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# Dielectric gels with ultra-high dielectric constant, low elastic modulus, and excellent transparency

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## Abstract

We designed dielectric gels, a new type of polymer-based dielectric material. By using solvents with high dielectric constants, the gels achieve a unique combination of ultra-high dielectric constant, low elastic modulus, and excellent transparency, which are extremely challenging or impossible to realize with traditional polymer dielectrics. The gels exhibit high stretchability (stretch of approximately 10) and low mechanical hysteresis. We demonstrated the use of the dielectric gels by fabricating a bioinspired tunable lens, the focal length of which can be adjusted by varying the applied voltage. We believe that the dielectric gels, as a new type of polymer dielectric, offer new opportunities for soft robotics, sensors, electronics, optics, and biomimetics.

## Introduction

As soft and flexible electrical insulators, polymer dielectrics have enabled diverse modern technologies, including electric power systems<sup>1</sup>, flexible electronics<sup>2</sup>, non-volatile memory devices<sup>3</sup>, electrocaloric cooling<sup>4</sup>, and soft robotics<sup>5–7</sup>, benefiting from their diverse functionality. Deformable polymer dielectrics can express mechanical motions in response to electrical stimulation: when a dielectric membrane is subjected to a voltage across its thickness, the membrane suffers an electrostatic force, the so-called Maxwell stress, and thus squeezes in the thickness direction and expands in the area direction. Such dielectrics behave as actuators, as demonstrated by ferroelectric polymers<sup>8</sup>, liquid crystal

elastomers<sup>9</sup>, electrostrictive polymers<sup>10</sup>, and dielectric elastomers<sup>5</sup>. However, the high voltage needed for actuation and the poor mechanical reliability hinder developments in practical applications. Improving the dielectric constant and lowering the elastic modulus of the polymer dielectrics are efficient ways of lowering the actuating voltage<sup>11,12</sup>. Conductive particles or high- $k$  ceramics have been filled into elastomers to increase their dielectric constants, but these rigid fillers have also dramatically increased the elastic modulus and reduced the extensibility<sup>13–16</sup>. Another approach is to use fluids as fillers; liquid–metal microdroplets have been integrated into elastomer substrates, which increased the dielectric constant by over 400% and avoided the internal compliance mismatch of rigid fillers<sup>17</sup>.

However, polymer dielectrics with optical transmittance functionality are emerging materials with practical significance in next-generation flexible displays and flexible touchscreen panels. By fabricating graphene interlayers to form polymer/graphene/polymer structures, Kim et al. prepared flexible and transparent dielectric films with a high dielectric constant ( $\epsilon$ ) of 51 and a transmittance of ~90%<sup>18</sup>. Park et al. used ultralong metal nanofibers as

