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Anthropogenic radionuclides in atmospheric air over Switzerland during the last few decades

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The atmospheric nuclear testing in the 1950s and early 1960s and the burn-up of the SNAP-9A satellite led to large injections of radionuclides into the stratosphere. It is generally accepted that current levels of plutonium and caesium radionuclides in the stratosphere are negligible. Here we show that those radionuclides are present in the stratosphere at higher levels than in the troposphere. The lower content in the troposphere reveals that dry and wet deposition efficiently removes radionuclides within a period of a few weeks to months. Since the stratosphere is thermally stratified and separated from the troposphere by the tropopause, radioactive aerosols remain longer. We estimate a mean residence time for plutonium and caesium radionuclides in the stratosphere of 2.5-5 years. Our results also reveal that strong volcanic eruptions like Eyjafjallajökull in 2010 have an important role in redistributing anthropogenic radionuclides from the stratosphere to the troposphere.

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nthropogenic radionuclides have been injected into the atmosphere by nuclear weapon tests (NWT) in the fifties, sixties and seventies, the burn-up of the SNAP-9A satellite¹, and accidents in nuclear power plants such as Chernobyl (1986) and Fukushima (2011). Radioactive debris became rapidly attached to ambient aerosol particles, which determined the mechanisms governing their atmospheric transport. In the troposphere, dry and wet deposition removes most contaminants within a few weeks to months. In the stratosphere, mean residence time for radioactive debris is longer due to the thermal stratification and separation from the troposphere by the tropopause and the smaller size distribution of the aerosols². It is generally accepted that the current level of plutonium (Pu) in the stratosphere is negligible and that the tropospheric level is mainly controlled by resuspension of Pubearing particles deposited on land³⁻⁹.

There are, however, limited available data on radionuclide transport and distribution in the stratosphere, the size distribution of the radioactive particles and its interaction with natural aerosols. Most information comes from early studies in the sixties and seventies on anthropogenic radionuclides in the atmosphere^{10–18}. In these studies, tropospheric and stratospheric aerosols were sampled with the aid of aircraft and balloons. Because of the significant decrease in the atmospheric level of most radionuclides, as a consequence of the introduction of the Nuclear Test Ban Treaty (NTBT) in 1963, most laboratories abandoned stratospheric surveillance in the early eighties. The collected radionuclide data were nonetheless very useful for investigating atmospheric processes¹⁹. Results show that the

transfer of radioactive debris from the stratosphere to the troposphere takes place mainly at the tropopause breaks, with a maximum in spring and a minimum in autumn 13,16,20

Radioactive particles in the size range of 1–10 µm settle rapidly out of the stratosphere with mean residence times of the order of weeks to months^{2,12,21}. However, the bulk of stratospheric radioactivity was associated with particles below a few tenths of a micrometre in radius ($<0.1\,\mu m$) and persisted in the stratosphere for years^{2,20}. Radioactive particles with radius below 0.02 µm were typically found above 27 km, and particles of larger radius were found between the tropopause and 21 km. Mean residence times of the order of 1-4 years were reported for radioactive particle aerosols in the stratosphere^{2,13,15,16,20}. These estimations were based on the rates of change of the stratospheric burden over short timescales (2-4 years), and numerous important factors/parameters were not taken into account. We could mention, for example, transport processes in the stratosphere (for example, horizontal mixing, interhemispheric transfer), the longer residence times for small particles (older tail of a residence time distribution) and the effect of stratospheric mixing of debris from different NWTs.

