueousChalmalamineCH, NH, CH, OH314 (a), 385 (a)AqueousChaline-1CH,AqueousAqueousChaline-1CH, CH,331 (a)AqueousStaccinateCH, CH,320 (a), 355 (n)AqueousBetaineCH, CH,355 (a), 354 (d), 365 (d), 341 (n), 405 (f), 421 (d), 541 (d), 543 (d)AqueousGycoreCH CH355 (a), 557 (d), 557 (d)AqueousGycoreCH-1, CH-6381 (n), 522 (d)Aqueousor glocoseCH-1, CH-6381 (n), 522 (d)Aqueousor glocoseCH-1, CH-6586 (d), 756 (d)AqueousUradinH5, F6586 (d), 756 (d)AqueousUradineH(a), H6, 60586 (d), 758 (d)AqueousNacloside/Nucleode71-79AqueousFormateCH345 (a)AqueousPhosphatidytchalineN(CH ₃), CH, OP322 (m), 427 (m)ChloroformPhosphatidytchalineN(CH ₃), CH, OP322 (m), 427 (m)ChloroformPhosphatidytchalineCH, OP324 (m), 63 (m)ChloroformPhosphatidytchalineCH, OP324 (m), 63 (m)ChloroformPhosphatidytchalineCH, OP324 (m), 63 (m) </td <td>,</td> <td></td> <td></td> <td>^</td>	,			^	
Gamma-anino N-butynaβ-CH2, α-CH2,191 (q), 2.1 (r)AqueousBayric acidamino10 (m), 2.1 (r), 30 (r)AqueousBayric acidaminoβ-CH2, α-CH221 (n), 3.5 (n), 3.7 (r)AqueousStactnate2-CCH1, CL1, CL1, CL1, 2.3 (n), 3.57 (n)AqueousBahmaloumineCH1, N12, CH1, OH3.14 (s), 3.85 (n)AqueousChalme-1CH2, CH2, N113.14 (s), 3.85 (n)AqueousChalme-2CH2, CH1, N113.20 (s), 5.55 (m), 5.66 (s), 5.81 (m), 4.05 (r), 4.21 (d), 5.4 (d)AqueousSucroseH16, 1F9, OH1, H17, 19, H5, H373.26 (s), 3.56 (s), 5.81 (m), 4.05 (r), 4.21 (d), 5.4 (d)AqueousGyerelCH3.55 (s), 5.6 (d), 5.6 (d), 5.81 (m), 4.05 (r), 4.21 (d), 5.4 (d)AqueousGyerelCH0H, CHOH, COH3.55 (d), 3.65 (d), 3.81 (m), 4.05 (r), 4.21 (d), 5.4 (d)AqueousGyerelCH1, CH-63.81 (m), 5.22 (d)AqueousGyerelCH1, CH-63.88 (r), 7.26 (d)AqueousUradiH3, H65.80 (r), 7.86 (d)AqueousUradiH3, H65.80 (r), 7.86 (d)AqueousFormateCHG, GaronAqueousFormateCHAds (d)AqueousFormateCHAds (d)AqueousChoroformAds (m)ChioroformPhosphatidythylorelGlycerol molety3.1 (m)ChioroformPhosphatidythylorelGlycerol molety3.1 (m)ChioroformPhosphatidythylorelCH1, OL2Ads (m)ChioroformPhosphatidythylorelCH1, OL	*			· ·	
Butyric acid 4- aminoIn (Ph., 2.3 (n), 3.03 (n)AquecousGlutanuteBCH ₂ , γ -CH, α CH2.1 (n), 2.33 (n), 3.77 (n)AquecousBahanolamineCH, NH, CH, α 2.40 (a)AquecousEhanolamineCH, NH, CH, OH3.14 (a), 3.85 (a)AquecousCholine-1CH, CH, CH3.19 (a)AquecousCholine-2CH, CH, CH3.20 (a), 3.55 (n)AquecousBetaineCH, CH, CH3.20 (a), 3.55 (n)AquecousSaccaseH10; H9; OH; H17, 19; H5; H3; H73.46 (b), 3.57 (n), Ads (c), 4.21 (d), 5.41 (d)AquecousGlycineCH3.35 (a)3.55 (a)AquecousGlycineCHOH, CHOH, COH3.55 (a)AquecousS-glucoseCH-1, CH-63.81 (m), 5.22 (d)Aquecousor glucoseCH-1, CH-65.84 (n), 7.52 (d)AquecousUradiuH5, H65.86 (n), 7.52 (d)AquecousFormatraCH6.51 (a)AquecousPostarita/glycerolCH-GH, CH, CHOP3.22 (m), 4.27 (m)ChloroformRodeoside/NucleosideCH-3.34 (do f)AquecousAquecousProsphatidyfycerolCH-GH, CH, CH, CH, CH3.1-39ChloroformPhosphatidyfycerolCH, eran4.31 (m), 1.16 (m)ChloroformDapolaphilyfycerolCH, eran4.31 (m), 1.16 (m)ChloroformDapolaphilyfycerolCH, eran4.31 (m), 1.16 (m)ChloroformDapolaphilyfycerolCH-GH3.8-53 (m)ChloroformDapolaphilyfycerolCH-GH3.8-53 (m)Chlor	Gamma-amino-N-butyrate			^	
Glutamate $ cH_{3}, \gamma_{c}CH_{3}, \alpha_{c}CH2.1 (m), 2.33 (dt), 3.77 (t)AqueousSucchate2xCH_{3}2.40 (s)AqueousSucchate2xCH_{3}2.40 (s)AqueousChalne-1CH_{1}, CH, OH3.14 (s), 3.85 (s)AqueousChalne-2CH_{1}, CH_{1}3.26 (s), 3.51 (m)AqueousSacroacH10, H9, OH; H17, 19, H5, H3, H73.46 (t), 3.54 (d), 3.66 (s), 3.81 (m), 4.05 (t), 4.21 (d), 5.44 (d)AqueousGlyceneCHOLI, CHOL3.55 (d), 3.52 (d), 3.53 (d), 3.66 (s), 3.81 (m), 4.05 (t), 4.21 (d), 5.44 (d)AqueousGlyceneCHOLI, CHOL3.55 (d), 3.52 (d), 3.57 (m)Aqueousor glocoseCH-1, CH-63.81 (m), 5.22 (d)AqueousOrglocosCH-14.63 (d)AqueousUraclH5, H65.89 (t), 7.86 (d)AqueousVincleoside/Nucleoside7.1-7.9AqueousNucleoside/Nucleoside7.1-7.9AqueousPiosphatidyfolmeNCH, J2, CH, OP3.22 (m), 4.27 (m)ChloroformPiosphatidyfolmeNCH, J2, CH, OP3.22 (m), 4.27 (m)ChloroformPiosphatidyfolmeNCH, J2, CH, OP3.1-3.9ChloroformPiosphatidyfolgerolfGH, ora3.1-3.9ChloroformPiosphatidyfolgerolfGH, ora, 3 (b), CH an.24.31 (m), 4.14 (m), 5.26 (m)ChloroformDacyglycerolhosphalpidCH an.1, CH an.24.31 (m), 5.13 (m)ChloroformDiacyglycerolhosphalpidCH an.3, Alb, CH an.4, Alb, CH an.24.31 (m), 5.14 (m), 5.26 (m)ChloroformStande$		1 2* 2*	*	^	
Succinate λCH_1 $\lambda Participant\lambda PareousEhanolamineCH2NH2CH2OH\lambda 14 (a), \lambda 85 (b)\lambda parcousCholine-1CH4\lambda 14 (b), \lambda 85 (c)\lambda parcousCholine-2CH2CH1NH\lambda 20 (c), \lambda 35 (m)\lambda quecousBetaineCH2CH1NH\lambda 20 (c), \lambda 31 (c)\lambda 4quecousSuccoseH10, H90, H17, 19, H5, H3, H7\lambda 46 (t), \lambda 34 (d), \lambda 66 (c), \lambda 81 (m), 405 (t), 4.1 (d), 5.4 (d)GlycendCHOH, CHOH, COH\lambda 55 (d), \lambda 55 (m)\lambda quecousGlycendCHOH, CHOH, COH\lambda 55 (d), \lambda 57 (m)\lambda quecous\rho glucoseCH-14.63 (d)\lambda quecous\rho glucoseCH-1, CHO\lambda 84 (d), \lambda 57 (m)\lambda quecous\rho glucoseCH-14.63 (d)\lambda quecous\rho glucoseCH-1, CHO\lambda 84 (d), \lambda 57 (m)\lambda quecous\rho glucoseCH-14.63 (d)\lambda quecous\rho glucoseCH-1\lambda 84 (d)\lambda quecous\rho glucoseCH\lambda 44 (d)\lambda quecous\rho glucoseCH\lambda 84 (d)\lambda quecous\rho glucoseCH\lambda 84 (d)\lambda quecous\rho glucoseCH\lambda 44 (d)\lambda quecous\rho glucose\lambda 44 (d), \lambda 27 (m)Chloroform\rho glucoseCH, \lambda 44 (d), \lambda 44 (m), \lambda 26 (m)<$		β-CH ₂ , γ-CH ₂ , α-CH		^	
EhanolamineCH_NH3_CH_Q0H3.14 (a) 3.85 (a)AqueousCholine-1CH3,3.19 (a)AqueousCholine-2CH5, CH3/NH3.20 (a) 3.55 (m)AqueousBetaineCH4, CH43.26 (a) 3.91 (a)AqueousSucroseH10; H9, OH; H17, J9; H5, H3, H73.66 (a) 3.81 (m), 4.05 (a), 4.21 (d), 5.4 (d)AqueousGlycerelCHOH1. CH0H3.55 (d) 3.66 (a) 3.81 (m), 4.05 (a), 4.21 (d), 5.4 (d)AqueousGlycerelCH01, CH0H, COH3.55 (d) 3.65 (d), 3.77 (m)AqueousSplicoseCH-14.63 (d)AqueousUraclH15, H65.89 (a), 7.52 (d)AqueousUraclH3, H-65.89 (b), 7.86 (d)AqueousNackoside/NucleosideCH6.51 (s)AqueousNackoside/NucleosideCH-03.17.99ChloroformGlyceroglycolpidsCH,-m3405 (m)ChloroformGlyceroglycolpidsCH,-m3405 (m)ChloroformPhosphalidylkyleroetGlycerol molety3.61 (m)ChloroformPhosphalidylkyleroetCH,0P3.22 (m),4.27 (m)ChloroformDiacylglycerophosphilpiCH,0P3.24 (m)ChloroformDiacylglycerophosphilpiCH,0P3.1.59ChloroformDiacylglycerophosphilpiCH,0P4.24 (m)ChloroformDiacylglycerophosphilpiCH,0P4.21 (m),4.14 (m),5.26 (m)ChloroformSquaeneCH,0P4.31 (m),4.14 (m),5.26 (m)ChloroformGlycerophosphilpiCH,3.13 (m)ChloroformChloroform <td></td> <td></td> <td></td> <td>· ·</td>				· ·	