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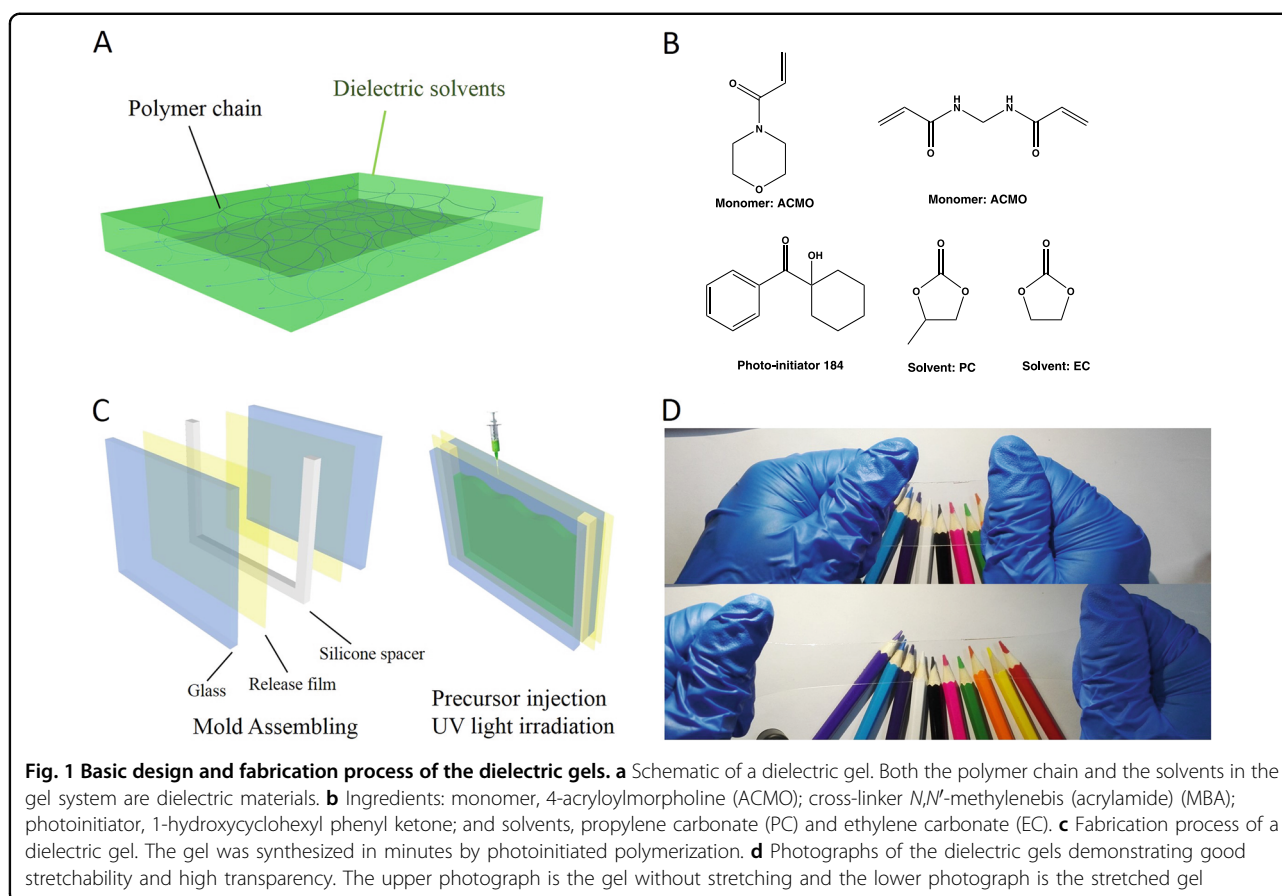
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fillers to increase  $\epsilon$  and obtained flexible and transparent dielectric cellulose nanofiber films ( $\epsilon$  above 9.2 with a high transmittance of 90%)<sup>19</sup>.

Herein, we introduce a new type of polymer dielectric, dielectric gels. The new materials achieve a unique combination of ultra-high  $\epsilon$  (30–50), low elastic modulus (from 20 to 60 KPa), and excellent transparency (~99%). A gel is a polymer composite with a three-dimensional polymer network that contains a large amount of solvent<sup>20,21</sup>. Gels are present as solid-state soft materials. We designed dielectric gels by using solvents with ultra-high  $\epsilon$  and a polymer network that matched well with the solvents. Dielectric gels offer new opportunities for soft robotics, sensors, electronics, optics, and biomimetics.

## Materials and methods

### Synthesis of the dielectric gels

We mixed appropriate amounts of 4-acryloylmorpholine (ACMO, monomer), *N,N'*-methylenebis(acrylamide) (MBA, cross-linker), and 1-hydroxycyclohexyl phenyl ketone (photoinitiator 184) with solvents to form gelation precursor solutions. Unless otherwise stated, the solvent content was fixed at 50 wt%. The precursor solution was poured into a transparent glass mold with a silicone spacer and release films. After irradiation with ultraviolet light