Other studies on Pu radionuclides and ¹³⁷Cs in the atmosphere are based mainly on monitoring their long-term evolution in surface aerosols and atmospheric depositions. These studies have consistently suggested that Pu activities in the stratospheric compartment may have decayed with mean residence times of 1.0-1.7 years²²⁻²⁵. The general mixing of stratospheric with tropospheric air is the main mechanism of NWT deposition. However, Pu activity in the stratosphere is not controlled by the

Sampling date	Altitude of sample (km)	Altitude of tropopause (km)	^{239,240} Pu (μBq m ^{– 3})	²³⁸ Pu/ ^{239,240} Pu	²⁴¹ Am/ ^{239,240} Pu	²⁴⁰ Pu/ ²³⁹ Pu
08.01.1973	12.1	11.6	15.3 ± 0.9	0.20 ± 0.02	0.47 ± 0.04	
05.04.1973	12.6	12.3	21 ± 1	0.14 ± 0.02	0.68 ± 0.06	
13.08.1973	13.0	12.5	14.3 ± 0.8	0.15 ± 0.02	0.36 ± 0.04	
11.03.1974	11.2	10.2	14.7 ± 0.8	0.04 ± 0.01	<dl< td=""><td></td></dl<>	
26.06.1974	11.2	10.2	93 ± 4	0.017 ± 0.002	0.54 ± 0.13	
02.09.1974	13.4	12.4	21.4 ± 0.9	0.046 ± 0.006	0.34 ± 0.04	
21.09.1976	12.1	11.6	0.68 ± 0.12	0.11 ± 0.06	1.2 ± 0.3	
29.11.1976	13.9	12.9	11.6 ± 0.72	0.034 ± 0.008	0.29 ± 0.04	
08.03.1977	13.0	12.0	97.4 ± 4.4	0.026 ± 0.002	0.36 ± 0.03	
14.09.1977	13.4	12.1	27.4 ± 0.1	0.010 ± 0.004	0.39 ± 0.04	
26.06.1978	10.1	9.1	128 ± 17	0.009 ± 0.006	0.39 ± 0.06	
10.02.1986	14.2	13.6	0.22 ± 0.06	0.045 ± 0.07	2.6 ± 1.0	
05.06.1986	NA	NA	0.24 ± 0.05	0.04 ± 0.04	<dl< td=""><td></td></dl<>	
13.10.2004	12.2	11.3	0.5 ± 0.1	0.08 ± 0.07	0.6 ± 0.3	
07.11.2005	13.1	12.2	0.32 ± 0.06	0.02 ± 0.03	0.4 ± 0.3	
24.11.2006	13.6	12.6	0.12 ± 0.04	0.07 ± 0.05	<dl< td=""><td></td></dl<>	
14.05.2007	10.8	11.7	3.9 ± 0.4	0.15 ± 0.04	0.4 ± 0.1	
02.07.2007	10.7	11.6	5.6 ± 0.4	0.11 ± 0.02	0.34 ± 0.06	
28.10.2008	12.5	11.5	9.7 ± 0.5	0.15 ± 0.01	0.43 ± 0.04	
28.10.2008	12.5	11.5	0.35 ± 0.06	0.11 ± 0.06	<dl< td=""><td></td></dl<>	
15.10.2009	10.8	9.8	0.08 ± 0.03	0.05 ± 0.02	<dl< td=""><td></td></dl<>	
20.04.2010	1.0	10.0	24 ± 2	0.13 ± 0.03	0.51 ± 0.08	0.181 ± 0.00
20.04.2010	1.0	10.0	15 ± 2	0.14 ± 0.04	0.56 ± 0.18	0.183 ± 0.00
20.04.2010	2.5	10.0	9 ± 2	0.17 ± 0.07	0.55 ± 0.24	0.178 ± 0.02
20.04.2010	3.0	10.0	8 ± 1	0.12 ± 0.06	0.51 ± 0.28	0.174 ± 0.02
21.04.2010	1.0	10.0	0.5 ± 0.2	0.21 ± 0.23	1.7 ± 1.0	0.132 ± 0.03
21.04.2010	3.0	10.0	0.6 ± 0.3	< DL	0.58 ± 0.5	0.140 ± 0.018
15.09.2010	3.0	13.4	1.3 ± 0.1	0.15 ± 0.03	0.5 ± 0.1	
30.03.2011	7.9	10.6	0.15 ± 0.03	0.04 ± 0.04	0.5 ± 0.2	
30.03.2011	5.2	10.6	0.19 ± 0.03	0.09 ± 0.05	0.8 ± 0.3	