Choline-1CH, b. 3.19 (a)AqueousCholine-2CH, CH, M 3.20 (b), 3.55 (m)AqueousBetaineCH, CH, 3.26 (b), 3.91 (a) $Aqueous$ SucroseHib, H9, OH; H17, 19, H5, H3, H7 3.46 (d), 3.54 (d), 3.66 (s), 3.81 (m), 4.05 (d), 4.21 (d), 5.44 (d)GlycineCH 3.55 (d), 3.55 (d), 3.56 (d), 3.77 (m)AqueousGlycrodCH0H, CHOH, COH 3.55 (d), 3.65 (d), 3.77 (m)Aqueous β -glucoseCH-1, CH-6 3.81 (m), 5.22 (d)Aqueous β -glucoseCH-1 4.54 (d)AqueousUraclH0, H-6 5.80 (d), 7.52 (d)AqueousVancoside/NucleosideCH $7.1-79$ AqueousFematateCH 6.51 (s)AqueousNacloside/NucleosideCH- 7.9 3.22 (m), 4.27 (m)ChloroformPhosphatidylcholineN(CH ₂), CH ₂ OP 3.22 (m), 4.27 (m)ChloroformPhosphatidylgverolGlycerol moiety 3.61 (m)ChloroformPhosphatidylgverolCH ₂ -cm (MeH) $3.1-3.9$ ChloroformPhosphatidylsverideCH ₂ -main 4.35 (m)ChloroformDiglycerideCH (G) 4.31 (m), 5.16 (m)ChloroformDiglycerideCH (S-1, CH-CH, CH) 6.14 (m), 6.36 (m)ChloroformDiglycerideCH (G) 2.35 (m)ChloroformCharoformCHacol, CH-CH, CH-CH, CH, CH, CH $5.8-6.8$ ChloroformChloroformCHacol, CH-CH-C-CH $5.8-6.8$ Chloroform		-		^	
Cholme-2 CH ₀ , CH ₂ NH 3.20 (a), 3.55 (m) Aqueous Betaine CH ₀ , CH ₂ 3.26 (a), 3.91 (a) Aqueous Sucrose H10, H9, OH; H17, 19, H5, H3, H7 3.46 (b), 3.54 (d), 3.66 (a), 3.81 (m), 4.05 (b), 4.21 (d), 5.40 (d) Aqueous Glycenol CHOH, CHOH, COH 3.55 (a), 3.65 (d), 3.57 (m) Aqueous α -glucose CH-1. 4.63 (d) Aqueous β -glucose CH-1. 4.63 (d) Aqueous Uradi H5, H6 5.88 (d), 7.52 (d) Aqueous Nacloside/Nacloside CH 5.88 (d), 7.52 (d) Aqueous Nacloside/Nacloside CH 3.22 (m), 4.27 (m) Aqueous Pionarate CH 8.45 (a) Aqueous Piospharidylendim NC(L), J., CH, OP 3.21 (m), 4.27 (m) Chioroform Phospharidylextrine CH ₃ on 4.05 (m) Chioroform Phospharidylextrine CH ₃ oP 3.1-3.9 Chioroform Phospharidylextrine CH ₃ OP 4.28 (m) Chioroform Disporbinid CH ₂ -m ³ 4.05 (m), 4.14 (m),				^	
Betaine CH ₂ , CH ₂ 3.26 (a), 391 (a) Aqueous Sucrose H10, H9, OH, H17, 19, H5, H3, H7 3.46 (b), 3.54 (d), 3.66 (b), 5.81 (m), 4.05 (b), 4.21 (d), 5.40 (d) Aqueous Glycrol CH OH, CHOH, COH 3.55 (d), 3.57 (d), 3.56 (d), 3.77 (m) Aqueous orguose CH-1, CH-6 3.81 (m), 5.52 (d) Aqueous β-glucose CH-1 6.63 (d), 7.57 (d) Aqueous Market H6, H-6 5.8 (d), 7.52 (d) Aqueous Vurale H3, H6 5.8 (d), 7.52 (d) Aqueous Nuclosid/CNacloside CH Aqueous Aqueous Nuclosid/CNacloside CH Aqueous Aqueous Nuclosid/CNacloside NCH ₃), CH ₂ OP 32 (m), 427 (m) Chloroform Opsphatidylcholine N(CH ₃), CH ₂ OP 32 (m), 427 (m) Chloroform Phosphatidylcholine CH ₂ -a3 435 (m) Chloroform Disphatidylcholine CH ₂ -a 435 (m) Chloroform Disphatidylcholine CH ₂ -a 431 (m), 514 (m), 526 (m) Chloroform Disphotidylchorine <td< td=""><td></td><td>-</td><td></td><td>^</td></td<>		-		^	
Sucrose H0: H9: OH: H17.19: H3, H3, H7 3.46 (1), 3.54 (d), 3.66 (a), 3.81 (m), 4.05 (t), 4.21 (d), 5.44 (d), 4queous Glycerol CH 3.55 (a) Aqueous Glycerol CHO, CHOH, COH 3.55 (d), 3.67 (m) Aqueous glucose CH-1, CH-6 3.81 (m), 5.22 (d) Aqueous glucose CH-1 4.63 (d) Aqueous Uracil H5, H6 5.89 (t), 7.52 (d) Aqueous Uracil H5, H6 5.89 (t), 7.52 (d) Aqueous Fumarate CH CH Aqueous Aqueous Formate Formate Aqueous Aqueous Phosphatidylcholine N(CH ₃), CH ₂ OP 3.22 (m), 4.27 (m) Chloroform Phosphatidylgycerol Glycerol molety 3.61 (m) Chloroform Phosphatidylgycerol Glycerol molety 3.61 (m) Chloroform Dispholidylgycerol CH ₂ As a 4.05 (m) Chloroform Dispholidylgycerol CH ₂ As a 4.05 (m) Chloroform Dispholidylgycerol CH ₂ As a 4.05 (m) 4.01 (m)				^	
GlycineCH3.55 (a)AqueousGlycorolCHOH, CHOH, COH3.55 (d), 3.57 (m)Aqueouso-glucoseCH-1, CH-63.81 (m), 5.22 (d)AqueousB-glucoseCH-14.63 (d)AqueousUraclH5, H65.8 (d), 7.52 (d)AqueousUradlH(a), H-65.89 (t), 7.86 (d)AqueousNucleoside/NucleosideCH5.81 (s)AqueousNucleoside/Nucleoside7.1-7.9AqueousFormateCH8.45 (s)AqueousGlyceroglycolipidsCH, sn34.05 (m)ChloroformPhosphatidylcholineN(CH ₃), CH ₂ OP3.22 (m), 4.27 (m)ChloroformPhosphatidylgycerolGlycerol molety3.61 (m)ChloroformPhosphatidylgycerolGlycerol molety3.61 (m)ChloroformPhosphatidylgycerolCH ₂ OP4.22 (m)ChloroformPhosphatidylgycerolCH ₂ OP4.31 (m), 4.14 (m), 5.26 (m)ChloroformDacydglycerolposhophipidCH ₃ .sn34.05 (m)ChloroformTriglycerideCH, sn1, 3(h), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualene4.CH, ya terminal CH ₃ =CHCH ₃ 1.61 (d), 1.64 (m), 6.35 (m)ChloroformSqualene4.CH, ya terminal CH ₃ =CHCH ₃ 1.61 (m) (m)ChloroformCaroenoidsC=CH, CH=CH, CH-C(CH ₃)6.14 (m), 6.44 (m), 6.31 (m)ChloroformSterolCH ₂ -Co2.32 (triplet of dublets)ChloroformDefinic protons (alkero)CH-CH2.86 (m)Chloroform				-	
GlycerolCHOH, CHOH, COH3.55 (d), 3.65 (d), 3.77 (m)Aqueous α -glucoseCH-1, CH-63.81 (m), 5.22 (d)Aqueous β -glucoseCH-14.63 (d)Aqueous β -glucoseCH-15.8 (d), 7.52 (d)AqueousUraclH5, H65.8 (d), 7.52 (d)AqueousUradineH(a), H-65.89 (h), 7.86 (d)AqueousFumardeCH6.51 (s)AqueousNackeoside/Nucleoside7.1-7.9AqueousFormateNCH_1) ₀ , CH_2OP3.22 (m), 4.27 (m)ChloroformGlyceroglycolipidsCH, -sn34.05 (m)ChloroformPhosphatidylgbyerolGlycerol moiety3.61 (m)ChloroformPhosphatidylgbyerolGlycerol moiety3.1-3.9ChloroformDiscylgyceropholpidCH, -sn34.05 (m)ChloroformDiscylgyceropholpidCH, sn34.05 (m)ChloroformDiscylgyceropholpidCH, sn34.05 (m)ChloroformDiscylgyceropholpidCH, sn34.05 (m)ChloroformSqualeneCH - GL4.31 (m), 5.13 (m)ChloroformSqualeneCH - GL5.8 - 6.8ChloroformSqualeneCH-CHCCH5.8 - 6.8ChloroformAldehydeCH-O2.32 (triplet of sublets)ChloroformAldehydeCH-O5.8 - 6.8ChloroformAldehydeCH-O2.32 (triplet of sublets)ChloroformPiaceCH_C-C-CH5.8 - 6.8ChloroformAldehydeCH_C-C-C2.32 (triplet					