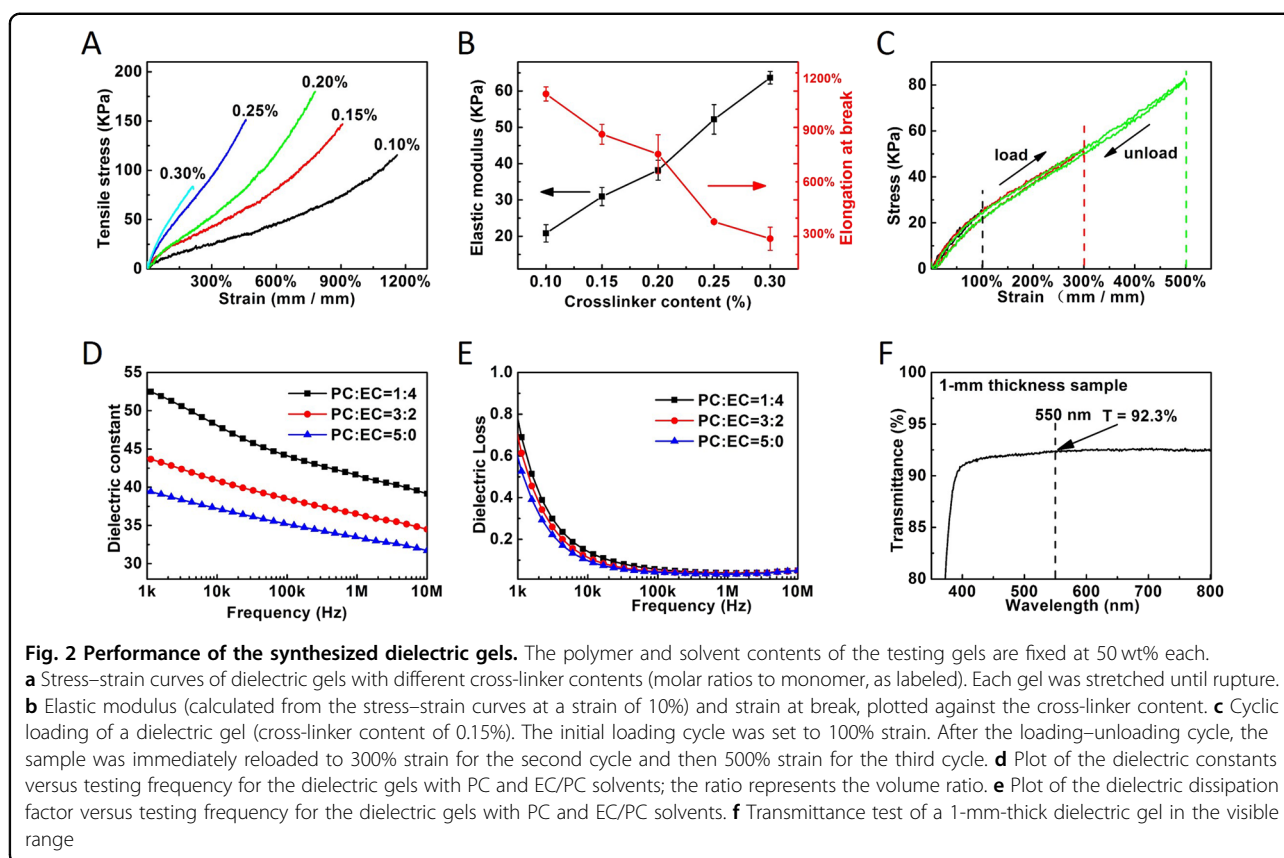
(365 nm, 400 W power) for 5 min, the dielectric gel was formed. We controlled the thickness of the dielectric gel by adjusting the thickness of the silicone spacer. Dielectric gels for electroactive actuation and tunable lens experiments were synthesized by a similar photocuring process. The solvent (PC:EC = 1:4) content was fixed at 20 wt%, the volume ratio of 4-acryloylmorpholine (ACMO) and 2-ethylhexyl acrylate (2-EHA) monomers was fixed at 1:1, and the molar ratios of the photoinitiator and monomer and the cross-linker and monomer were fixed at 1% and 0.1%, respectively.

### Mechanical tests

Mechanical tests were performed on an electronic tensile machine (CMT6503, MTS) with a 500 N load cell. The gels were cut into dumbbell shapes (testing dimensions of 12.0 × 2.0 × 2.0 mm<sup>3</sup>) for the tests.

### Dielectric tests

Dielectric tests were performed on a broadband dielectric/impedance spectrometer (Novocontrol GmbH). The gels were 1-mm thick, the testing copper electrodes were 30 mm in diameter, and the testing  $V_{\text{rms}}$  (Volt root mean square) was set at 1 V. The gel samples were treated without metal sputtering on their surfaces.



### Transparency tests

The transparency tests were performed on a ultraviolet–visible spectrophotometer (UV–vis) spectrophotometer (PE Lambda950, Instrument Analysis Center of Xi’an Jiaotong University). The gels for the tests were 1-mm thick.