DL, detection limit; NA, not available.
The ²³⁸Pu/^{239,240}Pu and ²⁴¹Am/^{239,240}Pu activity ratios and the ²⁴⁰Pu/²³⁹Pu isotope ratios in the samples are also given (uncertainties at 95% level of confidence). The altitude of the sample and the altitude of the tropopause during sampling and the volume of air sampled are also described

same processes as in the troposphere. Several studies have shown that Pu in ground-level air is strongly influenced by resuspension of Pu in soil particles^{6,7,26–29}. Hence, the estimation of a mean residence time of Pu in the stratosphere by this approach may have been biased to lower values.

High-altitude aerosols (from 8 to 13 km above sea level) have been collected periodically since 1970 under the framework of the environmental radioactivity survey of Switzerland with the aid of aircraft^{30,31}. Elevated concentrations of anthropogenic radionuclides were detected in tropospheric aerosol samples after the passage over Switzerland of the ash plume of the Eyjafjallajökull volcano eruption in 2010. Activities of Pu and ¹³⁷Cs were, for example, up to three orders of magnitude higher than the levels reported for ground-level aerosols³⁰. These were surprising results and at first glance contradicted previous studies on radionuclides in the atmosphere.

The purpose of this work is to measure Pu, ²⁴¹Am and ¹³⁷Cs in air filters from different layers of the troposphere (ground level to upper troposphere) and the lower level of the stratosphere with the aim to determine the residence time of both radionuclides in the atmosphere. We use a ²⁴¹Am/²⁴¹Pu age-dating model to determine the time of the introduction of Pu into the atmosphere. In addition, we study the effect of the volcanic eruption of the Eyjafjallajökull volcano of 2010 to reveal the potential role of volcanic eruptions on the redistribution of radioactive contamination present in the atmosphere.

Results

Anthropogenic radionuclides in the atmosphere. The activities of ^{239,240}Pu in stratospheric and tropospheric aerosols over Switzerland are displayed in Table 1. Graphical representation of ^{239,240}Pu and ¹³⁷Cs in stratospheric and tropospheric aerosols as a function of the year of collection over Switzerland are shown in Figs 1 and 2. ^{239,240}Pu activities in stratospheric aerosols of Switzerland in the seventies were comparable to the activities reported by the Stardust, Airstream and Ashcan high-altitude sampling programmes of the Environmental Measurements

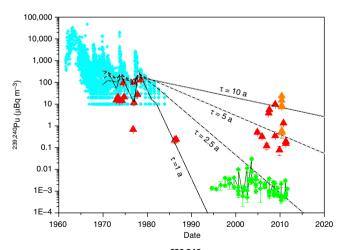


Figure 1 | Activity concentration of ^{239,240}Pu in the atmosphere.

 239,240 Pu in stratospheric aerosols from Switzerland (red traingles) and in aerosols samples from the high-altitude aerosol monitoring programmes of the Environmental Measurements Laboratory (light blue rhombus): stardust, airstream and ashcan^{11,14,17,18}. 239,240 Pu in the Eyjafjallajökull volcano ash plume (orange triangles) and at ground-level air in Switzerland (green rhombus) are also shown. The lines represent the Pu distribution obtained by the exponential model for different mean residence times (τ = 1, 2.5, 5 and 10 a). Uncertainties are expressed at 95% level of confidence (k = 2).

Laboratory (Fig. 1)^{11,14,17,18}. ^{239,240}Pu and ¹³⁷Cs in the stratosphere have since steadily decreased, and current levels are down to two orders of magnitude lower than the ones measured in 1974 (Fig. 1). ^{239,240}Pu activities in stratospheric aerosols are nonetheless up to five orders of magnitude higher than in ground-level aerosols at present days (Fig. 1). ¹³⁷Cs behaved similarly, with high-altitude aerosols having up to three orders of magnitude higher activities than ground-level aerosols (Fig. 2).