α-glucoseCH-1, CH-63.81 (m), 5.22 (d)Aqueousβ-glucoseCH-14.63 (d)AqueousUracilH5, H65.88 (d), 7.52 (d)AqueousUracinH(a), H-65.89 (t), 7.86 (d)AqueousFumarateCH6.51 (s)AqueousFormateCH6.51 (s)AqueousFormateMCH-3), CH, QP3.22 (m), 4.27 (m)ChloroformBrosphatidylcholineN(CH, 3), CH, QP3.22 (m), 4.27 (m)ChloroformBrosphatidylgyerolGlycerol mely3.61 (m)ChloroformPhosphatidylgyerolGlycerol mely3.61 (m)ChloroformPhosphatidylgyerolCH, Sn 34.05 (m)ChloroformPhosphatidylgyerolCH, Sn 34.05 (m)ChloroformDiscylgycerophospholipidCH, Sn 34.05 (m)ChloroformDiscylgyceroleCH sn 1, 3(a), CH sn 1, 3(b), CH sn 24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn 1, 3(a), CH sn 1, 3(b), CH sn 24.31 (m), 5.13 (m)ChloroformSqualene4xCH ₂ , 2x terminal CH ₃ =CHCH ₁ 1.61 (d), 1.69 (d), 2.04 (m)ChloroformSqualeneCH-CH, CH-C(CH, 1)6.14 (m), 6.64 (m), 6.35 (m)ChloroformOlefnic protons (alkene)CH=CH, CH-CCH5.8-6.8ChloroformOlefnic protons (alkene)CH=CH/-C=CH2.32 (triplet of doublets)ChloroformDiffer du addCH_CO2.32 (doublet of singlet)ChloroformDiffer du addCH_CO2.32 (doublet of singlet)Chloroform <t< td=""><td></td><td></td><td></td><td>·</td></t<>				·	
β_{-} glucoseCH-14.63 (d)AqueousUraclH5, H65.8 (d), 7.52 (d)AqueousUridineH(a), H-65.8 (d), 7.86 (d)AqueousNucleoside/NucleosideCH6.51 (s)AqueousNucleoside/Nucleoside7.1-7.9AqueousPormateImage: CH8.45 (s)AqueousOperateCH-8.45 (s)AqueousPhosphatidylgbycerolGlycerolycolydisCH ₂ -Sn34.05 (m)ChloroformDisophatidylgbycerolGlycerol noiety3.61 (m)ChloroformPhosphatidylgbycerolGlycerol noiety3.61 (m)ChloroformPhosphatidylgbycerolCH ₂ -Sn34.05 (m)ChloroformDiscrylgbyceroloshipidCH ₂ -Sn34.05 (m)ChloroformDiscrylgbyceroloshipidCH ₂ -Sn34.05 (m)ChloroformTiglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 5.13 (m)ChloroformSqualene4.5C μ_2 -St terminal CH ₂ -CHCH1.61 (d), 1.69 (d), 2.04 (m)ChloroformSqualene4.5C μ_2 -St terminal CH ₂ -CHCH5.8 - 6.8ChloroformPhacophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformPheaphytidCH ₂ -C-CH5.8 - 6.8ChloroformFere fatty acidCH ₂ -CO2.32 (tripted odublets)ChloroformPUFABs-allyt chainsi.0 1.82-2.77 (m) ilo 1.8-2.84 (m)ChloroformPUFABs-allyt chainsi.0 1.82-2.77 (m) ilo 1.8-2.92.84 (m)ChloroformAldehydeImageN-CH				· ·	
UracilH5,H658 (d), 7.52 (d)AqueousUridineH(a), H-65.89 (t), 7.86 (d)AqueousFumarateCH6.51 (s)AqueousFunceside/Vacleoside7.1-7.9AqueousFormateK8.45 (s)AqueousProsphatidy/tholineN(CH ₃), CH ₂ OP3.22 (m), 4.27 (m)ChloroformDisophatidy/thylerolGlycerol moiety3.61 (m)ChloroformPhosphatidy/thylerolGlycerol moiety3.61 (m)ChloroformPhosphatidy/thylerolCH ₂ OP4.28 (m)ChloroformDiacylglycerolnoshphilpidCH ₂ -sn34.05 (m)ChloroformDiacylglycerohoshphilpidCH ₃ -sn34.05 (m)ChloroformDiacylglycerohoshphilpidCH ₃ -sn34.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn1, X(a), CH sn1, X(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualene4xCH ₃ -2x terminal CH ₃ =CHCH ₃ 1.61 (d), 1.69 (d), 2.04 (m)ChloroformSqualene4xCH ₃ -2x terminal CH ₃ =CHCH ₃ 0.51 (m)ChloroformSqualeneCH ₂ -1, CH-CH, CH-CCH ₃ 6.41 (m), 6.64 (m), 6.35 (m)ChloroformPaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformPaeophytinN-H (pophyrin ring)1.61 (2.2-27 (m) ii) C18.3-2.81 (m) iii) C2.6-2.84 (m)ChloroformPUFABi-Ballytic Anian0.81 (m)ChloroformPUFAN=GL_GOO2.32 (tripted of oublets)ChloroformPUFAN=GL_GOO1.61 (m)Chloroform<	•			· ·	
UridineHa), H-6589 (b), 7.86 (d)AqueousFumarateCH6.51 (s)AqueousNucleoside/Nucleoside7.1-7.9AqueousFormateNCH_0, Jp, CH_OP3.22 (m), 4.27 (m)ChloroformBhosphatidylcholineN(CH_1, Jp, CH_OP3.22 (m), 4.27 (m)ChloroformGlyceroglycolipidsCH ₃ sn34.05 (m)ChloroformPhosphatidylgycerolGlycerol noiety3.61 (m)ChloroformPhosphatidylgycerolCH ₂ OP4.28 (m)ChloroformDisophatidylserineCH ₃ , sn34.05 (m)ChloroformDisophatidylserineCH ₃ , CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformDisophatidylserineCH sn1, CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualene4.54, Jp, z terminal CH ₂ =CHCH ₂ 4.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualeneCH, St, Sn3ChloroformChloroformSqualeneCH, CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformGlycerideCH sn1, CH ₂ , CHCH ₂ ChloroformChloroformSteridCH ₂ , CH-CH, CH-C(CH ₃)6.14 (m), 6.64 (m), 6.35 (m), 9.51 (m), 9				· ·	
Humarate CH 6.15 (s) Aqueous Nucleoside/Nucleoside 7.1-7.9 Aqueous Formate 8.45 (s) Aqueous Phosphatidylcholine N(CH ₂) ₂ , CH ₂ OP 3.22 (m), 4.27 (m) Chloroform Glyceroglycolpids CH ₂ -0 3.21 (m), 4.27 (m) Chloroform Phosphatidylgycerol Giycerol moiety 3.61 (m) Chloroform Diacylglycerophospholipid CH ₂ -0 4.28 (m) Chloroform Diacylglycerohospholipid CH _{3.0} , CH sn1, 3(b), CH sn2 4.31 (m), 5.13 (m) Chloroform Squalene 4K-2H _{3.2} Xerminal CH ₂ -CHCH ₂ 1.61(d), 1.69(d), 2.04(m) Chloroform Sterol CH ₂ -18 6.53 (m) Chloroform Glycerolynin N-H (opphyrin ring) -1.41 (s), -1.61 (s) Chloroform Defator CH ₂ -C=CH 2.32 (tripted doublets)				-	
Nucleoside/Nucleoside Instruct Instruct Aqueous Formate 8.45 (s) Aqueous Phosphatidylcholine N(CH,) ₃ , CH ₂ OP 3.22 (m), 4.27 (m) Chloroform Glyceroglycolipids CH ₃ -sn3 4.05 (m) Chloroform Phosphatidylghycerol Glycerol moiety 3.61 (m) Chloroform Phosphatidylghycerol Glycerol moiety 3.1-3.9 Chloroform Phosphatidylghycerol CH ₂ OP 4.28 (m) Chloroform Diacylghycerophospholipid CH ₃ -sn3 4.05 (m) Chloroform Diacylghycerophospholipid CH ₃ -sn3 4.05 (m) Chloroform Siglyceride CH sn1, 3(h), CH sn12 4.31 (m), 5.26 (m) Chloroform Siglacend 4xCH ₃ ax terminal CH ₃₉ -CHCH ₂ 1.61(d), 1.69 (d), 2.04(m) Chloroform Sterol CH ₃ -18 0.53 (m) Chloroform Diacylghytin N-H (pophyrin ring) -141 (s) -161 (s) Chloroform Aldehyde CH=CH - CCH S.8 -6.8 Chloroform Definer potons (alkere) CH=CH - CCH				· ·	
Formate Representation Representation Representation Representation Phosphatidylcholine N(CH ₃) ₃ , CH ₂ OP 3.22 (m), 4.27 (m) Chloroform Glyceroglycolipids CH ₃ -sn3 4.05 (m) Chloroform Phosphatidylgycerol Glycerol molety 3.61 (m) Chloroform Phosphatidylgycerol CH ₂ OP 4.28 (m) Chloroform Diacylglycerophospholipid CH ₃ -Sn3 4.05 (m) Chloroform Diacylglycerophospholipid CH ₃ -Sn3 4.05 (m) Chloroform Triglyceride CH sn1, 3(a), CH sn1, 3(b), CH sn2 4.31 (m), 5.13 (m) Chloroform Squalen 4x CH ₃ -2x terminal CH ₃ -eCH CH ₂ 1.61 (d), 1.69 (d), 2.04 (m) Chloroform Squalen 4x CH ₃ -2x terminal CH ₃ -eCH CH ₂ 1.61 (d), 1.69 (d), 2.04 (m) Chloroform Sterol CH ₂ -18 0.53 (m) Chloroform Chloroform Sterol CH ₂ -18 0.53 (m) Chloroform Chloroform Diadylyce CH CH CH, CH-CCH 5.8 - 6.8 Chloroform Chloroform Piter Attrigo Ch C		СН		· ·	