### Lens fabrication

The lenses were fabricated from two gel membranes with silicone oil sealed between them. The membranes were equally biaxially pre-stretched by  $\lambda = 2.5$  with an original thickness  $h = 1$  mm.

### Results

We chose propylene carbonate (PC) and ethylene carbonate (EC) as solvents for the dielectric gels (Fig. 1a) because they possess ultra-high  $\epsilon$  (65 for PC and 90 for EC)<sup>22</sup>, good chemical stability, low vapor pressure, and low toxicity<sup>23</sup>. They are widely used as electrolyte solvents in lithium-ion batteries. The melting points of PC and EC are  $-48.8$  °C and  $36.4$  °C, respectively<sup>24</sup>. Although EC is icy at room temperature, it can be mixed with PC to form a homogeneous solution. The mixed solution is liquid at room temperature and has a higher  $\epsilon$  than PC. We mixed appropriate amounts of 4-acryloylmorpholine (ACMO, monomer), *N,N'*-methylenebis (acrylamide) (MBA, cross-

linker), and 1-hydroxycyclohexyl phenyl ketone (photo-initiator 184) with the solvents to form gelation precursor solutions (Fig. 1b). Unless otherwise stated, the solvent content was fixed at 50 wt%. The precursor solution was poured into a transparent glass mold with a silicone spacer and release films. After irradiation with ultraviolet light (365 nm, 400 W power) for 5 min, the dielectric gel was formed (Fig. 1c, d).

The gels were then cut into dumbbell shapes with dimensions  $12.0 \times 2.0 \times 2.0$  mm<sup>3</sup>. Mechanical tests were performed on an electronic tensile machine with a 500 N load cell. The stretching rate was set at  $100$  mm min<sup>-1</sup> during both loading and unloading. The gels possess a low elastic modulus (tens of KPa) and good stretchability (with elongation at break ranging from  $\sim 200$  to  $\sim 1100\%$ ), as shown in Fig. 2a, b. The cross-linker content of a gel greatly affects its mechanical properties: as the cross-linker content increases, the elastic modulus of a gel increases and the elongation at break decreases. A poly(4-acryloylmorpholine) gel with a cross-linker content of 0.15% was chosen for the cyclic loading–unloading tests (Fig. 2c). The initial loading cycle was set with a strain of 100%. After the loading–unloading cycle, the sample was immediately reloaded to 300% strain and then to 500% strain. The gel showed negligible mechanical hysteresis at each strain cycle, and it fully recovered its original length

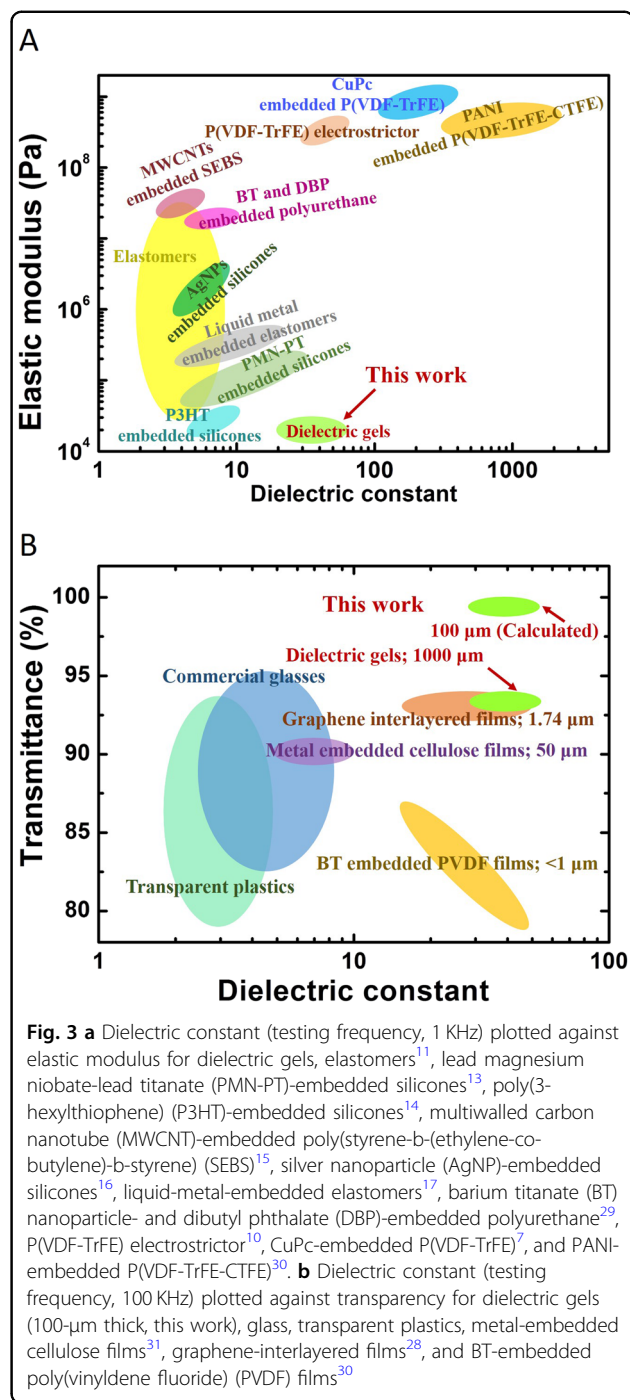
after unloading. A long-term cyclic loading–unloading test was also performed, in which the strain was set to 400%; the 100th cycle curve almost coincided with that of the initial cycle, and the elastic modulus remained unchanged (Figure S1).