Activity and isotopic ratios of radionuclides. We used the ²⁴¹Am/²⁴¹Pu age-dating method³² on selected aerosol samples to corroborate the time of the introduction of Pu in the atmosphere. The calculated dates showed consistently that the Pu contamination occurred between 1964 and 1982 (Table 2). Activity ratios, recalculated to the date of sampling, of 12–14 for ²⁴¹Pu/²³⁹,2⁴⁰Pu and ²⁴⁰Pu/²³⁹Pu isotope ratios close to 0.18 (Table 2) are also consistent with the ratios reported in atmospheric aerosols after the NTBT in 1963 (ref. 1). This is corroborated by ²⁴¹Am/²³⁹,2⁴⁰Pu ratios close to 0.5. ²³⁸Pu/²³⁹,2⁴⁰Pu ratios higher than expected for NWT fallout (typical ratios of 0.02–0.03) suggest additional Pu sources (Fig. 3). Enriched ²³⁸Pu in the stratosphere was observed during the years after the burn-up of the ²³⁸Pu-powered satellite SNAP-9A at

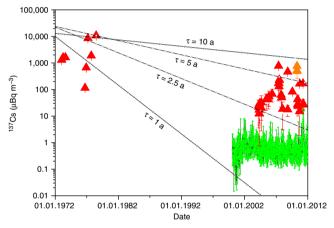


Figure 2 | Activity concentration of ¹³⁷Cs **in the atmosphere.** ¹³⁷Cs in low tropospheric (green rhombus) and stratospheric (red triangles) aerosols of Switzerland. ¹³⁷Cs in the Eyjafjallajökull volcano ash plume is also shown (orange triangles). The lines represent the ¹³⁷Cs distribution obtained by the exponential model for different mean residence times ($\tau = 1, 2.5, 5$ and 10 a). Uncertainties are expressed at 95% level of confidence (k = 2).

Table 2 | Activities of 241 Pu and 241 Pu/ 239,240 Pu activity ratios in high-altitude aerosol samples from Switzerland.

Sampling date	²⁴¹ Pu (μBq m ^{– 3})	²⁴¹ Pu/ ^{239,240} Pu	²⁴¹ Am/ ²⁴¹ Pu age (a)	Contamination date
05.04.1973	259 ± 27	12 ± 2	48 ± 6	1964 ± 6
13.08.1973	173 ± 18	12 ± 1	34 ± 6	1978 ± 6
02.09.1974	309 ± 31	14 ± 2	31 ± 4	1981 ± 4
08.03.1977	1342 ± 150	14 ± 2	30 ± 4	1982 ± 4
14.09.1977	341 ± 42	13 ± 1	31 ± 6	1981 ± 6
14.05.2007	6 ± 1	1.5 ± 0.2	47 ± 5	1965 ± 5
02.07.2007	16 ± 2	2.9 ± 0.3	32 ± 4	1980 ± 4
28.10.2008	31 ± 3	3.2 ± 0.3	34 ± 4	1978 ± 4
15.09.2010	4 ± 1	2.7 ± 0.3	39 ± 6	1973 ± 6

Data are reported for the sampling date (uncertainties at 95% level of confidence). The 241 Am/ 241 Pu ages and the contamination dates are also described.

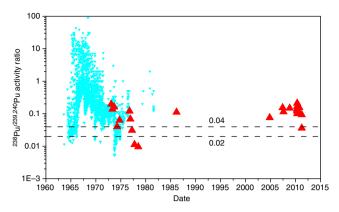


Figure 3 | ²³⁸Pu/^{239,240}Pu activity ratios in stratospheric aerosols.