Phosphatidylcholine NCH ₃)s, CH ₂ OP 3.22 (m), 4.27 (m) Chloroform Glyceroglycolipids CH ₃ -sn3 4.05 (m) Chloroform Phosphatidylgycerol Glycerol moiety 3.61 (m) Chloroform Phosphatidylgycerol Glycerol moiety 3.61 (m) Chloroform Phosphatidylgycerol Glycerol moiety 3.61 (m) Chloroform Phosphatidylgycerol CH ₂ OP 4.28 (m) Chloroform Diacylglycerophospholipid CH ₂ OP 4.31 (m), 5.13 (m) Chloroform Diacylglycerophospholipid CH ₃ -sn3 4.05 (m) Chloroform Squalen KxCH ₃ -2x terminal CH ₃ =CHCH ₂ 4.31 (m), 5.13 (m) Chloroform Squalen KxCH ₃ -2x terminal CH ₃ =CHCH ₂ 1.61 (d), 1.69 (d), 2.04 (m) Chloroform Sterol CH ₃ -1 CH-1/C-CCH 5.8-6.8 Chloroform Pheaophytin N-H (pophyrin ring) -1.41 (s), -1.61 (s) Chloroform Sterol CH ₂ -C=O 2.32 (tripted doublets) Chloroform Pter Aty CH ₂ C=O 2.32 (tripted of doublets) Chloroform				-	
GlyceroglycolipidsCH_3 sn 34.05 (m)ChloroformPhosphatidylgycerolGlycerol moiety3.61 (m)ChloroformPhospholipid3.1-3.9ChloroformPhospholipidCH_2OP4.28 (m)ChloroformDiacylglycerophospholipidCH sn 3 (a), CH sn 1, 3(b), CH sn 24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn 1, 3(a), CH sn 1, 3(b), CH sn 24.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualene4xCH ₃ 2x terminal CH ₃ .=CHCH ₂ 1.61 (d), 1.69 (d), 2.04 (m)ChloroformSterolCH180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH_3)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhacophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefnic protons (alkene)CH=CH-C=CH5.8-6.8ChloroformSterolCH_2C=O2.32 (triplet of doublets)ChloroformFree fatty acidCH_2C=O2.35 (doublet of singlet)ChloroformPUFABis-allylic chainsi) C18:2-2.77 (m) ii) C18:3-2.81 (m) iii) C2:2-6-2.84 (m)ChloroformDiff chainsi) C18:2-2.77 (m) ii) C18:3-2.81 (m) iii) C2:2-6-2.84 (m)ChloroformGlycadi Chainsi) C18:2-2.77 (m) ii) C18:3-2.81 (m) iii) C2:6-0.84 (m)ChloroformPUFA methyl endn-CH_30.88 (m)ChloroformJi Aldyl chain(CH_2-n1.61 (m)ChloroformAldyl chain(CH_2-n2.34 (m)ChloroformAlkyl chainCH_2OO-1.61 (m)ChloroformAlkyl chain <td></td> <td></td> <td></td> <td>· ·</td>				· ·	
DisplativlglycerolGlycerol moiety $3.61 (m)$ ChloroformPhosphativlglycerolGlycerol moiety $3.1-3.9$ ChloroformPhosphatidylserineCH ₂ OP $4.28 (m)$ ChloroformDiacylglycerophospholipidCH, sn3 $4.05 (m)$ ChloroformTriglycerideCH sn1, $3(a)$, CH sn1, $3(b)$, CH sn2 $4.31 (m)$, $5.13 (m)$ ChloroformSqualene $4xCH_9$, $2x$ terminal CH ₂ =CHCH ₂ $1.61 (d)$, $1.69 (d)$, $2.04 (m)$ ChloroformSterolCH, -18 $0.53 (m)$ ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH ₂) $6.14 (m)$, $6.64 (m)$, $6.35 (m)$ ChloroformPhacophytinN-H (pophyrin ring) $-1.41 (s)$, $-1.61 (s)$ ChloroformOlefinic protons (alkene)CH=CH/-C=CH $5.8-6.8$ ChloroformSterolCH ₂ C=O $2.32 (triplet of doublets)$ ChloroformFree fatty acidCH ₂ C=O $2.35 (doublet of singlet)$ ChloroformPUFABis-allylic chainsi) C18:2-2.77 (m) ii) C18:3-2.81 (m) iii) C2:2-2.84 (m)ChloroformDiff chainsi) C18:2-2.77 (m) ii) C18:3-2.81 (m) iii) C2:2-2.84 (m)ChloroformAlkyl chain $n-CH_3$ $0.98 (m)$ ChloroformAlkyl chain $n-CH_3$ $0.98 (m)$ ChloroformAlkyl chainCH_2CO- $2.44 (m)$ ChloroformCarboxylic end*CH_2CO- $2.44 (m)$ ChloroformAlkyl chain (allylic)CH_2CH=CHCH=CHCH_3 $2.06 (m)$ ChloroformCarboxylic end*CH_2CO- $2.44 (m)$ Chloroform <t< td=""><td>· · ·</td><td></td><td></td><td></td></t<>	· · ·				
Phospholipid3.1-3.9ChloroformPhospholipidCH_OP4.28 (m)ChloroformDiacylglycerophospholipidCH_3-sn34.05 (m)ChloroformTriglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformSqualene4xCH_3.2 x terminal CH_3=CHCH_21.61 (d), 1.69 (d), 2.04(m)ChloroformSterolCH_3-180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH_3)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPlaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CH/-C=CH5.8-6.8ChloroformAldehydeCH=O8.0 (m), 8.55 (m), 9.40 (m), 9.53 (m), 10.4 (m)ChloroformEsterCH_2C=O2.32 (triplet of dublets)ChloroformPUFABis-allylic chainsi) C 18.2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m)ChloroformPUFAn-CH_30.88 (m)ChloroformPUFA methyl endn-CH_30.88 (m)ChloroformAklyl chain(CH_2-)_n1.25 (m)ChloroformCarboxylic end*CH_4CO-2.34 (m)ChloroformCarboxylic end*CH_4CO-2.34 (m)ChloroformDiff comCH_4CO-2.34 (m)ChloroformOutre formCH_4CO-2.34 (m)ChloroformPUFA methyl endn-CH_50.98 (m)ChloroformCarboxylic end* <t< td=""><td></td><td></td><td></td><td></td></t<>					
PhosphaidylserineCH2OP4.28 (m)ChloroformDiacylglycerophospholipidCH3,-sn34.05 (m)ChloroformTriglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn1, CH sn24.31 (m), 5.13 (m)ChloroformSqualene4xCH3, 2x terminal CH3,=CHCH21.61 (d), 1.69 (d), 2.04 (m)ChloroformSterolCH3,-180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CHC, CH3)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CO8.0 (m), 8.55 (m), 9.40 (m), 9.53 (m), 10.4 (m)ChloroformEsterCH2C=O2.32 (riplet of doublets)ChloroformFree fatty acidCH2C=O2.35 (doublet of singlet)ChloroformPUFABis-allylic chainsi) C1 8:2-2.77 (m) ii) C1 8:3-2.81(m) iii) C22:6-2.84(m)ChloroformPUFAn-CH30.98 (m)ChloroformChloroformAlkyl chain(CH2-O)1.25 (m)ChloroformChloroformAlkyl chain (allylic)CH2,CO-1.61 (m)ChloroformChloroformJkyl chain (allylic)CH2,CO-2.34 (m)ChloroformCarboxylic end*CH2,CO-2.34 (m)ChloroformAlkyl chain (allylic)CH2,CH=CHCH=CHCH22.06 (m)ChloroformAlkyl chain (allylic)CH2,CO-2.34 (m)ChloroformCarboxylic end*CH2,CO-2.34 (m)Chloroform <td>1 , 0,</td> <td>Glycerol moiety</td> <td></td> <td></td>	1 , 0,	Glycerol moiety			
DiacylglycorophospholipidCH, sn34.05 (m)ChloroformTriglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn1, CH sn24.31 (m), 5.13 (m)ChloroformSqualene4xCH ₃ , 2x terminal CH ₃ ,=CHCH ₂ 1.61(d), 1.69 (d), 2.04(m)ChloroformSterolCH ₃ -180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH ₃)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CH/-C=CH5.8-6.8ChloroformEsterCH ₂ C=O8.0 (m), 8.55 (m), 9.40 (m), 9.5 (m), 9.53 (m), 10.4 (m)ChloroformFree fatty acidCH ₂ C=O2.32 (triplet of doublets)ChloroformPUFABis-allylic chainsi) C 18:2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m)ChloroformPUFAn-CH ₂ 0.88 (m)ChloroformPUFA methyl endn-CH ₃ 0.98 (m)ChloroformAlkyl chain(-CH ₂ -) _n 1.61 (m)ChloroformAlkyl chain (allylic)CH ₂ CO-2.34 (m)ChloroformCarboxylic end*CH ₂ CO-2.34 (m)ChloroformCarboxylic end*CH ₂ CO-2.34 (m)ChloroformAlkyl chain (bisallylic)CH ₂ CH=CH(-H ₂ CH=CH ₂)2.60 (m)ChloroformChloroformCH ₂ CO-2.34 (m)ChloroformAlkyl chain (bisallylic)CH ₂ CO-C2.34 (m)ChloroformCarboxylic end*C					