Dielectric properties were tested on a broadband dielectric/impedance spectrometer. The testing samples were 1-mm thick, and the testing copper electrodes were 30 mm in diameter. The gel samples were soft enough to be in good contact with the electrodes. The testing  $V_{\text{rms}}$  was set at 1 V. Figure 2d, e shows the dielectric properties of gels with solvents of different EC/PC volumetric ratios. The gels exhibit a high  $\epsilon$  (30–50) over a broad frequency range (1 K–10 MHz). The solvent composition greatly affects the  $\epsilon$  of the gels. A higher EC content results in a higher  $\epsilon$ . As the frequency increases, the  $\epsilon$  decreases slightly, as shown in Fig. 2d. This may be because the switching of the small-molecule dipoles is unable to match the switching of the electric field at high frequencies. The dielectric loss (loss tangent) of the gels was maintained at  $<0.1$  over a frequency range of 20 K–10 MHz. At low frequencies (10–20 KHz), as the frequency decreases, the dielectric loss increases sharply (Figure S2), while the  $\epsilon$  does not change much (Figure S3). We used pure PC as the solvent and studied the dielectric properties of gels with varying polymer contents of 50, 55, 60, and 65%. As the polymer content increases,  $\epsilon$  decreases, and the dielectric loss at low frequency decreases substantially (Figures S4, S5).

The transmittance was measured on an UV–vis spectrophotometer. The gels are highly transparent in the visible range, as shown in Fig. 2f. A sample with a thickness of 1 mm showed a transmittance of 92.3% at 550 nm, corresponding to a transmittance of 99.2% for a 100- $\mu\text{m}$ -thick gel. The transparency was calculated from the Lambert–Beer equation  $\log(T) = KI$ , where  $T$  is the transparency,  $I$  is the thickness of the sample, and  $K$  is a constant.