Activity ratios in stratospheric aerosols from Switzerland (red traingles) and in aerosol samples from the high-altitude aerosol monitoring programmes of the Environmental Measurements Laboratory (light blue triangles): stardust, airstream and ashcan^{11,14,17,18}. The dashed lines at 0.02 and 0.04 represent the current range of ²³⁸Pu/^{239,240}Pu activity ratios in ground-level aerosols and soil samples from Switzerland.

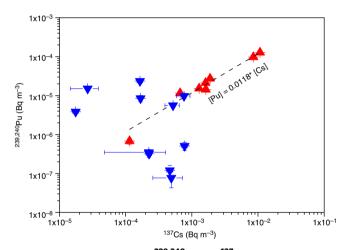


Figure 4 | Correlation between ^{239,240}Pu and ¹³⁷Cs in stratospheric aerosols. Red triangles represent the post-moratorium (after the NTBT in 1963) stratospheric aerosols (1973–1986). Blue triangles represent recent stratospheric samples taken after 2004. The correlation line for the post-moratorium samples is also shown.

about 50 km in the atmosphere in 1964 (ref. 15). This event increased significantly the amount of ²³⁸Pu in the atmosphere (Fig. 3). The relatively high ²³⁸Pu/²³⁹,2⁴⁰Pu activity ratios observed currently indicates that ²³⁸Pu originating mainly from the satellite accident and from NWT with high ²³⁸Pu/²³⁹,2⁴⁰Pu, possibly injected in the upper stratosphere, is still circulating in the stratosphere. A ²³⁹,2⁴⁰Pu/¹³⁷Cs mean activity ratio of 0.0118 was found in post-moratorium (NTBT) stratospheric aerosols (Fig. 4), which is in good agreement with the average ²³⁹,2⁴⁰Pu/¹³⁷Cs activity ratio of 0.012 reported for global fallout^{33,34}. Nevertheless, recent ²³⁹,2⁴⁰Pu/¹³⁷Cs activity ratios in stratospheric aerosols deviate significantly from the above correlation line (Fig. 4). This may be explained by their different chemical/physical properties (for example, volatilities, particle affinities in the atmosphere and so on).

Discussion

The determination of radioelement isotopic ratios and the ²⁴¹Am/²⁴¹Pu-dating method all confirmed that Pu contamination

in the stratosphere was mainly originated from the NWTs fallout and the burn-up of the satellite SNAP-9A. Over the past decades, stratospheric ^{239,240}Pu and ¹³⁷Cs activities showed a large variability that suggests a low mixing rate of stratospheric air masses and/or the input of these radionuclides from upper layers of the stratosphere. The lower variability observed in ground-level air may indicate that equilibrium or steady-state conditions have been reached for the sources (for example, resuspension, transfer from the stratosphere) and sinks (for example, deposition) of these radionuclides in the troposphere. Our results show that significant fractions of radioactive aerosols (possibly small particles < 0.1 µm) remain in the stratosphere for timescales of the order of several decades. Mean residence time of stratospheric air can be obtained from observations of tracer changes in the atmosphere³⁵⁻⁴¹. We used an exponential distribution model (box model) to interpret the trend of Pu and 137Cs in the stratosphere of Switzerland. This choice was justified by the sparse data set of Pu and ¹³⁷Cs in stratospheric aerosols, which is not suitable for estimating annual trends. In addition, the distributions of our data sets fit well the asymmetric shape of the exponential distribution, with a peak for short residence times and a long tail for long residence times. The radionuclide is assumed to be distributed homogeneously over the stratospheric domain. The model does not resolve the spatial distribution of the radionuclide within the stratospheric domain, but it describes the activity of the radionuclide (for example, ^{239,240}Pu, ¹³⁷Cs) inside a box representing the stratosphere over Switzerland. We assumed that horizontal inflow and outflow of the radionuclide into the box are similar. However, vertical downward transport from upper layers of the atmosphere into the stratospheric box and out of the box through the tropopause may have strong effects on the radioisotope activity and therefore on the mean age of the tracer. Best agreement between the measured and modelled Pu activities were obtained for mean residence times of Pu in the stratosphere of 2.5-5 years (Fig. 1). The mean residence times were further confirmed by the 137 Cs data (Fig. 2). These results are in good agreement with previous estimations of the mean residence time of stratospheric air using sampling methods like aircrafts, high-altitude balloons and satellite observations^{35,38,40}. Radioisotopes introduced in the atmosphere as a consequence of the nuclear bomb tests during the fifties and sixties are still useful tracers to study transport processes.