TriglycerideCH sn1, 3(a), CH sn1, 3(b), CH sn24.31 (m), 4.14 (m), 5.26 (m)ChloroformDiglycerideCH sn1, CH sn24.31 (m), 5.13 (m)ChloroformSqualene $4xCH_3$, $2x$ terminal CH $_{39}$ =CHCH $_2$ 1.61(d), 1.69 (d), 2.04(m)ChloroformSterolCH $_3$ -180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH $_3$)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CH/-C=CH5.8-6.8ChloroformAldehydeCH=O8.0 (m), 8.55 (m), 9.40 (m), 9.5 (m), 9.53 (m), 10.4 (m)ChloroformEsterCH,C=O2.32 (triplet of doublets)ChloroformFree fatty acidCH,C=O2.35 (doublet of singlet)ChloroformPUFABia-Puly in chainsi) C 18:2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m)Chloroformb) Fatty Acid Chainsn-CH_30.88 (m)ChloroformAlkyl chain(-CH_2-)n1.25 (m)ChloroformAlkyl chain(-CH_2-)n1.25 (m)ChloroformAlkyl chainCH_2,CO-2.34 (m)ChloroformAlkyl chain (allylic)CH_2,CH=CH $_3$ 2.06 (m)ChloroformCarboxylic end*CH_2,CO-2.34 (m)ChloroformAlkyl chain (allylic)CH_2,CH=CH $_3$ 2.06 (m)ChloroformCarboxylic end*CH_2,CO-2.34 (m)ChloroformAlkyl chain (allylic)CH=CH(CH_2,CH=CH)_n2.81 (m)ChloroformDouble bonds<		-			
DiglycerideCH sn1, CH sn24.31 (m), 5.13 (m)ChloroformSqualene $4xCH_{3\nu}$ 2x terminal CH_{3=}CHCH21.61(d), 1.69 (d), 2.04(m)ChloroformSterolCH_3-180.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH3)6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhaeophytinN-H (pophyrin ring)-1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CH/-C=CH5.8-6.8ChloroformAldehydeCH=O8.0 (m), 8.55 (m), 9.40 (m), 9.53 (m), 10.4 (m)ChloroformEsterCH2C=O2.32 (triplet of doublets)ChloroformFree fatty acidCH2C=O2.35 (doublet of singlet)ChloroformPUFABi-allylic chainsi) C 18:2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m)ChloroformDeff and n-CH30.88 (m)ChloroformChloroformPUFA methyl endn-CH30.98 (m)ChloroformAlkyl chain(-CH2-)n1.25 (m)ChloroformAlkyl chain (allylic)CH2-CO-2.34 (m)ChloroformAlkyl chain (bisallylic)-CH=CH(CH2CH=CH)_n2.81 (m)ChloroformAlkyl chain (bisallylic)-CH=CH(CH2CH=CH)_n2.81 (m)ChloroformDuble bonds-CH=CH-5.4 (m)ChloroformAlkyl chain(-CH3-)n2.81 (m)ChloroformAlkyl chainCH2_0n1.25 (m)ChloroformChloroform2.81 (m)ChloroformChloroformAlkyl chain (bisallylic)-CH=CH-5.4 (m)Chloroform					
Squalene $4xCH_3, 2x terminal CH_3=CHCH_2$ $1.61(d), 1.69(d), 2.04(m)$ Chloroform Sterol CH_3-18 $0.53 (m)$ Chloroform Carotenoids $C=CH, CH=CH, CH-C(CH_3)$ $6.14 (m), 6.64 (m), 6.35 (m)$ Chloroform Phaeophytin N-H (pophyrin ring) $-1.41 (s), -1.61 (s)$ Chloroform Olefinic protons (alkene) CH=CH/-C=CH $5.8-6.8$ Chloroform Aldehyde CH=Q $8.0 (m), 8.55 (m), 9.40 (m), 9.5 (m), 9.53 (m), 10.4 (m)$ Chloroform Ester CH_2C=Q $2.32 (triplet of doublets)$ Chloroform Free fatty acid CH_2C=Q $2.35 (doublet of singlet)$ Chloroform PUFA Bis-allylic chains i) C 18:2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m) Chloroform PUFA methyl end n-CH_3 0.98 (m) Chloroform PUFA methyl end n-CH_4 0.98 (m) Chloroform Carboxylic end CH_2CO- 1.61 (m) Chloroform Carboxylic end CH_2CO- 2.34 (m) Chloroform Carboxylic end* CH_2CO- 2.34 (m) Chloroform			4.31 (m), 4.14 (m), 5.26 (m)		
Sterol CH_3-18 0.53 (m)ChloroformCarotenoidsC=CH, CH=CH, CH-C(CH_3) 6.14 (m), 6.64 (m), 6.35 (m)ChloroformPhaeophytinN-H (pophyrin ring) -1.41 (s), -1.61 (s)ChloroformOlefinic protons (alkene)CH=CH/-C=CH $5.8-6.8$ ChloroformAldehydeCH=O 8.0 (m), 8.55 (m), 9.40 (m), 9.53 (m), 10.4 (m)ChloroformEsterCH_2C=O 2.32 (triplet of doublets)ChloroformFree fatty acidCH_2C=O 2.35 (doublet of singlet)ChloroformPUFABis-allylic chainsi) C 18:2-2.77 (m) ii) C 18:3-2.81(m) iii) C 22:6-2.84(m)Chloroform b Fatty Acid Chainsn-CH_30.98 (m)ChloroformPUFA methyl endn-CH_30.98 (m)ChloroformAlkyl chain(-CH_2-)_n1.25 (m)ChloroformCarboxylic end*CH_2COO-2.34 (m)ChloroformAlkyl chain (allylic)CH_2CH=CHCH=CHCH_22.06 (m)ChloroformAlkyl chain (bisallylic)-CH=CH(CH_2CH=CH)_n2.81 (m)ChloroformAlkyl chain (bisallylic)-CH=CHCH_5 44 (m)ChloroformAlkyl chaini-CH=CH-5.4 (m)ChloroformAlkyl chaini-CH=CH-5.4 (m)ChloroformAlkyl chaini-CH=CH-5.4 (m)ChloroformDouble bonds-CH=CH-5.4 (m)ChloroformAlkyl chaini-CH=CH-5.4 (m)ChloroformOutbe formi-CH=CH-5.4 (m)ChloroformDouble bonds-CH=CH-		CH sn1, CH sn2		Chloroform	
Carotenoids C=CH, CH=CH, CH=C(CH ₃) 6.14 (m), 6.64 (m), 6.35 (m) Chloroform Phaeophytin N-H (pophyrin ring) -1.41 (s), -1.61 (s) Chloroform Olefinic protons (alkene) CH=CH/-C=CH 5.8–6.8 Chloroform Aldehyde CH=O 8.0 (m), 8.55 (m), 9.40 (m), 9.5 (m), 9.53 (m), 10.4 (m) Chloroform Ester CH ₂ C=O 2.32 (triplet of doublets) Chloroform Free fatty acid CH ₂ C=O 2.35 (doublet of singlet) Chloroform PUFA Bis-allylic chains i) C18:2-2.77 (m) ii) C18:3-2.81(m) iii) C 22:6-2.84(m) Chloroform b Fatty Acid Chains n-CH ₃ 0.88 (m) Chloroform PUFA Bis-allylic chains 0.98 (m) Chloroform PUFA methyl end n-CH ₃ 0.88 (m) Chloroform Rubyl chain (-CH ₂ -) _n 1.25 (m) Chloroform Carboxylic end CH ₂ CO- 2.34 (m) Chloroform Chloroform CH ₂ CO- 2.34 (m) Chloroform Carboxylic end* CH ₂ CO- 2.34 (m) Chloroform Ch	Squalene	4xCH ₃ , 2x terminal CH ₃ ,=CHCH ₂	1.61(d), 1.69 (d), 2.04(m)	Chloroform	
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Carboxylic end*CH_2COO-2.34 (m)ChloroformAlkyl chain (bisallylic)-CH=CH(CH_2CH=CH)_n2.81 (m)ChloroformDouble bonds-CH=CH-5.4 (m)ChloroformPUFA methyl end $n-CH_3$ 0.98 (m)ChloroformAlkyl chain(-CH_2-)_n1.25 (m)Chloroform	Carboxylic end	C <u>H</u> ₂ COO-	1.61 (m)	Chloroform	
Alkyl chain (bisallylic) -CH=CH(CH ₂ CH=CH) _n 2.81 (m) Chloroform Double bonds -CH=CH(CH ₂ CH=CH) _n 5.4 (m) Chloroform PUFA methyl end n-CH ₃ 0.98 (m) Chloroform Alkyl chain (-CH ₂ -) _n 1.25 (m) Chloroform	Alkyl chain (allylic)	$C\underline{H}_2CH=CHCH=CHC\underline{H}_2$	2.06 (m)	Chloroform	
Double bonds -CH=CH- 5.4 (m) Chloroform PUFA methyl end n-CH_3 0.98 (m) Chloroform Alkyl chain (-CH_2-)_n 1.25 (m) Chloroform	Carboxylic end*	C <u>H</u> ₂ COO-	2.34 (m)	Chloroform	
PUFA methyl end n-CH ₃ 0.98 (m) Chloroform Alkyl chain (-CH ₂ -) _n 1.25 (m) Chloroform	Alkyl chain (bisallylic)	-CH=CH(CH ₂ CH=CH) _n	2.81 (m)	Chloroform	
Alkyl chain (-C <u>H</u> ₂ -) _n 1.25 (m) Chloroform	Double bonds	-CH=CH-	5.4 (m)	Chloroform	
	PUFA methyl end	n-C <u>H</u> ₃	0.98 (m)	Chloroform	
	Alkyl chain	(-C <u>H</u> ₂ -) _n	1.25 (m)	Chloroform	
	Double bonds		5.4 (m)	Chloroform	