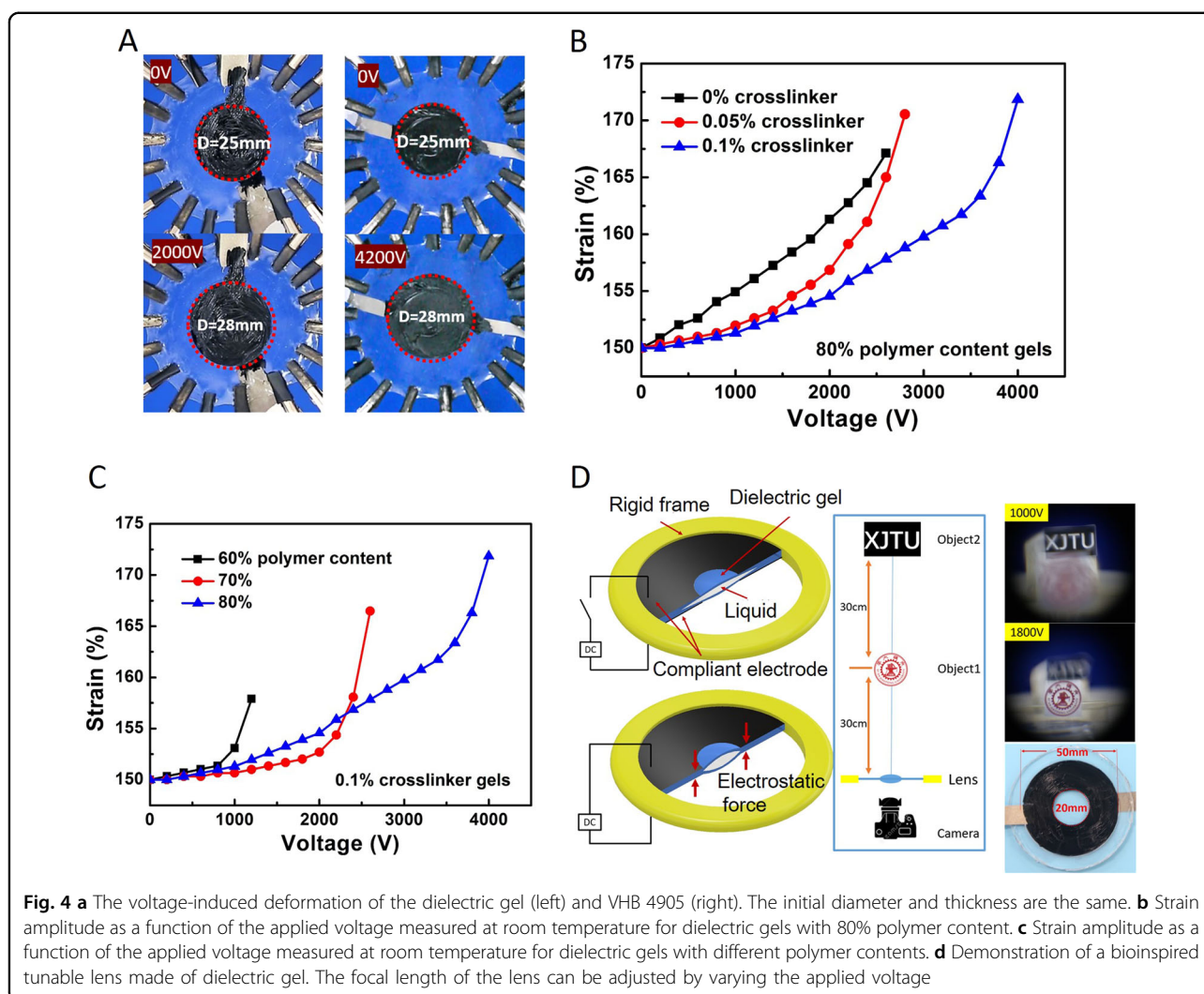
Figure 3a plots the elastic modulus versus  $\epsilon$  for various dielectric elastomers. Our dielectric gels possess a very low elastic modulus and ultra-high  $\epsilon$ . The diagram of transmittance versus  $\epsilon$  for different transparent materials is shown in Fig. 3b. We labeled the thickness of the materials for comparison in Fig. 3b and show the original transparency of the 1-mm-thick dielectric gel. We also show the transparency of our 0.1-mm-thick dielectric gel calculated according to the Lambert–Beer law. The dielectric gels showed very high transparency and high  $\epsilon$ .

We demonstrated the use of the dielectric gel as an electroactive actuator. Figure 4a shows the voltage-induced deformation of the dielectric gel and VHB 4905 (3M Company), the most frequently used material worldwide with a relative  $\epsilon$  of  $\sim 4$  and a shear modulus of a few tens of KPa. A voltage-induced 2200% areal actuation strain has been achieved, and many functional devices have been