The aerosol samples collected in the low tropopause (altitude: 1-3 km) during the Eyjafjallajökull event showed Pu activities and ²³⁸Pu/^{239,240}Pu activity ratios similar to the ones observed in stratospheric aerosols (Figs 1 and 3, Table 1). This result indicates that the Pu detected in the volcano ash plume may have come from the stratosphere. The Eyjafjallajökull is a sub-glacial volcano covered by an ice cap of about 77 km², which is about 250 m thick at the summit of 1,666 m above sea level. During the eruption of April 2010, thousands of tons of molten rock came into contact with ice creating a huge explosion throwing steam and particles high into the air. This explosive eruption introduced volcanic fine-grained ash and gases (for example, SO₂) up to the lowermost stratosphere 42,43. In the atmosphere, sulphur-bearing gases such as SO₂ transformed into tiny sulphuric acid particles^{2,44}. Aerosol size distributions within ash clouds showed a fine mode (0.1-0.6 µm) associated with sulphuric acid and/or sulphate, and a coarse mode (0.6-35 µm) associated with ash⁴⁵. Numerous studies showed that the Eyjafjallajökull ash particles were strongly depolarizing⁴⁶ and that particles were electrically charged^{47,48}. Harrison *et al.*⁴⁷ proposed radioactive decay as a potential source for this particle depolarization. Charging of aerosols has significant effects on their transport mechanisms like changes in the vertical deposition speeds and particle-particle agglomeration rates⁴⁹. Charged particles have,

for example, increased collision efficiencies and therefore large scavenging coefficients⁵⁰. We hypothesize that the introduction of ash particles and gases into the stratosphere by the Eyjafjallajökull volcano is responsible for an increased scavenging of Pu and ¹³⁷Cs aerosols in the stratosphere and an increased transport into the troposphere by sedimentation.

This hypothesis is further supported by the presence of cirrus-like formations over Europe after the volcano eruption. Volcanic ash particles favour ice nucleation (IN). Seifert et al.⁵¹ showed the existence of very efficient IN during the Eyjafjallajökull volcanic event. Bingemer et al.⁵² also observed a large increase of the IN concentration during this event, with IN surface densities of at least a factor two higher than that in other conditions (for example, air masses affected by mineral dust). Seifert et al.⁵¹ demonstrated the presence of ash-assisted formation of cirrus-like clouds in the troposphere over Leipzig and Maisach during several days after the first observational period (14-17 April 2010). This was further confirmed by the detection of a volcanic ash-induced cirrus cloud at altitudes of 8-11 km over Jülich during the passage of the Eyjafjallajökull ash plumes⁴⁶. The backward trajectory analysis confirmed that the cirrus cloud was originated from ash loadings of air masses near Iceland⁴⁶. The air masses had relative humidities slightly above 100% in which the formation of ice crystals was very favourable. The ash layer then descended with the cirrus cloud. Trickl et al. 43 concluded that a branch of the ash plume was present in the tropopause over Garmisch-Partenkirchen on 19 April, and that some volcanic contribution in the lowermost stratosphere cannot be excluded. Backward trajectories revealed Iceland as the closest source region for the air masses of the troposphere and lowermost stratosphere⁴³. Trickl et al.⁴³ suggested that the cirrus formation occurred between 14 and 16 April 2013 in the upper troposphere over Iceland where favourable conditions for aerosol growth existed (relative humidity exceeded 80% several times). Here we show that the strong volcanic eruption of the Eyjafjallajökull volcano has redistributed anthropogenic radionuclides in the lower atmosphere. Waugh³⁸ emphasized the need for more measurements of 'age tracers'. We believed that NWT radionuclides could be a very good proxy for the study of atmospheric air mass transfer during the last 50 years. In this respect, Pu is the perfect candidate because it has a long physical half-life and a high distribution coefficient on mineral and organic particles, and do not react with atmospheric components under light irradiation.