Table 1. Proton chemical shifts observed in *Scenedesmus regularis* treated with different nitrate concentration(a) metabolite and (b) fatty acid fraction.

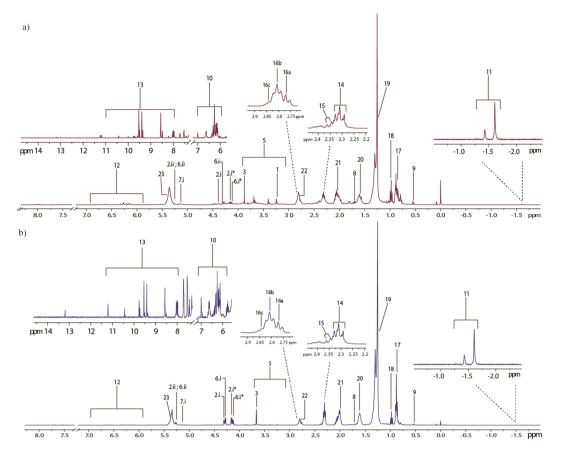


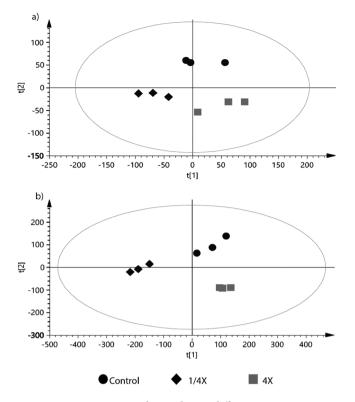
Figure 4. 1H NMR spectra of *S. regularis* extracted using chloroform solvent from (**a**) 1/4X Nitrate and (**b**) 4X Nitrate treatment. Metabolites marked with an asterisk appear only in the treatment sample and not in control. Key for spectra refer to Table 1.