**Fig. 3** **a** Dielectric constant (testing frequency, 1 KHz) plotted against elastic modulus for dielectric gels, elastomers<sup>11</sup>, lead magnesium niobate-lead titanate (PMN-PT)-embedded silicones<sup>13</sup>, poly(3-hexylthiophene) (P3HT)-embedded silicones<sup>14</sup>, multiwalled carbon nanotube (MWCNT)-embedded poly(styrene-*b*-(ethylene-co-butylene)-*b*-styrene) (SEBS)<sup>15</sup>, silver nanoparticle (AgNP)-embedded silicones<sup>16</sup>, liquid-metal-embedded elastomers<sup>17</sup>, barium titanate (BT) nanoparticle- and dibutyl phthalate (DBP)-embedded polyurethane<sup>29</sup>, P(VDF-TrFE) electrostrictor<sup>10</sup>, CuPc-embedded P(VDF-TrFE)<sup>7</sup>, and PANI-embedded P(VDF-TrFE-CTFE)<sup>20</sup>. **b** Dielectric constant (testing frequency, 100 KHz) plotted against transparency for dielectric gels (100- $\mu\text{m}$  thick, this work), glass, transparent plastics, metal-embedded cellulose films<sup>31</sup>, graphene-interlayered films<sup>28</sup>, and BT-embedded poly(vinylidene fluoride) (PVDF) films<sup>30</sup>

demonstrated using VHB<sup>25–27</sup>. The dielectric gel and VHB material originally had diameter  $D = 40$  mm and thickness  $h = 0.5$  mm, and both were equally biaxially pre-stretched 1.5 times by 18 clamps. Subsequently, a circular area at the center of the membrane was coated with carbon grease on both sides to act as electrodes. The effective actuation area was actuated from 25 to 28 mm with a linear strain of 12% (Supplementary Movie 1). The voltage needed for the dielectric gel is 2000 V, which is much lower than that for



VHB (4200 V). Figure 4b, c shows the relationship between the electric field and the areal actuation strain. The gels for the electroactuation tests were both equally biaxially pre-stretched 1.5 times by 18 clamps. Subsequently, a circular area at the center of the membrane was coated with carbon grease on both sides to act as electrodes. We used a camera to catch the effective actuation area as the applied voltage increased until breakdown. The ordinate values were the linear strain. In Fig. 4b, we fixed the polymer content of the gels at 80%, and as the cross-linker content increased, the maximum actuation strain increased. At a fixed actuation strain, the needed voltage increases, which may be due to the higher elastic modulus of the gels at higher cross-linker content. Figure 4c shows the electroactuation properties of gels with different polymer contents. At higher polymer content, the maximum actuation strain was higher, and high voltages were difficult to sustain for gels with high liquid content (low polymer content). Figure 4d shows a demonstration of a voltage-driven tunable lens made of dielectric gels, referring to a similar configuration<sup>28</sup>.

Transparent liquid was sandwiched between two membranes composed of the dielectric gels. The physical properties, including the mechanical and dielectric properties, of the dielectric gels used for electroactive actuation tests are shown in Figures S6, S7. When a voltage is applied, the two membranes tend to close and squeeze the liquid in the center, resulting in a change in the focal length of the biomimic lens. One can clearly resolve an object near the lens or an object far away from the lens by varying the applied voltage (Fig. 4d, Supplementary Movie 2). The design of the lens is inspired by human eyes. For human eyes, the ciliaris contracts to change the focal length of the eye. In our design, the dielectric gel acts as the muscle. When the dielectric gel is subjected to a voltage, it contracts to change the focal length of the lens.

## Discussion

In conclusion, by using high  $\epsilon$  solvents, i.e., PC and EC/PC mixtures, we designed and fabricated a series of high-performance dielectric gels that simultaneously

displayed ultra-high  $\epsilon$  (30–50), low elastic modulus (from 20 to 60 KPa), and excellent transparency. The dielectric gels showed high stretchability (stretch of  $\sim 10$ ) and low mechanical hysteresis. We demonstrated that the voltage needed for the same actuation strain is reduced by a half using our dielectric gels, which are electroactive materials, compared with the most commonly used dielectric elastomer VHB. We demonstrated the use of the dielectric gel by fabricating a bioinspired tunable lens, the focal length of which can be adjusted by varying the applied voltage. As a new type of polymer dielectric, we believe that these dielectric gels offer new opportunities for soft robotics, sensors, electronics, optics, and biomimetics.

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#### Authors' contributions

L.S., R.Y., S.L., T.L., and S.D. designed the study and interpreted the results. L.S. developed the protocol for fabrication of the gels and prepared the initial samples. R.Y. designed and performed the electroactive actuation tests and tunable lens tests. T.W. and W.W. contributed to the discussion of the results. L.S., T.L., and S.D. wrote the manuscript. All authors commented on the manuscript.

#### Conflict of interest

The authors declare that they have no conflict of interest.

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