Methods

Sampling and laboratory methods. High-altitude aerosols were collected six times a year with especially prepared military airplanes. This sampling programme started in the seventies but it was interrupted between the end of the eighties and 2004 (ref. 31). Air volumes between 200 and 3,000 m³ were filtered through cellulose filters ($555 \times 526 \, \text{mm}^2$, type 0048, Whatman). The filters were immediately analysed by gamma spectrometry 31,53 . Ground-level aerosols (at 1 m above the soil surface) were collected with high volume samplers ($500-800 \, \text{m}^3 \, \text{h}^{-1}$) at five different locations of Switzerland. The samples were analysed by gamma spectrometry on a weekly basis.

Since 1994, the cumulative samples of 1 year for each sampling location (annual air volumes > 200,000 m³) was analysed for alpha emitters. After the eruption of the Eyjafjallajökull volcano in 2010, aerosol samples were collected during the passage over Switzerland of the ash plume at the altitude of 1–3 km (ref. 30). In this study, we selected several samples of high-altitude aerosol for analysing Pu and ²⁴¹Am. The air filters were ashed at 550 °C for 48 h before the radiochemical analysis. The radiochemical method combined high-pressure microwave digestion for the dissolution of the ashes with the highly selective extraction chromatographic resins Tetravalent (TEVA, the extractant is a quaternary ammonium salt) and N,N,N',N'-tetra-n-octyldiglycolamide (DGA) (supplied by Triskem International, Bruz, France) to purify Pu and americium⁵⁴. The alpha sources were prepared by electrodeposition on stainless steel discs. High-resolution alpha spectrometry was performed on an alpha spectrometer with passivated implanted planar silicon (PIPS) detectors (Alpha Analyst, Canberra Electronic, France). Selected Pu alpha sources were redissolved in concentrated HNO₃ and HCl as described in ref. 32 for determining either the activity of ²⁴¹Pu by liquid scintillation counting (Quantulus 1220, Wallac) or the concentration of ²⁵⁹Pu and

 240 Pu by sector-field inductively coupled plasma mass spectrometry equipped with a desolvation unit for the reduction of 238 U 1 H molecular interference isobaric to 239 Pu. The inductively coupled plasma mass spectrometry analysis was carried out on an Element 2 at the Federal Office of Public Health according to ref. 55.

Age distribution model. We have chosen an exponential distribution model to interpret our radionuclide data in stratospheric air. The response function of the exponential model (g(t)) is represented by

$$g(t) = T^{-1} \exp^{\frac{(-t)}{T}}$$

where T represents the mean residence time of the distribution and t is the time. The whole residence time distribution, also named age spectra³⁸, is only characterized by T, which represents in our study case the mean residence time of the radionuclide in the stratosphere. The initial activity of the radionuclide in the stratospheric box is set for the year 1970, a few years after the entry into force of the NTBT. The input of the radionuclide into the stratospheric layer in the seventies calculated from the activity values reported by the Environmental Measurements Laboratory 11,14,17,18 . After 1980, any additional input of 239,240 Pu is assumed to be negligible.

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Author contributions

P.S., P.F. and J.A.C.A. organized the research. P.S. and S.E. provided the samples and performed gamma spectrometry analysis. J.A.C.A. and P.F. performed the radiochemical analyses. M.H. was in charge of the ICP-MS analysis. P.S., P.F., F.B. and J.A.C.A. interpreted the data. P.S., P.F. and J.A.C.A. wrote the paper.

Additional information

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