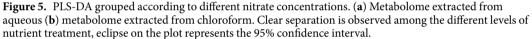
Methyl group of C-18 sterol structures of different sterols often resonate at different ppm. β -sitosterol for example, resonates at δ 0.68 while stigmasterol produces a signal at δ 0.7^{40,41}. As a general rule, C-18 moiety of methyl groups belonging to sterol compounds resonates at δ 0.6– δ 0.8⁴¹. Our study furnished a signal at δ 0.53, pointing to presence of ergosterol-like sterols in the mixture. Microalgae may synthesis either glycerides or fatty alcohols as their main neutral lipid storage⁴³. Triacylglycerides (TAG) are the major storage lipids in a majority of microalgae⁴⁴. Spectral ID returns glycerol moieties detected at δ 4.31, δ 4.14 and δ 5.26.

Chemical shifts coinciding with signals related to phospholipids are usually detected in the region δ 3.0– δ 4.05^{40,45}. Higher intensity signals (characteristic of phospholipid head group signals) were detected between the region of δ 3.1– δ 4.05 in nutrient abundant and control spectra in comparison to nutrient limited spectra suggesting reduced phospholipid production under nutrient limitation. Signals corresponding to phosphate head groups (CH₂OP) connected to phosphatidylcholine and phosphatidylserine were also detected at δ 4.27 and δ 4.28 respectively. The phosphate head group signal at δ 4.02 corresponding to phosphatidylethnolamine^{41,46} however was not detected. Diacylglycerophospholipids (glyceroglycolipids) presented very weak signals at δ 4.05 and is almost not visible in nutrient limited spectra. This shows the flexibility of the microalgae's metabolome to shift from membrane-lipid production to storage lipid and vice versa under different environmental cues.

Literature data show alkenes/fatty alcohols (olefinic protons) and aldehydes producing characteristic signals at δ 5.8– δ 6.8 and δ 8.0– δ 11.0 respectively⁴⁰. These signals are a mixture of signals from fatty hydrocarbons and conjugated/non-conjugated protons of oxylipins of PUFAs such as DHA⁴⁰. Fatty esters and free fatty acids were also detected in the spectra. In regards to their physiological roles, not much is known⁴⁷. Some secondary fatty aldehydes and alkenes from microalgae have been studied for their antimicrobial properties and thus may have been synthesised by microalgae as an ecological defence system^{48,49}.

Multivariate analysis. The supervised PLS-DA plot the variation that occurs by separating between groups observed and segregation of the studied group. Both aqueous and chloroform sample showed similar pattern, in which clear cluster were observed between three treatment (Fig. 5). Clustering among groups of treatments was more apparent in the chloroform extracts of nitrate (Fig. 5b; $R^2X = 0.896$). Lower R^2X values in aqueous extracts of nitrate treatment (Fig. 5a; $R^2X = 0.76$), signifies a partial diversion in the degree of response towards different nutrient conditions. High predictability values from the PLS-DA score plots (>0.7) shows high reproducibility of data from treatments.

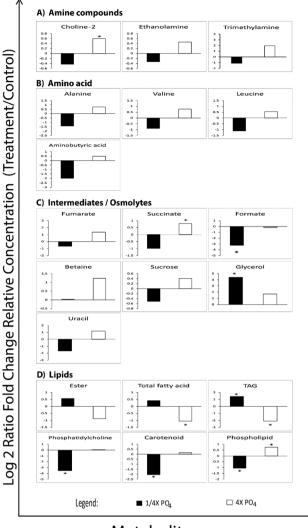




Following PLSDA analysis, the data represent each metabolites were all log2 transformed and tested with student T-test between treatments. Metabolite that show significant different p < 0.05 were then plotted as relative to control into bar charts (Fig. 6). Significant difference between treatments and control were observed in membrane-lipid related metabolites (Choline-2, ethanolamine, betaine), amine compounds and amino acids (trimethylamine, alanine, valine, leucine, aminobutyric acid), TCA cycle intermediates (succinate, fumarate, uracil), energy-related/osmolytes metabolites (succose, formate, glycerol) and lipid fragments (ester, fatty acids, TAG, phosphatidycholine, carotenoid and phospholipid) (Fig. 6). These metabolites are involved in Calvin cycle, TCA cycle, Urea cycle and Fatty acid synthesis cycle (Fig. 7). Fluctuations in aforementioned metabolites prove that different nutrient regimes are able to affect the up-regulation and down-regulation of several important pathways.

In 20 metabolites that detected significant versus to control, apart of lipids fragments, all were increased under 4X nitrate treatment, while mostly reduced its content under limit nitrate treatment. Amino acid synthesis pathways were most severely affected in low nitrate treatments. Amino acid contents (alanine, valine, leucine, aminobutyric acid) were significantly reduced in nitrate limited cultures. This in turn would explain that under nitrate limitation, the rate of protein synthesis was greatly reduced thus affecting the synthesis of chlorophyll proteins due to the low nitrate substrate to support synthesis of amino-containing compounds. Example is the reduction of chlorophyll content (Fig. 1) as a direct consequence of reduced chlorophyll protein under nitrate limitation, as the synthesis of chlorophyll requires 4 molecules of nitrogen. However, there seems to be another route of amino acid synthesis and catabolism adopted by microalgae by proteolysis of ribulose-1,5-bisphosphate carboxylase oxygenase (Rubisco) and conversion of other N-containing materials such as nucleotides and their derivatives to maintain amino acids synthesis⁴⁴. The increment of amino acid and amine compounds were observed in nitrate abundance condition. This is in agreement with treatment of diatom in nitrogen replete treatment, where all amino acids were found increased except histidine and methionine¹⁸. Similar explanation for the succinate and fumarate which may be a direct consequence of increased nitrate availability thus providing the cell enough substrate to support photosynthesis, which increases energy output of the cells. This increased energy-production coincides with increased sucrose levels in 4X NO₃ cultures. Luxury uptake of nitrate increases photosynthetic rate thus diverting the excess carbon into storage components within the cell.

Nitrate is an important content for the production of RNA material in cells⁵⁰. Uracil, an amino containing nucleotide, was observed to be reduced considerably under nitrate limitation but significantly increased under nitrate abundance condition. Limitations on DNA/RNA material synthesis may inhibit cell growth or reduce cell division rates in nitrate limited microalgal cultures. Reduction of osmolytes or osmoprotectants may be a common response to nitrate limited stress⁴⁴. Osmolytes accumulate in the cytoplasm and mainly play an active role in maintaining the ionic balance, help in scavenging ROS, regulating pH and for maintaining the osmotic pressure (osmoregulation) within the microalgae cell^{43,51}. Production of osmolytes are dependent on environmental cues and may be synthesised at any stage of growth⁵¹. Betaine levels were significantly elevated in 4X NO₃. Betaine may play dual roles as a non-phosphorus containing cell wall component or as an osmolyte in the microalgae cell⁵¹.



Metabolites

Figure 6. Fold changes of metabolites plotted against relative concentration of the control. Asterisks represents significant different between each nutrient treatment (*t*-test; P < 0.05).

Glycerophospholipid pathways leading towards formation of phospholipids and pyrimidine biosynthesis pathways were found up-regulated in nitrate abundance. Contradictory, low nutrient condition stimulates phospholipid hydrolysis and the activation of acyl-hydrolase therefore increasing fatty acid content in the cell⁵². The cell switches its mode of synthesis from forming polar lipids to storage lipids. Increased in fatty acid and TAG was associated with degradation of phospholipids and phosphatidycholine, suggesting that TAG was recycled from membrane proteins⁴⁴. Nevertheless, while nitrate limitation has been known to achieve higher lipid to protein ratio within the cell, behaviour of polar metabolites such as carbohydrates, proteins and phosphatases are not popularly known to be affected^{3,53-56}. Alternatively, exogenous bioactive molecules and light intensity can be utilised in growth promotion and lipid content accumulation in microalgae^{57,58}.

Materials and Method

Microalgae cultures and nitrate treatment. Scenedesmus regularis (KS-MC1) was obtained from Institute of Marine Biotechnology (IMB) of Universiti Malaysia Terengganu. The culture and maintenance of Scenedesmus regularis was adopted from³⁴. In nitrate treatment, the concentration for NaNO₃ was altered while the other components remained the same as in Guillard's F2 media. The concentrations for NaNO₃ in control cultures is 8.824×10^{-1} mM, in 4X nitrate treatment, concentration of NaNO₃ was modified to 3.528 mM while in 1/4X treatment, NaNO₃ was modified to 2.205×10^{-1} mM. An inoculum concentration of 1×10^5 cells/ml was transferred to freshly prepared treatment media. The treatment cut off point is on Day 8 of culture. For each treatment, three biological samples were used and the experiment were repeated three time.

Quantification of Chlorophyll Content. Chlorophyll content was extracted by taking 1.5 ml aliquots of culture media containing microalgae from each treatment and centrifuged at 10,000 rpm for 5 minutes to get

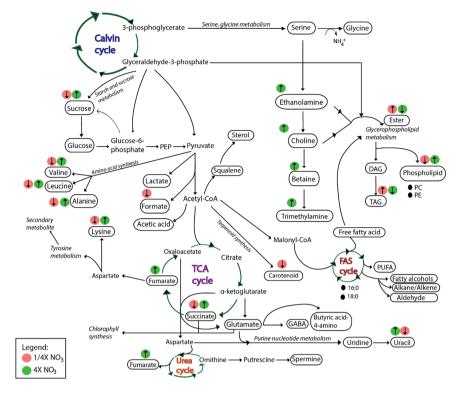


Figure 7. Metabolite regulation pathways involved in *S. regularis* under nitrate treatment. Key for abbreviations: (i) FAS: Fatty acid synthesis; (ii) PC: Phosphatidylcholine; (iii) PE: Phosphatidylethanolamine; (iv) PEP: Phosphoenulpyruvate; (v) TCA: Tricarboxylic cycle; (vi) TAG: Triacylglyceride; (vii) DAG: Diacylglyceride; (viii) GABA: γ-aminobutyric acid.

 $200 \,\mu$ l of supernatant. Pellet was rehydrated with 1 ml of methanol (Merk, ACS grade) and left to shake at 300 rpm on heating plate at 60 °C for 5 minutes. Experiments were carried out under dark conditions. Tubes were then removed from the heating plates and left to stand in the dark for 24 hours at 4 °C. The tubes were subsequently centrifuged for 5 minutes at 10 000 rpm and the supernatant were collected. Chlorophyll content was determined using a spectrophotometer and measured the absorbance at wavelength 665 and 652 nm⁵⁹. Chlorophyll content calculated in µg/ml was determined according to Taylor and Fletcher⁶⁰ formula as follows:

Chlorophyll a: $c_a(\mu g/ml) = 16.72 A_{665.2} - 9.16 A_{652.4}$

Chlorophyll b: $c_a(\mu g/ml) = 34.09 A_{652.4} - 15.28 A_{665.2}$

Total carotenoid: $c_{(x+c)}(\mu g/ml) = (1000 A_{470} - 1.63 c_a - 104.96c_b)/221$

Biomass yield. The wet biomass were harvested from 1.2 L of liquid culture then frozen overnight at -80 °C before subjected to freeze drying (LABCONCO) at -80 °C until constant weight were obtained. The biomass yield was expressed as difference between final dried weight minus initial fresh weight.

SEM preparation and analysis. Fresh microalgae samples in 50 ml was collected by centrifuged and the pellet was underwent general fixation overnight at room temperature using 2.5% glutaraldehyde in 0.1 M sodium cacodyle. Solution was then discarded and pellet rinsed with 0.1 M sodium cocadylate buffer (pH 7.2). Pellet was then submerged in 1% osmiumtetraoxide prepared in 0.1 M sodium cocadylate buffer and left to stand for 4 hours. The pellet was then rinsed with buffer and underwent a series of dehydration steps using various concentrations of ethanol. Pellet was then mounted on specimen stub and left to dry in a fume hood with the aid of hexamethyldisilazane (HMDS – Merk, ACS grade). Specimen stub was then coated with gold using auto fine coater (INOS, UMT).

Extraction procedure and NMR spectra acquisition. A total of 0.215 g of freeze-dried algal biomass was used for (homogenizer) extraction³⁴. Extraction solvents were prepared by adding Chloroform (99% absolute chloroform, ACS grade) with aqueous solvent that prior mixed from 99% absolute methanol (ACS, ISO, Reag grade) and pure distilled water into 1:1 ratio. Two layers of supernatant formed following each extraction and each layer was collected separately and dried up in freeze drier. The aqueous phase of extract was re-suspended in 600 μ l of phosphate buffer (0.1 M, pH 7) containing 10% D₂O and the chloroform phase extract was re-suspended

in 600 μ l of chloroform –d 99.8% containing 0.01% of sodium tetramethysilane (TMS). Both sample were then subjected to 10 min of 10,000 rpm centrifugation before transferred into 5 mm NMR tube³⁴.

NMR analysis and data reduction. Analysis and further furnishing of NMR peaks (baseline correction, phasing, solvent peak removal) were carried out in Mnova (Ver 9.0 – Mestrelab) and Topspin (Ver 3.2 – Bruker). TMS (δ 0.0) was used as the reference point in chloroform spectra while alanine (δ 1.48; doublet) was used as reference peaks in aqueous spectra. The interfering signal of suppressed H₂O resonance and CHCl₃ in aqueous and chloroform spectra was removed. NMR peaks were then identified based on their integration, chemical shifts and peak multiplicity.

Multi-variant analysis. All variables were log-centred and scaled to unit variance before proceeding with supervised and unsupervised multivariated analysis, namely Principal Component Analysis (PCA) and Partial List Squares Discriminant Analysis (PLS-DA) using SIMCA (SIMCA-P + 1.3 – Umetrics) software. The validity of the models was assessed by statistical parameters R2 (correlation coefficient) and Q2 (cross-validation correlation coefficient). Average of each variable were first Log2 transformed and Two tailed student's T-tests were carried out on metabolites to determined statistical differences between input variable and cut out point at p < 0.05.

Conclusion

Cell cultures respond to nutrient fluctuations through a myriad of strategies. Under N deficiency, production of photosynthetic pigments is severely stunted. NO₃ addition lead to chlorophyll pigments and amino acid accumulation while NO₃ limitation had very low amino acid content but was very effective at increasing neutral lipid content. Multi-variate analysis showed that more changes were observable in non-polar extracts when microalgae are exposed to nutrient deficiency.

A basic fingerprint of the biochemical changes using NMR technology coupled with multi-variate analysis proved the capability of NMR as a time-saving and efficient data mining analytical tool as biochemical changes in both polar and non-polar could be captured simultaneously. More studies need to be carried out in order to determine the correct proportions of nutrient content in which to stimulate production of favourable metabolites without compromising productivity. Metabolite activity are important indicators in pathway alterations which would come in handy when determining biomarkers or designing nutrient regimes for economical output.

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Acknowledgements

The authors acknowledge the financial support by the Ministry of Higher Education and Universiti Malaysia Terengganu for the conduct of the research under the Fundamental Research Grant Scheme (Vot 59264).

Author Contributions

N.L., T.S. and A.A. designed the study; K.Y. performed and collected the data; Y.L., S.S. and K.Y. analysed the data; N.L., K.Y. and S.S. drafted the manuscript. All authors read and approved the final manuscript.

Additional Information

Competing Interests: The authors declare no competing interests.

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